



光催化的对映选择性的1,4-共轭加成 反应

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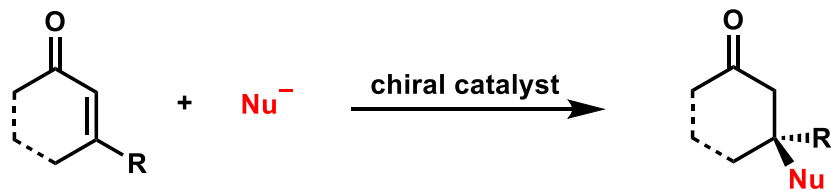
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3. 总结

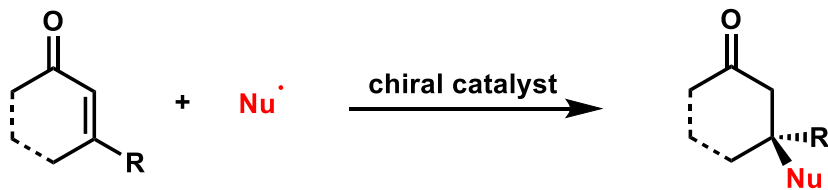
背景介绍

双电子型亲核试剂



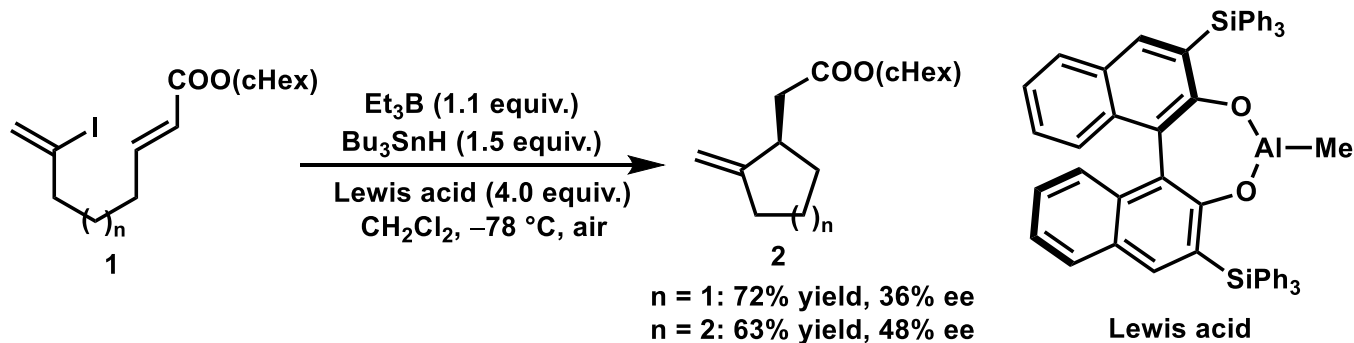
1. 体系发展成熟
2. 构建手性中心及新的化学键最重要的方法之一

单电子型亲核试剂

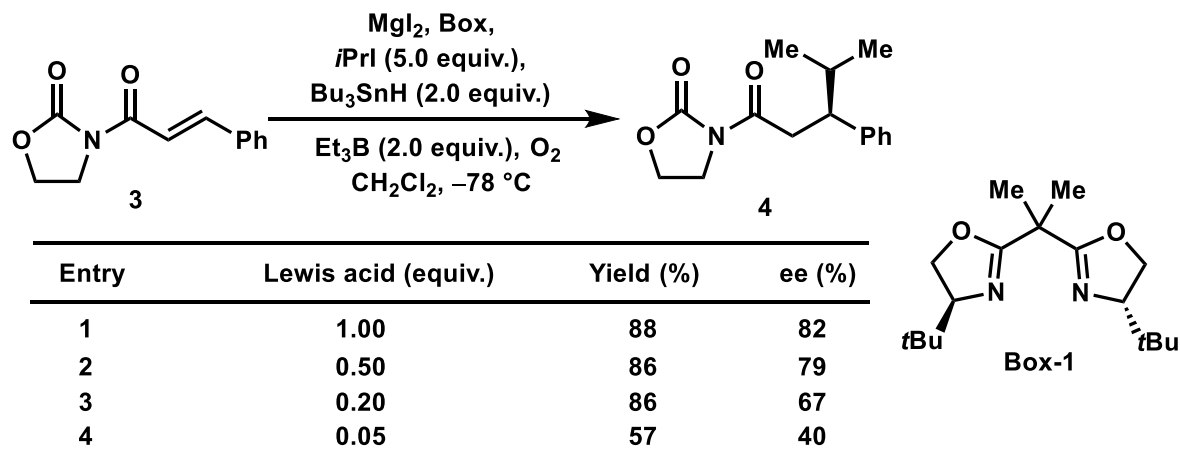


1. 自由基反应活性高，寿命短
2. 存在消旋背景反应

背景介绍

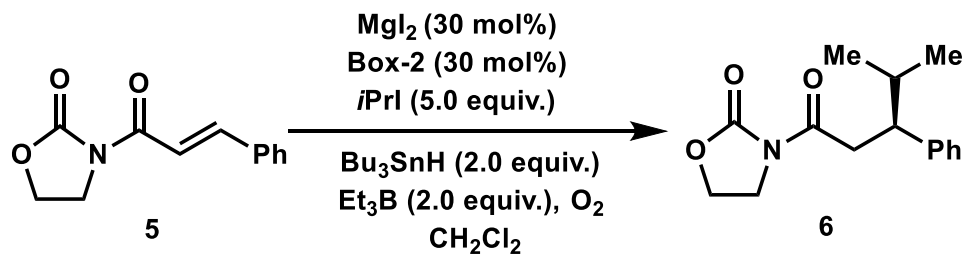


Hayashi, H.; Nishida, A.; Kawahara, N.; Nishida, M. *Chem. Commun.* **1996**, 579.

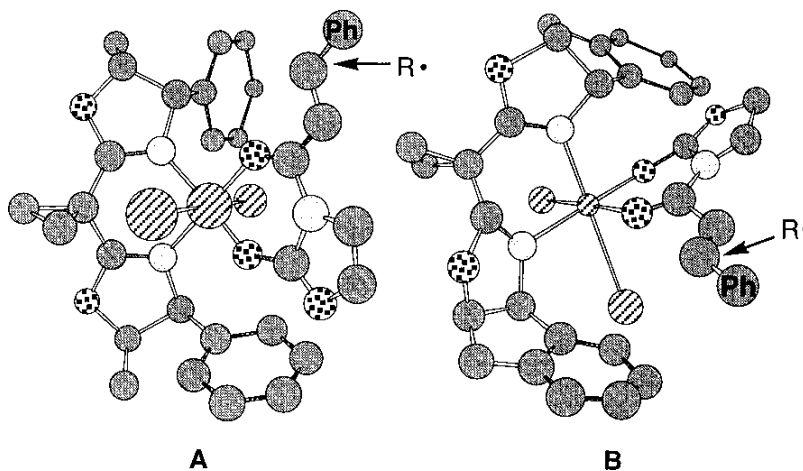
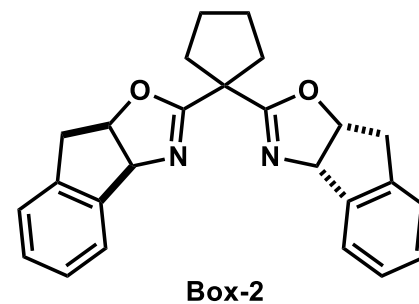


Ji, J.; Sibi, M. P. *J. Am. Chem. Soc.* **1996**, *118*, 9200.

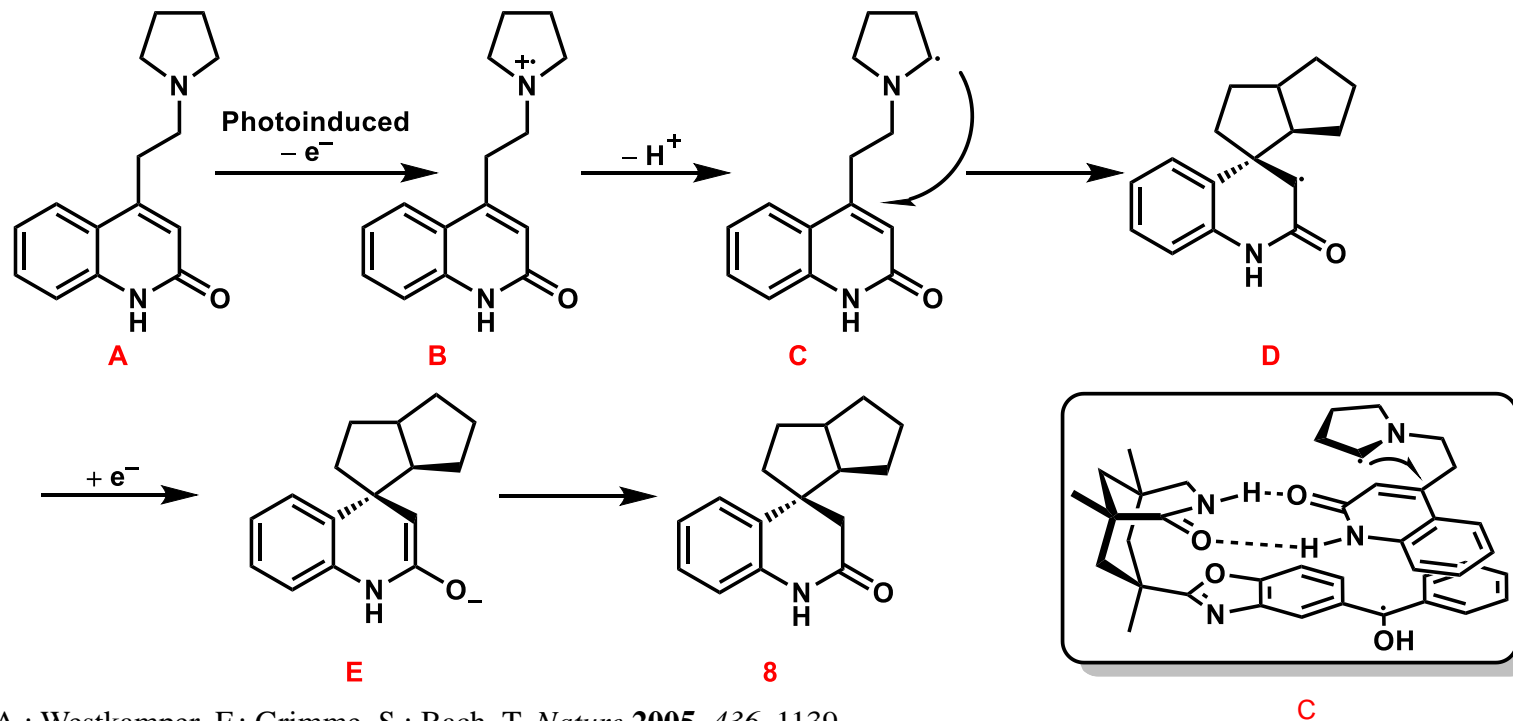
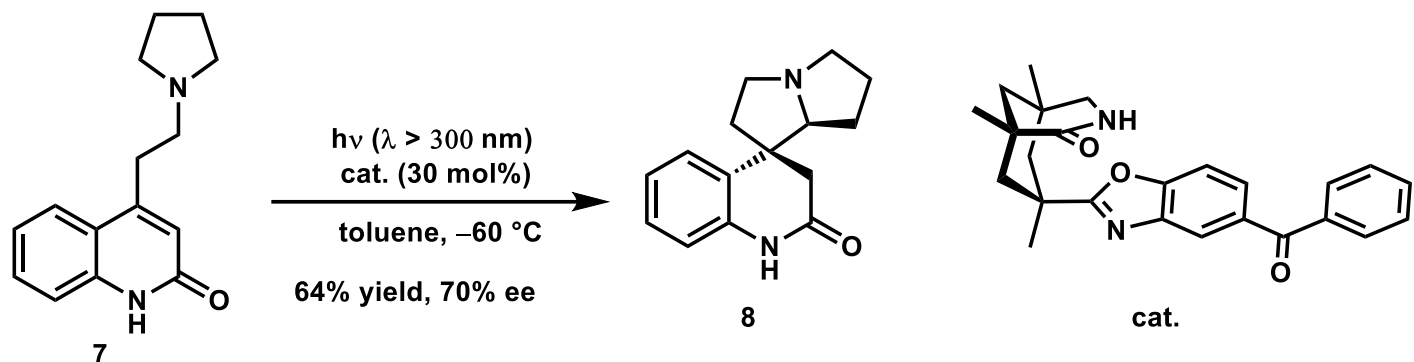
背景介绍



Entry	T (°C)	Yield(%)	ee(%)
1	-78 °C	91	97
2	-20 °C	93	95
3	0 °C	91	94
4	25 °C	87	93



背景介绍



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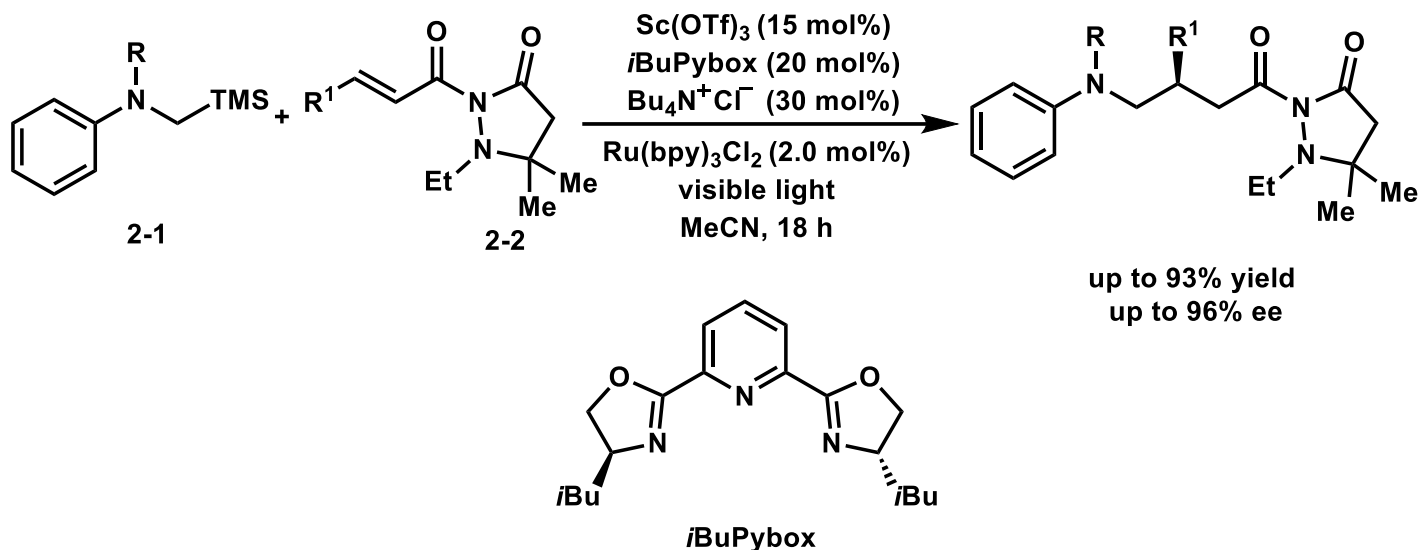
2.1 手性路易斯酸催化的共轭加成反应

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手性路易斯酸催化的共轭加成反应



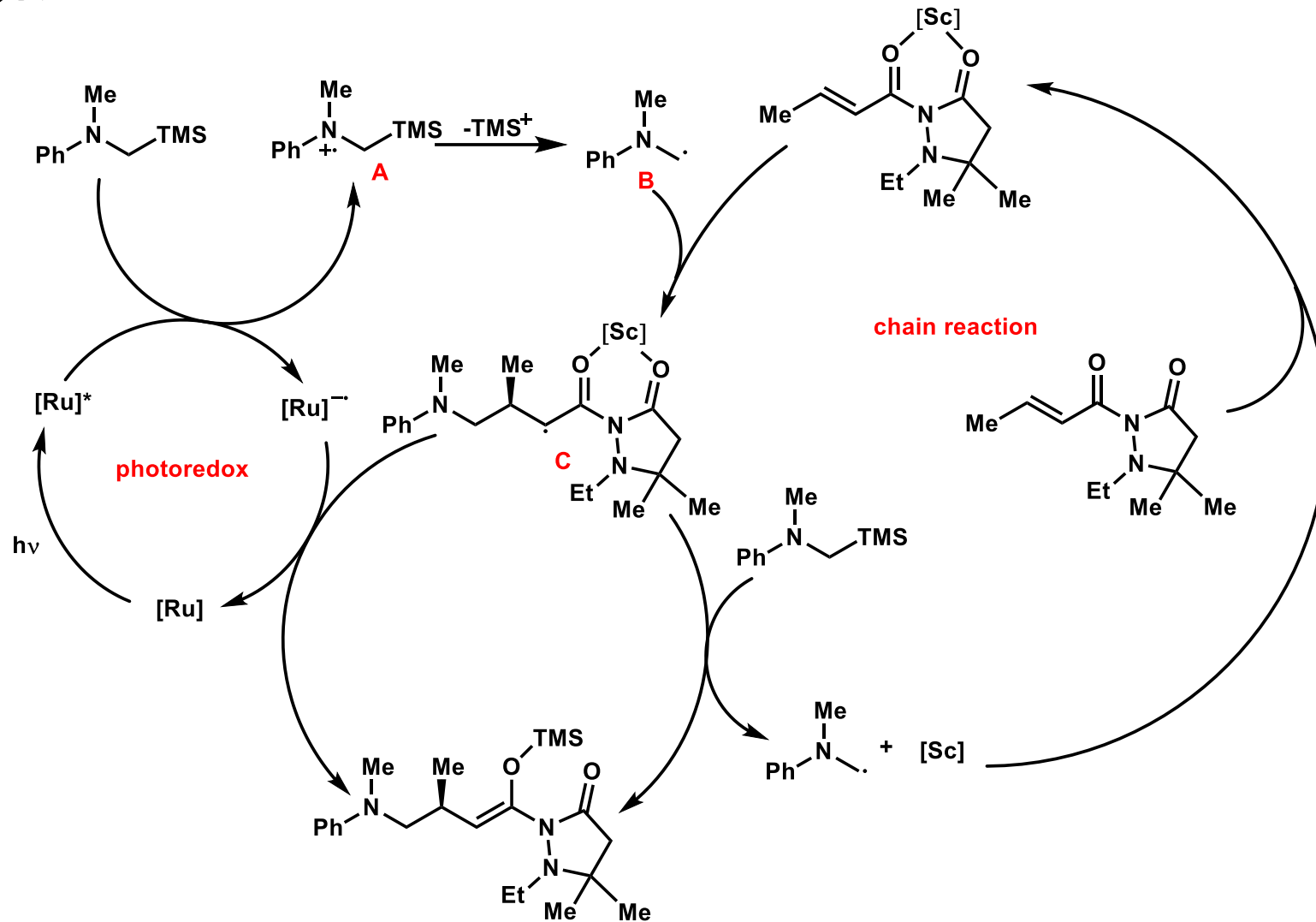
氯离子效应

Entry	Lewis acid	Additive	Time	Yield(%)
1	none	none	3 h	25
2	none	30 mol% $\text{Bu}_4\text{N}^+\text{Cl}^-$	3 h	28
3	$\text{sBuPybox} \cdot \text{Sc}(\text{OTf})_3$	none	2 h	67
4	$\text{sBuPybox} \cdot \text{Sc}(\text{OTf})_3$	30 mol% $\text{Bu}_4\text{N}^+\text{Cl}^-$	2 h	98
5	$\text{sBuPybox} \cdot \text{ScCl}_3$	none	6 h	61

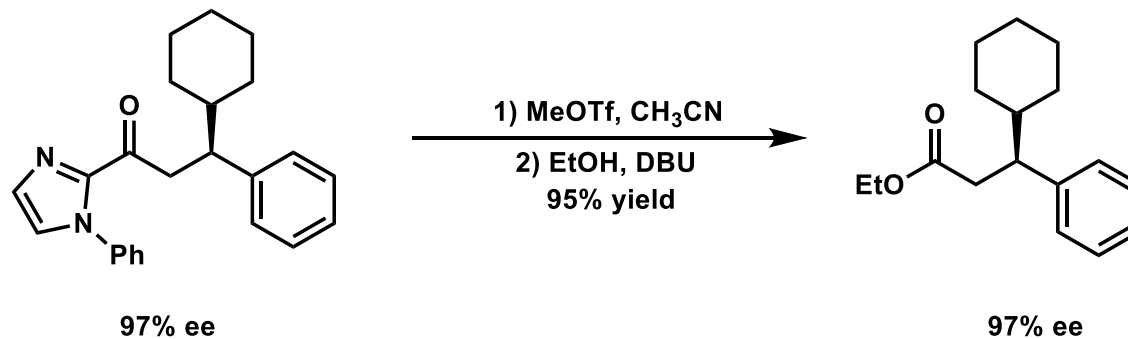
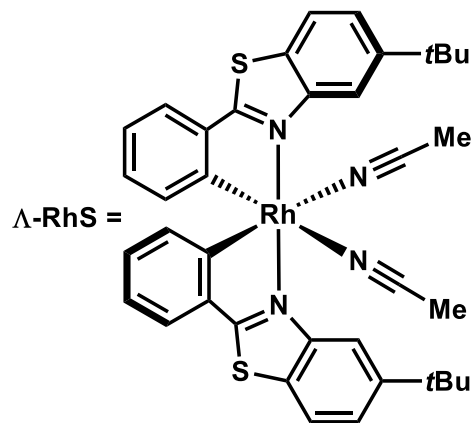
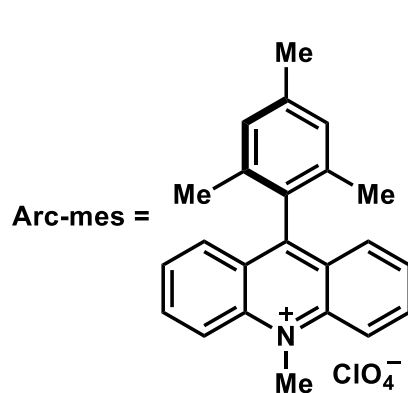
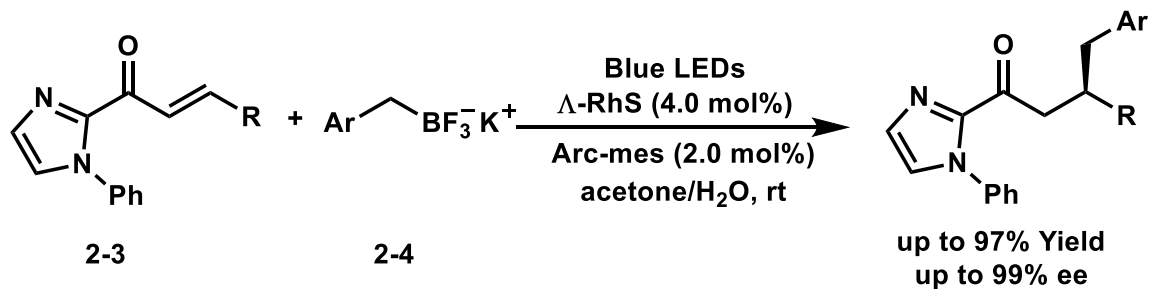
氯离子促进了路易斯酸的催化循环次数

手性路易斯酸催化的共轭加成反应

反应机理

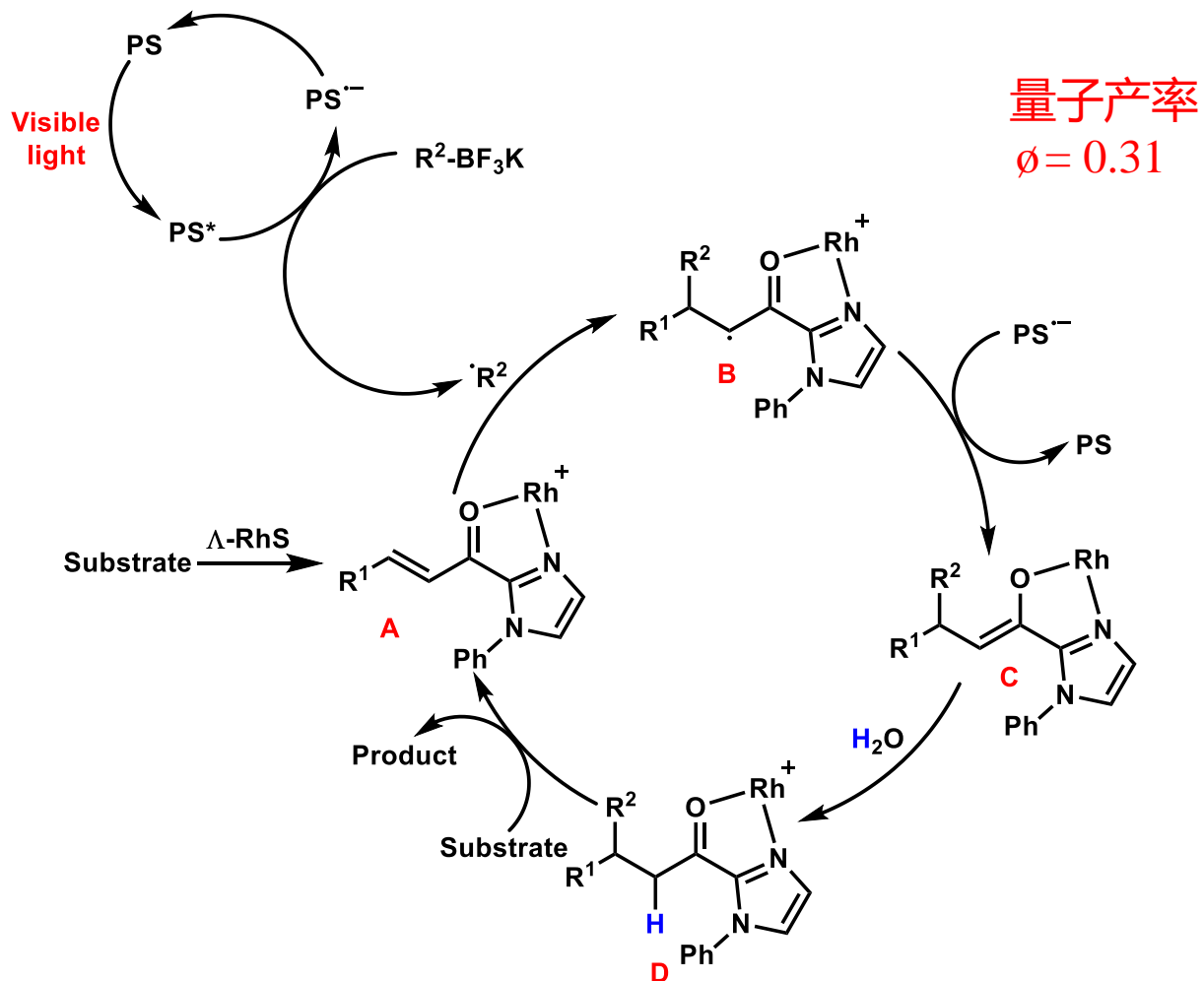


手性路易斯酸催化的共轭加成反应



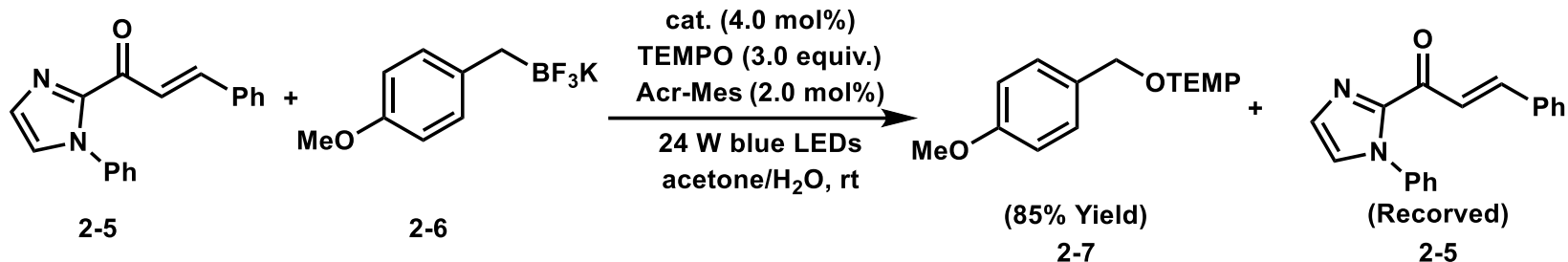
手性路易斯酸催化的共轭加成反应

反应机理

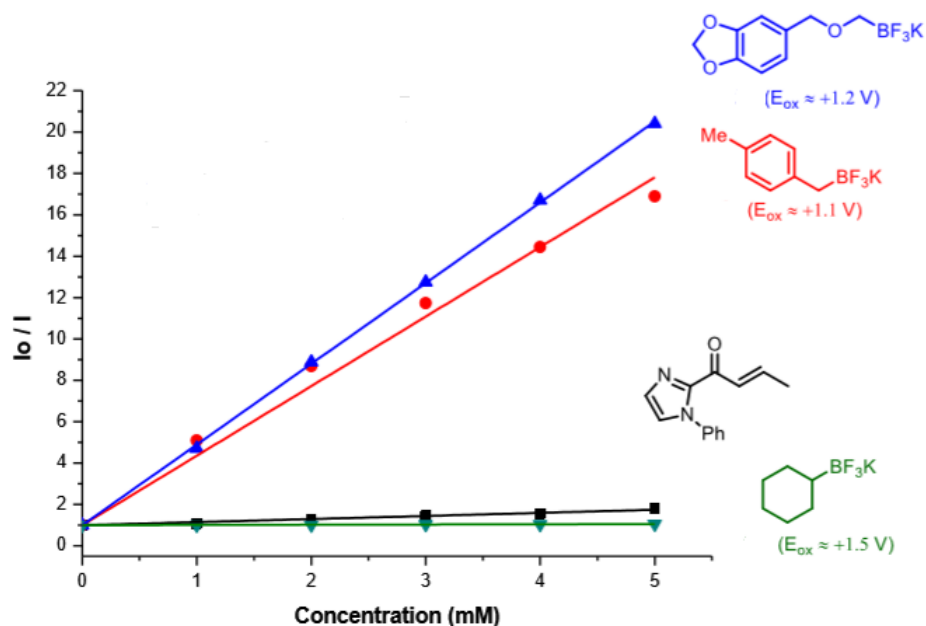


手性路易斯酸催化的共轭加成反应

对照试验

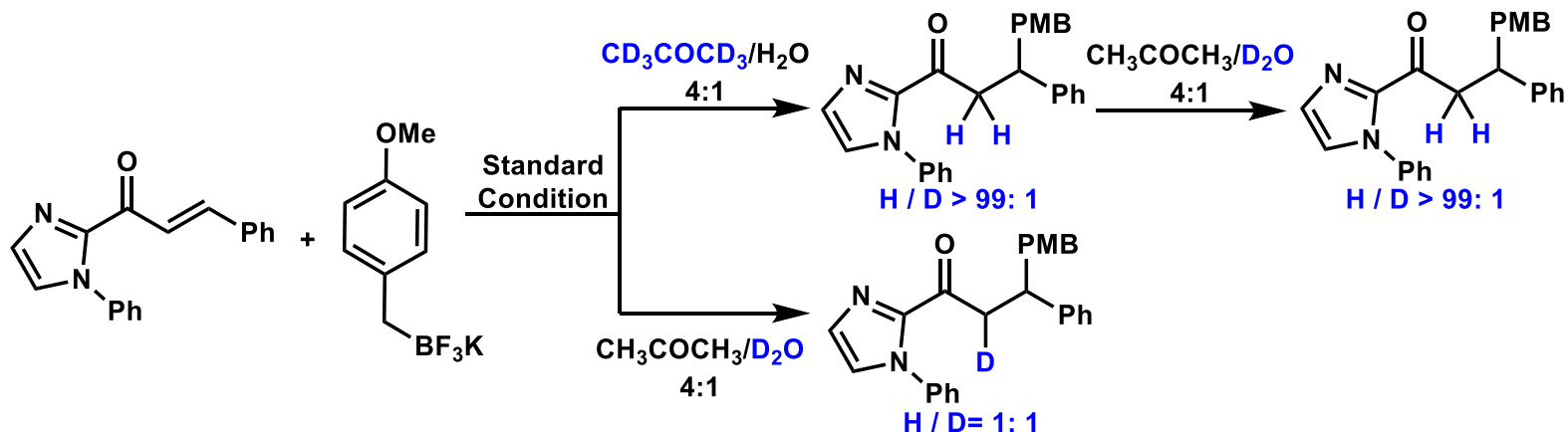


荧光淬灭实验

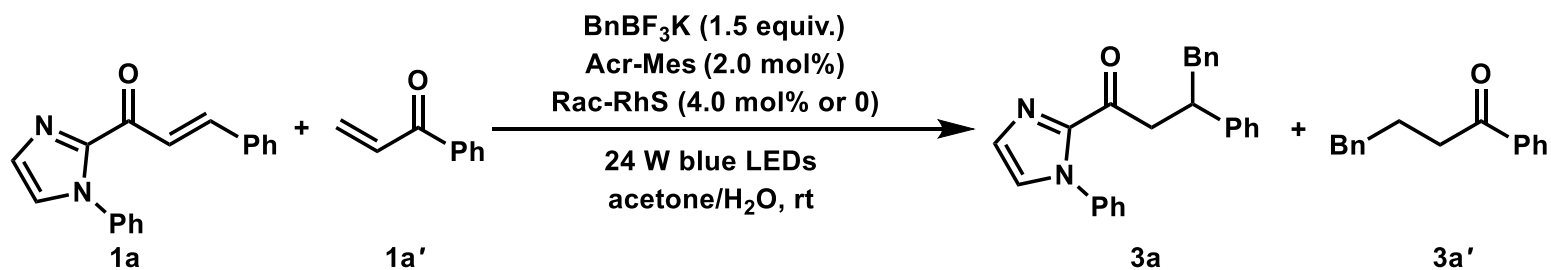


手性路易斯酸催化的共轭加成反应

对照试验



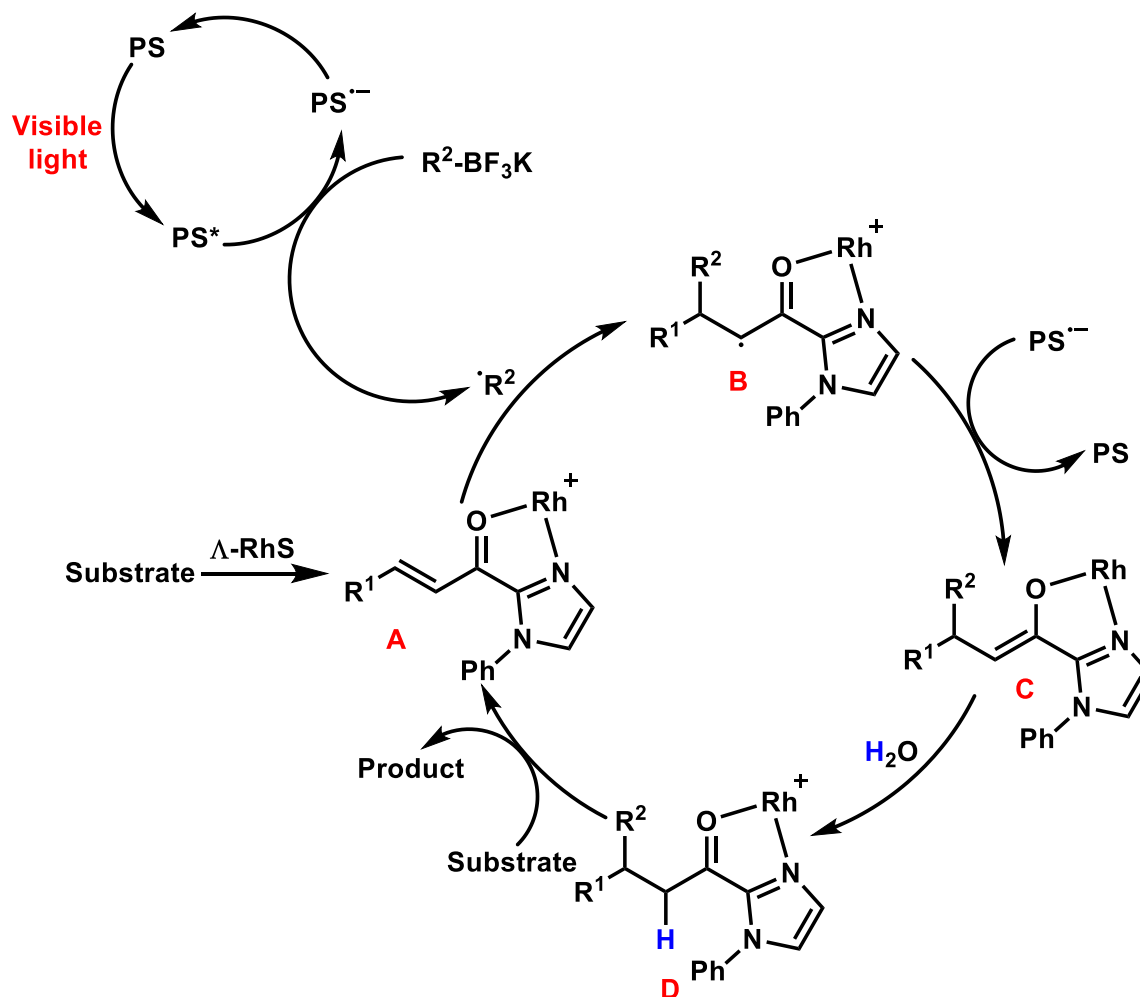
竞争试验



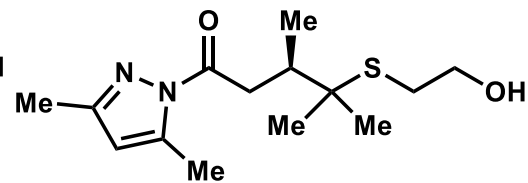
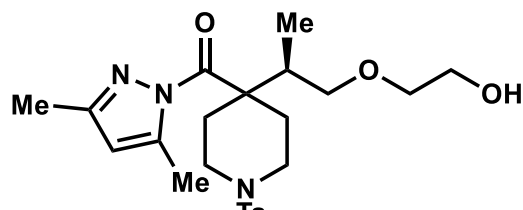
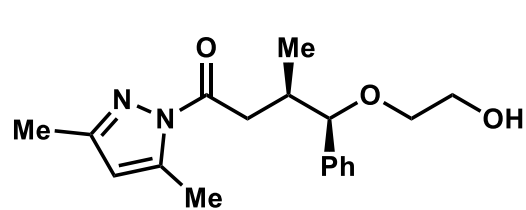
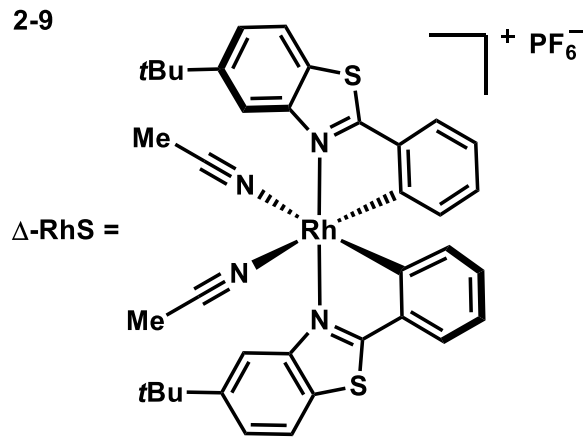
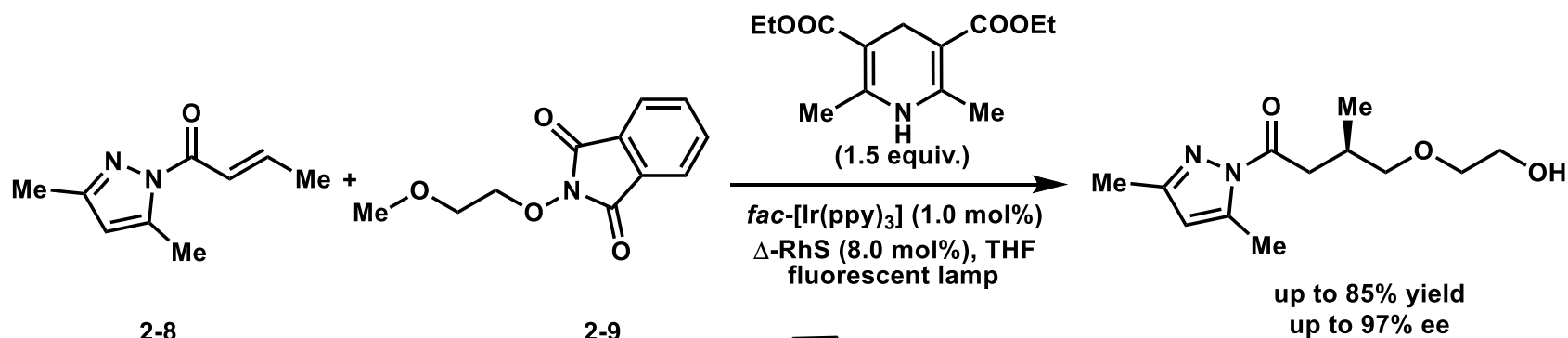
RhS (%)	Yield (3a)	Yield (3a')	3a:3a'
0	8%	20%	0.4: 1
4.0	55%	not observed	> 550: 1

手性路易斯酸催化的共轭加成反应

反应机理

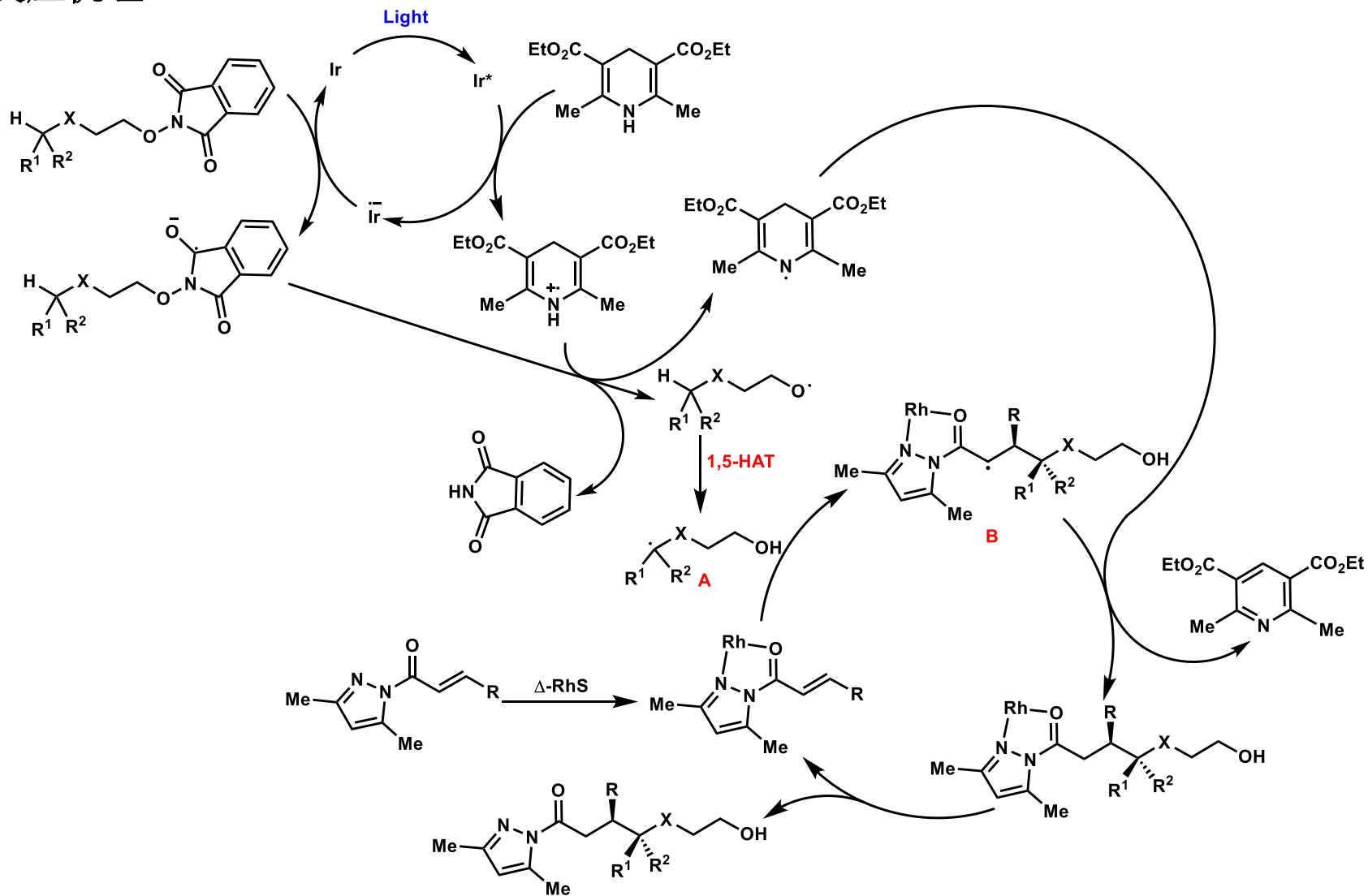


手性路易斯酸催化的共轭加成反应

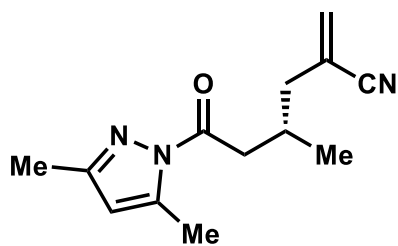
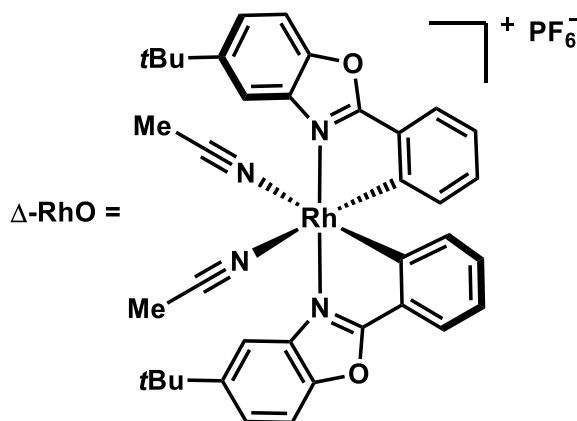
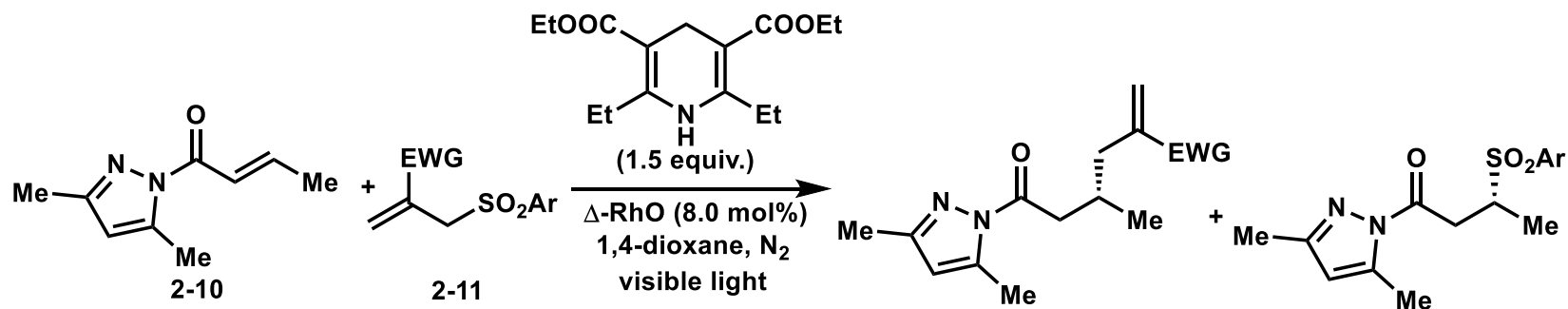


手性路易斯酸催化的共轭加成反应

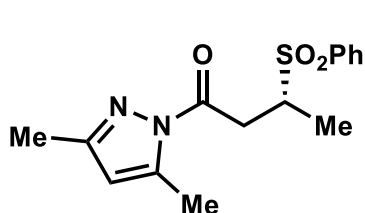
反应机理



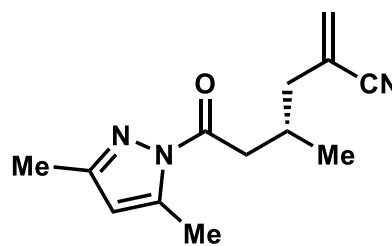
手性路易斯酸催化的共轭加成反应



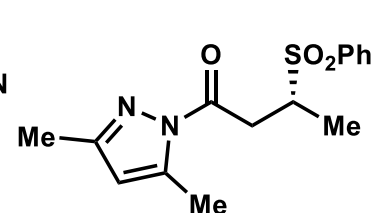
85% yield, 96% ee



92% yield, 85% ee



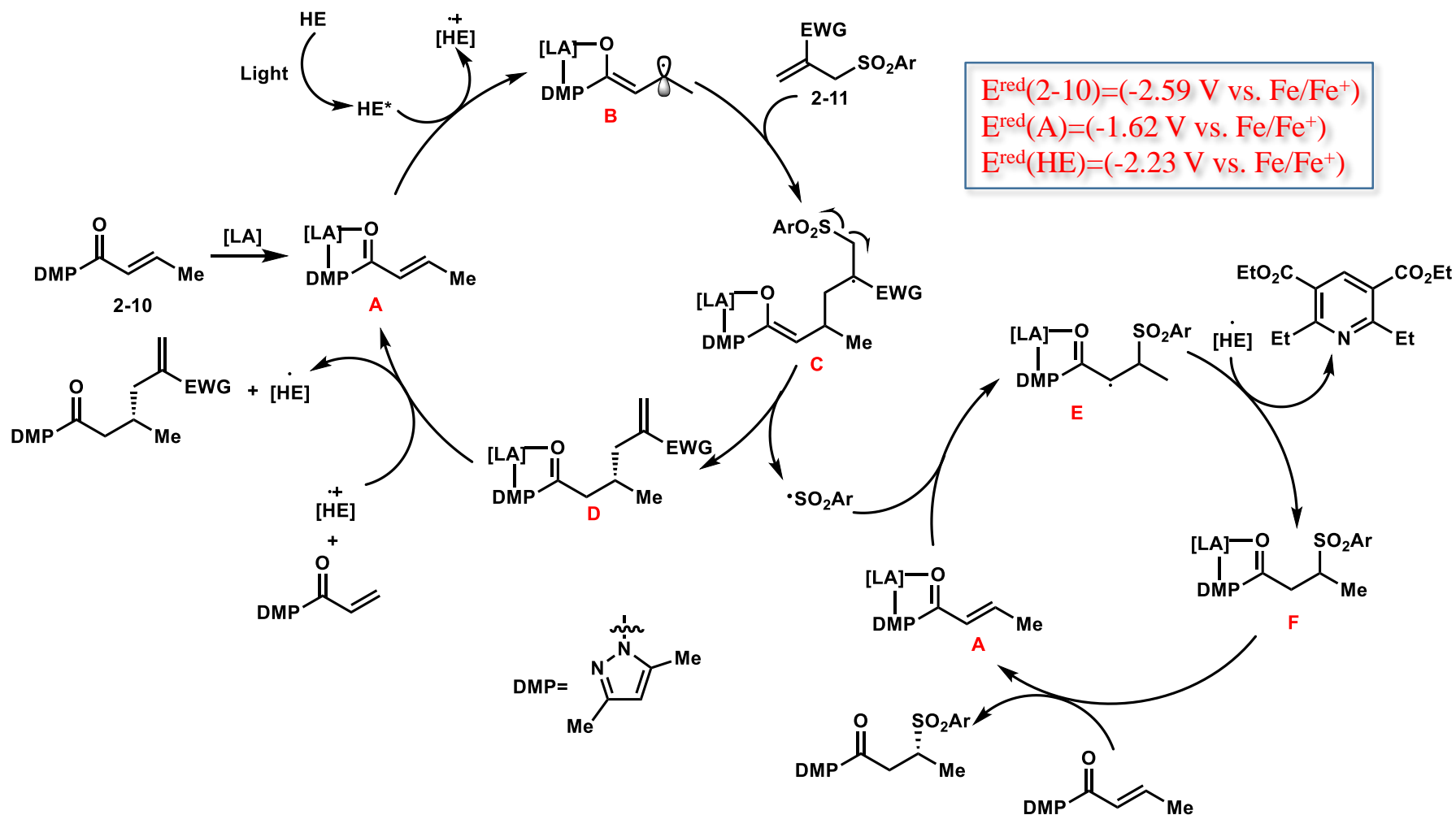
79% yield, 95% ee



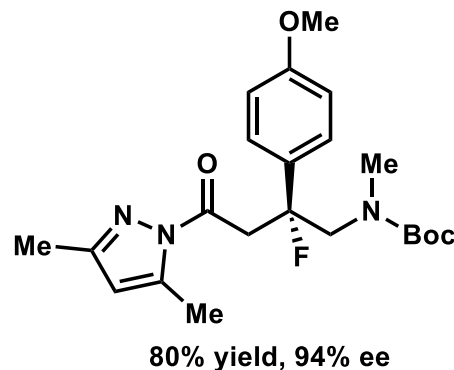
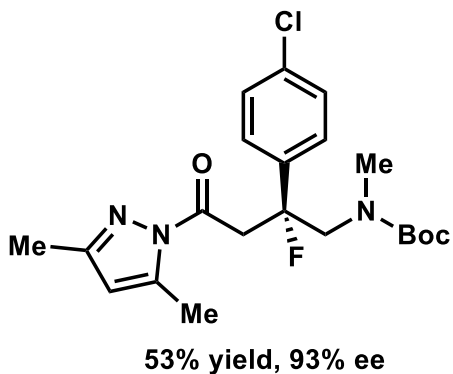
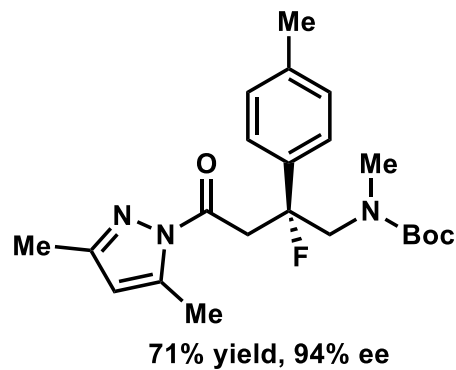
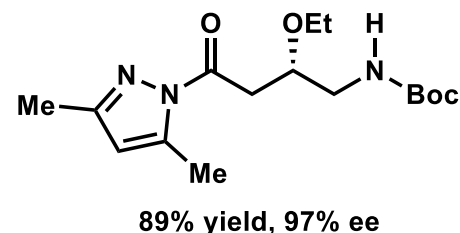
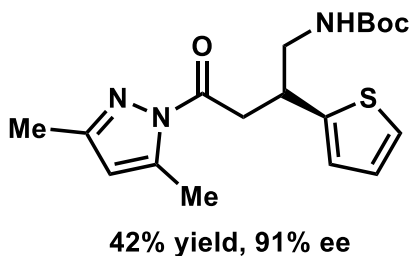
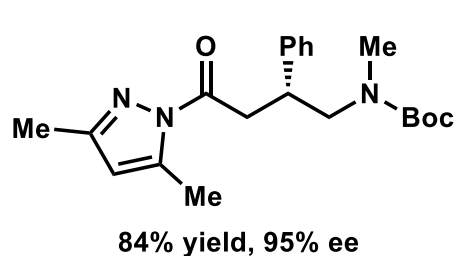
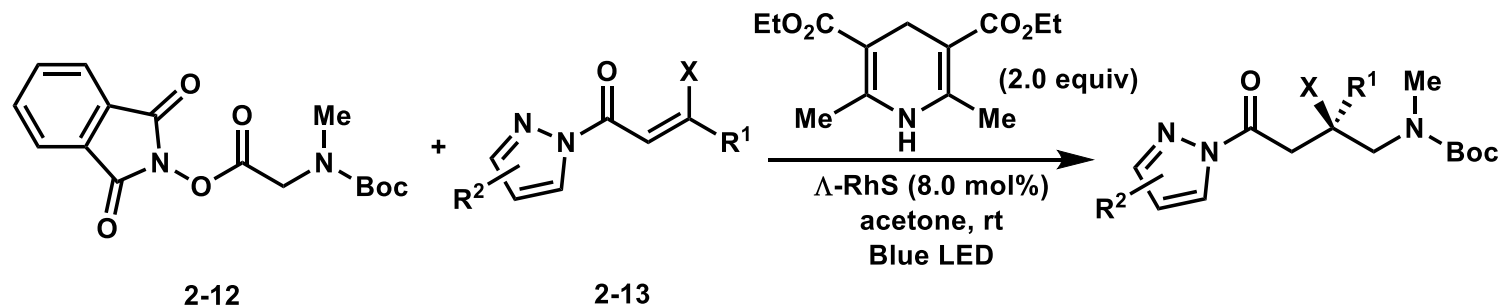
68% yield, 84% ee

手性路易斯酸催化的共轭加成反应

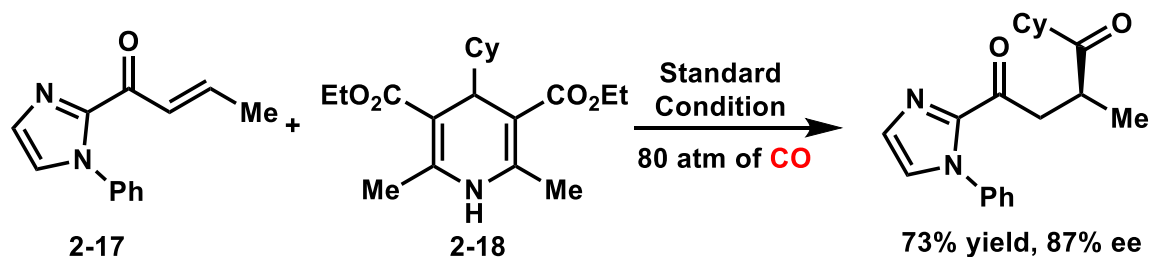
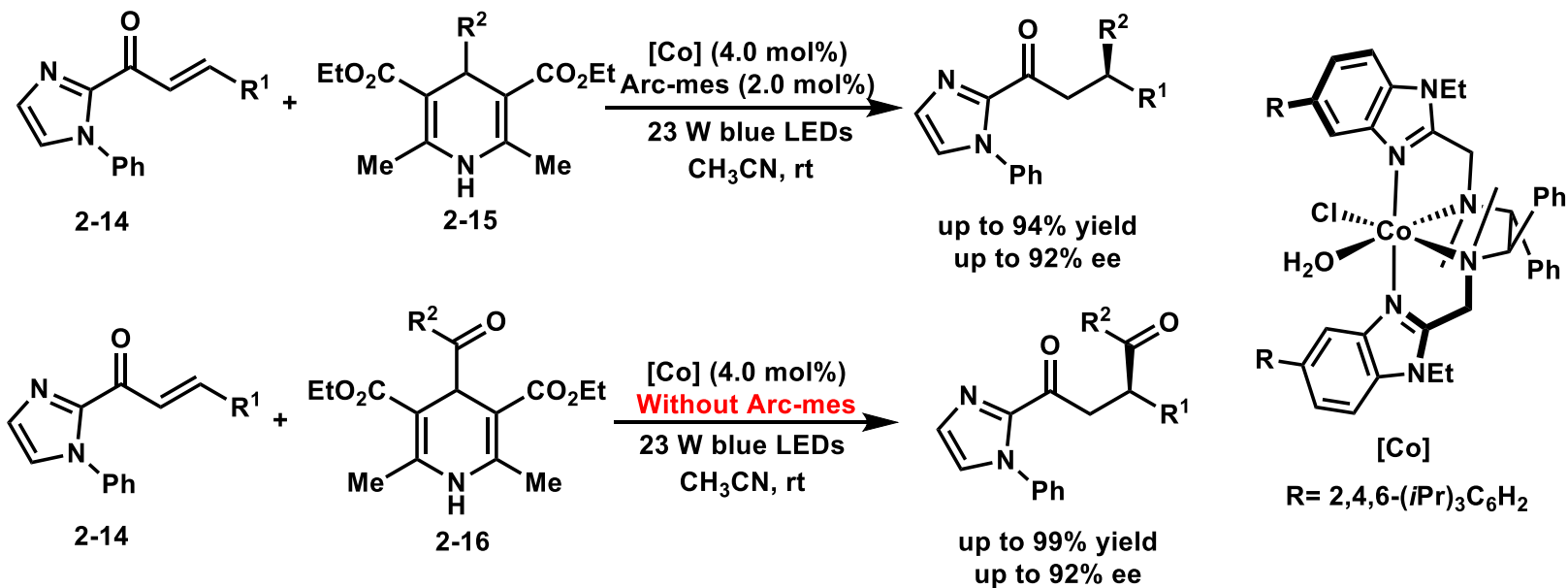
反应机理



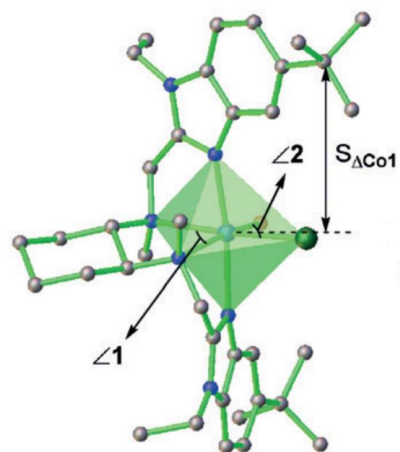
手性路易斯酸催化的共轭加成反应



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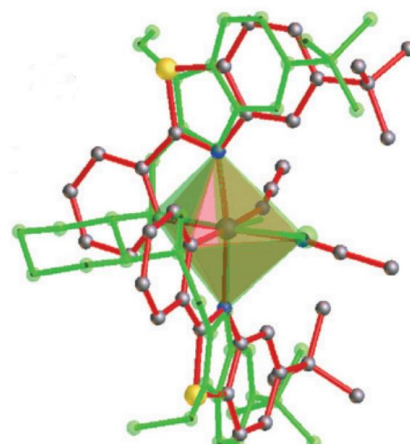


手性路易斯酸催化的共轭加成反应



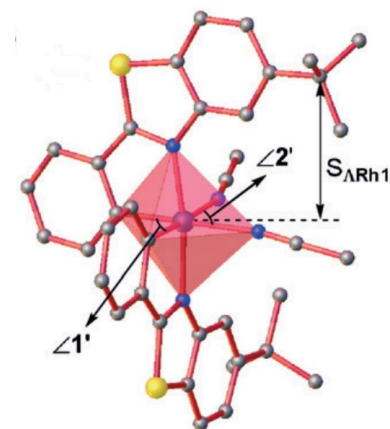
ΔCo1 (green)

$S_{\Delta\text{Co1}} = 5.05 \text{ \AA}$ vs $S_{\Delta\text{Rh1}} = 4.57 \text{ \AA}$



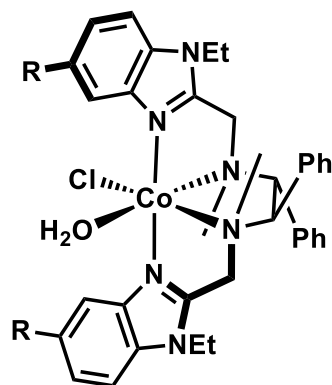
overlay of ΔCo1 and ΔRh1

$\angle 1 = 81.1^\circ$ vs $\angle 1' = 87.8^\circ$



ΔRh1 (red)

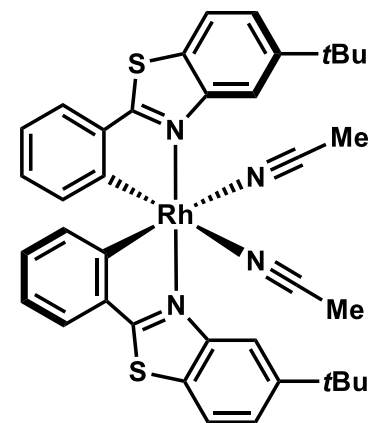
$\angle 2 = 95.8^\circ$ vs $\angle 2' = 88.3^\circ$



[Co]

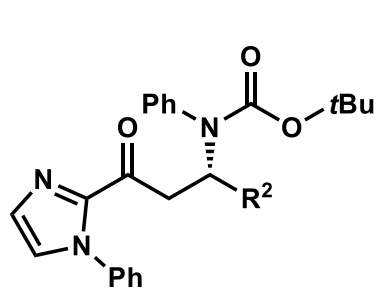
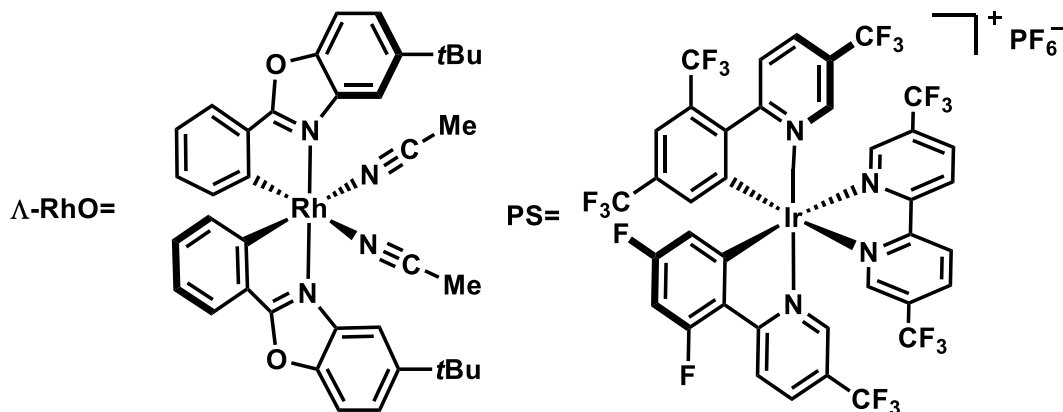
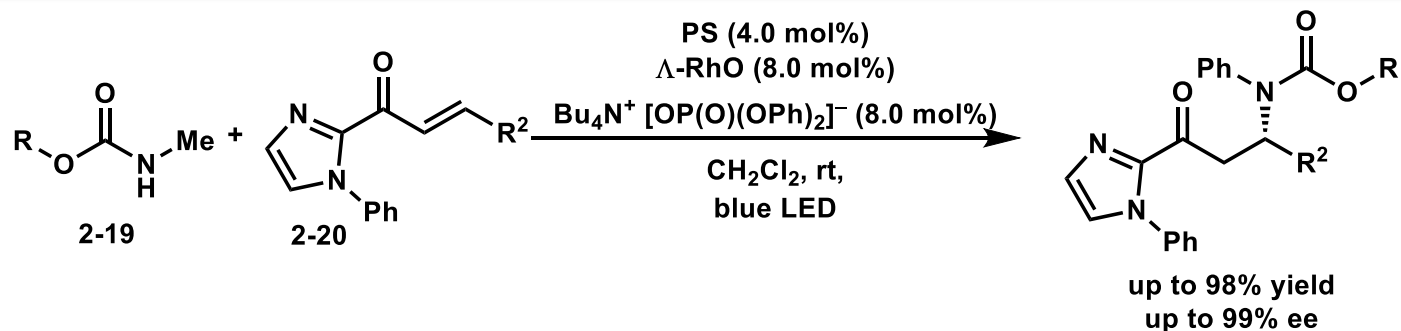
R = 2,4,6-(iPr)₃C₆H₂

Co cat	yield(%)	ee
R = tBu	80	40
R = 2,4,6-(iPr) ₃ C ₆ H ₂	92	92

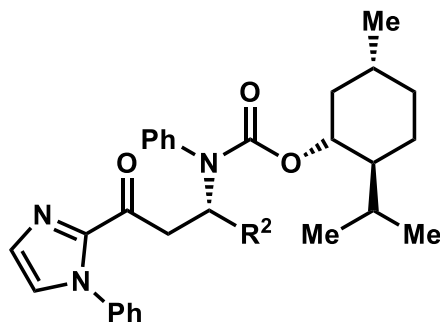


$\Delta\text{-RhS}$

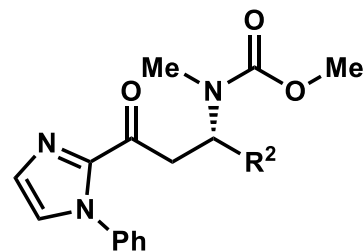
手性路易斯酸催化的共轭加成反应



92% yield, 98% ee



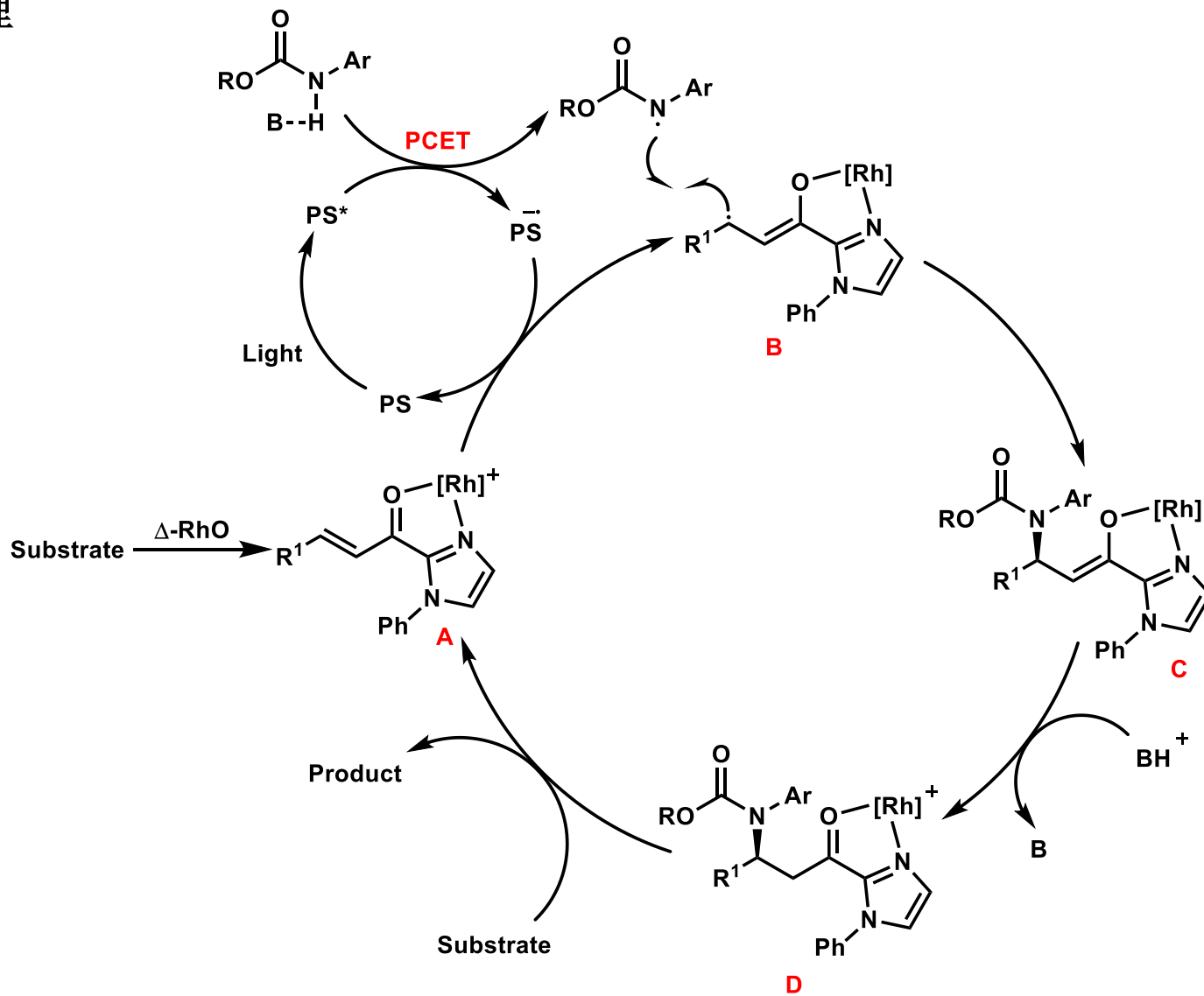
85% yield, > 99:1 d.r.



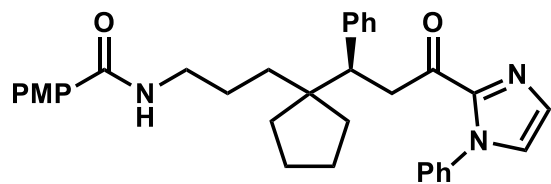
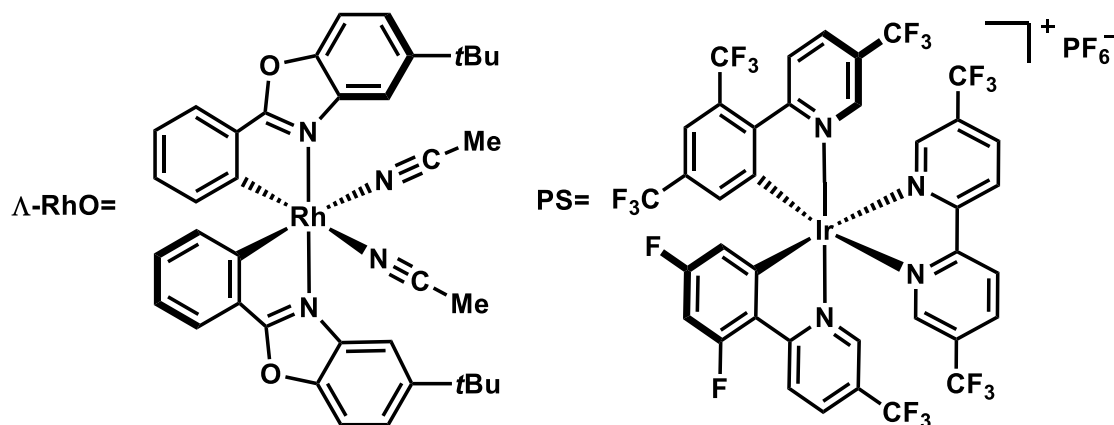
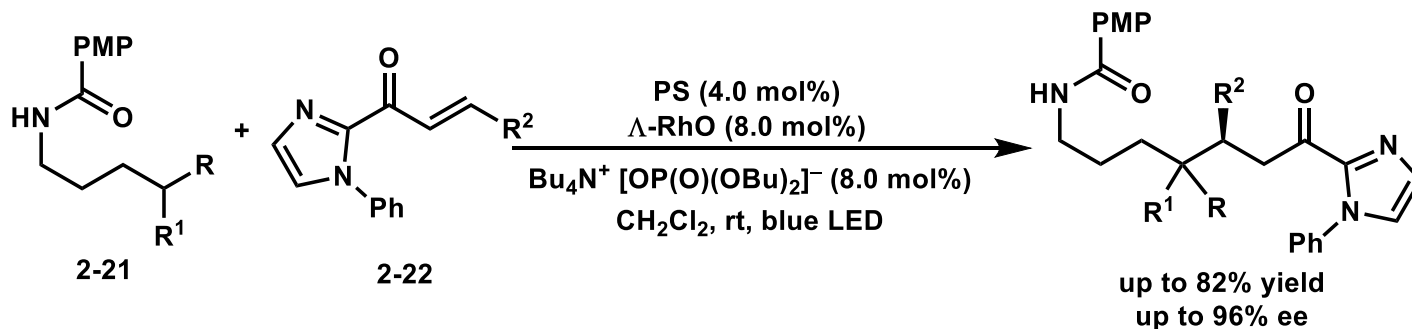
0% yield, 0% ee

手性路易斯酸催化的共轭加成反应

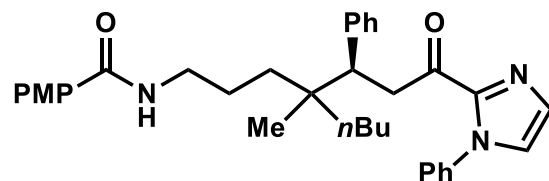
反应机理



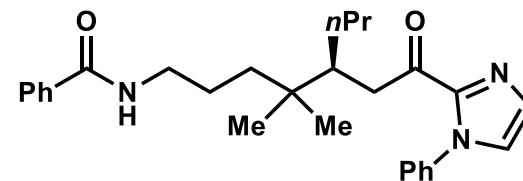
手性路易斯酸催化的共轭加成反应



64% yield, 97% ee



62% yield, 1.2:1, 95% ee, 93% ee

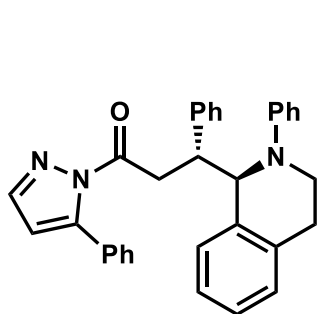
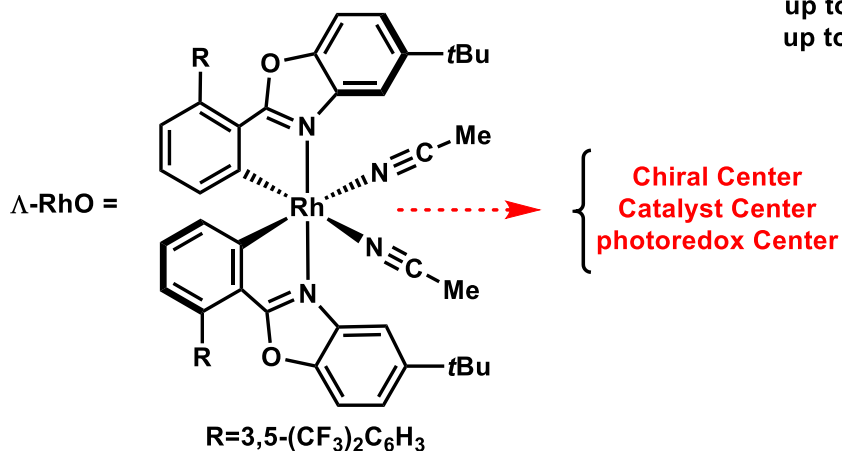


0% yield, 0% ee

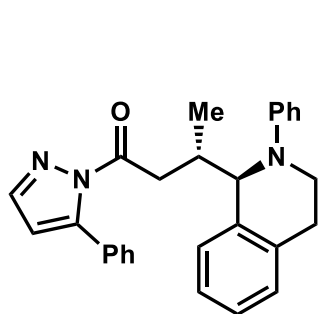
手性路易斯酸催化的共轭加成反应



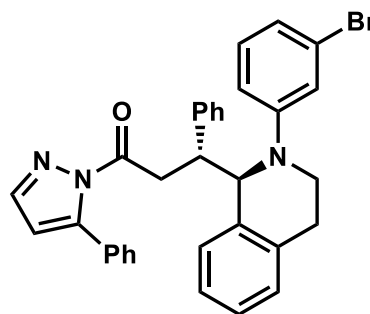
up to 96% yield
up to 82% dr
up to 99% ee



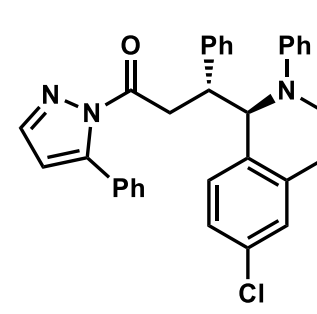
90% yield, 83:17 d.r. 95% ee



54% yield, 66:34 d.r. 97% ee



80% yield, 81:19 d.r. 93% ee

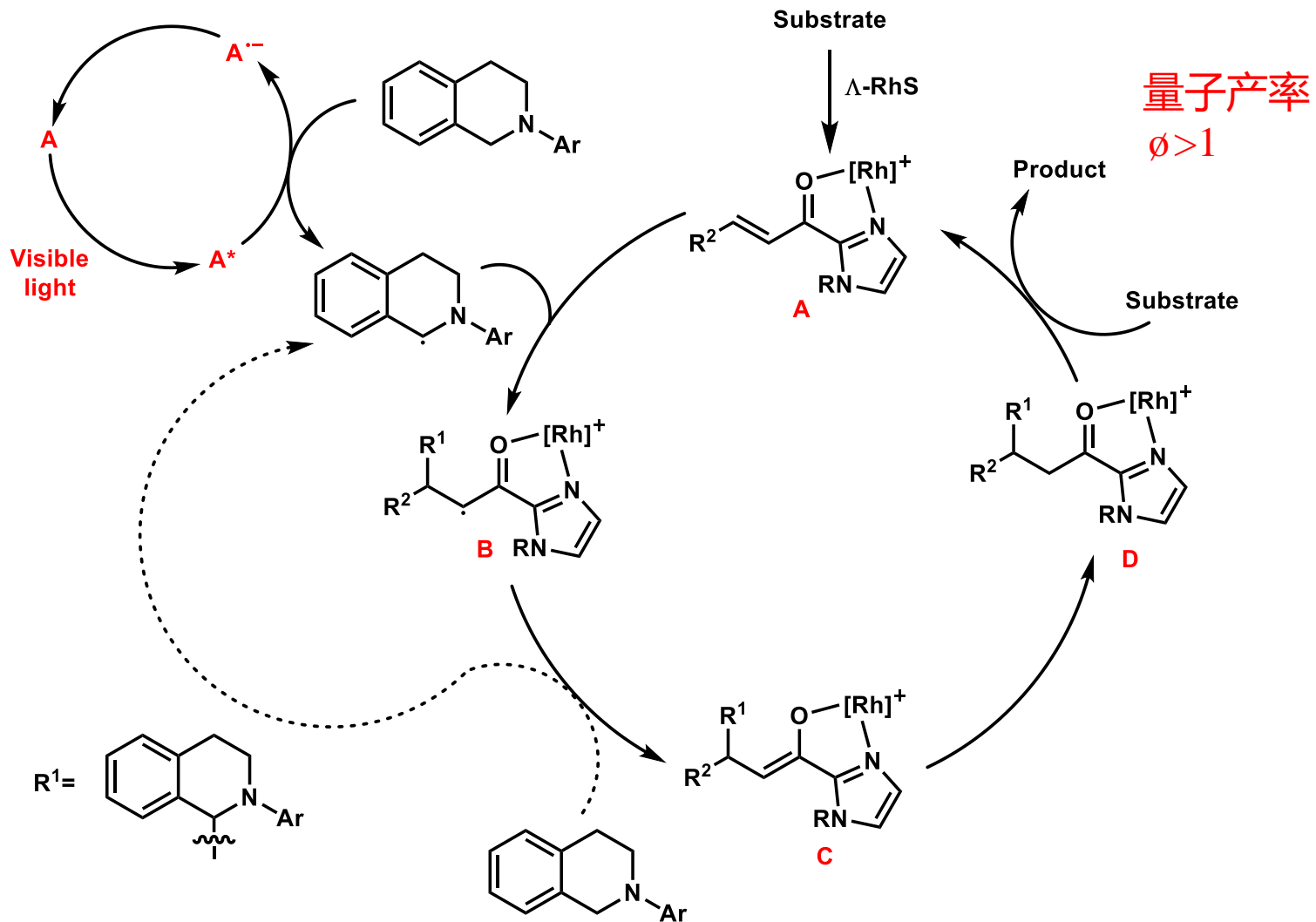


96% yield, 59:41 d.r. 99% ee

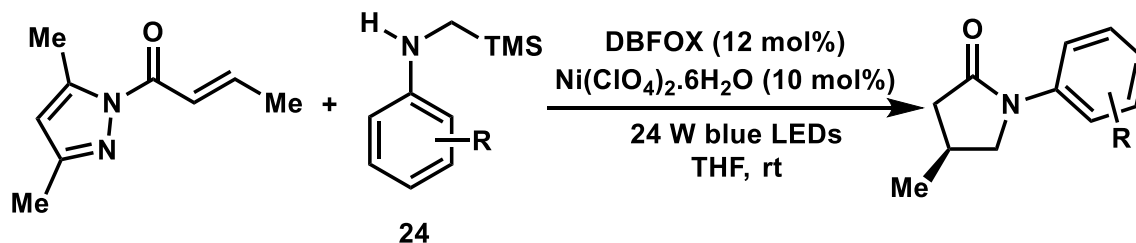
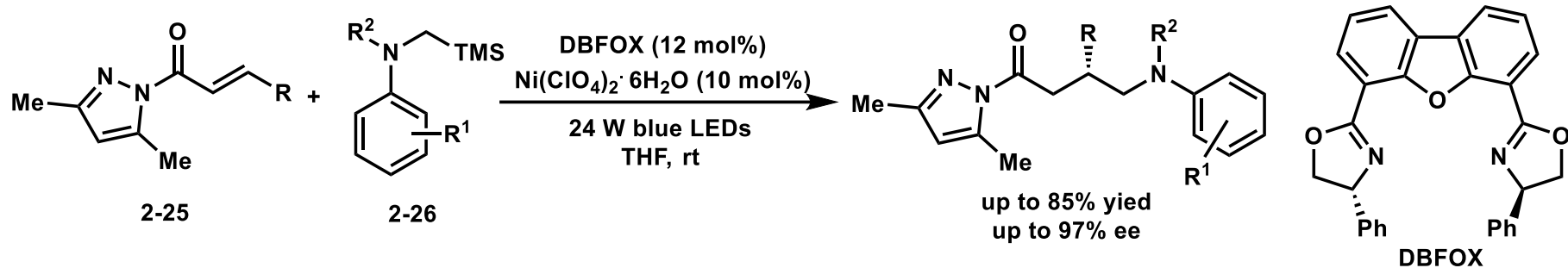
Huo, H.; Shen, X.; Wang, C.; Zhang, L.; Röse, P.; Chen, L.; Harms, K.; Marsch, M.; Hilt, M.; Meggers, E. *Nature*, **2014**, *515*, 100.
Lin, S.; Sun, G.; Kang, Q. *Chem. Commun.* **2017**, *53*, 7665.

手性路易斯酸催化的共轭加成反应

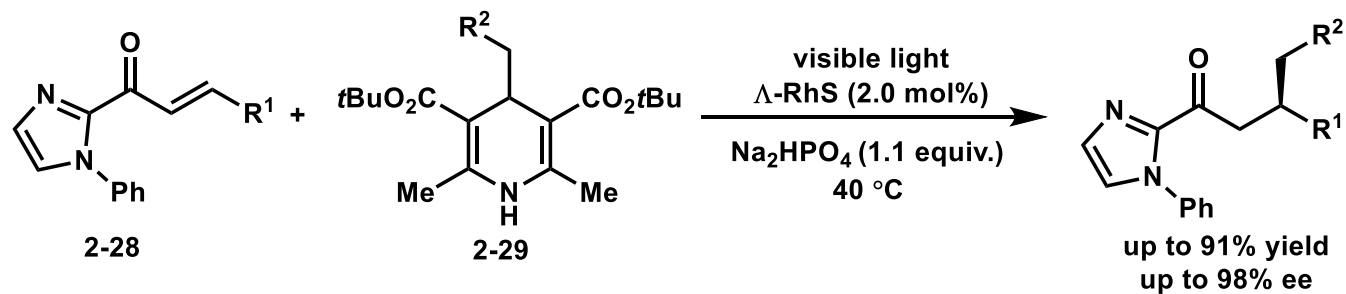
反应机理



手性路易斯酸催化的共轭加成反应

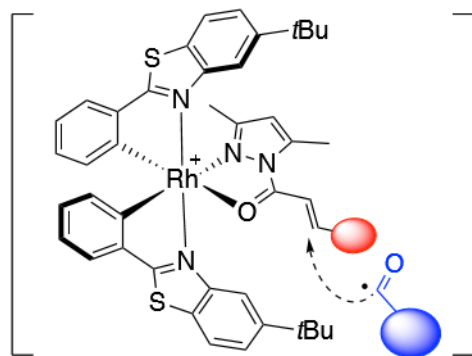
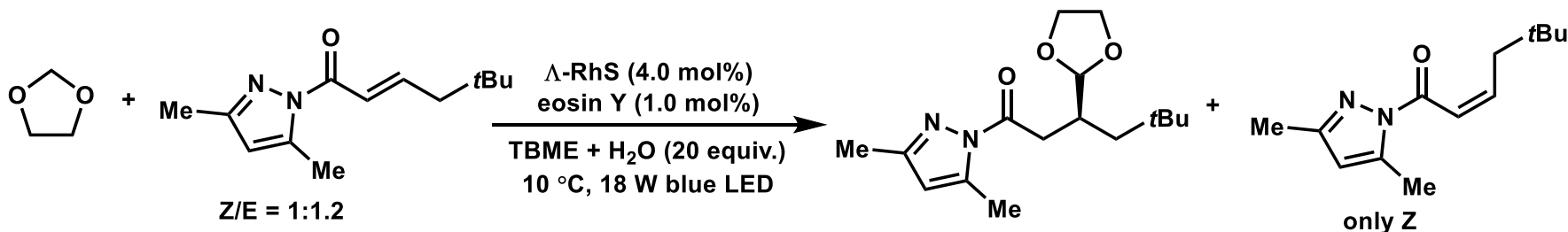
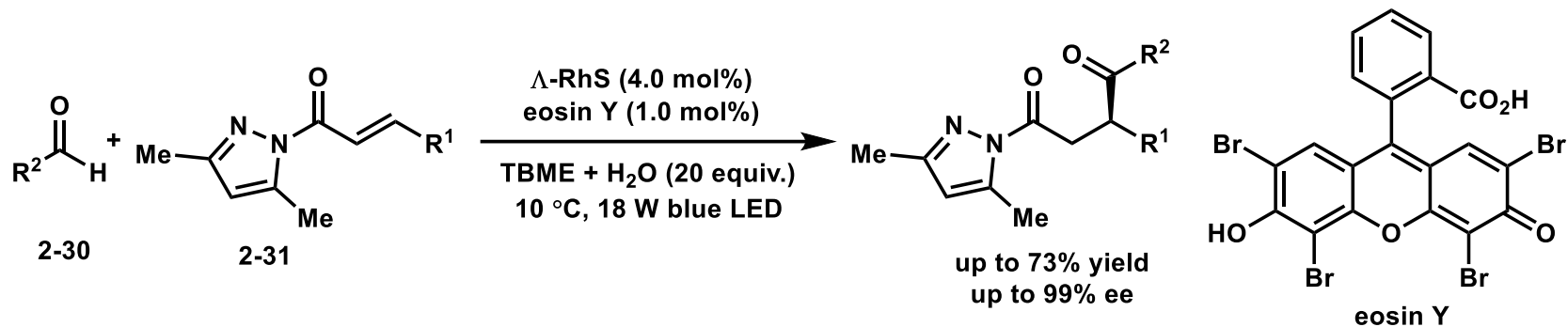


Shen, X.; Li, Y.; Wen, Z.; Cao, S.; Hou, X.; Gong, L. *Chem. Sci.* **2018**, *9*, 4562.



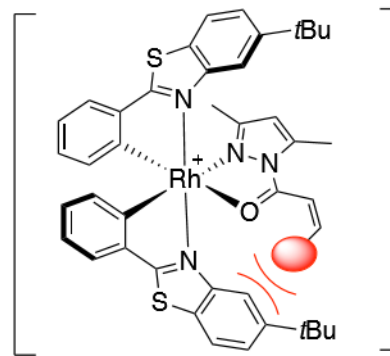
Assis, F. F.; Huang, X.; Akiyama, M.; Pilli, R. A.; Meggers, E. *J. Org. Chem.* **2018**, *83*, 10922–10932.

手性路易斯酸催化的共轭加成反应



Favored Transition state of *E* isomer

vs



Disfavored activation model of *Z* isomer

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1. 背景介绍

2. 内容

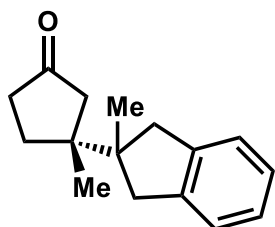
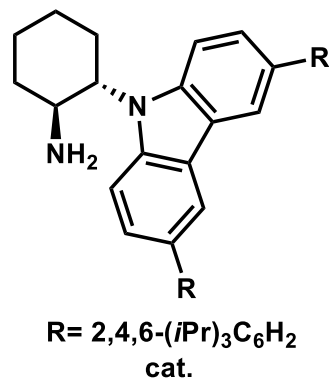
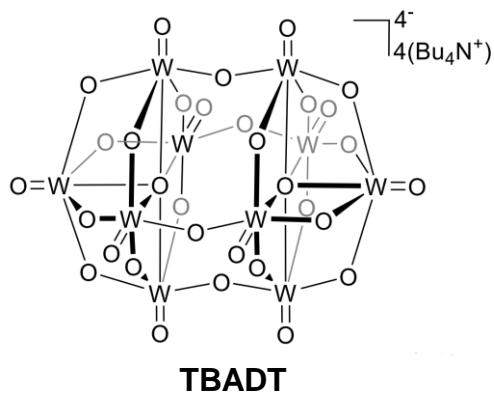
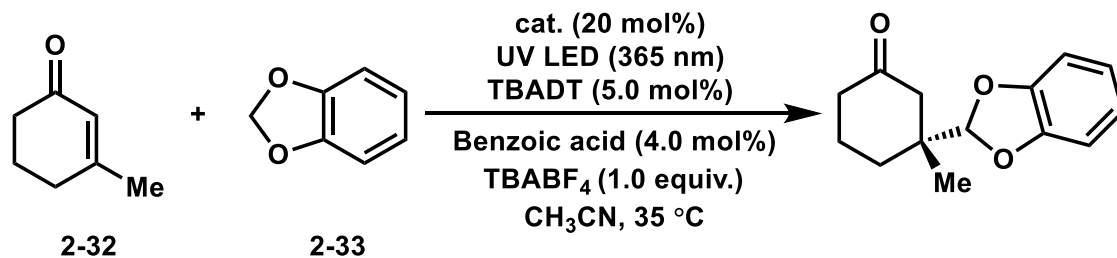
2.1 手性路易斯酸催化的共轭加成反应

2.2 手性胺催化的共轭加成反应

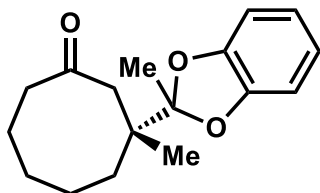
2.3 手性磷酸催化的共轭加成反应

3. 总结

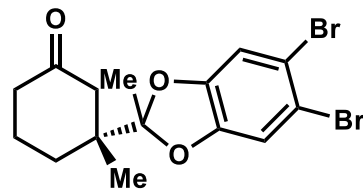
手性胺催化的共轭加成反应



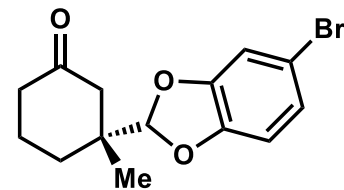
99% Yield, 88% ee



63% Yield, 91% ee



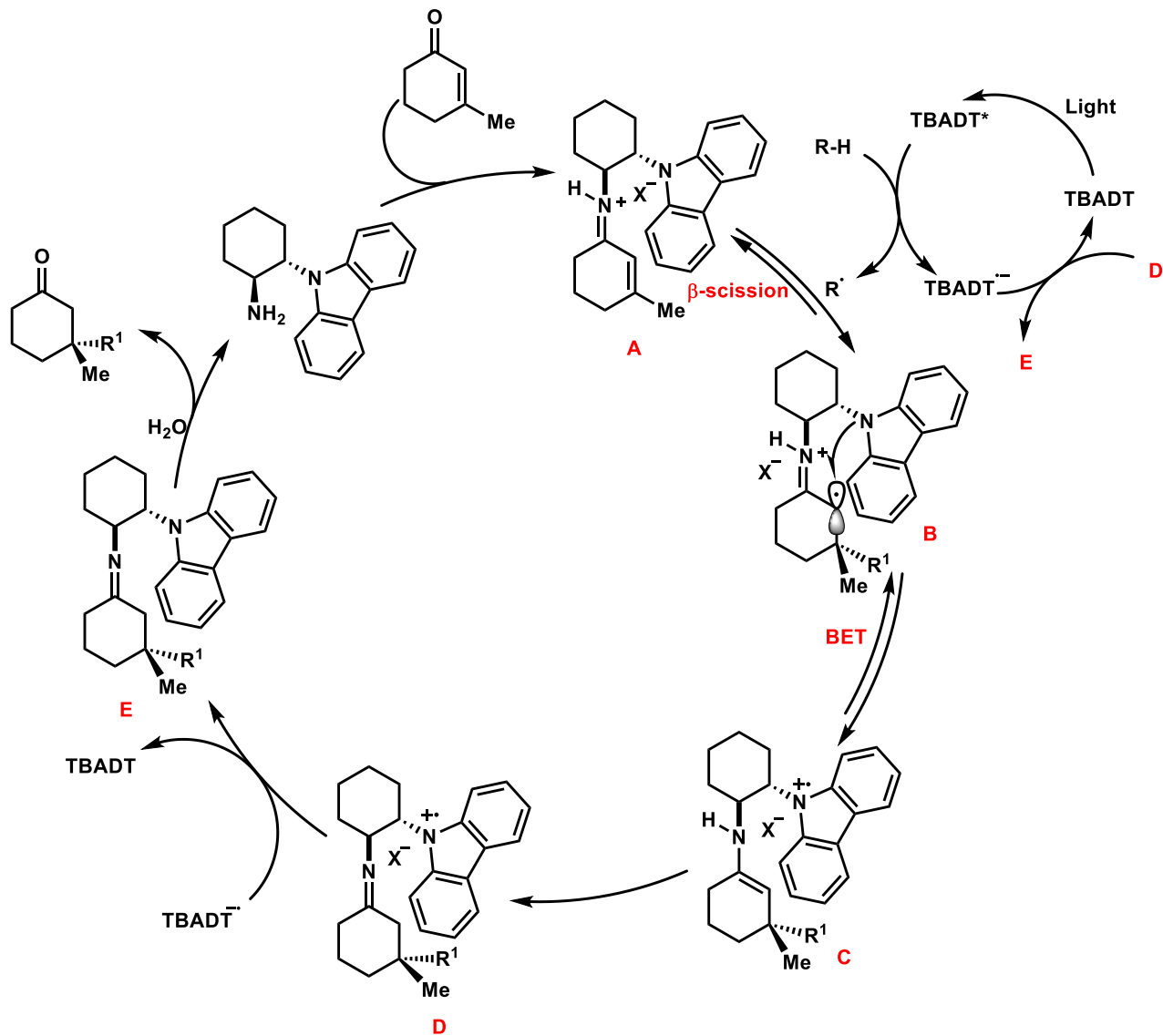
67% yield, 1.5:1 d.r., 93% ee



49% Yield, 91% ee

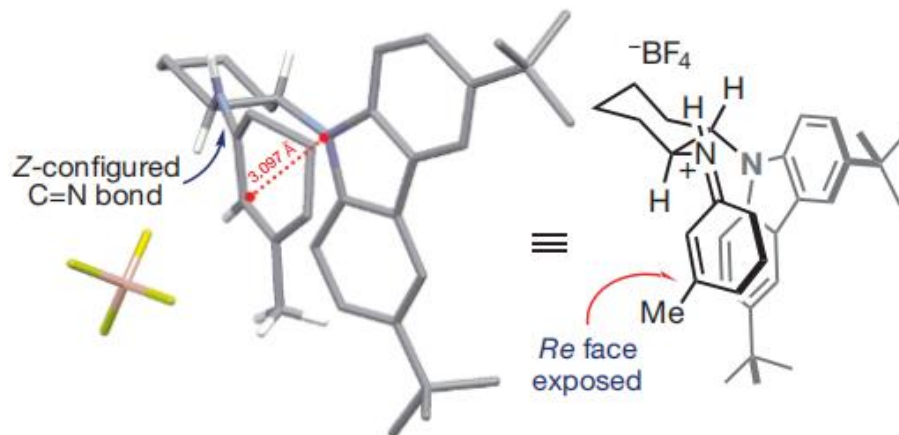
手性胺催化的共轭加成反应

反应机理

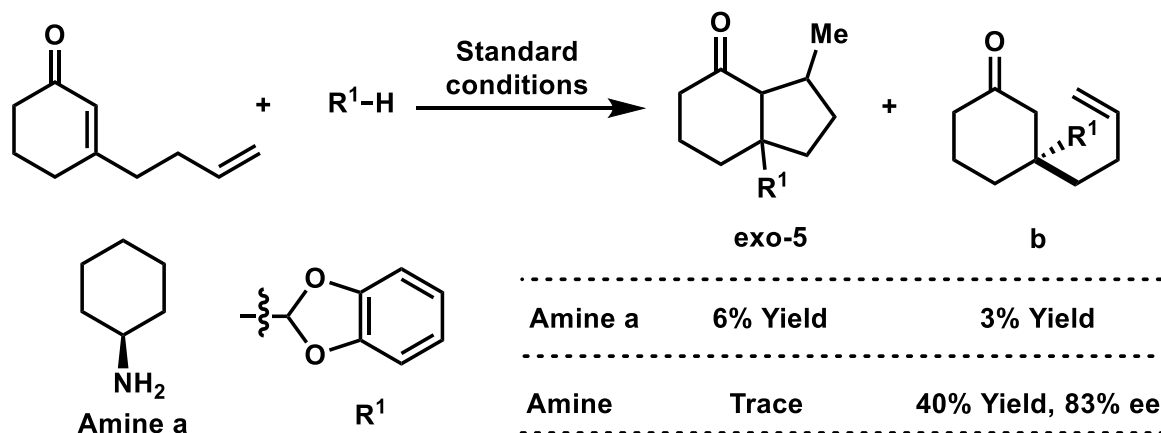


手性胺催化的共轭加成反应

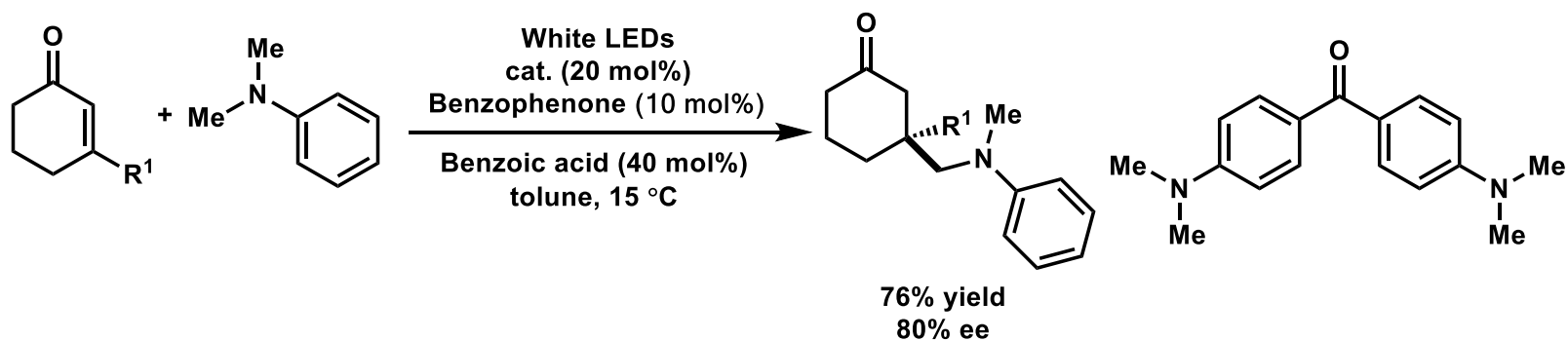
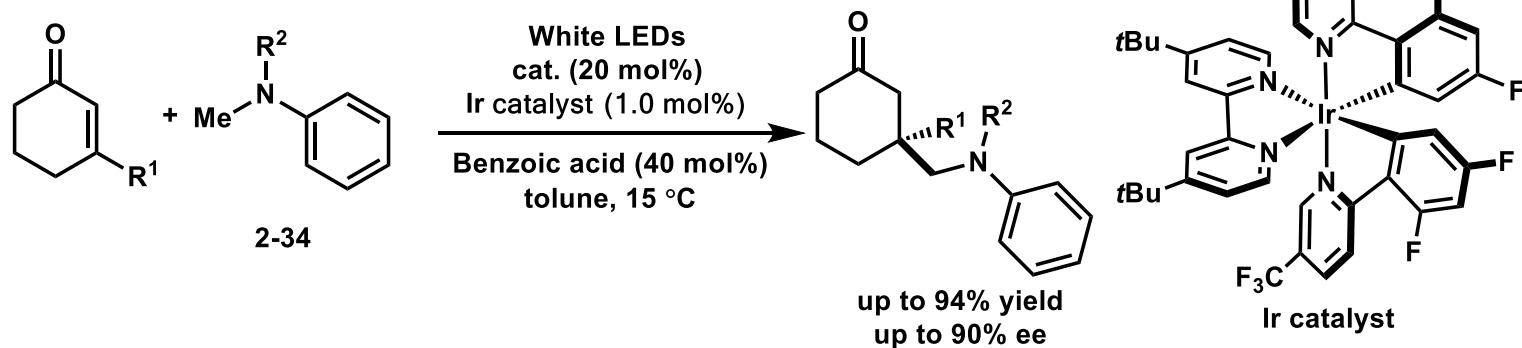
单晶结构



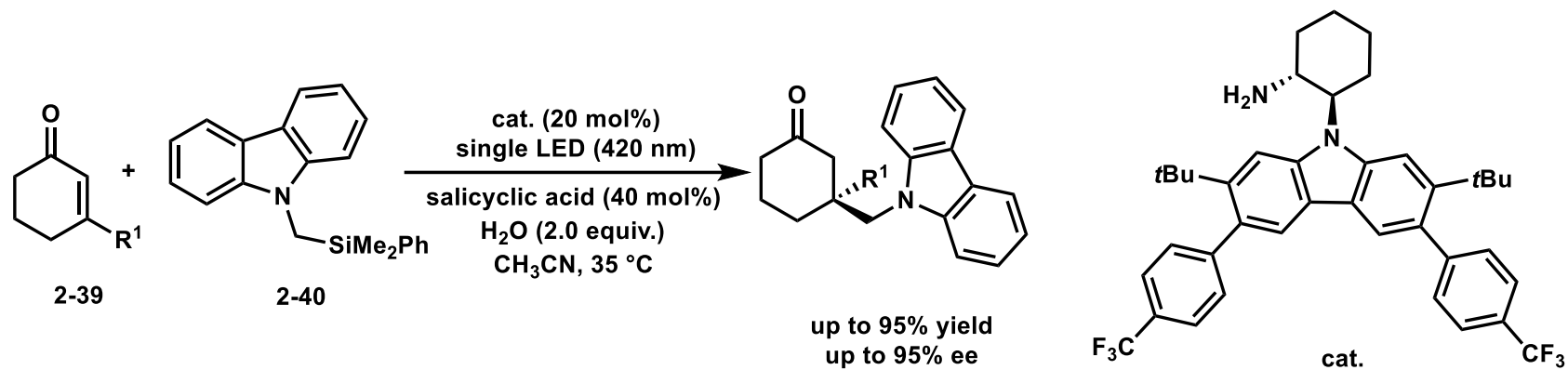
对照试验



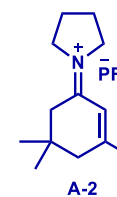
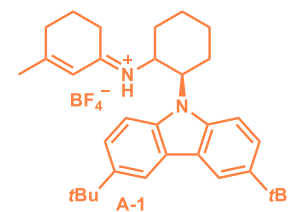
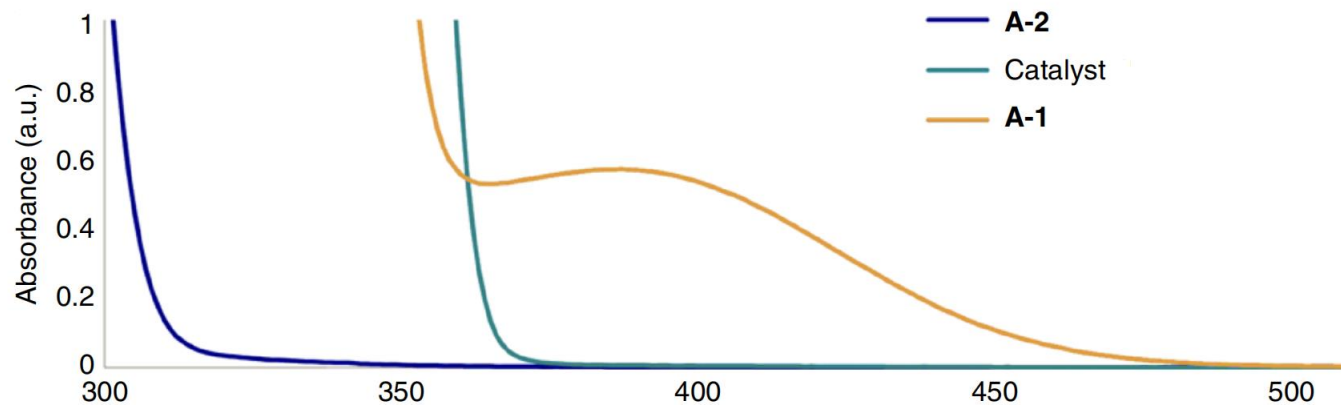
手性胺催化的共轭加成反应



手性胺催化的共轭加成反应

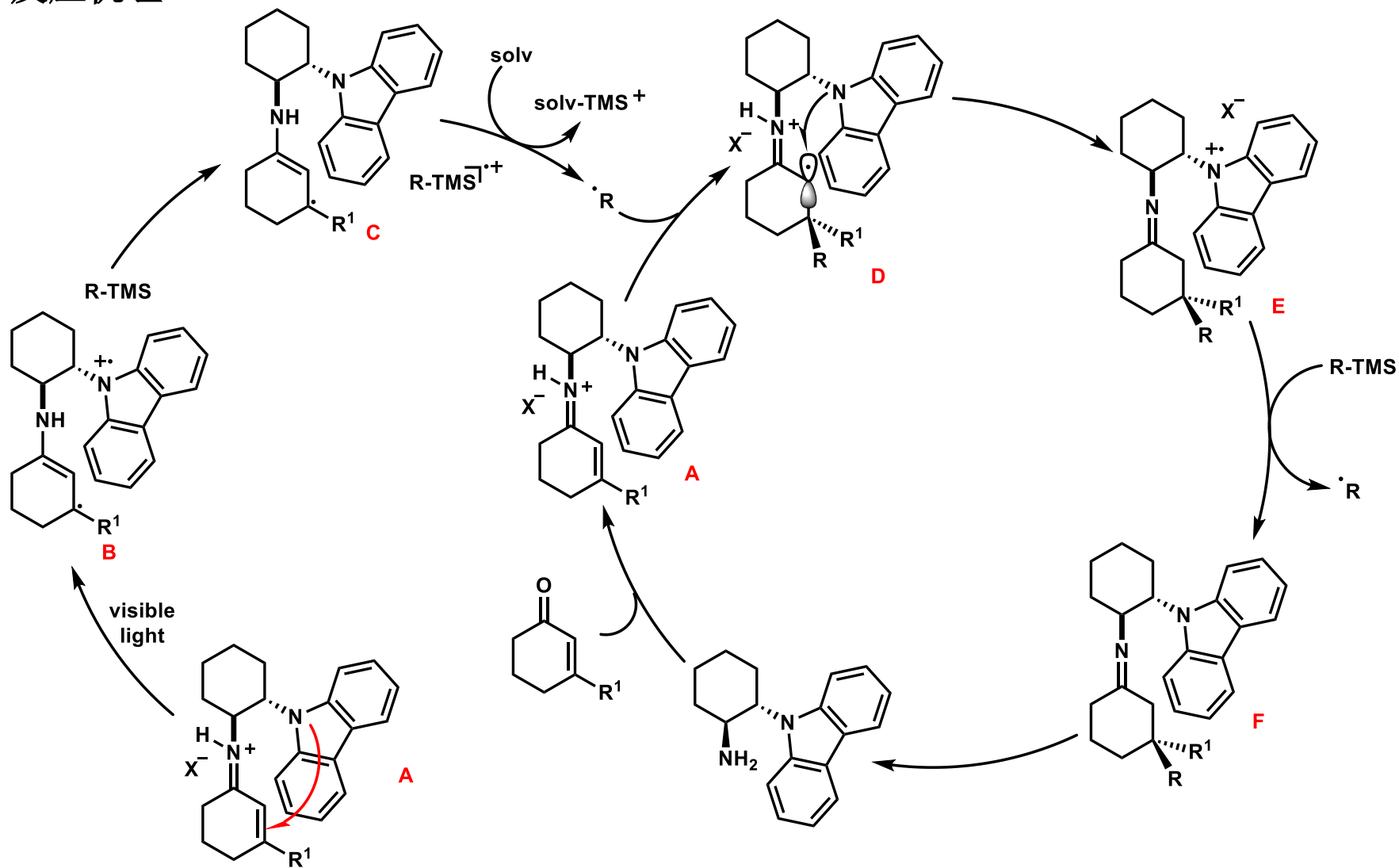


紫外可见光谱



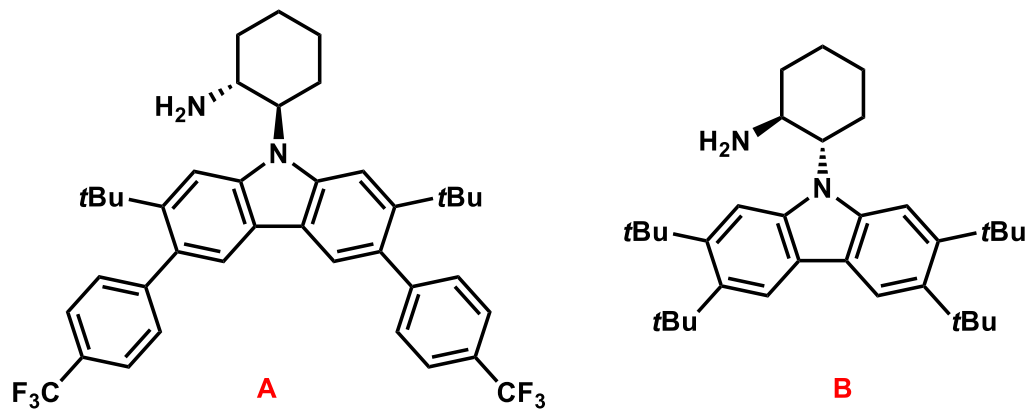
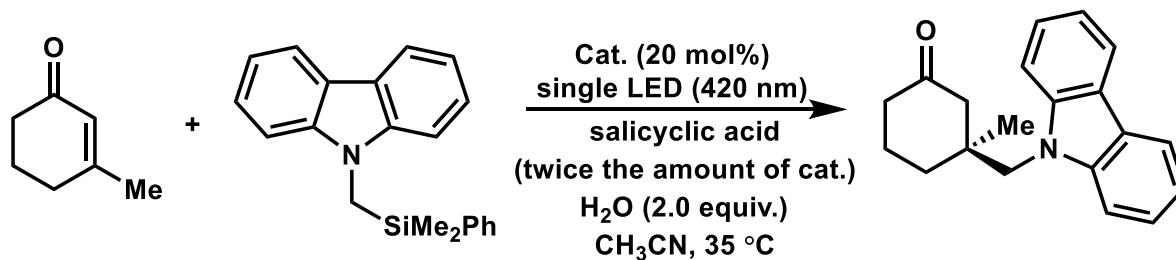
手性胺催化的共轭加成反应

反应机理



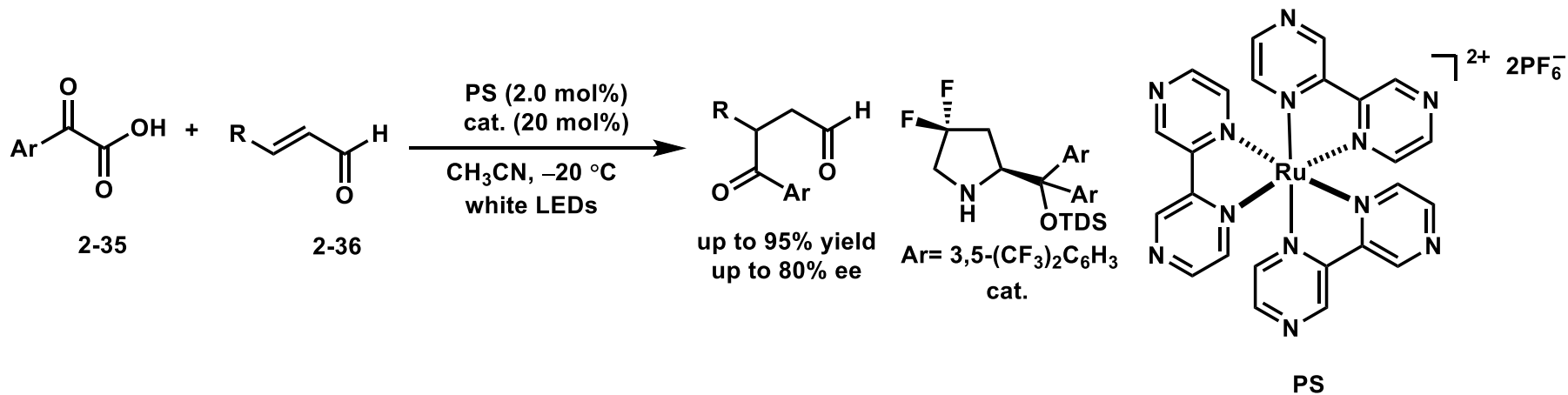
手性胺催化的共轭加成反应

对照试验

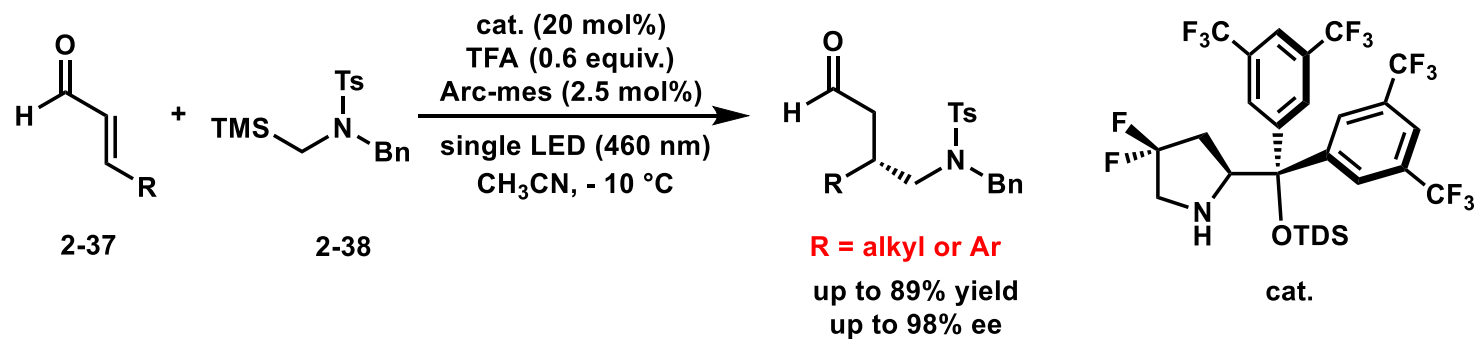


Cat.	Yield(%)	ee(%)
A (R, R) (15 mol%)	44	80
B (S, S) (5 mol%)	-	-
A (R, R) (15 mol%)+ B (S, S) (5 mol%)	40	42

手性胺催化的共轭加成反应

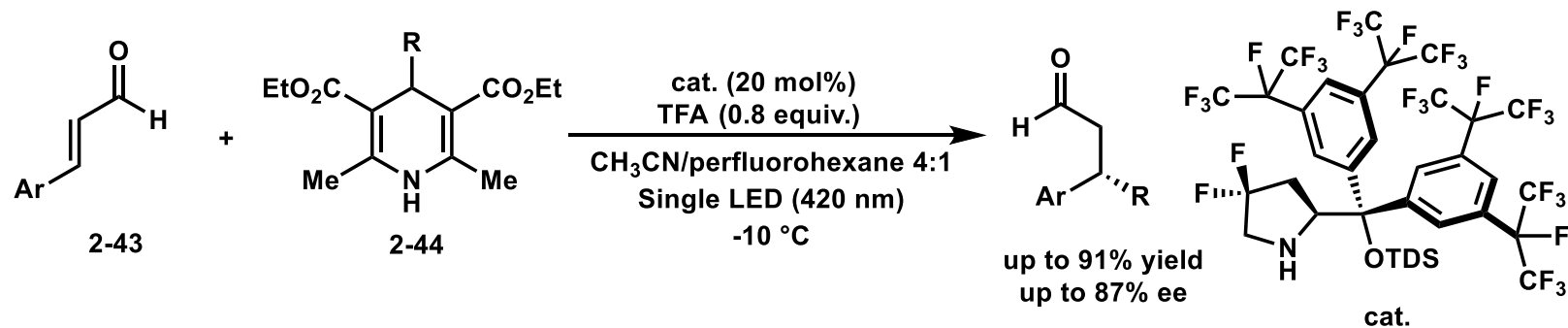


Zhao, J.; Zhang, H.; Shen, X.; Yu, S. *Org. Lett.* **2019**, *21*, 913.

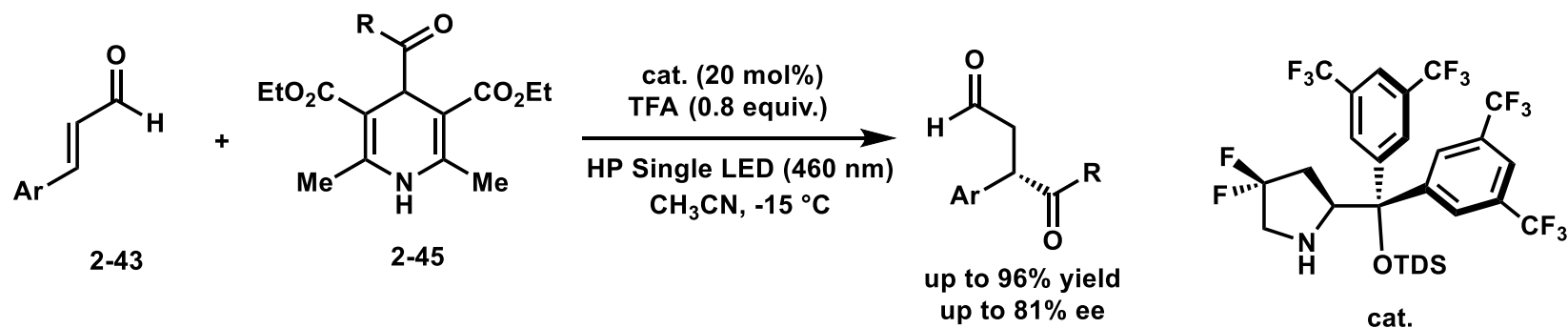


Saux, E. L.; Ma, D.; Bonilla, P.; Holden, C. M.; Lustosa, D.; Melchiorre, P. *Angew. Chem., Int. Ed.* DOI:10.1002/anie.202014876.

手性胺催化的共轭加成反应



Melchiorre, P et al. *ACS Catal.* **2018**, 8, 1062.



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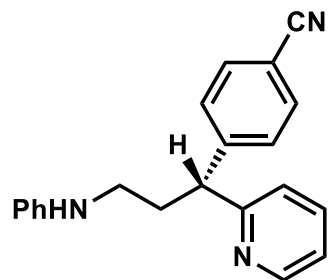
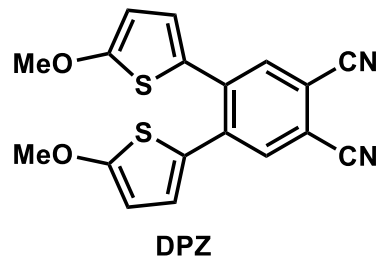
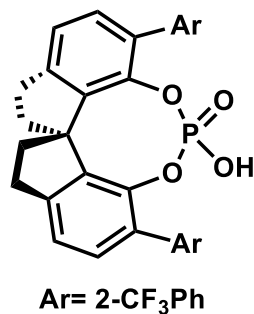
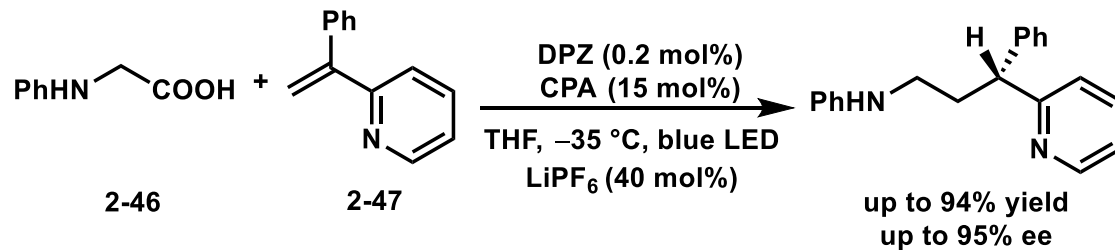
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2.2 手性胺催化的共轭加成反应

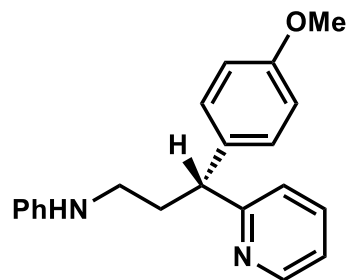
2.3 手性磷酸催化的共轭加成反应

3. 总结

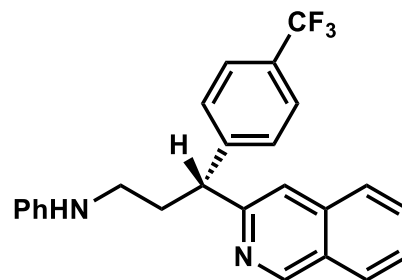
手性磷酸催化的共轭加成反应



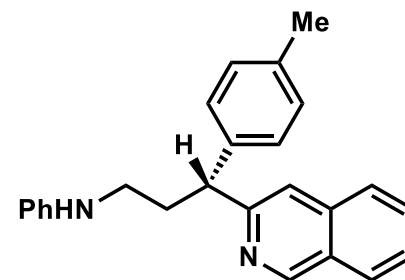
86% yield, 88% ee



82% yield, 90% ee



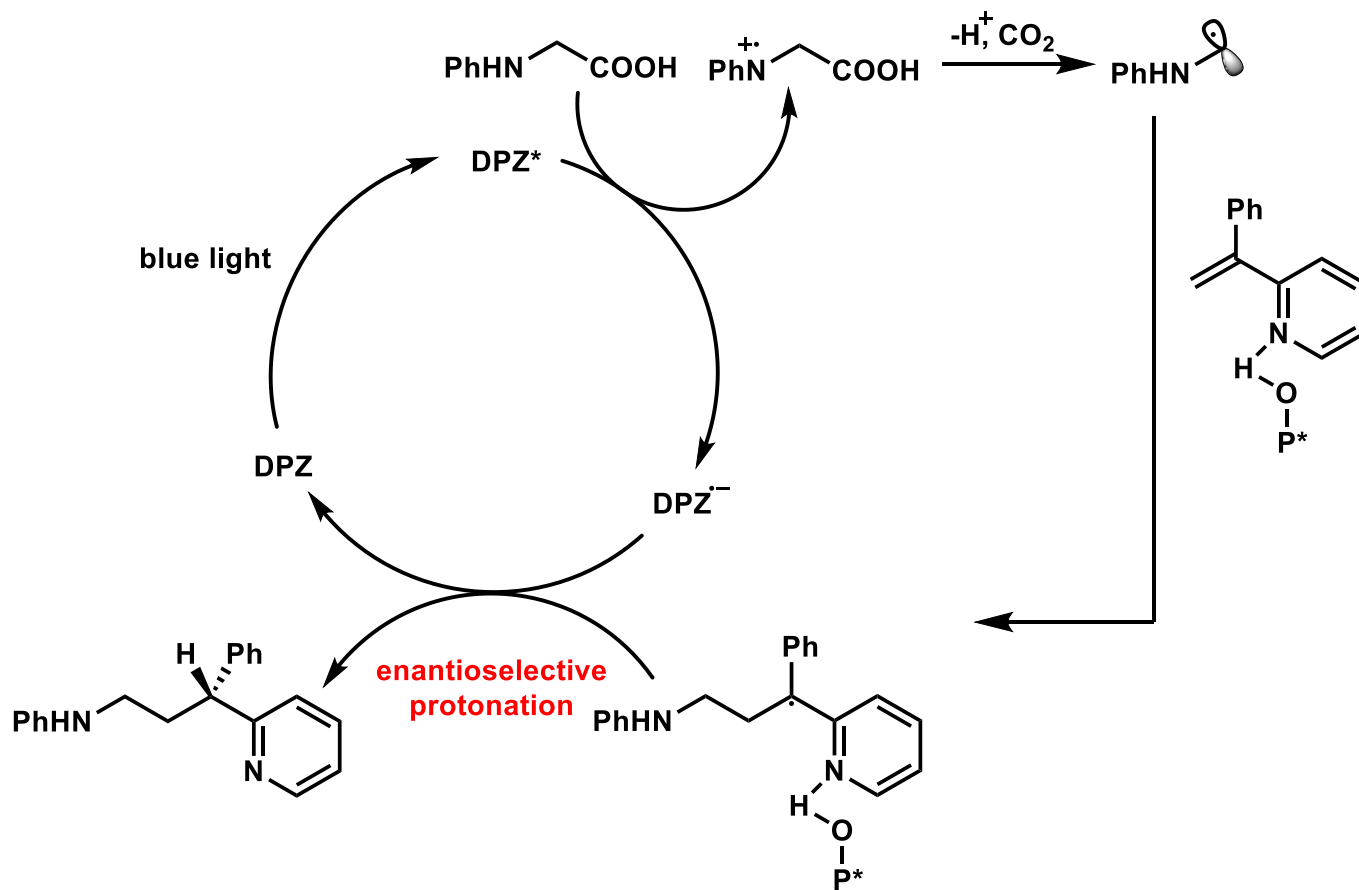
72% yield, 97% ee



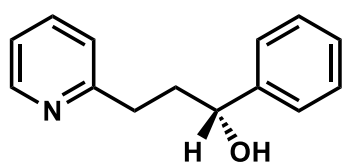
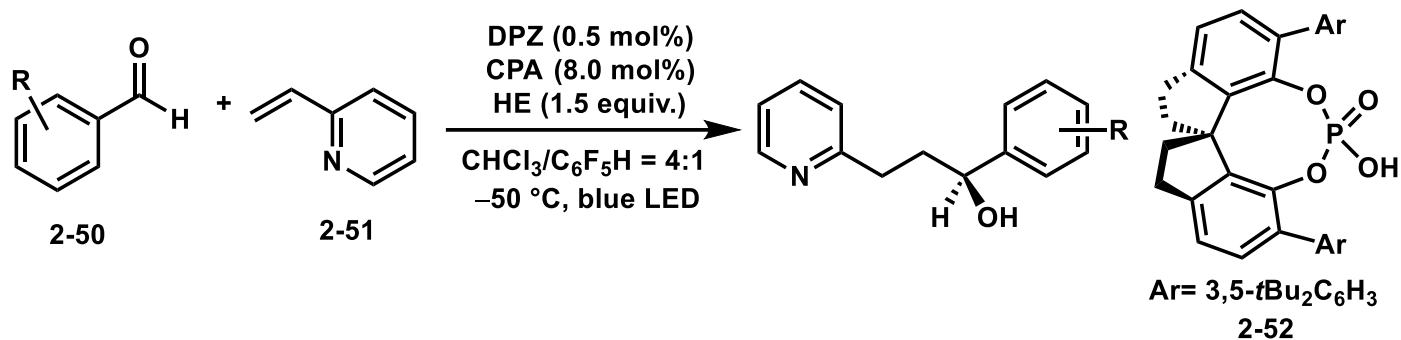
72% yield, 97% ee

手性磷酸催化的共轭加成反应

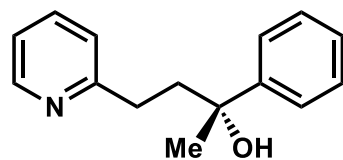
反应机理



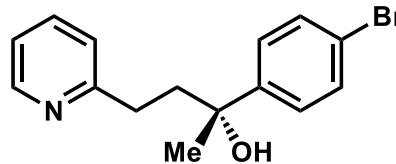
手性磷酸催化的共轭加成反应



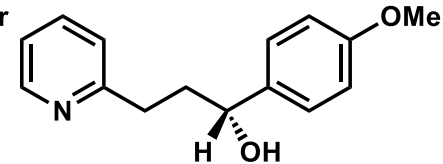
90% yield, 90% ee



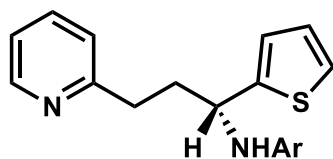
89% yield, 94% ee



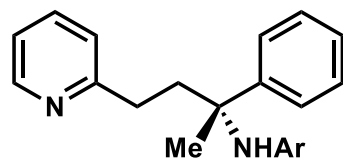
63% yield, 93% ee



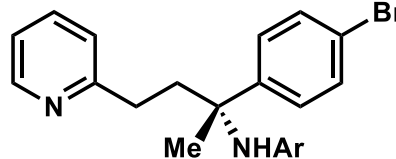
79% yield, 95% ee



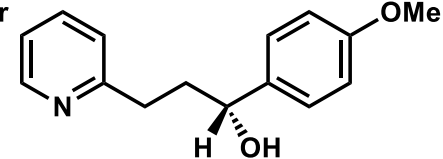
50% yield, 91% ee



65% yield, 92% ee



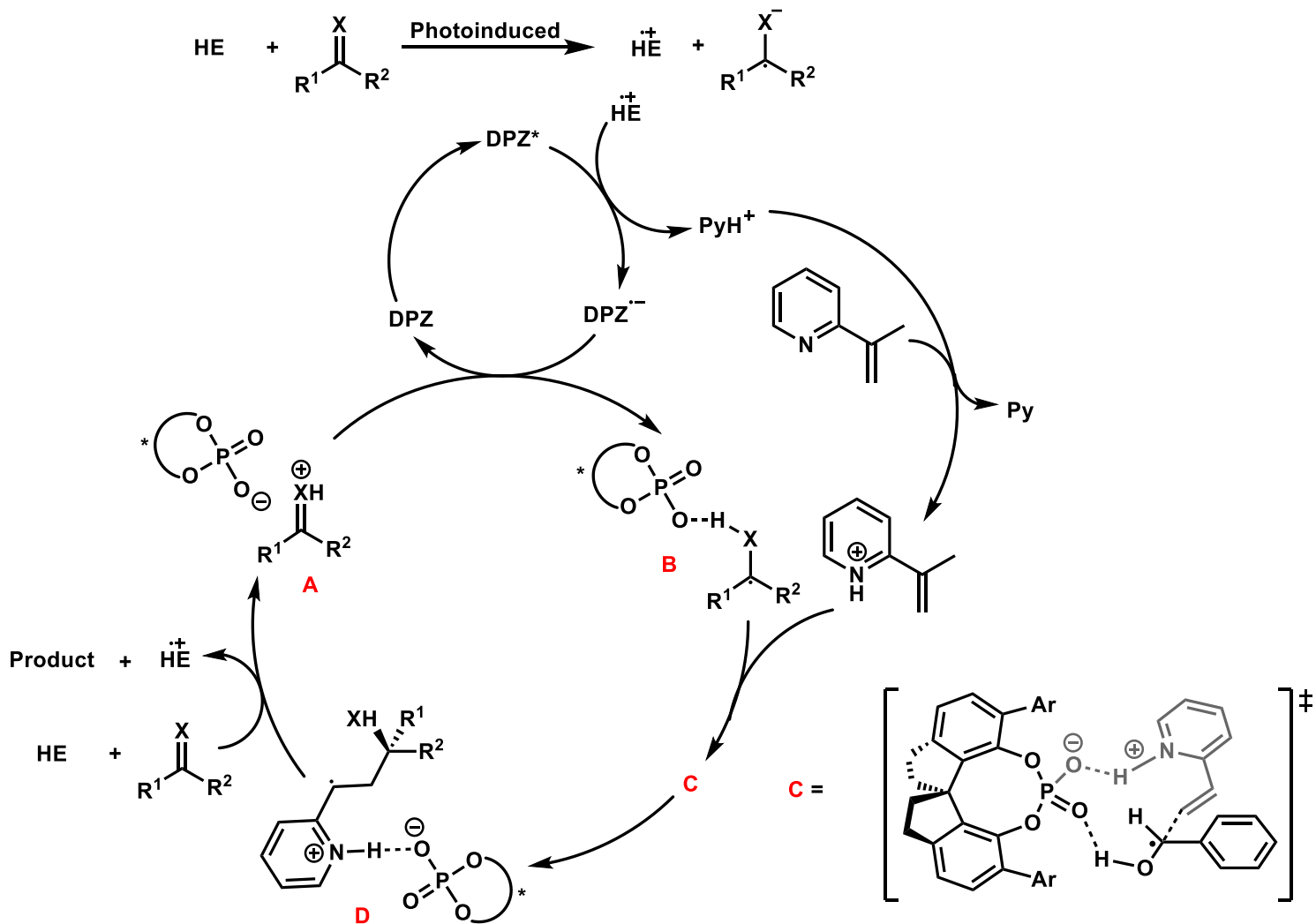
62% yield, 92% ee



55% yield, 96% ee

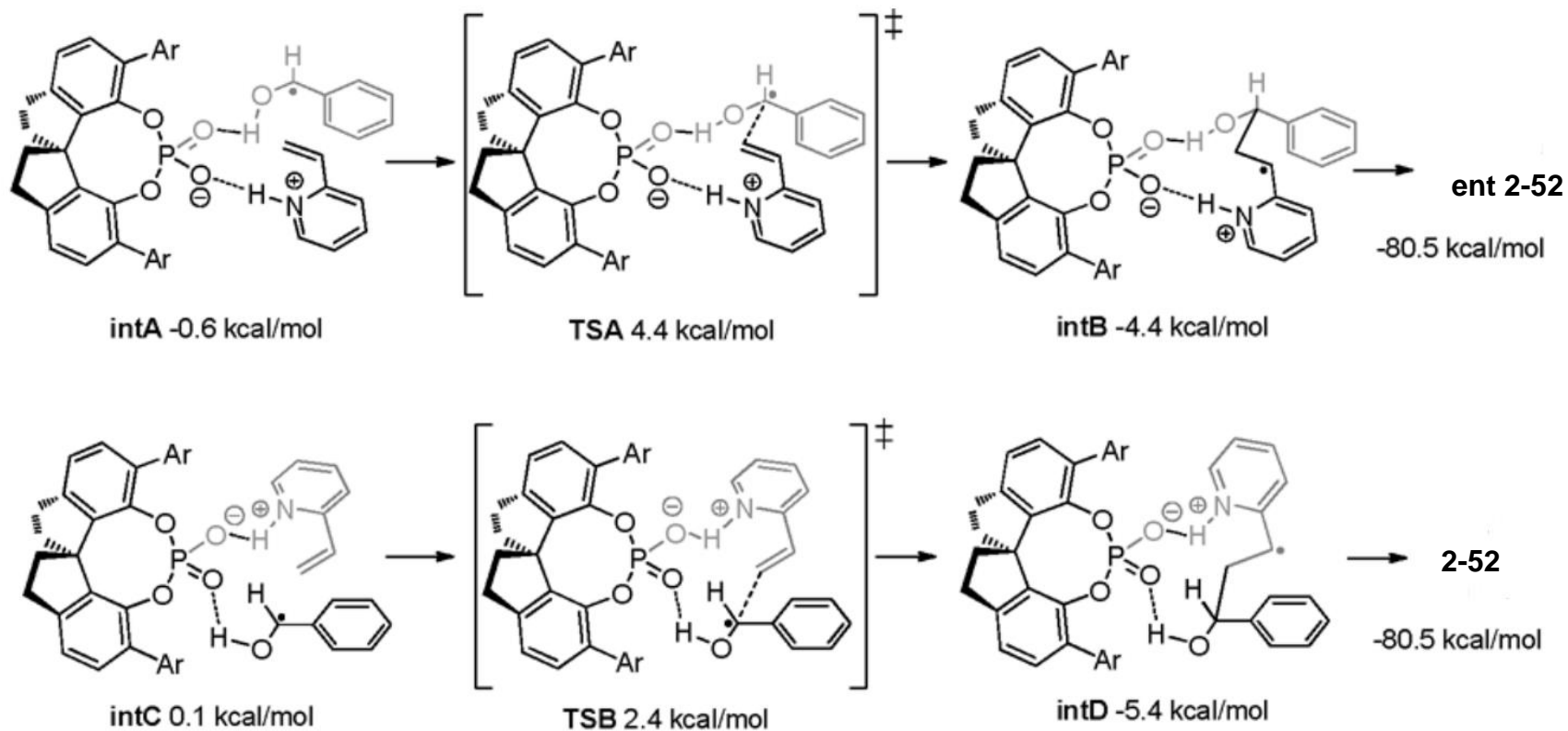
手性磷酸催化的共轭加成反应

反应机理



Zhiyong Jiang et al. *J. Am. Chem. Soc.* **2019**, *141*, 5437–5443.

手性磷酸催化的共轭加成反应



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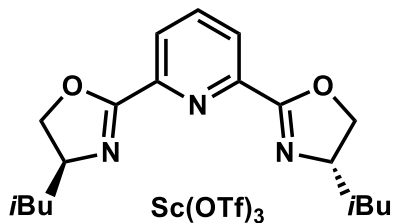
2.2 手性胺催化的共轭加成反应

2.3 手性磷酸催化的共轭加成反应

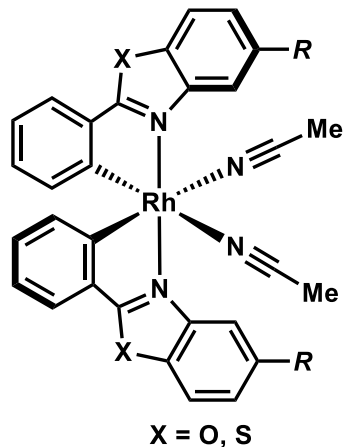
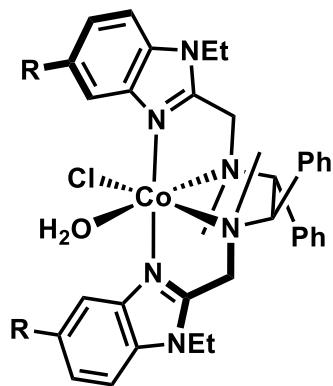
3. 总结

总结

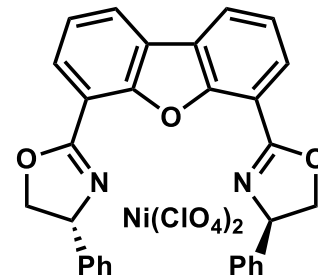
手性路易斯酸



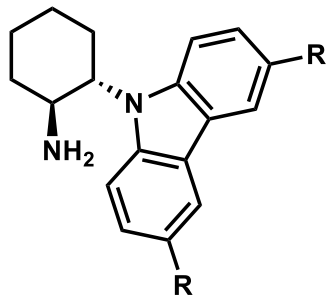
只控制手性



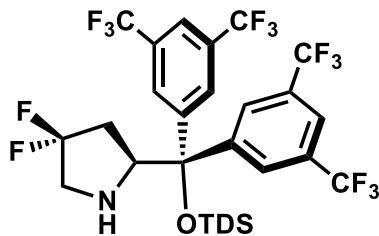
既控制手性，又有光化学性



手性胺

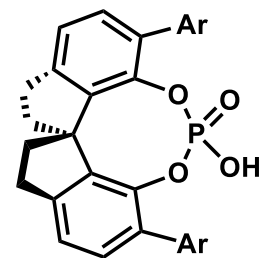


只控制手性，
EDA复合物



只控制手性，
亚胺中间体强氧化性

手性磷酸



对映选择性质子化，
氢键整合底物控制手性

**谢谢大家！
希望大家批评指正！**