



含氟C-1烷烃在氟烷基化 反应中的应用

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导师：施章杰 教授

2021-11-05



1. 背景介绍

1.1 氟化学背景

1.2 氟烷基化反应简介

1.3 四类含氟C-1烷烃分子简介

2. 含氟C-1分子在氟烷基化反应中的应用

2.1 三氟甲烷 (CF_3H)

2.2 三氟碘甲烷 (CF_3I)

2.3 一氯二氟甲烷 (ClCF_2H)

2.4 一溴二氟甲烷 (BrCF_2H)

3. 总结与展望



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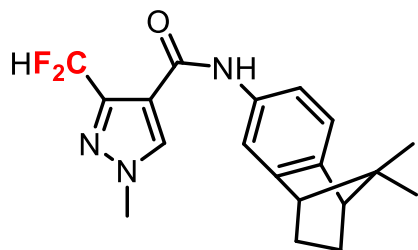
2.3 一氯二氟甲烷 (ClCF_2H)

2.4 一溴二氟甲烷 (BrCF_2H)

3. 总结与展望

1.1 氟化学背景

农药

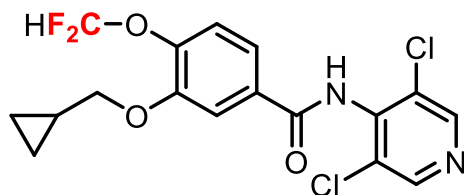


吡唑萘菌胺
(杀菌剂)



- 高选择性
- 化学稳定性

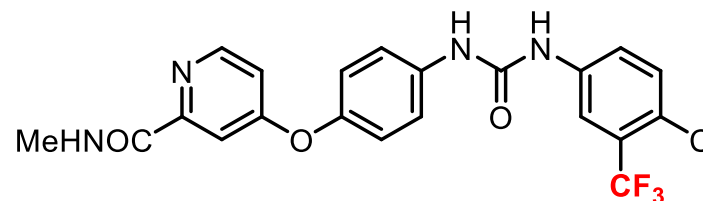
医药



罗氟司特
(治疗慢性阻塞性肺疾病)



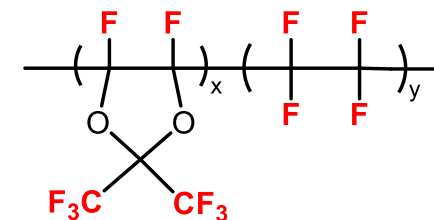
- 代谢稳定性
- 亲脂性
- 生物利用度



索拉非尼
(第一个治疗肿瘤口服药物)



材料

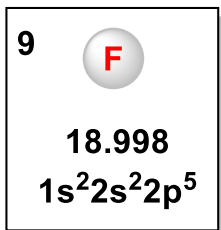


聚四氟乙烯



- 稳定
- 耐腐蚀

1.1 氟化学背景



电负性: 3.98

范德华半径: F 135 pm

H 120 pm

C-F 键 490 kJ/mol

C-CF₃ 键 431 kJ/mol



热稳定性
化学稳定性
抗氧化性



冰晶石 (Na₃AlF₆)



萤石 (CaF₂)

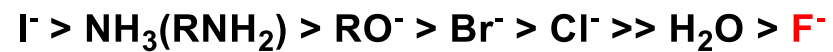
萤石是唯一一种可以
提炼大量氟元素矿物



氟磷灰石 (Ca₁₀(PO₄)₆F₂)

水温为18 °C时, CaF₂ 100 g水中只溶解0.0016 g

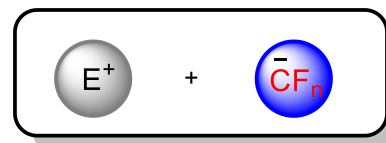
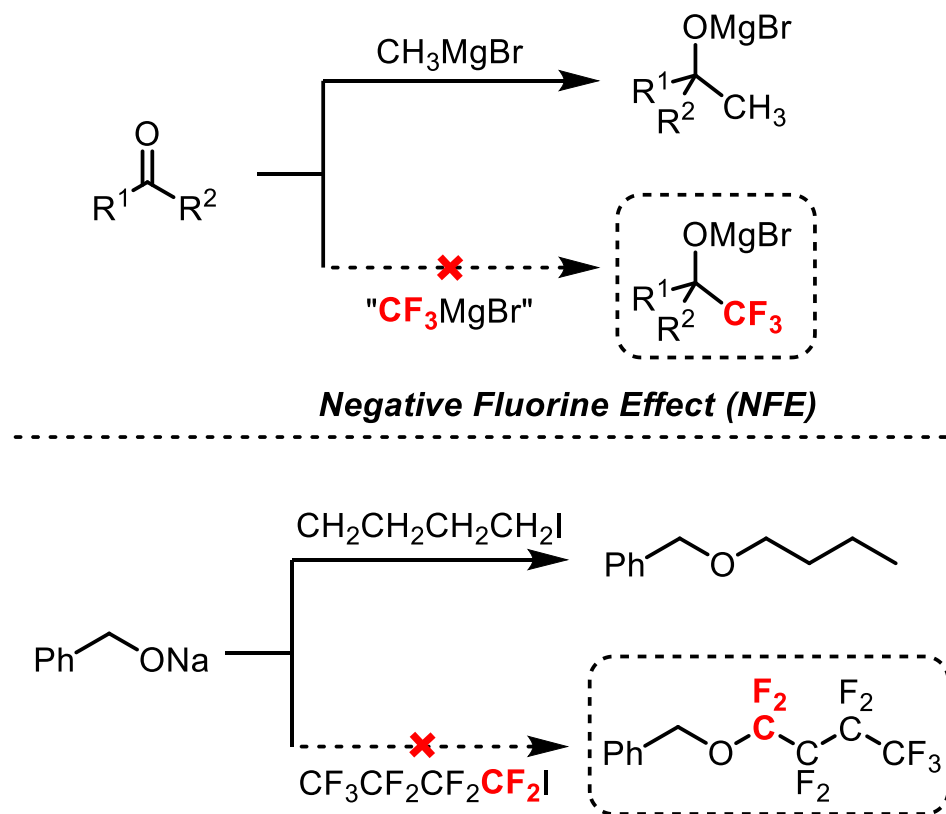
质子性溶剂亲核性:



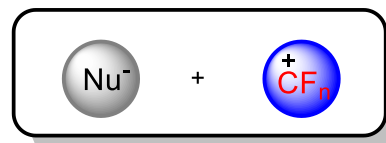
天然含氟有机化合物稀少

1.1 氟烷基化反应简介

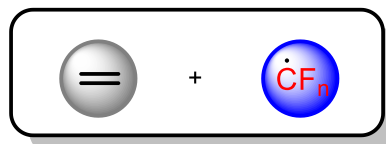
氟烷基化反应的特殊性



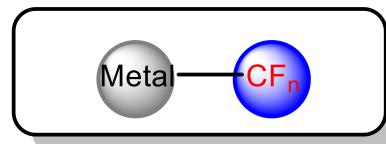
- 亲核氟烷基化反应



- 亲电氟烷基化反应



- 自由基氟烷基化反应

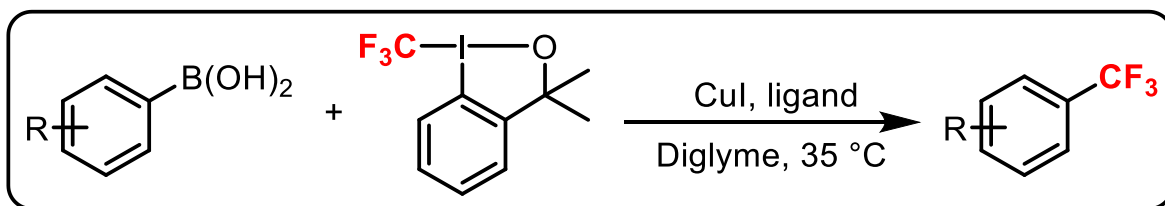


- 过渡金属参与的氟烷基化反应

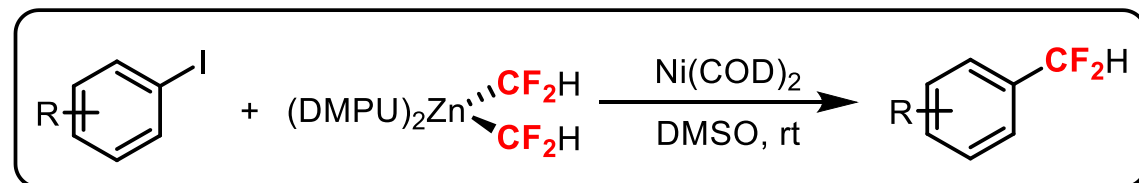
1.2 氟烷基化反应简介



Vicic, D. A. *J. Am. Chem. Soc.* **2008**, *130*, 8600.



Shen, Q. et al. *Org. Lett.* **2011**, *13*, 2342.



Vicic, D. A. et al. *J. Am. Chem. Soc.* **2016**, *138*, 2536.

Postigo, A. et al. *Chem. Eur. J.* **2017**, *23*, 14676. Li, C. et al. *Chem. Soc. Rev.* **2021**, *50*, 6308.

三氟甲基化试剂



(X = Na, K, Me, etc)

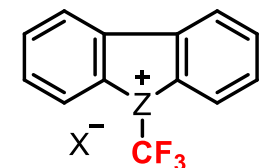
1981, Kondo



1984, Ruppert-Prakash

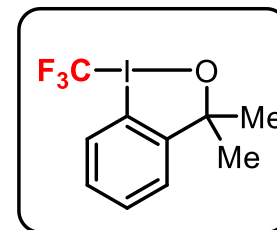


1989, Chen

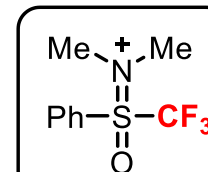


Z = S, Se, Te;
X = BF₄, OTf, PF₆

1990, Umemoto

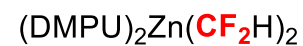


2006, Togni

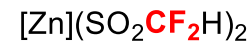


2008, Shibata

二氟甲基化试剂



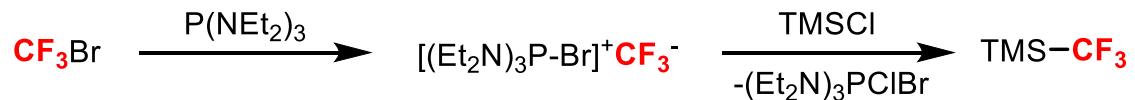
2016, Vicic



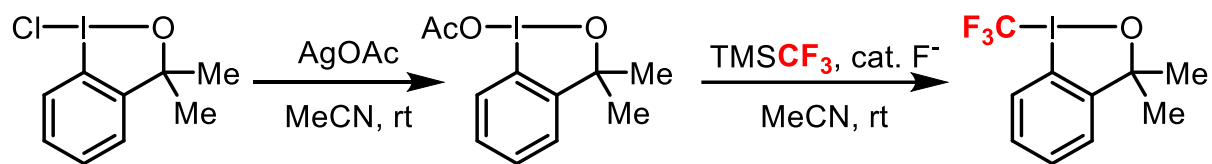
2012, Baran



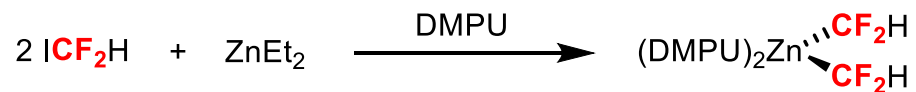
1.2 氟烷基化反应简介



Ruppert, I. et al. *Tetrahedron* **1984**, 25, 2159.



Togni, A. et al. *Chem. Eur. J.* **2006**, 12, 2579.



Vicic, D. A. et al. *J. Am. Chem. Soc.* **2016**, 138, 2536.

CF_3I

300 RMB/kg

CF_3H

90 RMB/kg

ClCF_2H

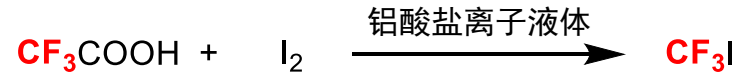
293 RMB/kg

BrCF_2H

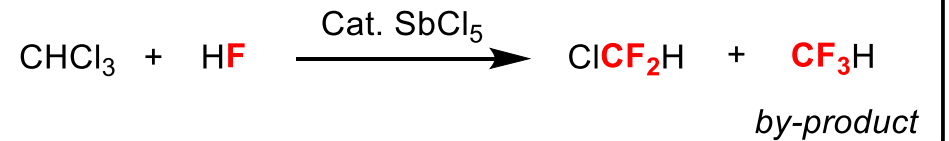
1.2 四类含氟C-1烷烃分子简介



- 沸点: -22.5 °C; 主要用于灭火剂和制冷剂
- 低毒, 无色, 无味



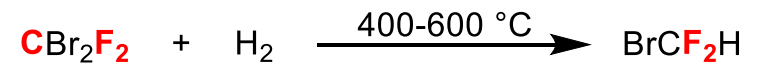
- 沸点: -82 °C; 溶于大多有机溶剂
- 无毒, 无色, 无味



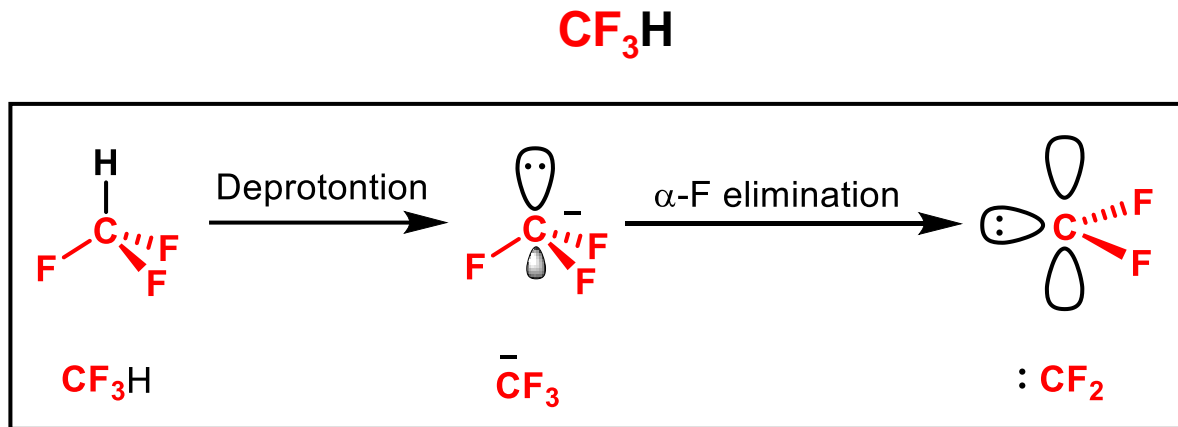
- 沸点: -40.8 °C; 溶于大多数有机溶剂
- 低毒, 无色, 轻微的甜气味
- 氟利昂-22; 是合成聚四氟乙烯的原料



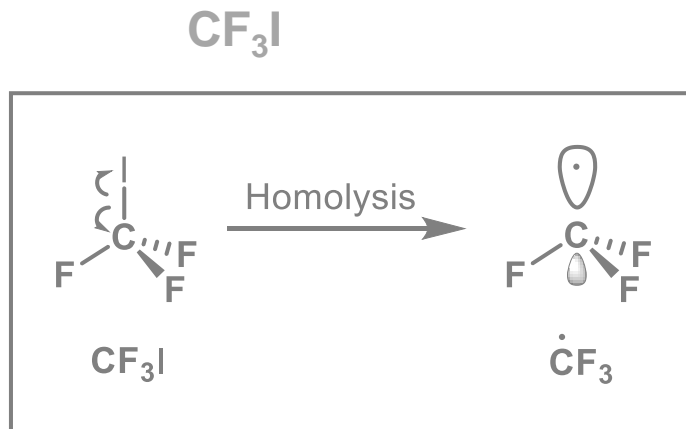
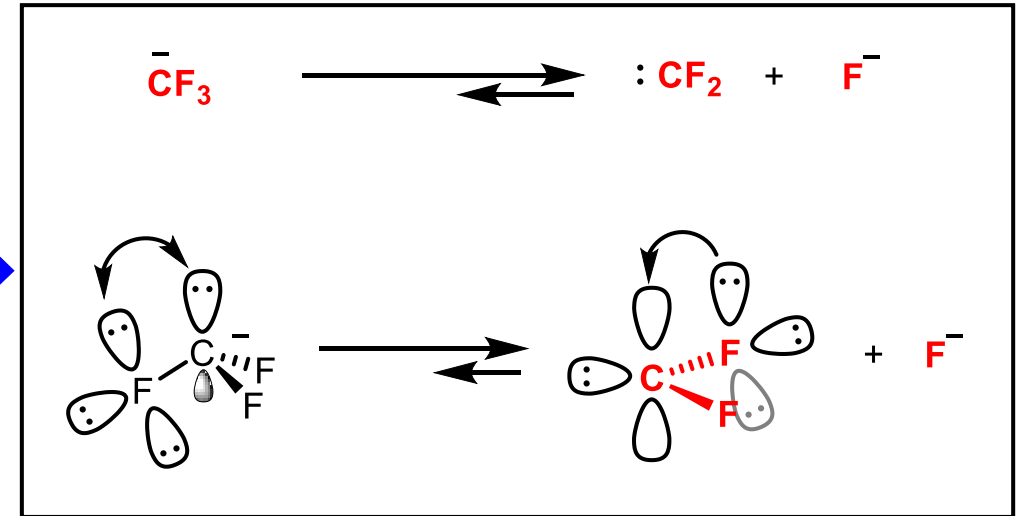
- 沸点: -14.5 °C
- 有毒, 无色, 无味



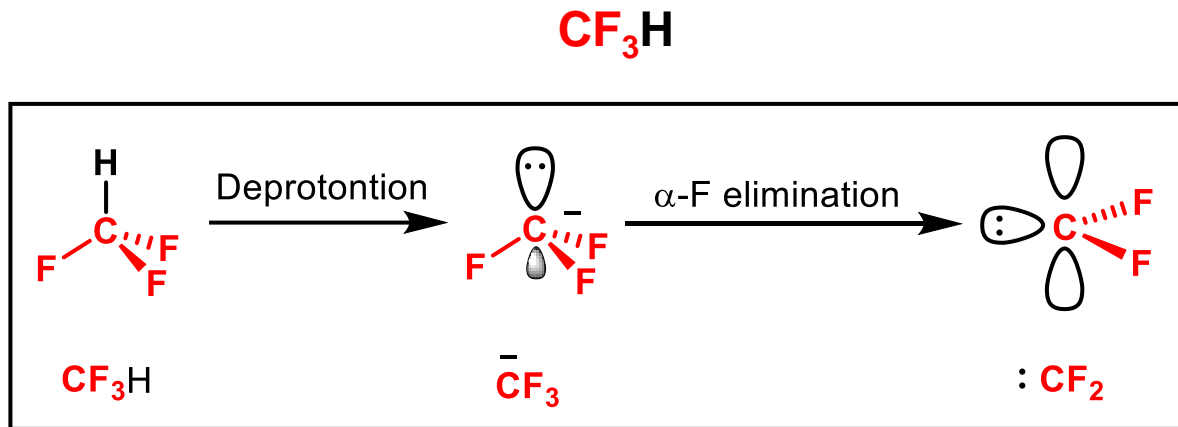
1.2 四类含氟C-1烷烃分子简介



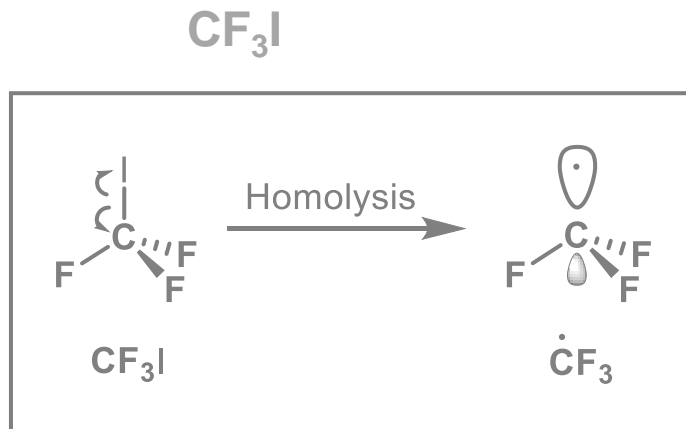
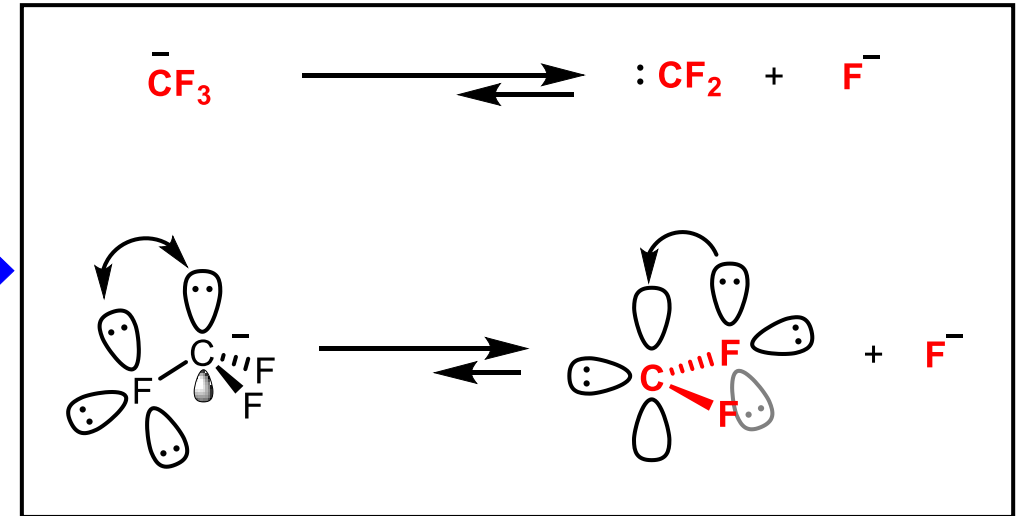
- C-H键酸性较弱 ($\text{pK}_a = 27$ in water)



1.2 四类含氟C-1烷烃分子简介



● C-H键酸性较弱 ($\text{pK}_a = 27$ in water)



1. DMF

用强的亲电试剂捕获

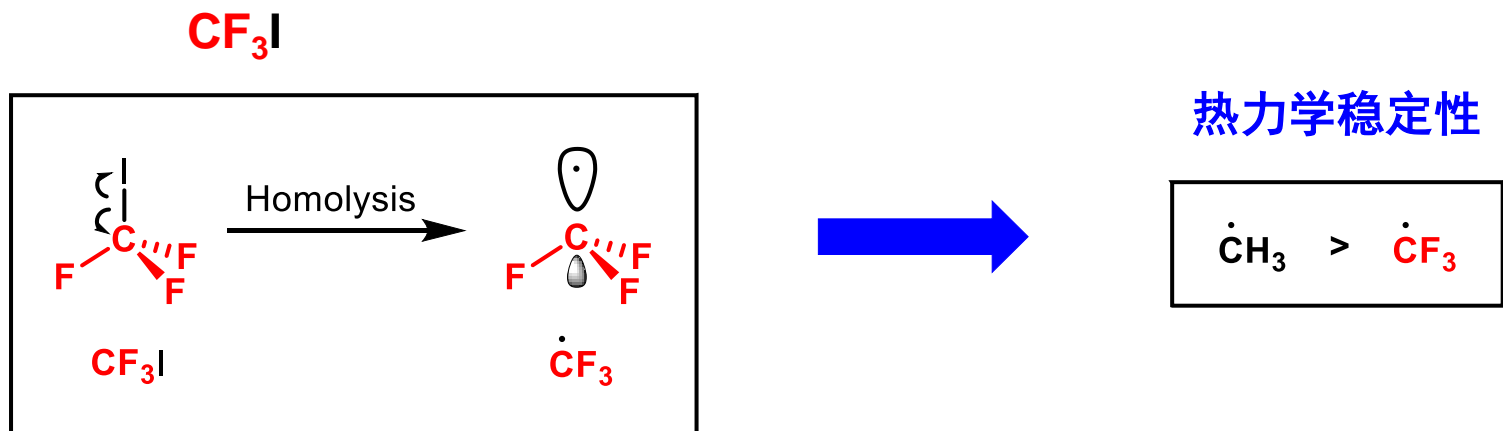
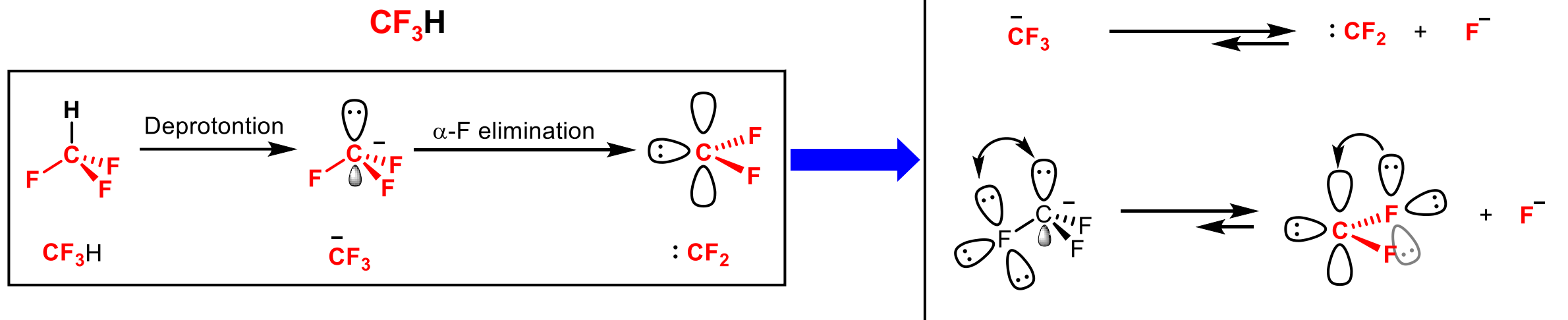
2. CuCF₃

将电荷分散到金属的空轨道

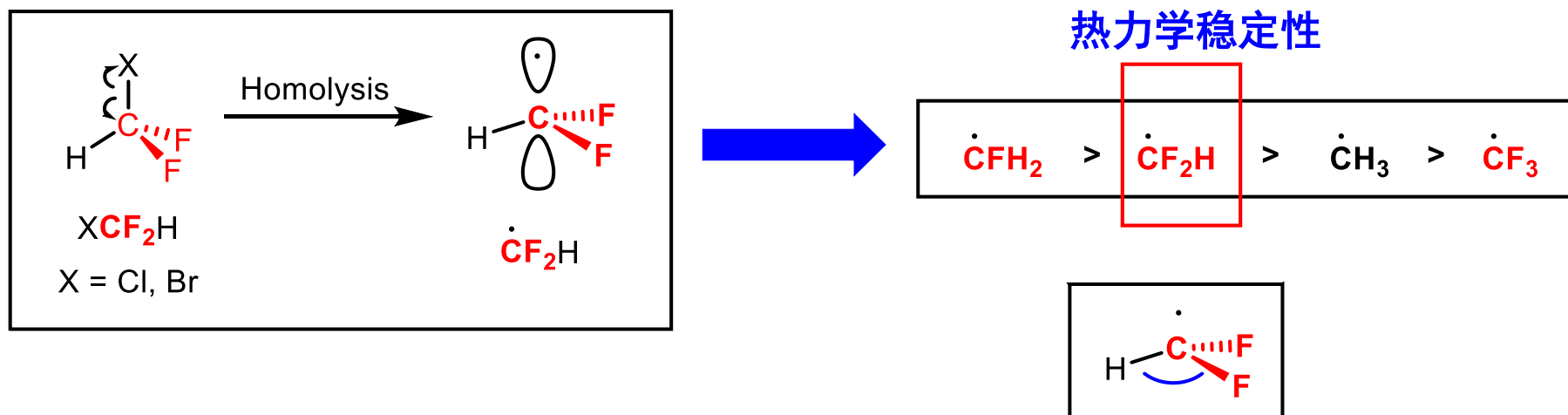
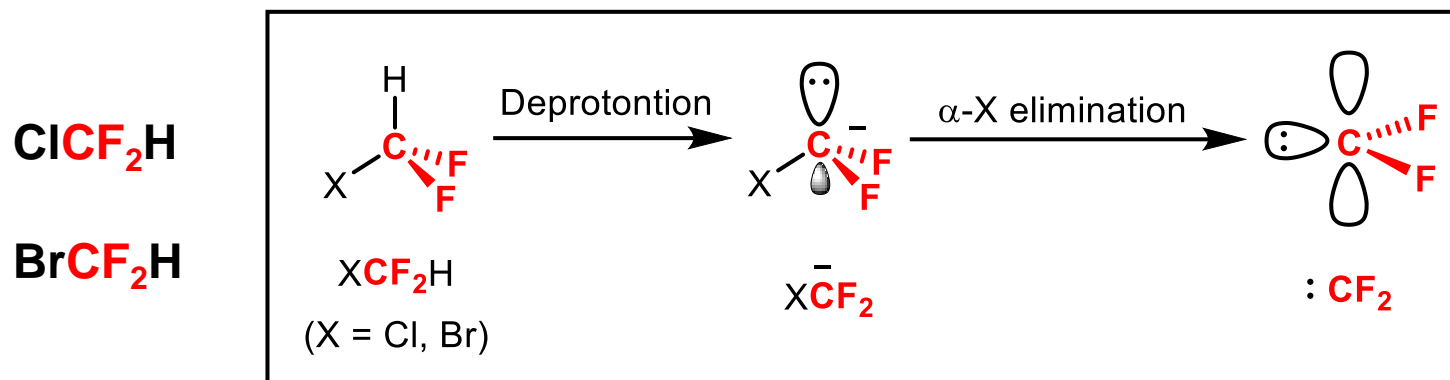
3. TMSCF₃

形成容易断裂的 σ 键

1.2 四类含氟C-1烷烃分子简介



1.2 四类含氟C-1烷烃分子简介





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2. 含氟C-1分子在氟烷基化反应中的应用

2.1 三氟甲烷 (CF_3H)

2.2 三氟碘甲烷 (CF_3I)

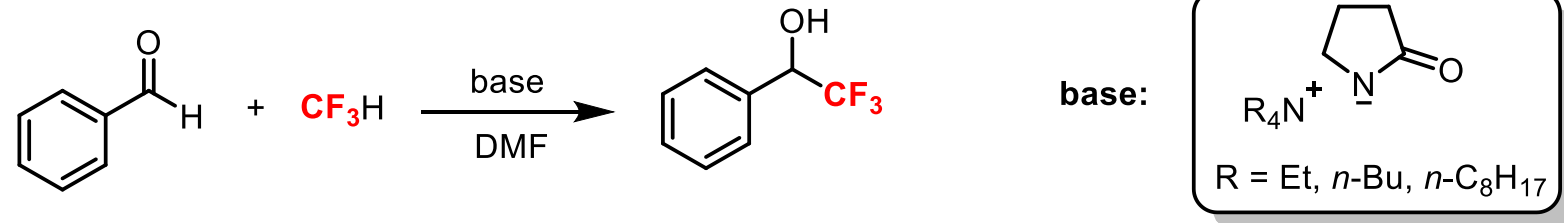
2.3 一氯二氟甲烷 (ClCF_2H)

2.4 一溴二氟甲烷 (BrCF_2H)

3. 总结与展望

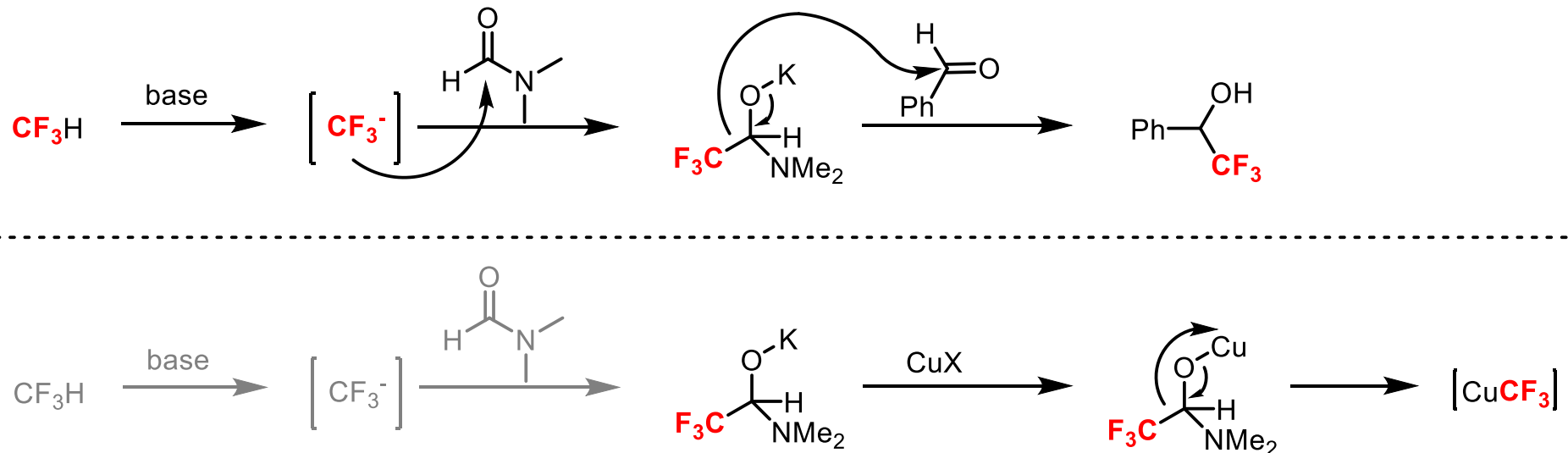
2.1 三氟甲烷 (CF₃H)

2.1.1 首例使用CF₃H进行的亲核三氟甲基反应



Shono, T. et al. *J. Org. Chem.* **1991**, 56, 2.

2.1.2 溶剂DMF作为亲电试剂的关键作用

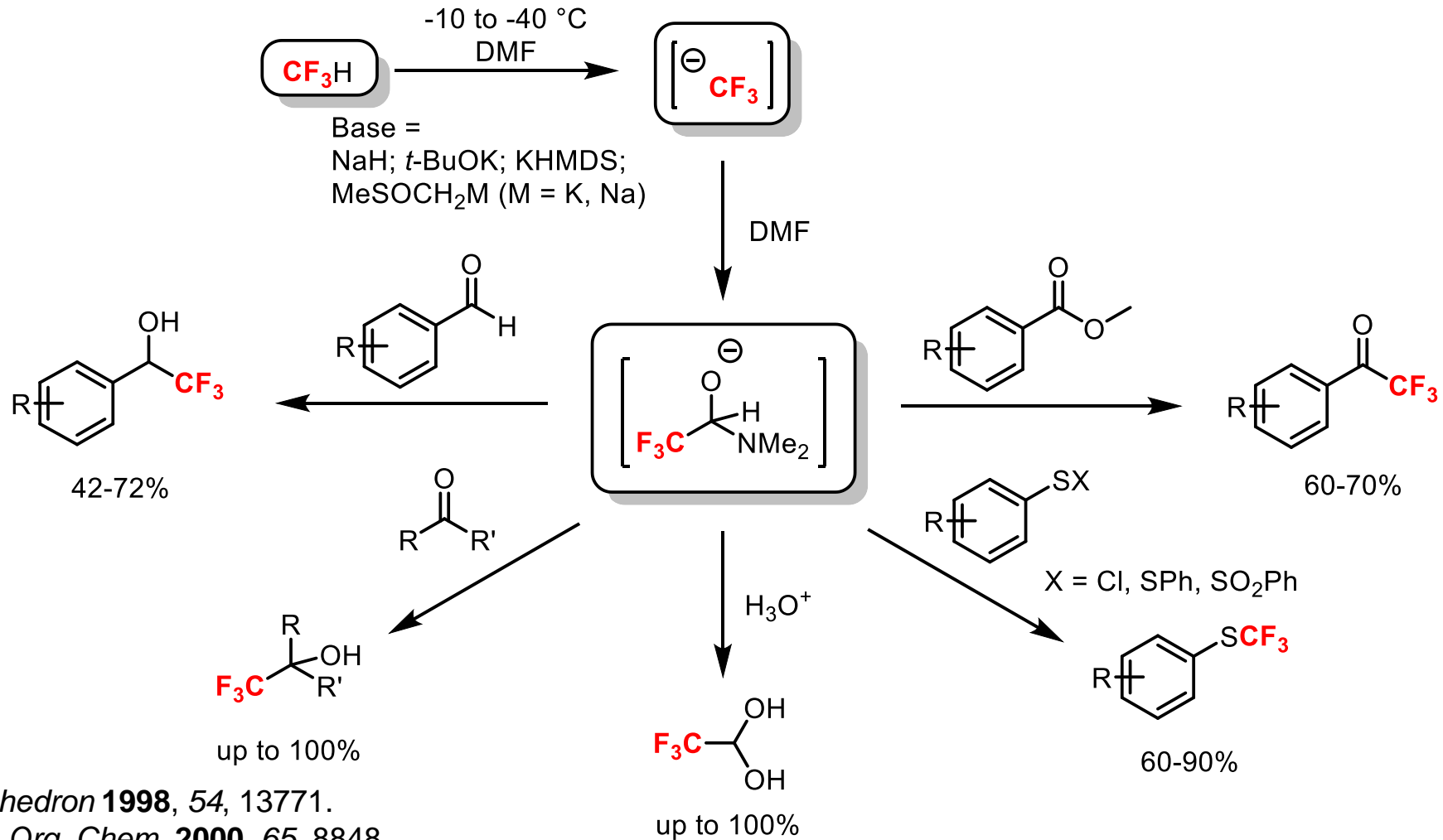


Roques, N. et al. *Tetrahedron* **1998**, 54, 13771.

Normant, J. F. et al. *Tetrahedron* **2000**, 56, 275.

2.1 三氟甲烷 (CF₃H)

2.1.3 用CF₃H对不同底物的三氟甲基化反应



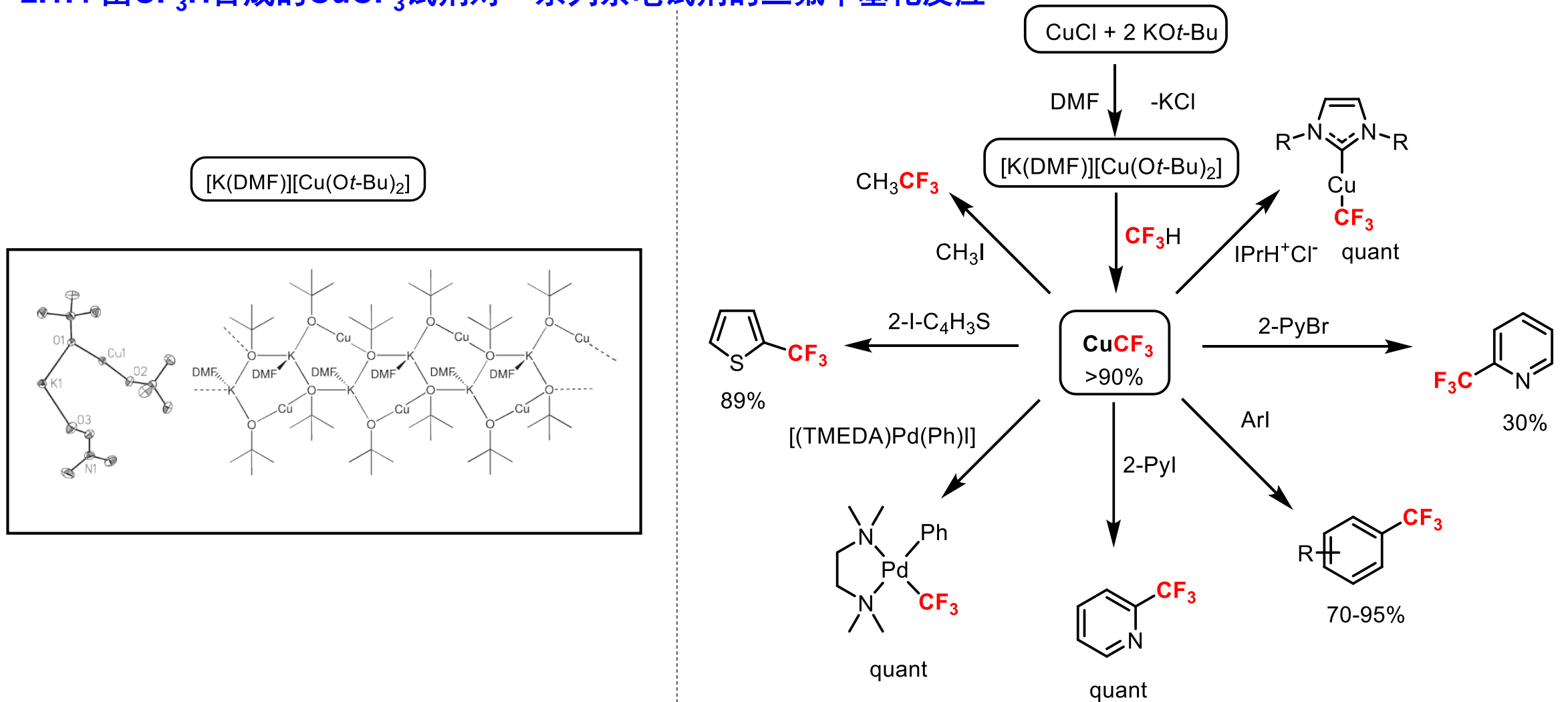
Roques, N. et al. *Tetrahedron* **1998**, *54*, 13771.

Langlois, B. R. et al. *J. Org. Chem.* **2000**, *65*, 8848.

Normant, J. F. et al. *Tetrahedron* **2000**, *56*, 275.

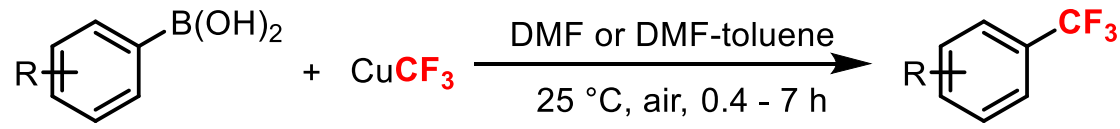
2.1 三氟甲烷 (CF₃H)

2.1.4 由CF₃H合成的CuCF₃试剂对一系列亲电试剂的三氟甲基化反应

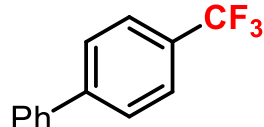


2.1 三氟甲烷 (CF₃H)

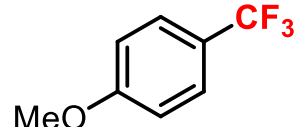
2.1.5 CuCF₃对芳基硼酸的三氟甲基化反应



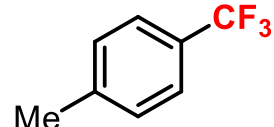
99% yield



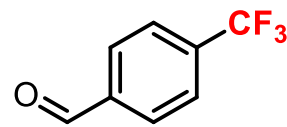
92 (92)% yield



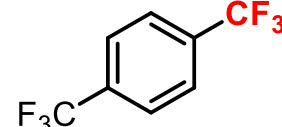
91% yield



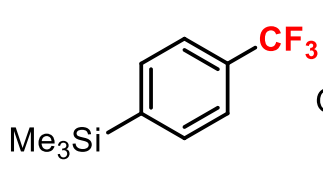
97% yield



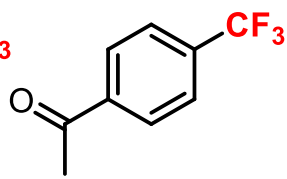
82% yield



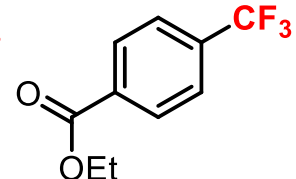
73% yield



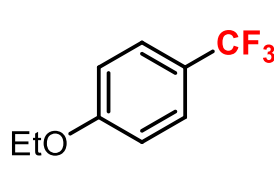
91 (81)% yield



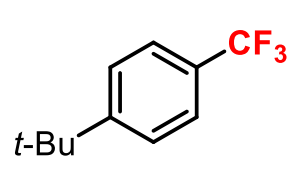
94 (59)% yield



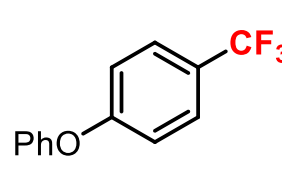
91 (61)% yield



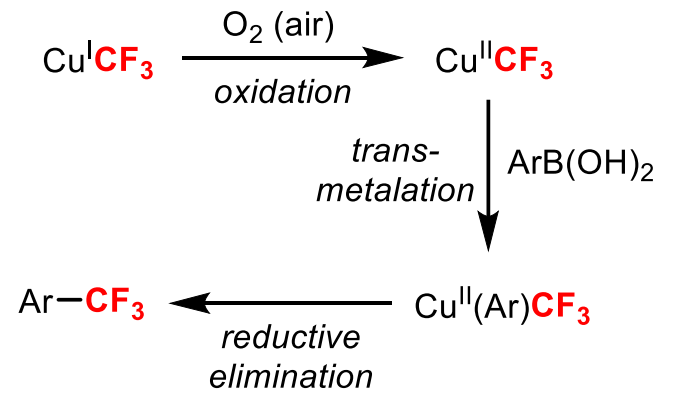
91% yield



96 (81)% yield

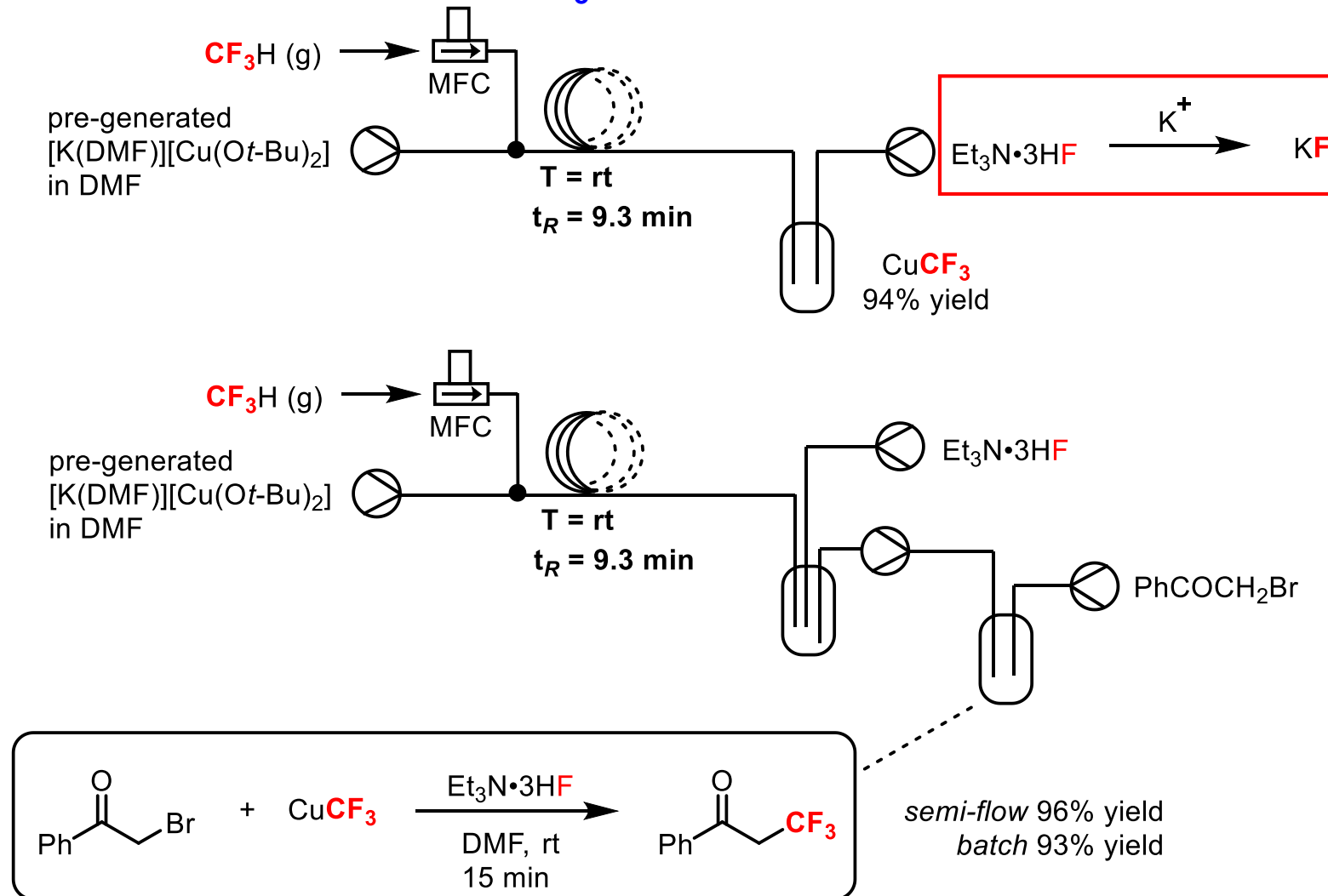


95 (94)% yield



2.1 三氟甲烷 (CF₃H)

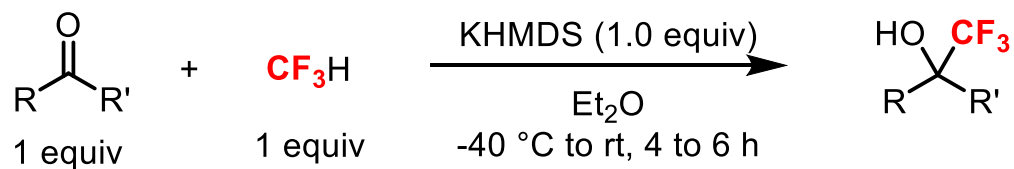
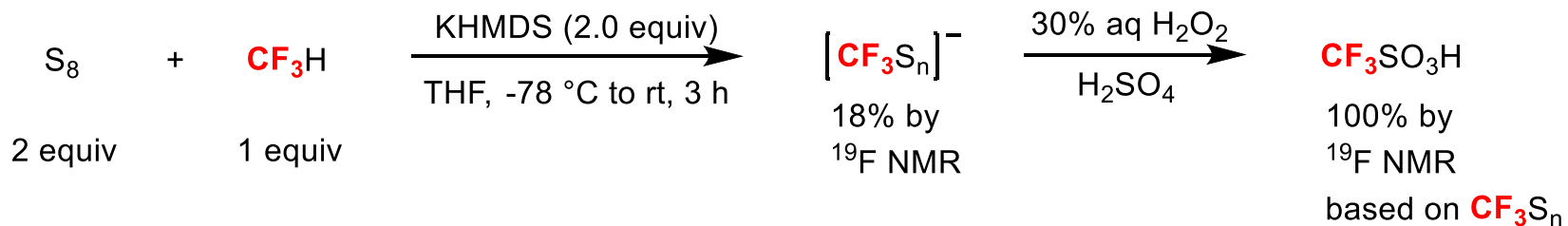
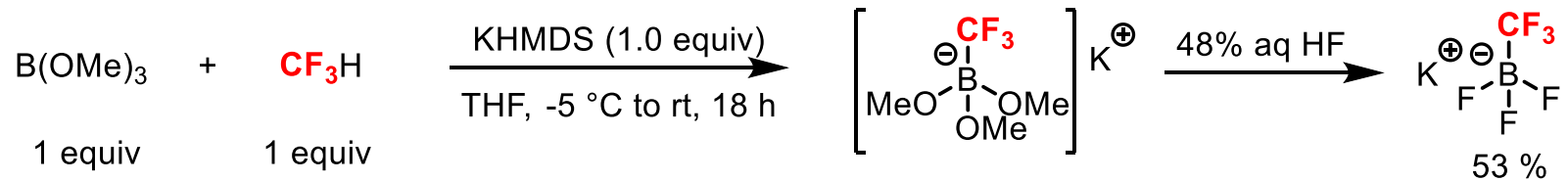
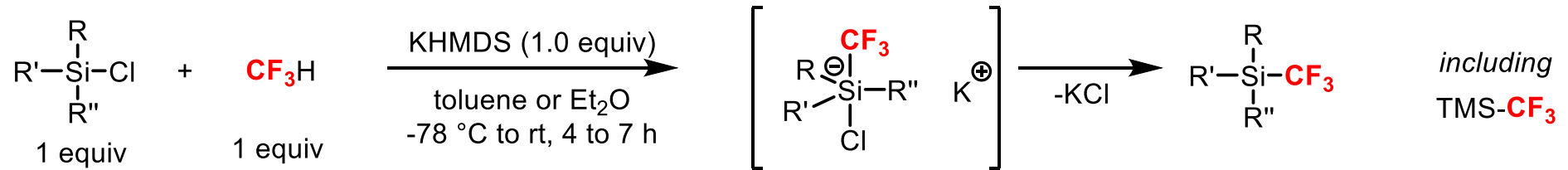
2.1.6 首次使用流动化学技术制备CuCF₃



- 精确的温度控制
- 产物完全与反应物分离
- 高效、快速
- 安全使用活性或有毒试剂
- 易于进行多相反应
- 易于自动化、模块化

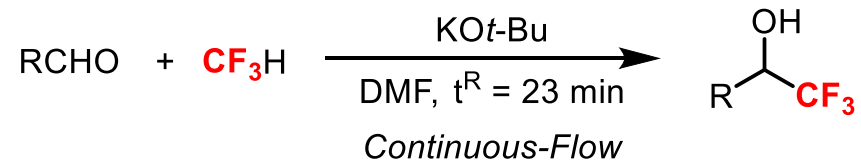
2.1 三氟甲烷 (CF₃H)

2.1.7 使用CF₃H直接进行亲核三氟甲基化反应

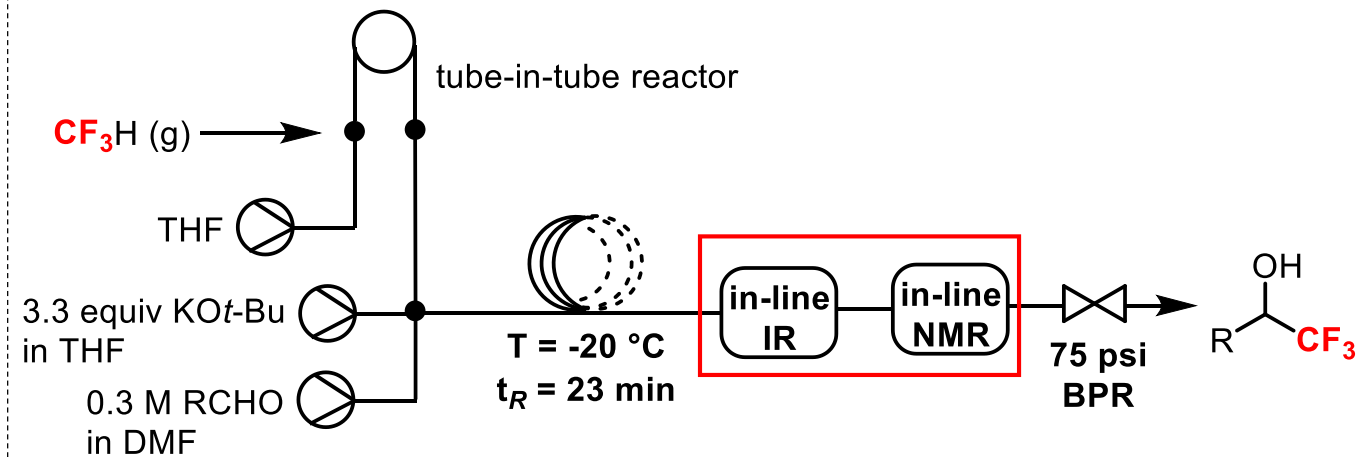
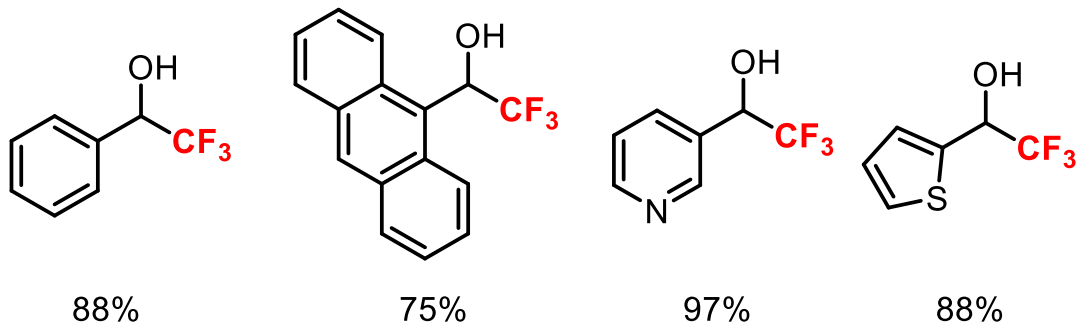


2.1 三氟甲烷 (CF₃H)

2.1.8 使用流动化学技术合成三氟烷基化合物

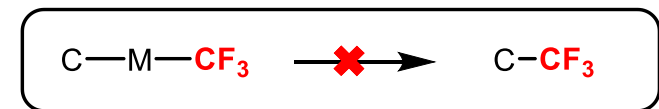
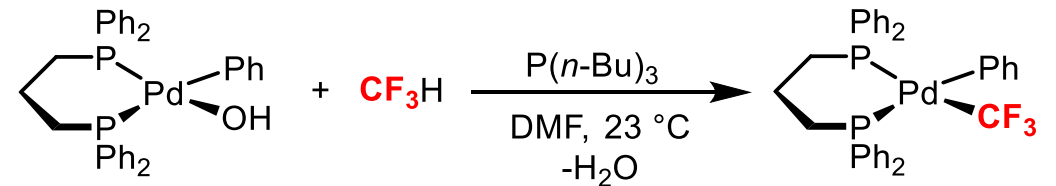
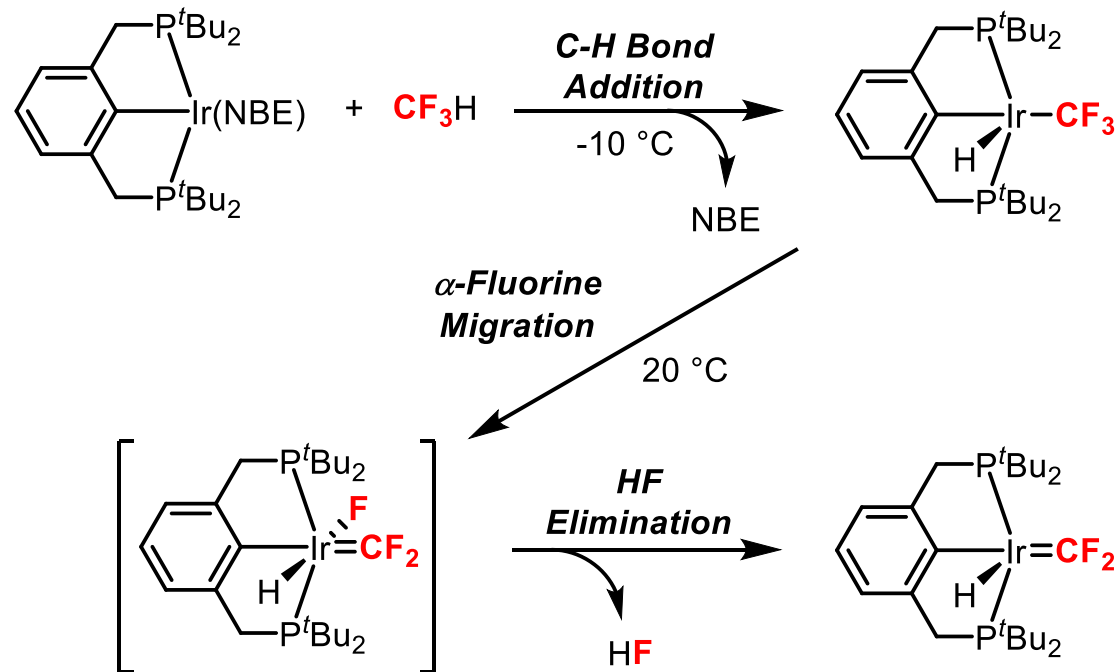


Selected examples



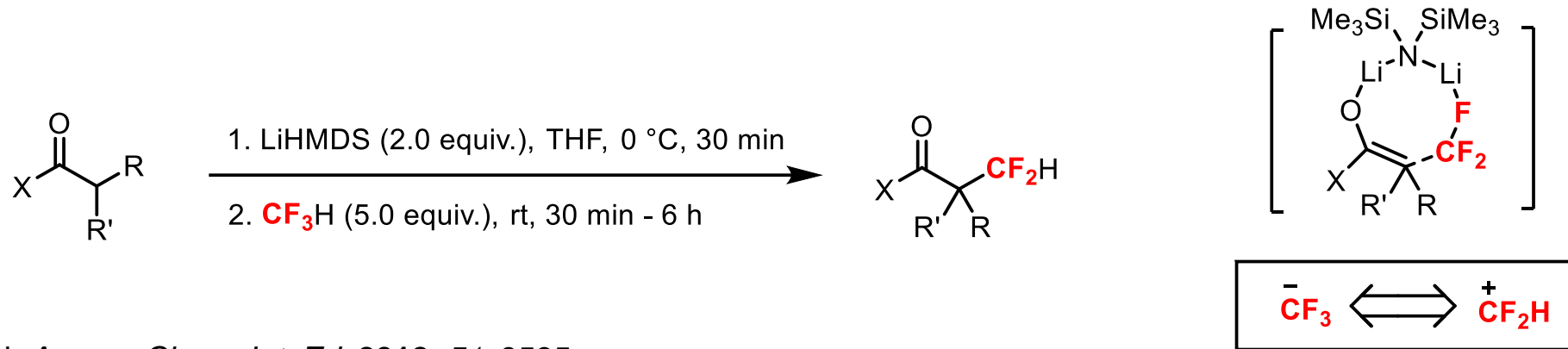
2.1 三氟甲烷 (CF₃H)

2.1.9 过渡金属对三氟甲烷的活化

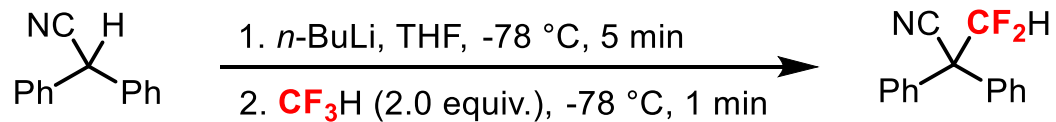


2.1 三氟甲烷 (CF₃H)

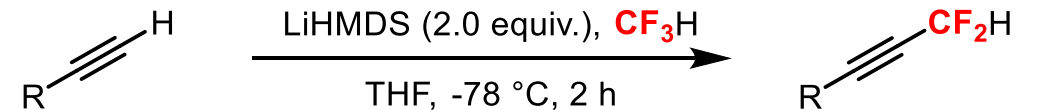
2.1.10 利用C-F键活化策略对羰基α位进行二氟甲基化



Mikami, K. et al. *Angew. Chem. Int. Ed.* **2012**, *51*, 9535.



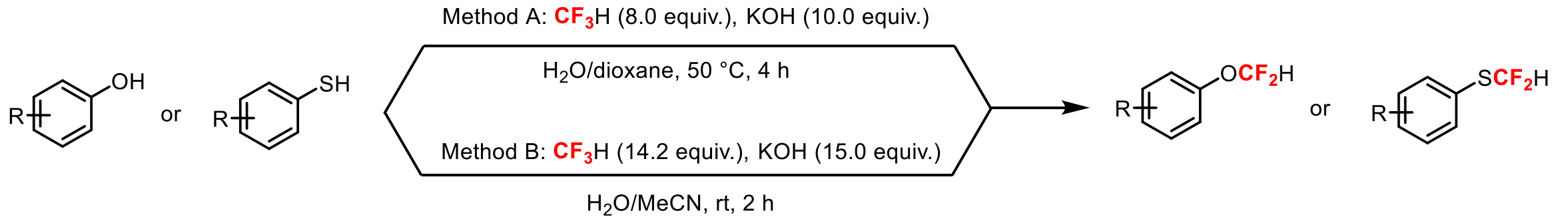
Mikami, K. et al. *Org. Lett.* **2015**, *17*, 4882.



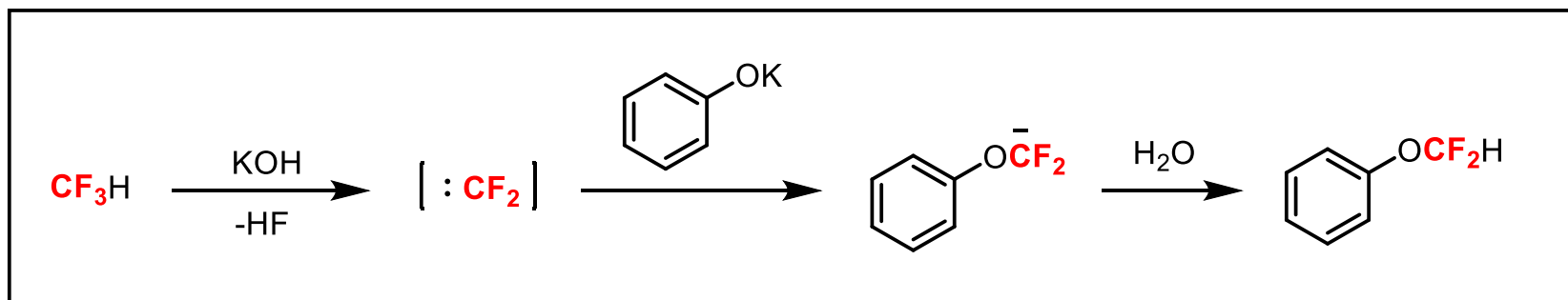
Mikami, K. et al. *Org. Lett.* **2016**, *18*, 3354.

2.1 三氟甲烷 (CF₃H)

2.1.11 CF₃H作为二氟卡宾源合成芳基二氟甲基醚/硫醚



机理





1. 背景介绍

1.1 氟化学背景

1.2 氟烷基化反应简介

1.3 四类含氟C-1烷烃分子简介

2. 含氟C-1分子在氟烷基化反应中的应用

2.1 三氟甲烷 (CF_3H)

2.2 三氟碘甲烷 (CF_3I)

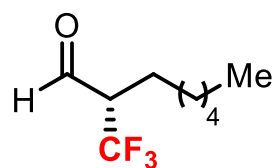
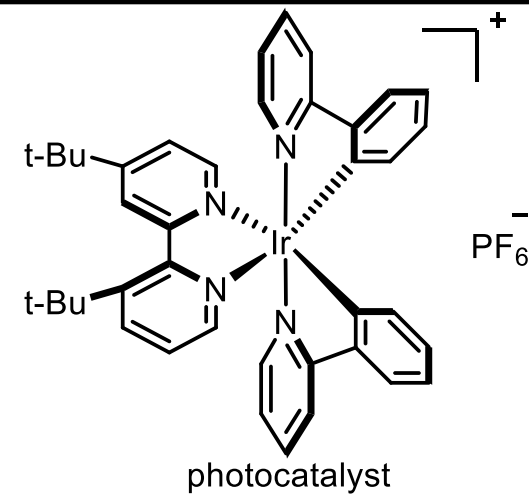
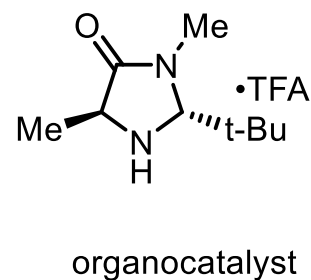
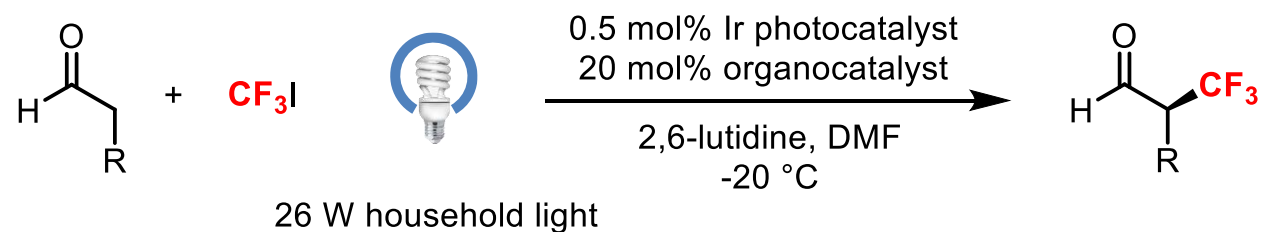
2.3 一氯二氟甲烷 (ClCF_2H)

2.4 一溴二氟甲烷 (BrCF_2H)

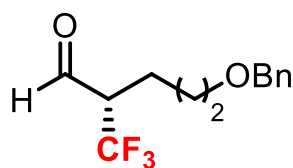
3. 总结与展望

2.2 三氟碘甲烷 (CF₃I)

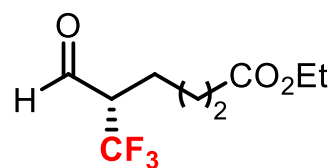
2.2.1 对映选择性的醛羰基α位三氟甲基化反应



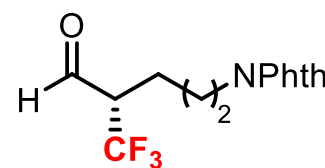
79% yield, 99% ee



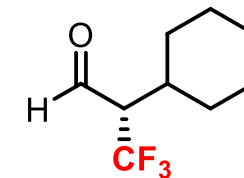
72% yield, 95% ee



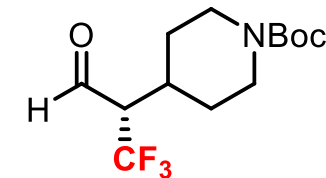
86% yield, 97% ee



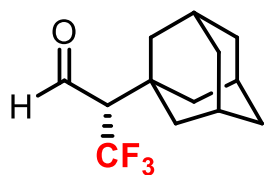
78% yield, 98% ee



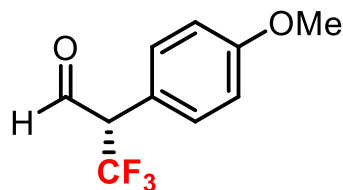
70% yield, 99% ee



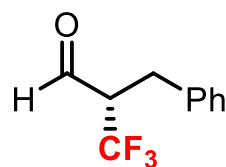
70% yield, 98% ee



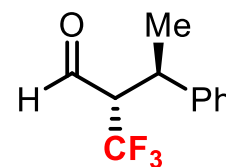
73% yield, 90% ee



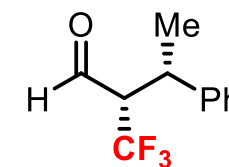
61% yield, 93% ee



75% yield, 97% ee



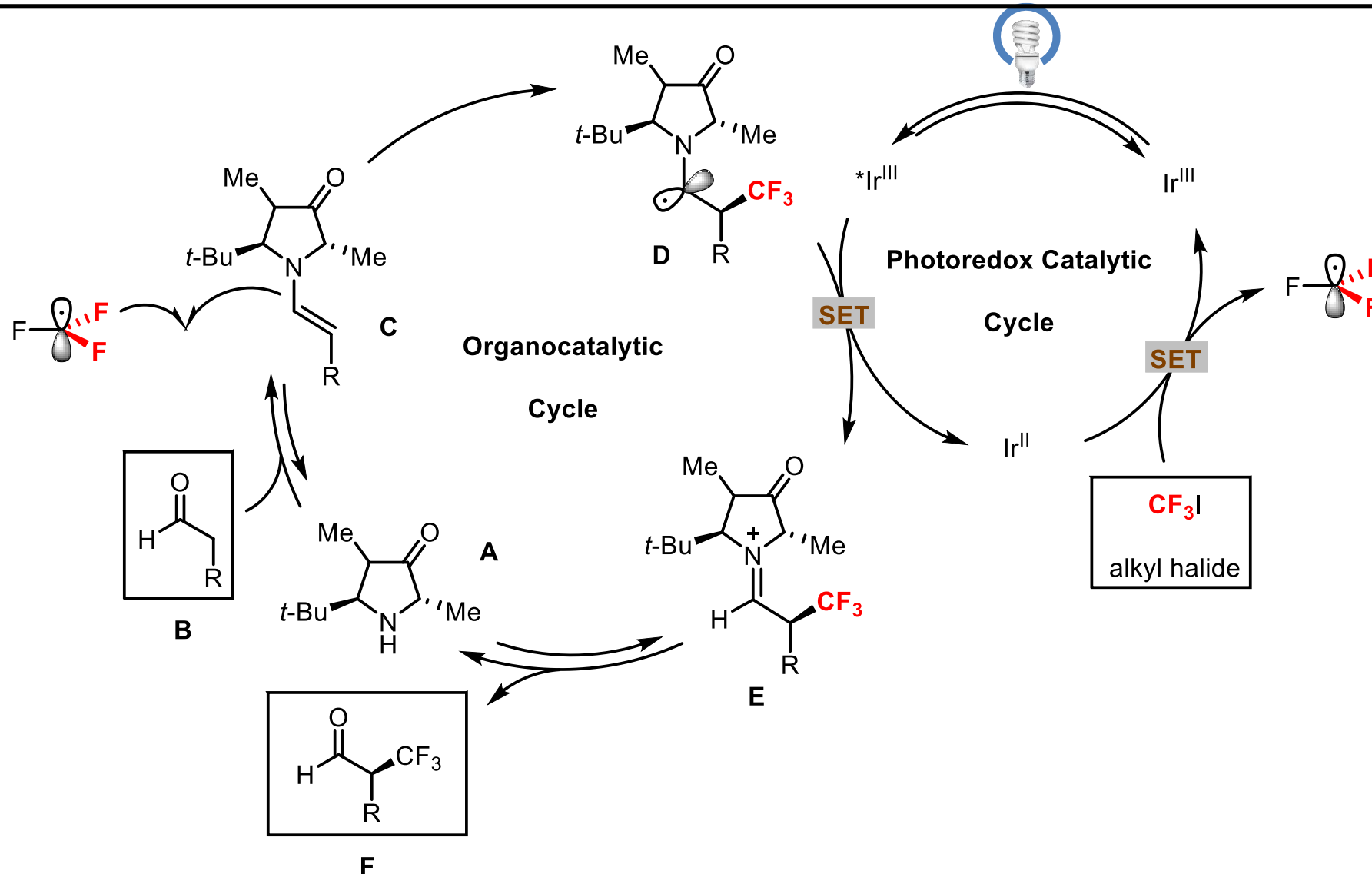
68% yield, 90% ee
>20:1 dr



62% yield, 99% ee
>20:1 dr

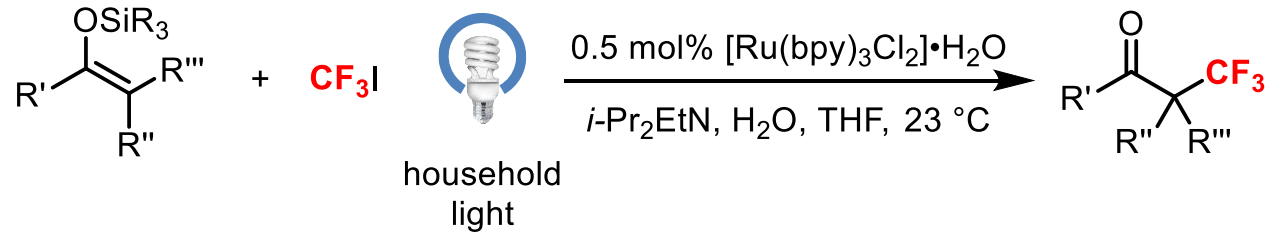
2.2 三氟碘甲烷 (CF₃I)

可能的机理

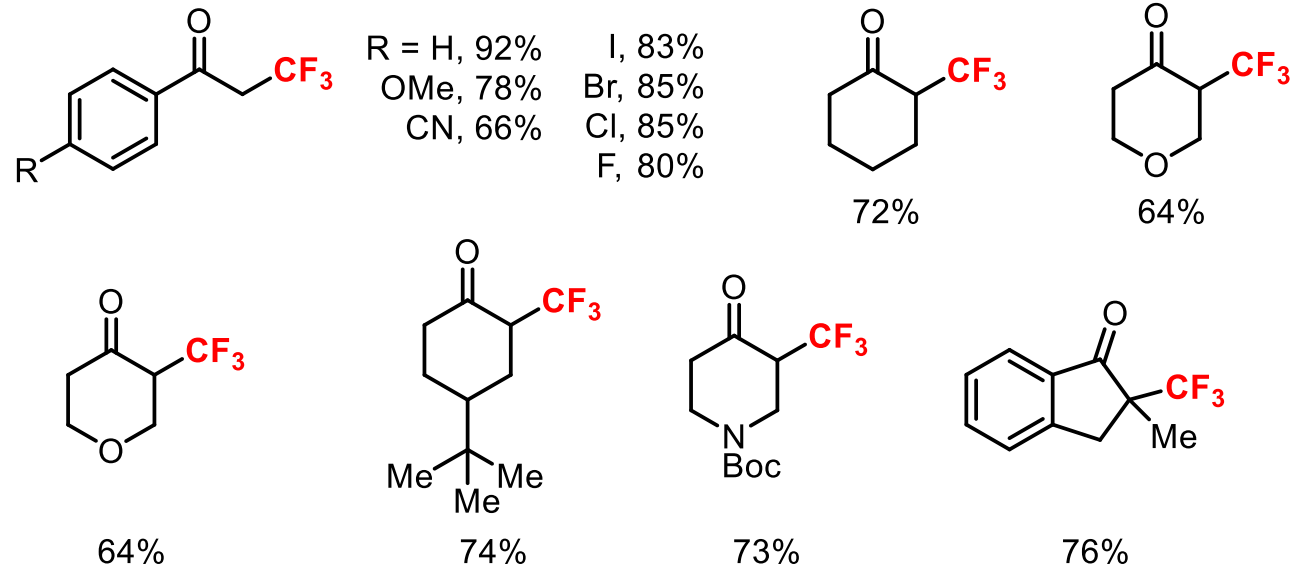


2.2 三氟碘甲烷 (CF₃I)

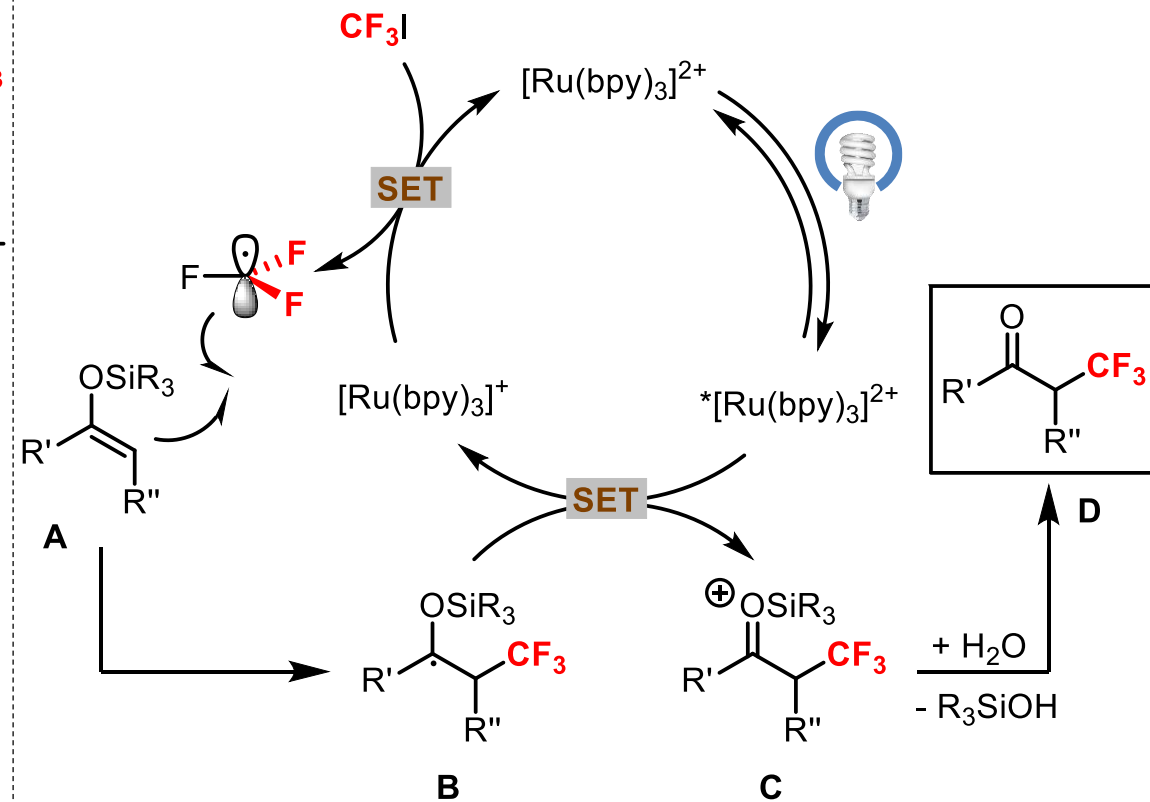
2.2.2 酮羰基α位的三氟甲基化反应



Selected examples

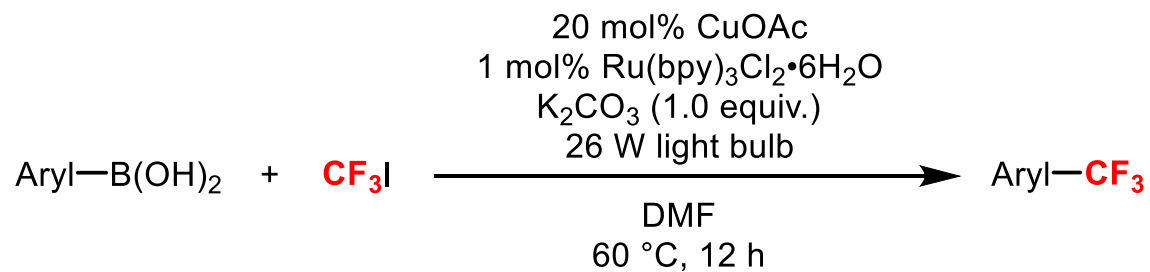


可能的机理

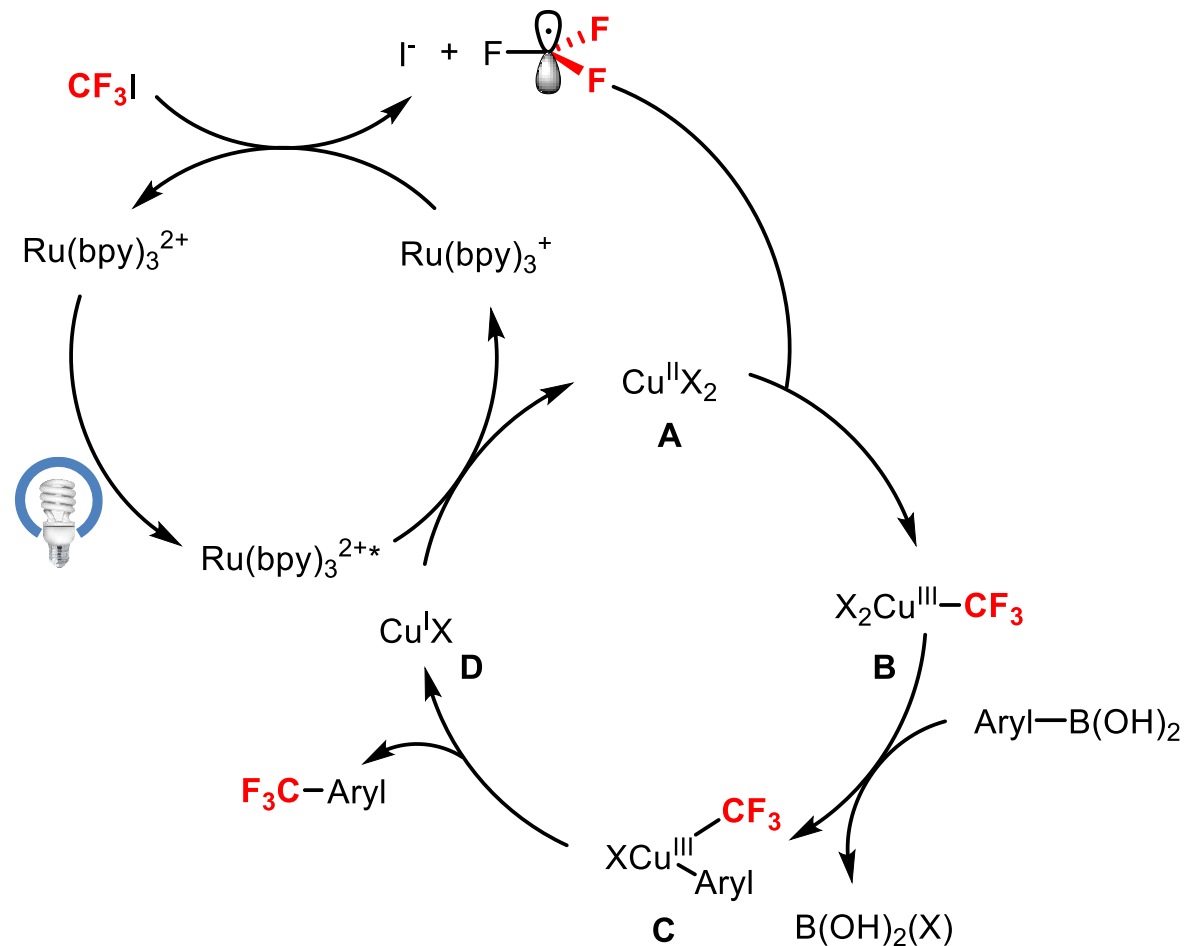
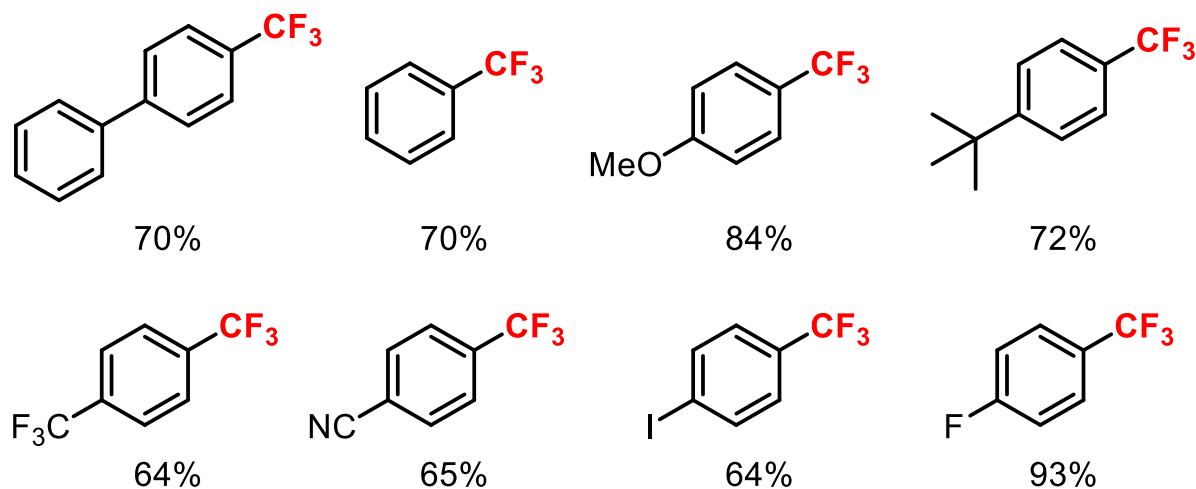


2.2 三氟碘甲烷 (CF₃I)

2.2.3 Cu/光催化的芳基硼酸的三氟甲基化反应

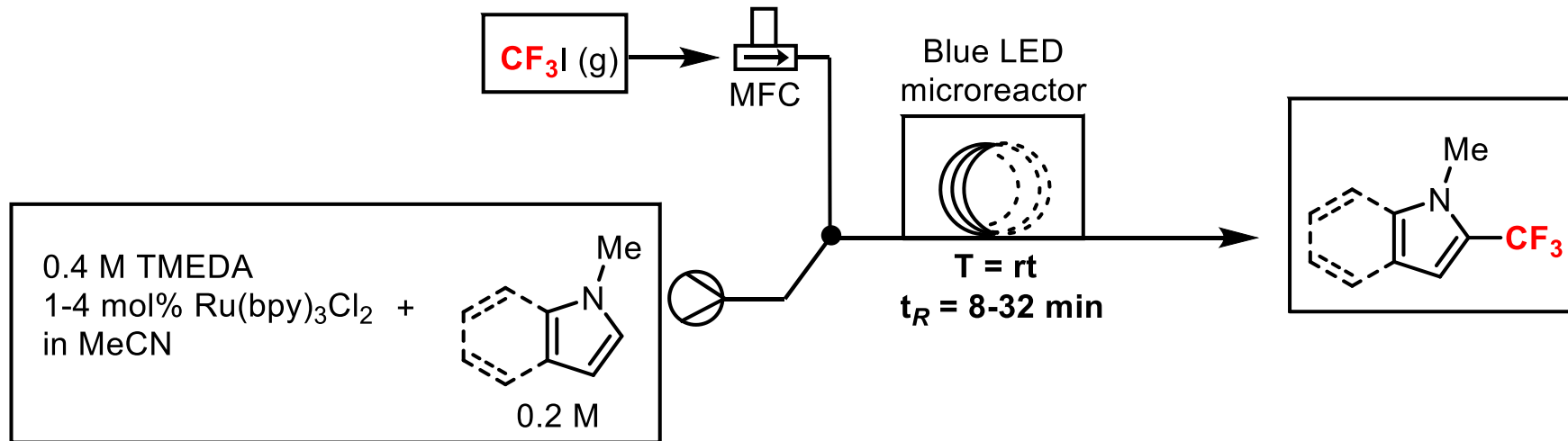


Selected examples

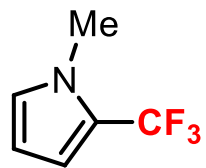


2.2 三氟碘甲烷 (CF₃I)

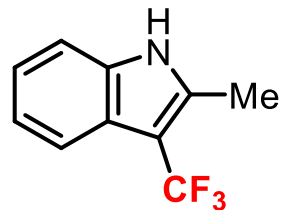
2.2.4 流动化学光催化条件下对五元杂芳环的三氟甲基化反应



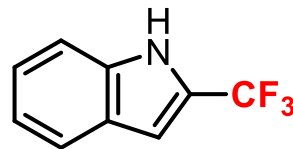
Selected examples



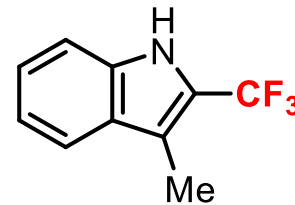
95%
t_R = 8 min



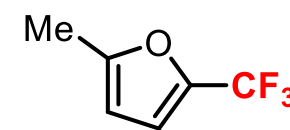
79%
t_R = 8 min



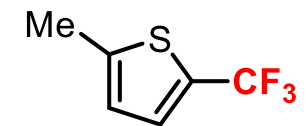
80%
t_R = 8 min



95%
t_R = 8 min



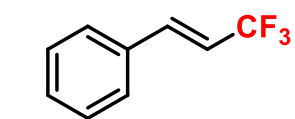
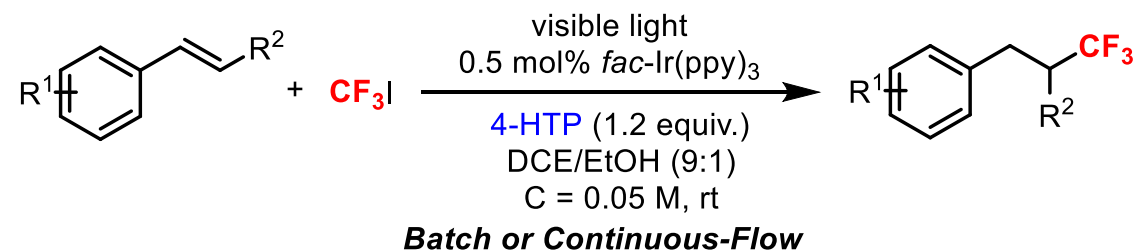
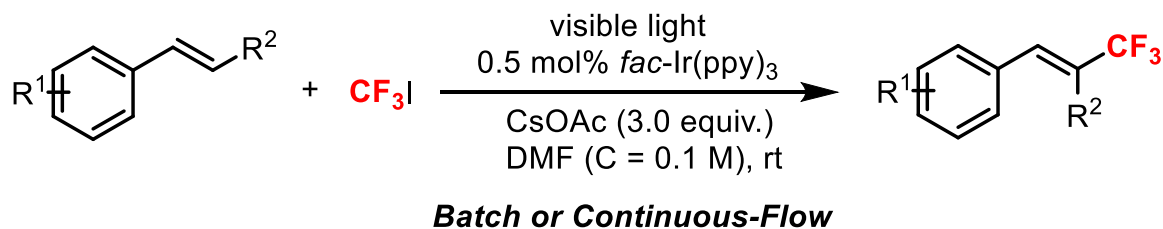
75%
t_R = 16 min



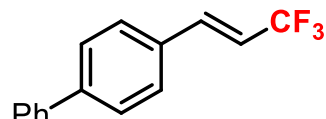
65%
t_R = 16 min

2.2 三氟碘甲烷 (CF₃I)

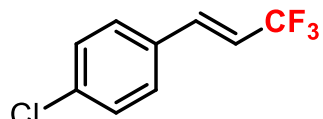
2.2.6 苯乙烯类化合物的三氟甲基化和氢三氟甲基化反应



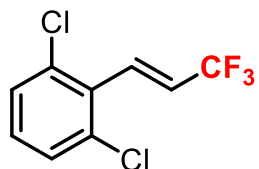
Batch: 75% yield
(72:28 *E/Z*)
Flow: 95% yield
(98:2 *E/Z*)



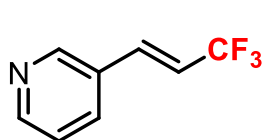
Batch: 89% yield
(83:17 *E/Z*)



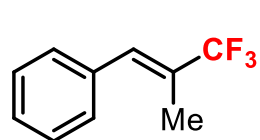
Batch: 88% yield
(54:46 *E/Z*)
Flow: 91% yield
(96:4 *E/Z*)



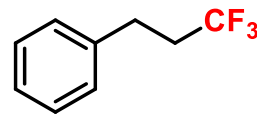
Batch: 75% yield
(68:32 *E/Z*)
Flow: 71% yield
(*E* only)



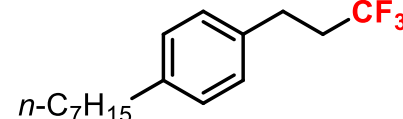
Batch: 75% yield
(75:25 *E/Z*)
Flow: 95% yield
(*E* only)



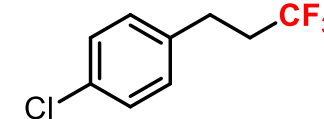
Batch: 75% yield
(69:31 *E/Z*)
Flow: 95% yield
(*E* only)



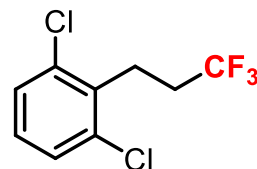
Batch: 79% yield
Flow: 77% (*t_R* = 50 min)



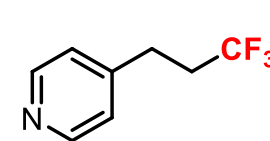
Batch: 88% yield



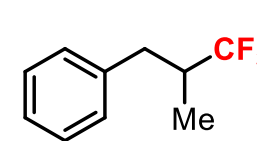
Batch: 71% yield



Batch: 75% yield



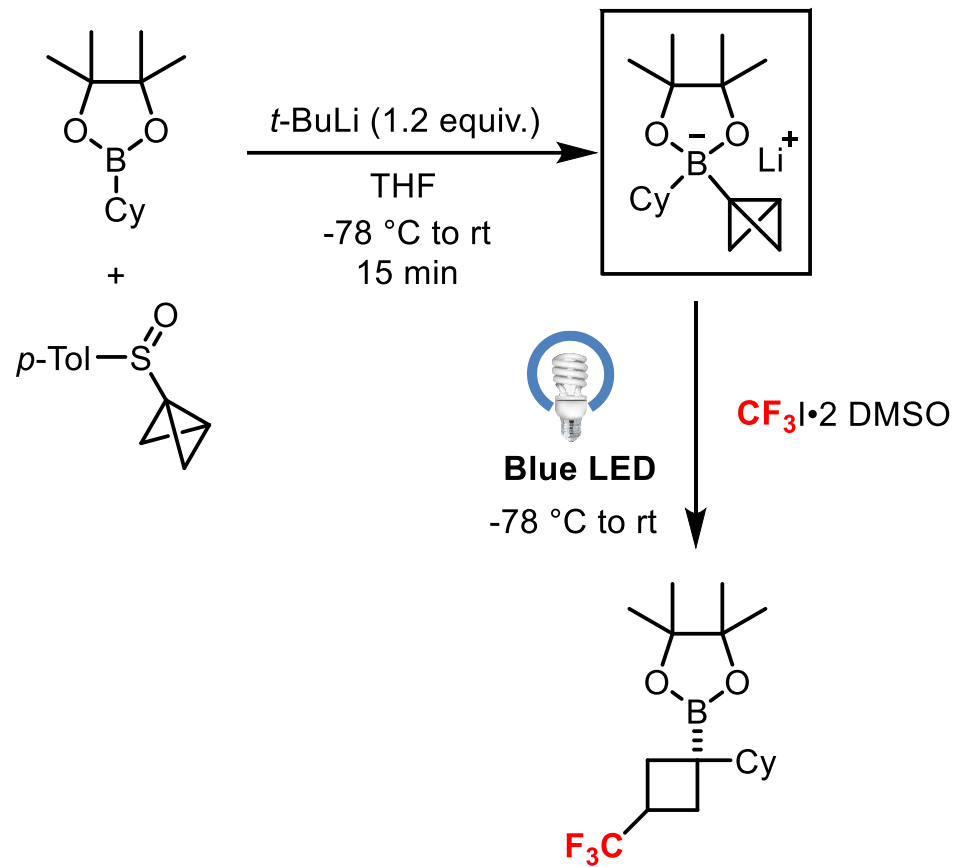
Batch: 72% yield



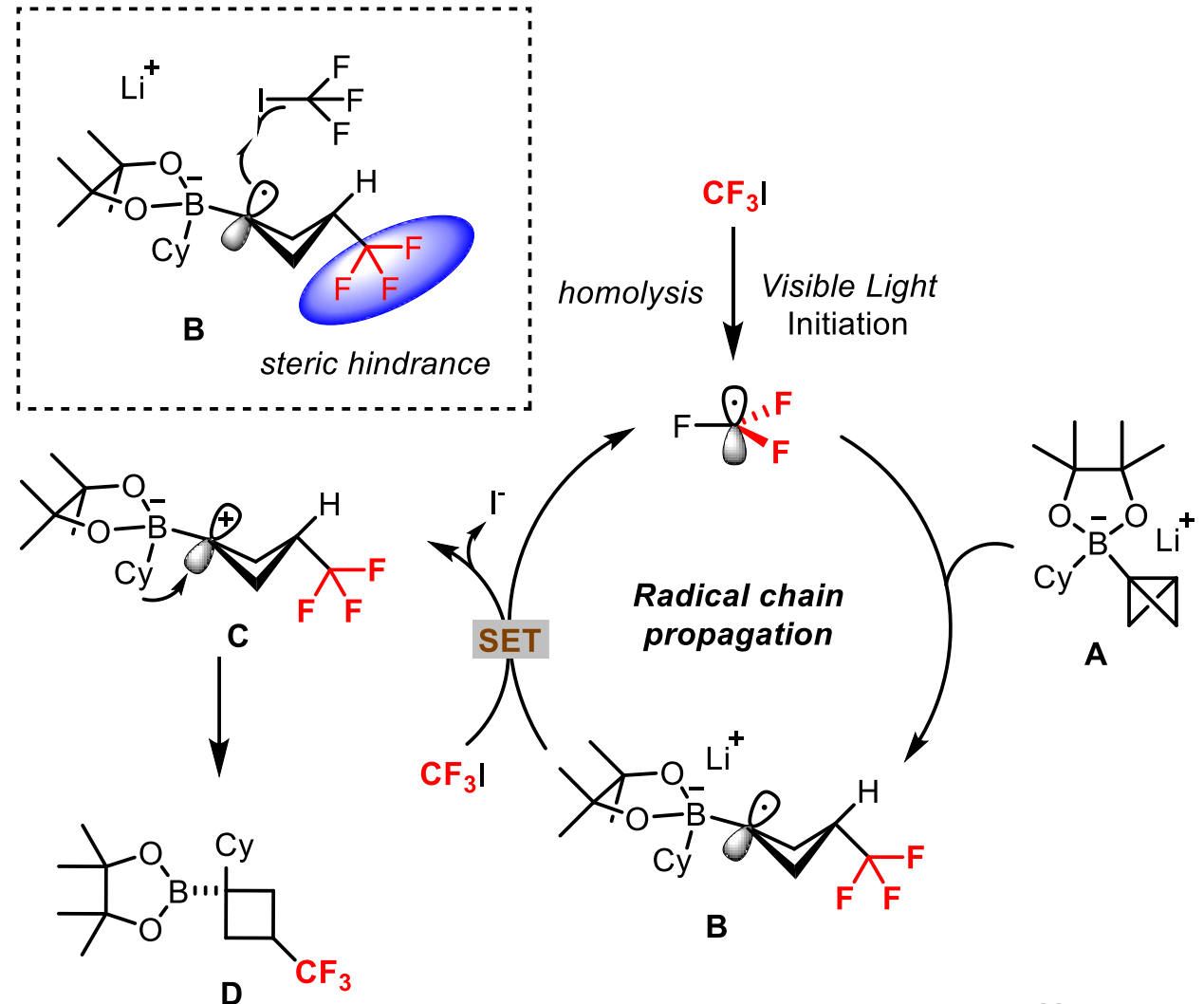
Batch: 82% yield

2.2 三氟碘甲烷 (CF₃I)

2.2.7 CF₃自由基对含张力σ-键的加成反应

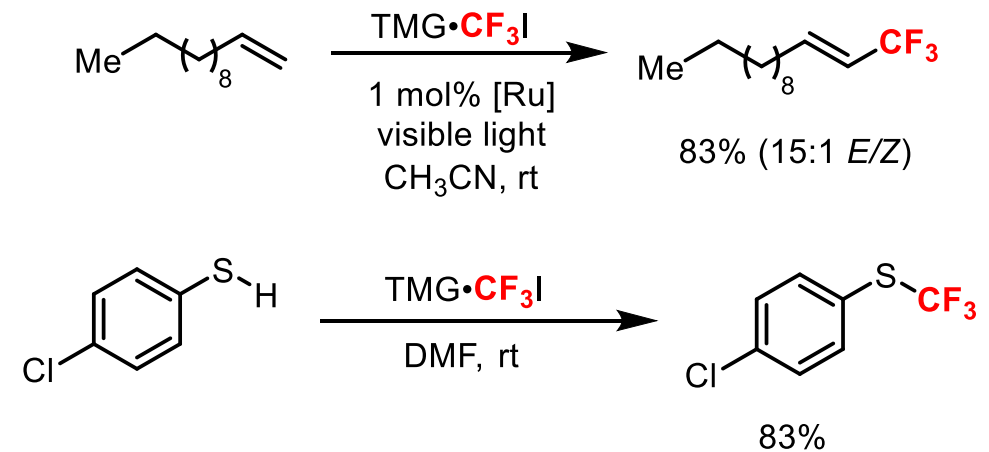
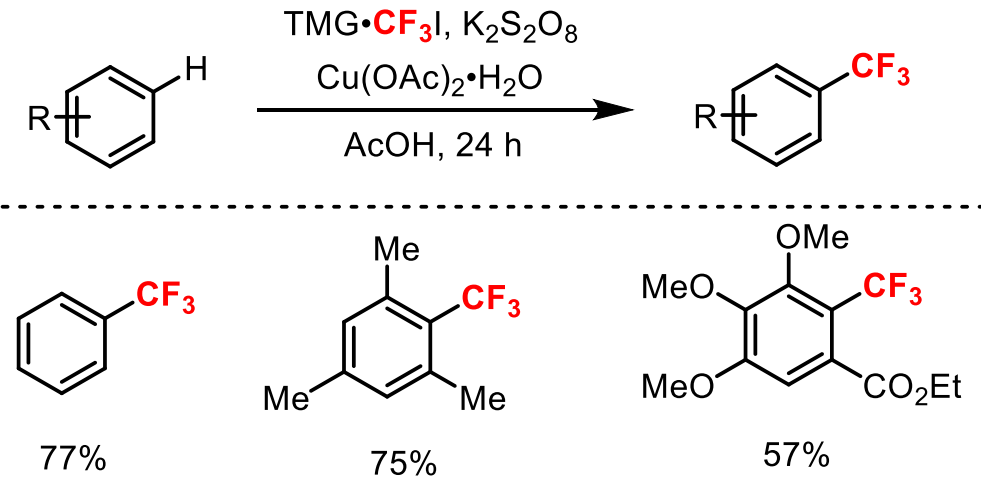
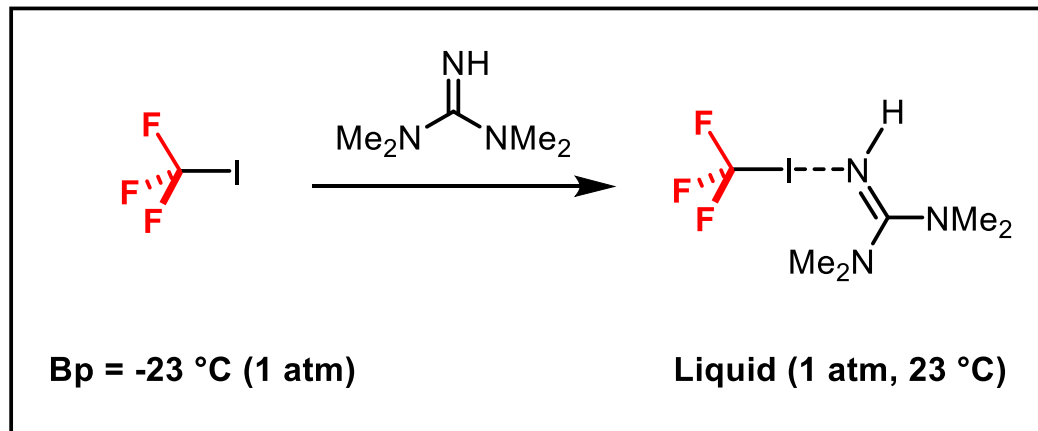
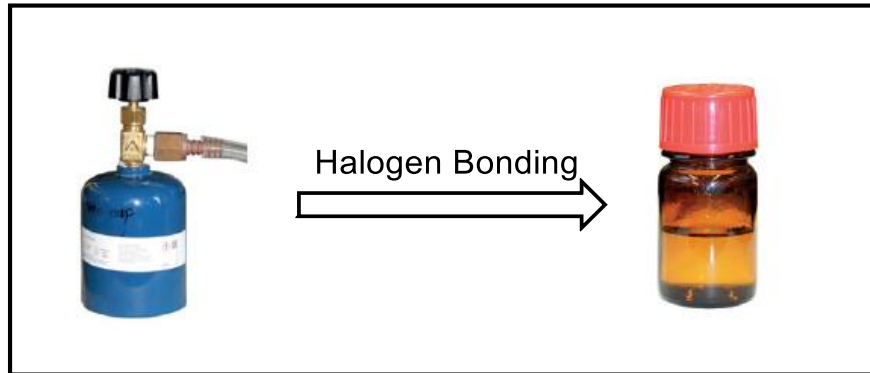


可能的机理



2.2 三氟碘甲烷 (CF₃I)

2.2.8 利用卤键作用制备含CF₃I的商业化溶液





1. 背景介绍

1.1 氟化学背景

1.2 氟烷基化反应简介

1.3 四类含氟C-1烷烃分子简介

2. 含氟C-1分子在氟烷基化反应中的应用

2.1 三氟甲烷 (CF_3H)

2.2 三氟碘甲烷 (CF_3I)

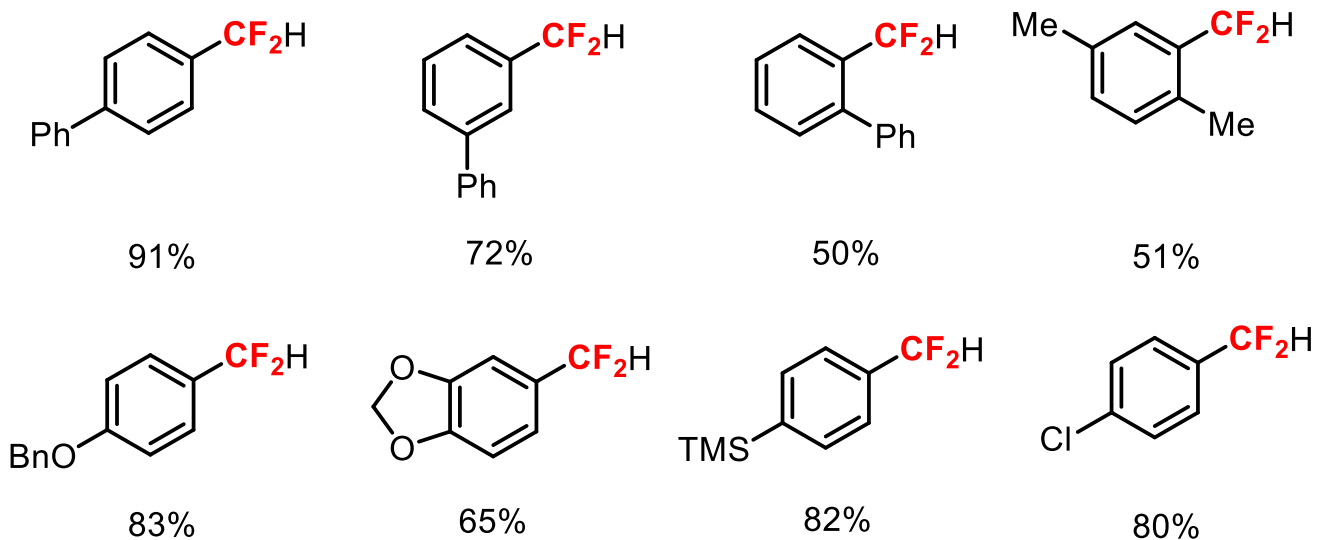
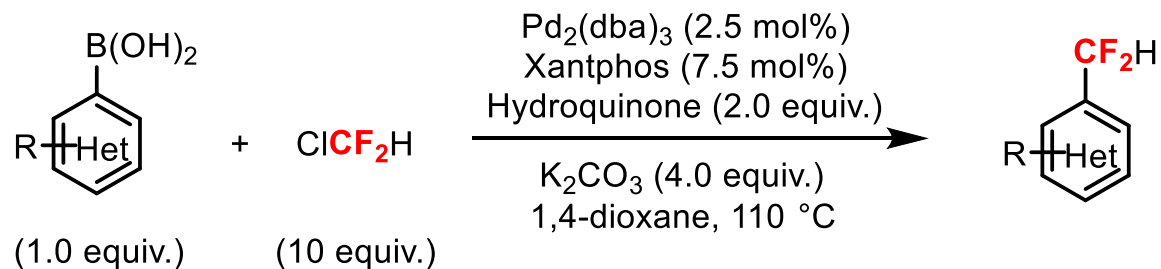
2.3 一氯二氟甲烷 (ClCF_2H)

2.4 一溴二氟甲烷 (BrCF_2H)

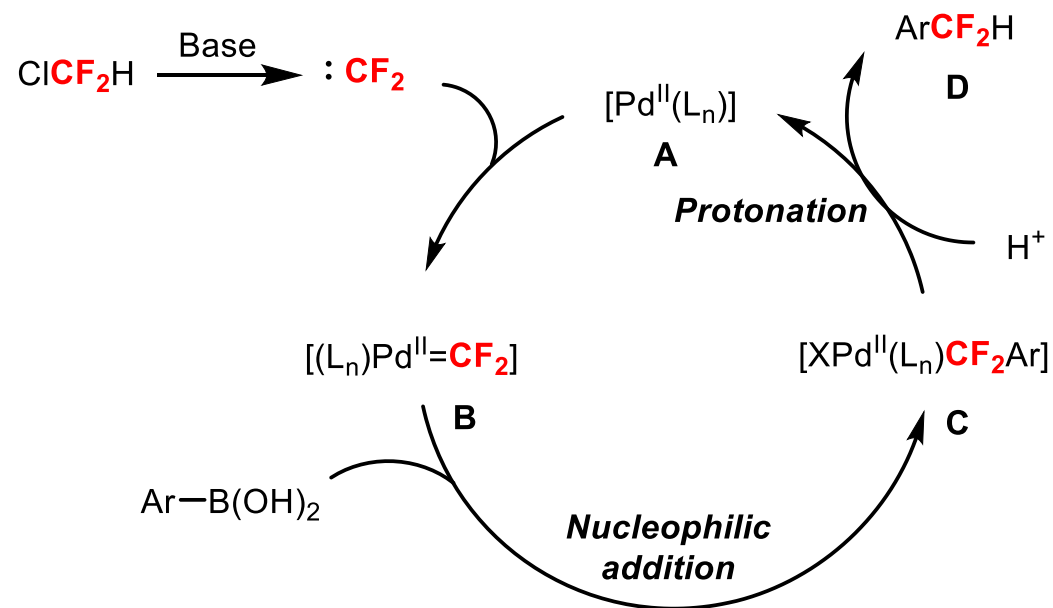
3. 总结与展望

2.3 一氯二氟甲烷 (ClCF₂H)

2.3.1 Pd催化的ClCF₂H对(杂)芳基硼酸或硼酸酯的二氟甲基化反应

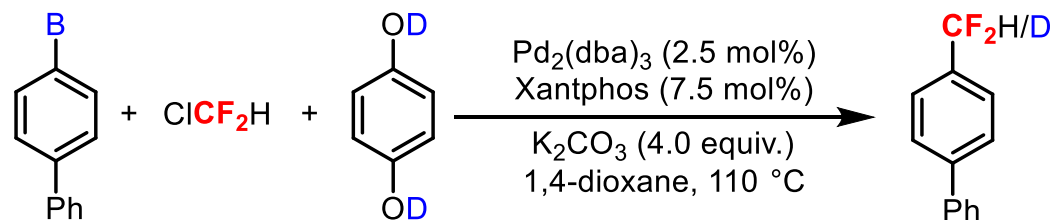
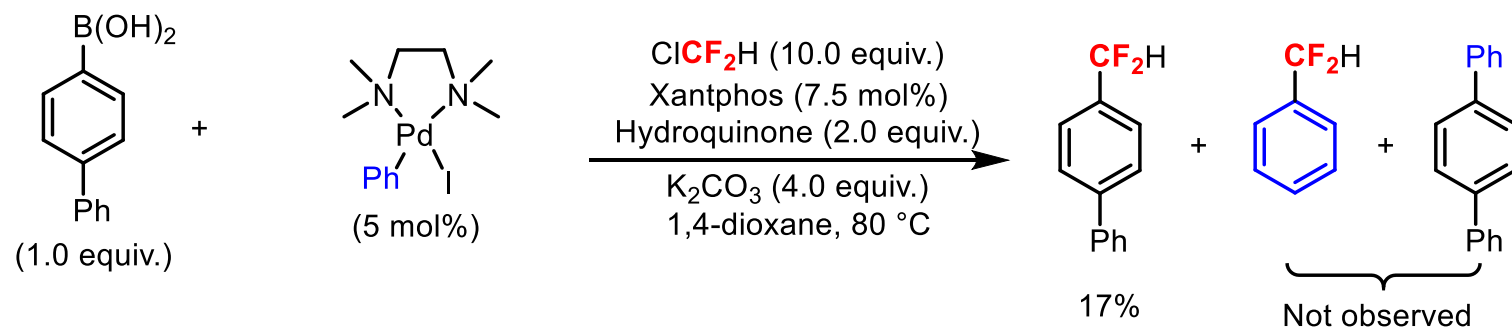
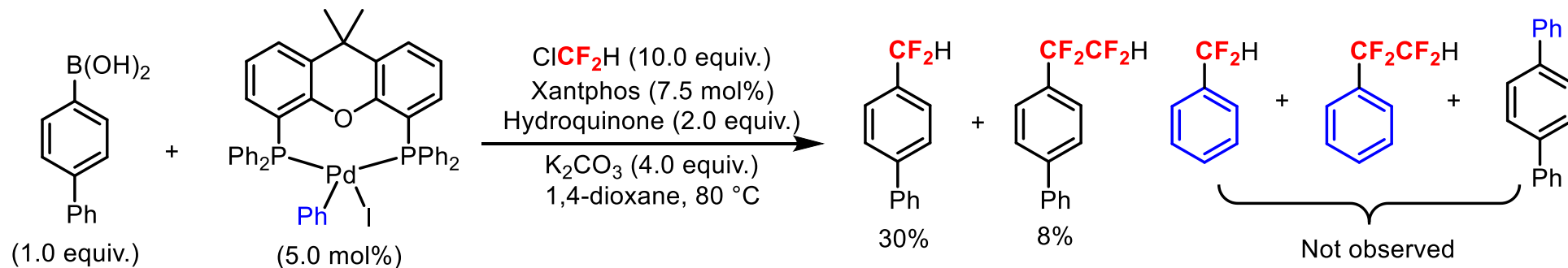


可能的机理



2.3 一氯二氟甲烷 (CICF₂H)

控制实验

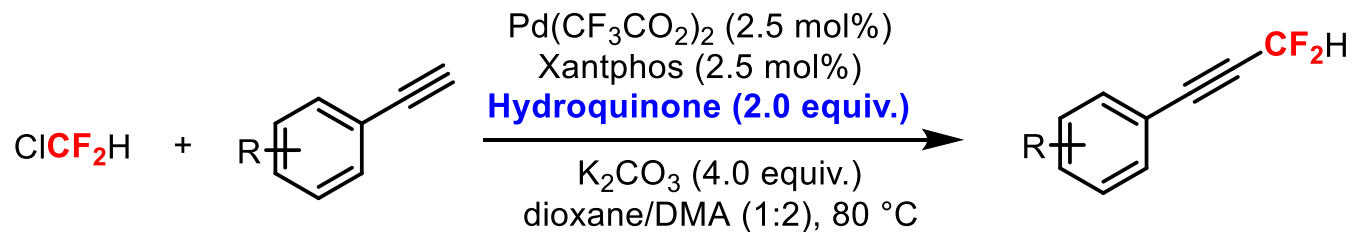


| Entry | Reaction conditions | Yield (%), 3-H/3-D |
|-------|--|--------------------|
| 1 | B = B(OH) ₂ , CICF ₂ H | 57/22 |
| 2 | B = Beg, CICF ₂ H | 40/27 |
| 3 | No hydroquinone | 1/- |

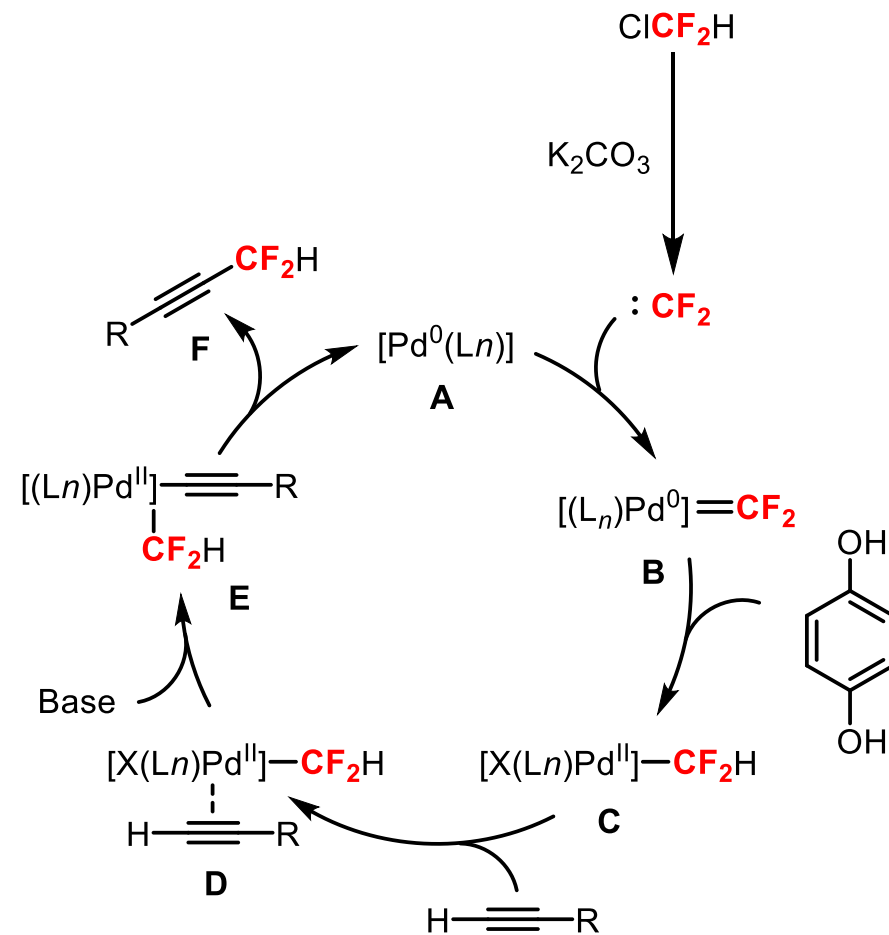
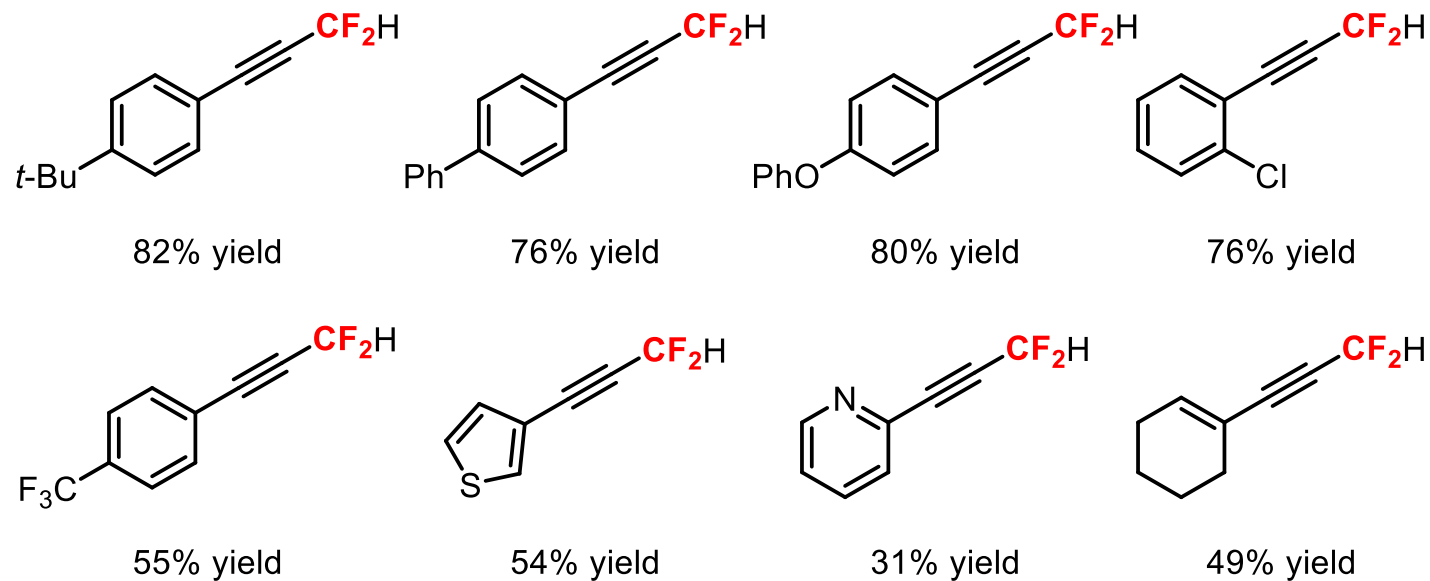
2.3 一氯二氟甲烷 (ClCF₂H)

2.3.2 Pd催化的ClCF₂H和端炔的交叉偶联反应

可能的机理

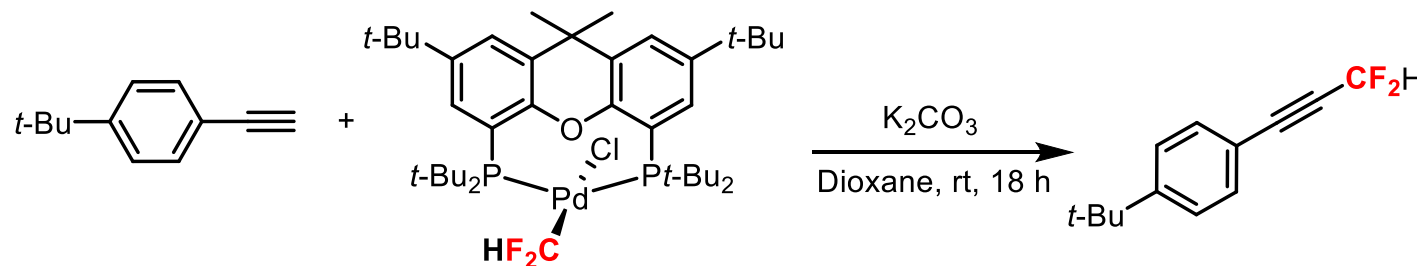
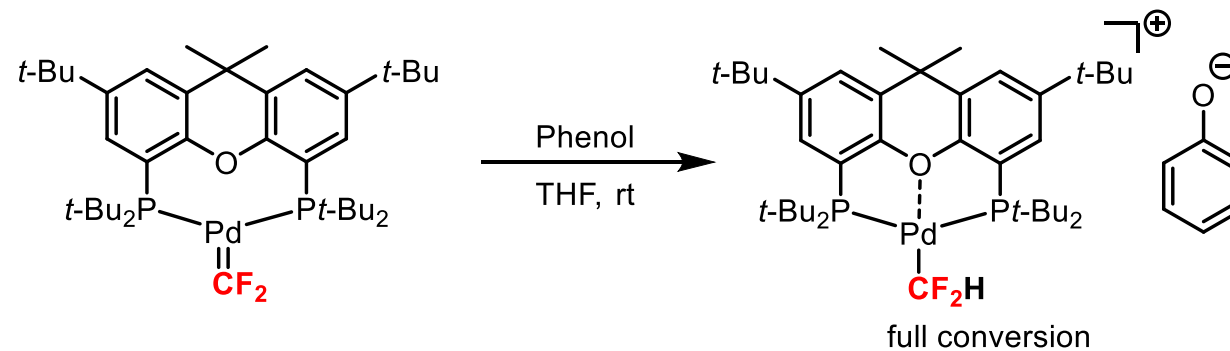
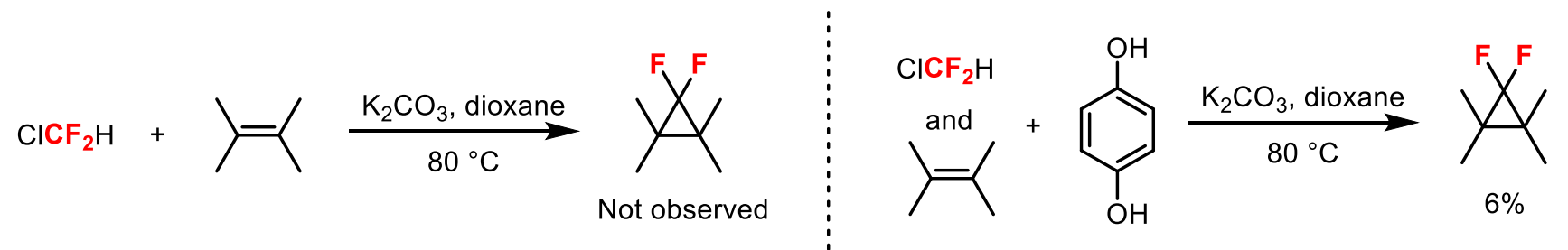


Selected examples



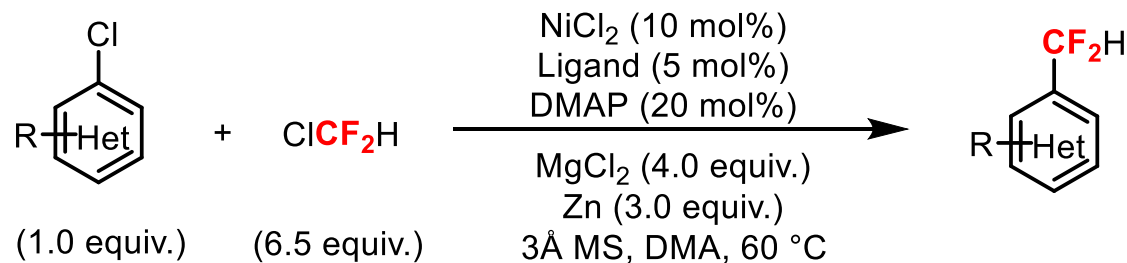
2.3 一氯二氟甲烷 (ClCF₂H)

控制实验

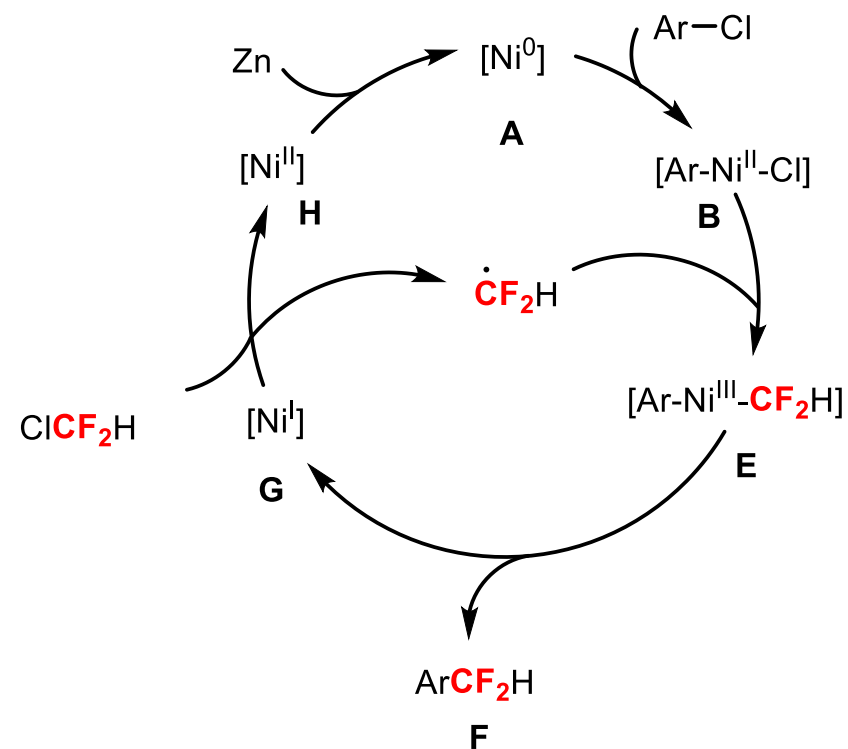
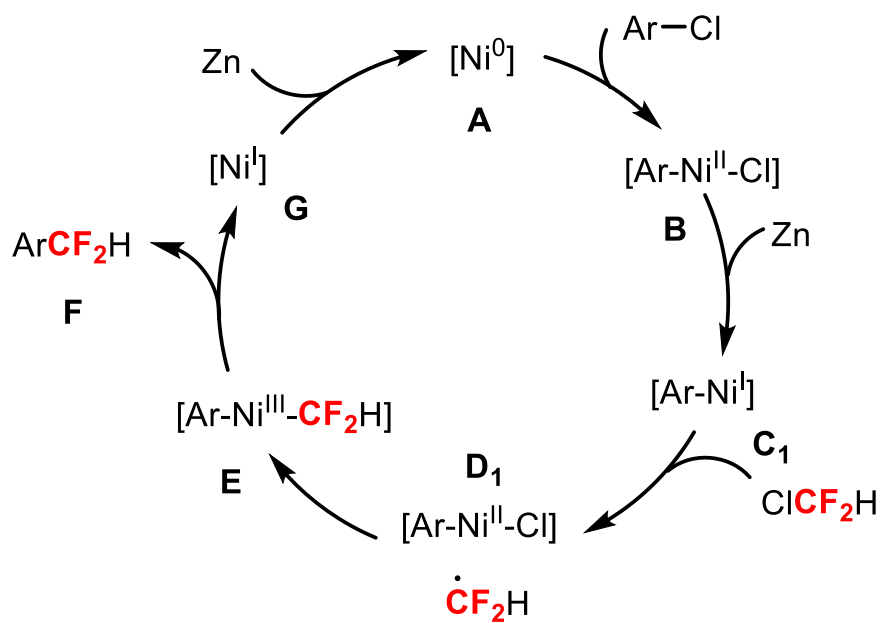


2.3 一氯二氟甲烷 (CICF₂H)

2.3.3 Ni催化(杂)芳基氯化物与CICF₂H的二氟甲基化反应

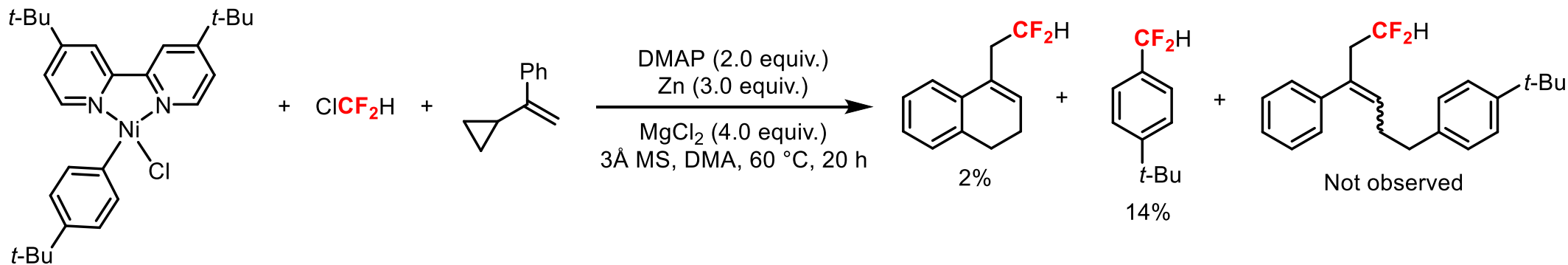
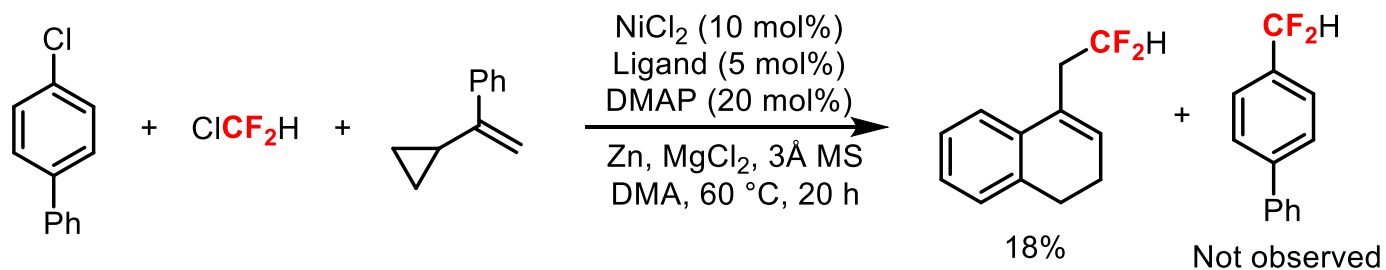
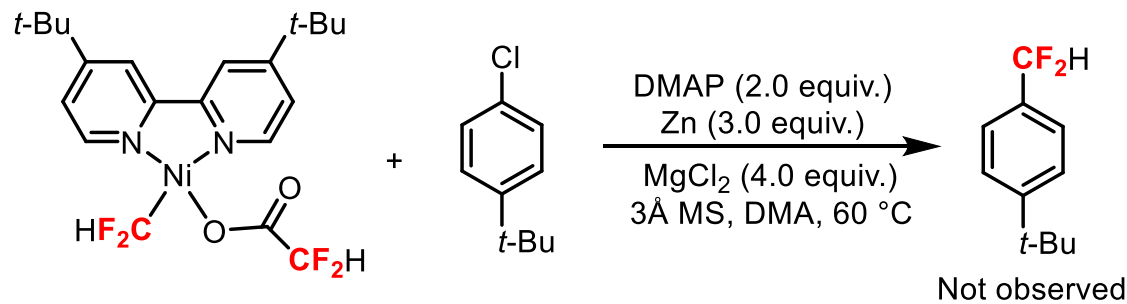


可能的机理



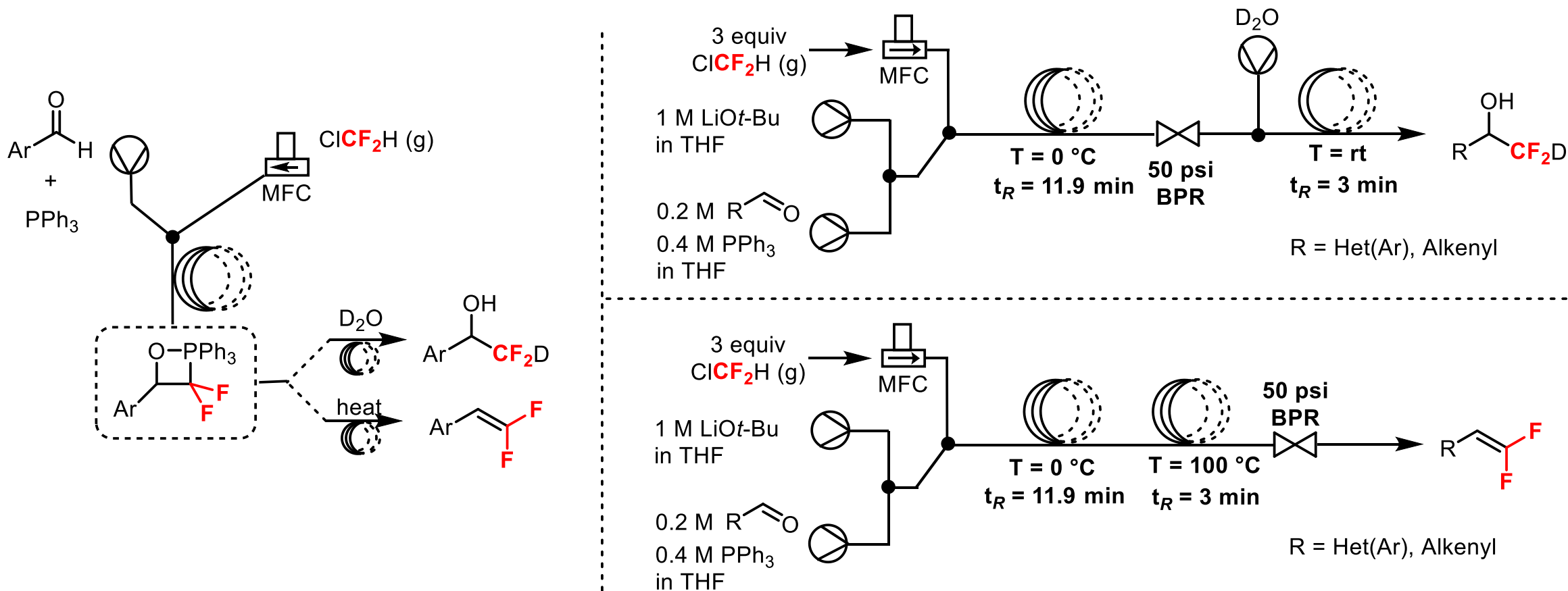
2.3 一氯二氟甲烷 (ClCF₂H)

控制实验



2.3 一氯二氟甲烷 (ClCF₂H)

2.3.4 连续流条件下醛的氘二氟甲基化和偕二氟烯基化反应





1. 背景介绍

1.1 氟化学背景

1.2 氟烷基化反应简介

1.3 四类含氟C-1烷烃分子简介

2. 含氟C-1分子在氟烷基化反应中的应用

2.1 三氟甲烷 (CF_3H)

2.2 三氟碘甲烷 (CF_3I)

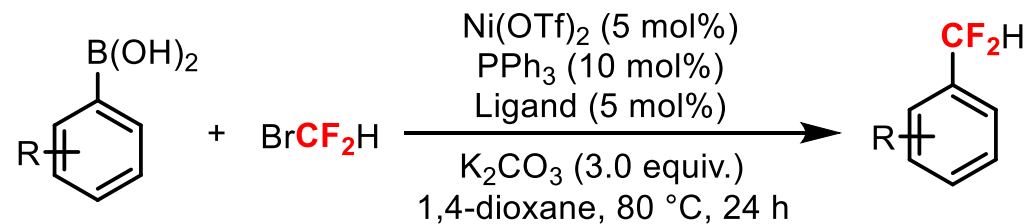
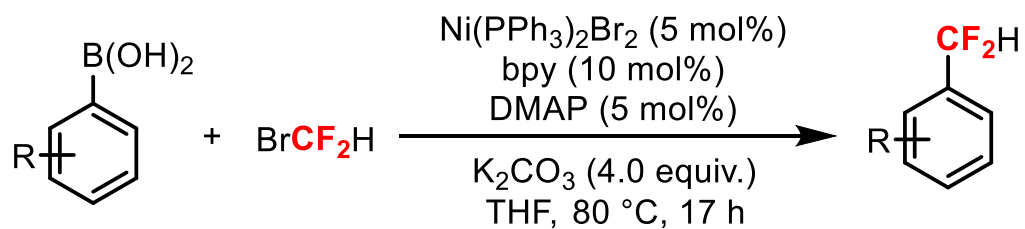
2.3 一氯二氟甲烷 (ClCF_2H)

2.4 一溴二氟甲烷 (BrCF_2H)

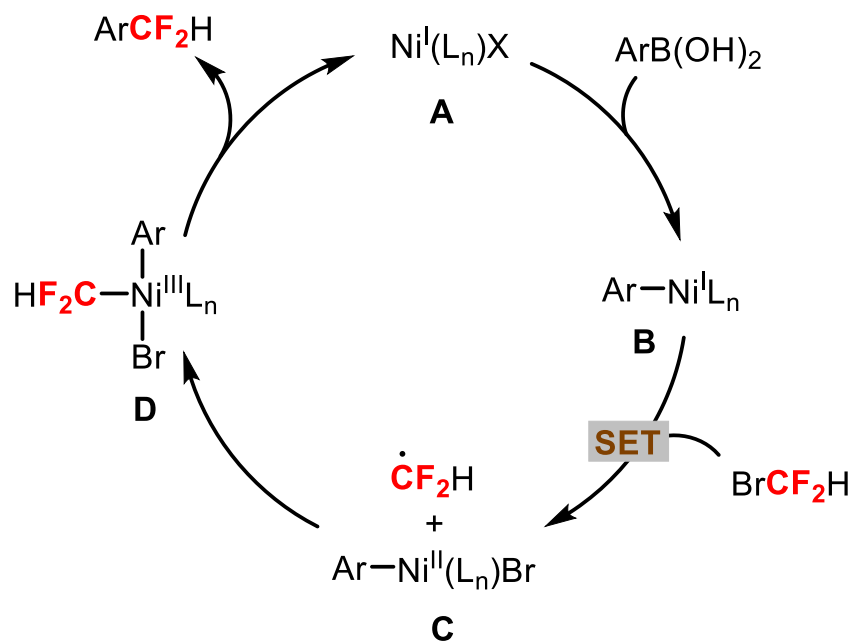
3. 总结与展望

2.4 一溴二氟甲烷 (BrCF₂H)

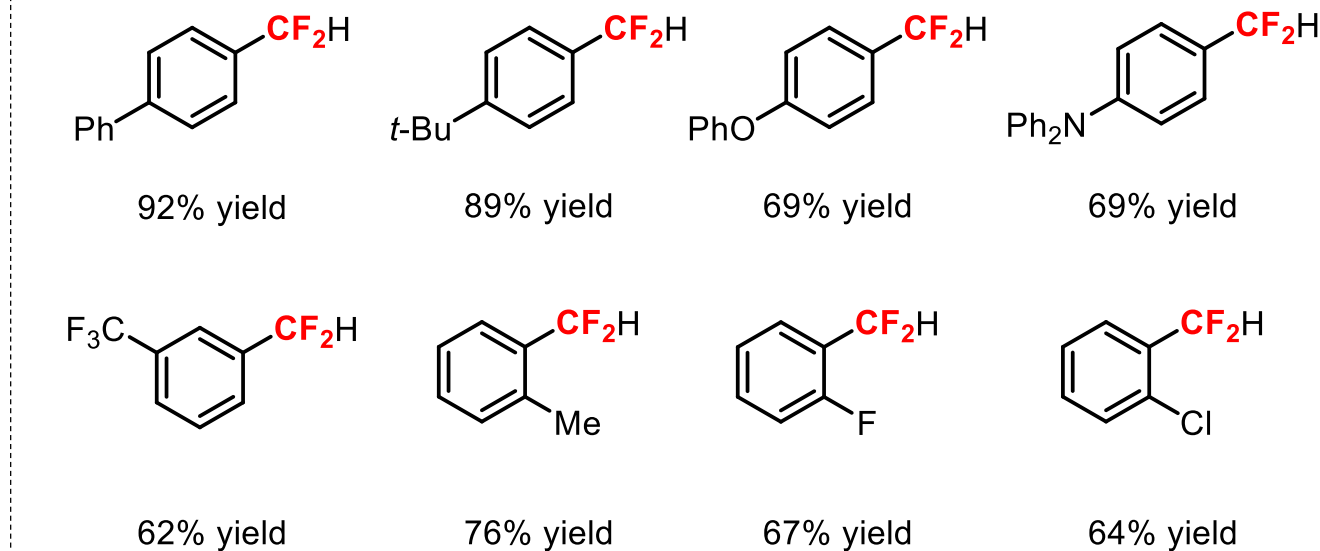
2.4.1 Ni催化的芳基硼酸与BrCF₂H的交叉偶联反应



可能的机理



Zhang, X. et al. *Chin. J. Chem.* **2018**, 36, 143.

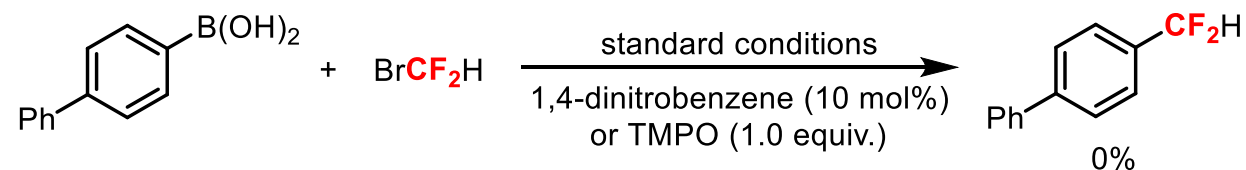


Wang, X. S. et al. *Org. Chem. Front.* **2018**, 5, 606.

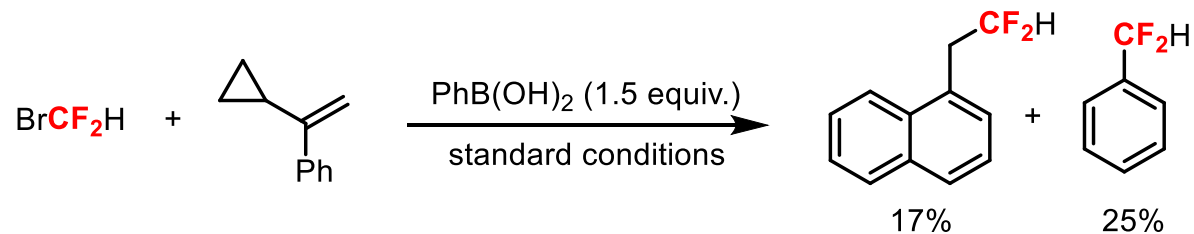
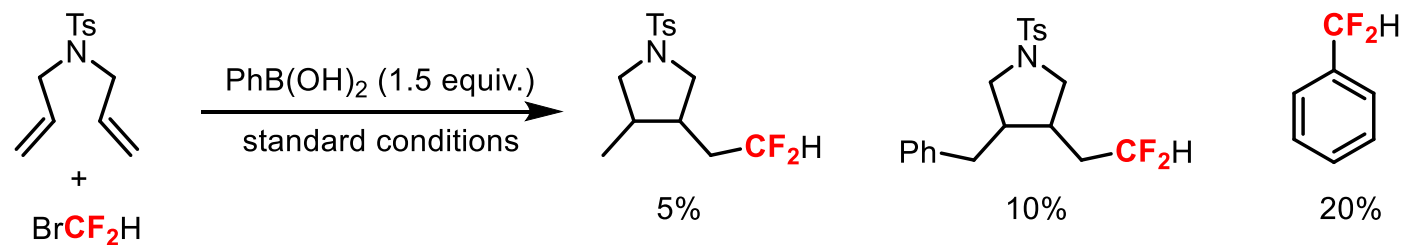
2.4 一溴二氟甲烷 (BrCF₂H)

控制实验

Radical inhibition experiment

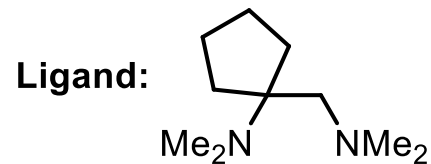
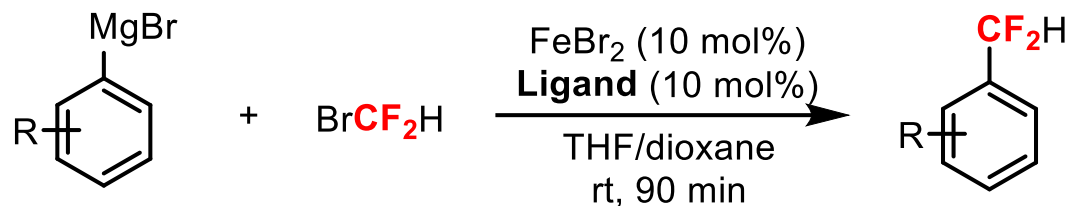


Radical clock experiments

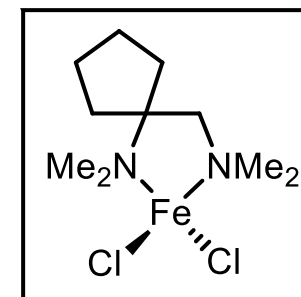
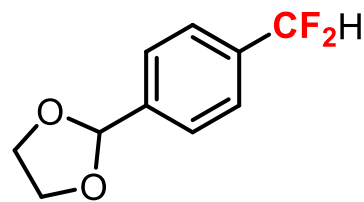
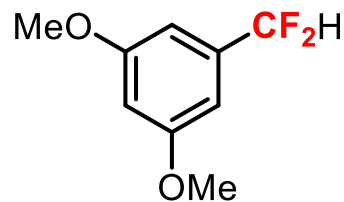
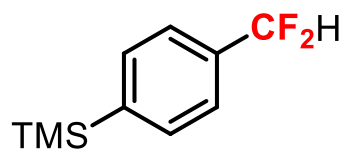
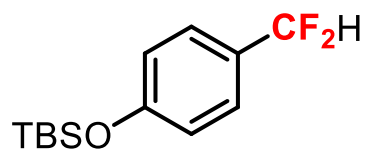
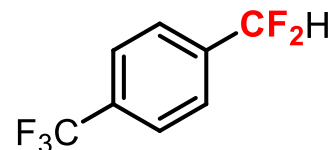
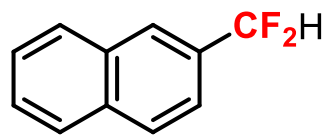
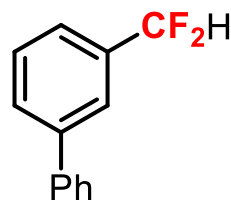
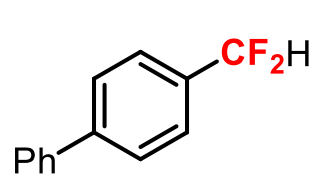


2.4 一溴二氟甲烷 (BrCF₂H)

2.4.2 Fe催化的芳基溴化镁和BrCF₂H的交叉偶联反应



Selected examples

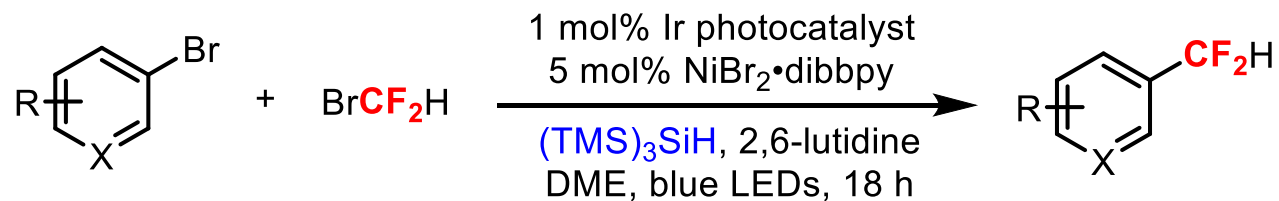


N-Fe-N angle:
82.09(11)°-82.81(10)°

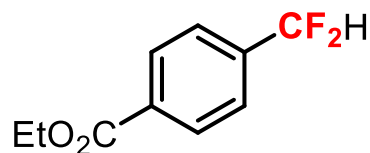


2.4 一溴二氟甲烷 (BrCF₂H)

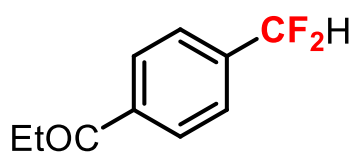
2.4.3 Ni/光催化的芳基溴化物的二氟甲基化反应



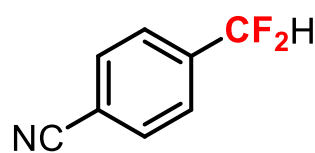
Selected examples



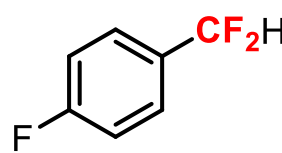
83%



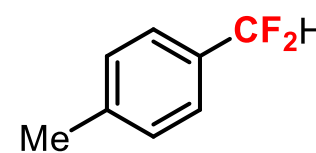
81%



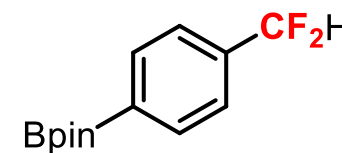
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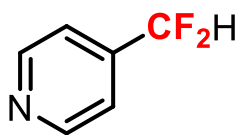
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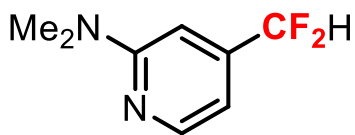
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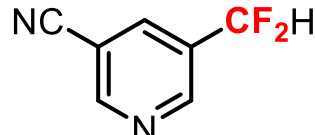
85%



54%



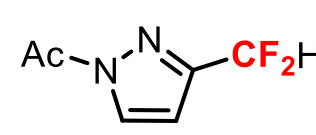
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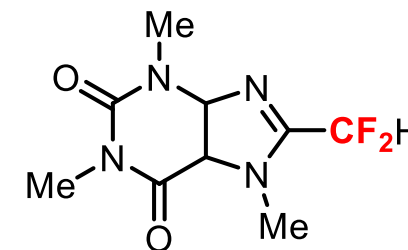
69%



65%



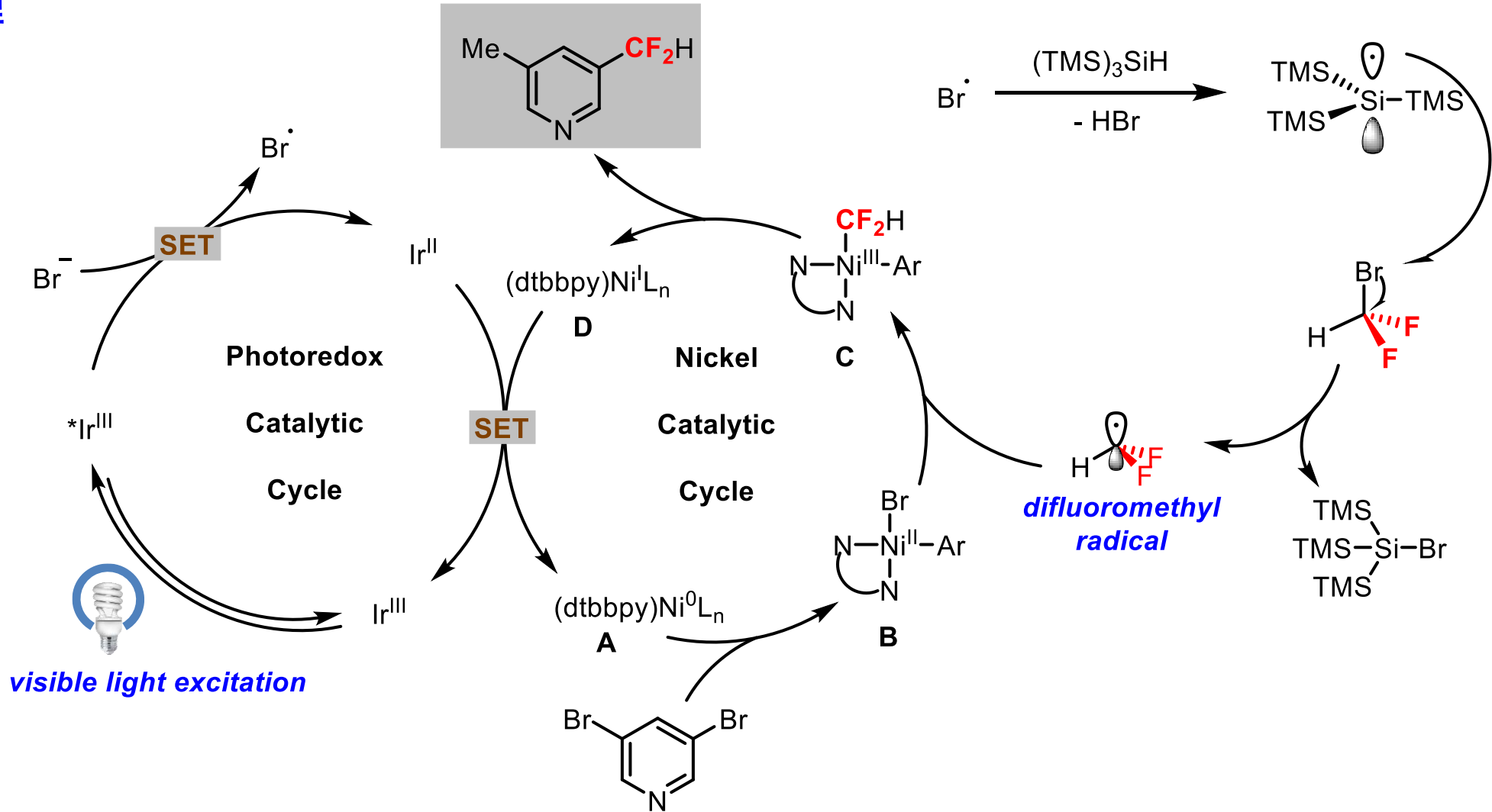
51%



75%

2.4 一溴二氟甲烷 (BrCF₂H)

可能的机理



3. 总结与展望



- 原料廉价，毒性低
- 无需多步合成，原子经济性高

- 气液反应操作起来困难
- 沸点低，安全性问题

 流动化学

展 望

- 发展更多的试剂作为含氟气体储存体系。
- 发展新的含氟气体活化模式。
- 各个反应体系可经流动化学提高效率，扩大反应规模。

谢谢大家！

请各位老师同学批评指正！