

Asymmetric Intermolecular Carbene Insertion to Non-activated C(sp³) - H Bonds

Reporter: Sheng Jiang

Supervisor: Prof. Shengming Ma

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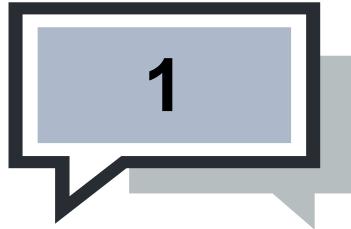
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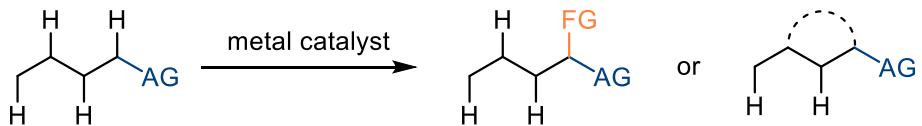


Introduction

Strategies for C(sp³)-H functionalization

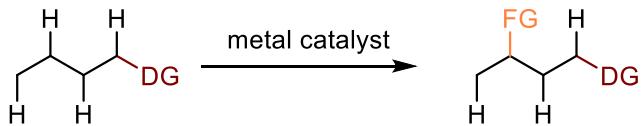
Site-selective C-H functionalization

- ✓ Controlled selectivity with an activating group (AG):

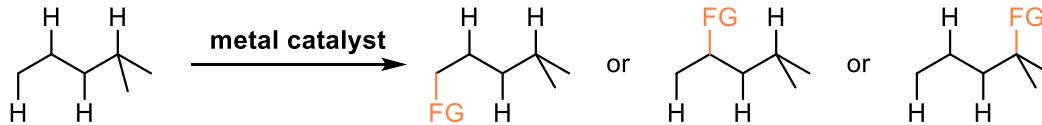


FG: functional group
AG: activating group
DG: directing group

- ✓ Controlled selectivity with a directing group (DG):

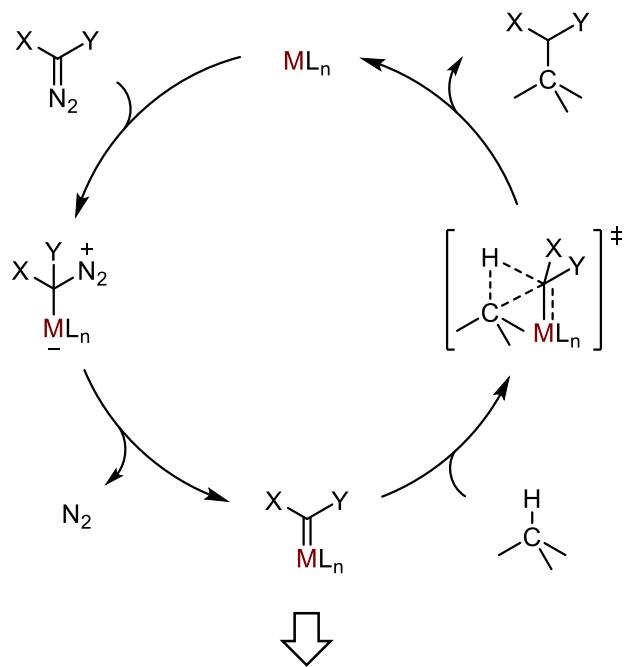


- ◆ How to control the selectivity in non-activated C(sp³)-H bonds?

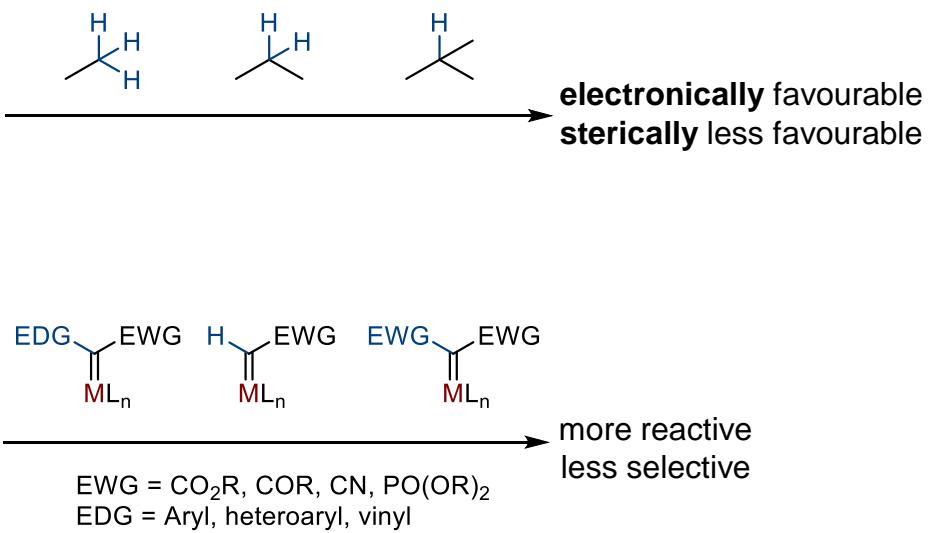


General mechanism and reactivity

Catalytic cycle



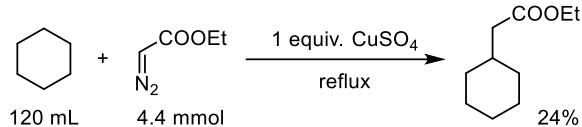
Regular reactivity



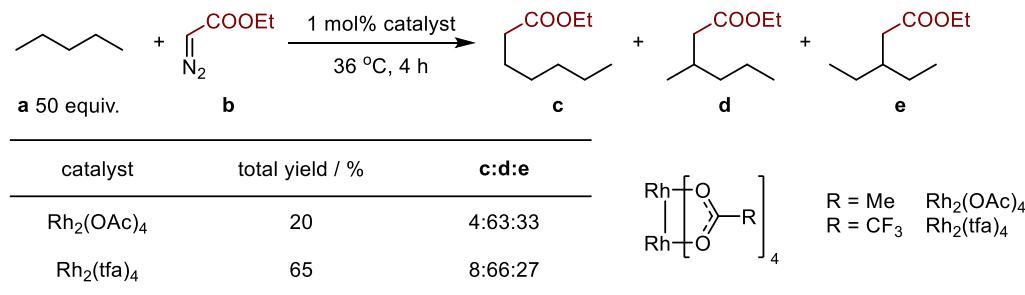
Electrophilic metal carbene

Site-selectivity and stereoselectivity

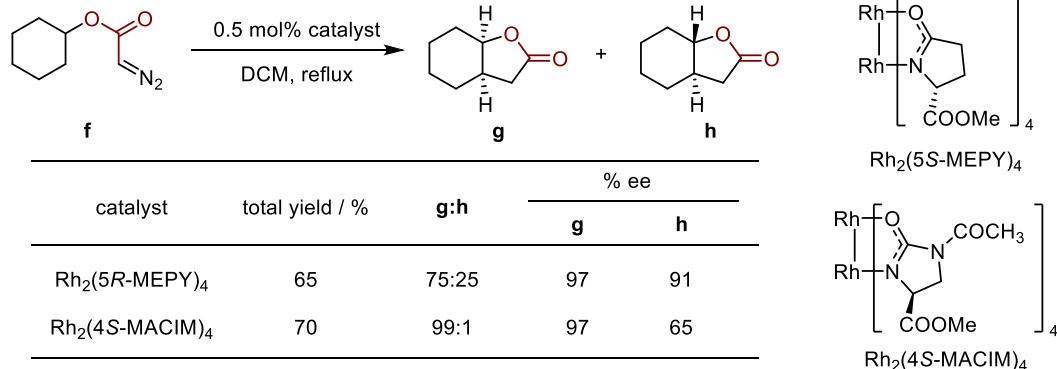
- The first example of intermolecular carbene insertion to non-activated C-H bonds (Scott, 1974)



- Pioneer studies with moderate site-selectivity (Teyssié, 1984)



- Asymmetric intramolecular C-H insertion (Doyle, 1994)



➤ Can we control the site- and stereoselectivity in an intermolecular reaction?

Scott, L. T.; Decicco, G. J., *J. Am. Chem. Soc.* **1974**, *96*, 322-323.

Demonceau, A.; Noels, A. F.; Hubert, A. J.; Teyssié, P., *Bull. Soc. Chim. Belg.* **1984**, *93*, 945-948.

Doyle, M. P.; Dyatkin, A. B.; Roos, G. H. P.; Canas, F.; Pierson, D. A.; Vanbasten, A.; Muller, P.; Polleux, P., *J. Am. Chem. Soc.* **1994**, *116*, 4507-4508.



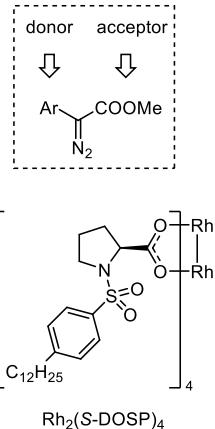
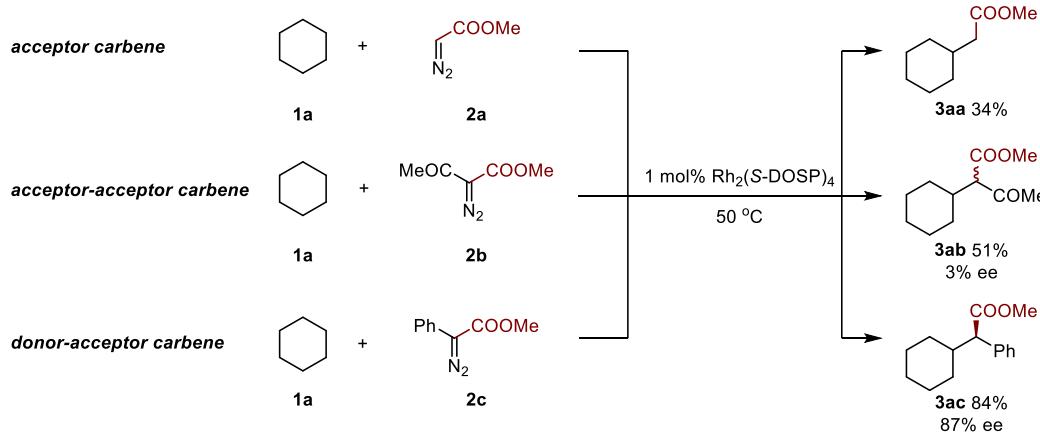
2.1

Seminal works

Seminal works: the first example

The first example of asymmetric intermolecular carbene insertion to non-activated C–H bonds

Reaction of cyclohexane



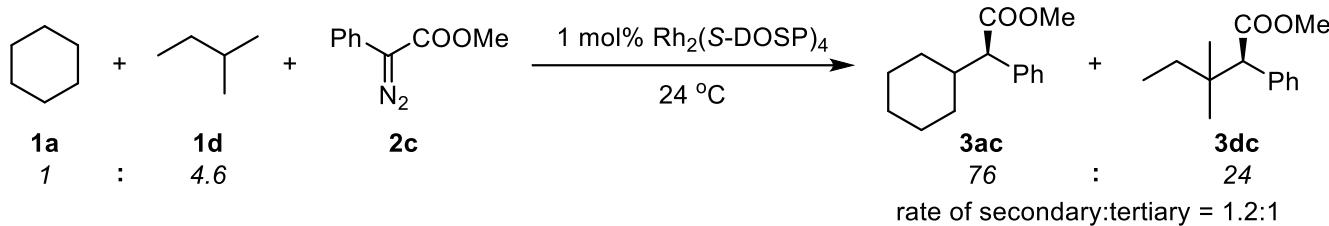
Reaction of acyclic alkanes

The reaction scheme shows the asymmetric intermolecular carbene insertion of acyclic alkanes (1b, 1c, and 1d) with diazophenyl acetate (2c) in a solvent (1) under reflux conditions, catalyzed by 1 mol% $\text{Rh}_2(\text{S-DOSP})_4$. The products are substituted alkanes (3) with yields and enantiomeric excess (ee) values.

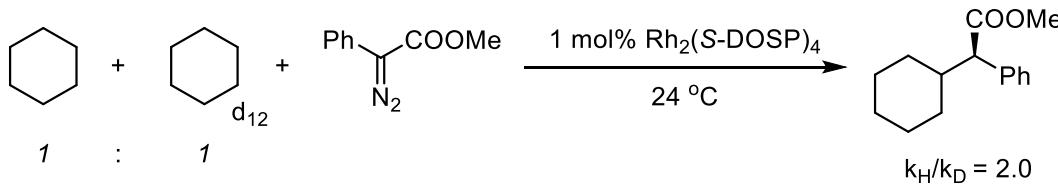
entry	substrates	products	yield / %	ee / %
1	1b	3bc, 4bc	75 (3bc:4bc=58:42)	26(3bc), 86(4bc, 4:1 dr)
2	1c	3cc	24	46
3	1d	3dc	60	68

Seminal works: the first example

The rate of reaction between secondary and tertiary sites



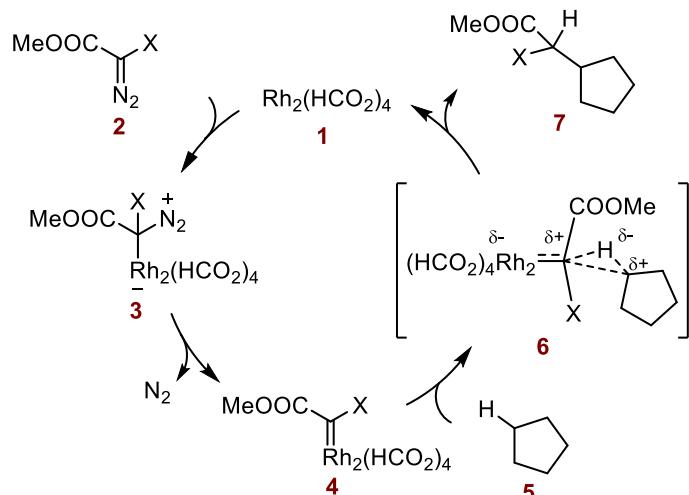
KIE experiment



Seminal works: mechanism

Reaction pathway

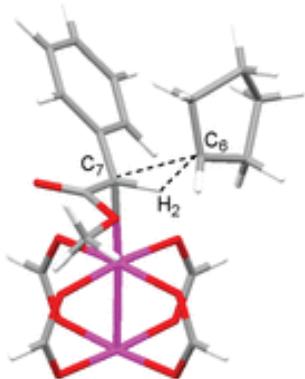
$X = H$ or Ph



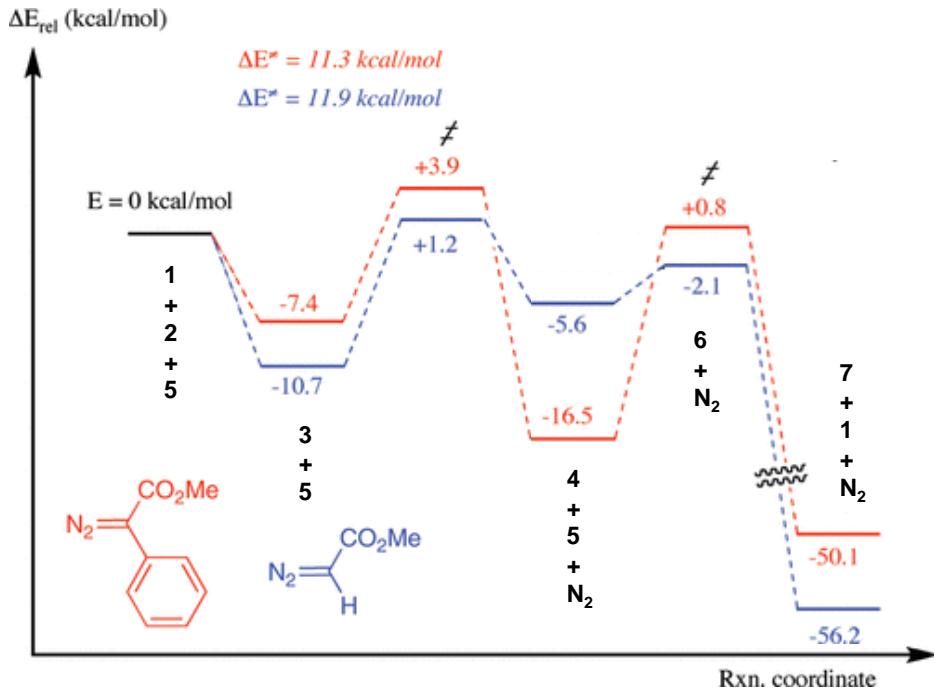
Calculated structures

for $X = \text{Ph}$ $\angle(\text{C}_6\text{-C}_2\text{-C}_7) = 126.5^\circ$

Bond	d (Å)	BO
C ₆ -H ₂	1.543	0.19
C ₇ -H ₂	1.148	0.69
C ₆ -C ₇	2.409	0.36



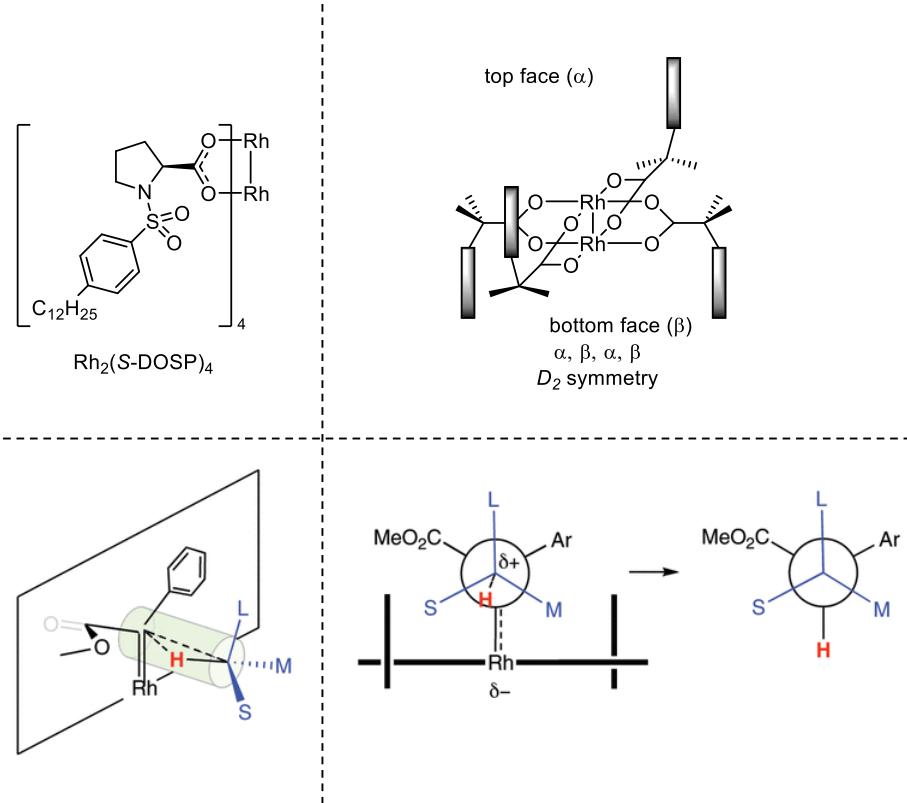
Relative energies



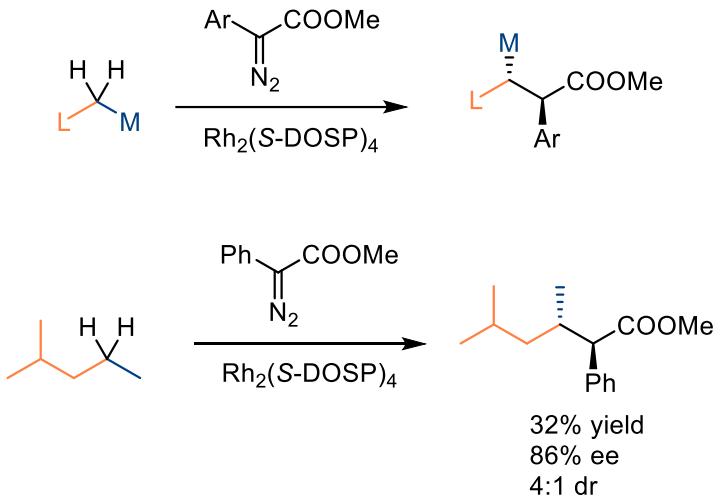
➤ transition state 6: concerted and asynchronous

Seminal works: model

Newman-projection model

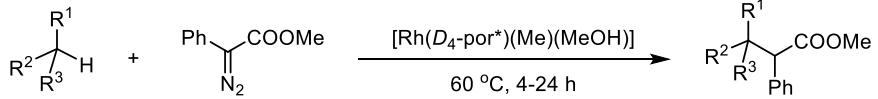


Enantio- and diastereoselectivity



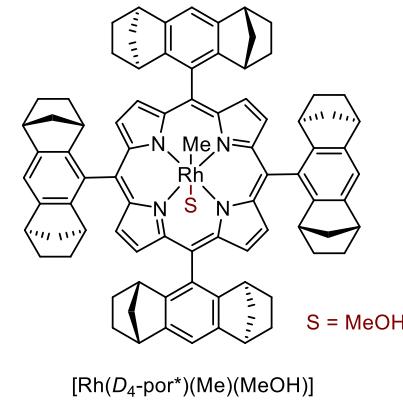
Seminal works: bulky catalysts

Insertion to primary C-H bonds



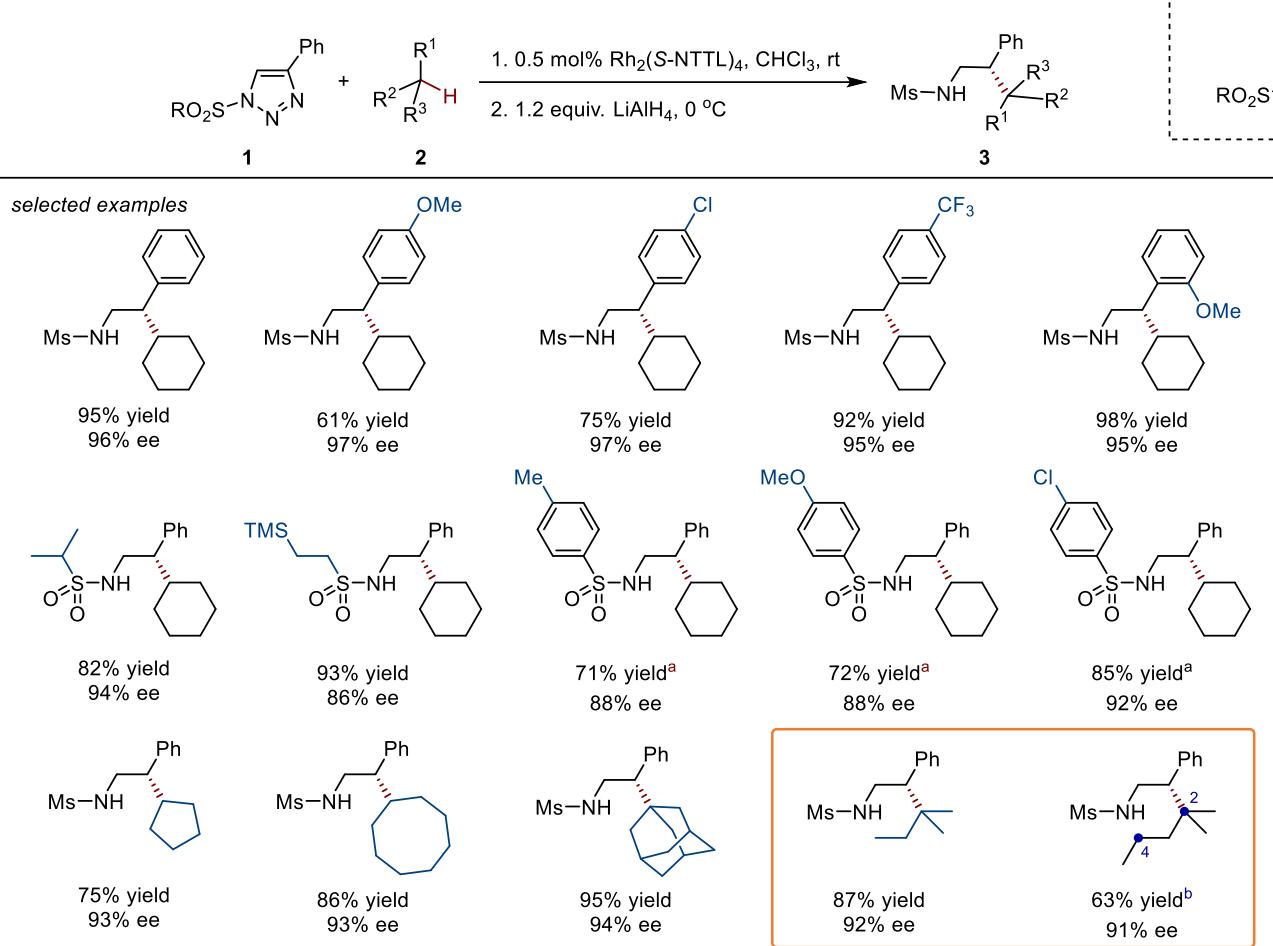
entry	substrates	products			total yield / %	ee / % [a]	$1^\circ : 2^\circ$ [b]
1	1a	1b	1c	1d	66	68 (1a)	3.5 : 1
2	2a	2b	2c		55	65 (2a)	3.8 : 1
3	3a	3b			80	92	
4	4a	4b			64	93	
5	5a	5b			78	88	

Reaction conditions: $N_2C(Ph)CO_2Me$ (0.1 mmol); **1a**, **2a**, **3a**, **4a** (4 mL); **5a** (2 equiv.) in DCE (4 mL); catalyst (0.1 mol% for **3a**, **4a**, **5a**, 1.5 mol% for **1a** and **2a**). [a] Absolute configuration not determined. [b] Normalized for the relative number of hydrogen atoms.

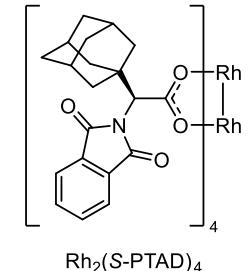
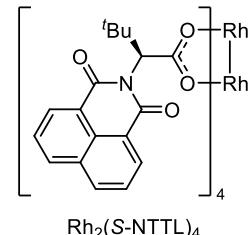
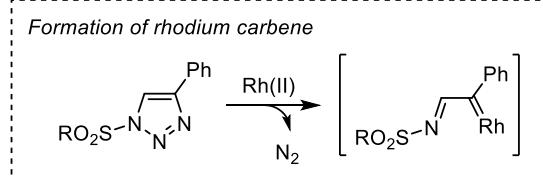


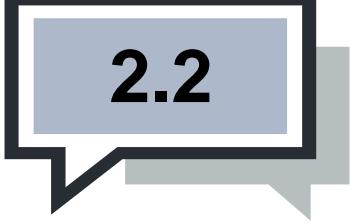
Seminal works: azavinyl carbenes

Selected examples



Reaction conditions: **1** (1 mmol), **2** (2.5 mL), CHCl₃ (2.5 mL) at room temperature. ^a Rh₂(S-PTAD)₄ (0.5 mol%), 40 °C. ^b 12% of C-4 insertion product was observed.





2.2

Acyclic Systems

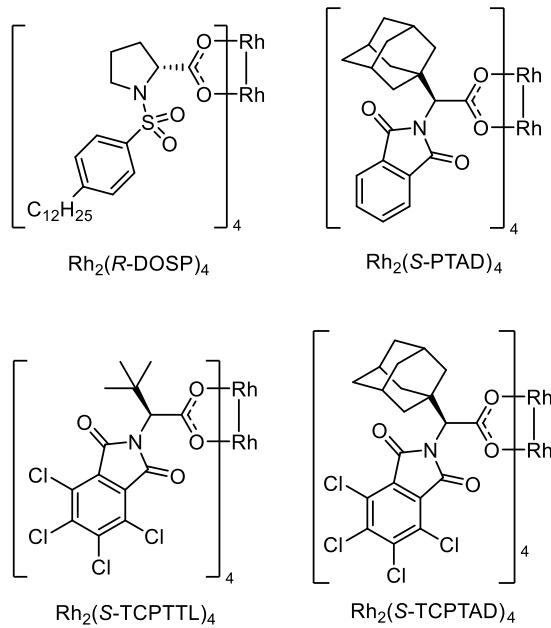
Acyclic Systems: tertiary C-H bonds

Optimization of reaction conditions

Reaction scheme: Compound **1** (ROOC-*p*-BrC₆H₄-N₂) reacts with 3 equivalents of a chiral alkene in DCM with 1 mol% catalyst to yield products **2** and **3**.

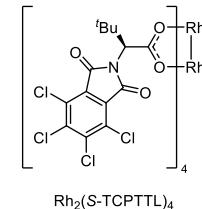
entry	catalyst	R	t / °C	yield / %	rr of 2 : 3	ee of 2 / %
1	Rh ₂ (R-DOSP) ₄	CH ₃	39	83	85:15	-43
2	Rh ₂ (S-PTAD) ₄	CH ₃	39	50	74:26	-34
3	Rh ₂ (S-TCPTTL) ₄	CH ₃	39	86	86:14	77
4	Rh ₂ (S-TCPTAD) ₄	CH ₃	39	89	87:13	79
5	Rh₂(S-TCPTAD)₄	CH₂CF₃	39	92 (83)	90:10	77
6	Rh ₂ (S-TCPTAD) ₄	CH ₂ CF ₃	24	85 (77)	91:9	80
7	Rh ₂ (S-TCPTAD) ₄	CH ₂ CF ₃	0	85 (79)	93:7	82
8	Rh₂(S-TCPTAD)₄	CH₂CF₃	-40	80 (77)	96:4	86

The percentage yields refer to the combined yield of products **2** and **3**. The isolated yield of **2** is given in parentheses.



Acyclic Systems: tertiary C-H bonds

Scopes



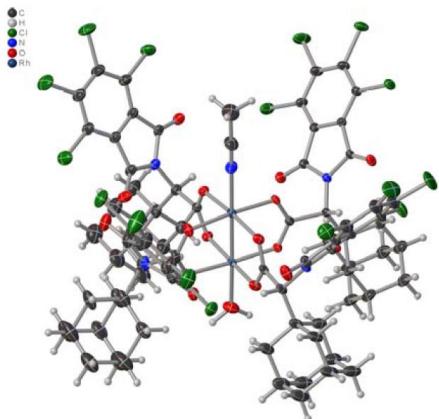
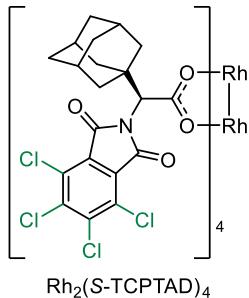
selected examples

 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{(p-Br)C}_6\text{H}_4$ $>98:2 \text{ rr}$ $81\% \text{ ee}$ $86\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ 39°C $87:13 \text{ rr}$ $77\% \text{ ee}$ $76\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ -40°C $96:4 \text{ rr}$ $87\% \text{ ee}$ $65\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ 39°C $92:8 \text{ rr}$ $82\% \text{ ee}$ $73\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ -40°C $98:2 \text{ rr}$ $92\% \text{ ee}$ $64\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ $>98:2 \text{ rr}$ $86\% \text{ ee}$ $66\% \text{ yield}$
 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ $>98:2 \text{ rr}$ $78\% \text{ ee}$ $87\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ $>98:2 \text{ rr}$ $83\% \text{ ee}$ $93\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ $>98:2 \text{ rr}$ $84\% \text{ ee}$ $70\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ $>98:2 \text{ rr}$ $77\% \text{ ee}$ $90\% \text{ yield}$		
 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{CH}_2-\text{CH}_2-\text{CH}_3$ 39°C $88:12 \text{ rr}$ $76\% \text{ ee}$ $73\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{C}_6\text{H}_4-\text{Cl}$ $96:4 \text{ rr}$ $68\% \text{ ee}$ $60\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{C}_6\text{H}_4-\text{CF}_3$ $>98:2 \text{ rr}$ $74\% \text{ ee}$ $89\% \text{ yield}$	 $\text{F}_3\text{CH}_2\text{COOC}-\text{C}(\text{R}^1)-\text{C}(=\text{N}_2)-\text{C}_6\text{H}_4-\text{OMe}$ $>98:2 \text{ rr}$ $63\% \text{ ee}$ $94\% \text{ yield}$		

Some of the C–H functionalization occurred at the most accessible secondary position (**marked in blue**) of the substrates.

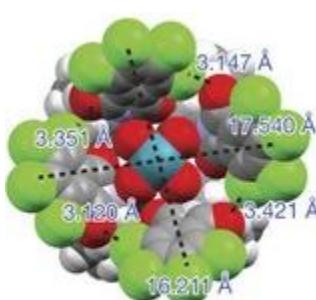
Acyclic Systems: chiral pocket

Structural information

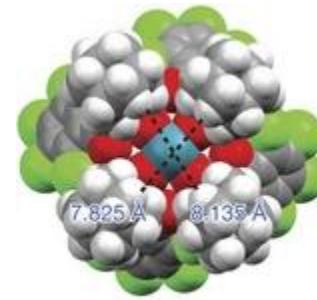


X-ray structure

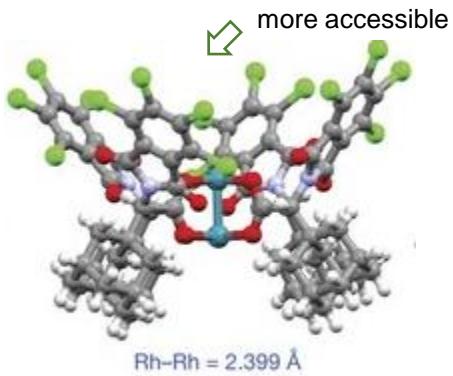
- phthalimido groups on the top side
- slightly distorted C_4 symmetric structure



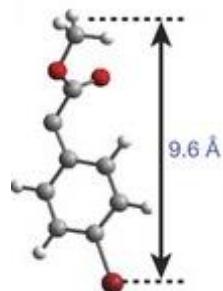
top view



bottom view

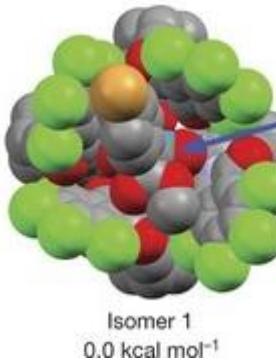


side view

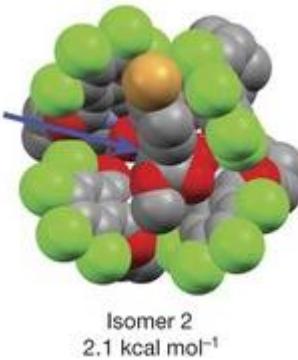


length of carbene

- π -stacking between phthalimido and aryl groups
- preferred attack at the *Re* face



Isomer 1
0.0 kcal mol⁻¹

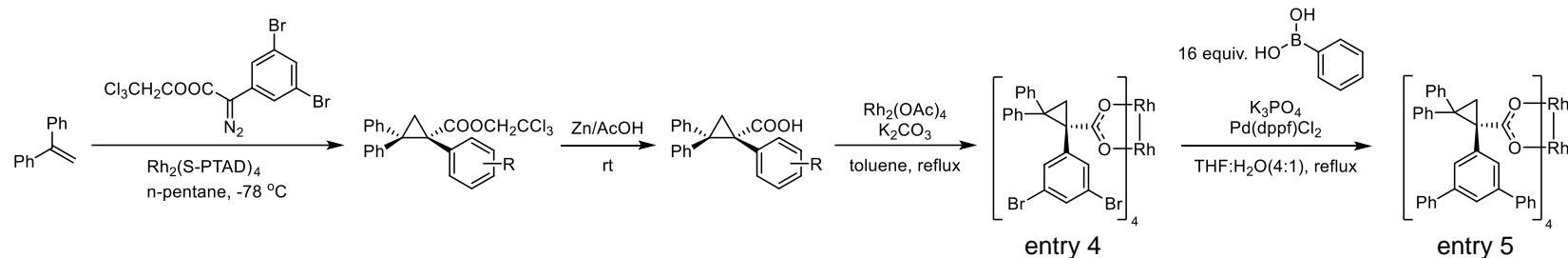


Isomer 2
2.1 kcal mol⁻¹

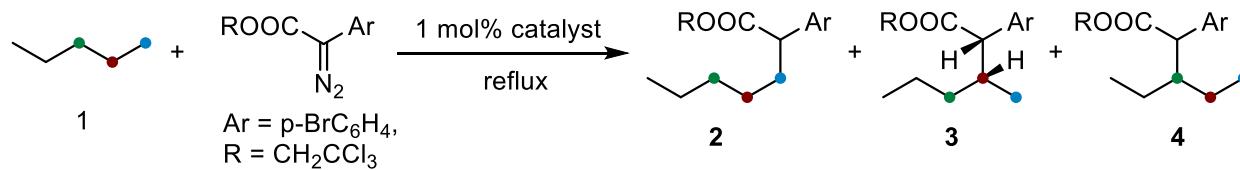
two most stable isomers

Acyclic Systems: secondary C-H bonds

Synthesis of catalysts



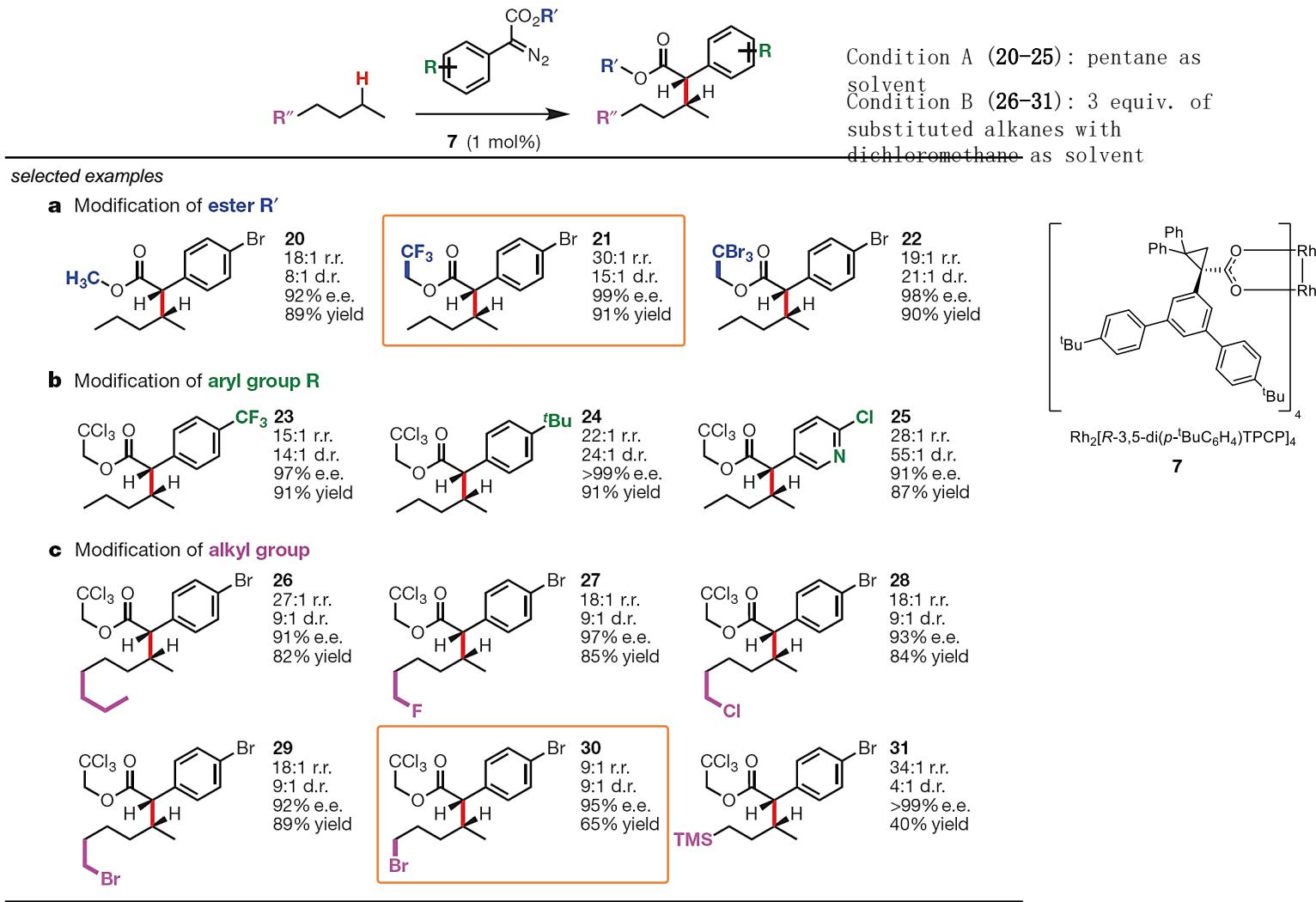
Optimization



entry	catalyst	2:3:4	dr of 3	ee% of 3	yield / %
1	$\text{Rh}_2(\text{S-DOSP})_4$	nd:29:1	3:1	82	98
2	$\text{Rh}_2(R-p\text{-PhTPCP})_4$	1:2:nd	14:1	92	97
3	$\text{Rh}_2(R-p\text{-BrTPCP})_4$	1:4:nd	6:1	91	98
4	$\text{Rh}_2(R\text{-3,5-di-BrTPCP})_4$	1:26:nd	5:1	78	95
5	$\text{Rh}_2(R\text{-3,5-di-PhTPCP})_4$	1:22:nd	5:1	96	95
6	$\text{Rh}_2[R\text{-3,5-di-(3,5-diPhC}_6\text{H}_3)\text{TPCP}]_4$	1:5:nd	16:1	97	98
7	$\text{Rh}_2[R\text{-3,5-di}(p\text{-tBuC}_6\text{H}_4)\text{TPCP}]_4$	1:25:nd	20:1	99	99

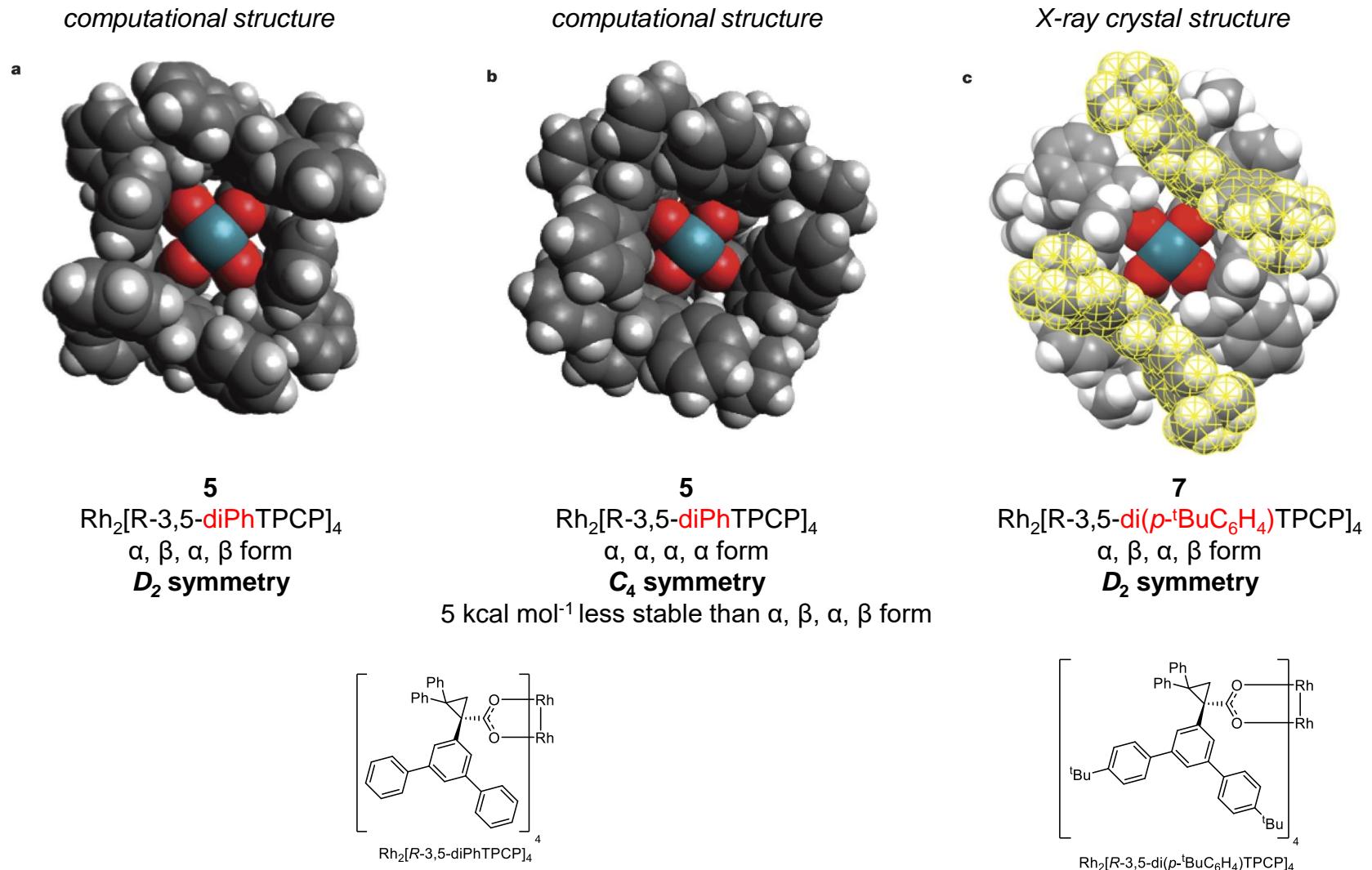
Acyclic Systems: secondary C-H bonds

Scopes



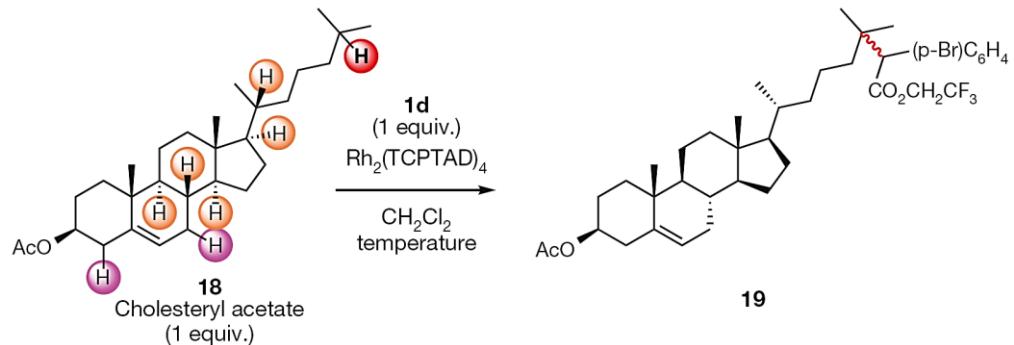
Acyclic Systems: secondary C-H bonds

Structural information

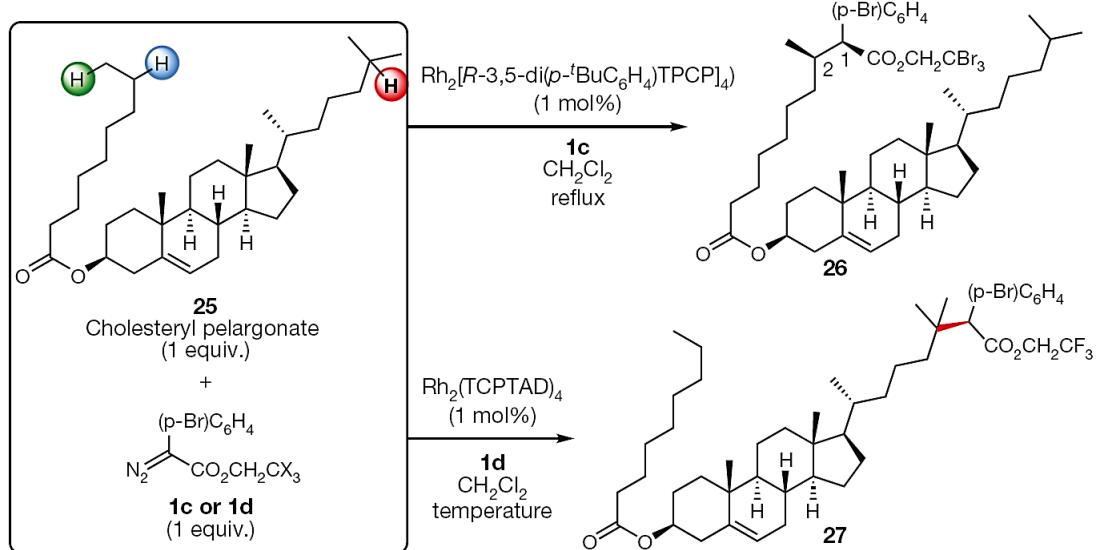


Acyclic Systems: secondary vs tertiary

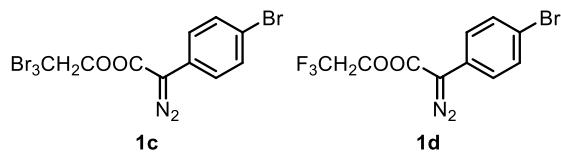
Controlled site-selectivity in elaborate substrates



Rh ₂ (TCPTAD) ₄	19	Temperature	r.r.	d.r.	Yield (%)
S, 1 mol%	<i>R</i>	39 °C	>98:2	11:1	78
<i>R</i> , 1 mol%	<i>S</i>	39 °C	>98:2	11:1	86
<i>R</i> , 0.1 mol%	<i>S</i>	39 °C	>98:2	9:1	83
<i>R</i> , 1 mol%	<i>S</i>	24 °C	>98:2	13:1	80
<i>R</i> , 1 mol%	<i>S</i>	0 °C	>98:2	16:1	60



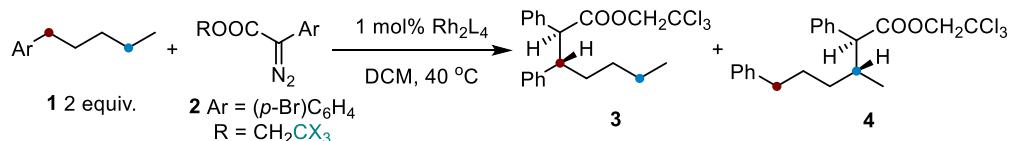
r.r.	d.r. of C1/steroid	d.r. of C1/C2	Yield (%)
87:10:3	39:1	9:1	68



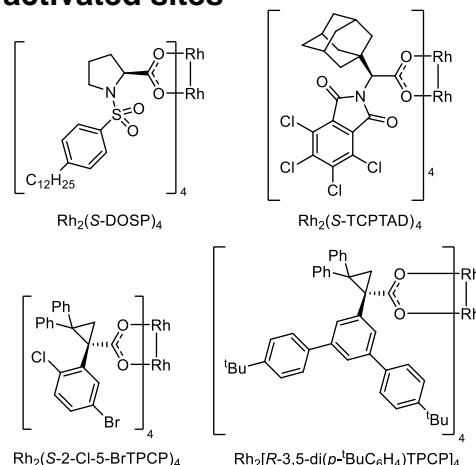
Rh ₂ (TCPTAD) ₄	27	Temperature	r.r.	d.r.	Yield (%)
<i>R</i>	<i>S</i>	39 °C	87:13	6:1	74
<i>R</i>	<i>S</i>	24 °C	89:11	>20:1	68
<i>R</i>	<i>S</i>	0 °C	92:8	>20:1	65

Acyclic Systems: secondary vs benzylic

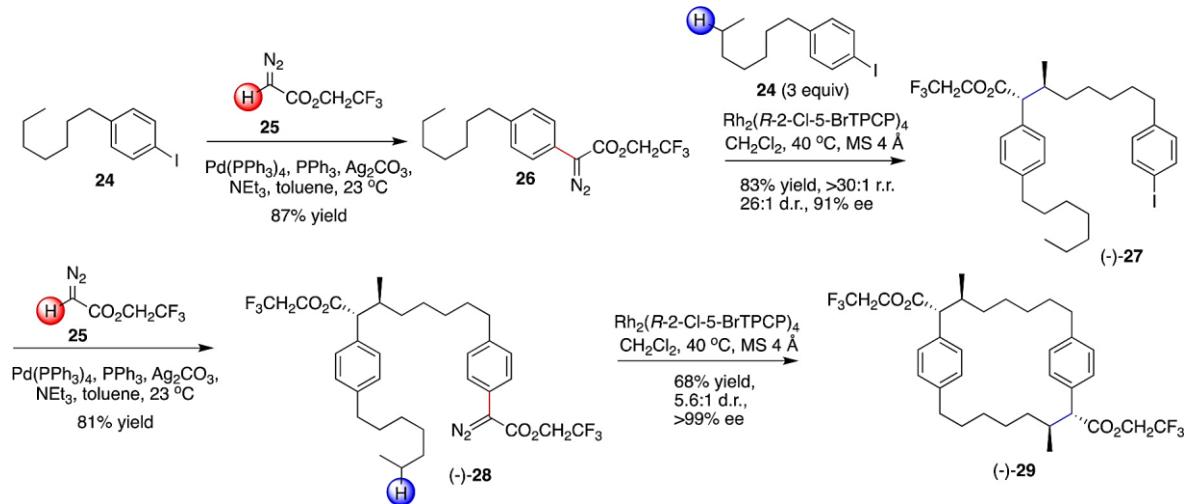
Optimization: non-activated C-H bonds in the presence of electronically activated sites



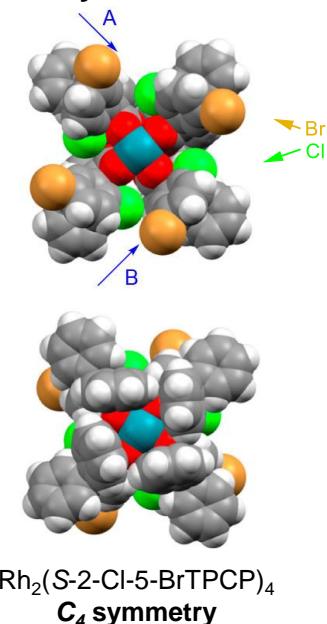
entry	Rh ₂ L ₄	X	yield / %	rr of 3:4	major product (3 or 4)	
					dr	ee / %
1	Rh ₂ (S-DOSP) ₄	Cl	68	6:1	4:1	77
2	Rh ₂ (S-TCPTAD) ₄	Br	75	11:1	16:1	90
3	Rh ₂ [R-3,5-di(<i>p</i> -tBuC ₆ H ₄)TPCP] ₄	Cl	69	1:3	7:1	89
4	Rh ₂ (S-2-Cl-5-BrTPCP) ₄	Cl	87	1:20	30:1	89
5	Rh ₂ (S-2-Cl-5-BrTPCP) ₄	F	86	1:24	28:1	91



Application: synthesis of macrocyclic core

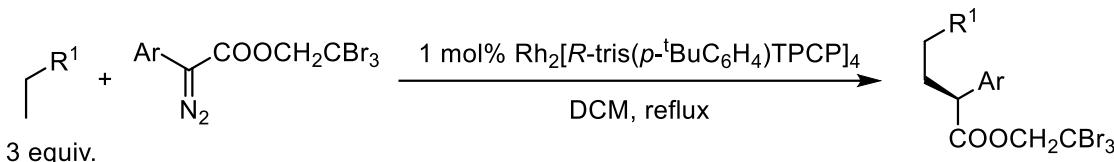


X-ray structure

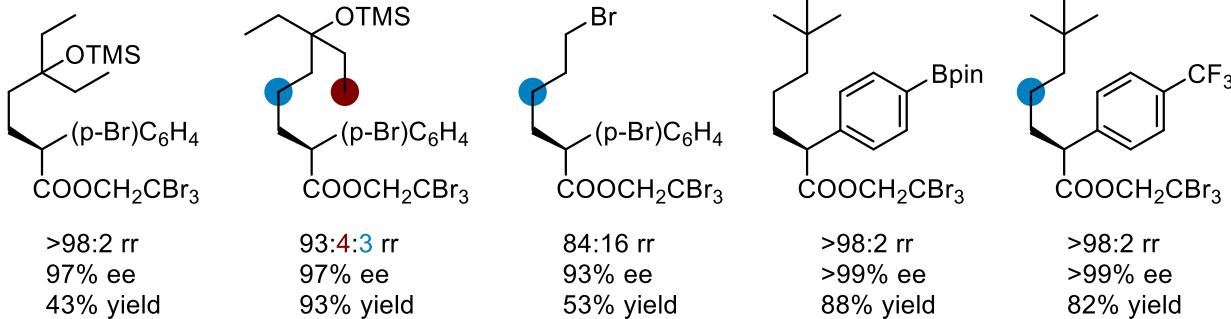
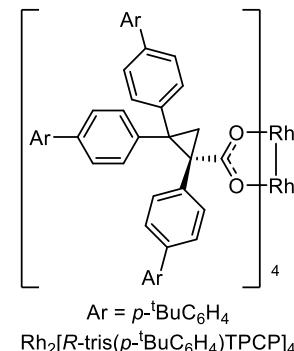
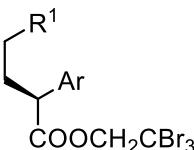
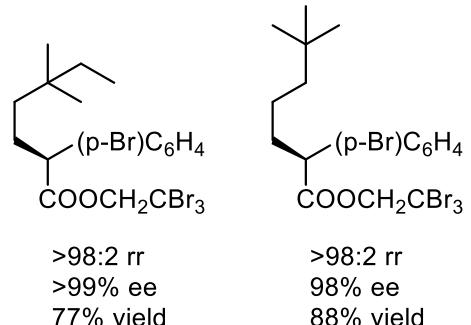
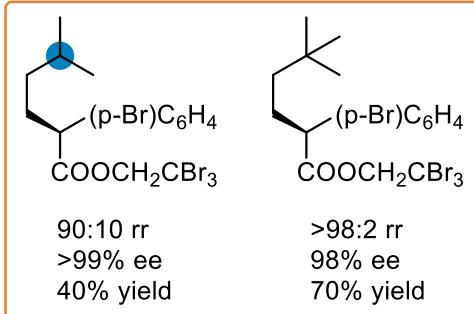


Acyclic Systems: primary C-H bonds

Scopes



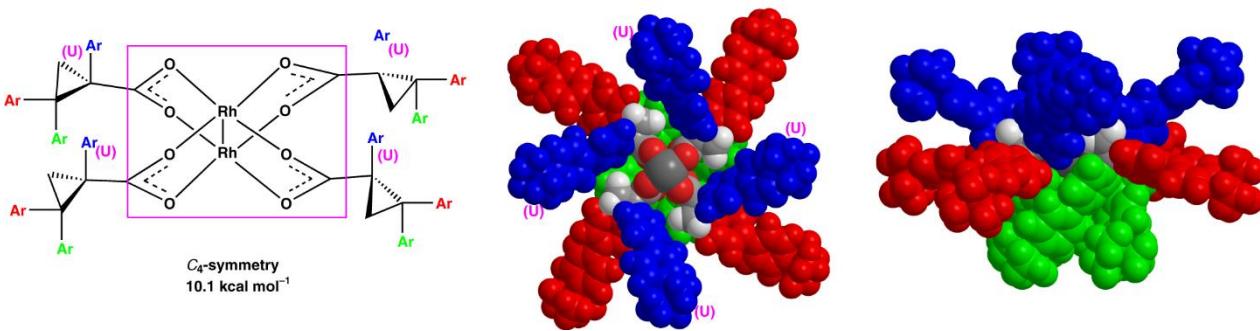
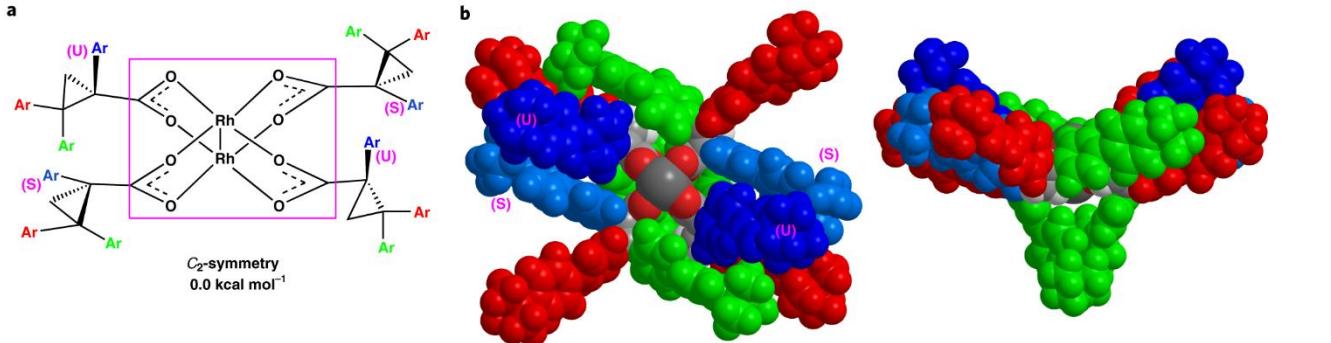
selected examples



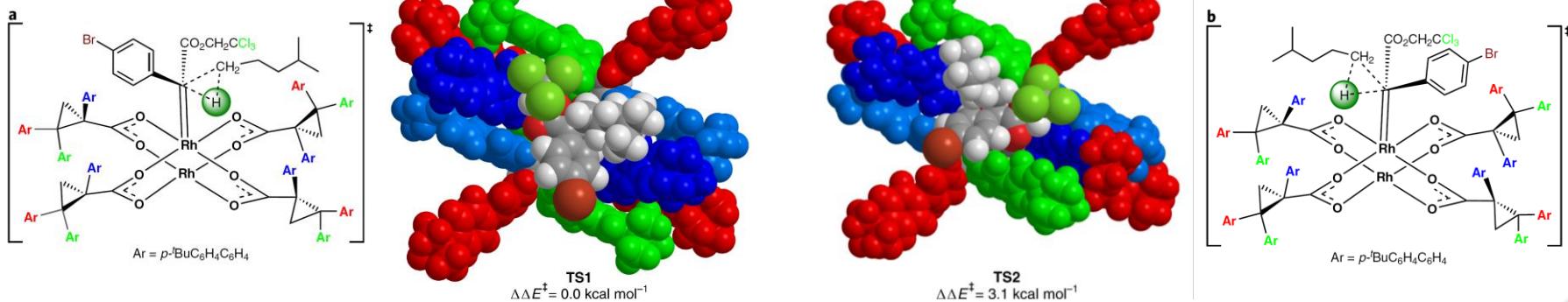
Some of the C–H functionalization occurred at the tertiary or secondary position (**marked in blue**) of the substrates. Some of the C–H functionalization occurred at the primary position (**marked in red**) of the substrates.

Acyclic Systems: primary C-H bonds

Computational structure



Transition state



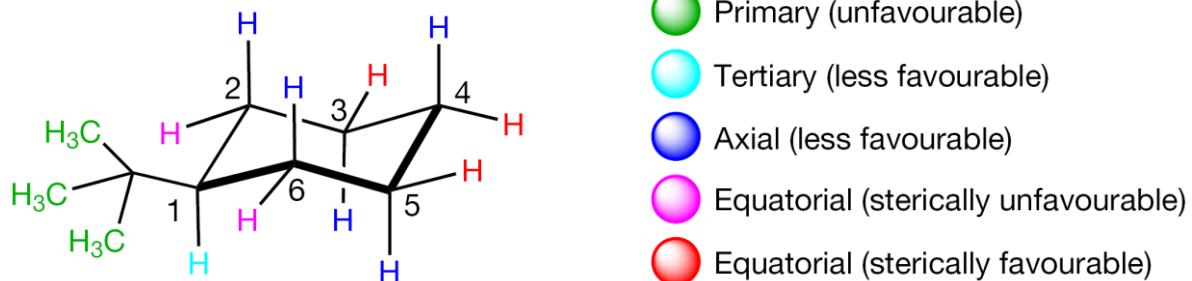


2.3

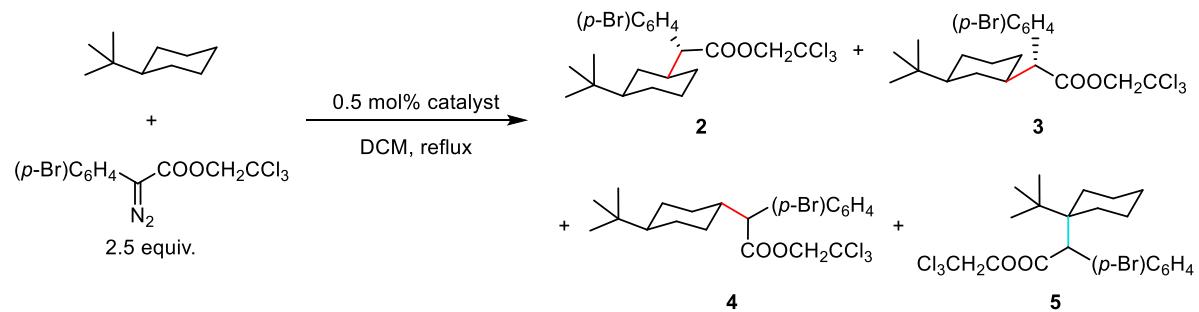
Cyclic Systems

Cyclic Systems: desymmetrization of cyclohexanes

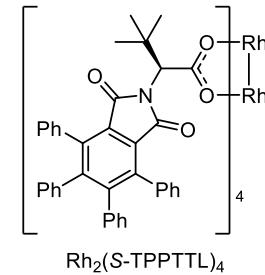
The structure of *tert*-butyl cyclohexane



Screen of catalyst

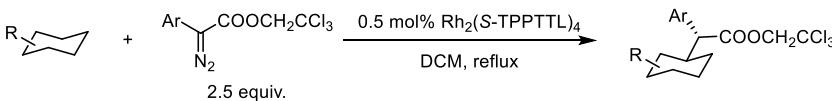


catalyst	product ratio			
	2	3	4	5
Rh ₂ (S-DOSP) ₄	60.8	9.7	24.1	5.3
Rh ₂ (S-TCPTT) ₄	49.6	9.3	40.6	nd
Rh ₂ [R-3,5-di(<i>p</i> - ^t BuC ₆ H ₄)TPCP] ₄	29.4	8.4	62.2	nd
Rh ₂ [R-tris(<i>p</i> - ^t BuC ₆ H ₄)TPCP] ₄	70.3	17.2	12.4	nd
Rh ₂ (S-TPPTT) ₄	91.3	8.7	nd	nd



Cyclic Systems: cyclohexanes

Scopes



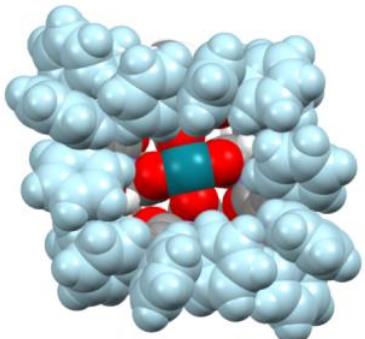
selected examples

 1 , 85%, 97% ee 12.5:1 rr. 4.0:1 dr ^a	 2 , 81%, 96% ee >50:1 rr 4.2:1 dr ^a	 3 , 77%, 90% ee >50:1 rr 3.7:1 dr ^a	 4 , 85%, 95% ee 40:1 rr 12:1 dr ^a	 5 , 80%, 95% ee >50:1 rr 11:1 dr ^a	 6 , 58%, 96% ee 9.9:1 rr 7.7:1 dr ^a
 7 , 80%, 97% ee 19.0:1 rr 14:1 dr ^a	 8 , 64%, 92% ee 50:1 rr 15:1 dr ^a	 9 , 89%, 97% ee >50:1 rr 11:1 dr ^a	 10 , 61%, 96% ee >50:1 rr 16:1 dr ^a	 11 , 81%, 98% ee >50:1 rr 11 :1 dr ^a	 12 , 58%, 98% ee >50:1 rr 9.8:1 dr ^a
 13 , 70%, 98% ee 16:1 rr 3.7:1 dr ^a	 14 , 75%, 85% ee >50:1 rr 25:1 dr ^a	 15 , 62%, 59% ee 15:1 rr >50:1 dr ^b	 16 , 80%, 91% ee >50:1 rr >50:1 dr ^b	 17 , 77%, 94% ee 4.8:1 rr >50:1 dr ^b	 18 , 68%, 98% ee >50:1 rr 6.7:1 dr ^a

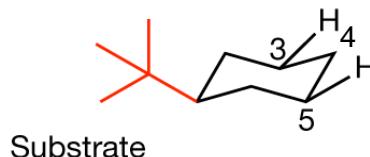
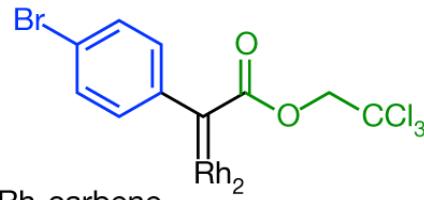
^a No ring diastereomers were observed. ^b Owing to symmetry, there are no side-chain diastereomers.

Cyclic Systems: cyclohexanes

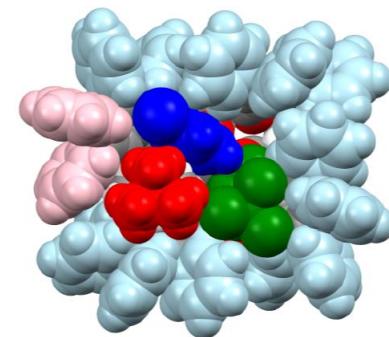
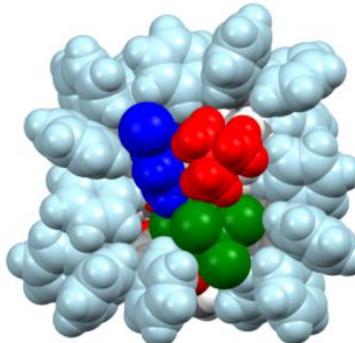
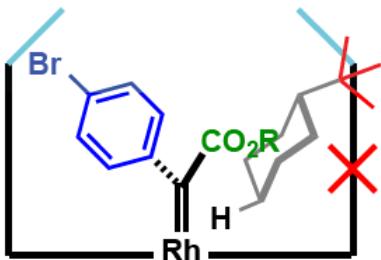
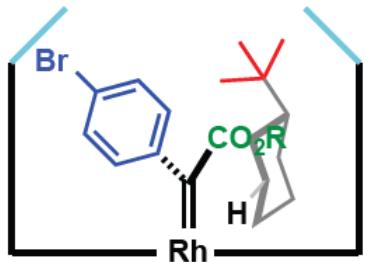
Computational studies



Isomer **47b**
 C_4 symmetry

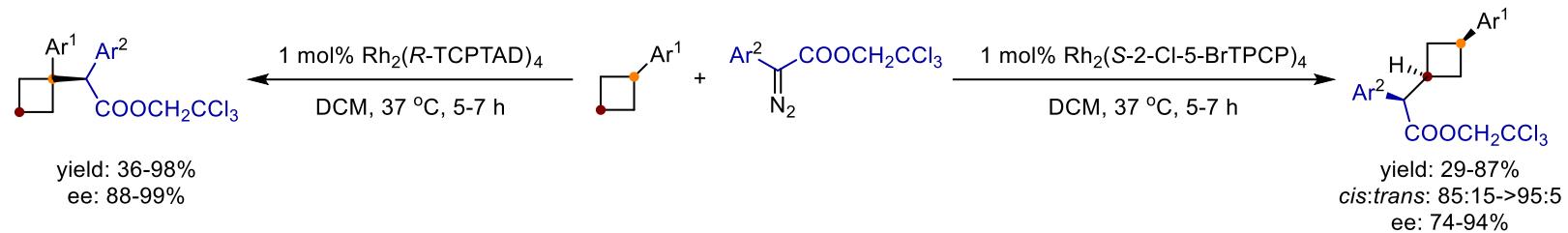


Rh-carbene-substrate complex



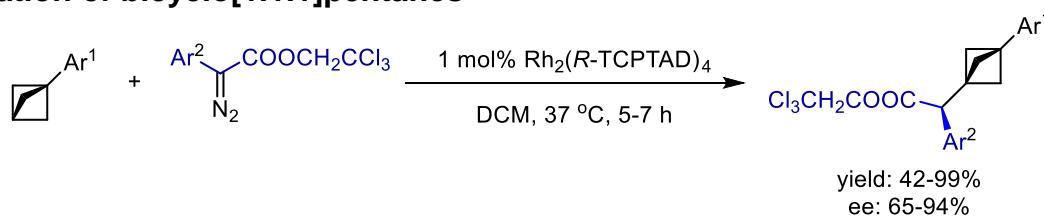
Cyclic Systems: BCPs and cyclobutanes

C-H functionalization of cyclobutanes

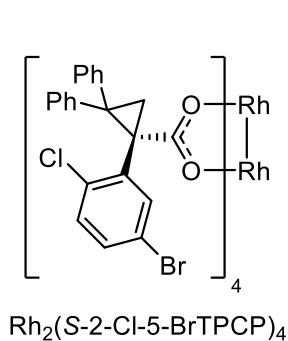
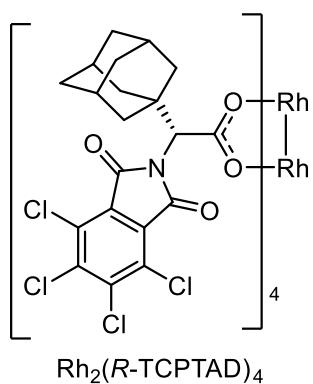


Garlets, Z. J.; Wertz, B. D.; Liu, W. B.; Voight, E. A.; Davies, H. M. L., *Chem* **2020**, *6*, 538-538.

C-H functionalization of bicyclo[1.1.1]pentanes



Garlets, Z. J.; Sanders, J. N.; Malik, H.; Gampe, C.; Houk, K. N.; Davies, H. M. L., *Nat. Catal.* **2020**, *3*, 351-357.





2.3

Summary and Outlook

Summary: catalyst-controlled reactivity

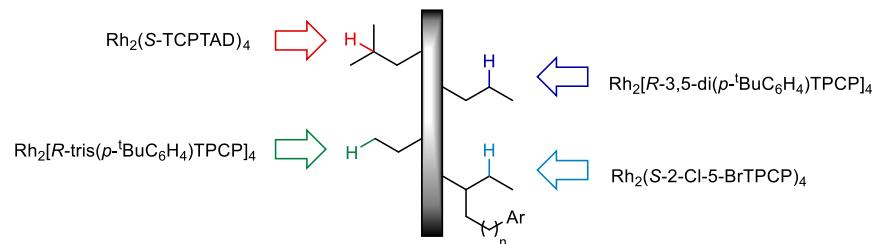
Seminal works

- Well-established catalysts
- Preliminary investigation on site- and stereoselectivity



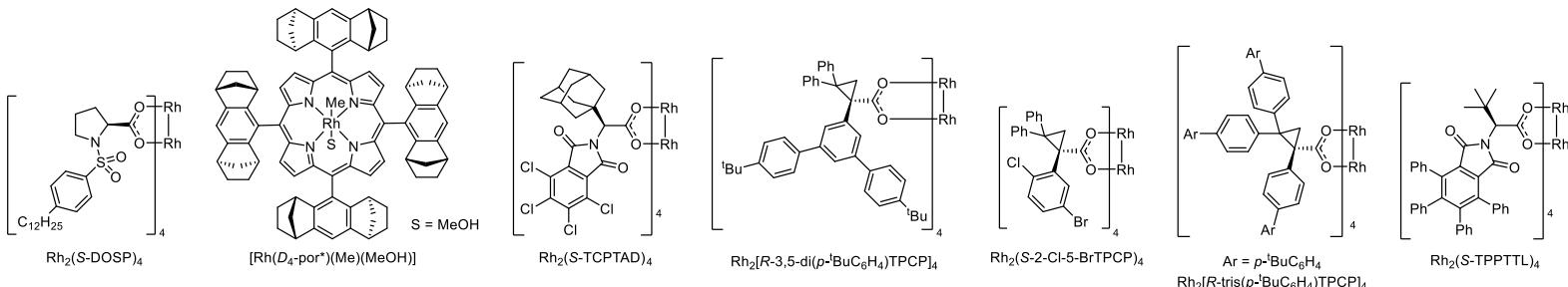
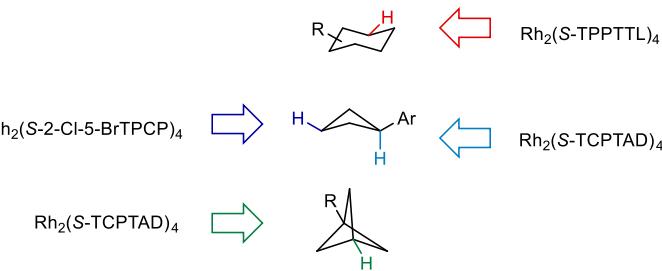
Comprehensive studies on acyclic systems

- Design of **new catalysts**
- C-H functionalization at **different sites**
- Excellent **enantio- and diastereoselectivity**



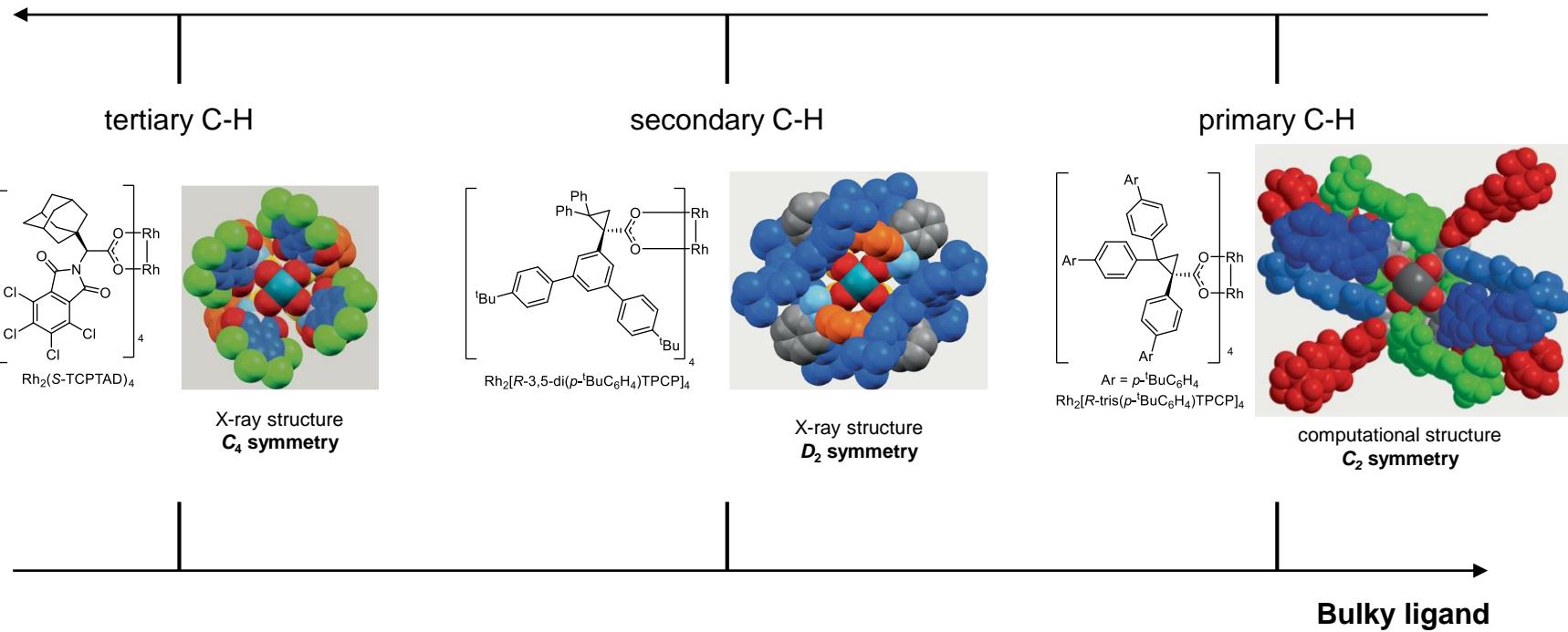
Further studies on cyclic systems

- Application of catalyst family
- Discrimination in **different conformation**

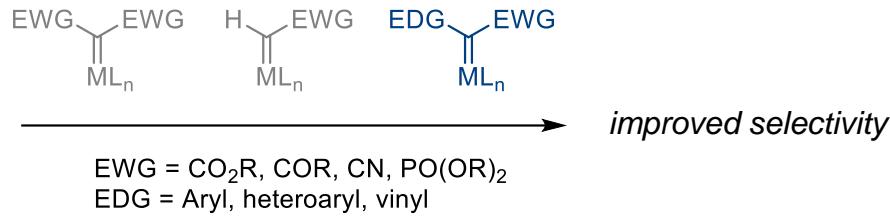


Summary: overview of the catalyst

Sterically hindered / electronically favoured site

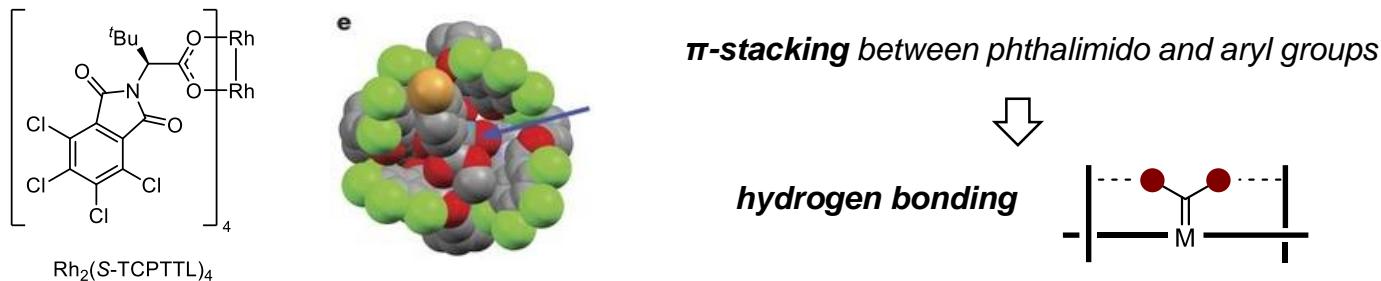


Donor-acceptor carbene



Outlook

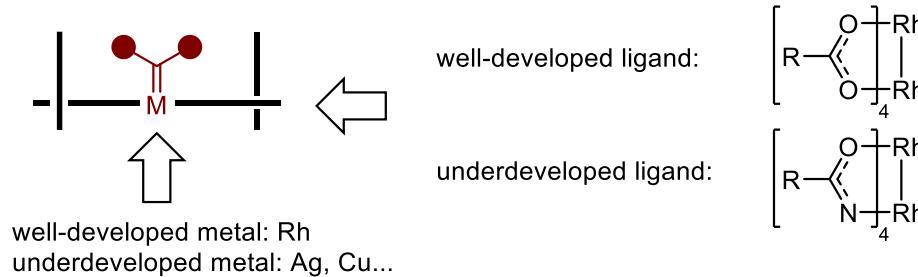
- Incorporation of other noncovalent interactions into the catalyst



- Application of other types of carbenoid



- Development of new catalysts



THANKS FOR YOUR ATTENTION!