



Deaminative Functionalizations of Unactivated Aliphatic Amines

Reporter: Yuzhen Zhang

Supervisor: Prof. Quan Cai

Fudan University
2023-04-28

● Background

● Deaminative Functionalizations via Katritzky-type Pyridinium Salts

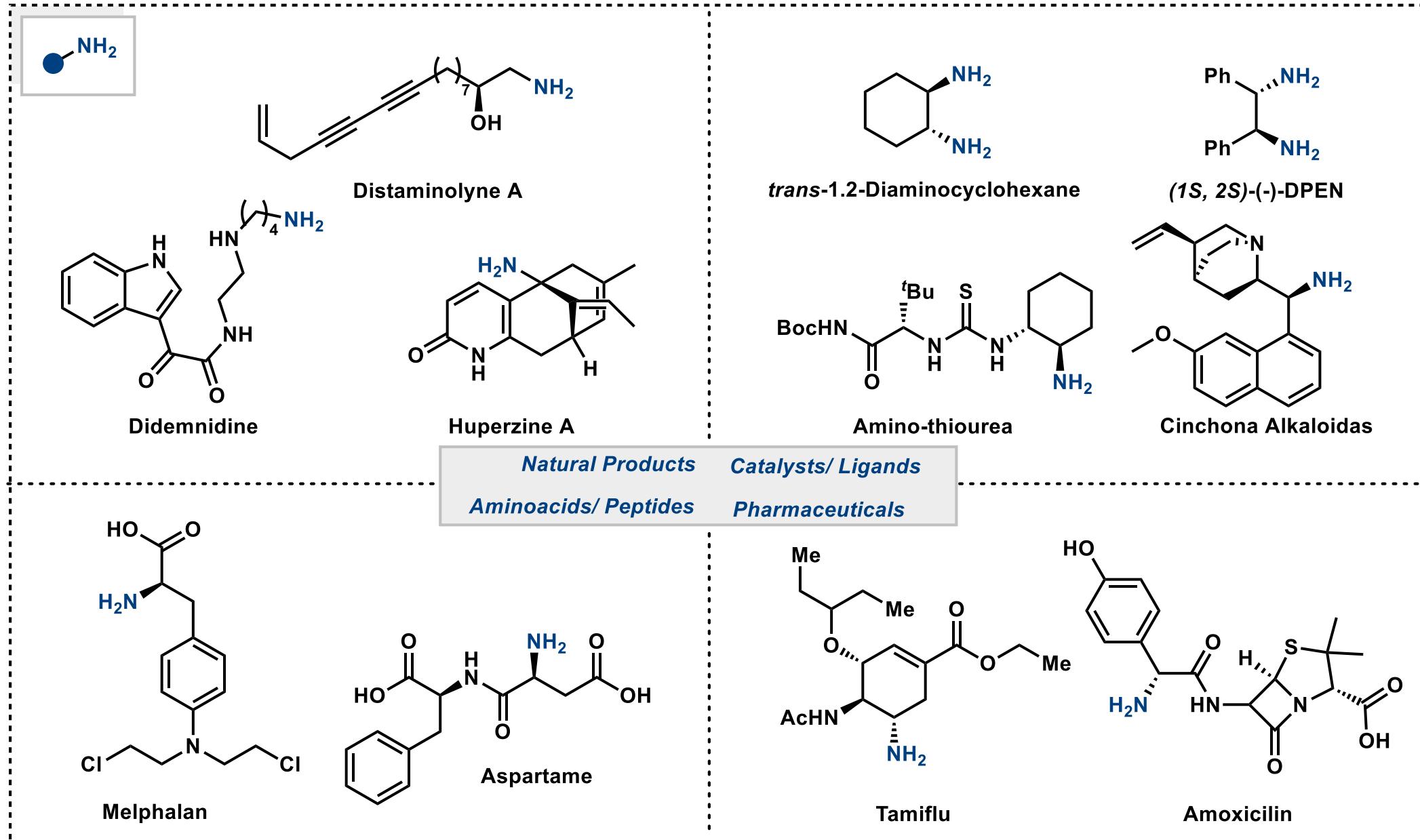
- By Transition Metal Catalysis and Lewis Base Catalysis
- By Photoredox Catalysis and Electrochemical Catalysis

● Deaminative Functionalizations via Electron-rich Imines

● Direct Deaminative Functionalizations

● Summary and Outlook

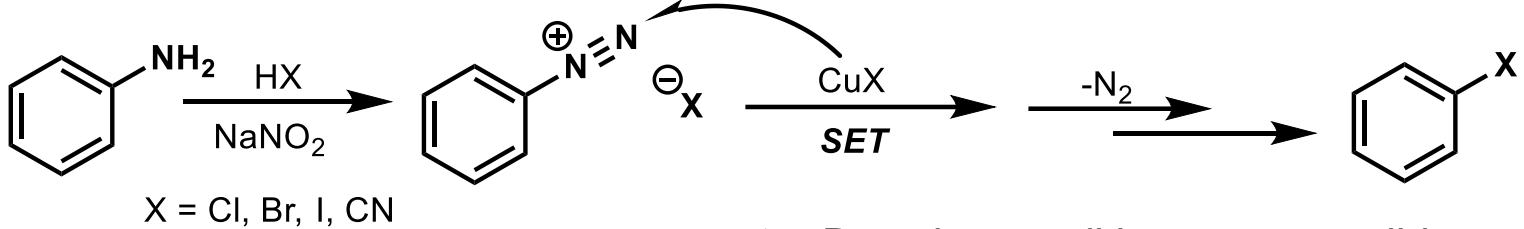
Background



C—H	414 kJ/mol
C—I	240 kJ/mol
C—C	332 kJ/mol
C—N	305 kJ/mol

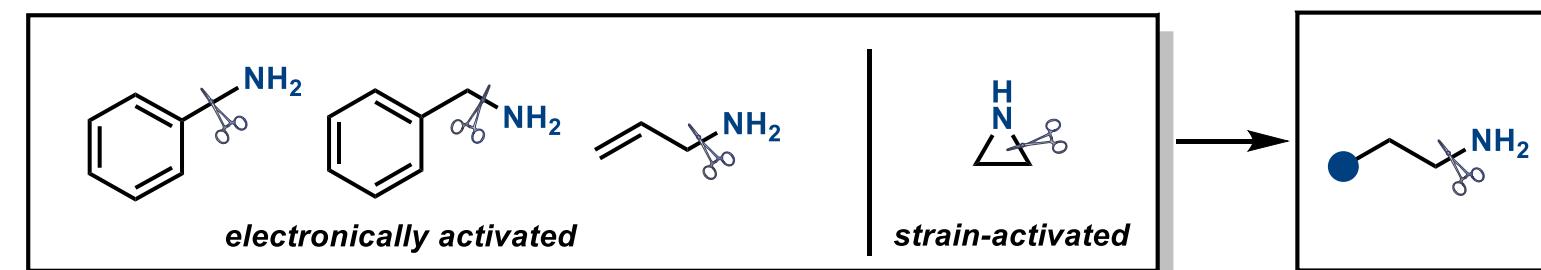
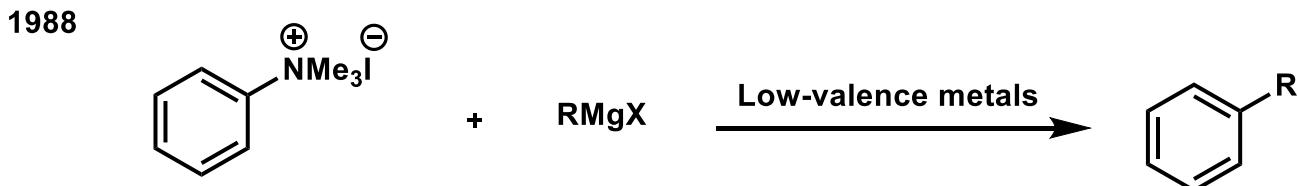


Sandmeyer Reaction
(*sp*² C-N)



- Reaction conditions are not mild
- Strong acids and oxidants
- Instability of diazonium salts

Quaternary ammonium salts
(*sp*² C-N & activated *sp*³ C-N)



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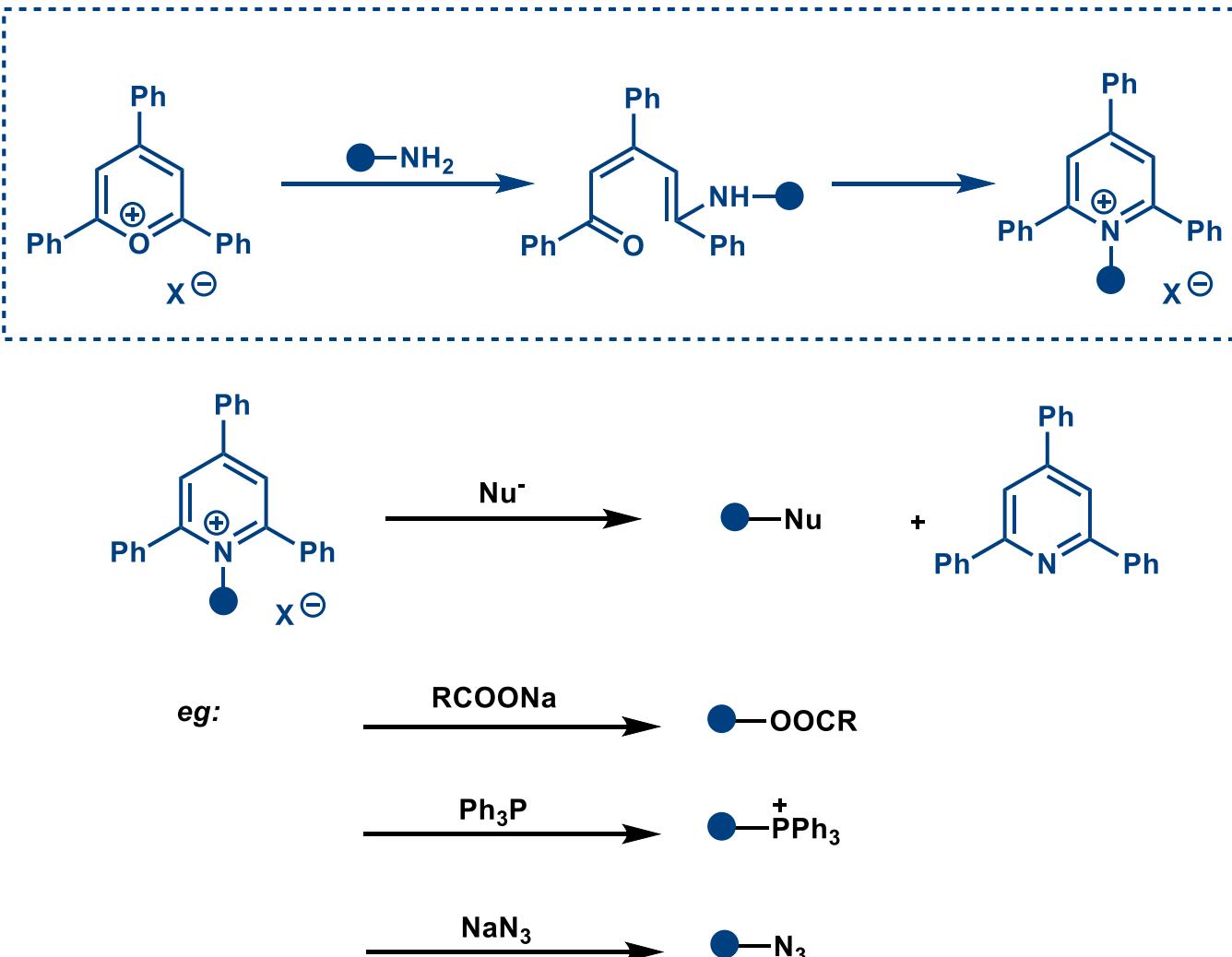
Katritzky-type Pyridinium Salts



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Alan Roy Katritzky
1928-2014



Katritzky, A. R.; Gruntz, U.; Kenny, D. H.; Rezende, M. C.; Sheikh, H. J. *J. Chem. Soc., Perkin Trans. 1* **1979**, 1, 430.

Katritzky, A. R.; Leahy, D.E. *J. Chem. Soc., Perkin Trans. 2* **1985**, 171.

Katritzky, A. R.; Marson, C. M. *Angew. Chem., Int. Ed. Engl.* **1984**, 23, 420.

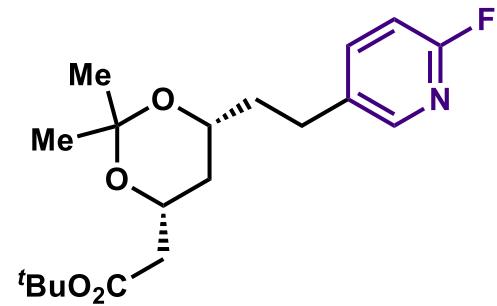
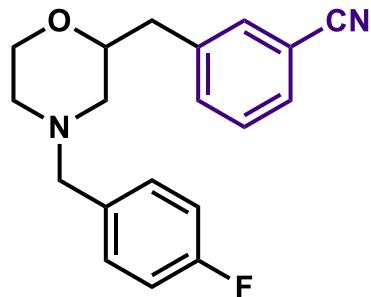
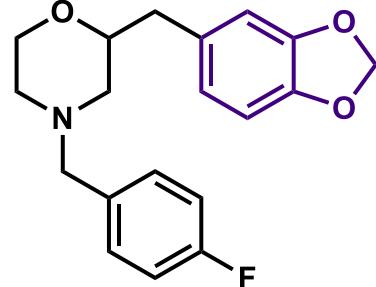
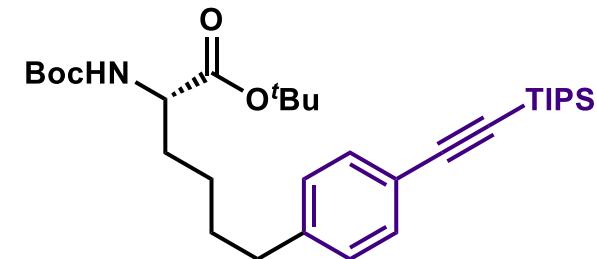
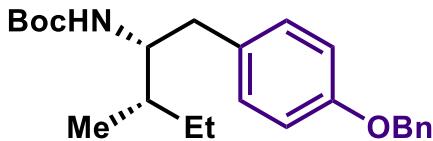
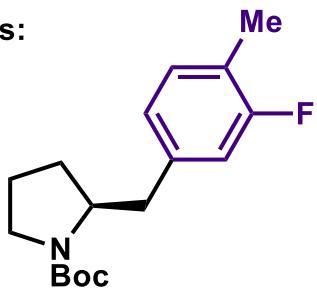
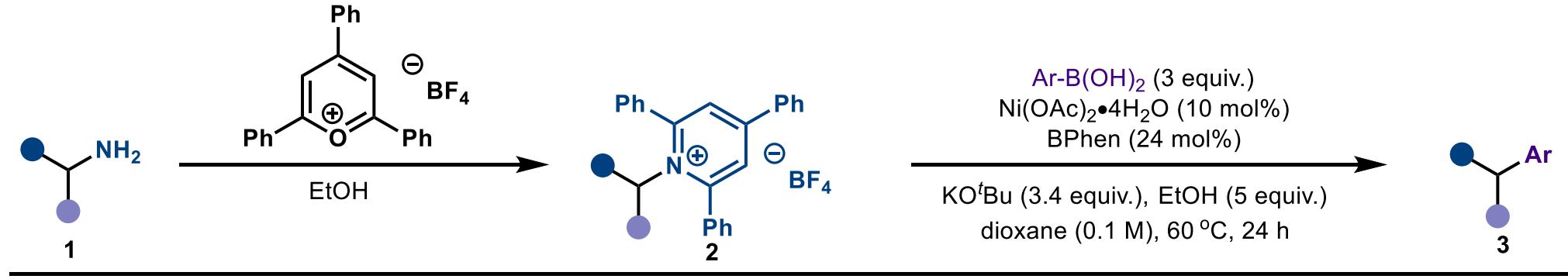
Said, S. A.; Fiksdahl, A. *Tetrahedron: Asymmetry* **2001**, 12, 1947.

Deamination of Pyridinium Salts by TM Catalysis



Arylation

M. P. Watson

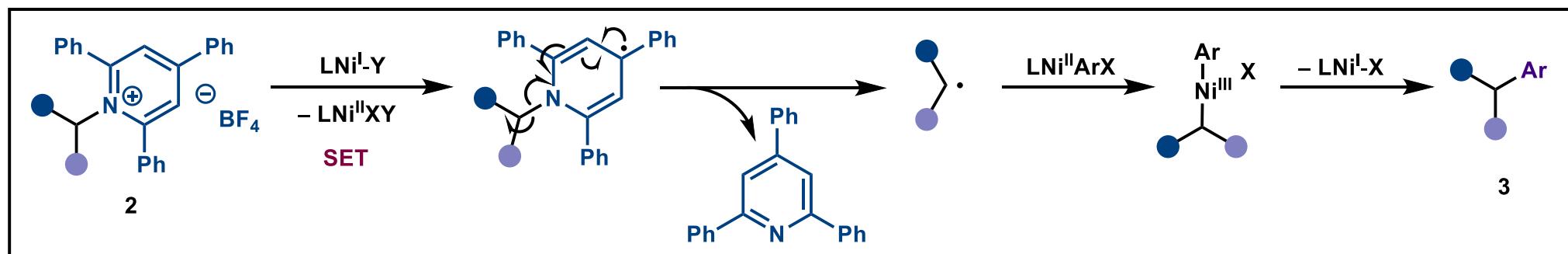
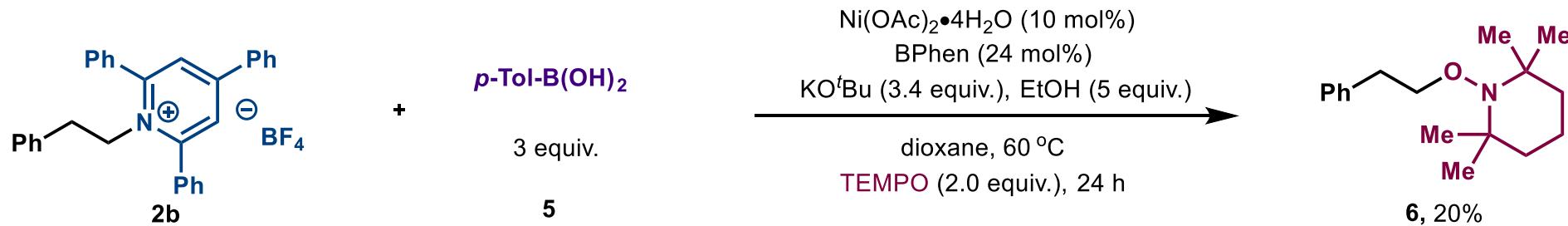
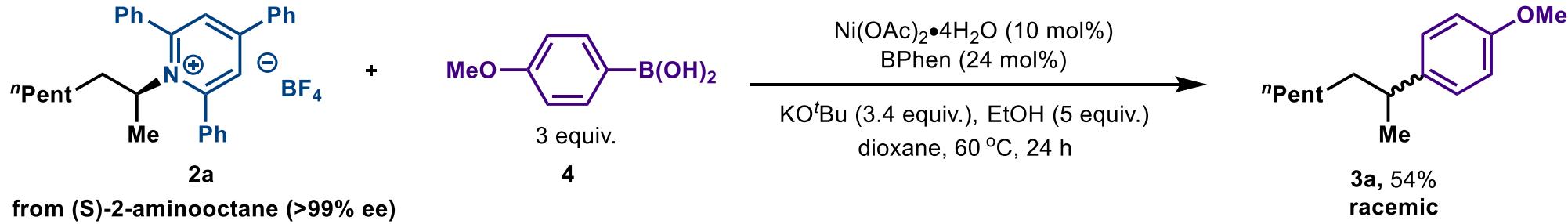


Deamination of Pyridinium Salts by TM Catalysis



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Investigation of the reaction mechanism



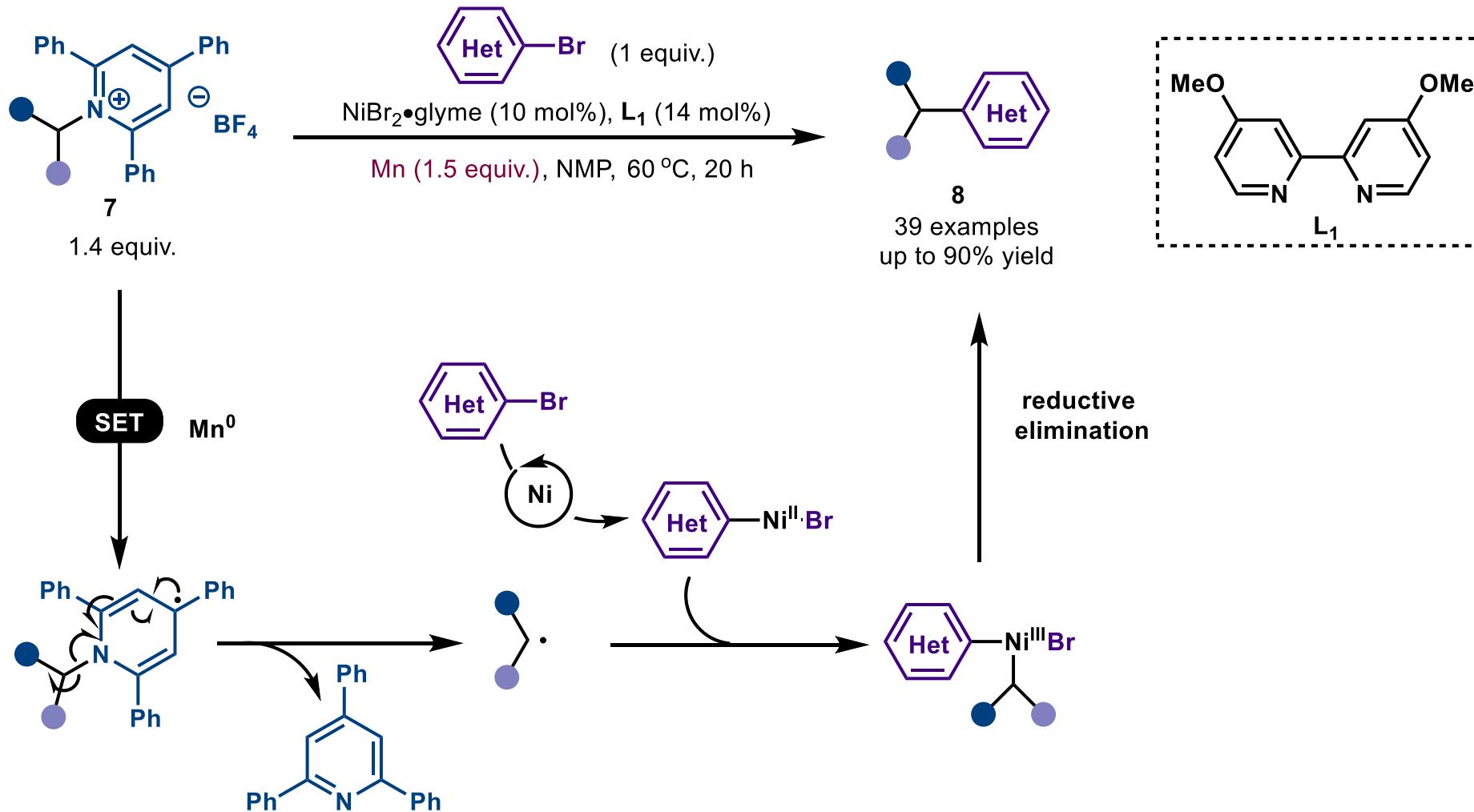
Deamination of Pyridinium Salts by TM Catalysis



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Arylation

R. Martin



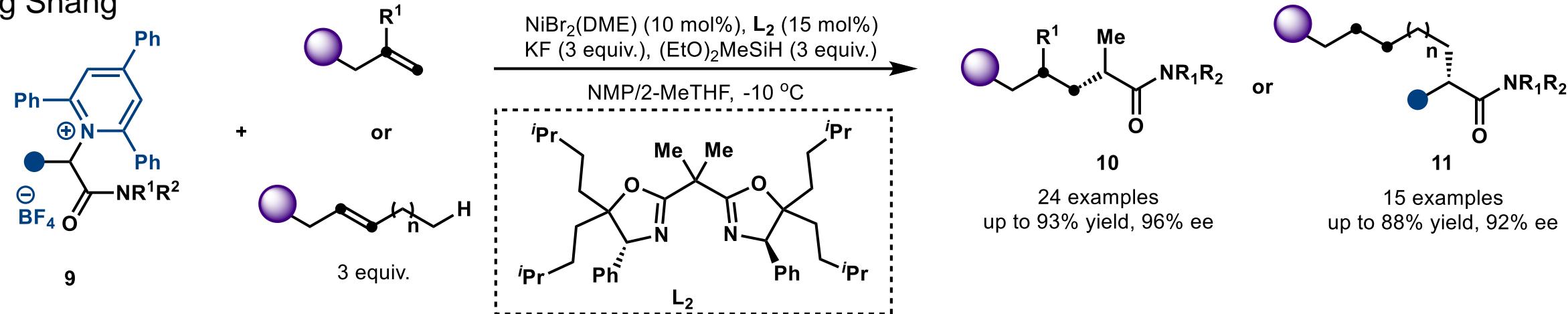
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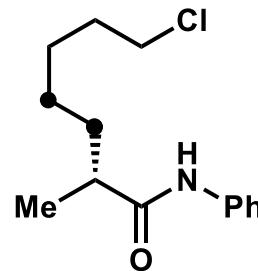
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Dicarbon functionalization

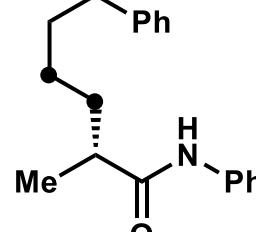
Ming Shang



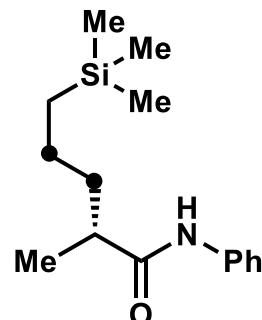
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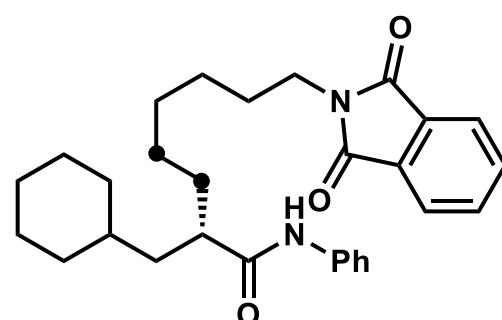
75% yield
92% ee



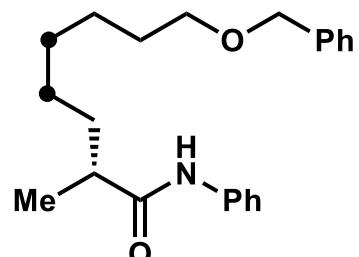
52% yield
92% ee



60% yield
92% ee



83% yield
84% ee



56% yield
92% ee

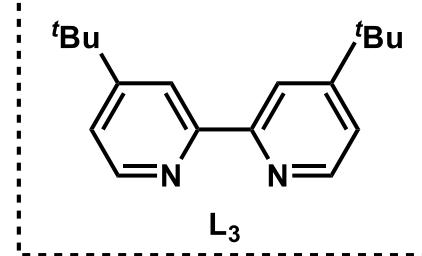
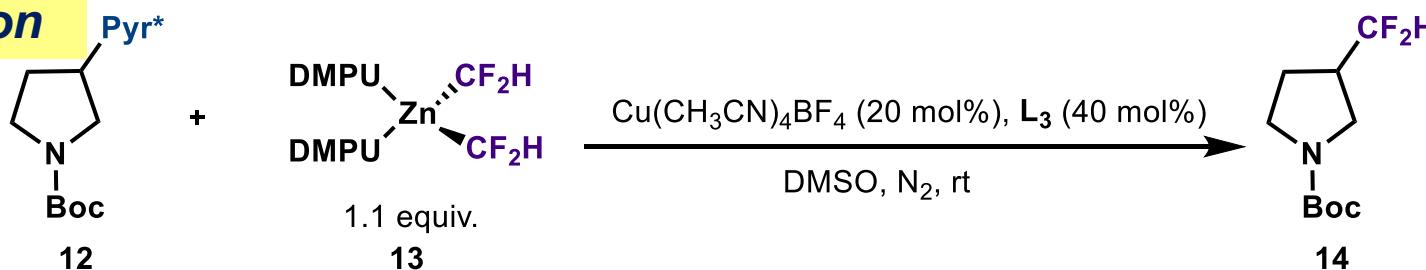
Deamination of Pyridinium Salts by TM Catalysis



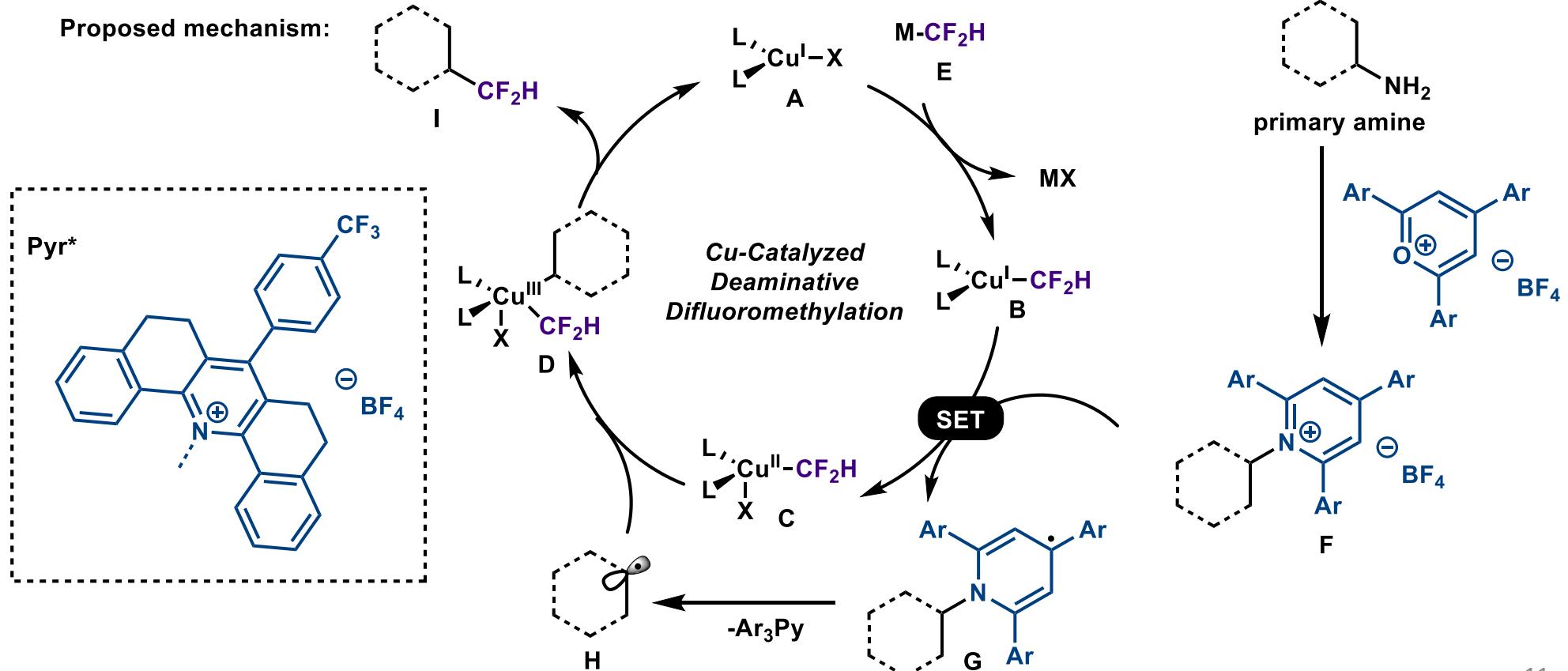
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Difluoromethylation

Liu Wei



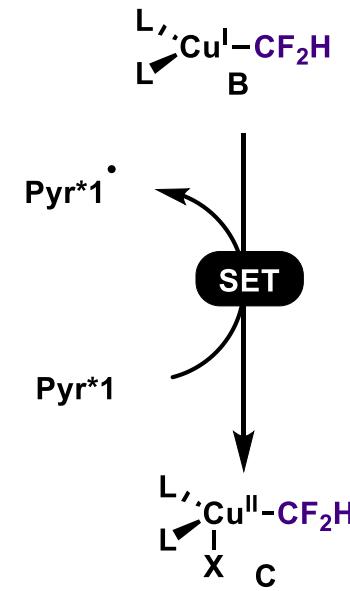
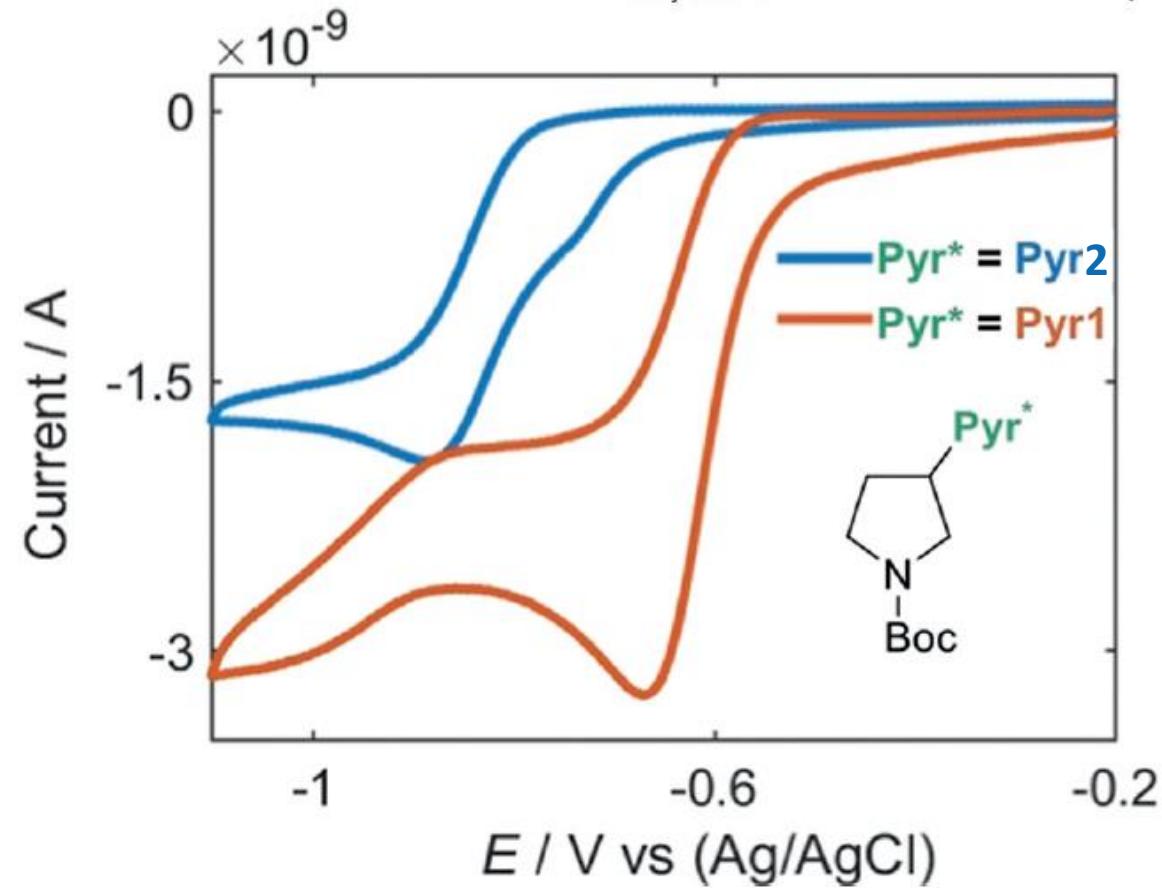
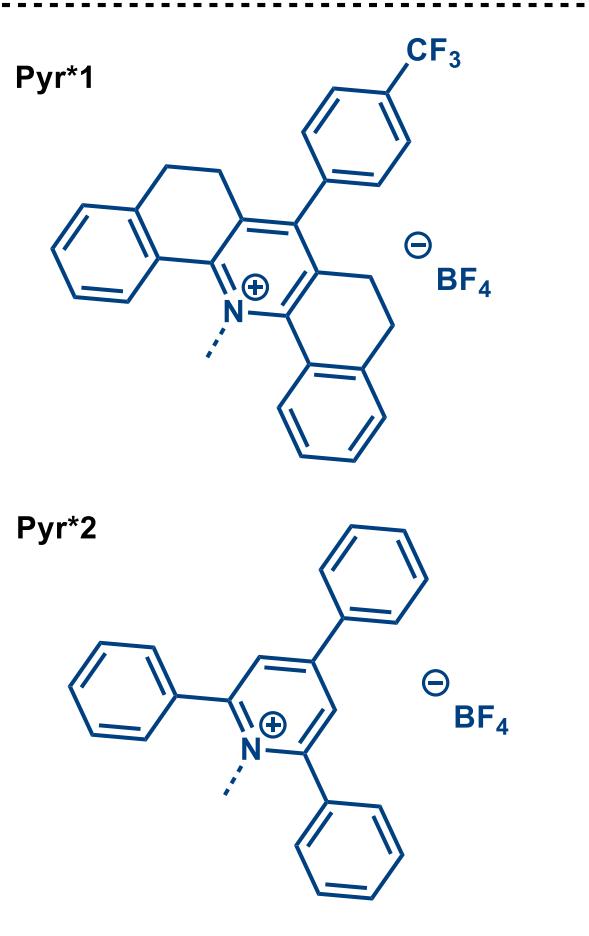
Proposed mechanism:



Deamination of Pyridinium Salts by TM Catalysis



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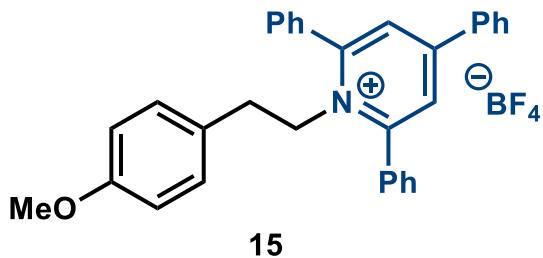
Deamination of Pyridinium Salts by LB Catalysis



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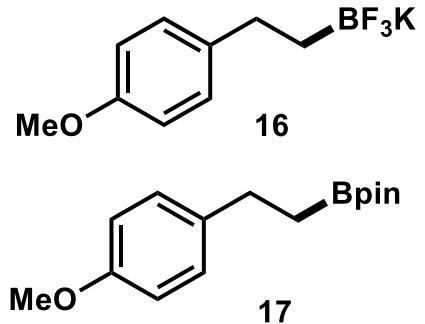
Borylation

Shi Zhuangzhi
&
Li Shuhua



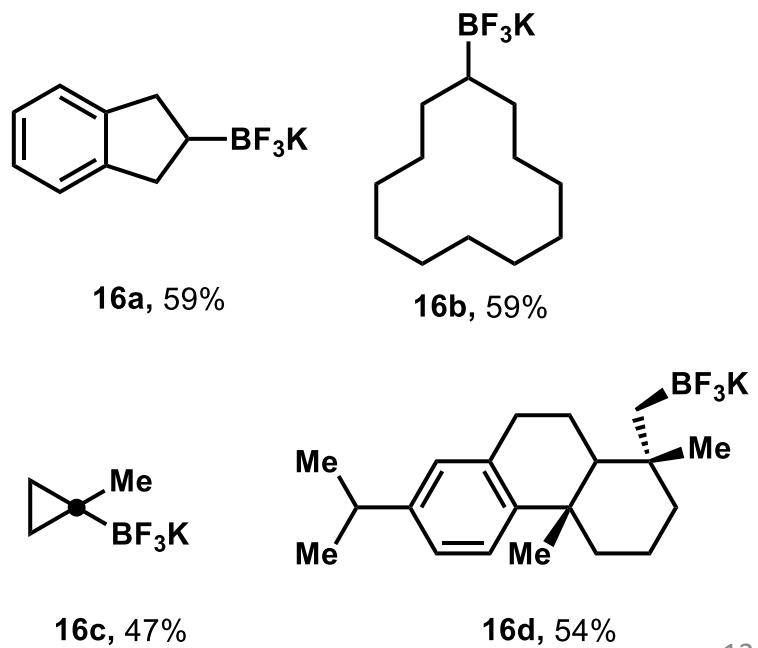
dtbpy (10 mol%), B_2cat_2 (1.5 equiv.)
DMA, 100 °C, 6 h

then KHF_2
then pinacol
 Et_3N , rt



selected examples:

Entry	Variation from the standard conditions	yield
1	None	84
2	Without Lewis base	28
3	Without Lewis base, 24 h	49
4	B_2pin_2 instead of B_2cat_2	0
5	In the dark	80

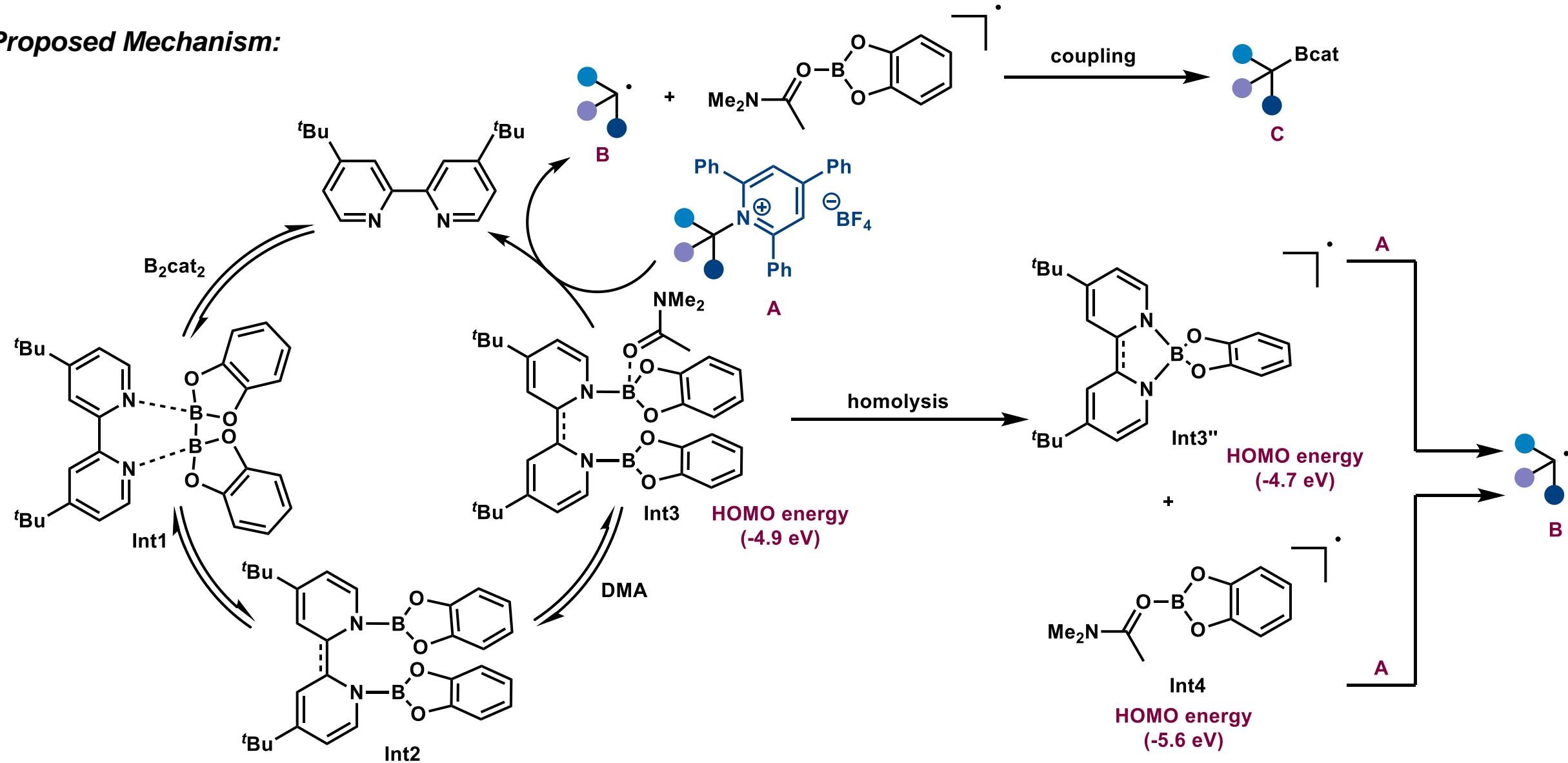


Deamination of Pyridinium Salts by LB Catalysis



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Proposed Mechanism:



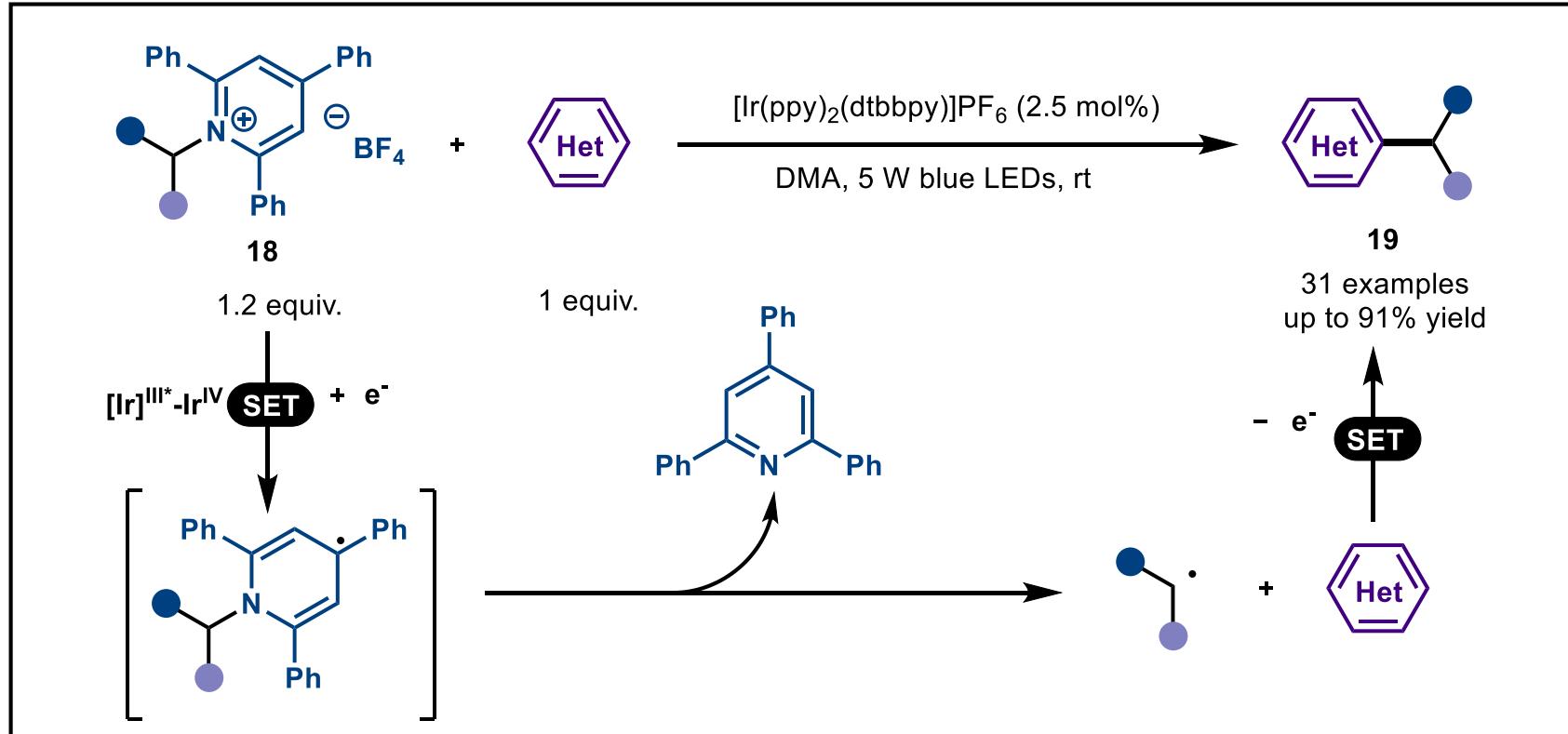
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Deamination of Pyridinium Salts by Photocatalysis



Heteroarylation

F. Glorius

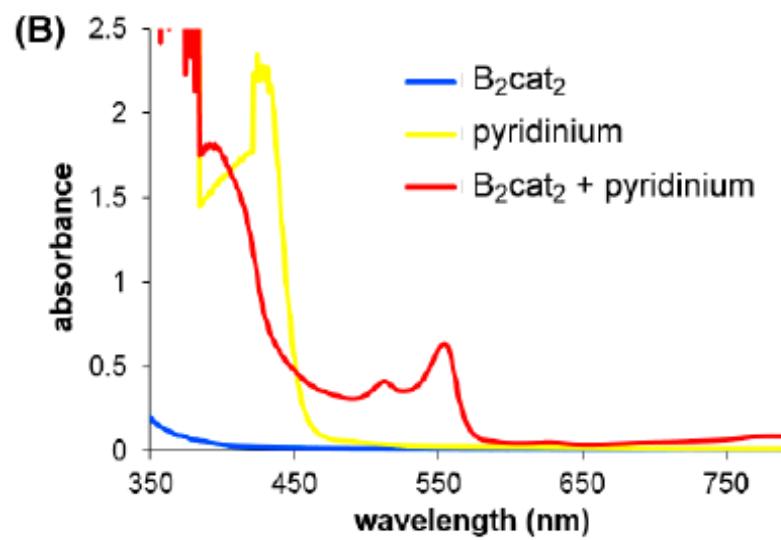
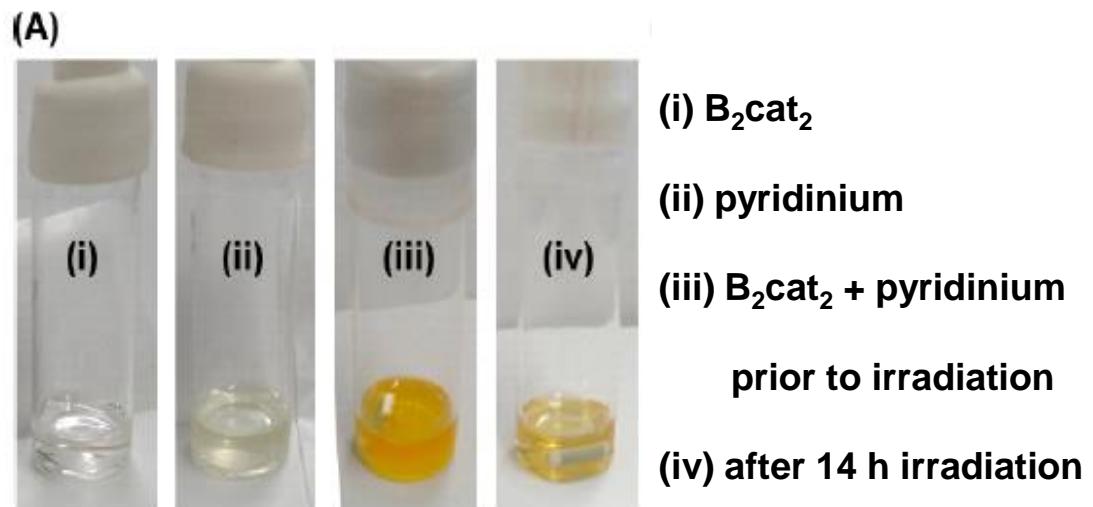
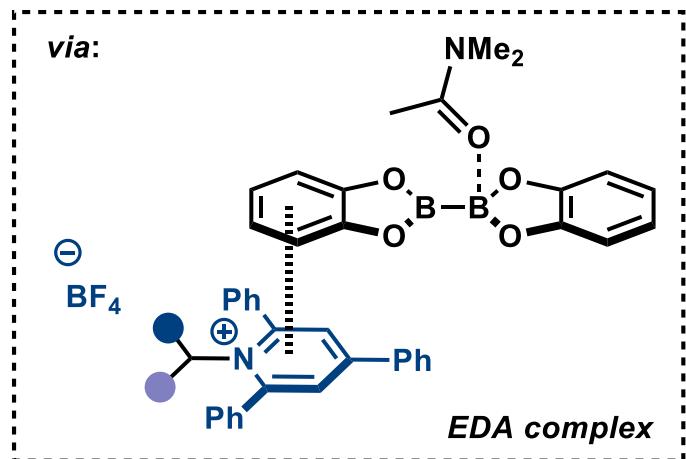
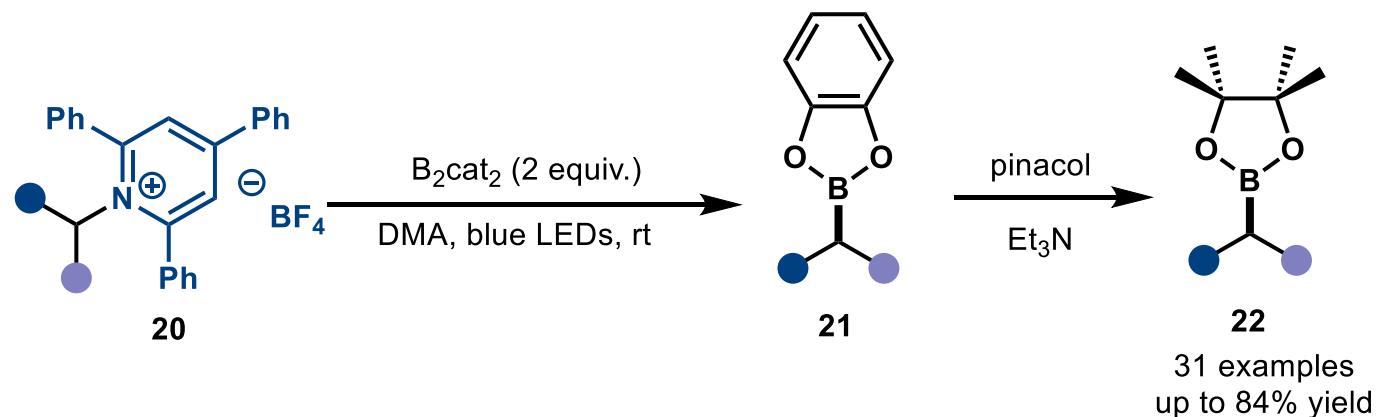


Deamination of Pyridinium Salts by Photocatalysis



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V. K. Aggarwal

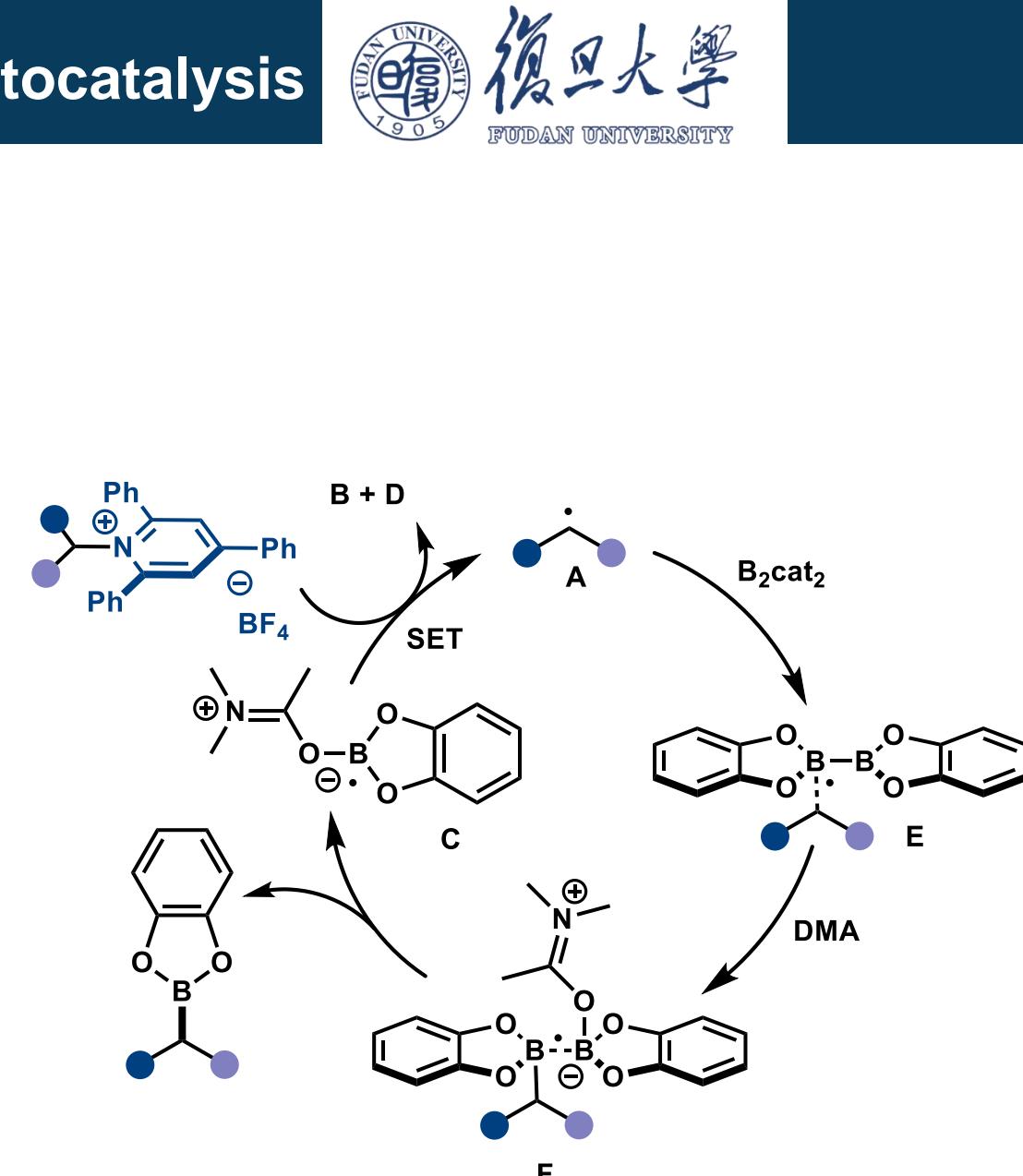
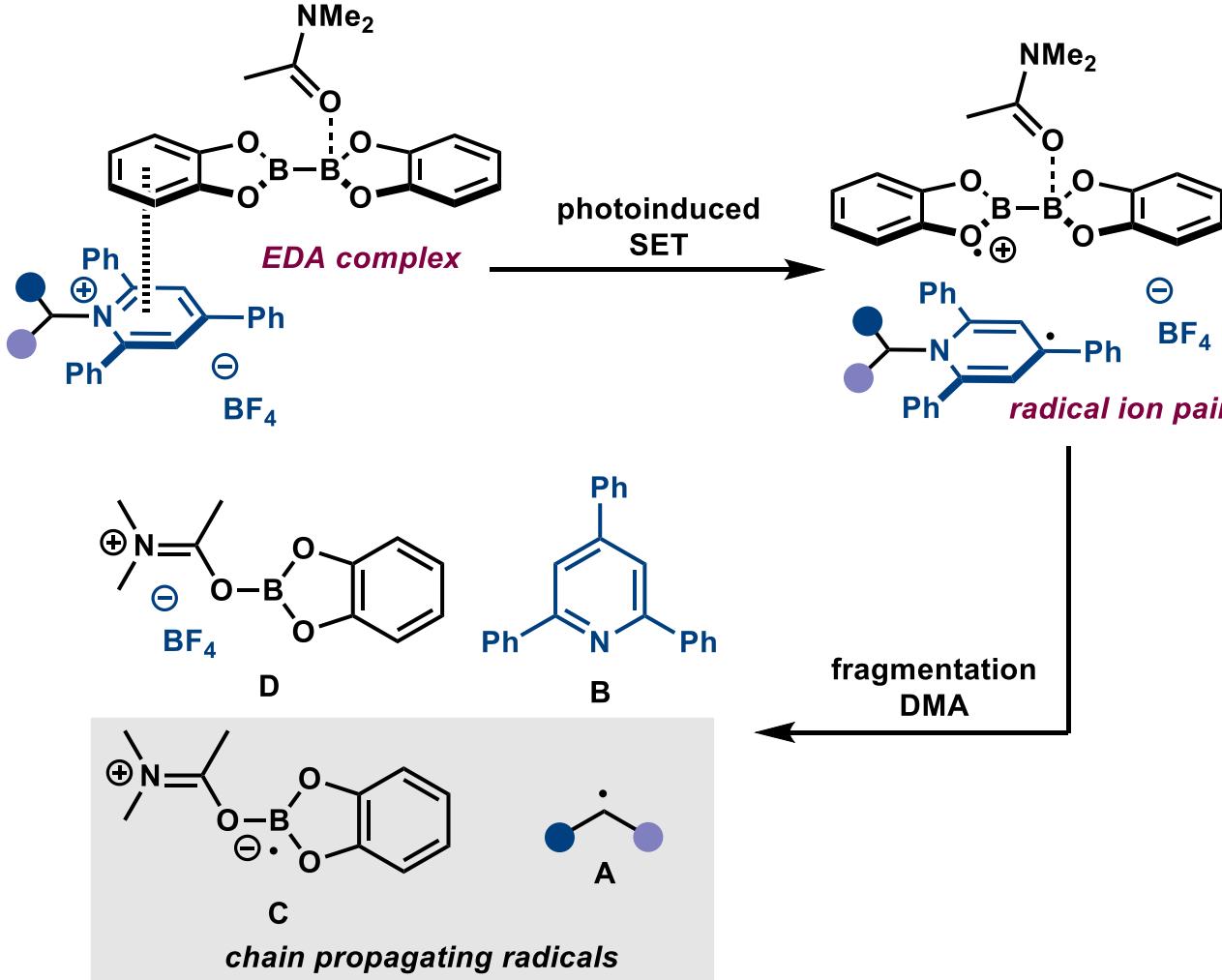


Deamination of Pyridinium Salts by Photocatalysis



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Proposed Mechanism:

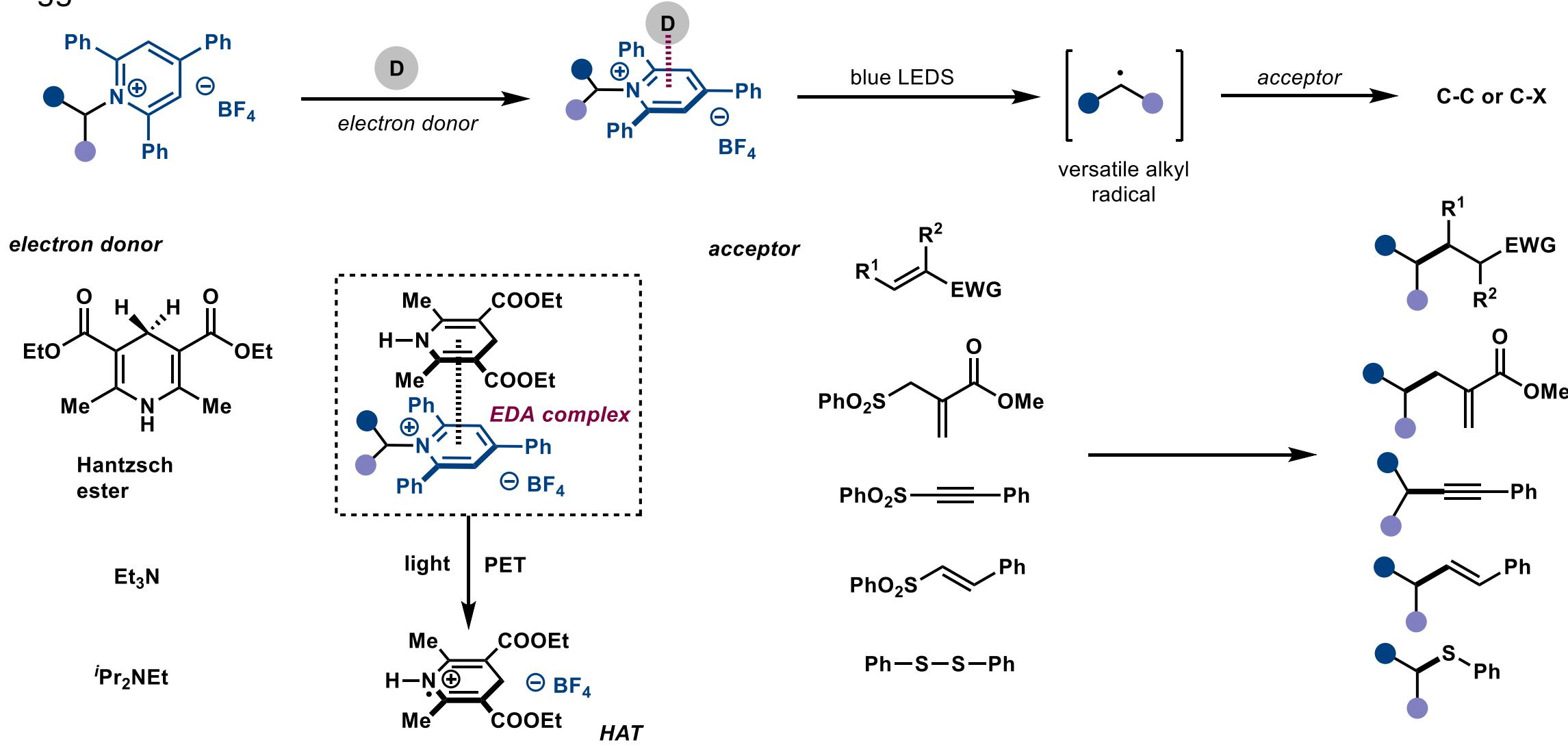


Deamination of Pyridinium Salts by Photocatalysis



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V. K. Aggarwal

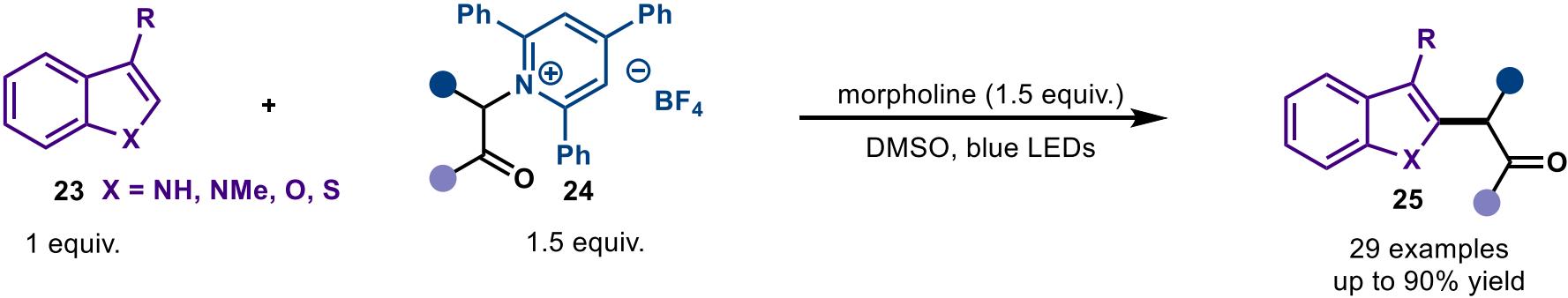


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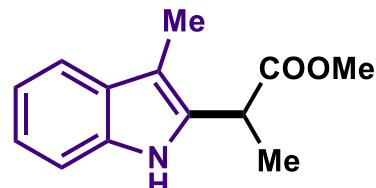


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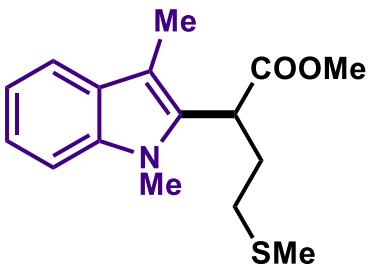
F. Glorius



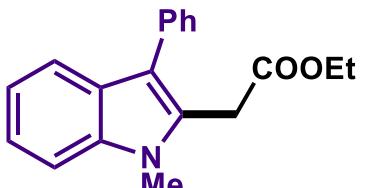
selected examples:



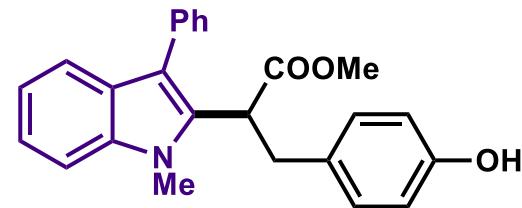
25a, 84%



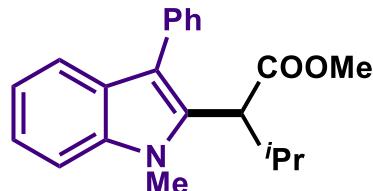
25b, 76%



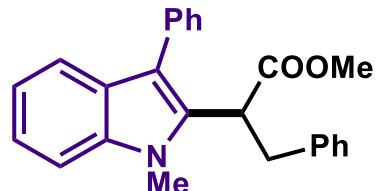
25c, 70%



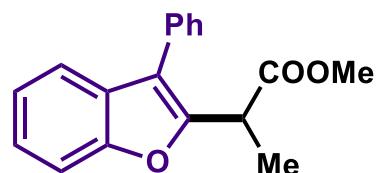
25d, 78%



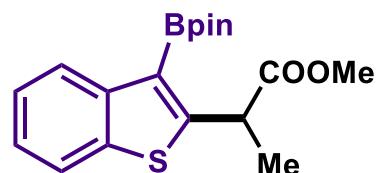
25e, 61%



25f, 61%



25g, 83%

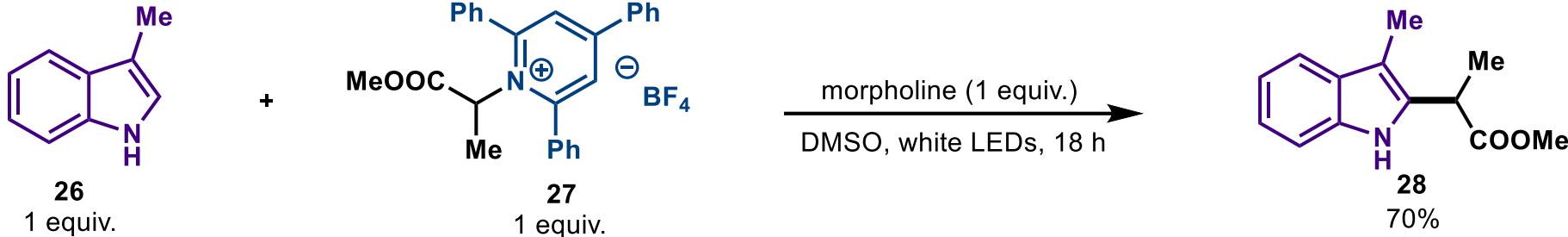


25h, 44%

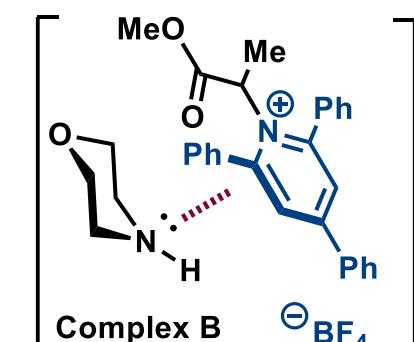
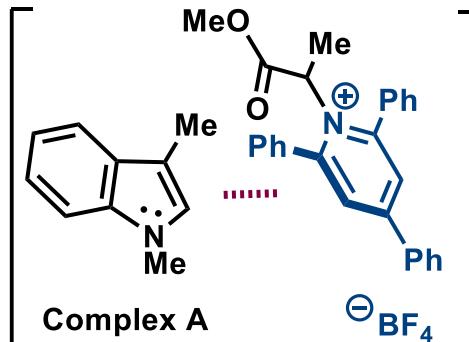
Deamination of Pyridinium Salts by Photocatalysis



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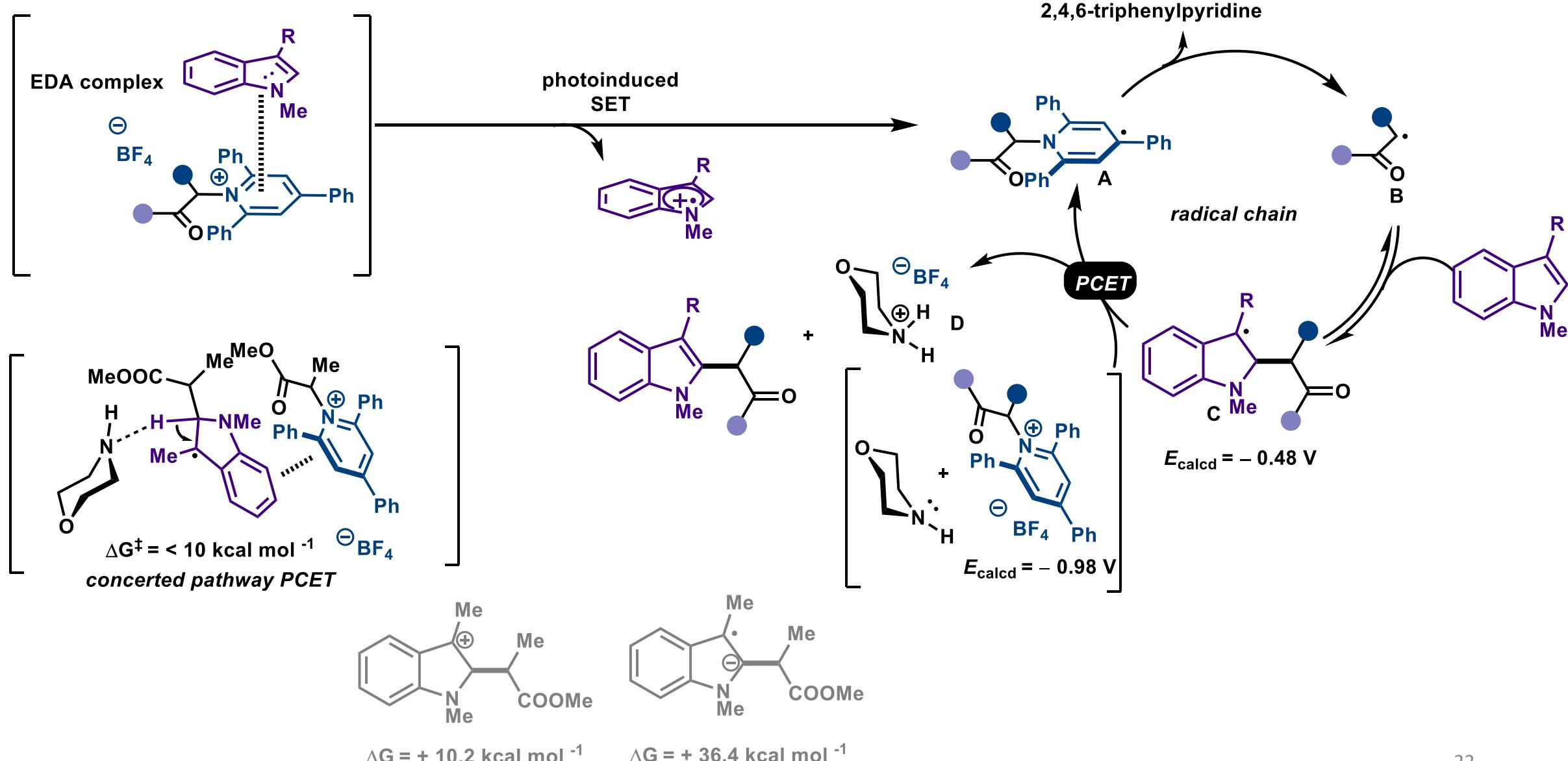
Base	Yield
morph. No light	8%
morph. + O ₂	0%



Deamination of Pyridinium Salts by Photocatalysis



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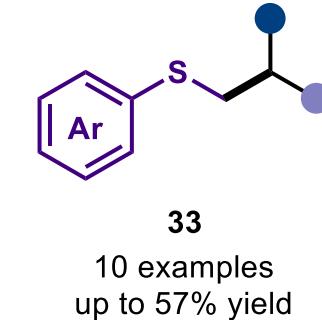
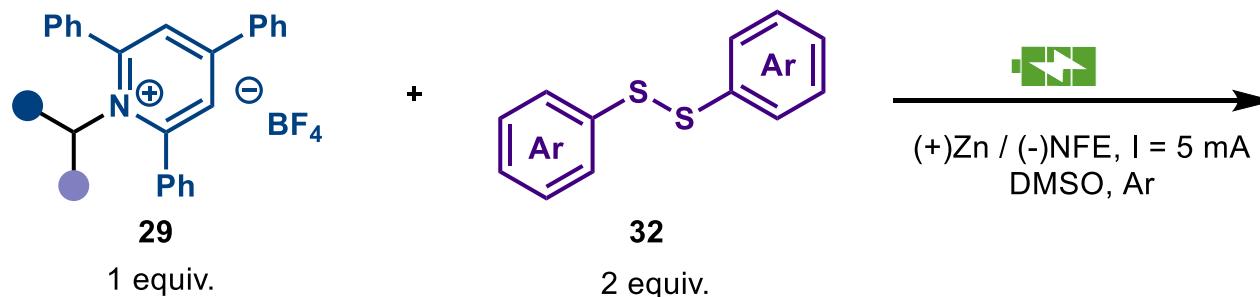
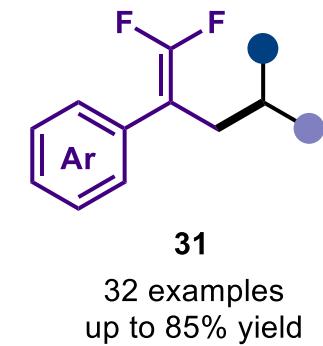
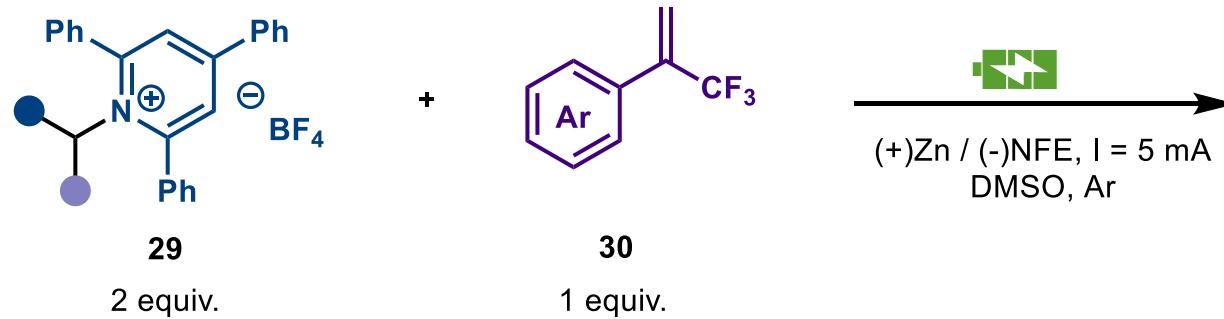
Deamination of Pyridinium Salts by Electrochemical



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Fluoroalkenylation and Thiolation

Wang Yi
&
Pan Yi

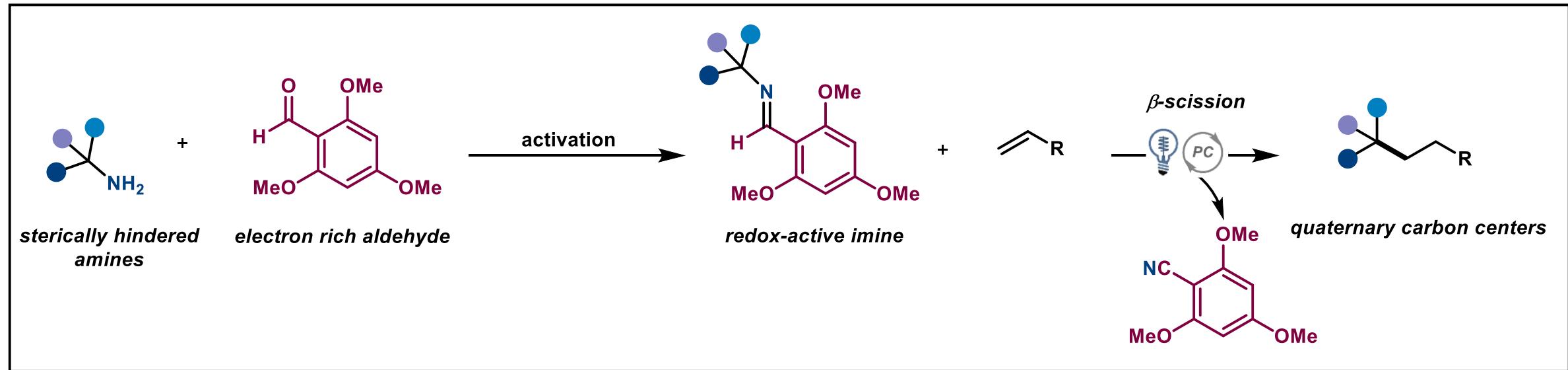


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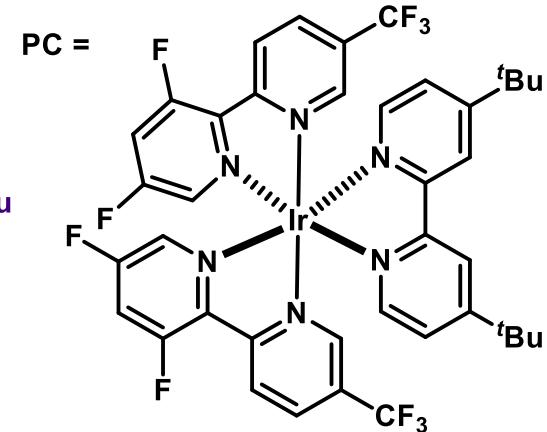
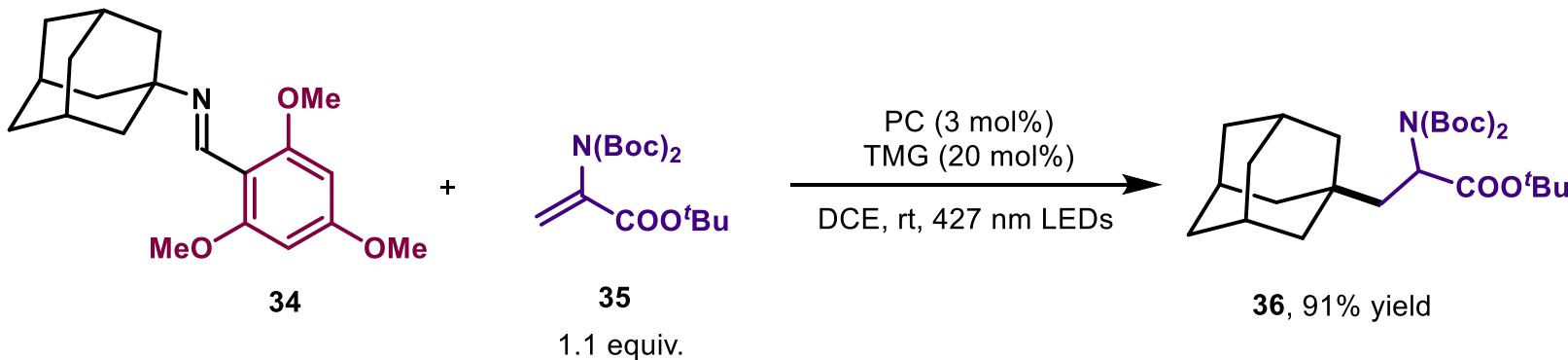
Electron-rich Imines By Photocatalysis



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T. Rovis

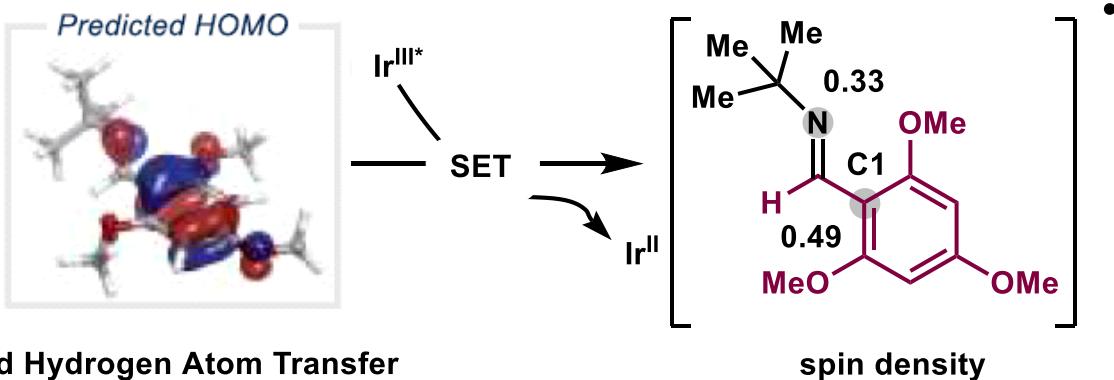


Electron-rich Imines By Photocatalysis

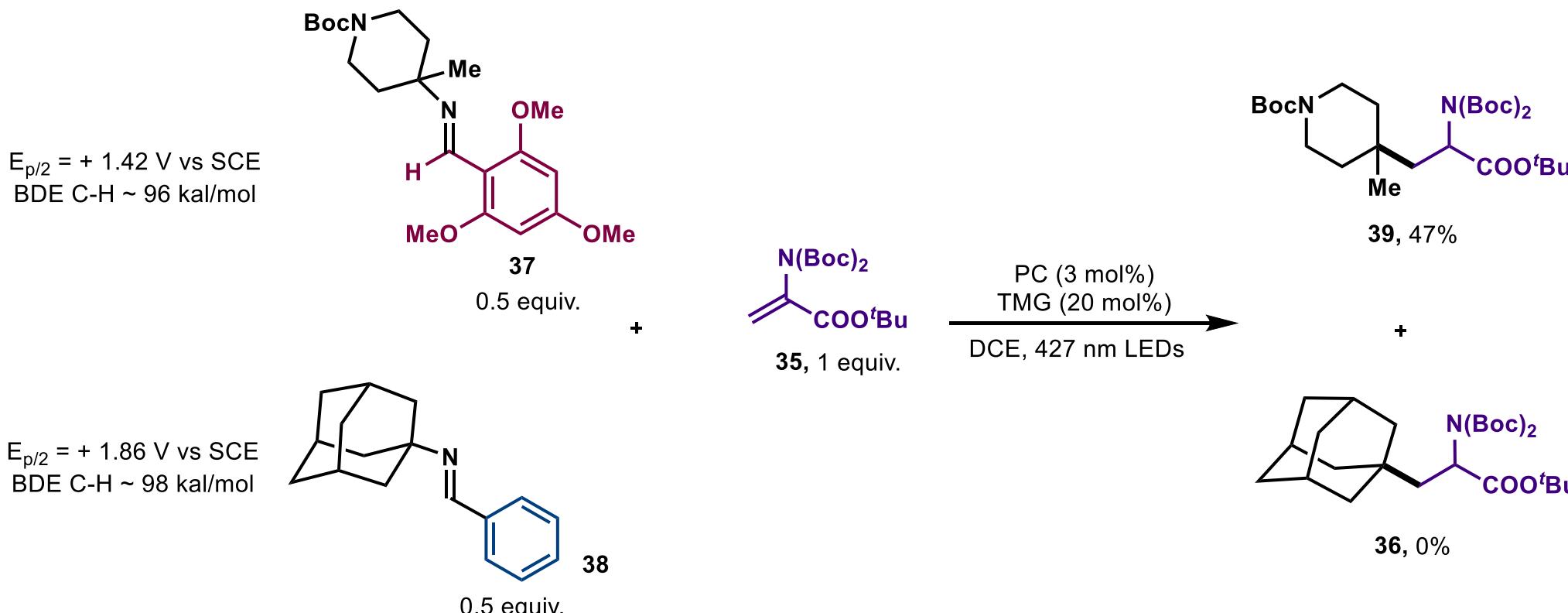


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A. Potential Modes of Activation



B. Testing for Iminyl Radical Promoted Hydrogen Atom Transfer

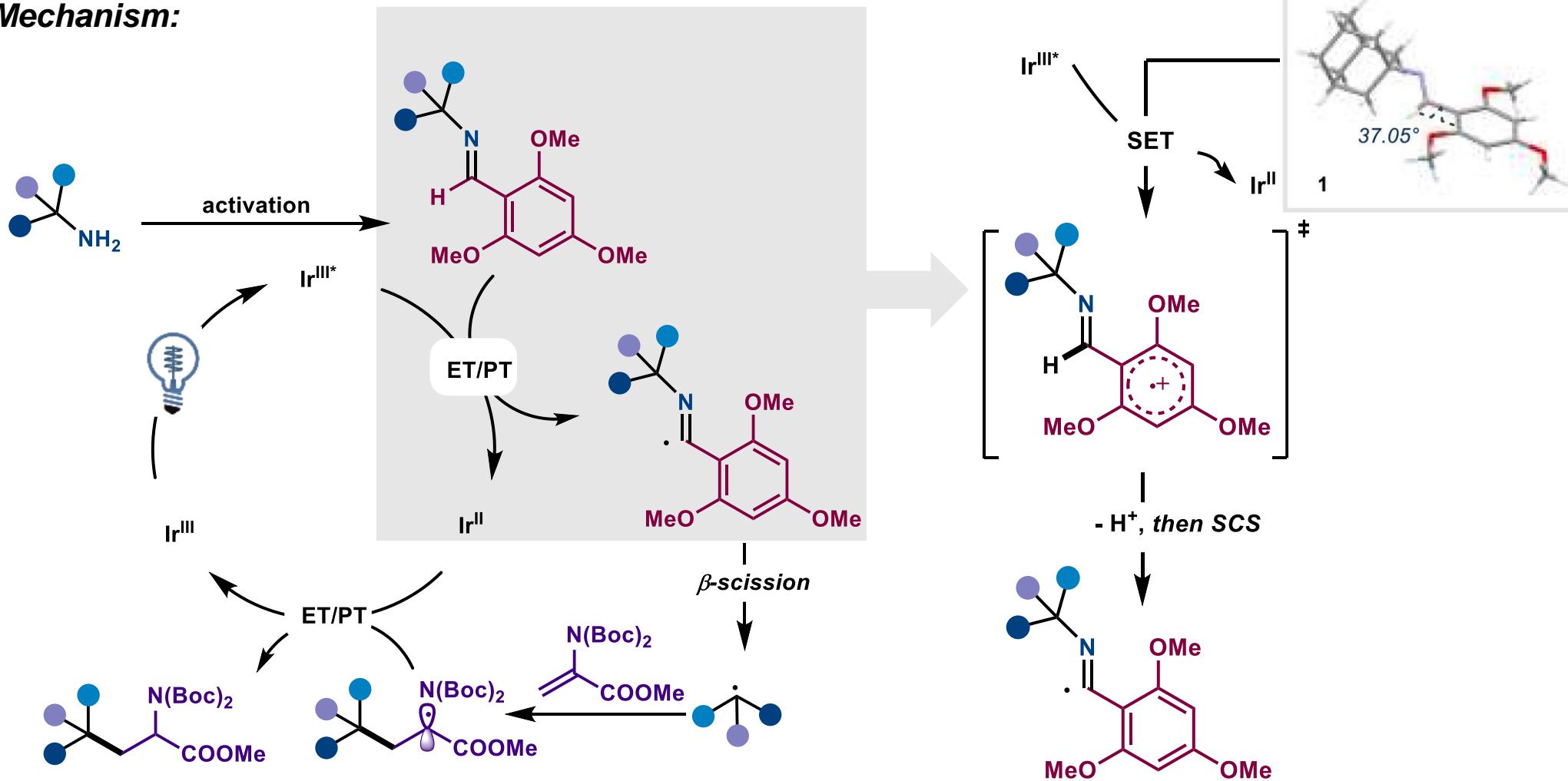


Electron-rich Imines By Photocatalysis



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Proposed Mechanism:

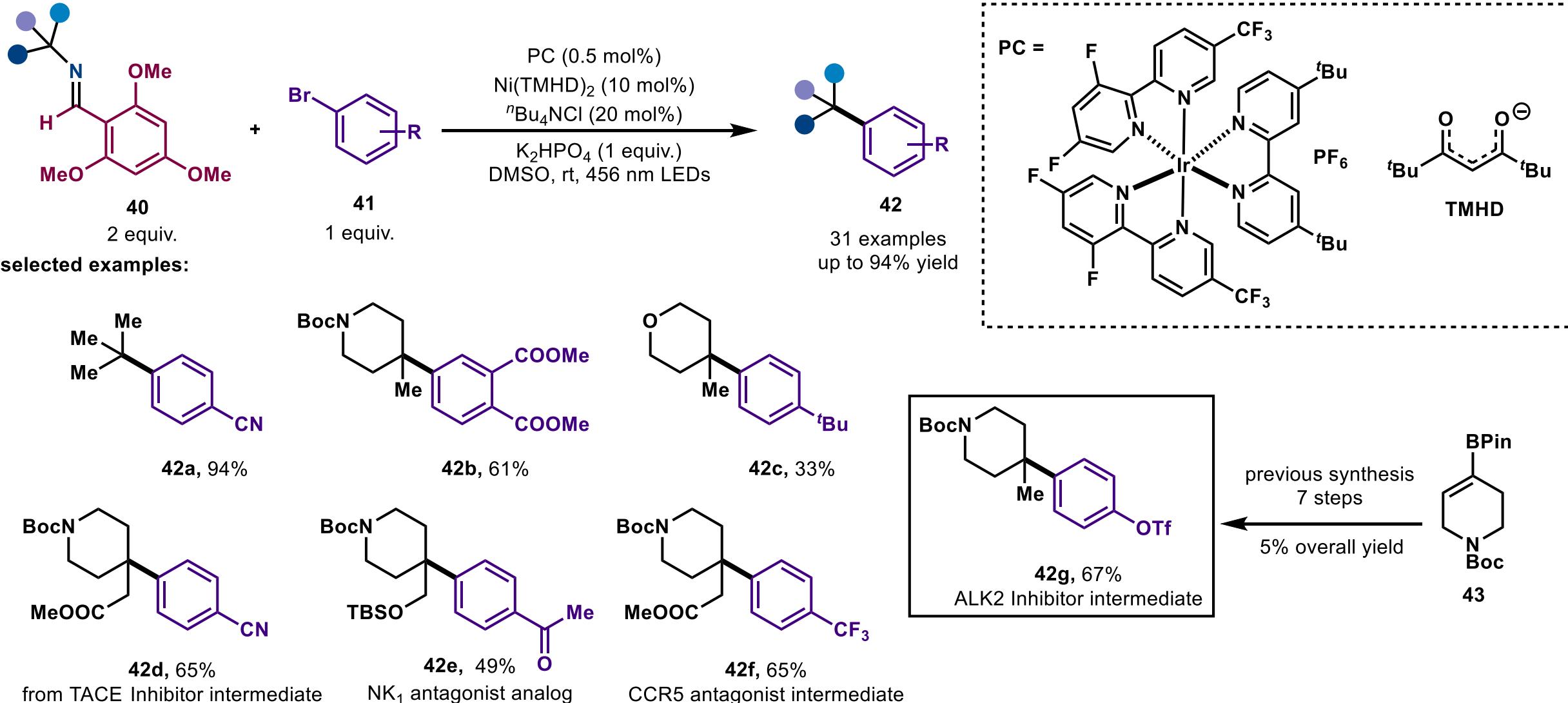


Electron-rich Imines By Nickel/Photo-Catalyzed



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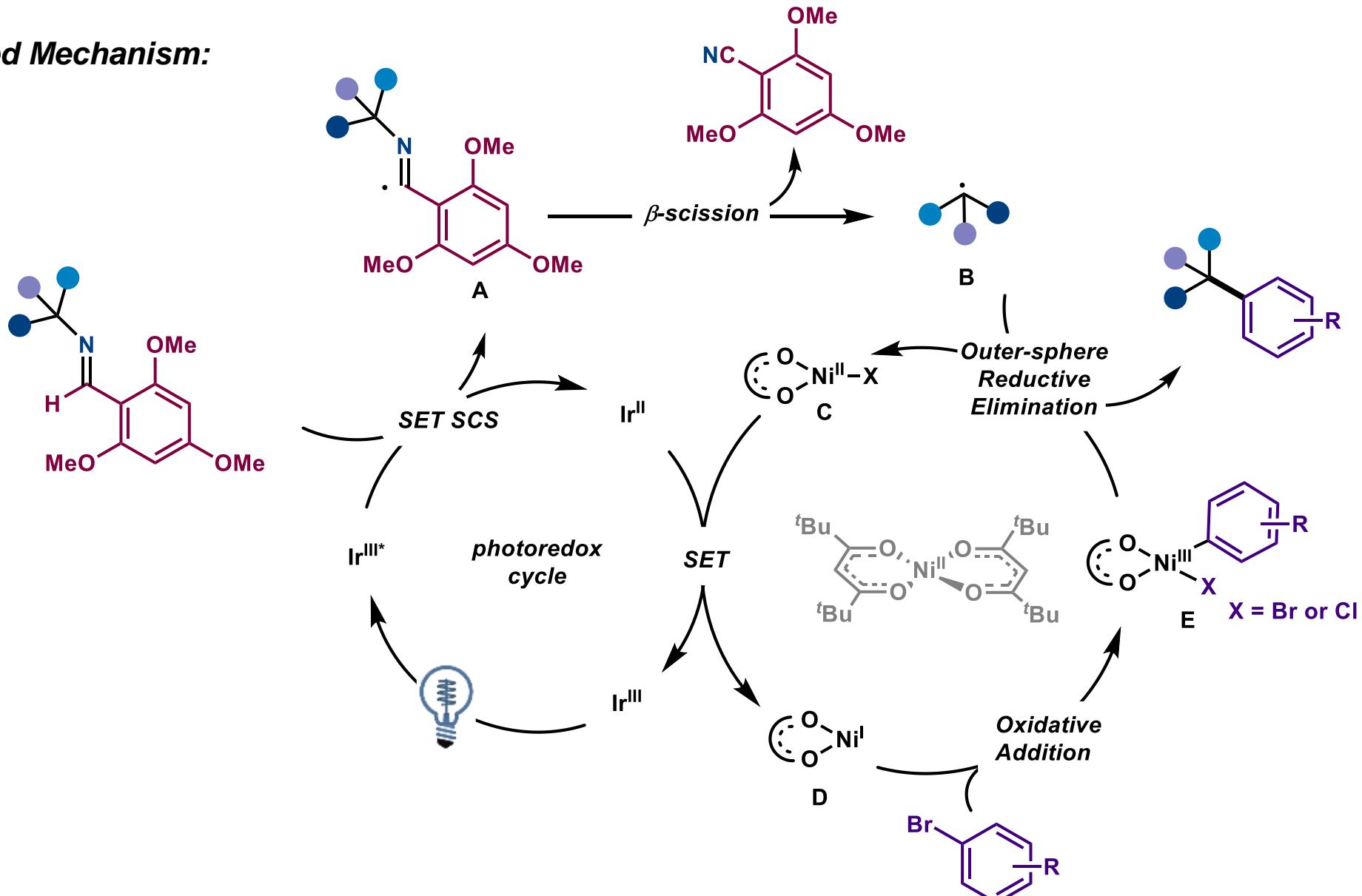
T. Rovis



Electron-rich Imines By Nickel/Photo-Catalyzed



Proposed Mechanism:



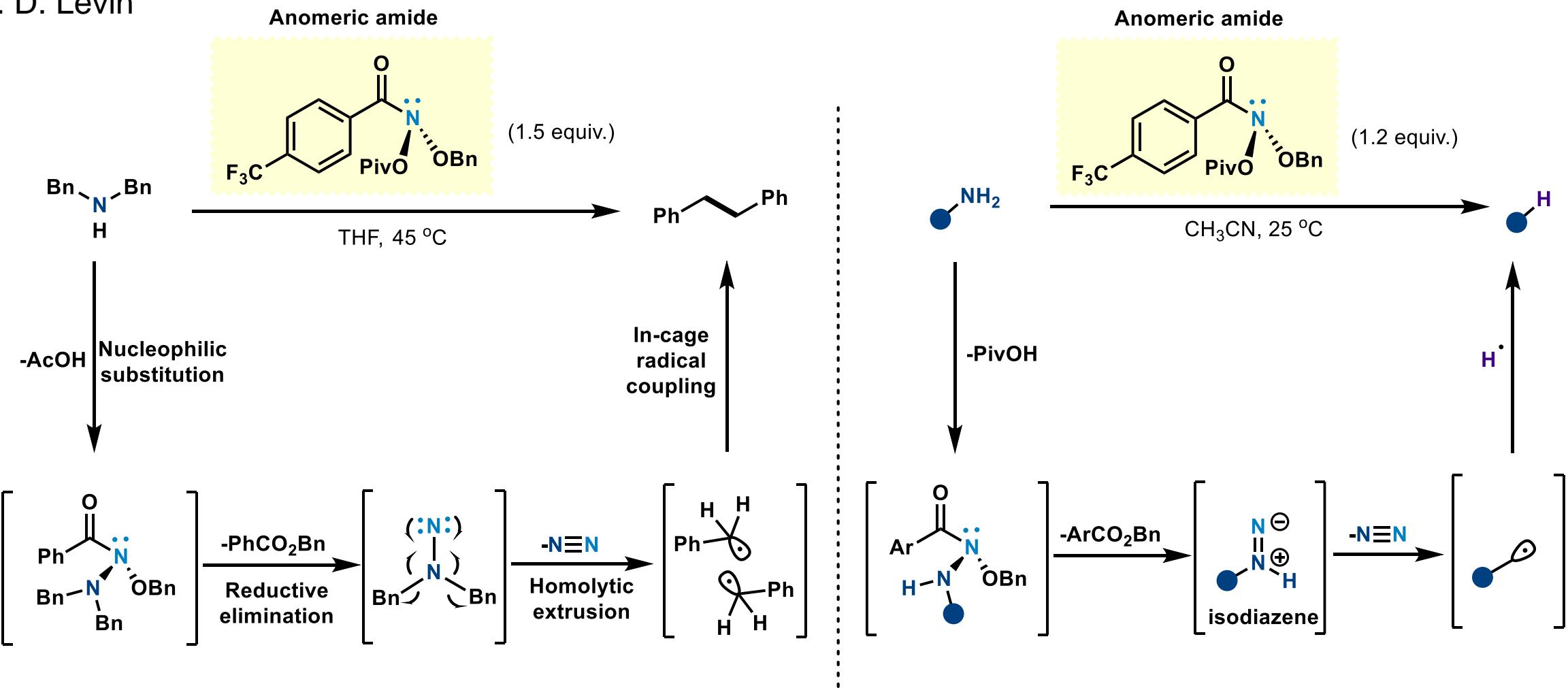
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Direct Deaminative Functionalizations

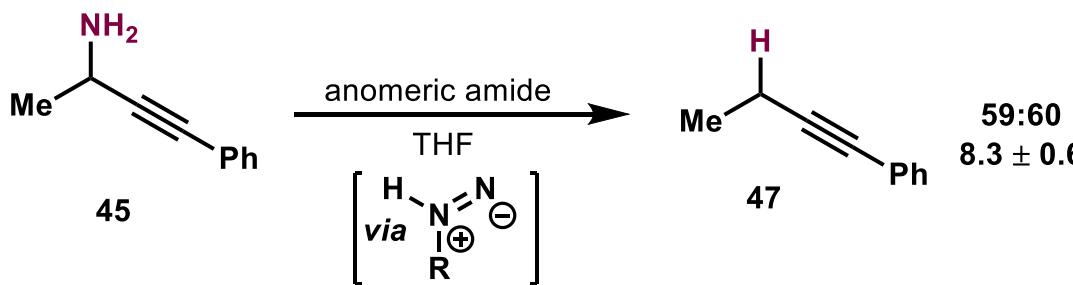
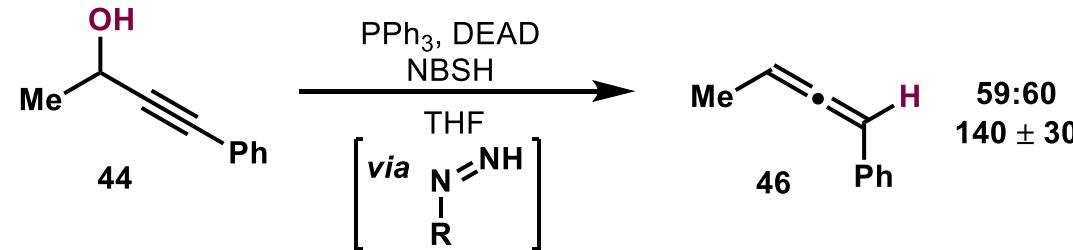


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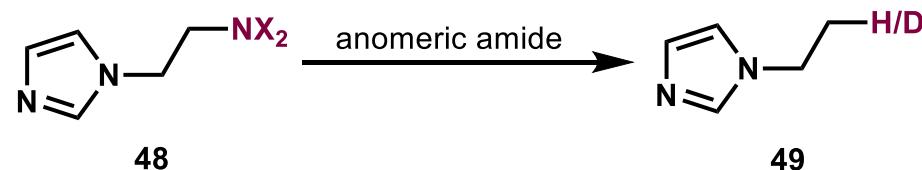
M. D. Levin



Comparison with the Myers allene synthesis

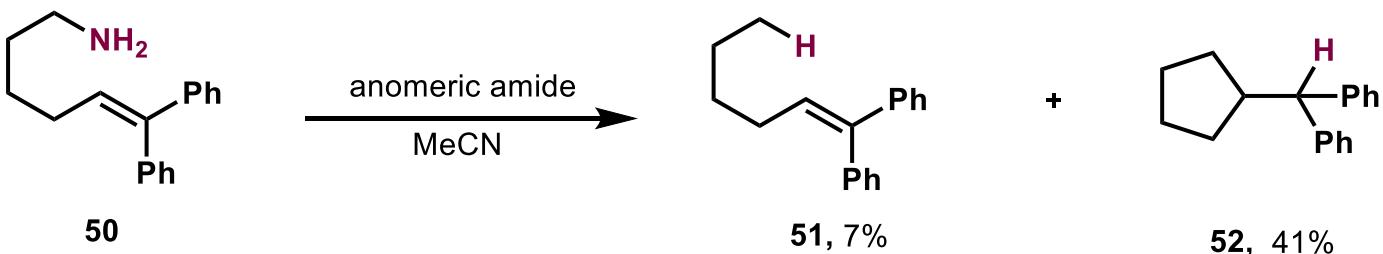


Deuterium labeling studies



X ₂	Solvent	Product
X = H	MeCN-d ₃	0% D
X = D	MeCN	63 ± 9% D
X = H	96:4 MeCN/D ₂ O	73 ± 7% D
X = D	96:4 MeCN/D ₂ O	101 ± 4% D

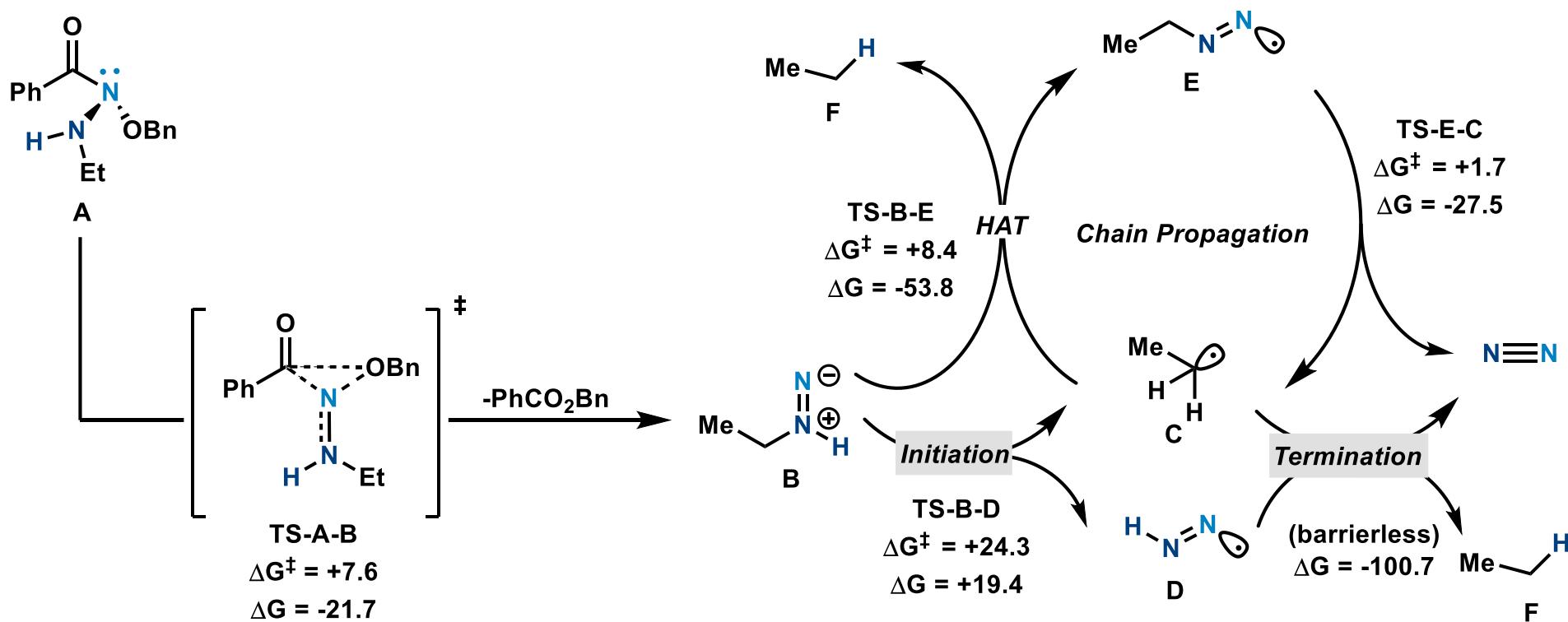
Radical clock experiments



Direct Deaminative Functionalizations



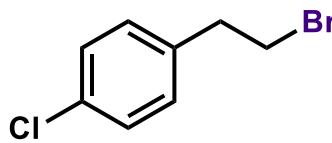
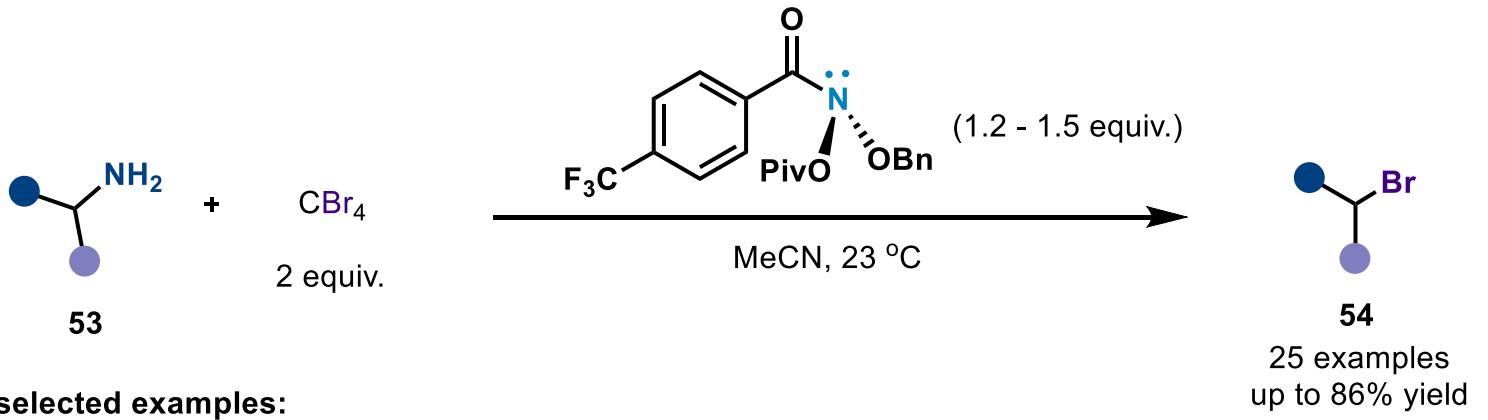
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Direct Deaminative Functionalizations



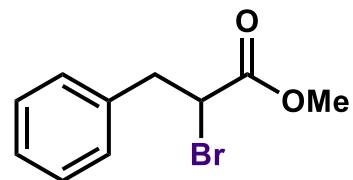
M. D. Levin



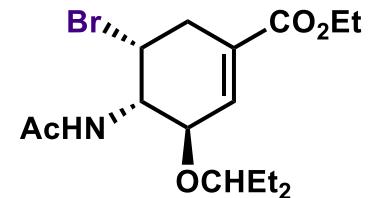
54a, 78%



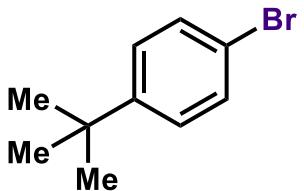
54b, 55%



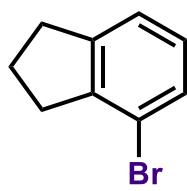
54c, 58%



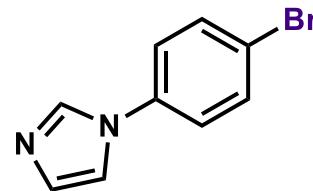
54d, 59%



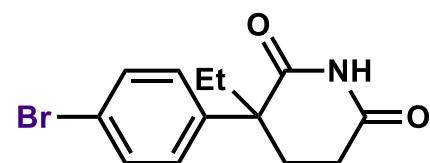
54e, 84%



54f, 86%



54g, 83%



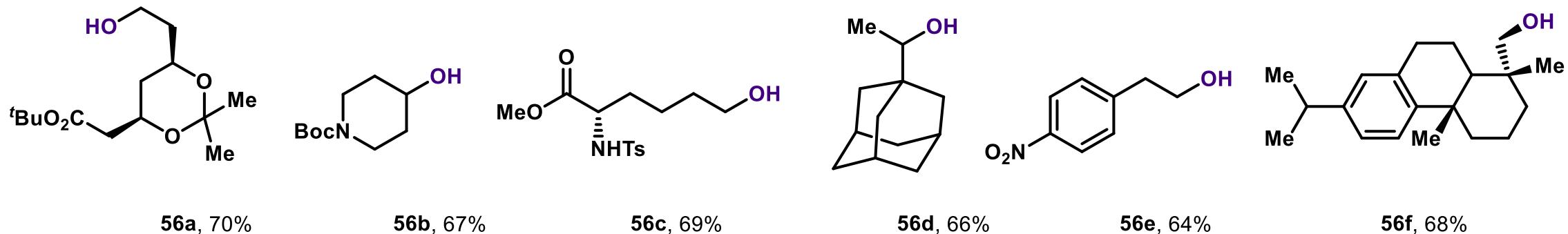
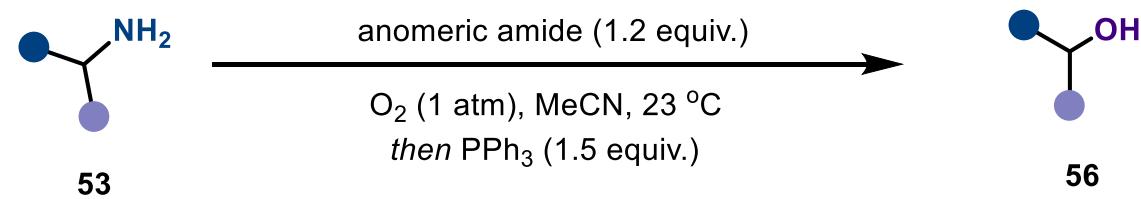
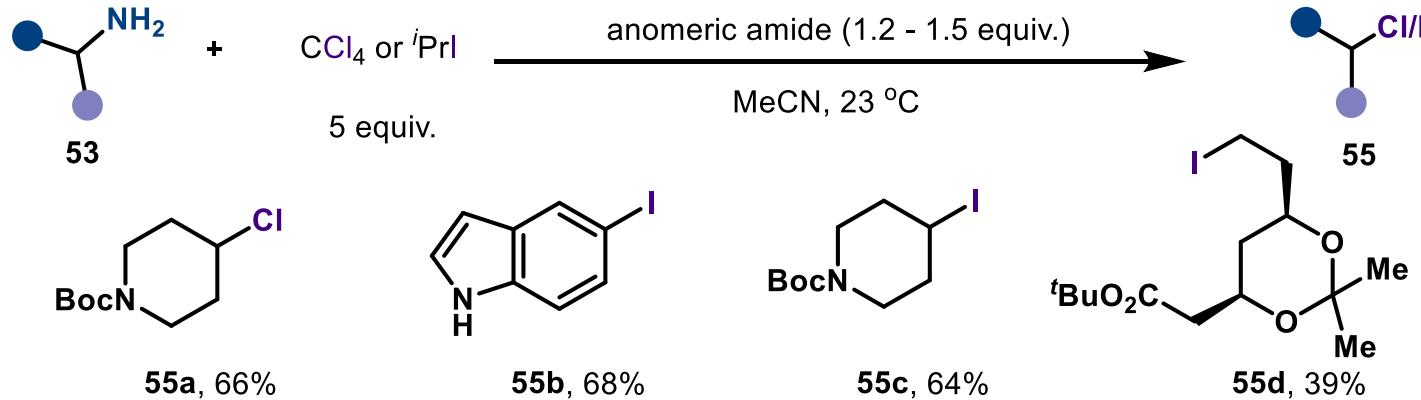
54h, 77%

Direct Deaminative Functionalizations

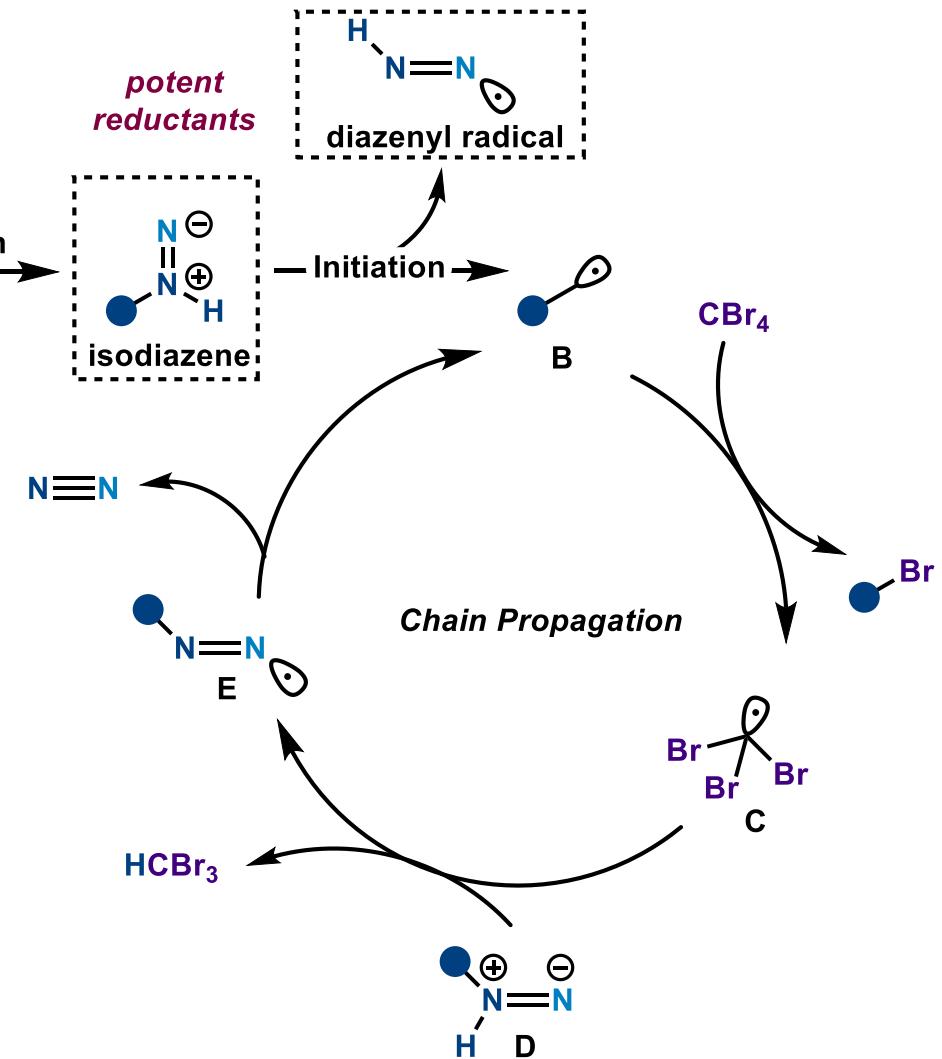
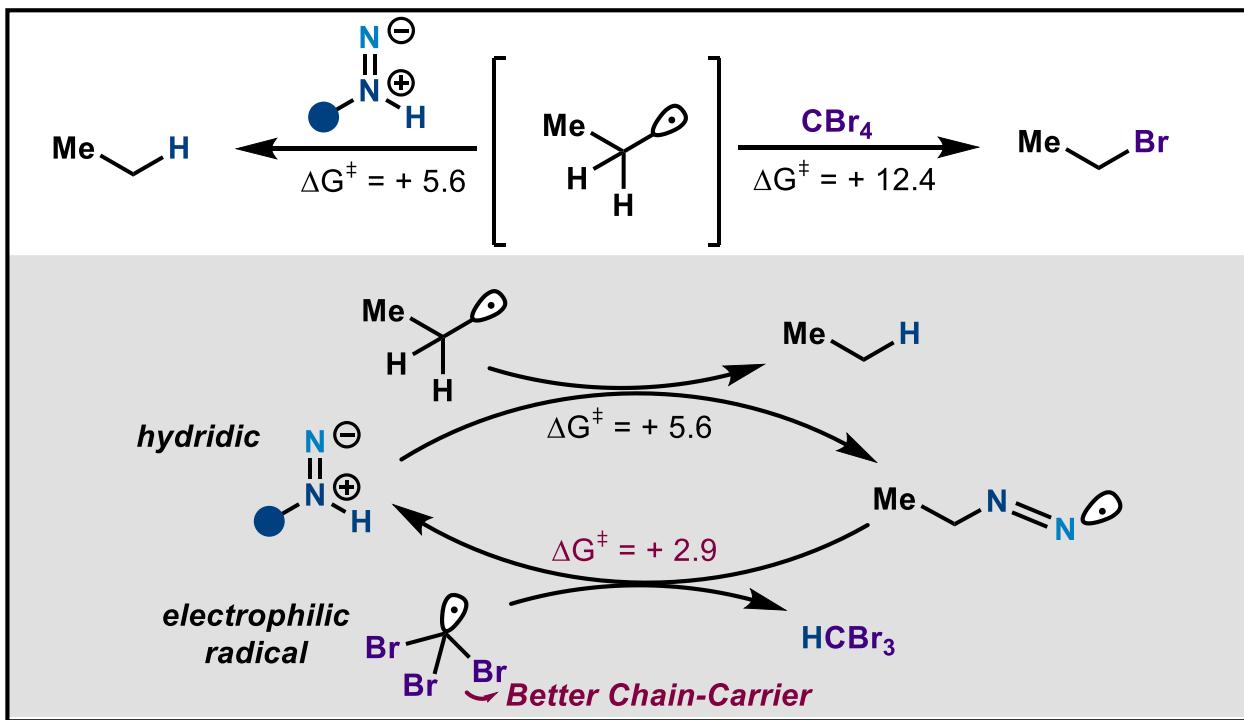
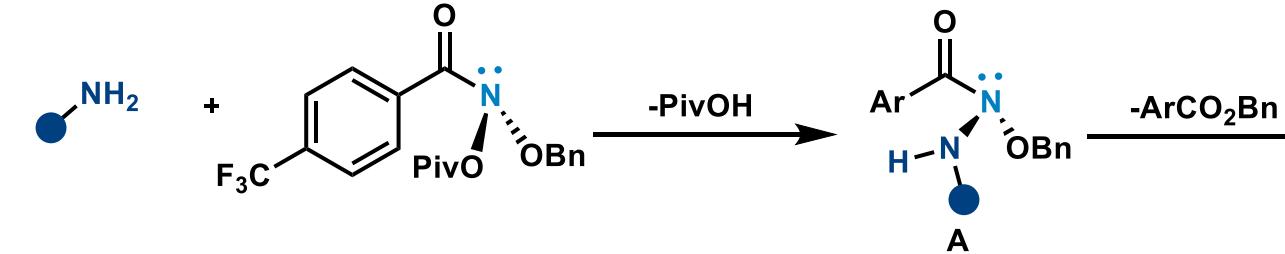


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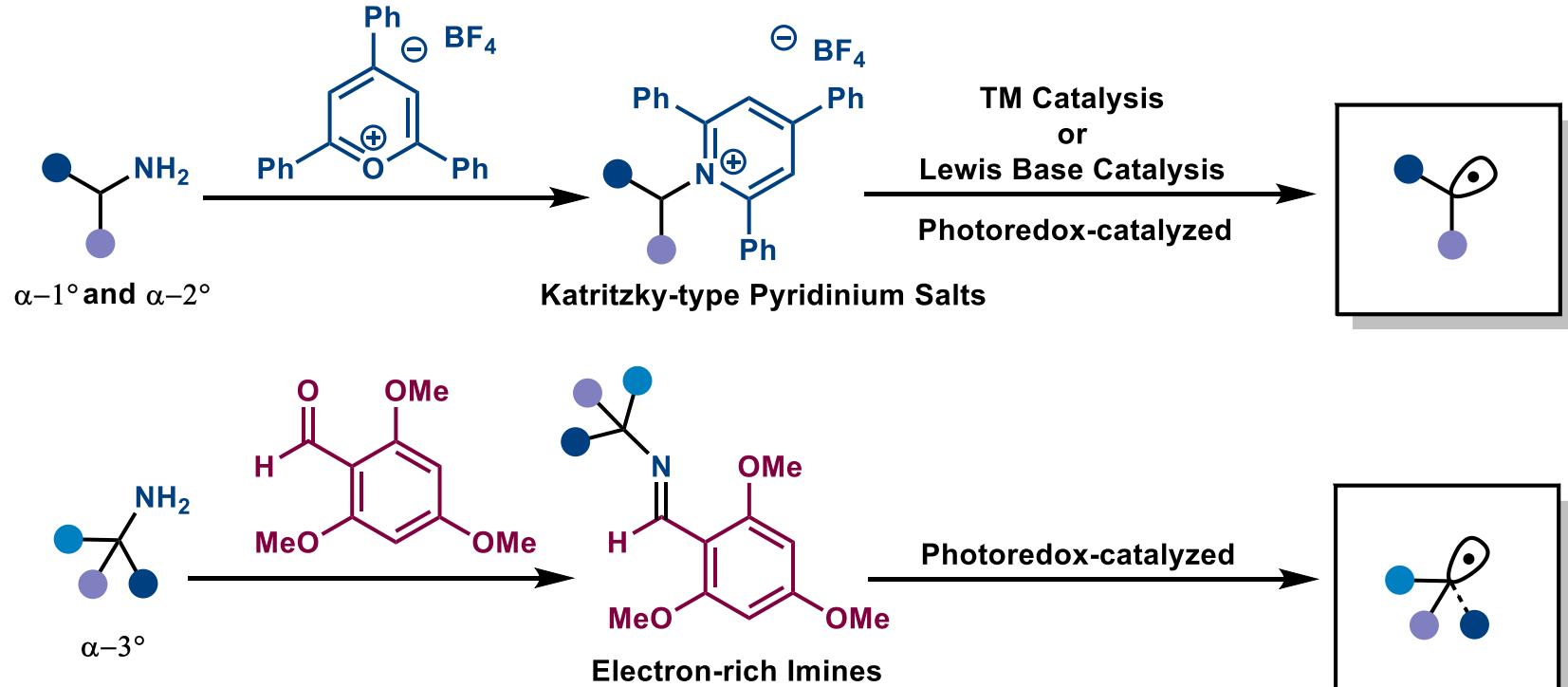
Direct Deaminative Functionalizations



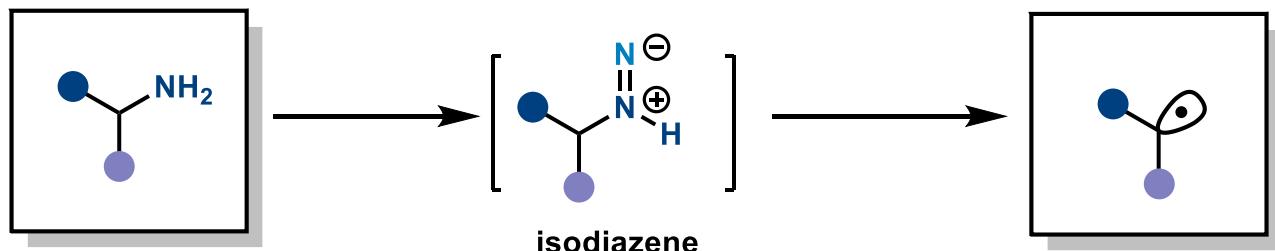
- **Background**
- **Deaminative Functionalizations via Katritzky-type Pyridinium Salts**
 - By Transition Metal Catalysis and Lewis Base Catalysis
 - By Photoredox Catalysis and Electrochemical Catalysis
- **Deaminative Functionalizations via Electron-rich Imines**
- **Direct Deaminative Functionalizations**
- **Summary and Outlook**

Summary and Outlook

Indirect direct deaminative functionalizations via condensation of pyran or aldehyde

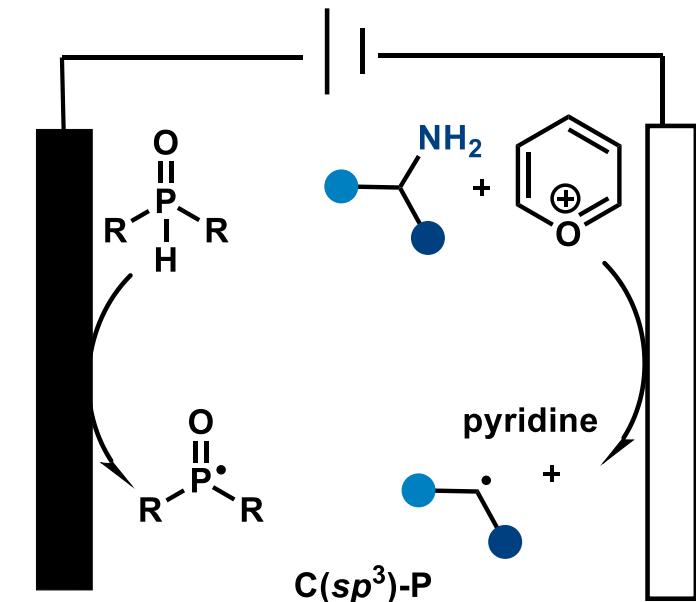


Direct deaminative functionalizations



Outlook

(i) In-situ generation (one pot)



(ii) Asymmetric version

(iii) Other reagent



Thanks for your
attention!