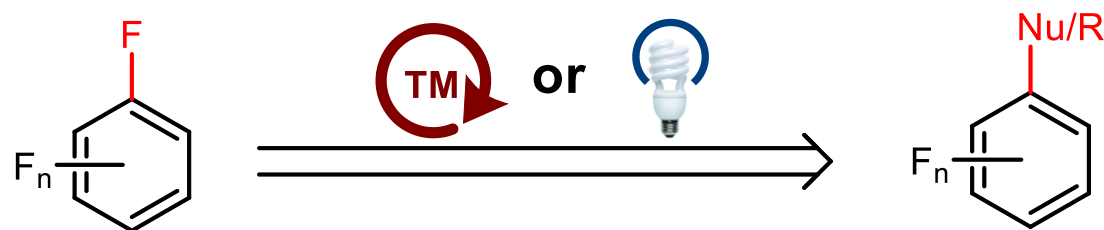


# 多氟芳烃的脱氟官能团化反应

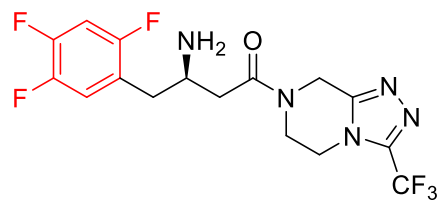


汇报人：赵佳奥  
指导老师：麻生明 教授  
钱辉 副教授  
汇报时间：2025年10月24日

1. 背景介绍
2. 多氟芳烃的脱氟官能团化反应
  - 2.1 过渡金属参与的脱氟官能团化反应
  - 2.2 光催化的脱氟官能团化反应
3. 总结与展望

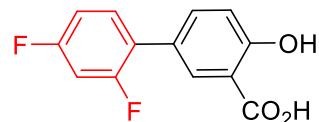
1. 背景介绍
2. 多氟芳烃的脱氟官能团化反应
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3. 总结与展望

# 1、背景介绍



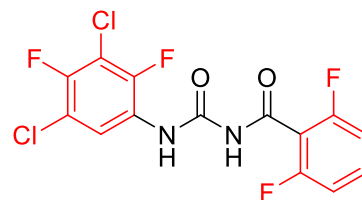
Sitagliptin

用于治疗II型糖尿病



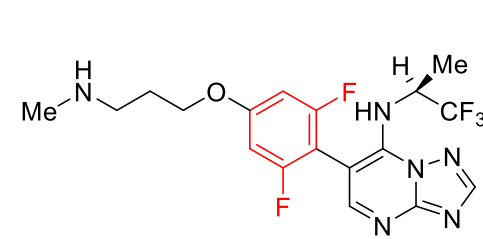
Diflunisal

抗炎药



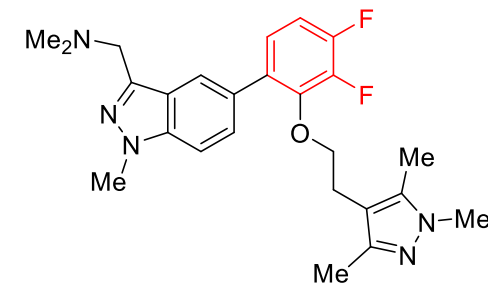
Teflubenzuron

杀虫剂



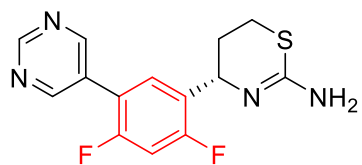
Cevipabulin

具有抗肿瘤活性



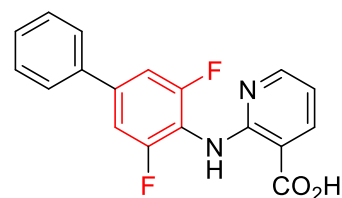
IMP-1088

NMT1、NMT2 抑制剂



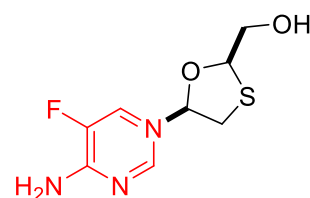
LY2811376a

BACE1抑制剂



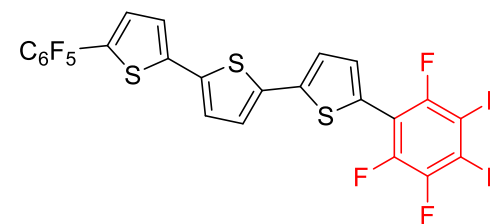
ASLAN003

DHODH抑制剂

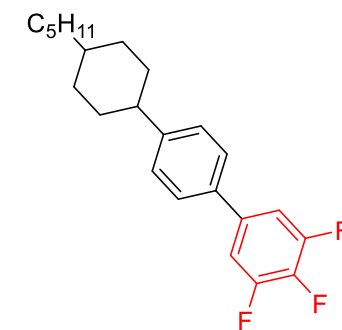


Emtricitabine

抗病毒活性



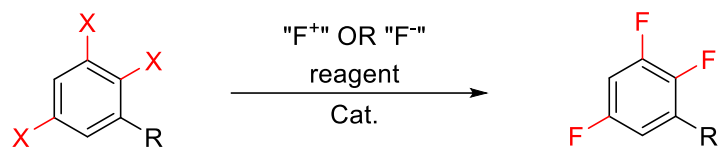
N-型半导体



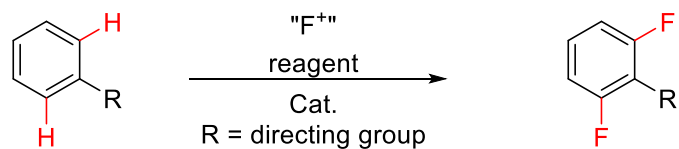
液晶材料

# 1、背景介绍

## 1、采用F<sup>+</sup>或F<sup>-</sup>实现直接氟化：

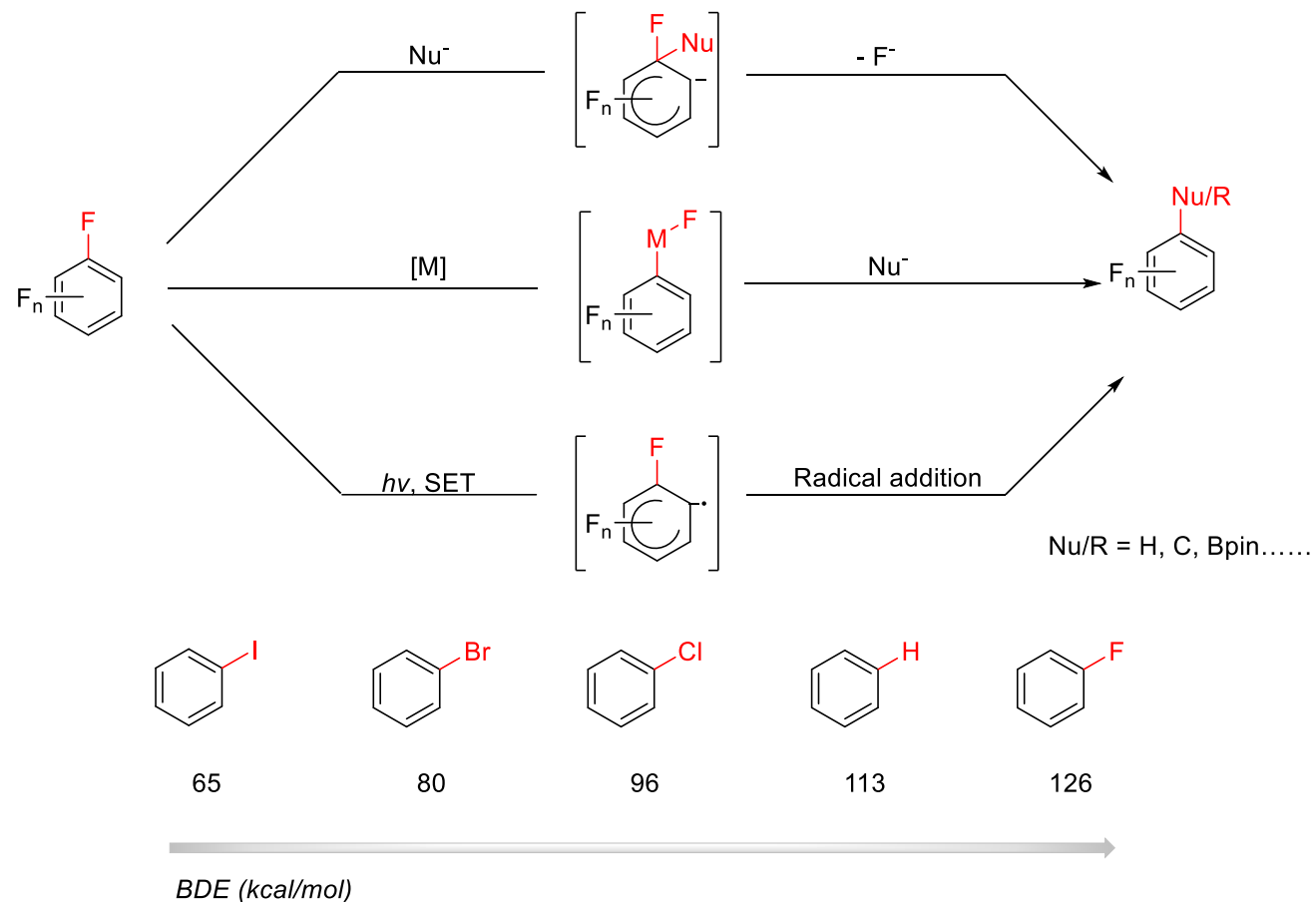


For F<sup>-</sup>, X = -I, -OTf, -IAr. For F<sup>+</sup>, X = B(X)<sub>3</sub>, SnR<sub>3</sub>, SiR<sub>4</sub>



R = directing group

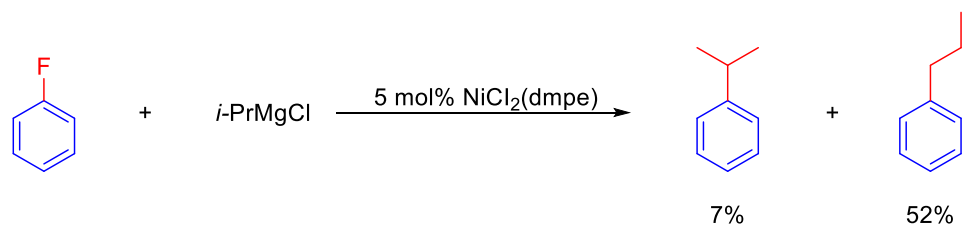
## 2、利用多氟芳烃进行脱氟官能团化制备：



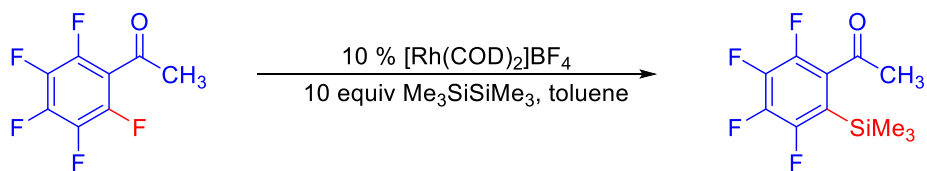
1. 背景介绍
2. 多氟芳烃的脱氟官能团化反应
  - 2.1 过渡金属参与的脱氟官能团化反应
  - 2.2 光催化的脱氟官能团化反应
3. 总结与展望

## 2.1、过渡金属参与的脱氟反应

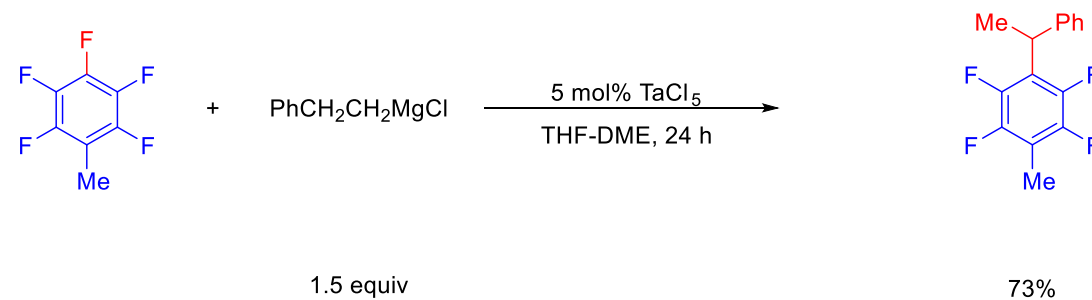
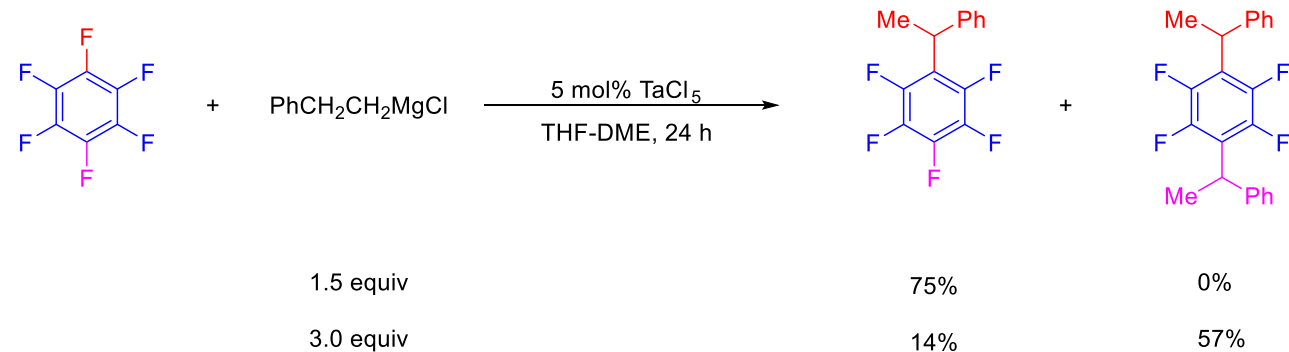
**Kumada, 1973:**



**Murai, 1998:**

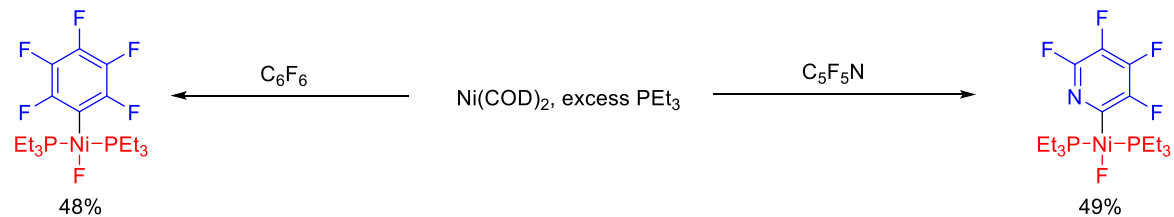


**Takahashi, 2006 :**

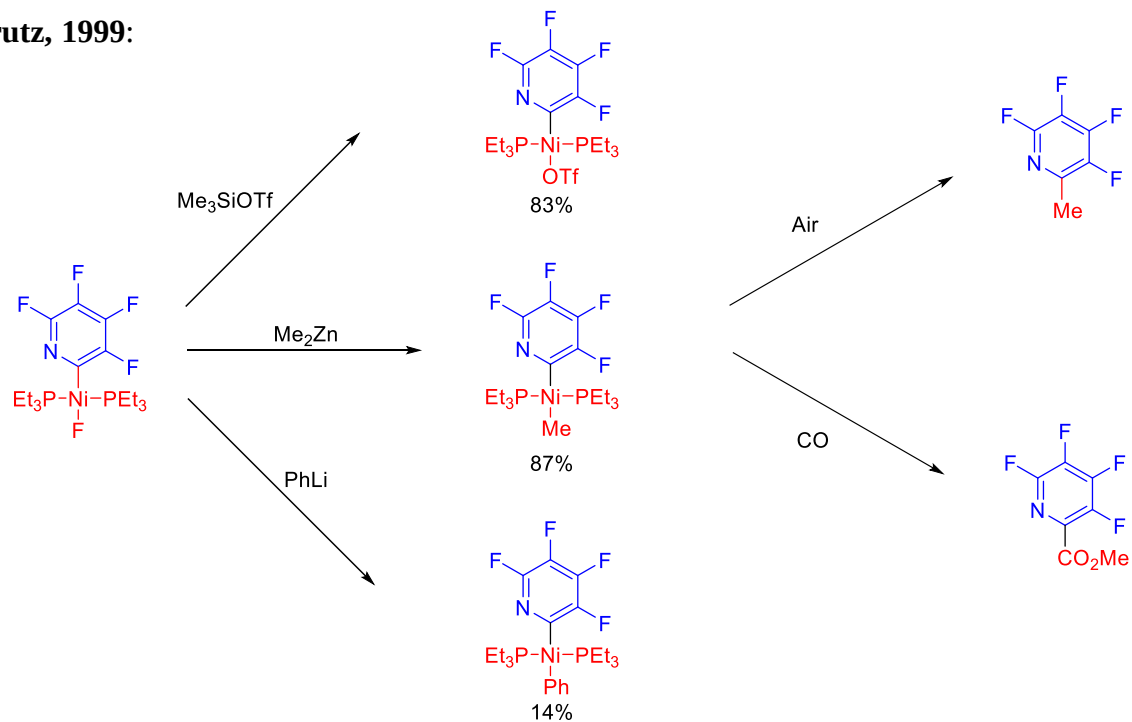


## 2.1、过渡金属参与的脱氟反应

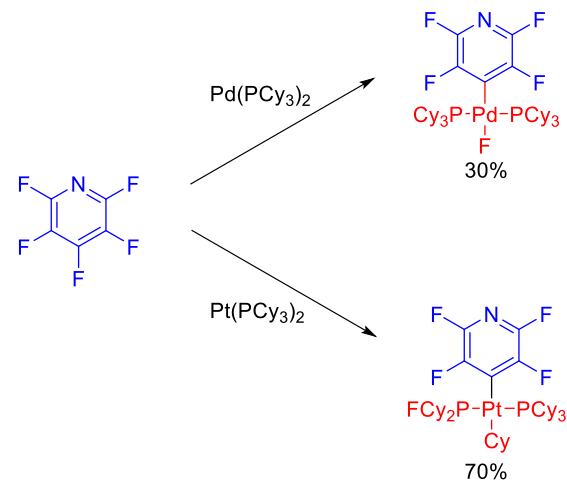
Perutz, 1997:



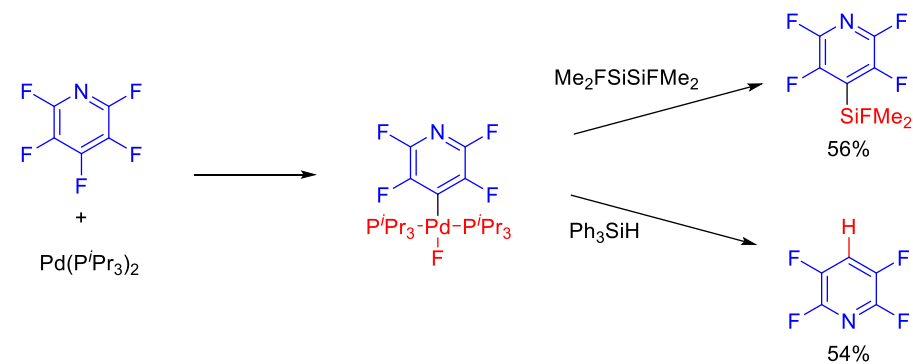
Perutz, 1999:



Perutz, 2004:

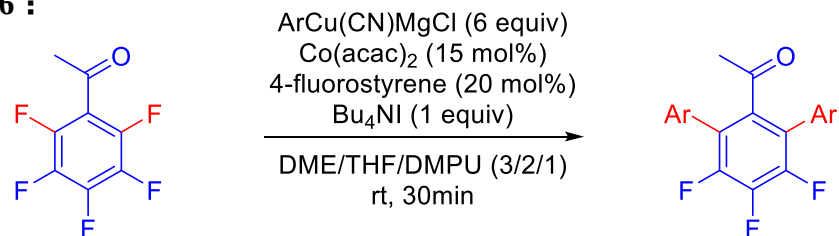


Braun, 2006:

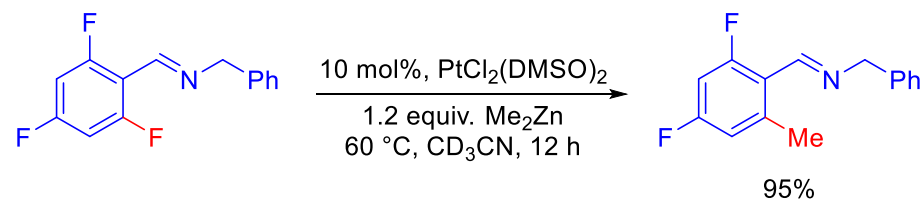


## 2.1、过渡金属参与的脱氟反应

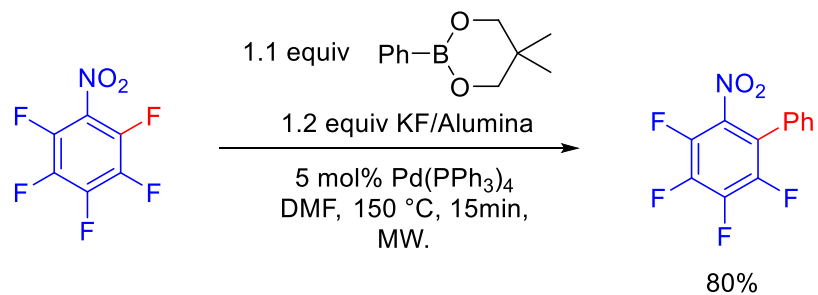
**Knochel, 2006 :**



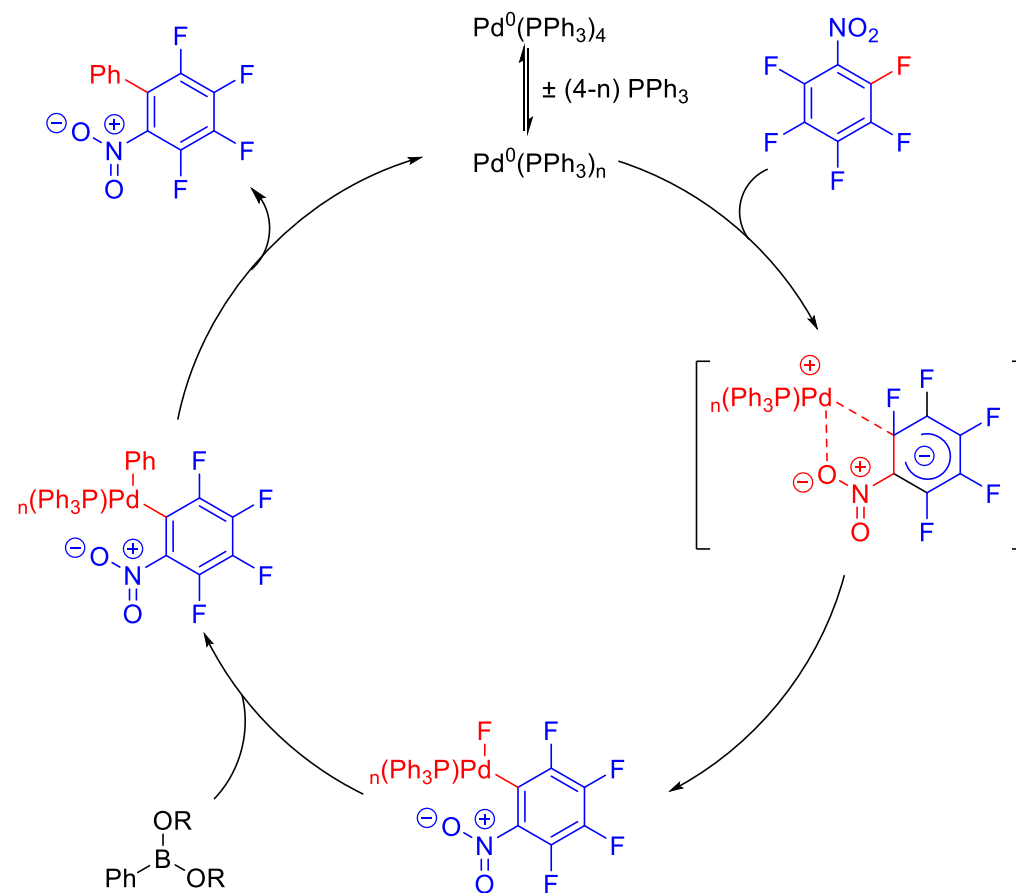
**Love, 2010 :**



**Nelles, 2010 :**

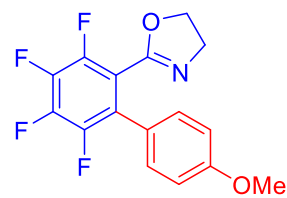
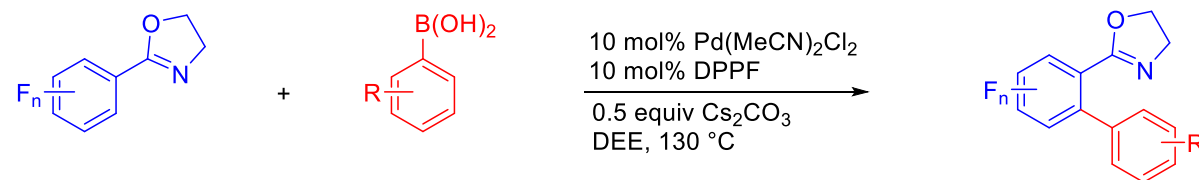


**Nelles, 2010 :**

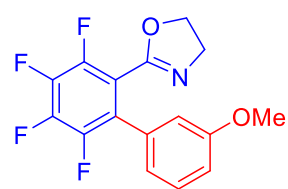


## 2.1、过渡金属参与的脱氟反应

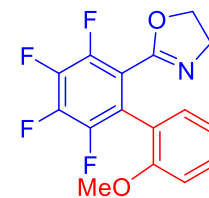
Lu, 2012:



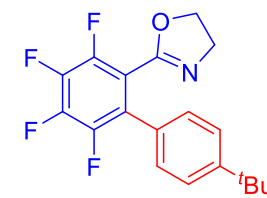
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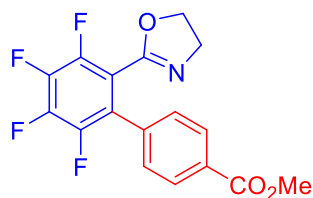
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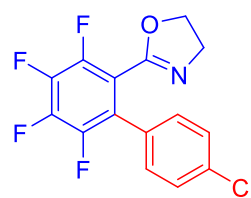
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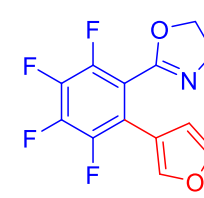
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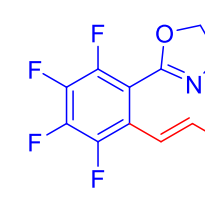
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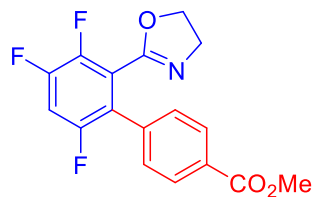
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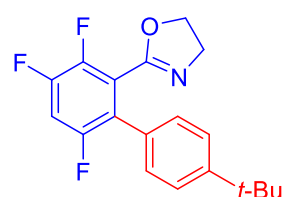
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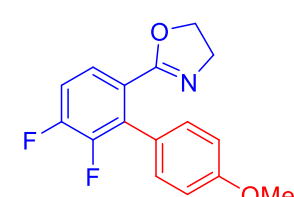
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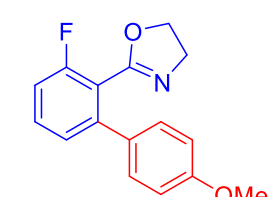
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56%



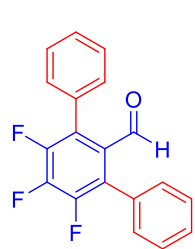
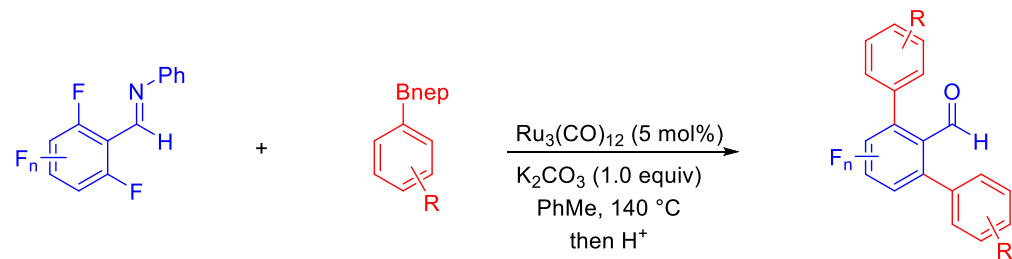
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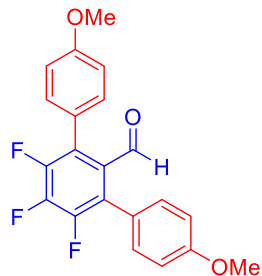
No Reaction

## 2.1、过渡金属参与的脱氟反应

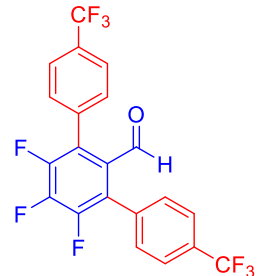
Zhang, 2022:



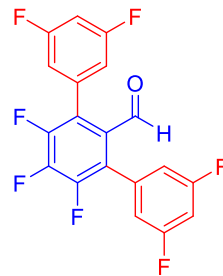
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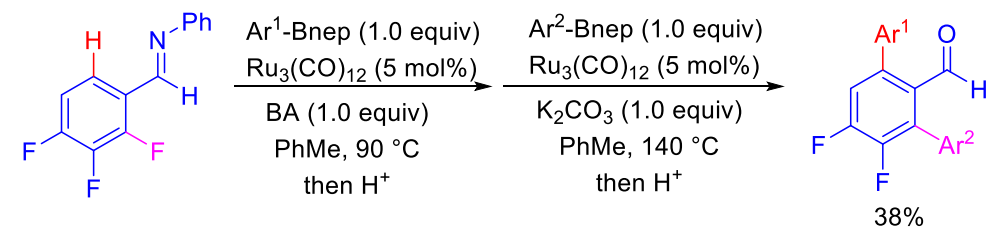
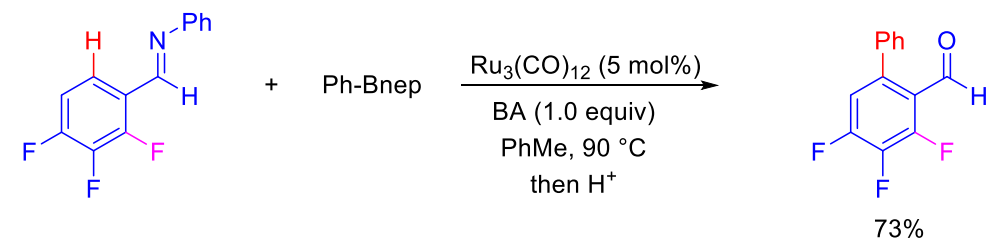
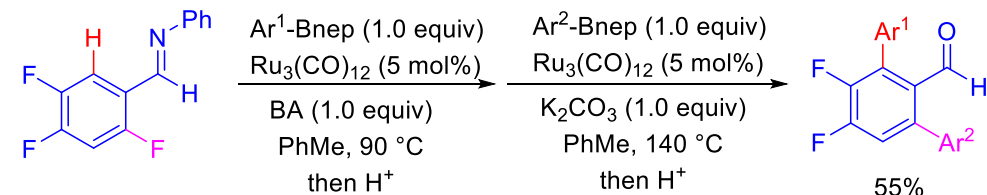
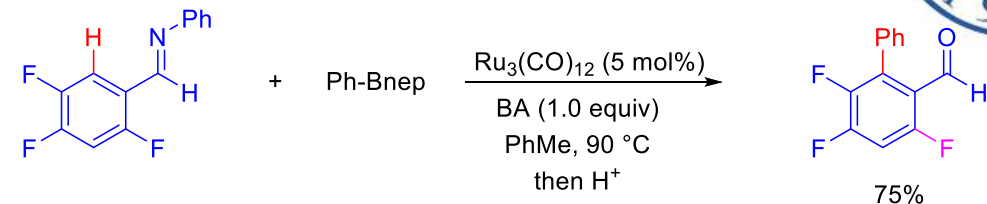
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84%



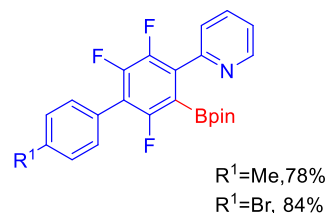
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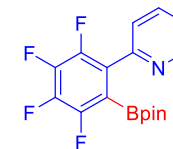
BA= Benzylideneacetone

$\text{Ar}^1 = \text{C}_6\text{H}_5$

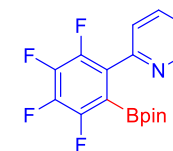
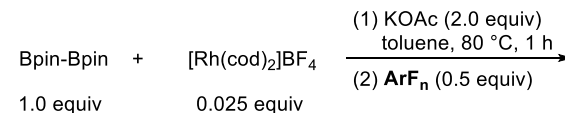
$\text{Ar}^2 = 4\text{-MeO-C}_6\text{H}_4$



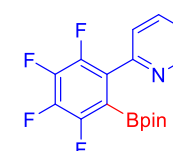
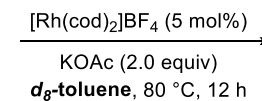
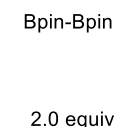
$\text{Bpin-Bpin} + [\text{Rh}(\text{cod})_2]\text{BF}_4 \xrightarrow[\text{(2) ArF}_n \text{ (0.5 equiv)}]{\text{(1) KOAc (2.0 equiv), toluene, 80 }^\circ\text{C, 1 h}}$ 

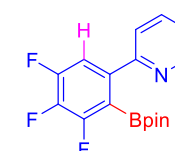
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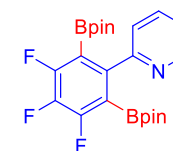
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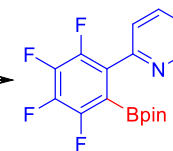
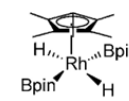
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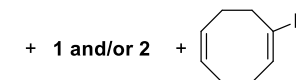
11%



23%

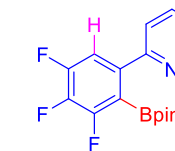


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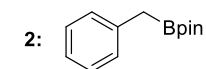
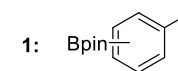
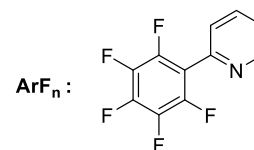


detected by  
GC-MS

detected by  
 $^2\text{D}$  NMR

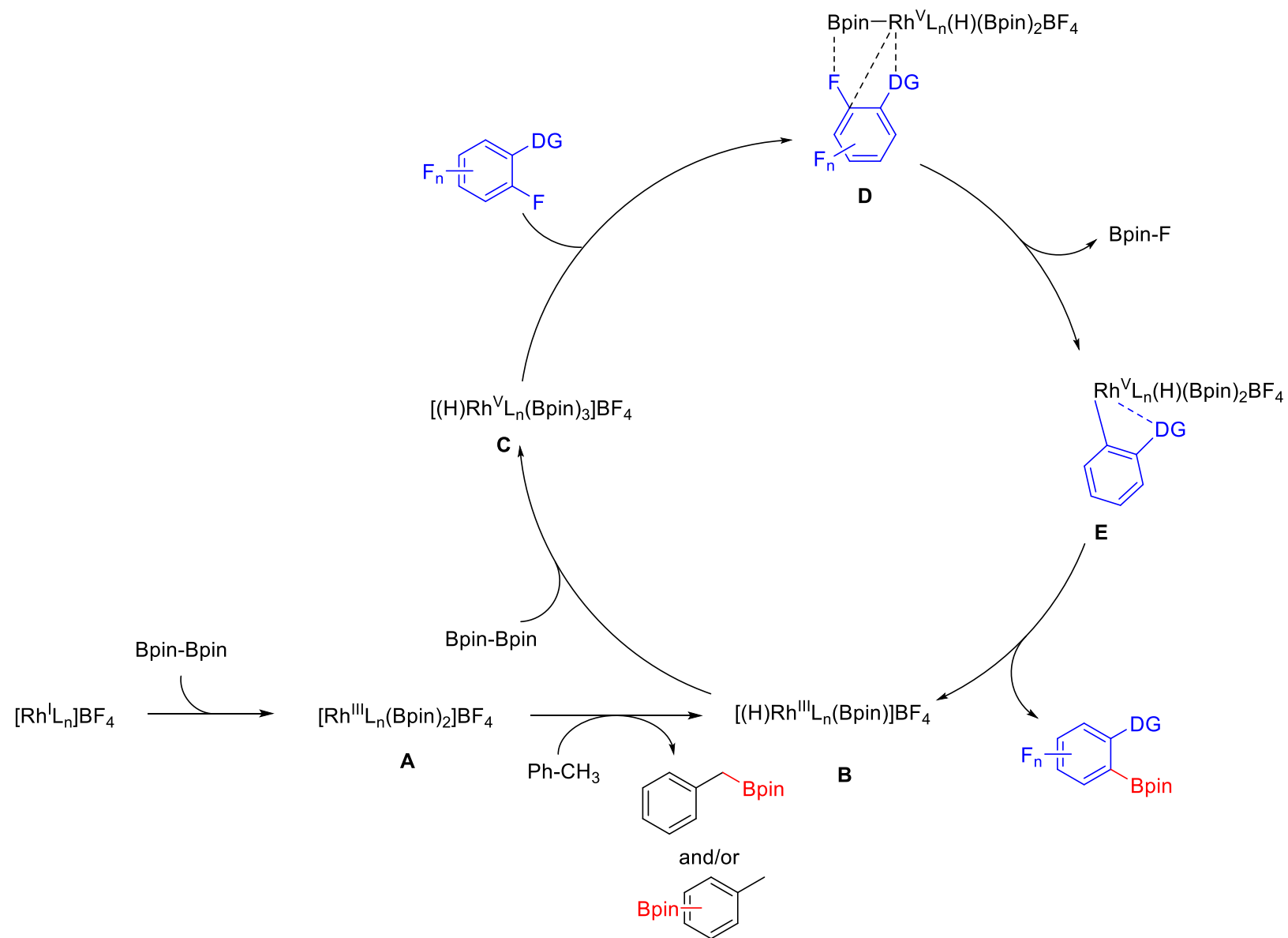


0%



## 2.1、过渡金属参与的脱氟反应

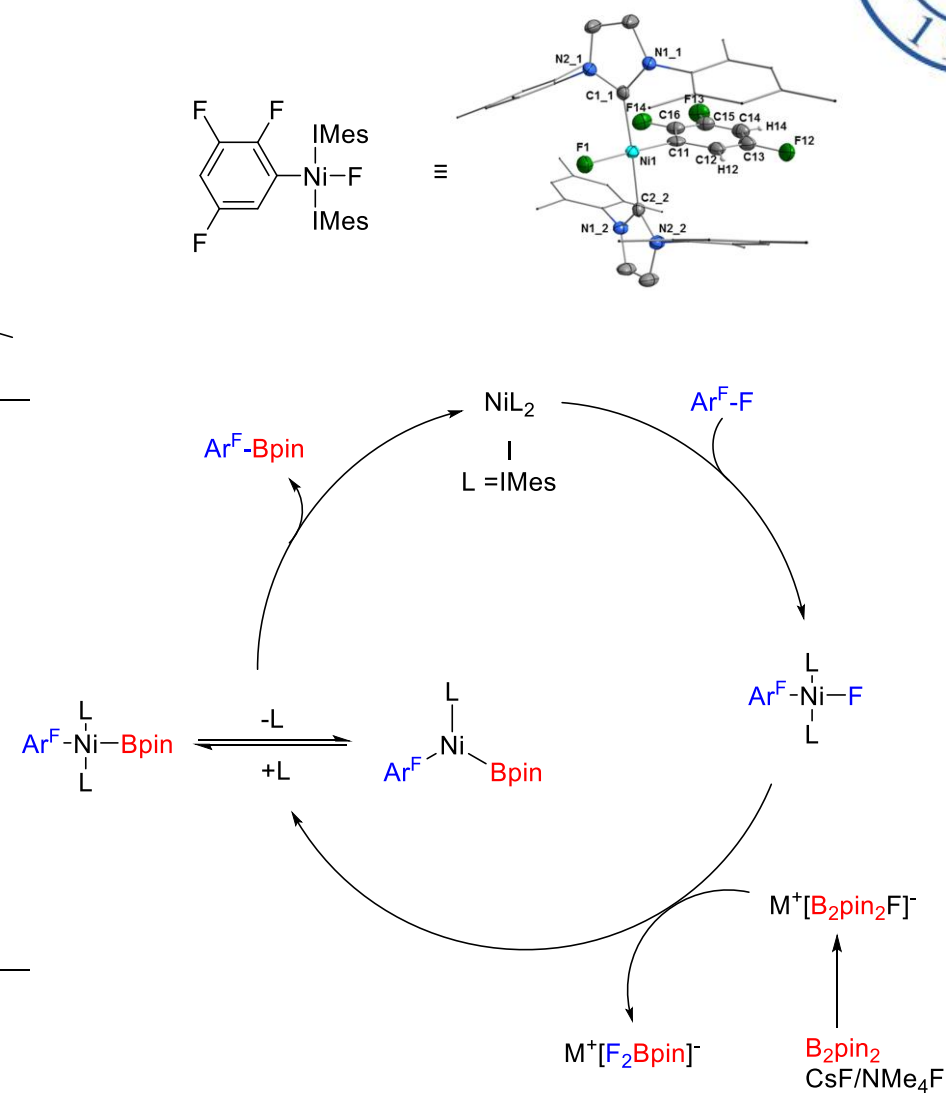
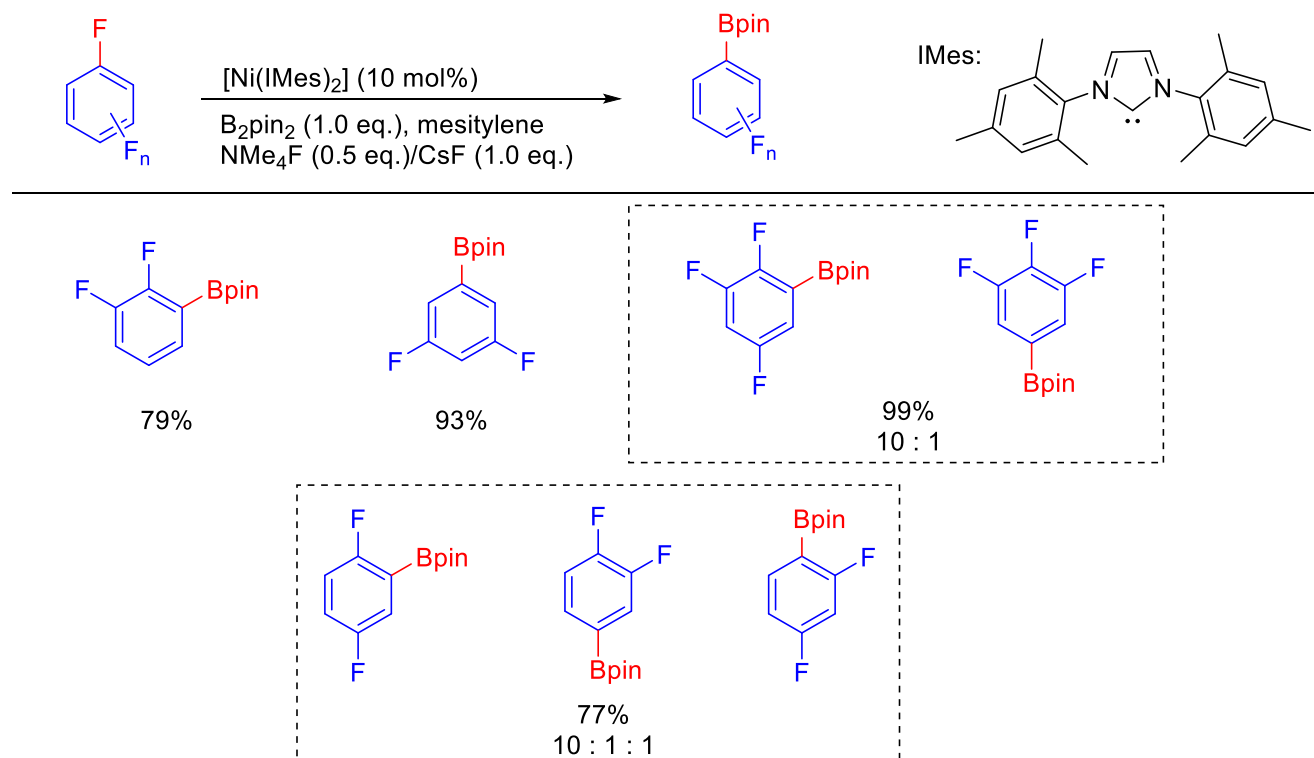
Zhang, 2015:



| Configuration | Ligand                                                                    | $\Delta E$ (kJ/mol) |
|---------------|---------------------------------------------------------------------------|---------------------|
| trans         | $[\text{Ni}(\text{Im})_2(\text{F})(4-(\text{CF}_3)\text{C}_6\text{F}_4)]$ | 0.0                 |
| trans         | $[\text{Ni}(\text{Im})_2(\text{F})(3-(\text{CF}_3)\text{C}_6\text{F}_4)]$ | 2.1                 |
| trans         | $[\text{Ni}(\text{Im})_2(\text{F})(2-(\text{CF}_3)\text{C}_6\text{F}_4)]$ | 33.6                |
| cis           | $[\text{Ni}(\text{Im})_2(\text{F})(4-(\text{CF}_3)\text{C}_6\text{F}_4)]$ | 42.2                |
| cis           | $[\text{Ni}(\text{Im})_2(\text{F})(3-(\text{CF}_3)\text{C}_6\text{F}_4)]$ | 43.9                |
| cis           | $[\text{Ni}(\text{Im})_2(\text{F})(2-(\text{CF}_3)\text{C}_6\text{F}_4)]$ | 64.0                |

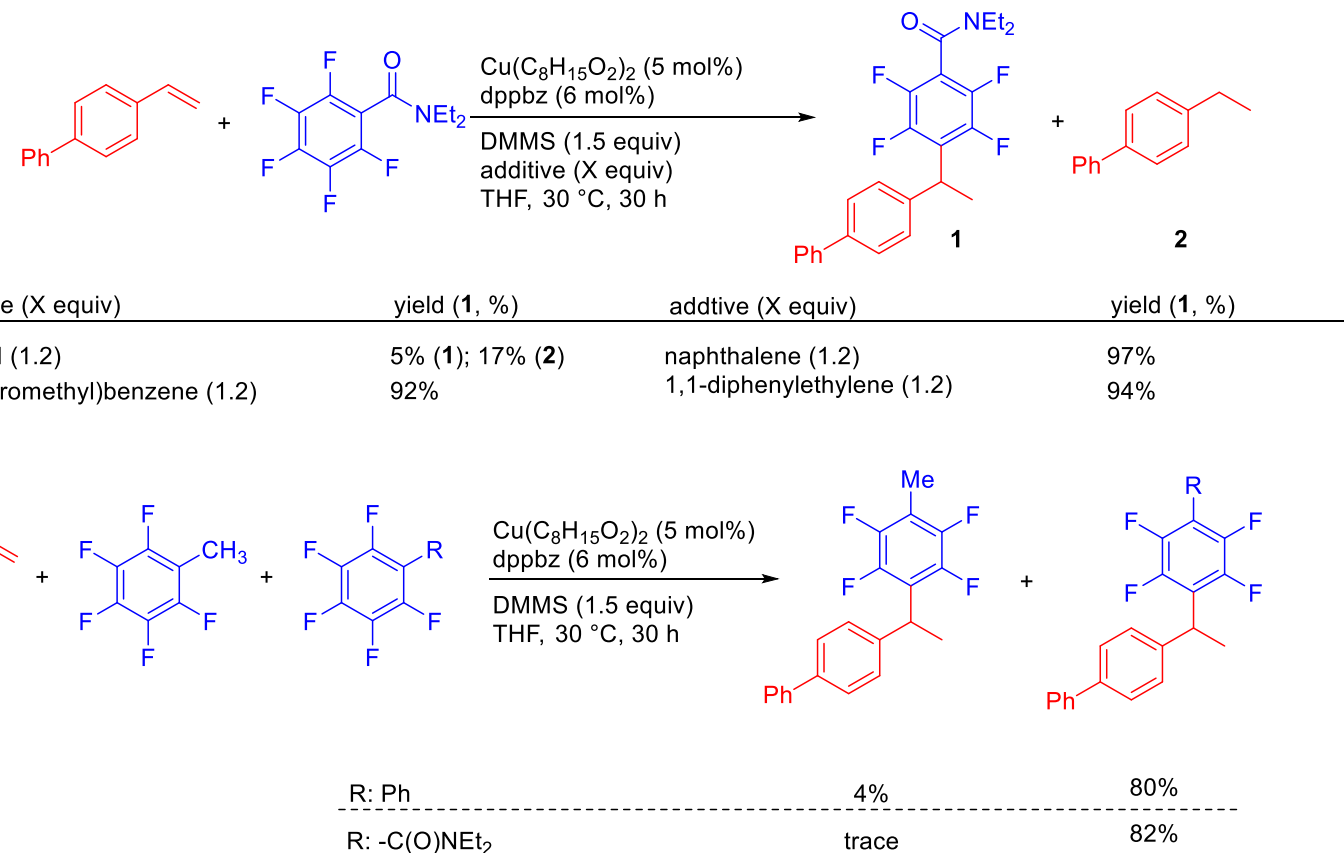
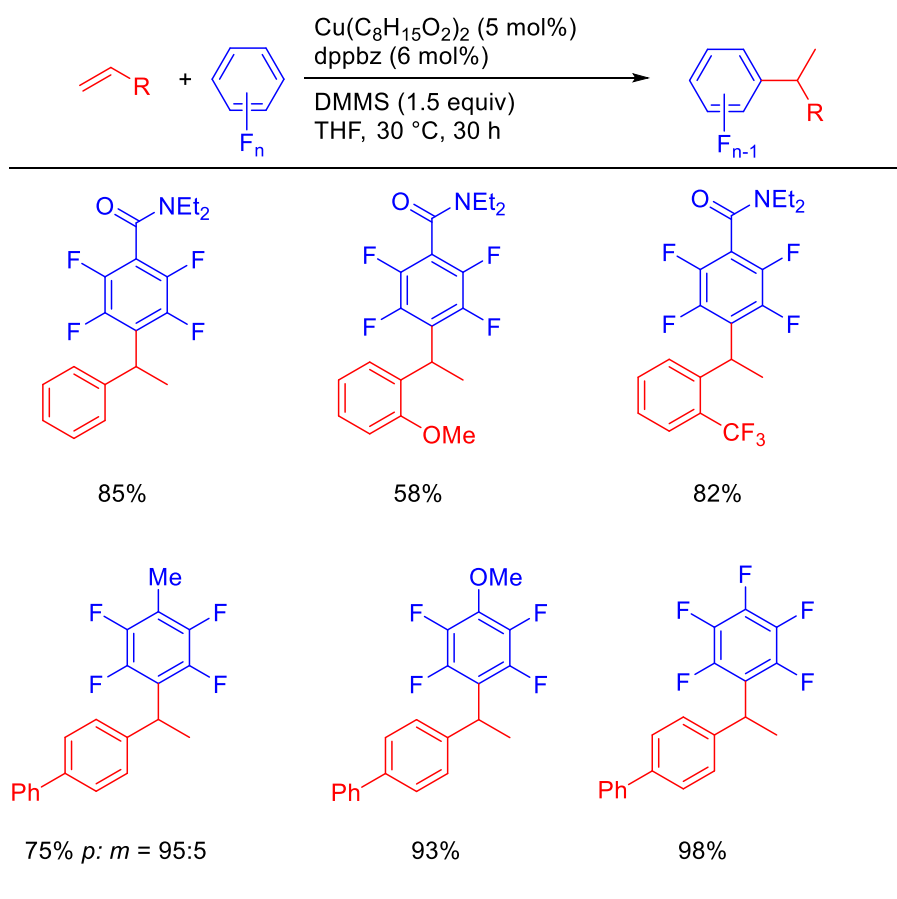
## 2.1、过渡金属参与的脱氟反应

Radius, 2016:



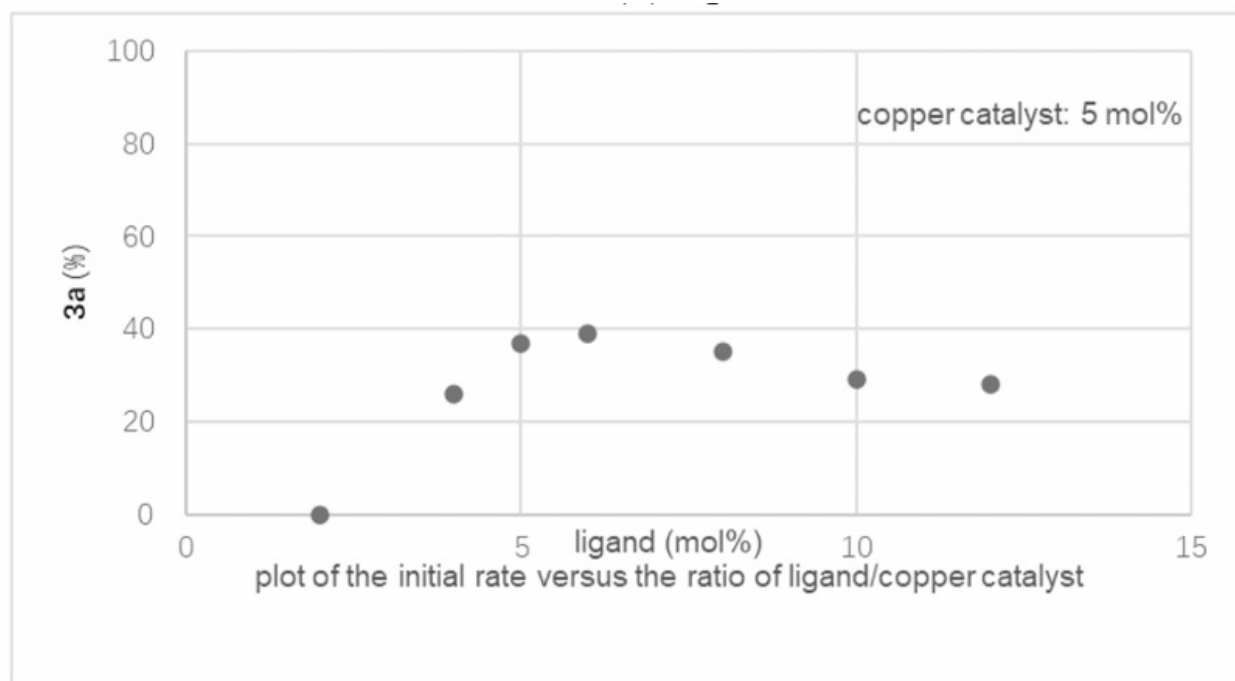
## 2.1、过渡金属参与的脱氟反应

Zhang, 2020 :

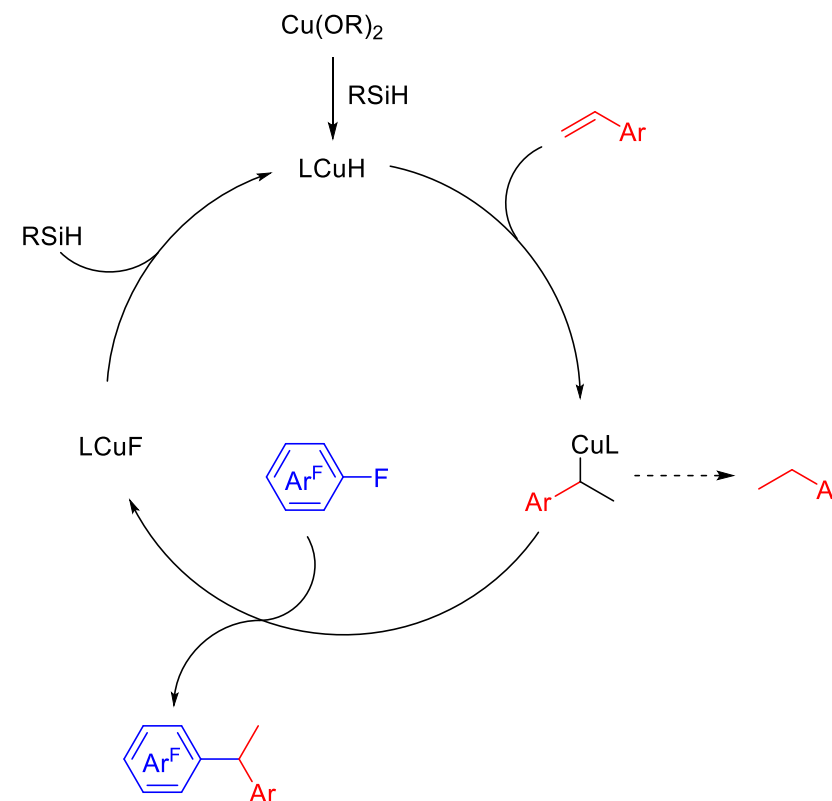


## 2.1、过渡金属参与的脱氟反应

Zhang, 2020 :

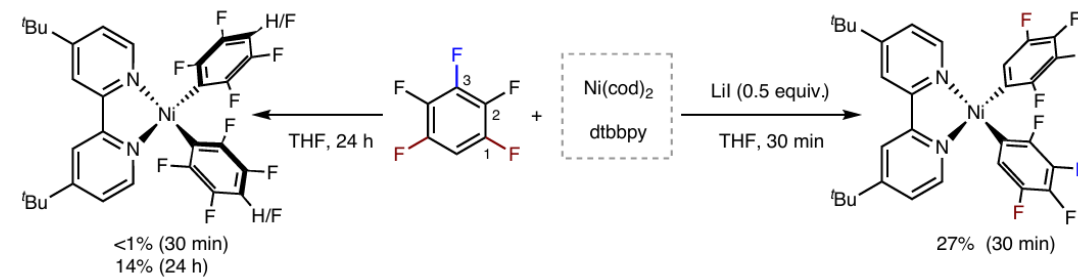
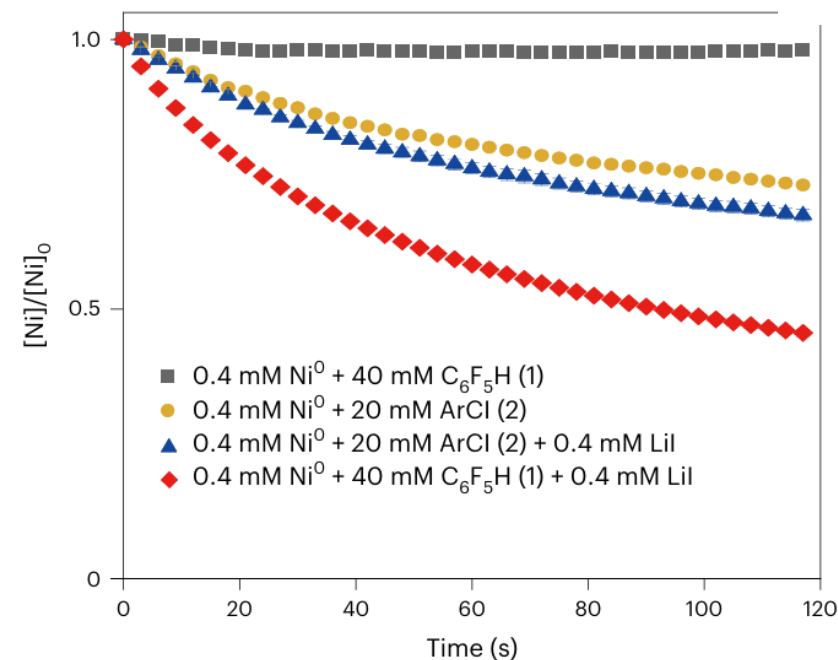
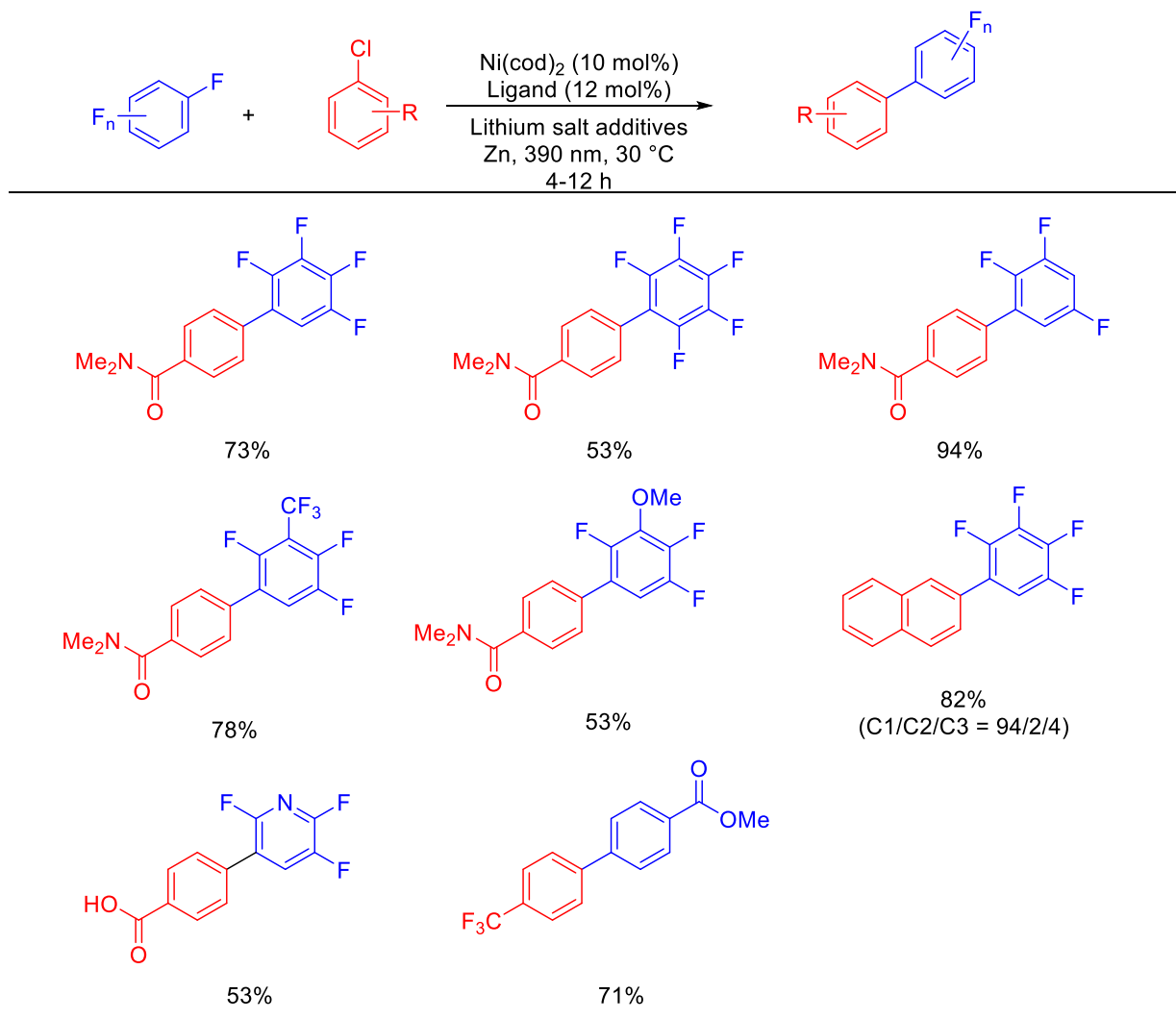


Proposed mechanism :



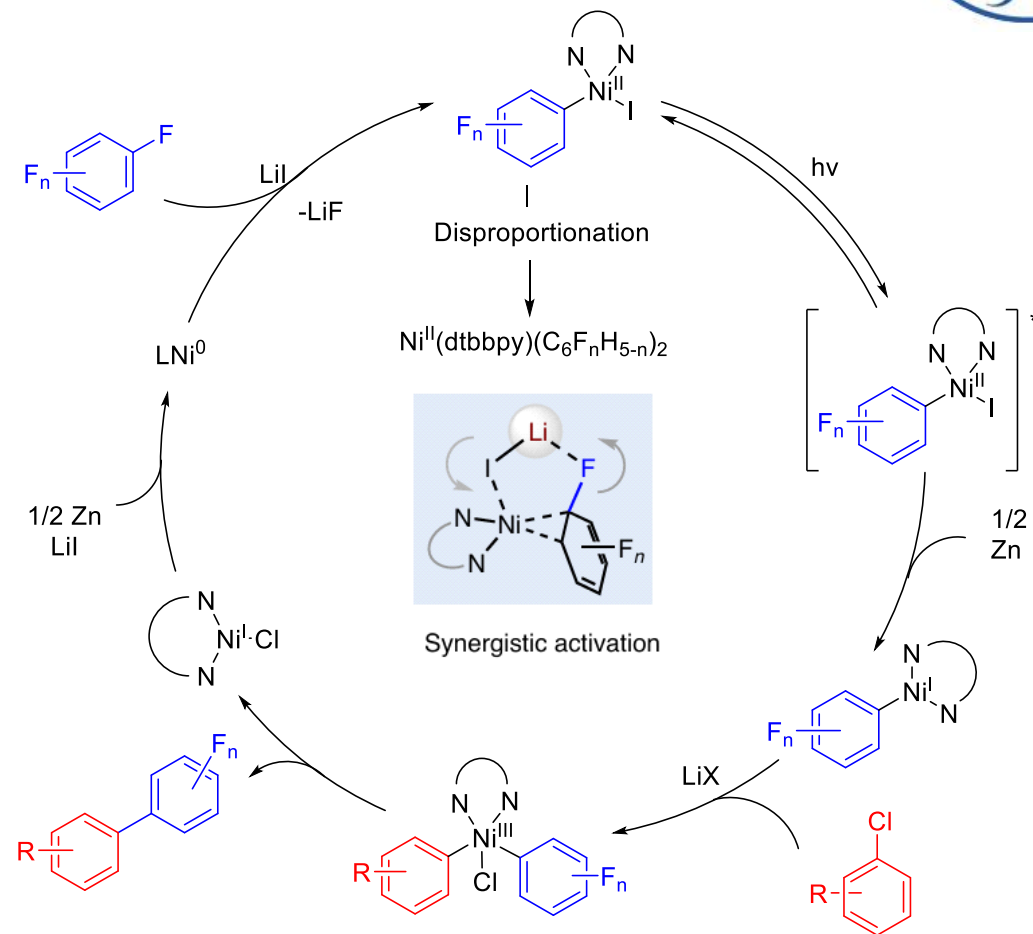
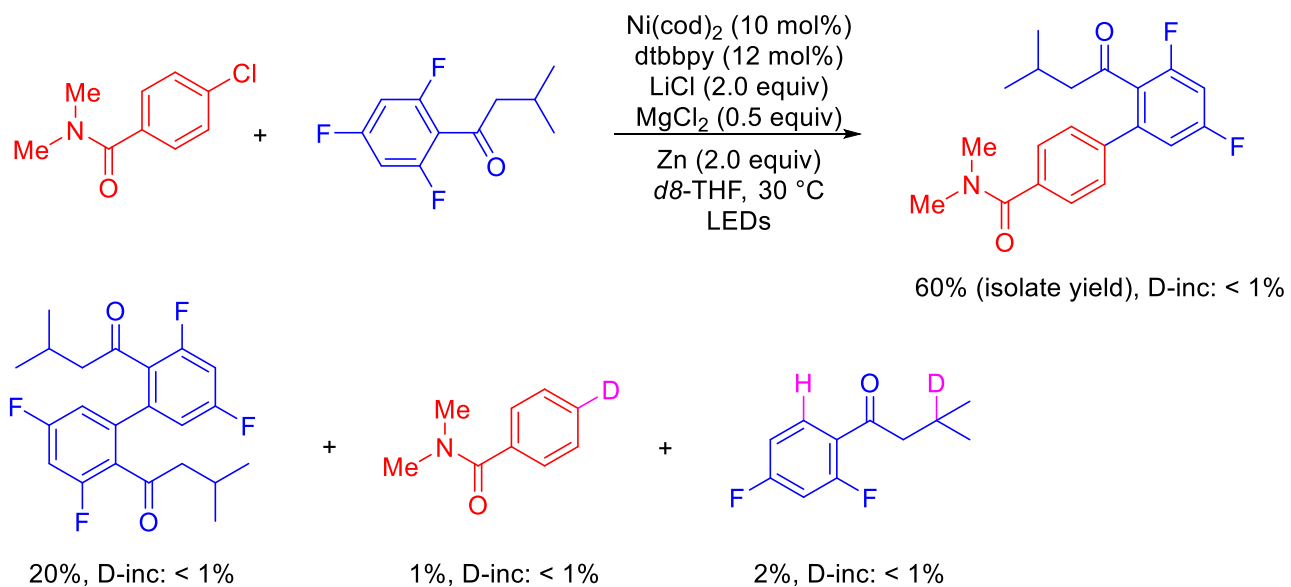
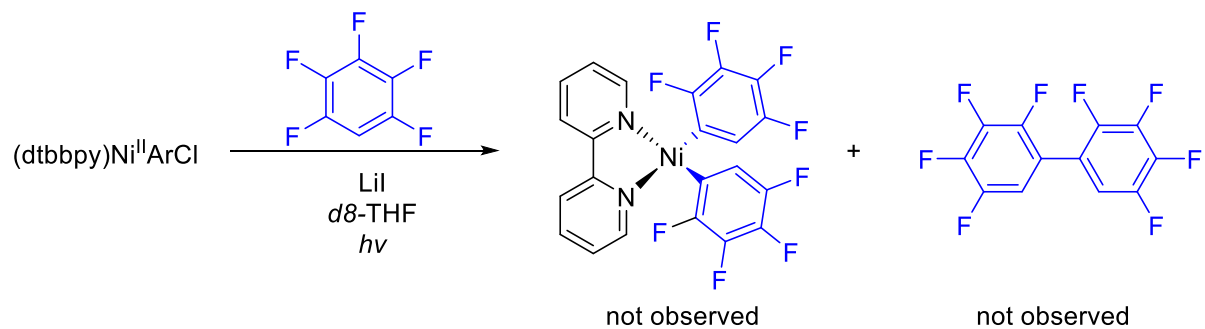
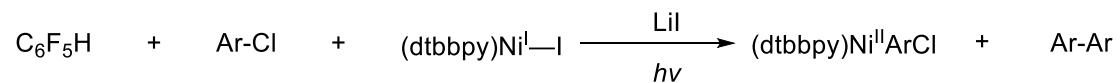
## 2.1、过渡金属参与的脱氟反应

Xie, 2025 :



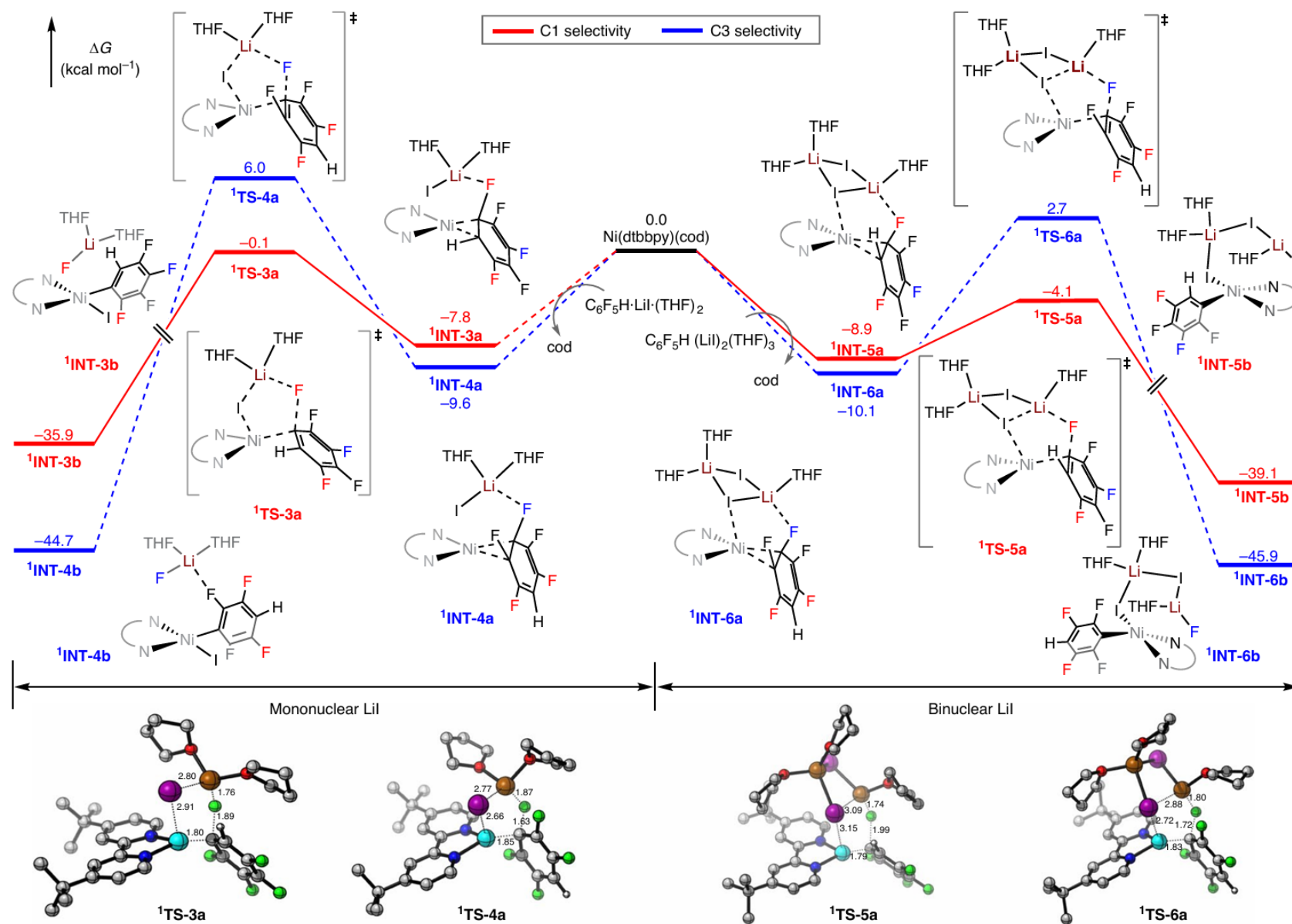
## 2.1、过渡金属参与的脱氟反应

Xie, 2025 :



## 2.1、过渡金属参与的脱氟反应

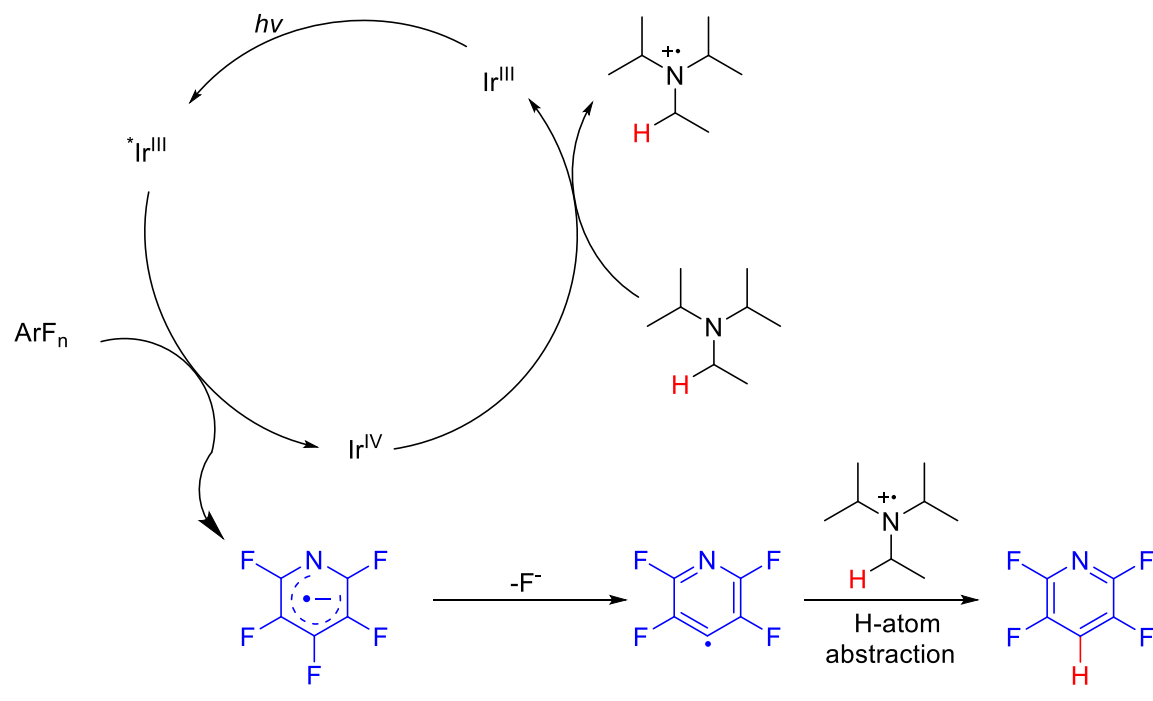
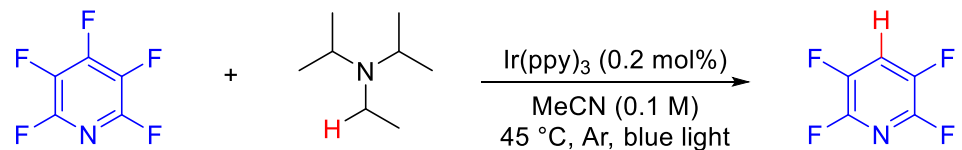
Xie, 2025 :



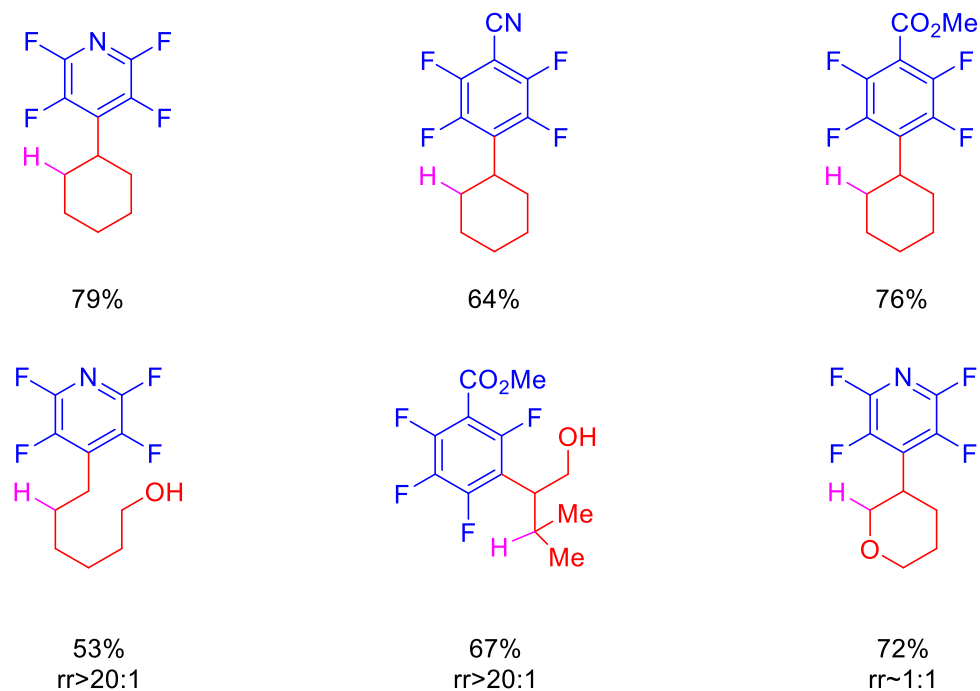
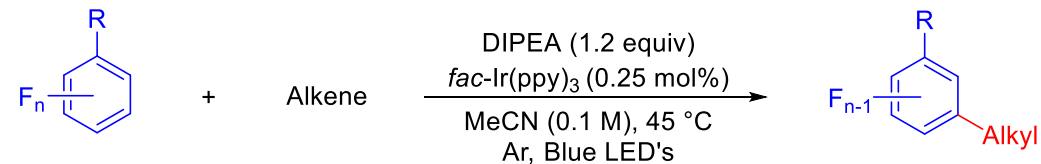
1. 背景介绍
2. 多氟芳烃的脱氟官能团化反应
  - 2.1 过渡金属参与的脱氟官能团化反应
  - 2.2 光催化的脱氟官能团化反应
3. 总结与展望

## 2.2、光催化的脱氟反应

Weaver, 2014 :

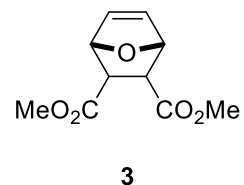
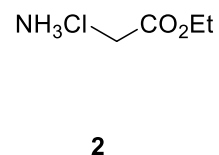
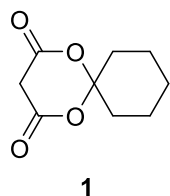
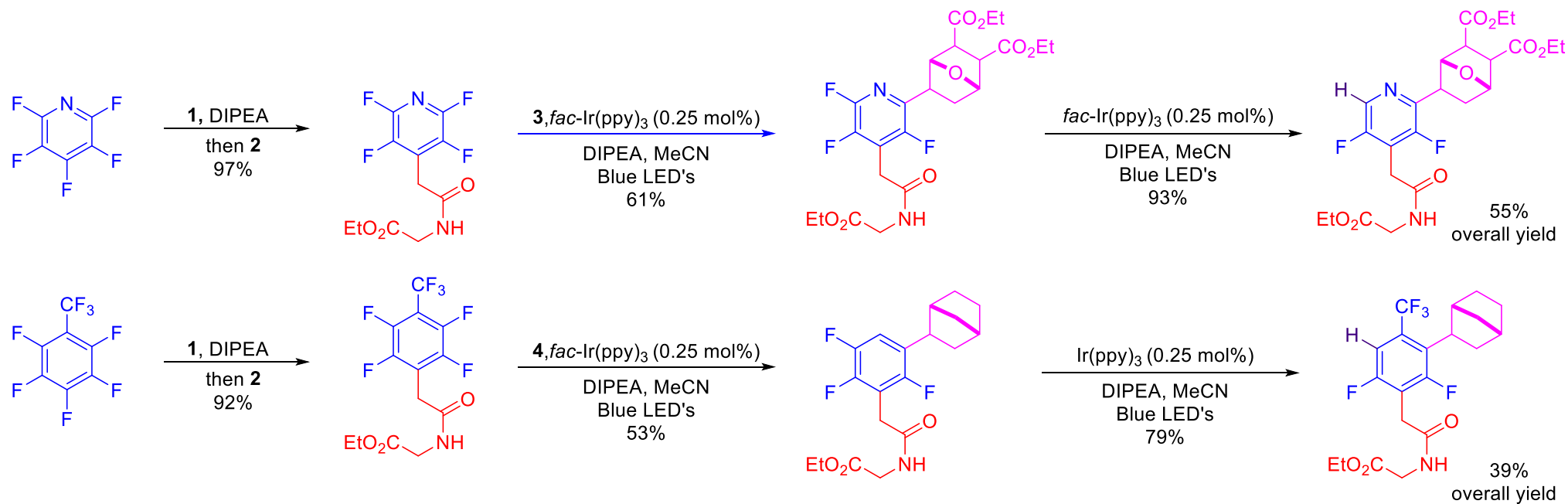
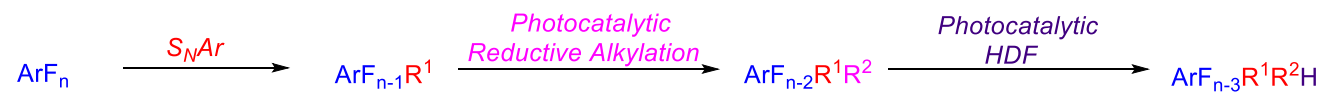


Weaver, 2015 :



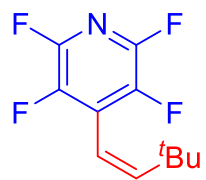
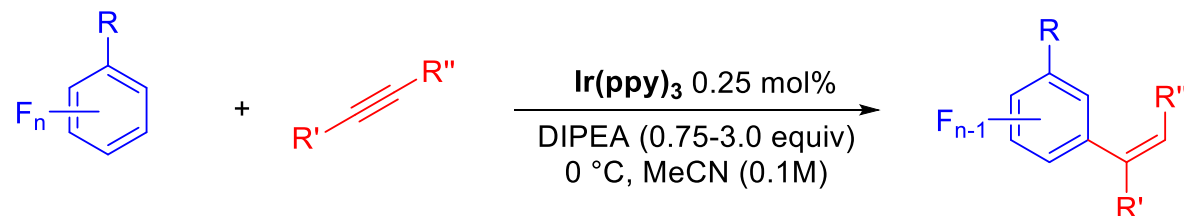
## 2.2、光催化的脱氟反应

Weaver, 2015 :

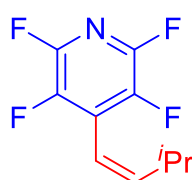


## 2.2、光催化的脱氟反应

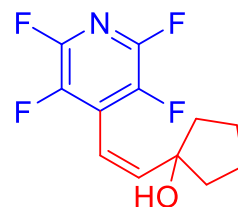
Weaver, 2016 :



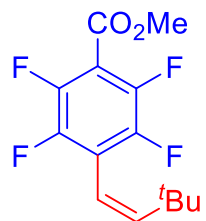
85%, Z:E = 24:1



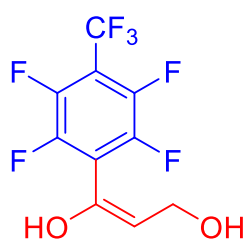
74%, Z:E > 25:1



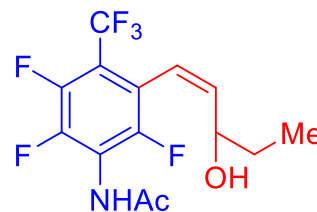
61%, Z:E = 25:1



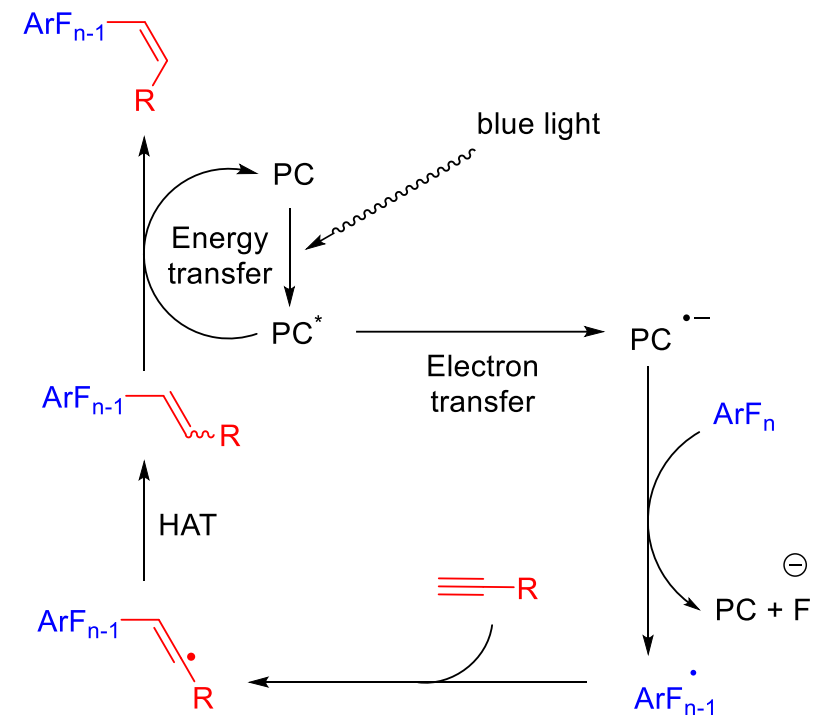
86%, Z:E = 24:1



63%, Z:E > 25:1

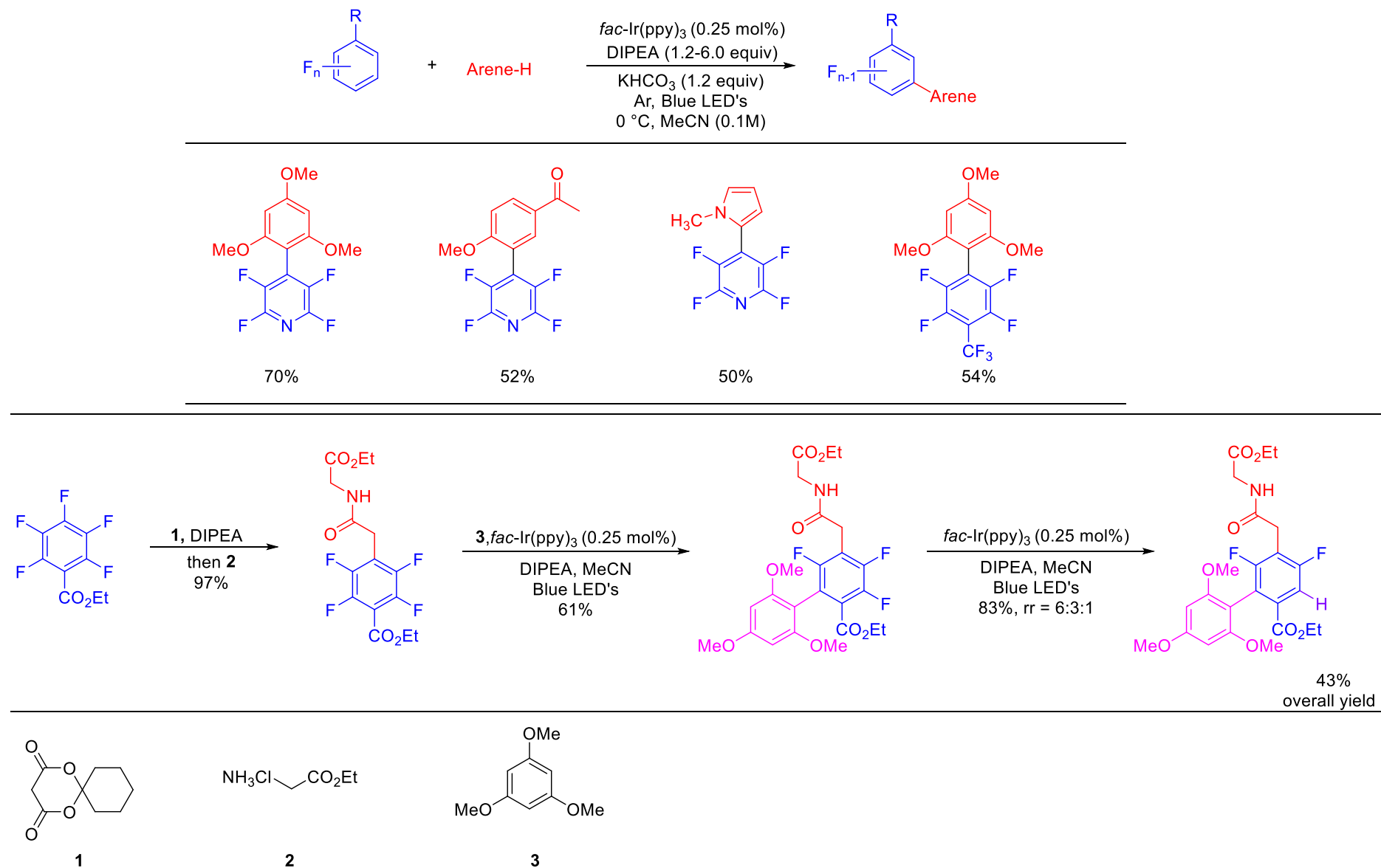


51%, Z:E > 25:1



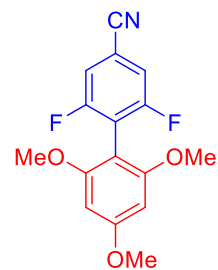
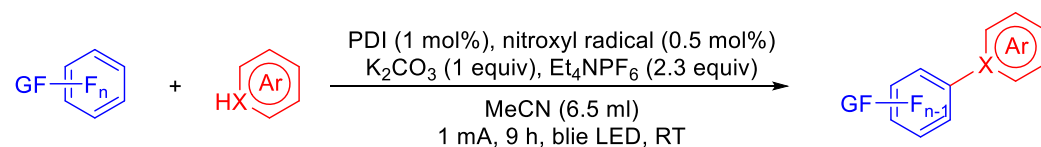
## 2.2、光催化的脱氟反应

Weaver, 2016 :

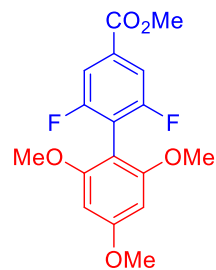


## 2.2、光催化的脱氟反应

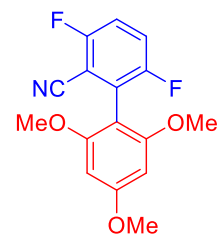
Wu, 2022 :



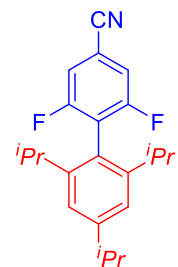
73%



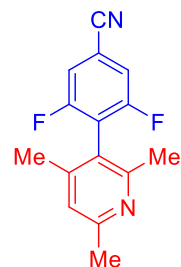
71%



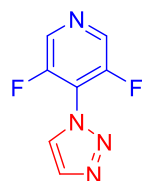
55%



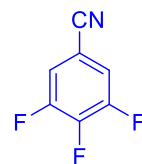
50%



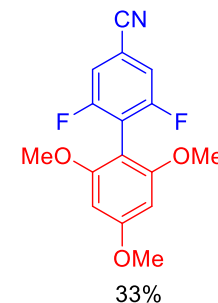
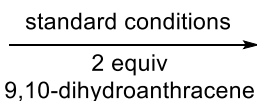
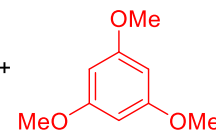
43%



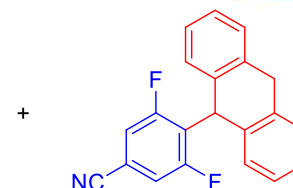
66%



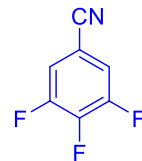
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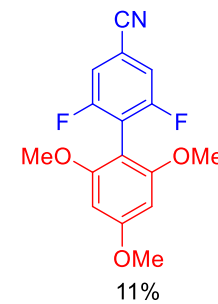
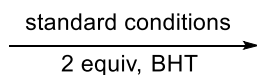
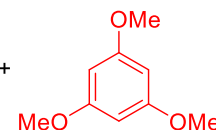
33%



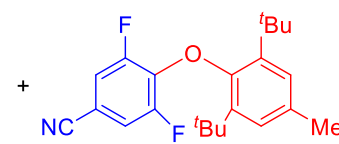
detected by GC-MS



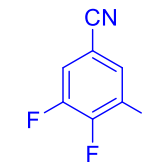
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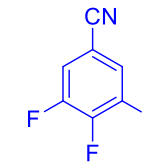
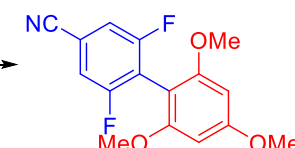
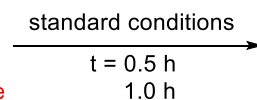
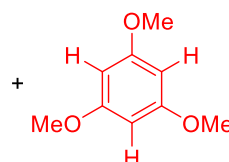
11%



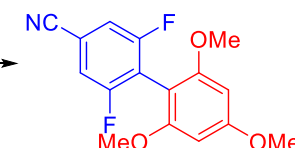
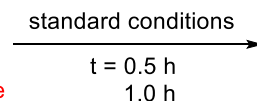
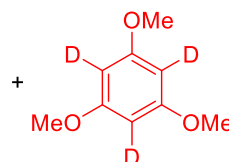
detected by GC-MS



+



+



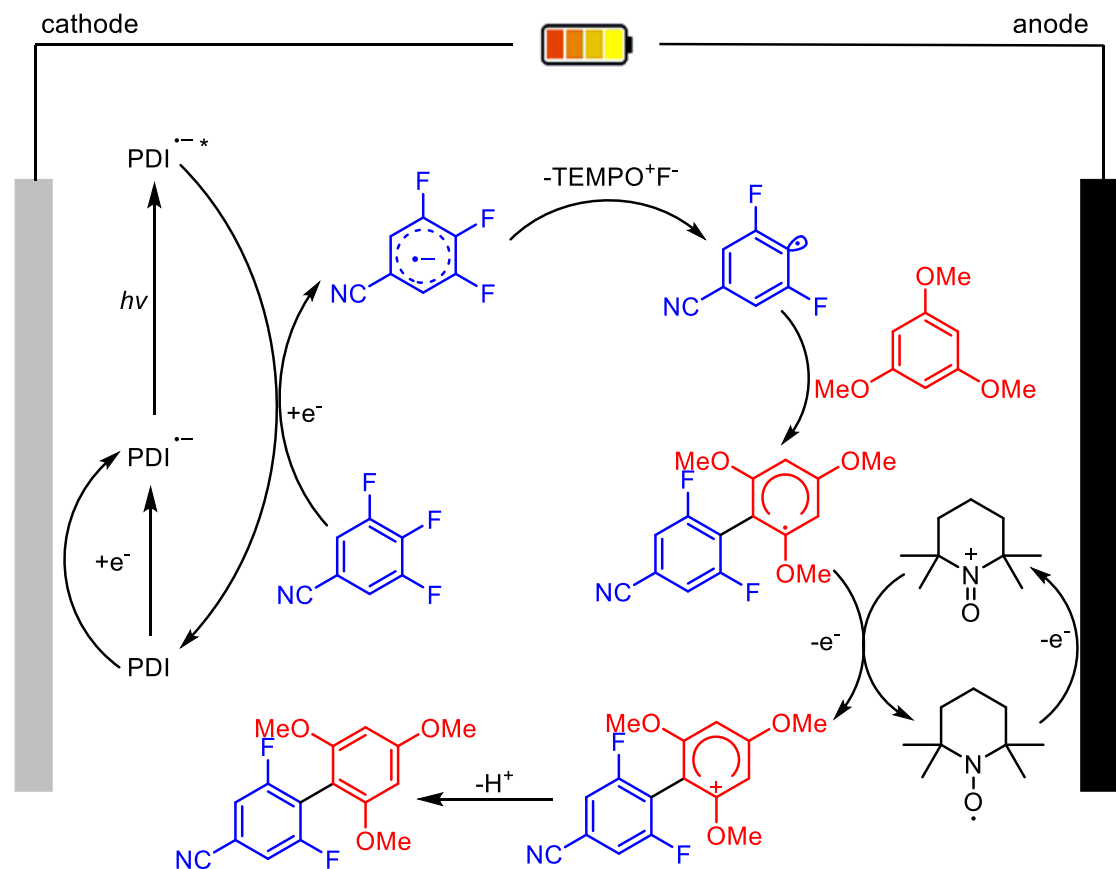
t = 0.5 h  $k_H/k_D = 1.33$

t = 1.0 h  $k_H/k_D = 1.33$

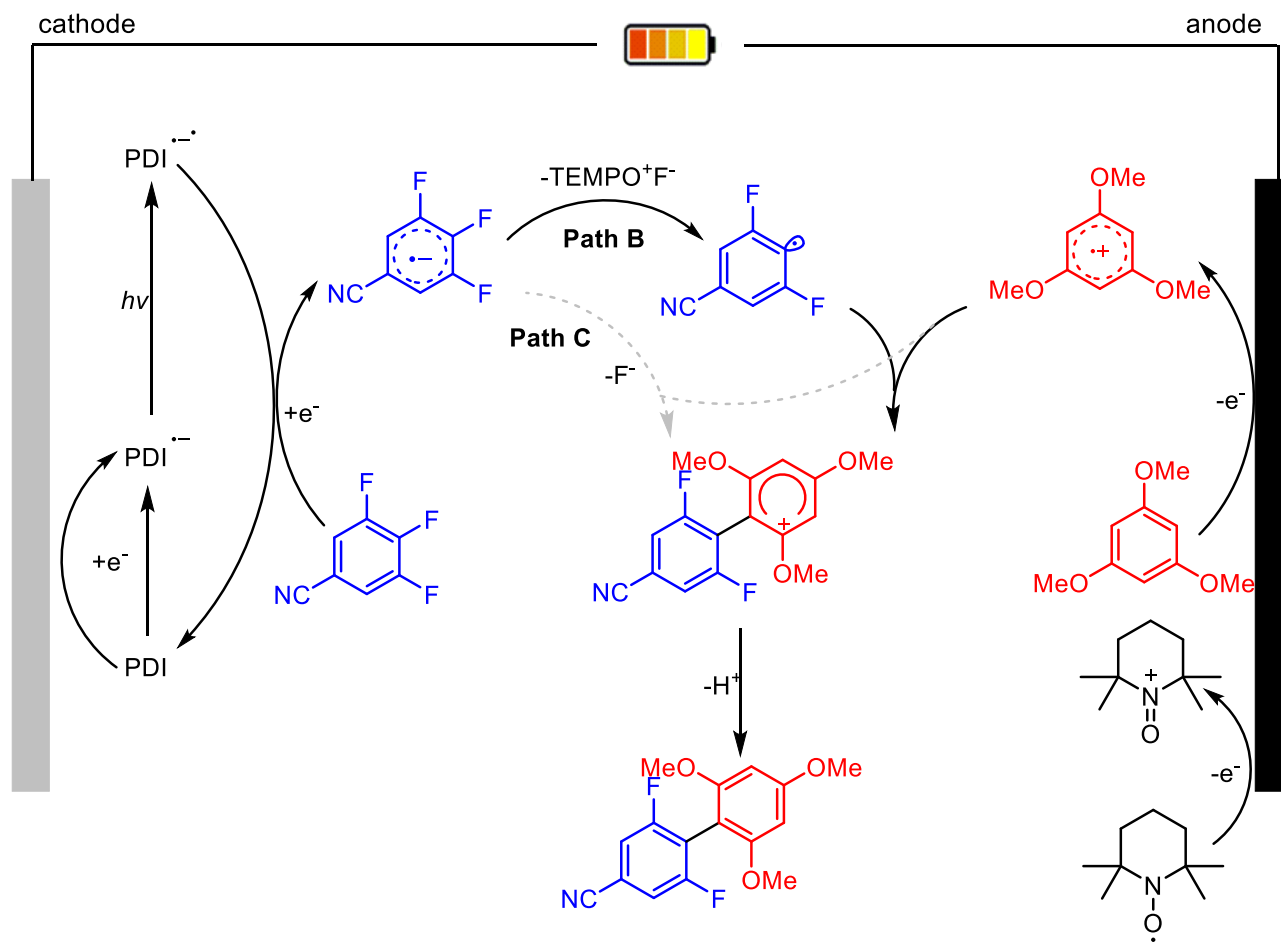
## 2.2、光催化的脱氟反应

Wu, 2022 :

Path A :

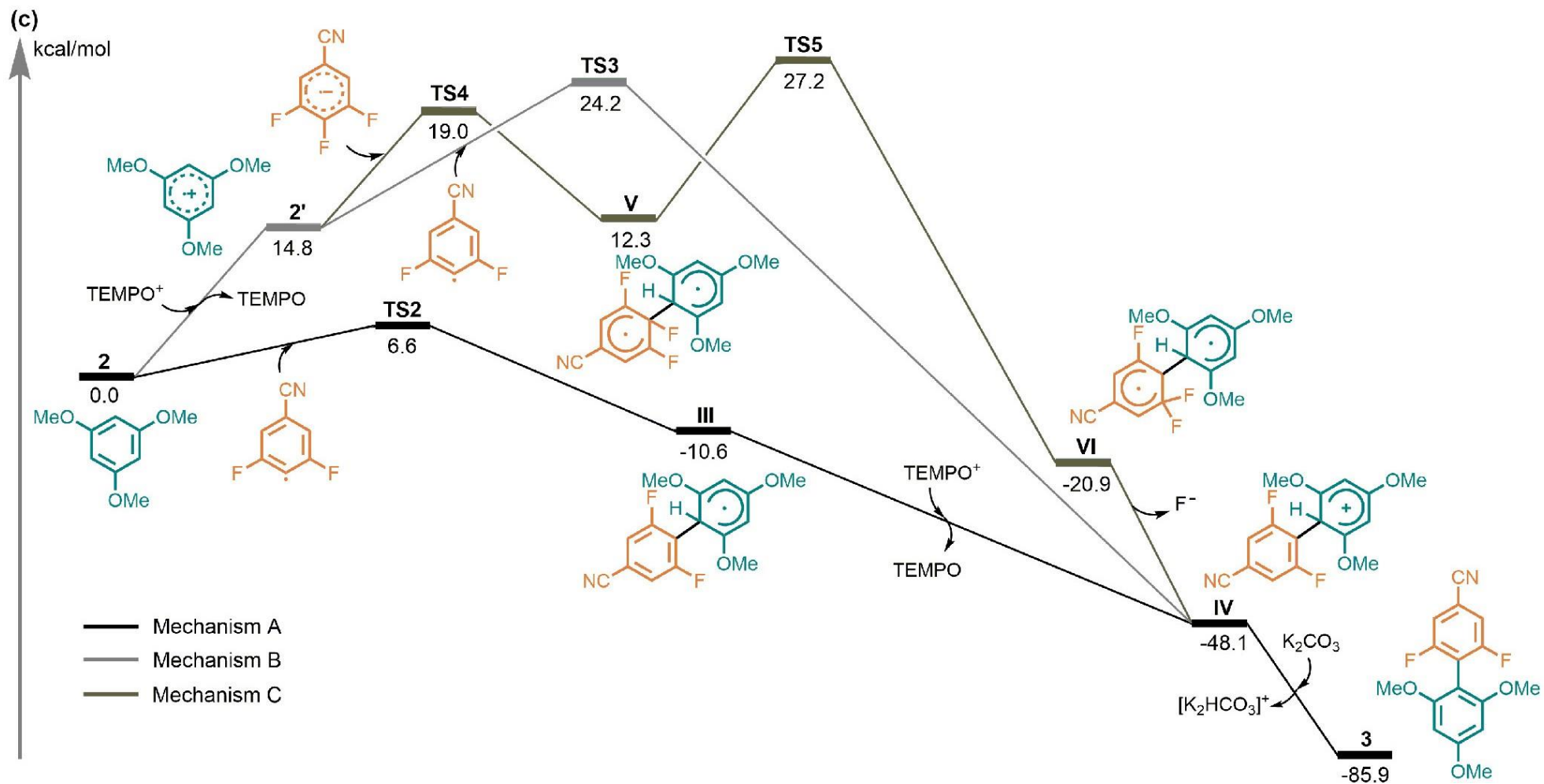


Path B and C :



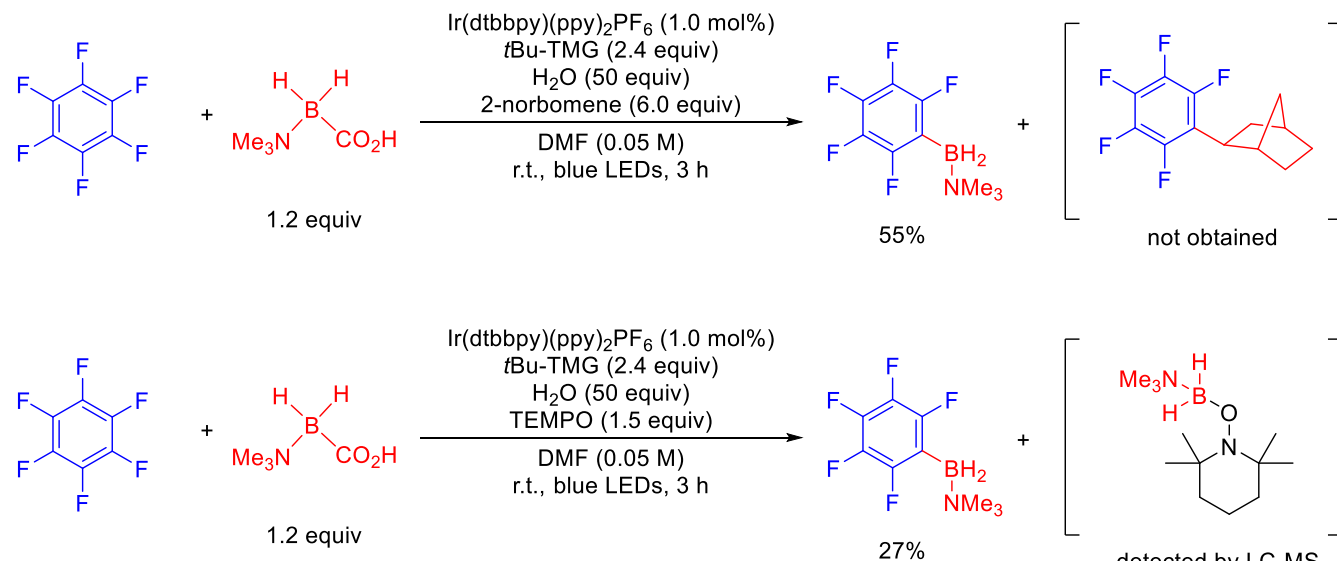
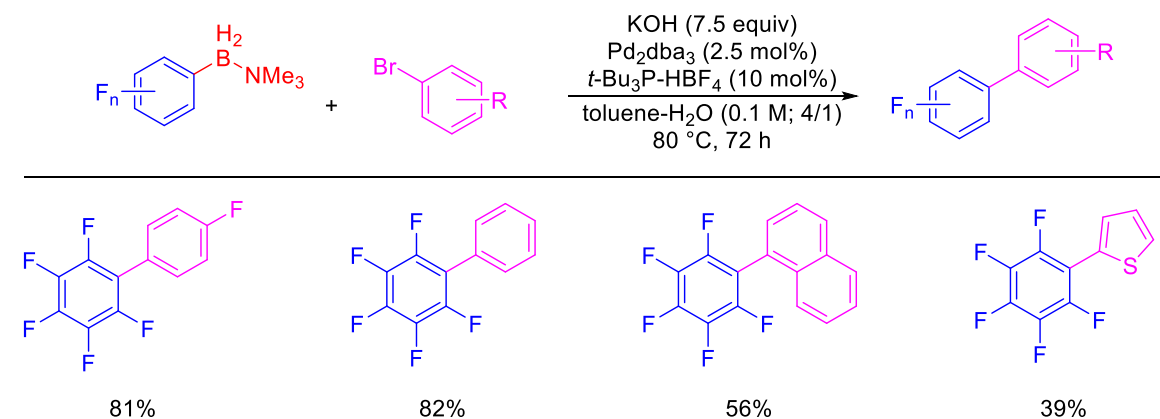
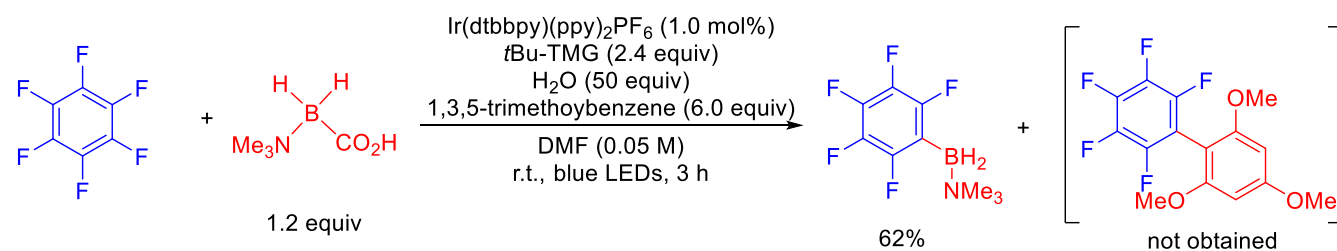
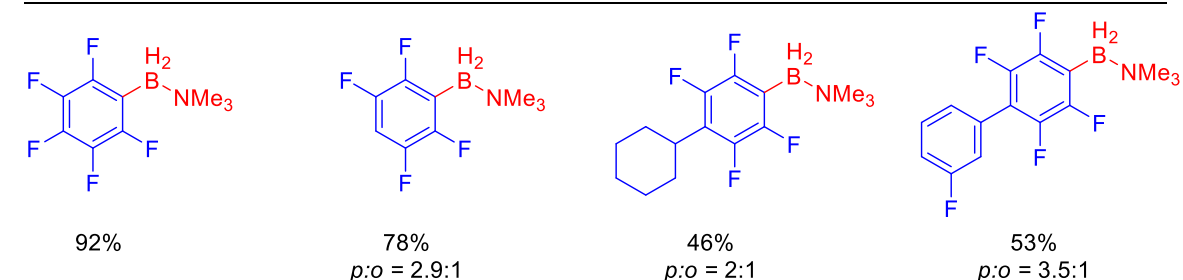
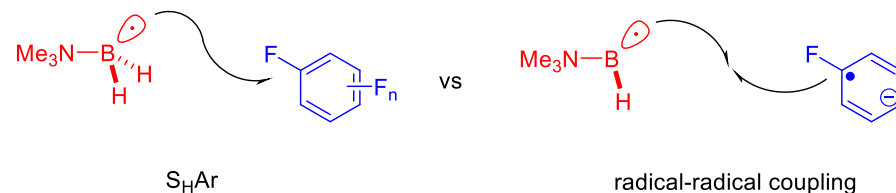
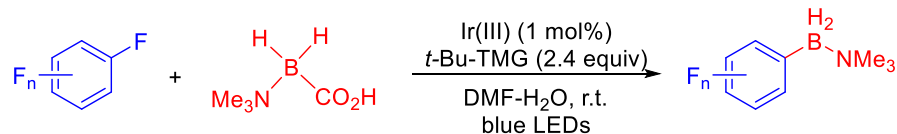
## 2.2、光催化的脱氟反应

Wu, 2022 :



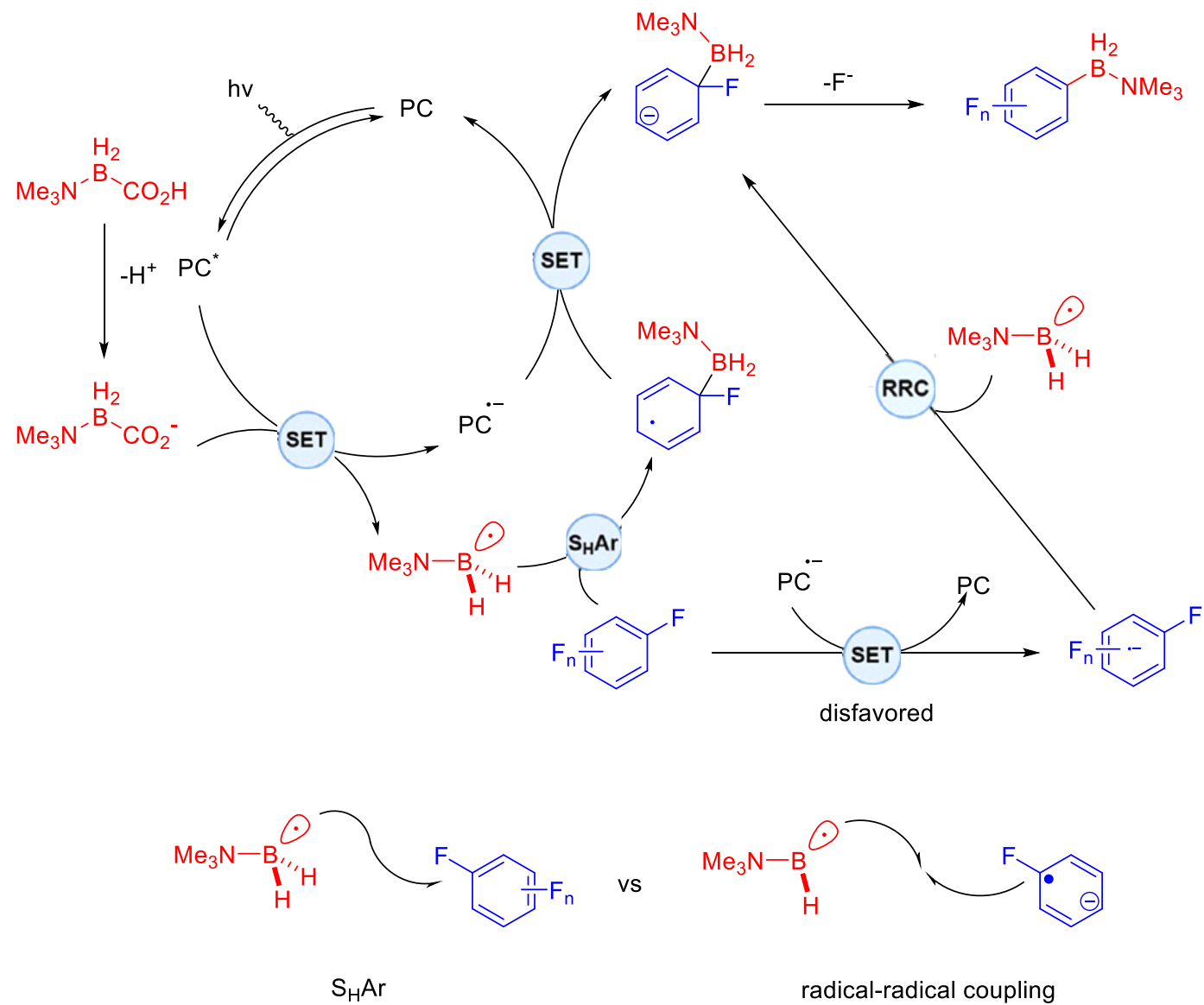
## 2.2、光催化的脱氟反应

Nakamura, 2025:

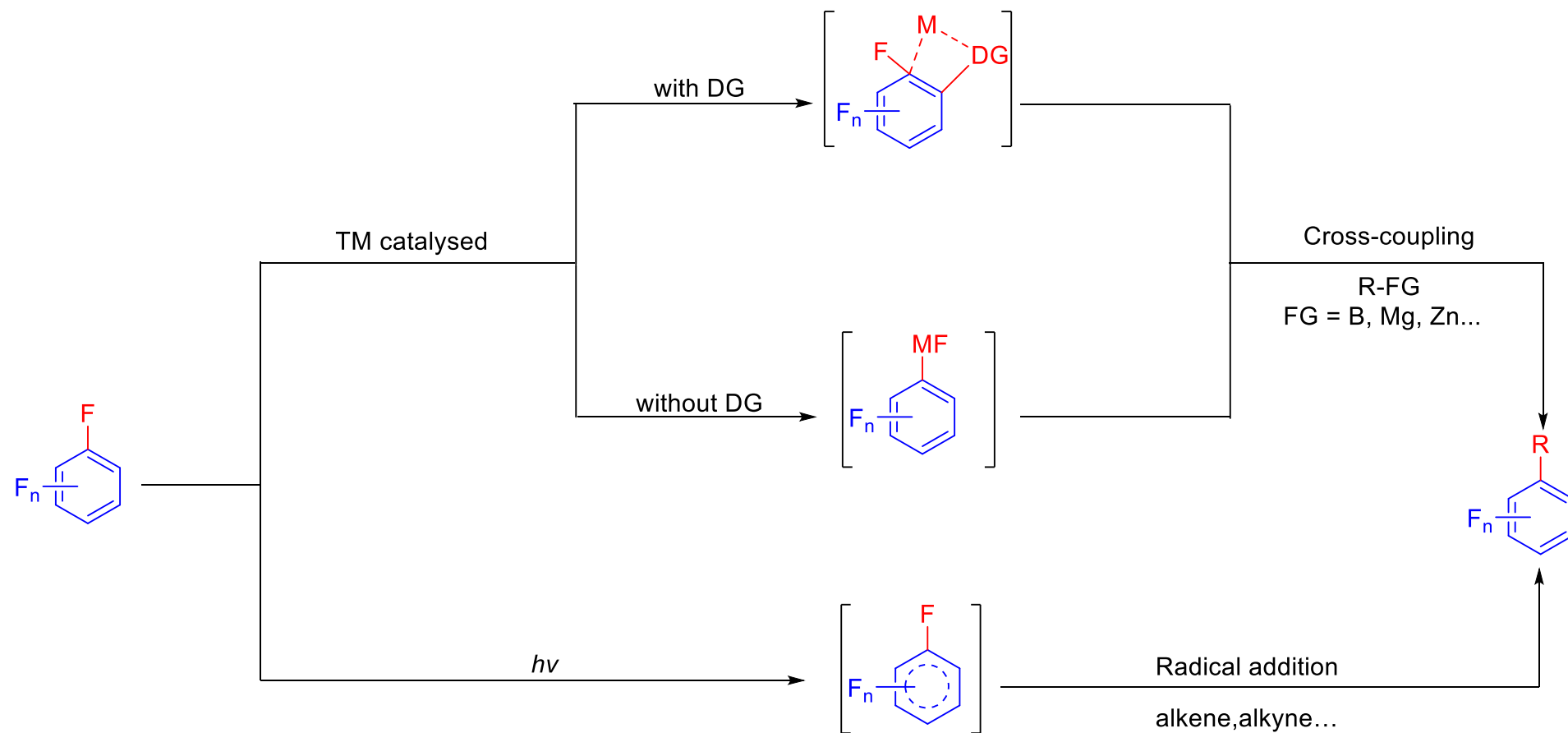


## 2.2、光催化的脱氟反应

Nakamura, 2025:

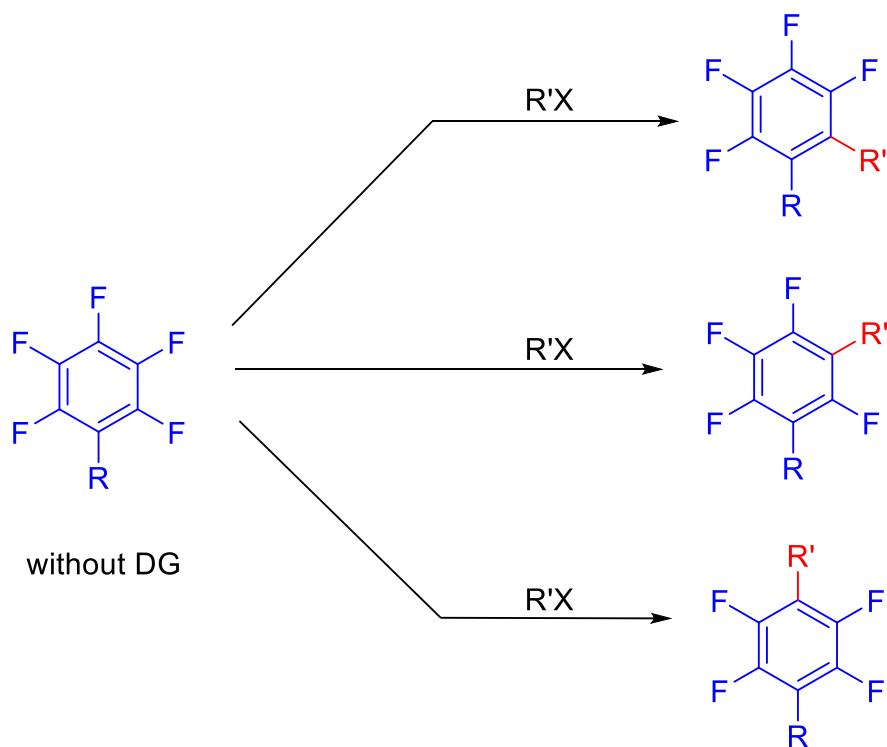


1. 背景介绍
2. 多氟芳烃的脱氟官能团化反应
  - 2.1 过渡金属参与的脱氟官能团化反应
  - 2.2 光催化的脱氟官能团化反应
3. 总结与展望

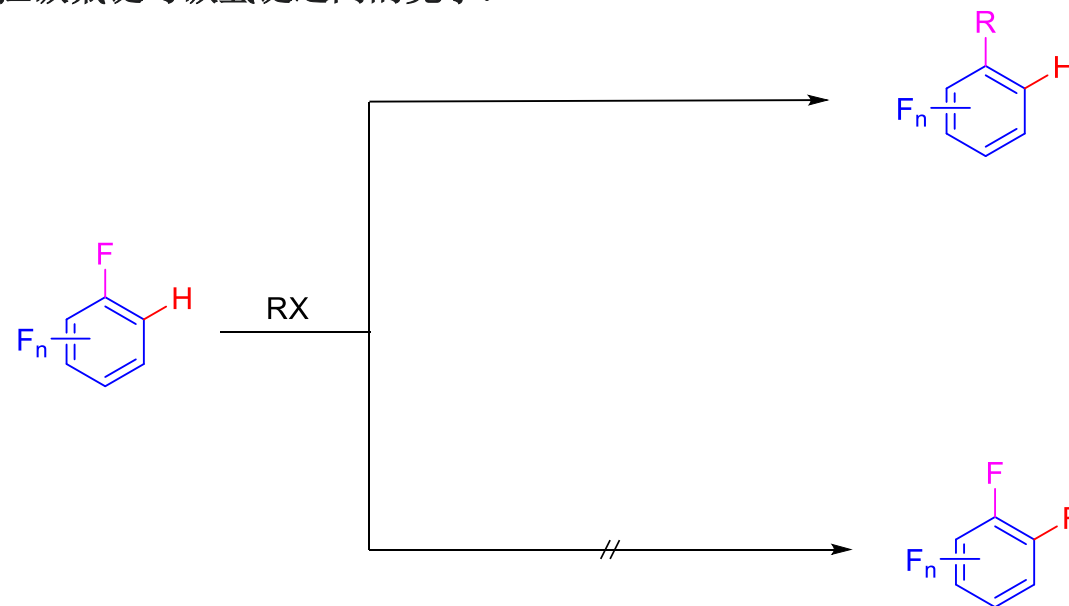


### 3、总结与展望

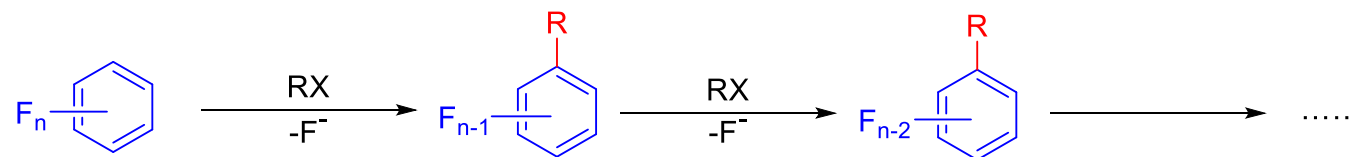
实现无导向基时的精准脱氟：



调控碳氟键与碳氢键之间的竞争：



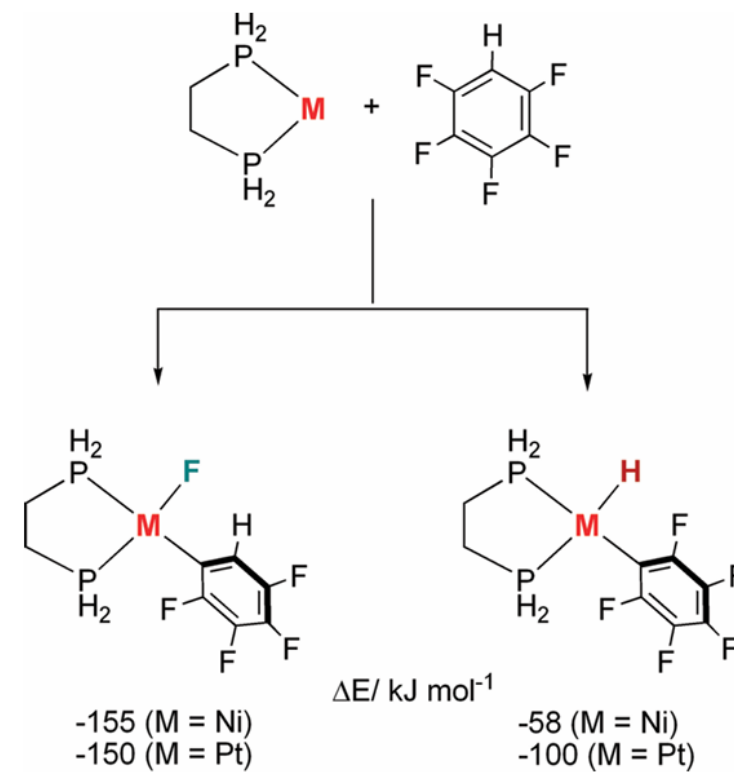
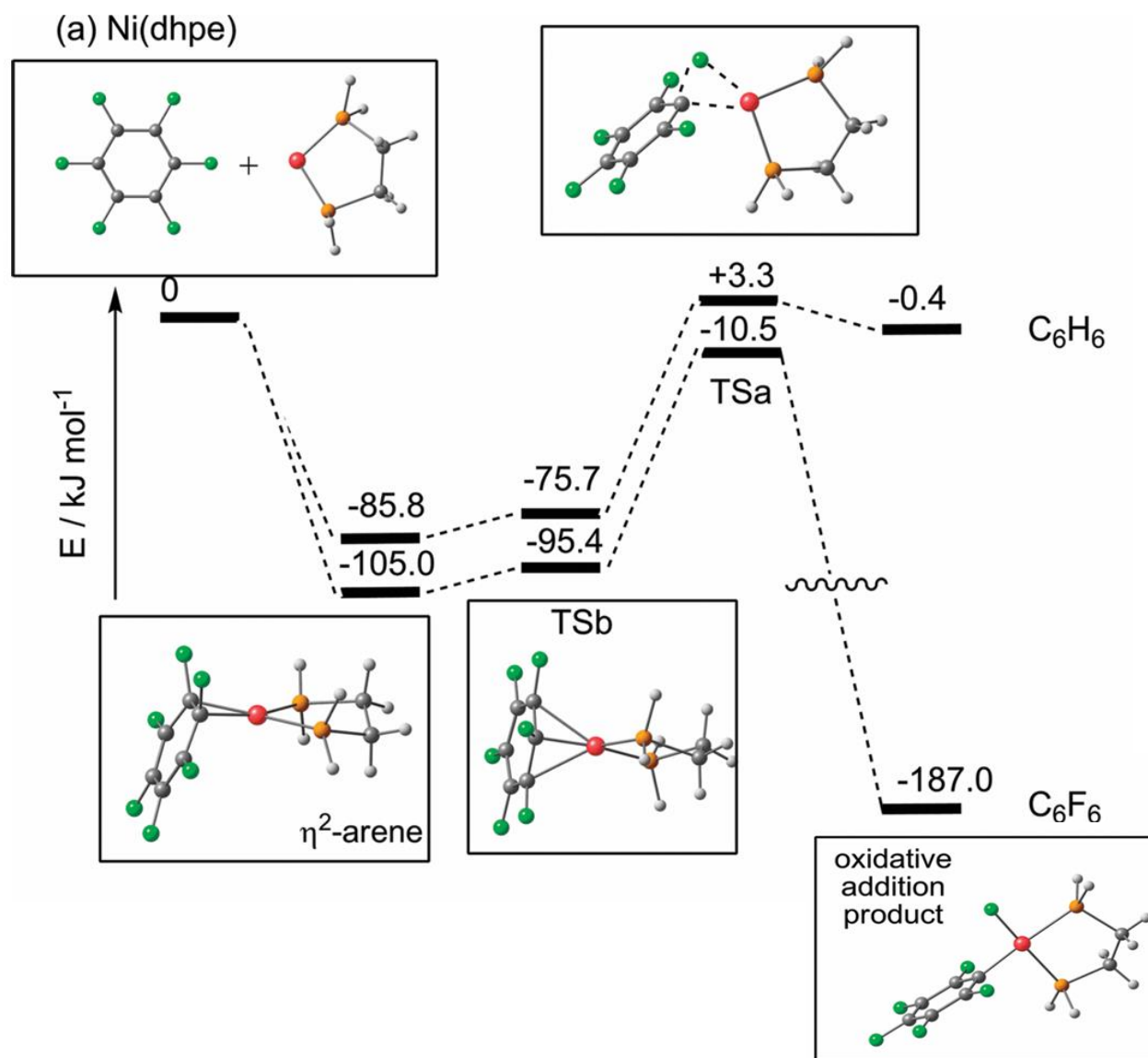
通过连续可控脱氟官能团化构建复杂分子：

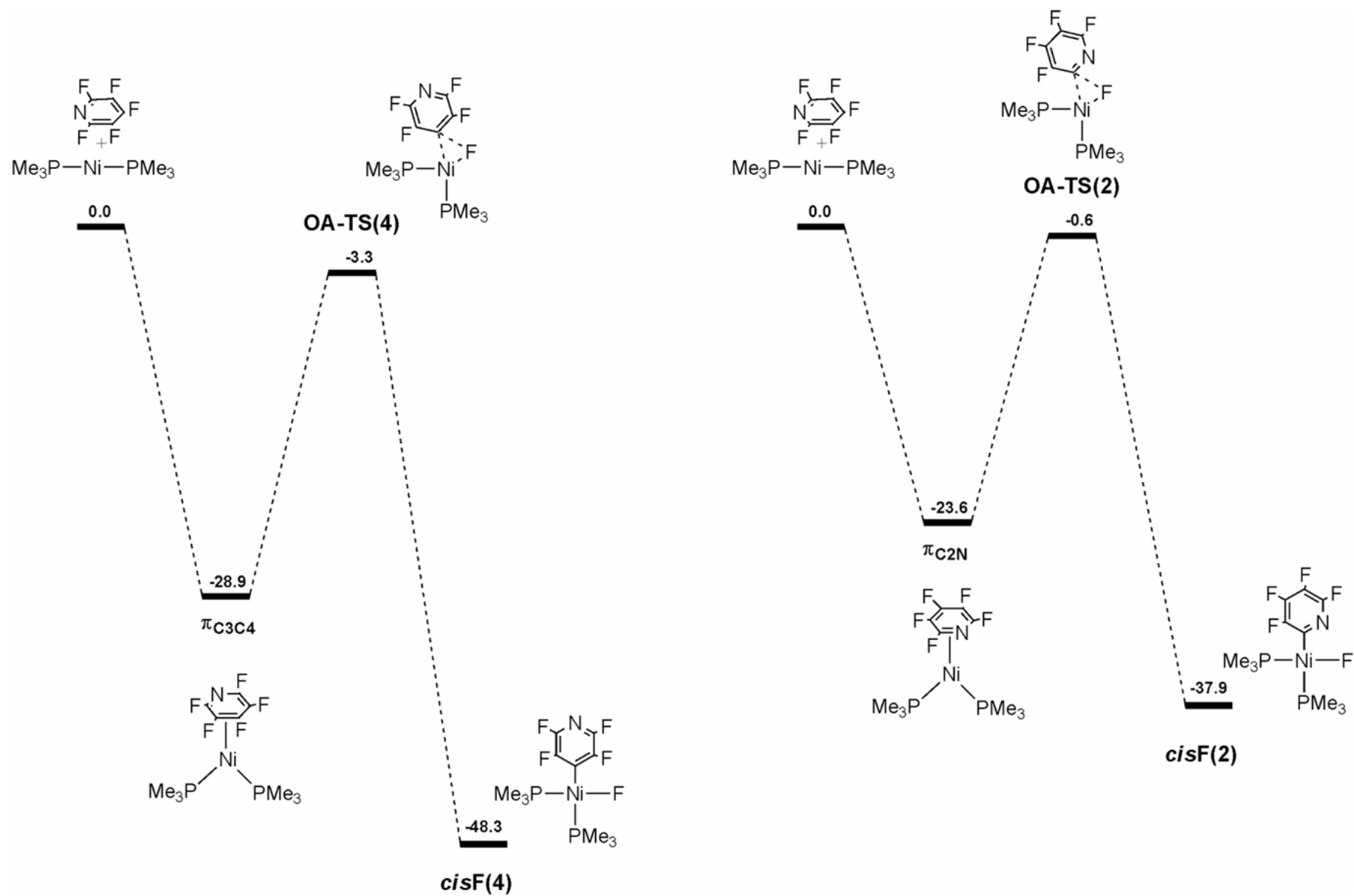


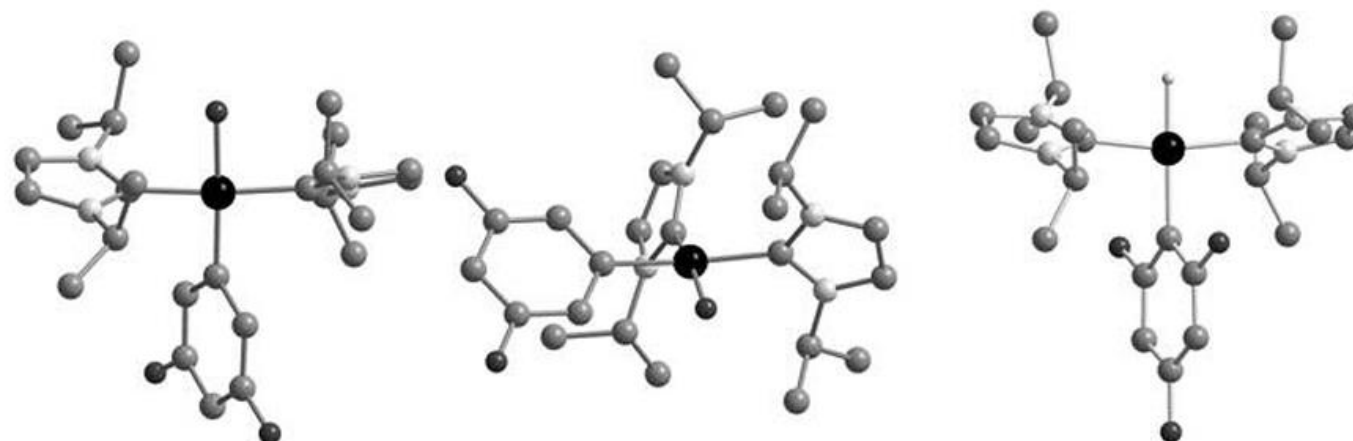
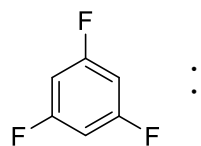


*Thanks for your listening*

# 4、Supporting Information





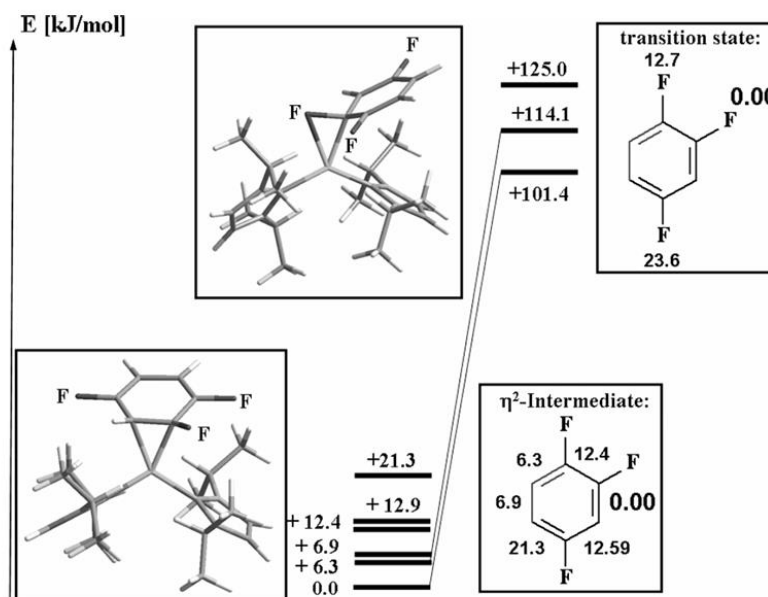


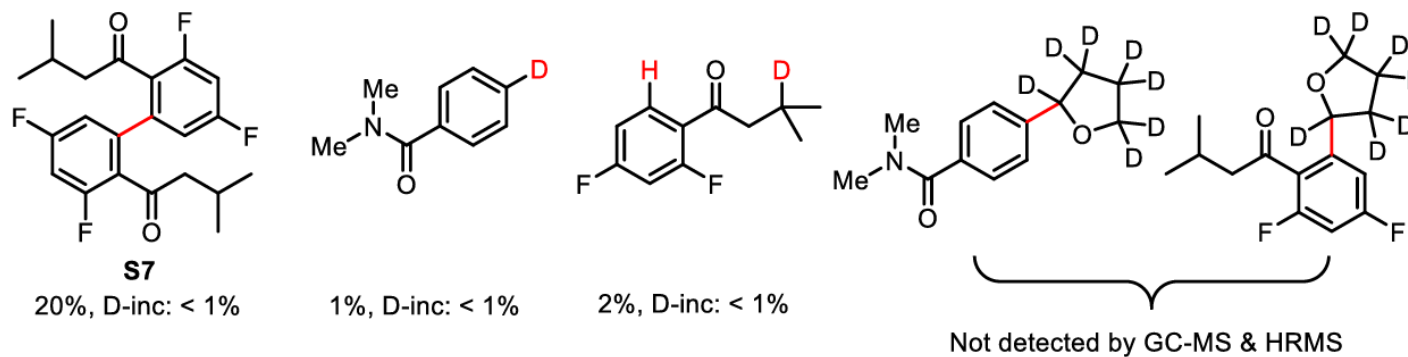
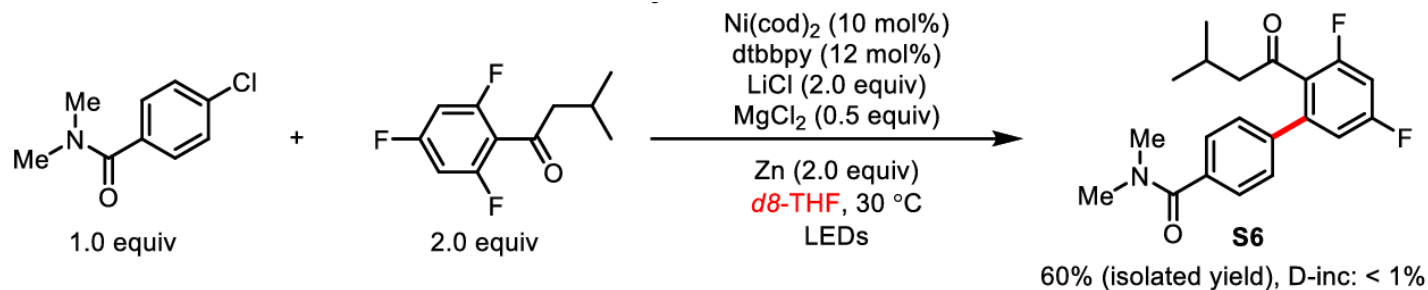
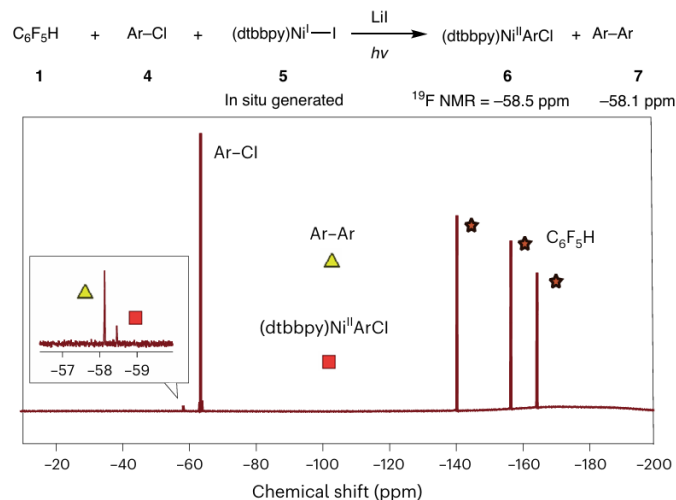
*trans*-[Ni(Pr<sub>2</sub>Im)<sub>2</sub>(F)(3,5-C<sub>6</sub>F<sub>2</sub>H<sub>3</sub>)]    *cis*-[Ni(Pr<sub>2</sub>Im)<sub>2</sub>(F)(3,5-C<sub>6</sub>F<sub>2</sub>H<sub>3</sub>)]    *trans*-[Ni(Pr<sub>2</sub>Im)<sub>2</sub>(H)(2,4,6-C<sub>6</sub>F<sub>3</sub>H<sub>2</sub>)]

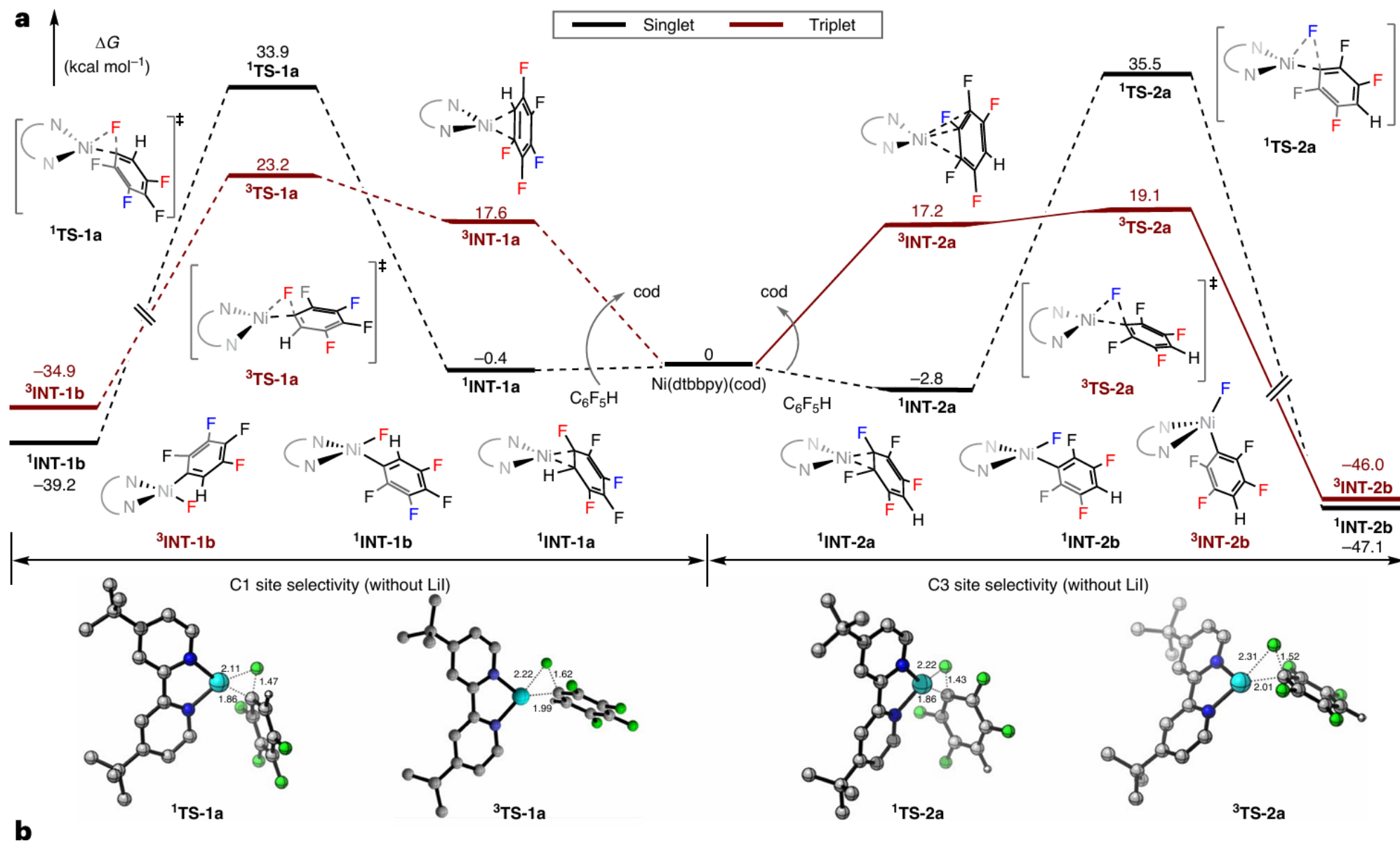
$\Delta E = 0.0$  kJ/mol

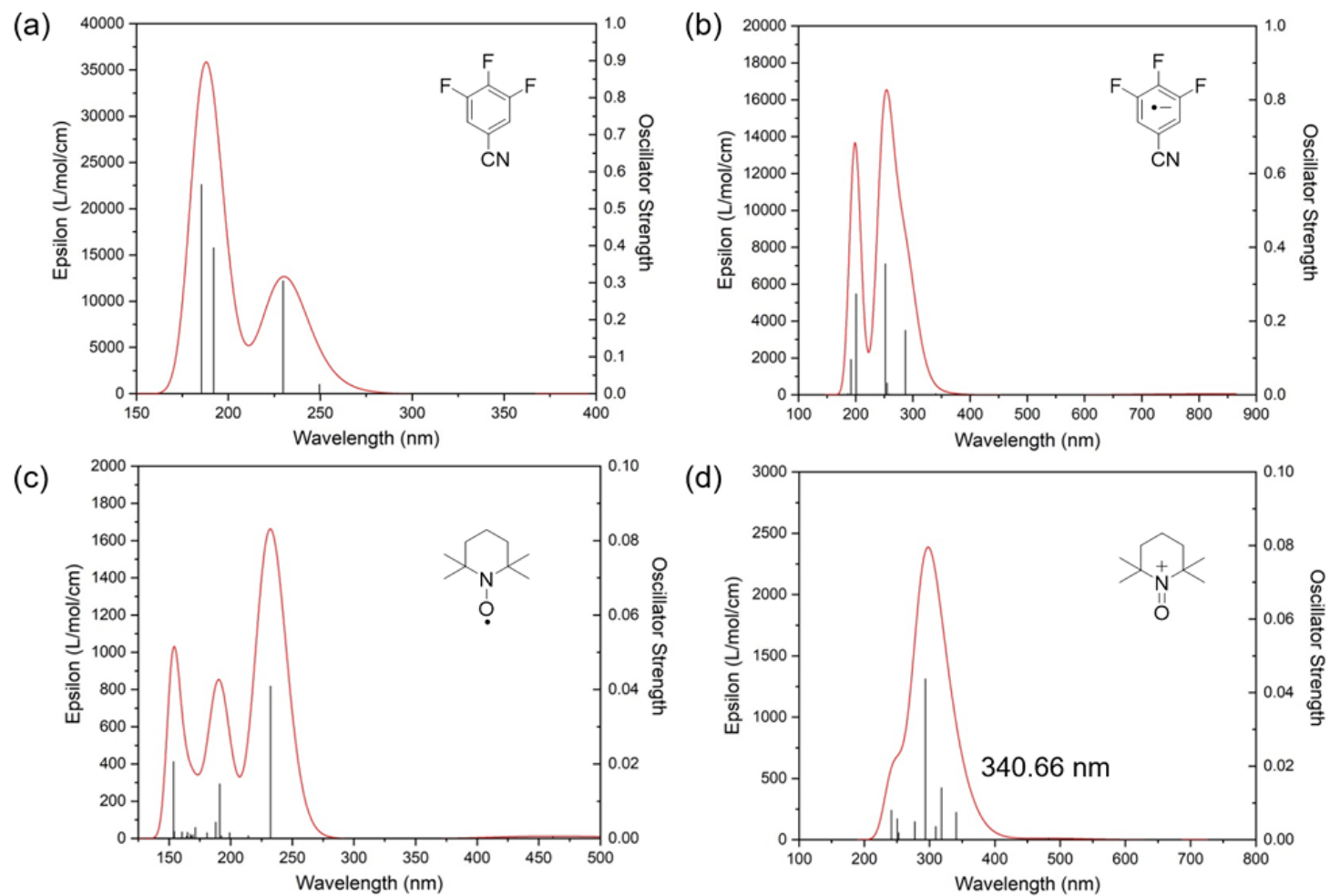
$\Delta E = 28.8$  kJ/mol

$\Delta E = 86.4$  kJ/mol









S25

