



Carbon Isotope Exchange (CIE) with Labeled Carbon Source

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3 Summary and Outlook



Introduction

1 Introduction

Isotope labeling technology

Tracing organic compounds

- Disease diagnosis
- Agrochemicals
- Environmental fate
- Pharmacokinetic studies
-

Non-radioisotope labeling

MS/NMR detection

Radioisotope labeling

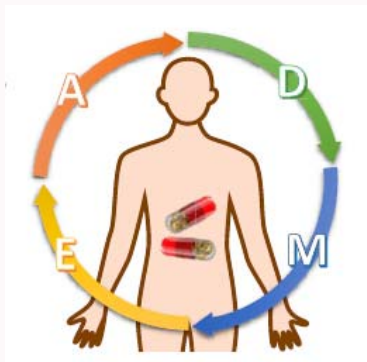
ADME studies,
PET imaging, etc

Absorption

Distribution

Metabolism


Excretion



Positron Emission Tomography (PET)

1 Introduction

^{11}C


- **radioactive** 
- $t_{1/2} = 20.33 \text{ min}$
- PET imaging

^{12}C

^{13}C

- non-radioactive
- MS/NMR detection

^{14}C

- **radioactive** 
- $t_{1/2} = 5730 \text{ years}$
- ADME studies

(Commonly used carbon isotopes)

Introduction of carbon isotopes is still challenging, **especially radioactive ^{11}C and ^{14}C .**

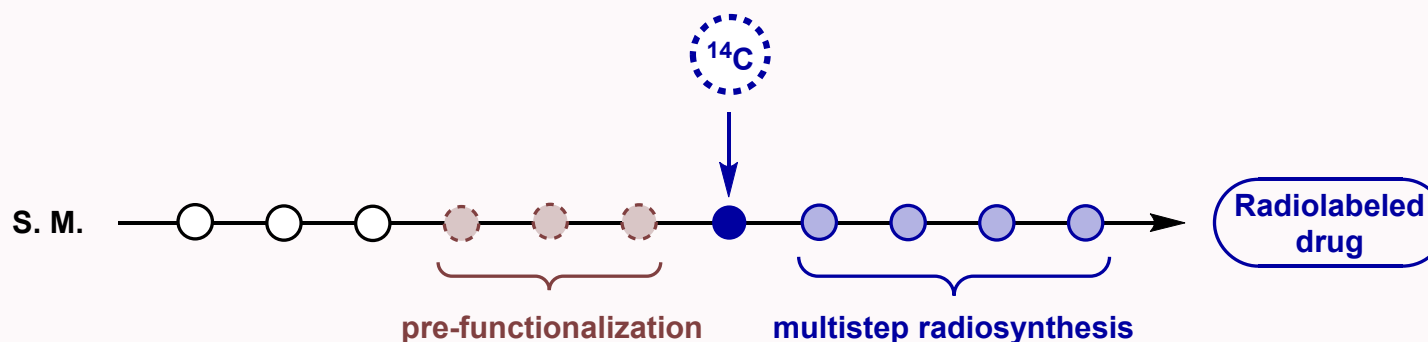
Cost and **radioactive waste** must be taken into account seriously.



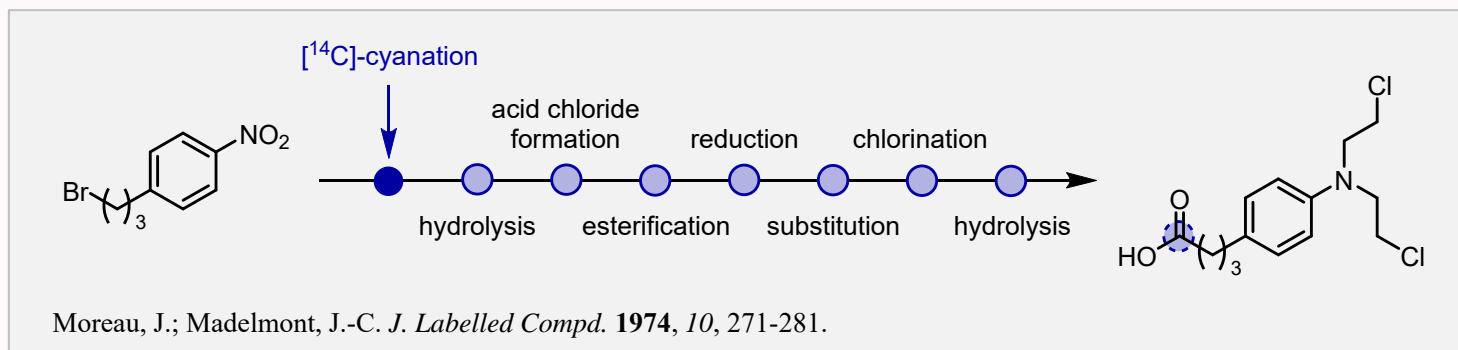
1 Introduction

Traditional approaches for the introduction of carbon isotopes Take [^{14}C] for example:

(1) *De novo* synthesis

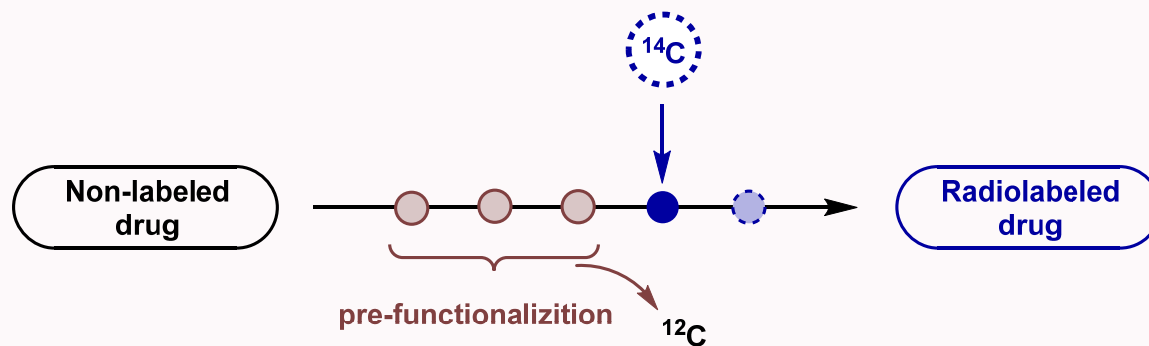


- Time-consuming
- High costs
- Generating large amounts of radioactive waste (extremely difficult and expensive to dispose of)

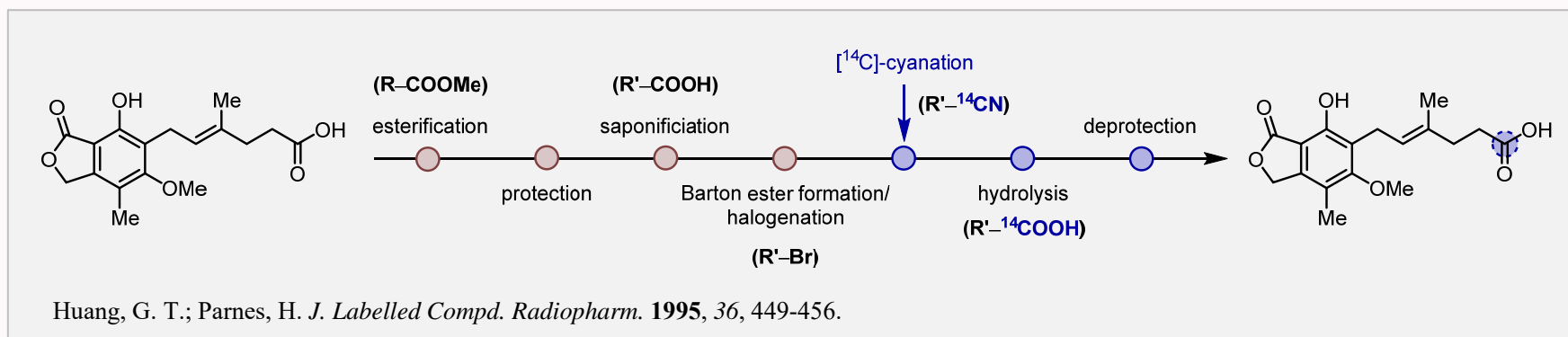


1 Introduction

(2) Degradation-reconstruction strategy



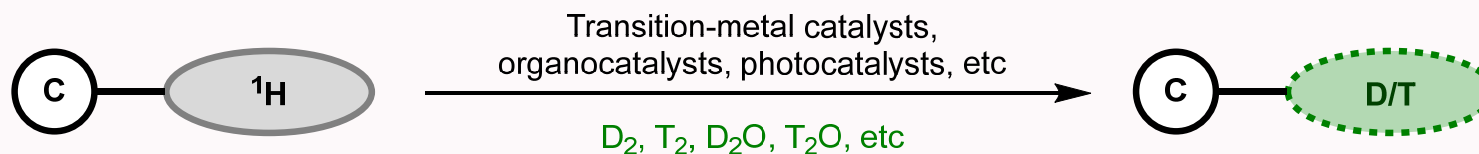
- Reducing the production of radioactive waste
- Time-consuming



1 Introduction

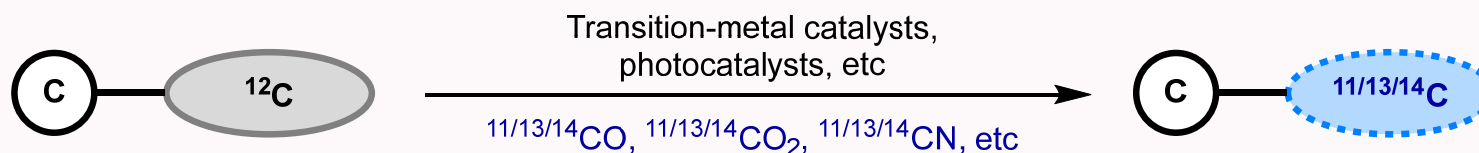
➤ Conceptual development: Carbon Isotope Exchange (CIE)

Hydrogen Isotope Exchange (HIE): well-established



- Hundreds of reports
- Broad scope of application
- Multiple incorporation of isotopes
- Mild reaction conditions

Carbon Isotope Exchange (CIE):



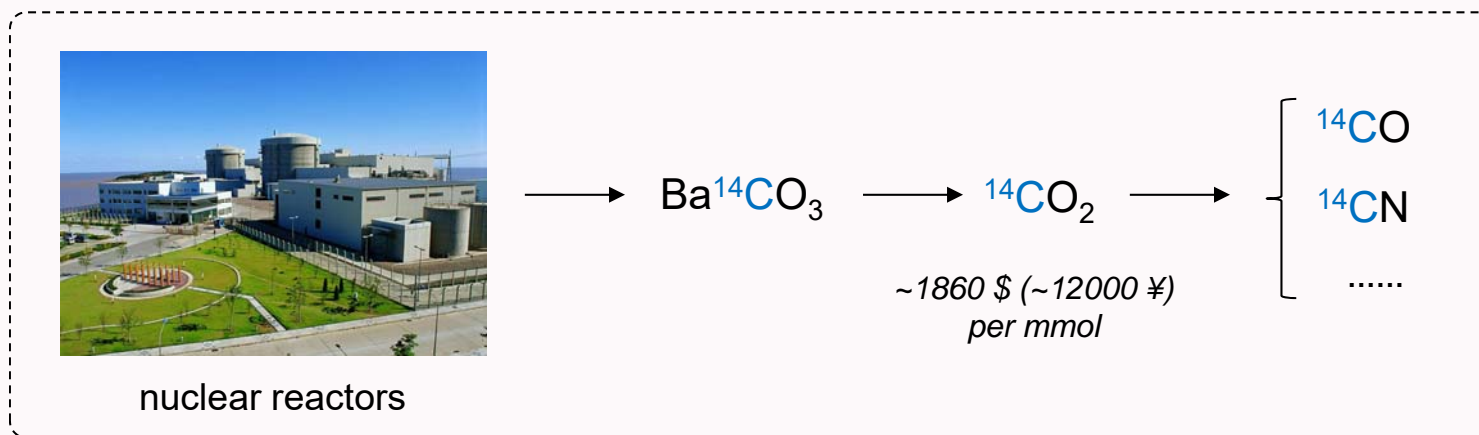
For selected reviews on HIE, see:

- (1) Atzrodt, J.*; Derdau, V.*; Fey, T.*; Zimmermann, J.* *Angew. Chem. Int. Ed.* **2007**, *46*, 7744-7765.
- (2) Atzrodt, J.*; Derdau, V.*; Kerr, W. J.*; Reid, M.* *Angew. Chem. Int. Ed.* **2018**, *57*, 3022-3047.

1 Introduction

➤ The challenges of CIE:

- (1) Carbon-carbon bond cleavage (*activation*)
- (2) Carbon-carbon bond reconstruction (*avoiding degradation*)
- (3) Limited collection of commercially available ^{14}C raw materials



Industrial production of [^{14}C] raw materials



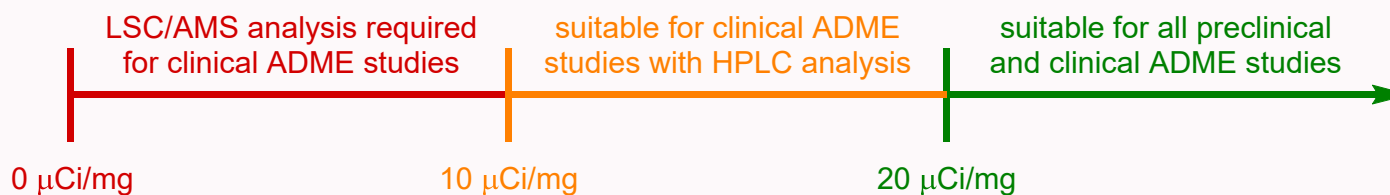
2

Carbon Isotope Exchange (CIE) with Labeled Carbon Source

2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source

Instructions before starting

■ For ^{14}C labeling:



■ For optimization and scope studies, [^{13}C] is usually used as a surrogate for precious [$^{11/14}\text{C}$].

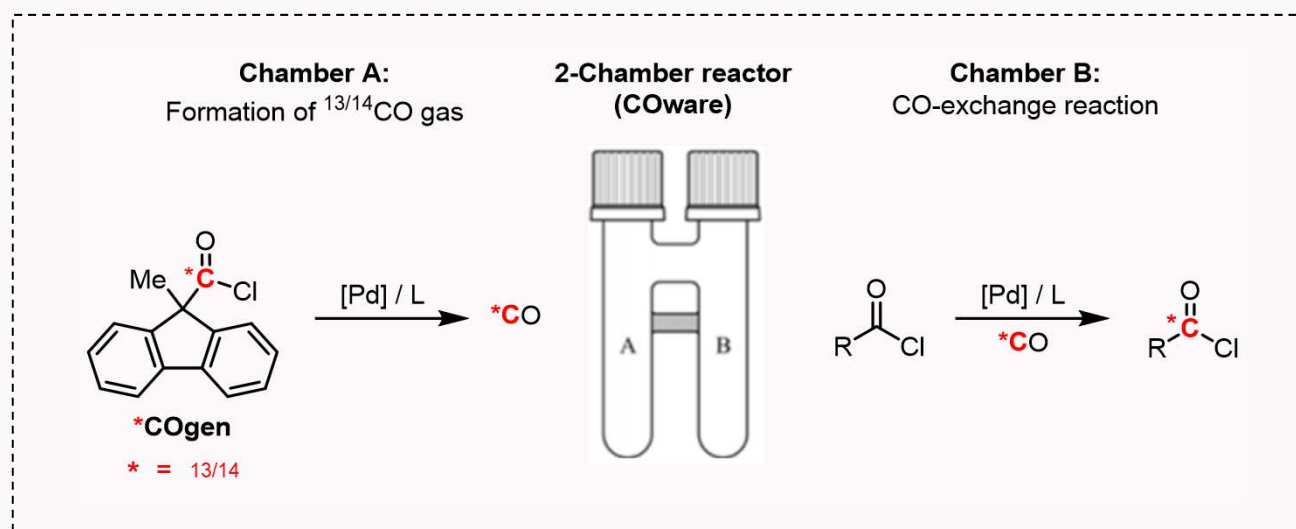
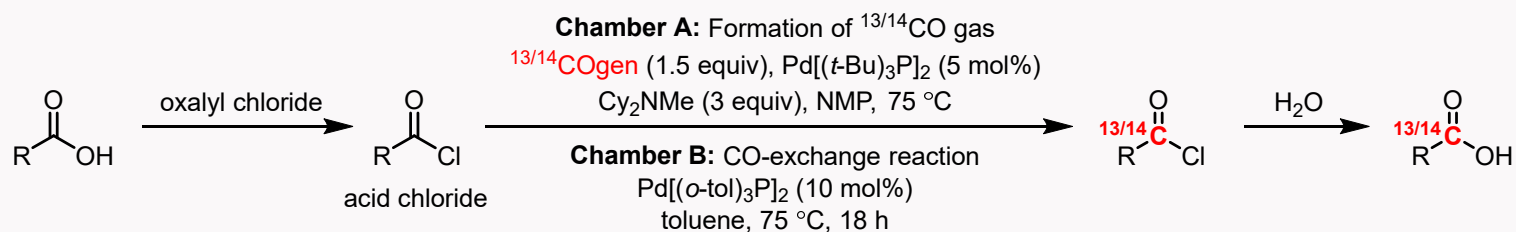
Activity: the number of atoms that a radioactive element or isotope decays per second. Units: bq, Ci (1 bq = 1 s⁻¹, 1 Ci = 3.7 × 10¹⁰ bq)
Specific activity (SA): radioactivity per unit mass of a stated element or compound. Units: mCi/mmol, μCi/mg, Mbq/mmol, etc.

2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source

2.1 CIE with Labeled CO

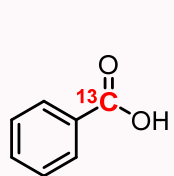


Gauthier, Jr., 2018

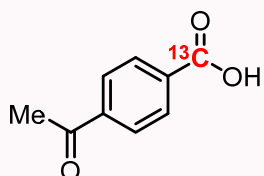


2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source

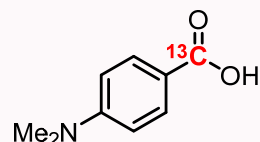
Selected examples:



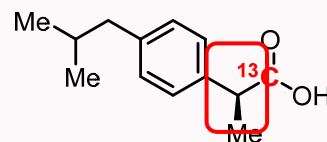
41% (88% yield)



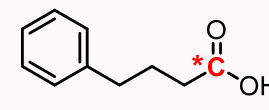
30% (90% yield)



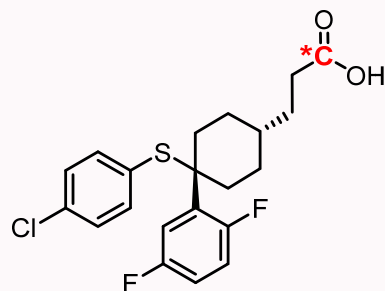
29% (82% yield)



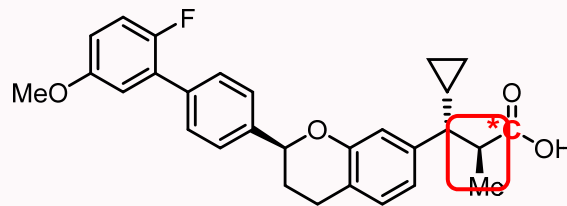
(S)-Ibuprofen (98% ee)
32% (86% yield)



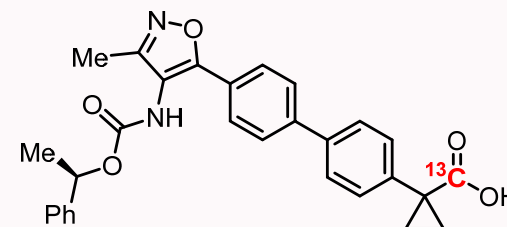
^{13}C : 48% (90% yield)
 ^{14}C : 27% (16% yield)
SA = 16.8 mCi/mmol



^{13}C : 53% (67% yield)
 ^{14}C : 37% (20% yield)
SA = 23.3 mCi/mmol



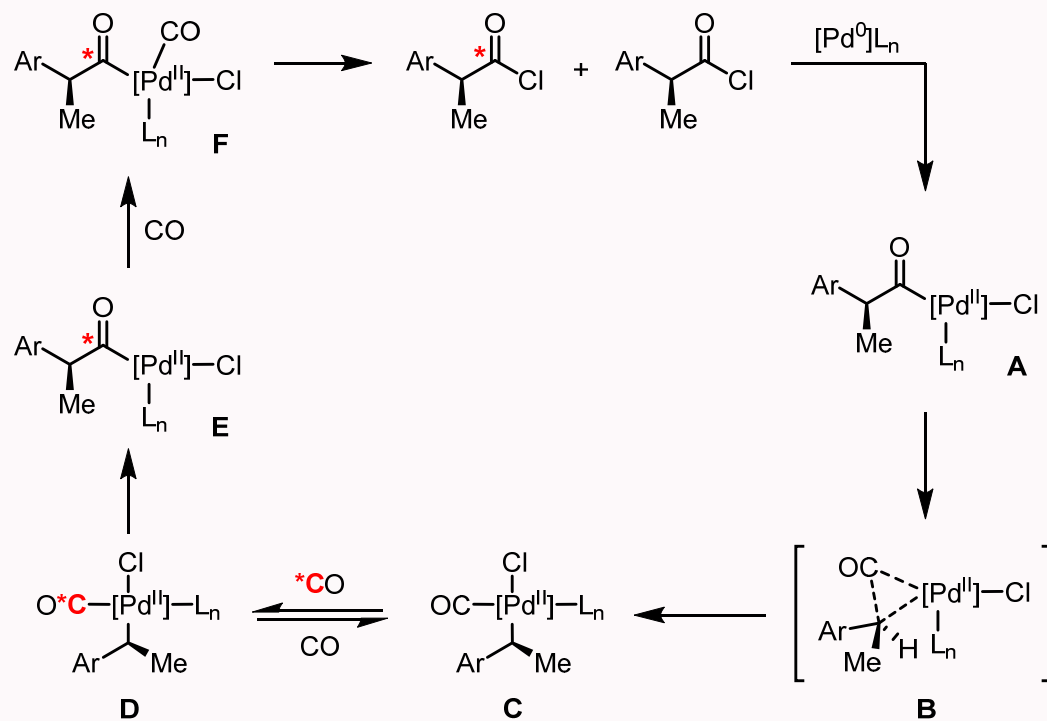
^{13}C : 43% (39% yield)
 ^{14}C : 40% (24% yield)
SA = 25.2 mCi/mmol



45% (30% yield)

2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source

The proposed mechanism:

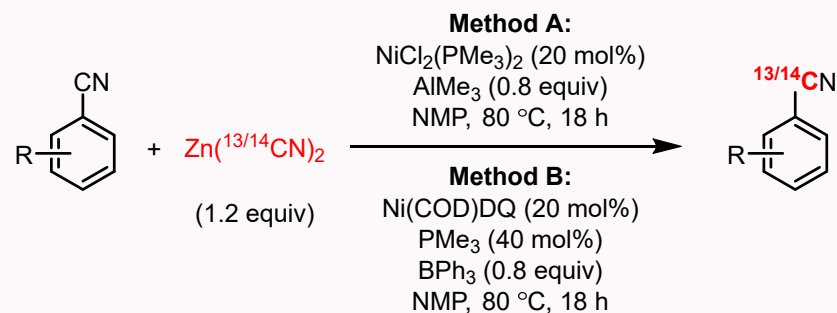


2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source

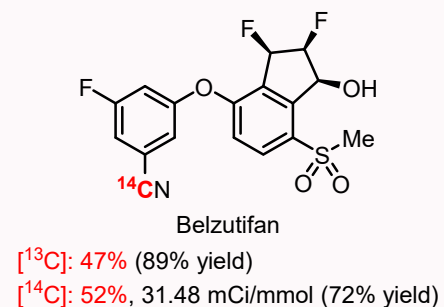
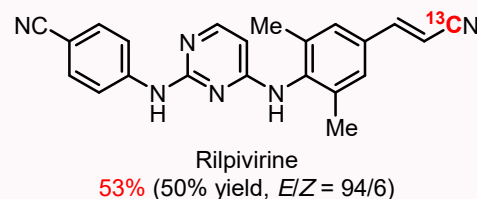
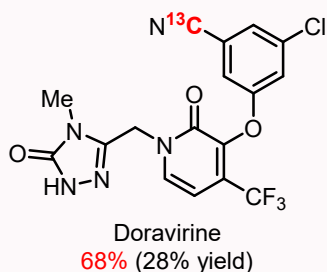
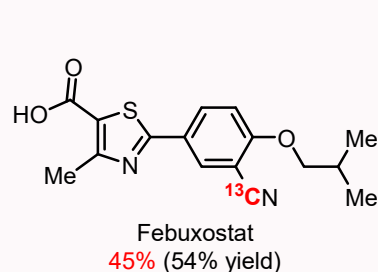
2.2 CIE with Labeled Cyanide



Reilly & Strotman, 2021



Selected examples:

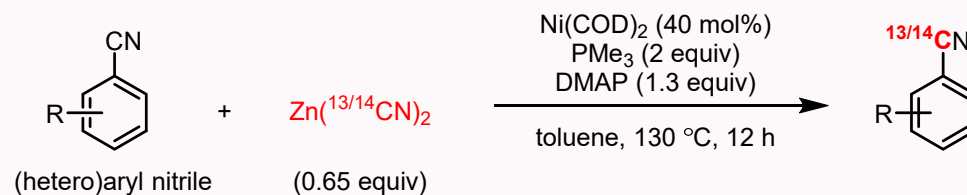


Non-labeled Belzutifan: 15 steps
No efficient route to label CN site

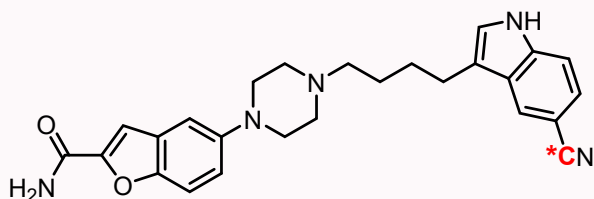
2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source



Audisio, 2021



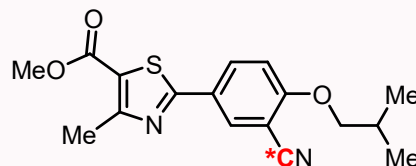
Selected examples:



Vilazodone

[${}^{13}\text{C}$]: 35% (91% yield)

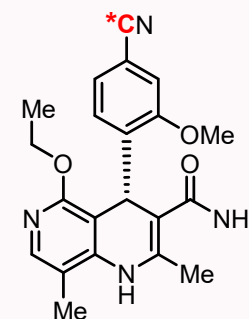
[${}^{14}\text{C}$]: 23%, 540 MBq/mmol (21% yield)
(14.59 mCi/mmol)



Febuxostat ester

[${}^{13}\text{C}$]: 50% (61% yield)

[${}^{14}\text{C}$]: 42%, 949 MBq/mmol (39% yield)
(25.65 mCi/mmol)



Finerenone

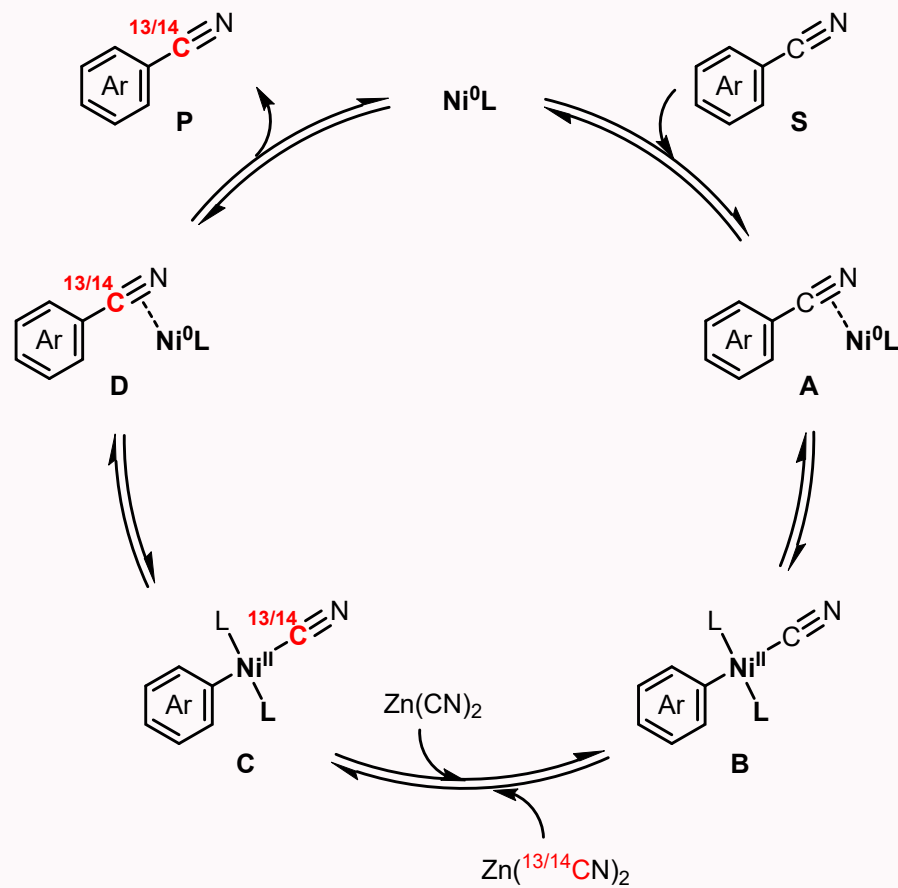
[${}^{13}\text{C}$]: 44% (74% yield)

[${}^{14}\text{C}$]: 45%, 1032 MBq/mmol (37% yield)
(27.89 mCi/mmol)

Previous report: multistep
117 MBq/mmol
(3.16 mCi/mmol)

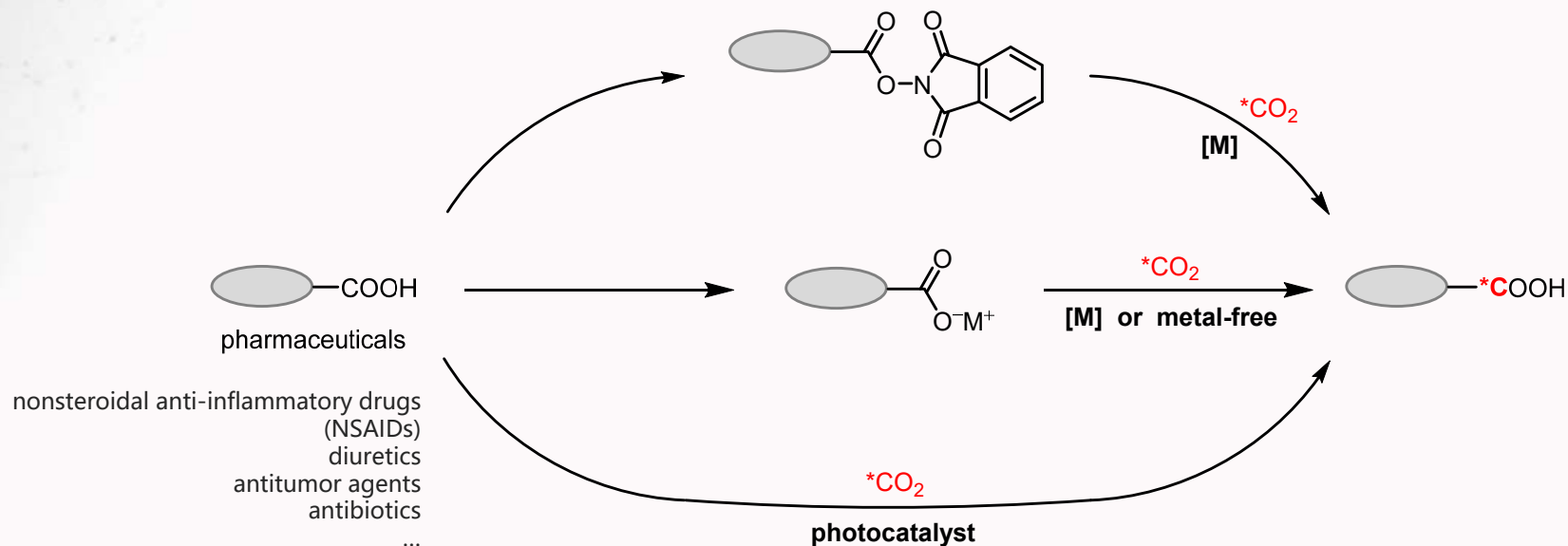
2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source

The proposed mechanism:



2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source

2.3 CIE with Labeled CO₂



Key: activation of carboxylic acids

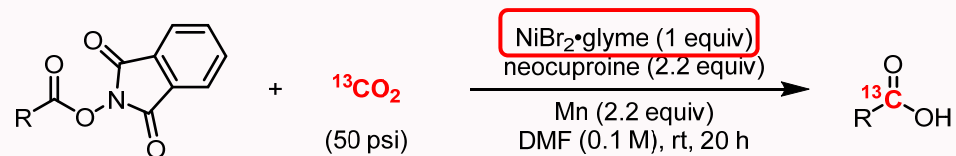
- Decarboxylative carboxylation of redox-active esters
- Decarboxylative carboxylation of carboxylates
- Organic photoredox catalysis

2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source

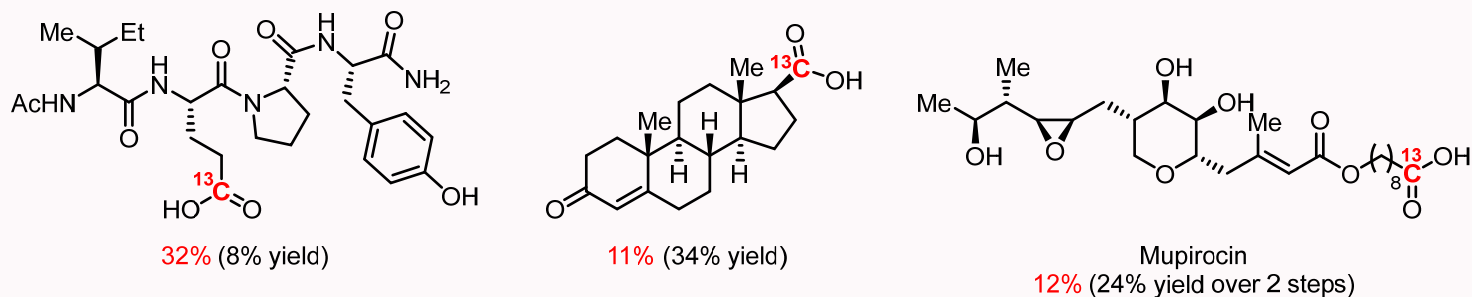
■ Decarboxylative carboxylation of redox-active esters



Baran, 2019



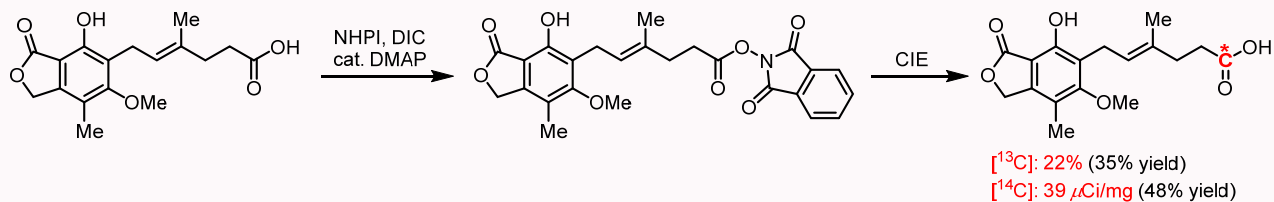
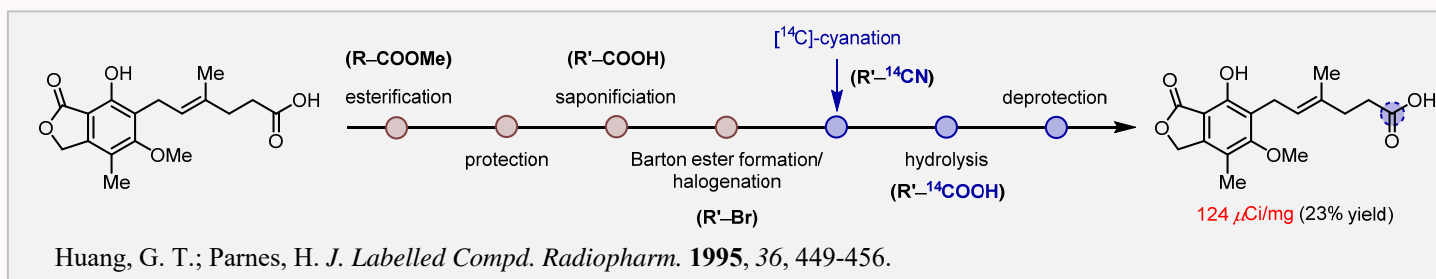
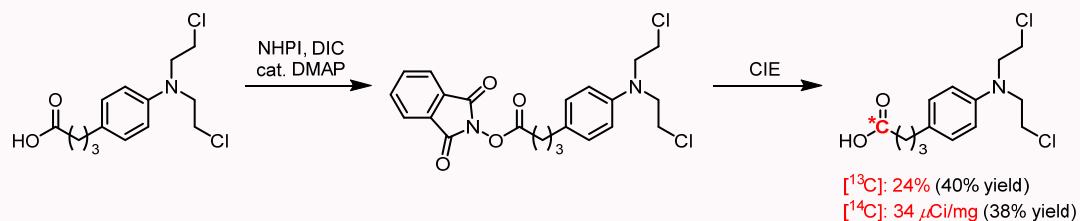
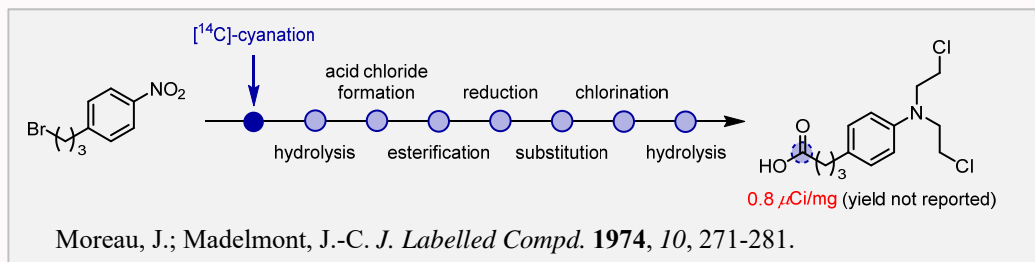
Selected examples:



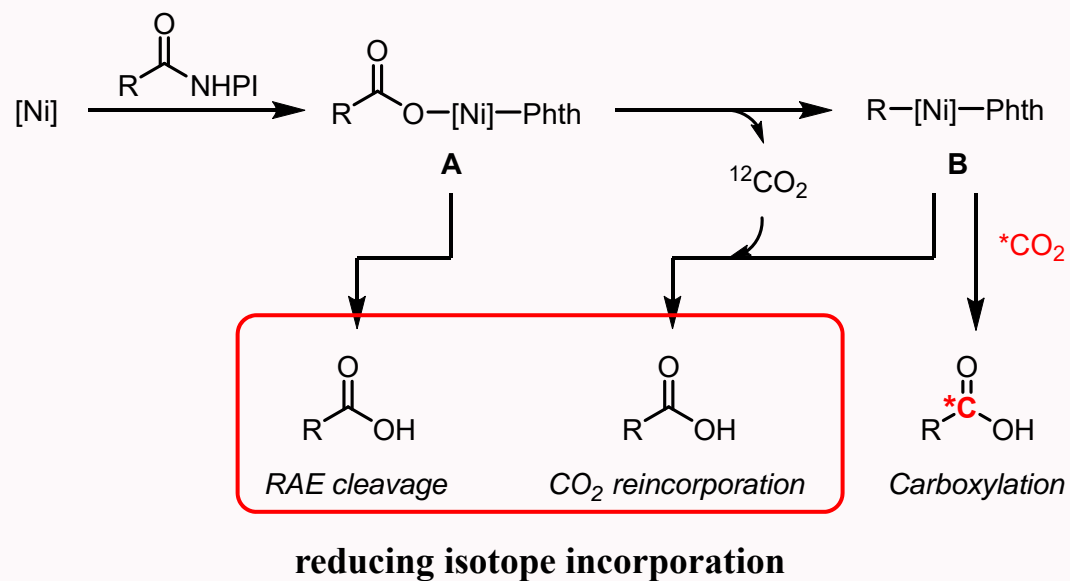
(1 psi = 6.895 kPa, 50 psi = 344.7 kPa = 3.40 atm)

Baran, P. S.* et al. *J. Am. Chem. Soc.* **2019**, *141*, 774-779.

2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source



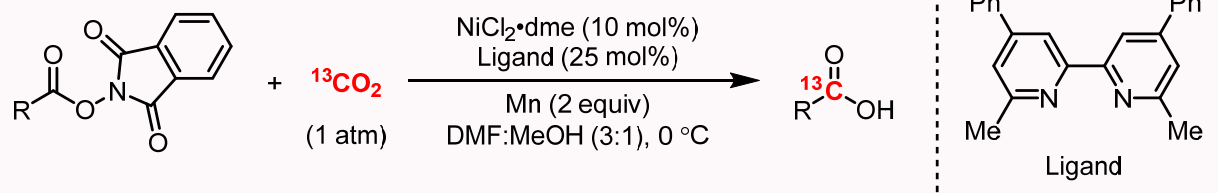
2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source



2 Carbon Isotope Exchange (CIE) with Labeled Carbon Source



Martin, 2019



Selected examples:

