



金属-氮宾插入烯烃下的不对称氮杂环丙烷化反应

Enantioselective Azirdination By Addition of Metal–Nitrenes to Olefins

汇报人：张思淼 导师：麻生明 教授

2025 年 09 月 26 日

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2 Enantioselective Azirdination By Addition of Metal–Nitrenes to Olefins

2.1 Metal–Nitrenes from Iminoiodinanes

2.2 Metal–Nitrenes from Amides Derivatives

2.3 Metal–Nitrenes from Hydroxylamine Derivatives

2.4 Metal–Nitrenes from Azides

3 Summary and outlook

1 Background

2 Enantioselective Azirdination By Addition of Metal–Nitrenes to Olefins

2.1 Metal–Nitrenes from Iminoiodinanes

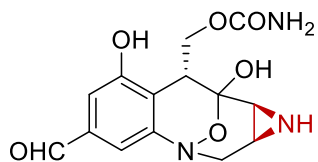
2.2 Metal–Nitrenes from Amides Derivatives

2.3 Metal–Nitrenes from Hydroxylamine Derivatives

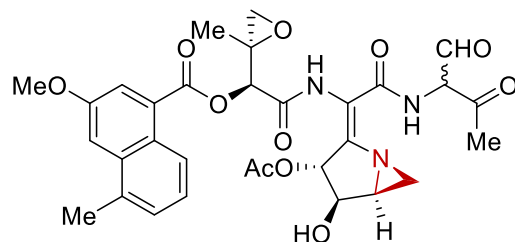
2.4 Metal–Nitrenes from Azides

3 Summary and outlook

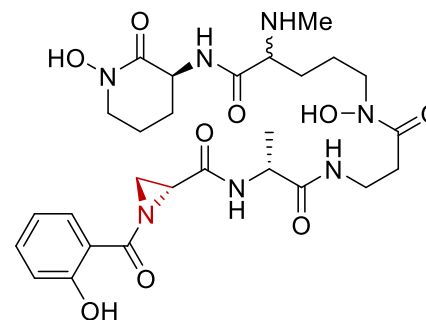
1 Background



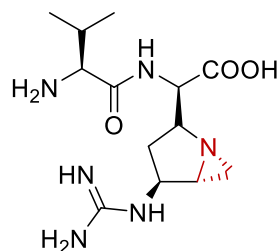
FR-9000482
潜在抗肿瘤活性



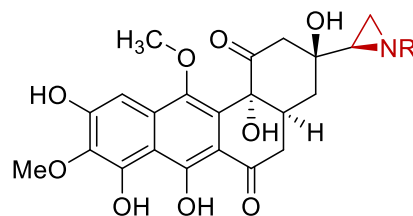
Azinomycin B
潜在抗肿瘤活性



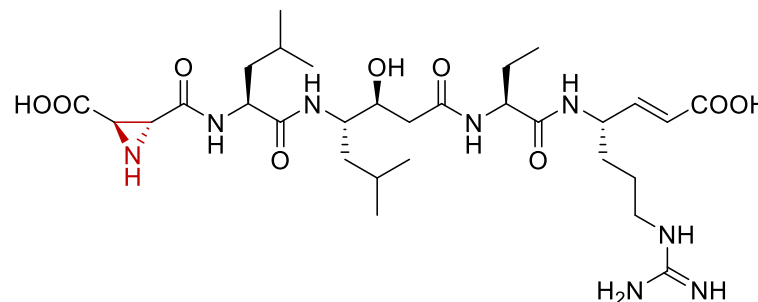
Madurastatin A1
抗菌活性



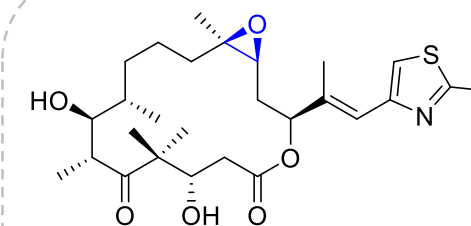
Ficellomycin
抗革兰氏阳性菌活性



Azicemicin A: R = H;
Azicemicin B: R = Me
抗革兰氏阳性菌活性

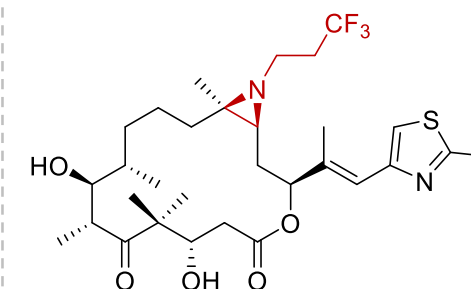


Miraziridine
Cathepsin B抑制剂



Epothilone B

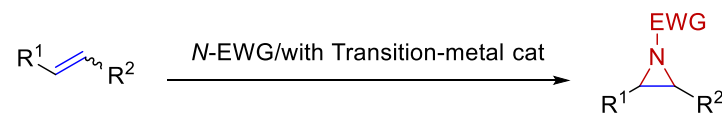
$IC_{50} = 1.27 \text{ nM (SKOV3)}$



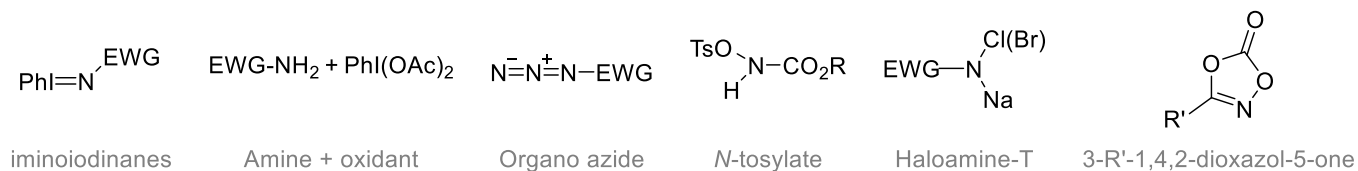
$IC_{50} = 0.01 \text{ nM (SKOV3)}$

Komaki, H. et al. *J. Antibiot.* **2004**, 57, 125–135. Furmeier, S. et al. *Eur. J. Org. Chem.* **2003**, 2003, 649–659.
Takeuchi, T. et al. *J. Antibiot.* **1995**, 48, 1148–1152. Takeuchi, T. et al. *J. Antibiot.* **1995**, 48, 217–221.
Genisson, Y. et al. *Tetrahedron.* **2011**, 67, 2570–2578. Dembitsky, V. M. et al. *Eur. J. Med. Chem.* **2009**, 44, 3373–3387.
Nicolaou, K. C. et al. *J. Org. Chem.* **2020**, 85, 2865–2917.

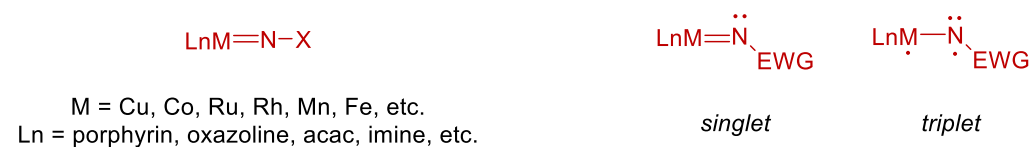
1 Background



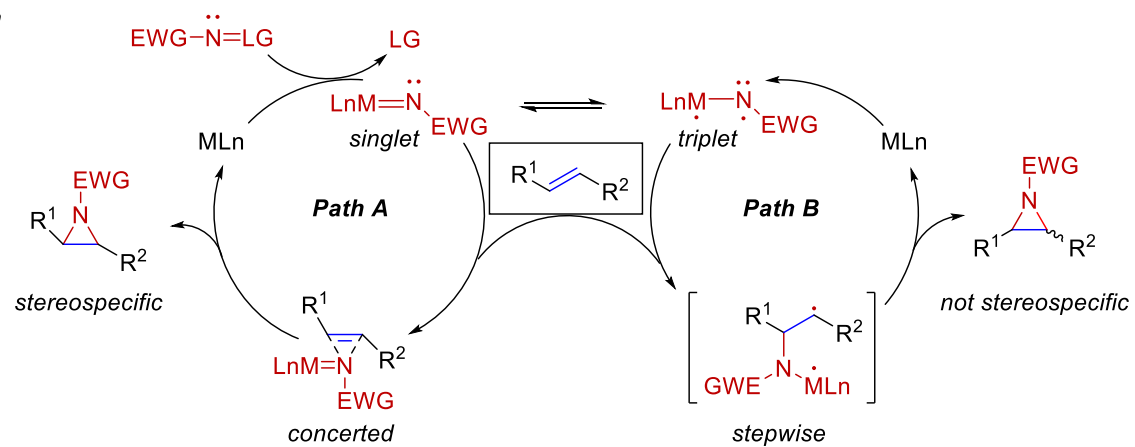
Metal-nitrenoid sources



Metal-nitrenoid species



Mechanism



1 Background

2 Enantioselective Azirdination By Addition of Metal–Nitrenes to Olefins

2.1 Metal–Nitrenes from Iminoiodinanes

2.2 Metal–Nitrenes from Amides Derivatives

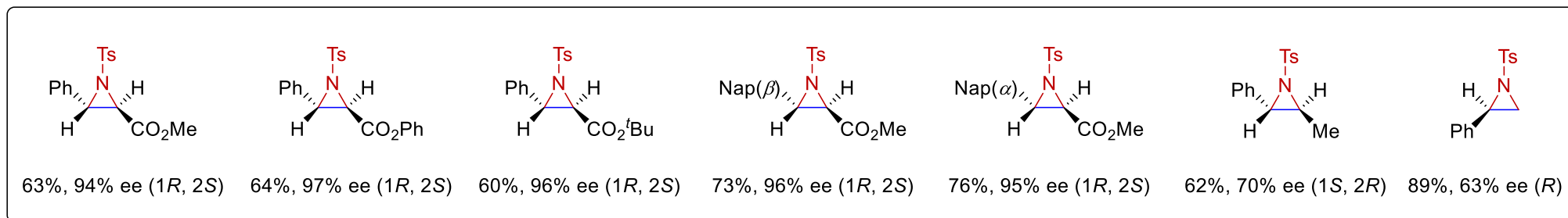
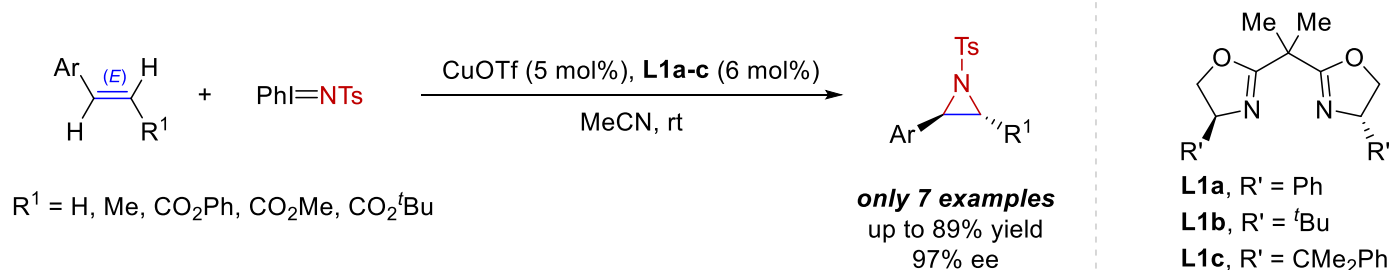
2.3 Metal–Nitrenes from Hydroxylamine Derivatives

2.4 Metal–Nitrenes from Azides

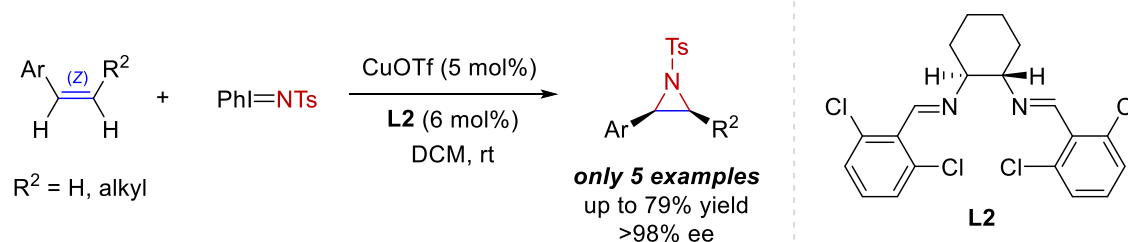
3 Summary and outlook

2.1 Metal–Nitrenes from Iminoiodinanes

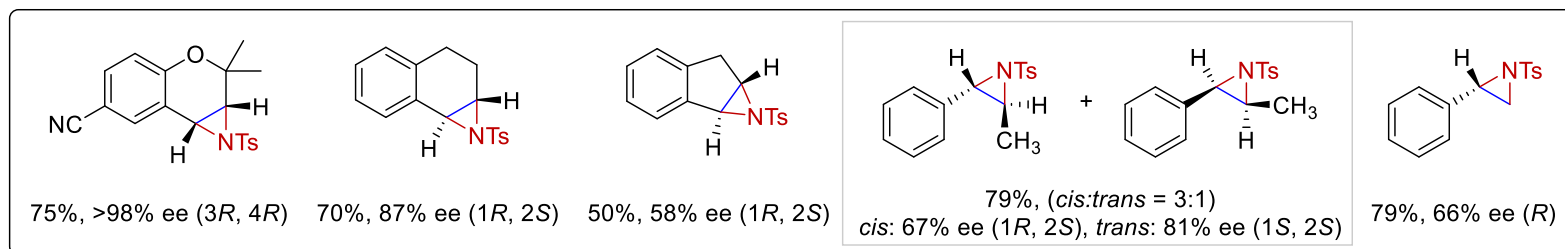
a. Evans (1993)



b. Jacobsen (1993)

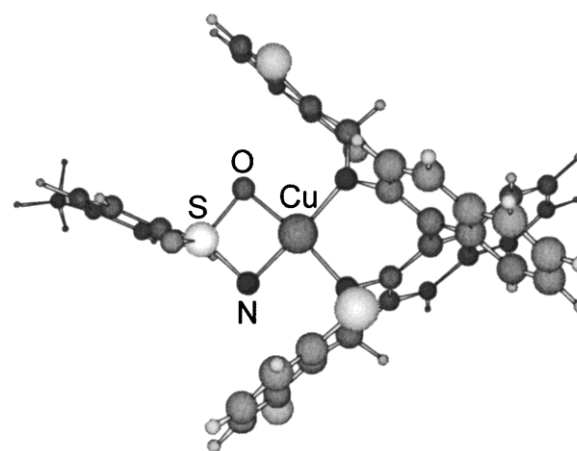
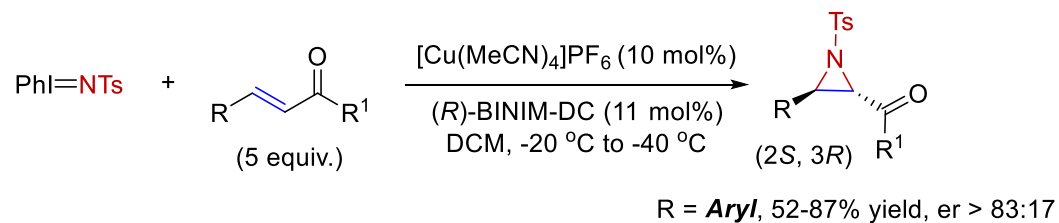


首次实现氮杂环丙烷
骨架的不对称构建

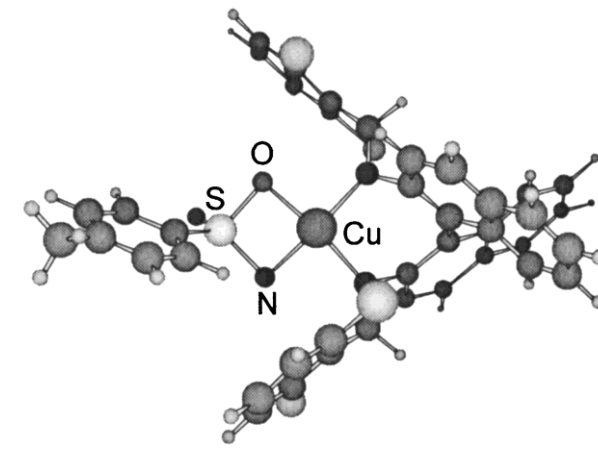


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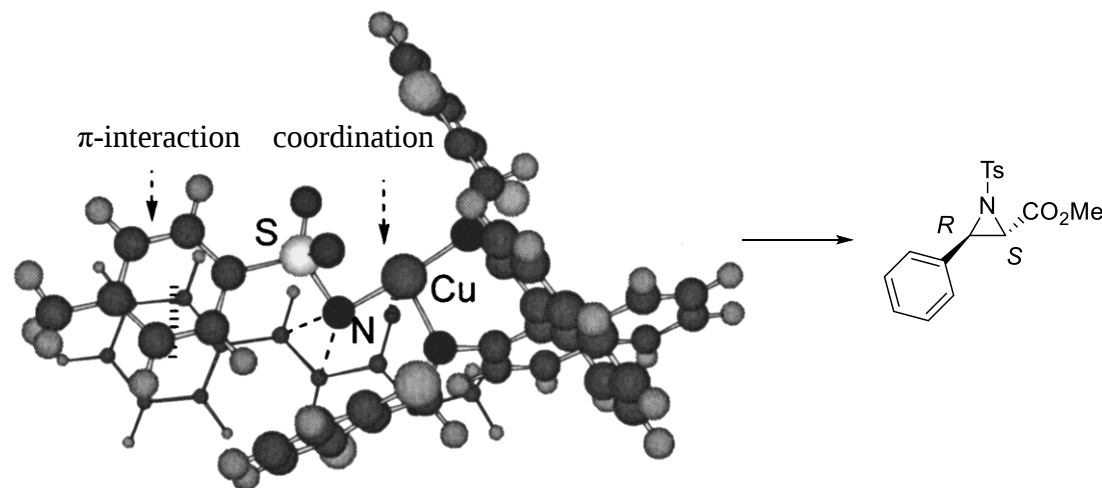
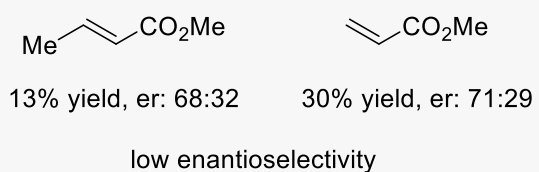
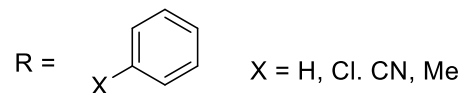
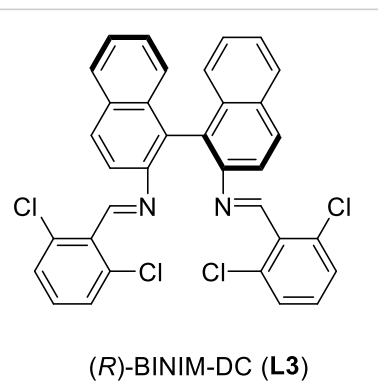
Suga (2003)



Intermediate A

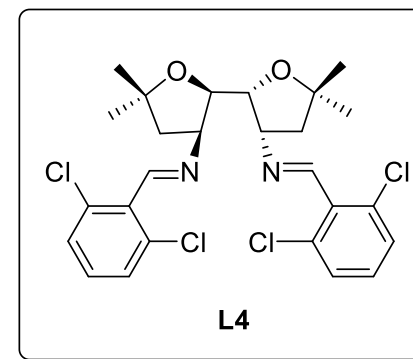
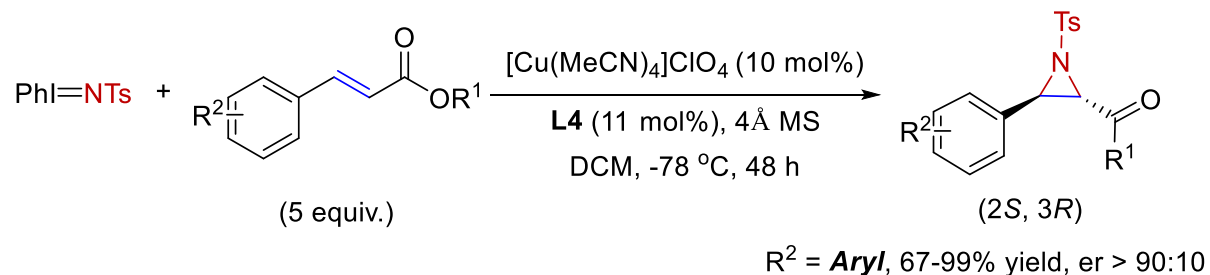


Intermediate B

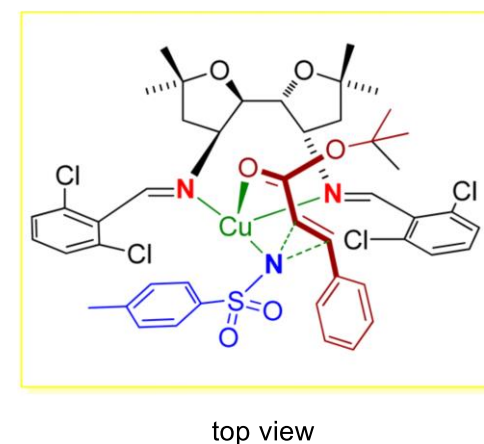


2.1 Metal–Nitrenes from Iminoiodinanes

Ding (2006)



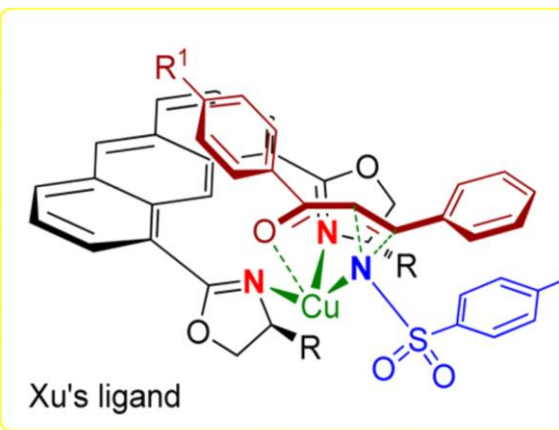
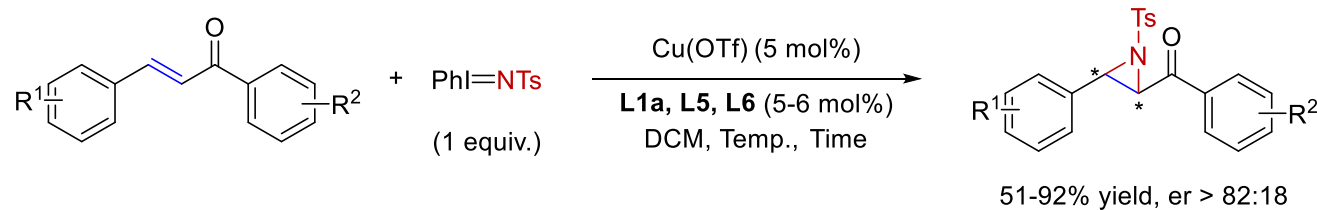
Entry	R ¹	R ²	Yield (%)	ee (%)	Config.
1	Me	H	97	87.6	2S, 3R (-)
2	Ph	H	96	87.1	2S, 3R (-)
3	<i>t</i> Bu	H	99	> 99	2S, 3R (-)
4	<i>t</i> Bu	4-F	99	97.8	(-)
5	<i>t</i> Bu	4-Cl	97	98.1	(-)
6	<i>t</i> Bu	4-Br	97	98.3	2S, 3R (-)
7	<i>t</i> Bu	4-Me	86	94.4	(-)
8	<i>t</i> Bu	4-MeO	95	80.1	(+)
9	<i>t</i> Bu	2-NO ₂	63	98.6	(+)
10	<i>t</i> Bu	2,3-(MeO) ₂	67	97.0	(+)



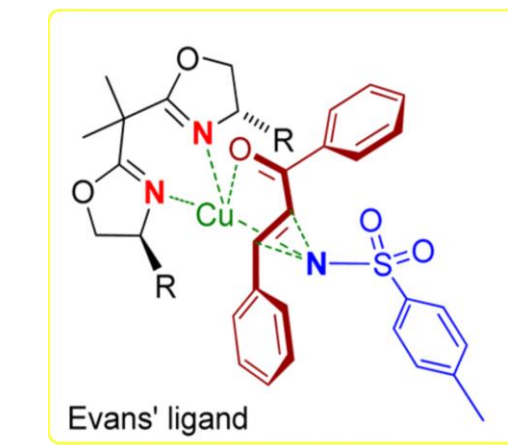
Entries 1-3 通过对照文献旋光度确定绝对构型；Entry 6 通过单晶确定绝对构型，表格中其他产物均为未知产物，未明确给出绝对构型，仅测定旋光

2.1 Metal–Nitrenes from Iminoiodinanes

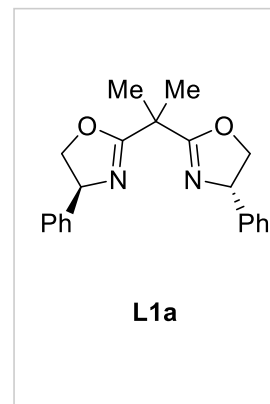
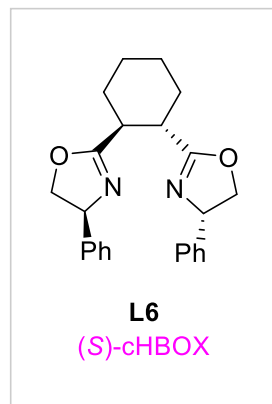
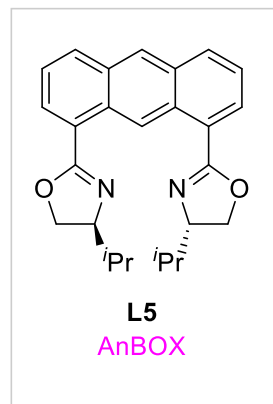
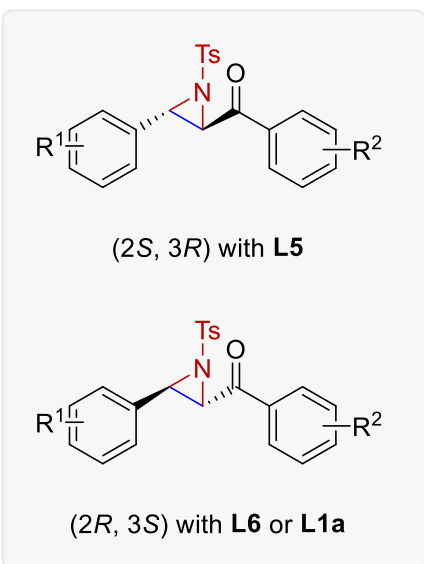
Xu (2005 and 2007)



底物苯环和配体骨架
存在相互作用
底物苯环靠近配体



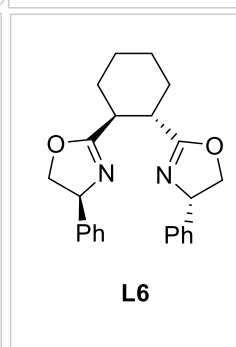
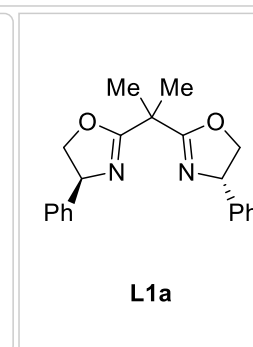
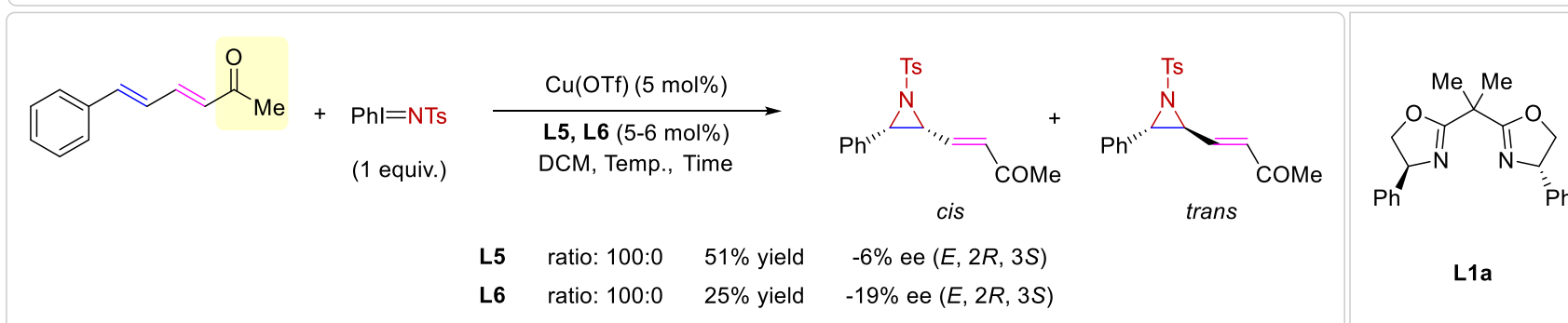
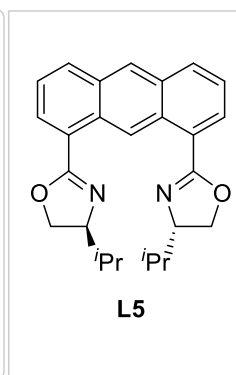
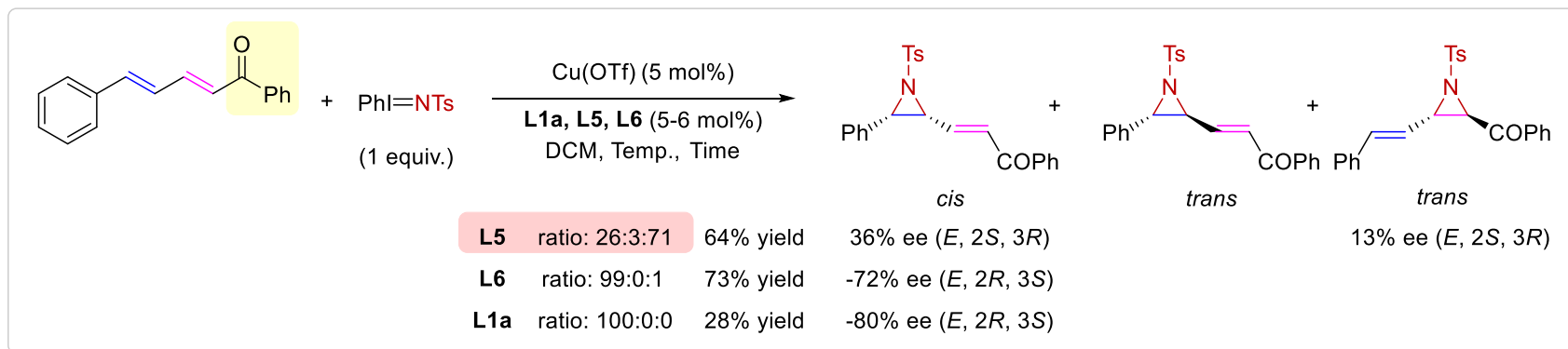
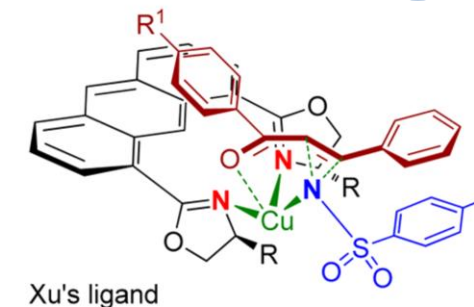
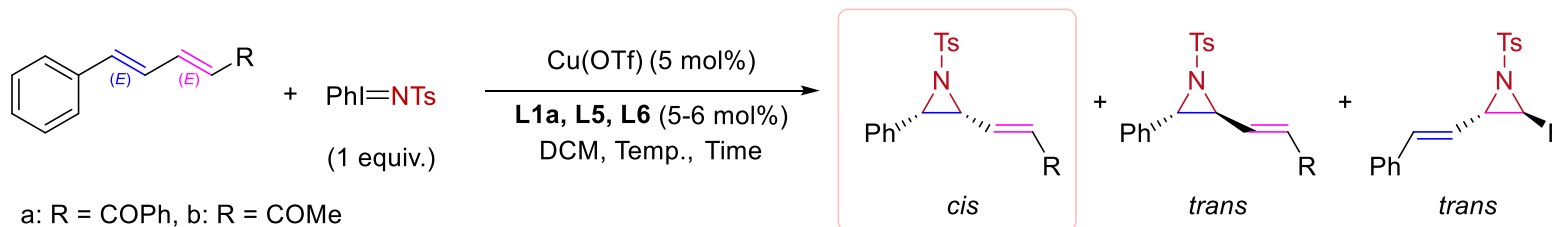
受到位阻影响
底物苯环远离配体



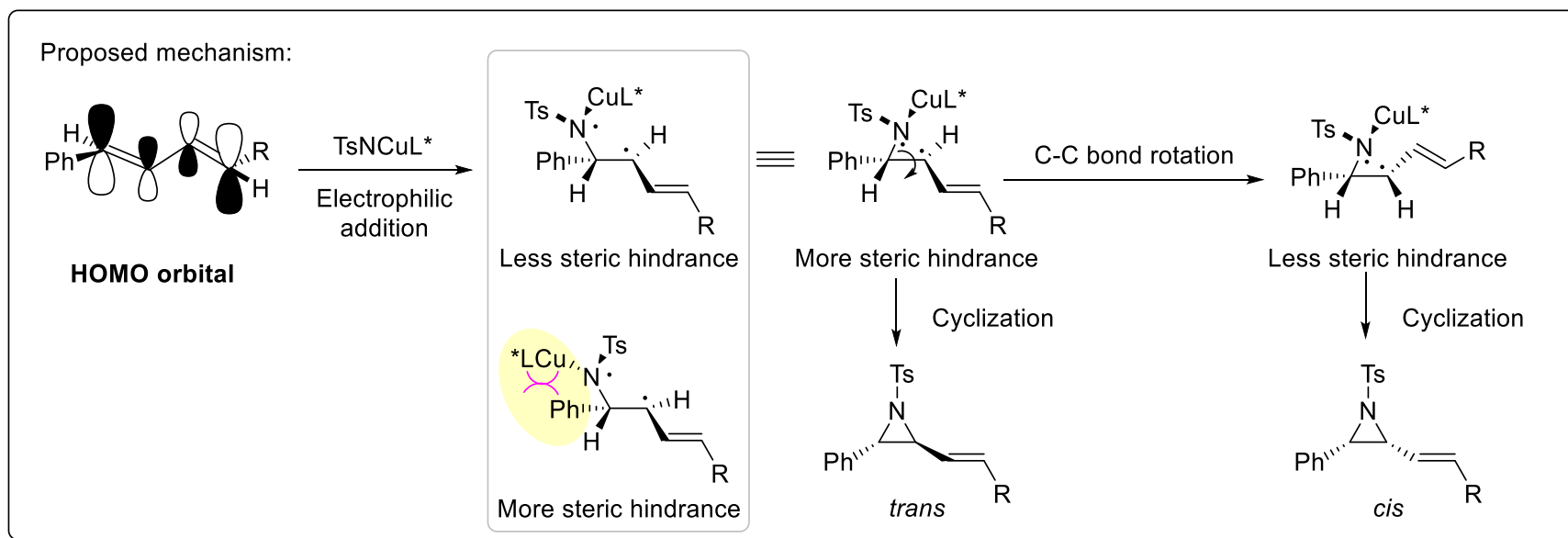
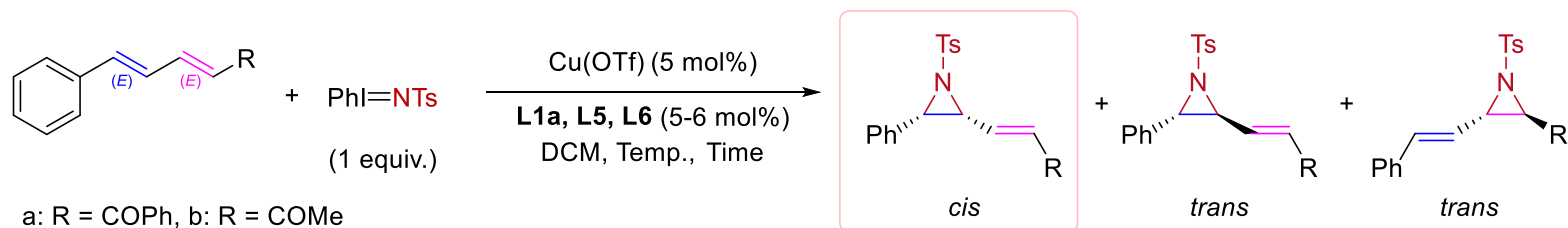
2.1 Metal–Nitrenes from Iminoiodinanes



Xu (2006)



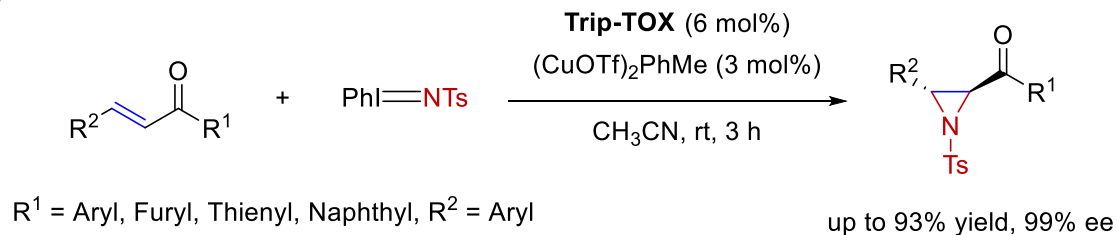
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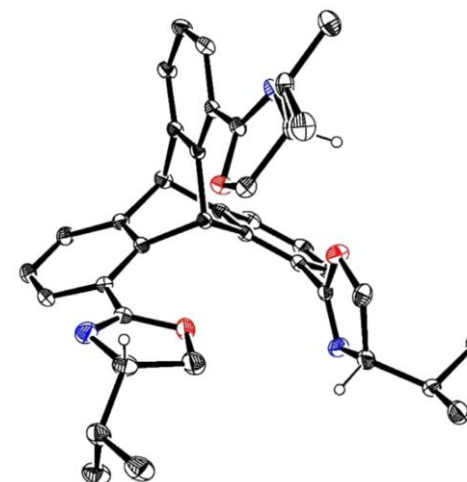
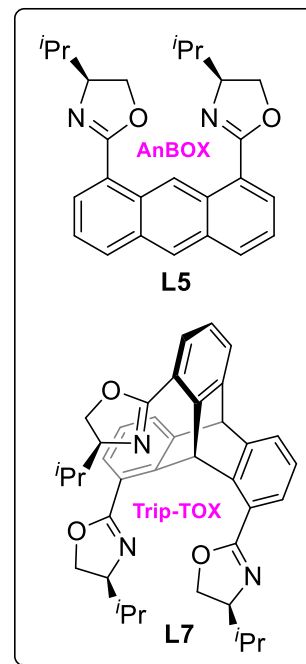
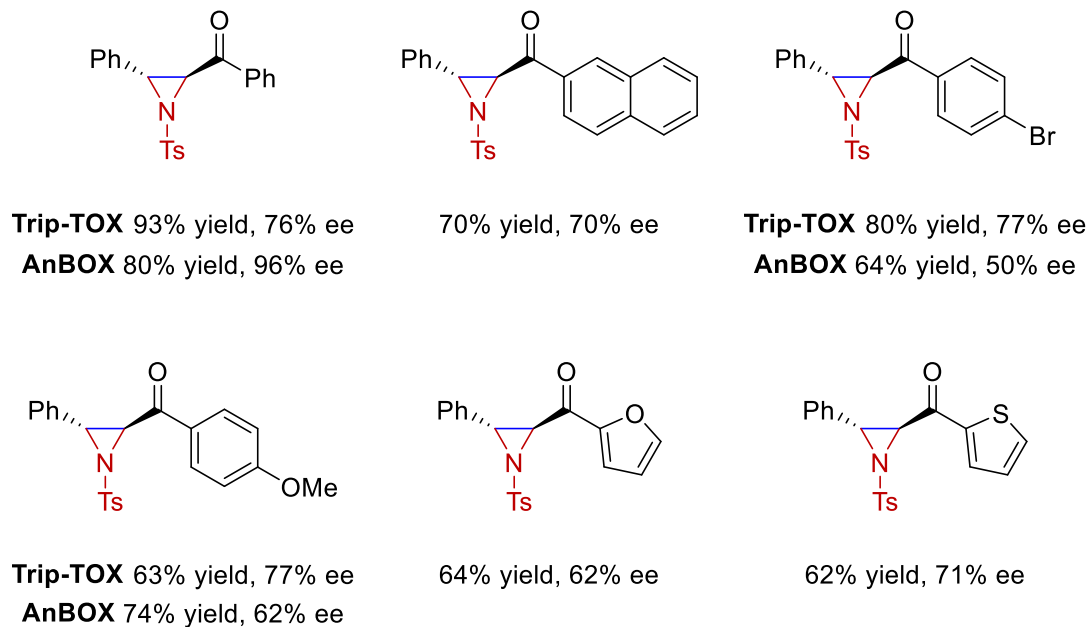
2.1 Metal–Nitrenes from Iminoiodinanes



Gelman (2021)

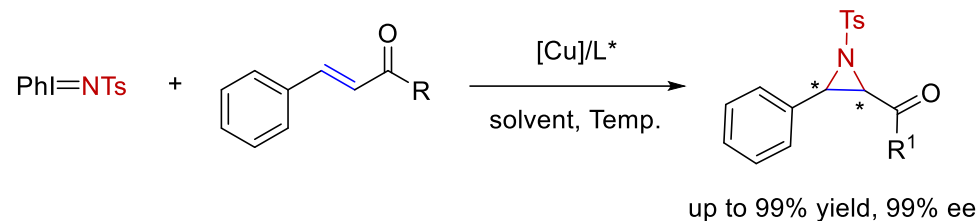


C_3 对称性进一步
降低底物与催化剂空间自由度



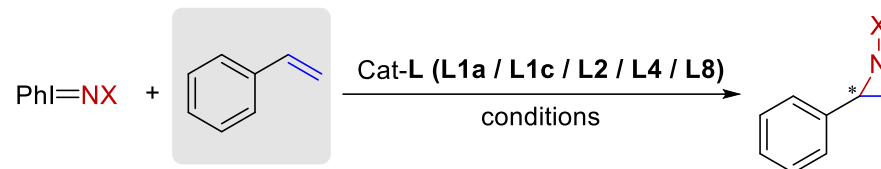
2.1 Metal–Nitrenes from Iminoiodinanes

查尔酮类底物

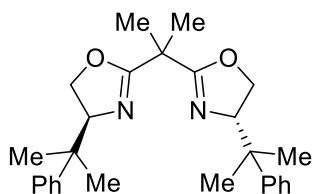


活性高、通过底物配位、
手性配体修饰控制立体选择性

更具挑战性的苯乙烯底物



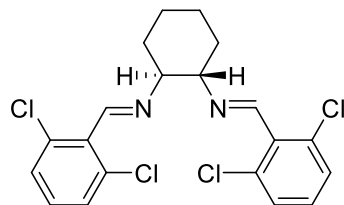
Evans (1993)



L1c

Cat: CuOTf (5 mol%), L1c (6 mol%)
styrene, 0 °C, 2.5 h
X = Ts, 89%, er: 81.5:18.5
(R) configuration

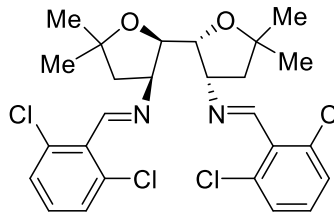
Jacobsen (1993)



L2

Cat: CuOTf (10 mol%), L2 (11 mol%)
DCM, -78 °C
X = Ts, 79%, er: 83:17
(R) configuration

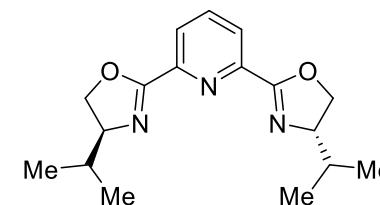
Ding (2006)



L4

Cat: [Cu(MeCN)₄]ClO₄ (10 mol%), L4 (5 mol%)
4 Å MS, DCM, -78 °C, 48 h
X = Ts, 99%, er: 64:36
(S) configuration

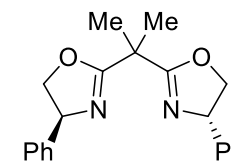
Bolm (2008)



L8

Cat: Fe(OTf)₂ (5 mol%), L5 (30 mol%)
4 Å MS, MeCN, rt, 1 h
X = Ts, 72%, er: 70:30
(R) configuration

Hutchings (2003)



L1a

Cat: Zeolite CuHY, L1a (7 mol%)
4 Å MS, MeCN, rt, 1 h
X = Ts, 78%, er: 88:12
X = Ns, 78%, er: 92:8

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2.1 Metal–Nitrenes from Iminoiodinanes

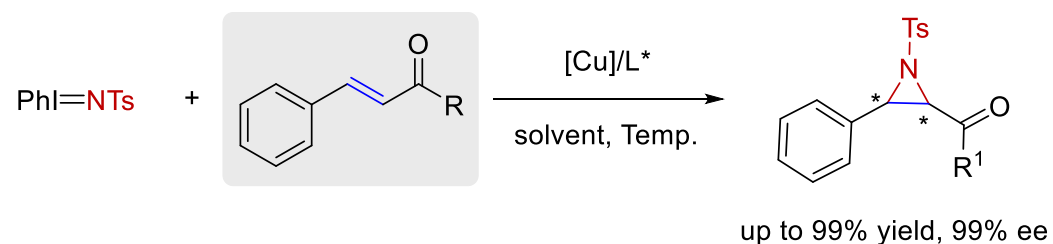
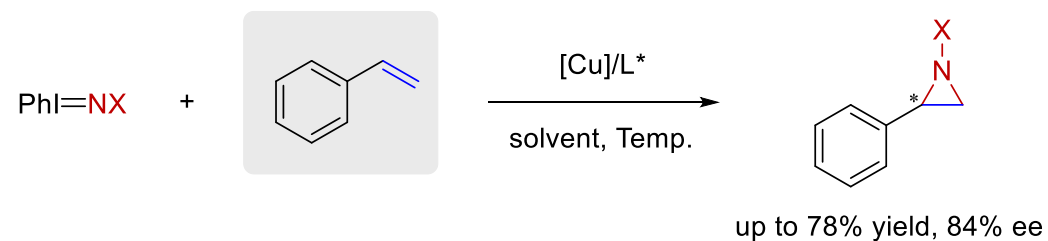
2.2 Metal–Nitrenes from Amides Derivatives

2.3 Metal–Nitrenes from Hydroxylamine Derivatives

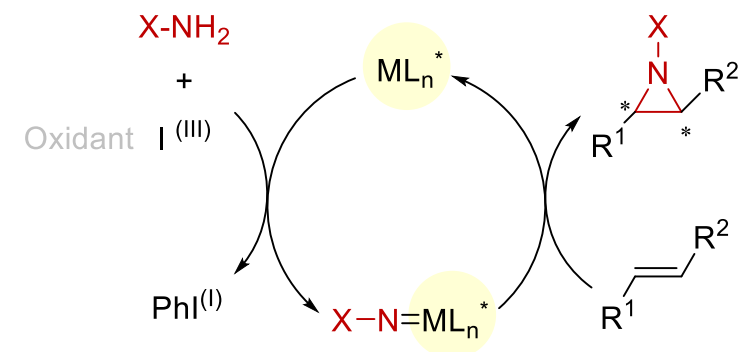
2.4 Metal–Nitrenes from Azides

3 Summary and outlook

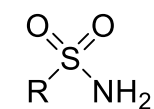
2.2 Metal–Nitrenes from Amides Derivatives



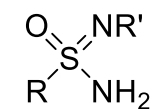
底物类型相对单一、亚胺类前体不稳定



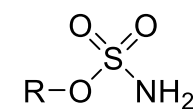
Nitrogen donors X-NH_2



sulfonamide



sulfonimidamide



sulfamate

M = Cu, Rh, Ag

Oxidant: $\text{PhI}(\text{OAc})_2$, $\text{PhI}=\text{O}$

CC(C)(C)C(=O)O[C@H](C=C)Nc1ccccc1
 $\xrightarrow[\text{MeCN, rt, -20 } ^\circ\text{C}]{\text{PhI=O (1.2 equiv), SesNH}_2 \text{ (1.2 equiv), Cu(MeCN)}_4\text{PF}_6 \text{ (25 mol\%), L1b (35 mol\%)}$
CC(C)(C)C(=O)O[C@H](C=C)Nc1ccccc1
+
CC(C)(C)C(=O)O[C@H](C=C)Nc1ccccc1

16% yield
 dr (*R, R*)/(*R, S*): up to 75/25

SesNH₂ = CC(C)(C)S(=O)(=O)N

L1b

Reaction scheme showing the synthesis of spirocyclic sulfonamides:

Starting material: R-CH=CH-CH2-CH2-SO2NH2 (where R = H, Alkyl, Ph)

Reagents and Conditions:

- PhI=O (1.5 equiv)
- Cu(MeCN)4PF6 (5 mol%)
- L1b** (5.5 mol%)
- MeCN, 3 Å MS, rt, -20 °C

Product: R-CH2-CH2-CH2-CH2-SO2-N(R)-CH2-CH2-CH2-CH2-R (spirocyclic sulfonamide)

Yield: 24% - 86% yield, ee up to 92:8

Structure of **L1b** (Ligand):

L1b is a chiral ligand, specifically a spirocyclic sulfonamide derivative, shown in a grey box.

Mechanism of the reaction:

The reaction proceeds via a cyclic intermediate (a spirocyclic sulfonamide) which is then attacked by nucleophiles (Nu^1 and Nu^2) to form the final product.

Structure of the final product (spisulosine derivative):

Nu^2-CH2-CH2-CH2-CH2-NH2 (where Nu^1 and Nu^2 are nucleophiles)

Structure of the final product (spisulosine derivative):

n-C13H27-CH2-CH2-CH2-CH2-NH2 (where $n-C13H27$ is a long alkyl chain)

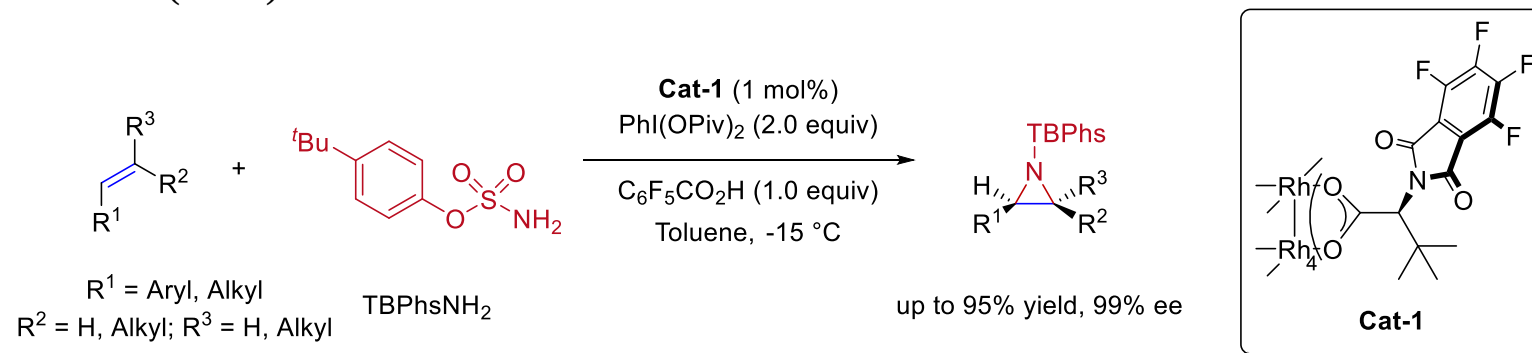
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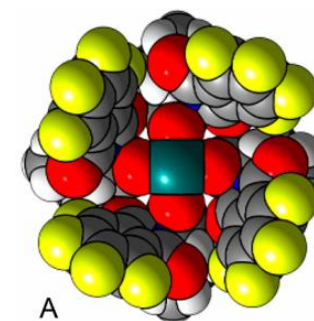
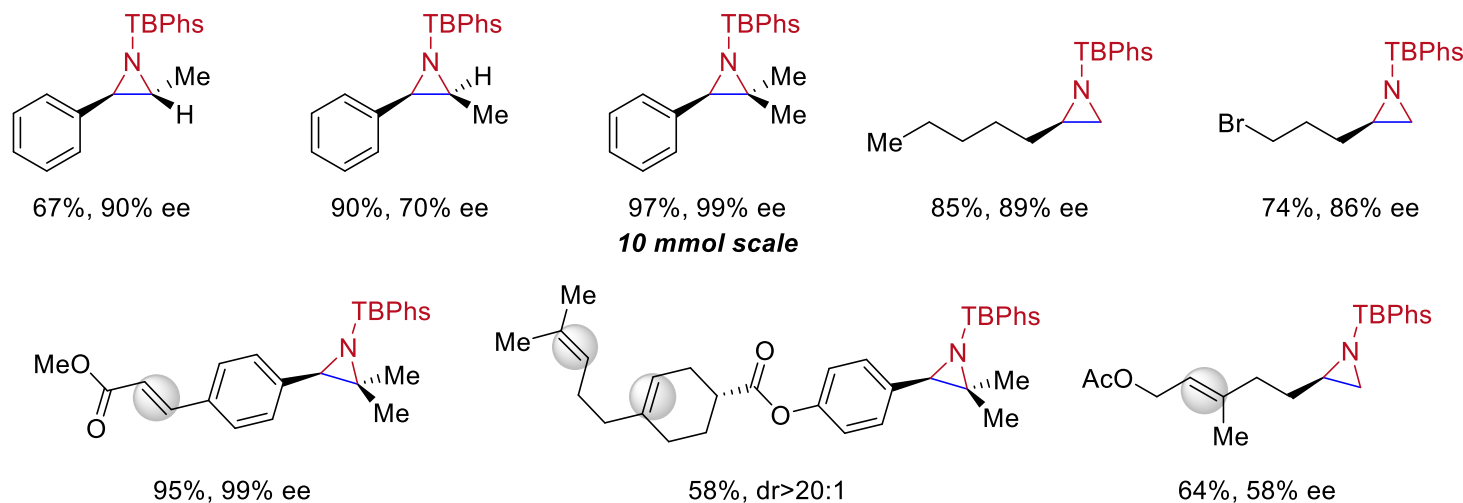
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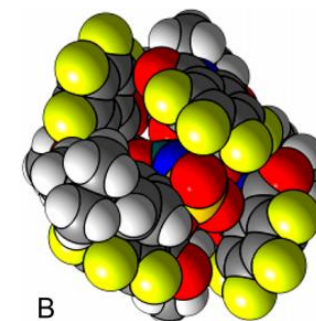
Dauban (2022)



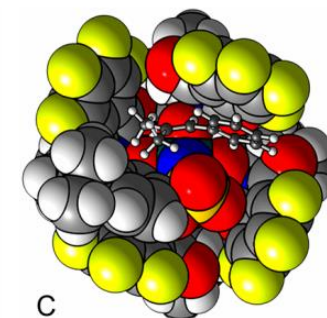
Selected examples



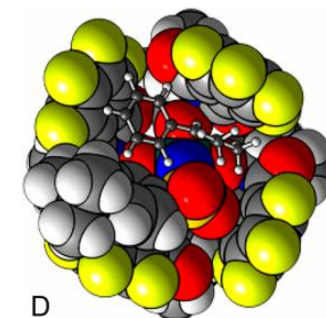
3Rh
2



3Rh_2N



$^3RhN:Styr_{Si}$

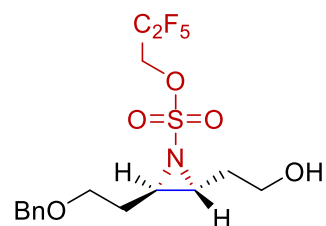
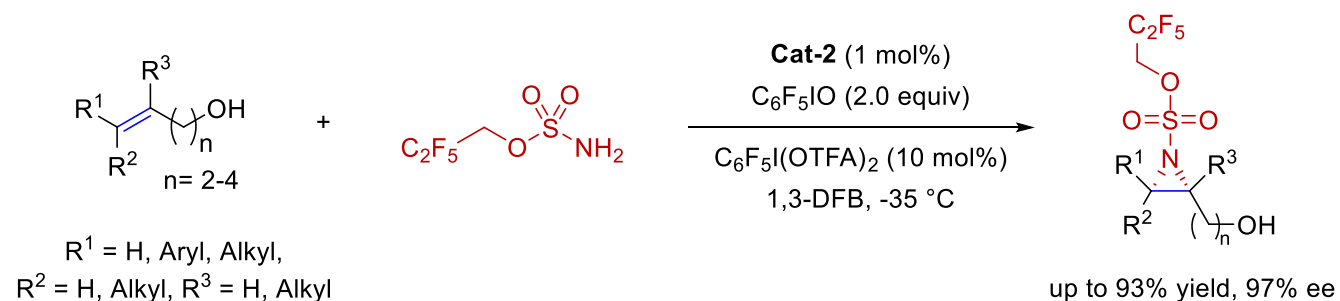


$^3RhN:Styr_{Re}$

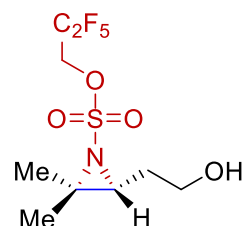
2.2 Metal–Nitrenes from Amides Derivatives



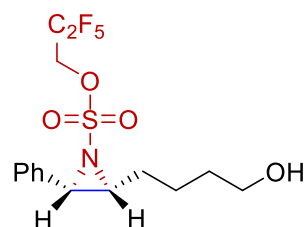
Phipps (2023)



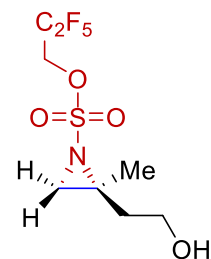
from *Trans*-substrate
53%, 91% ee



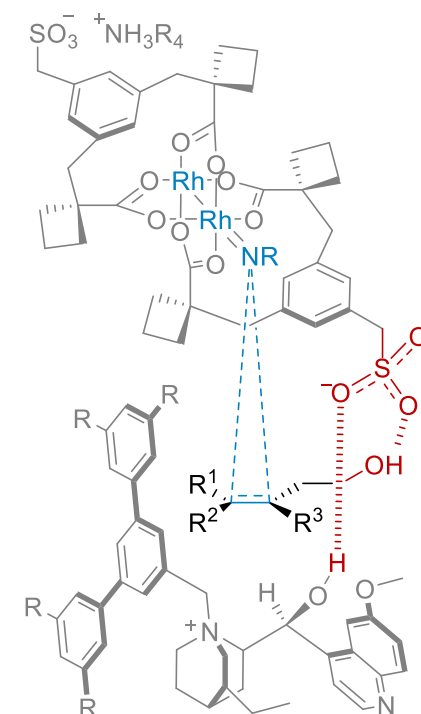
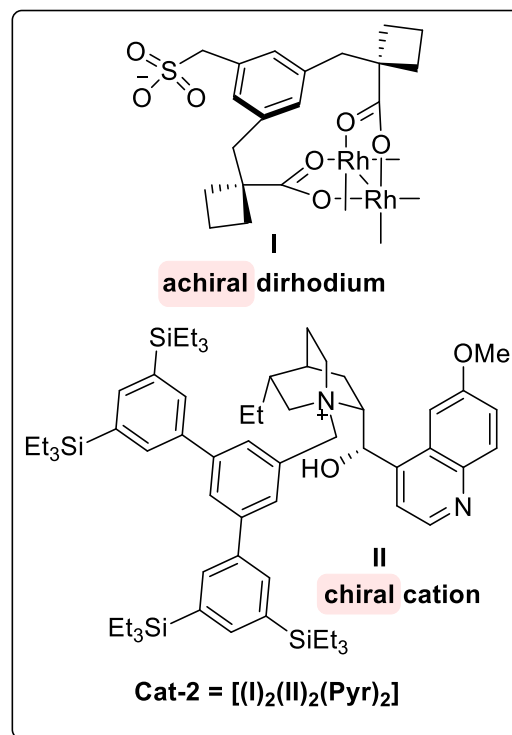
73%, 86% ee



from *Cis*-substrate
72%, 96% ee



72%, 96% ee



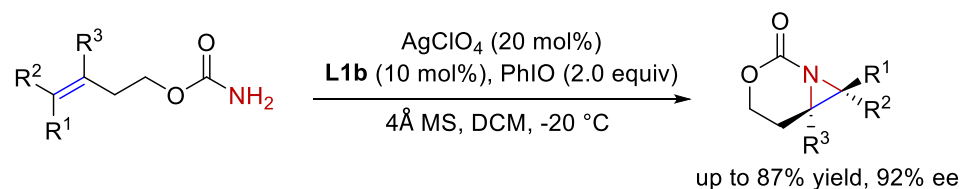
离子对作用构造出
更加牢固的手性空腔

2.2 Metal–Nitrenes from Amides Derivatives

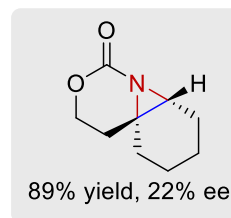
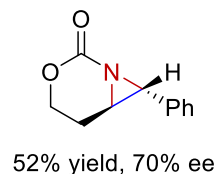
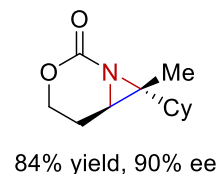
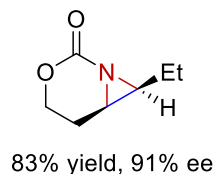
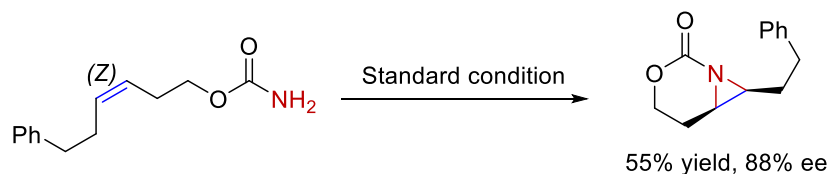


Schomaker (2017)

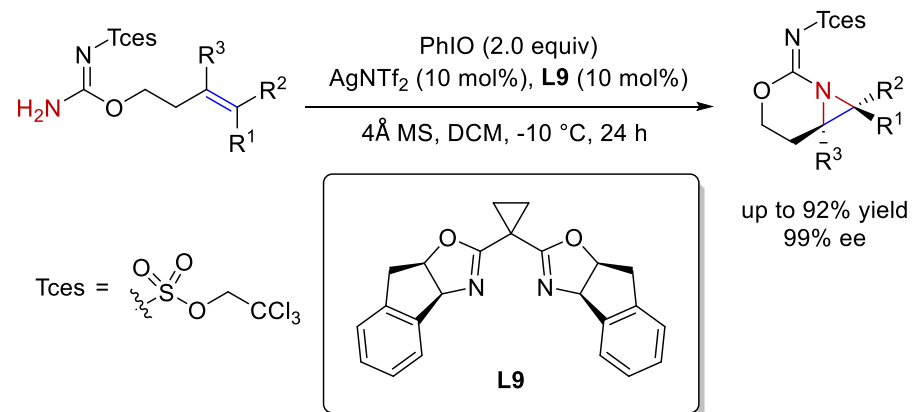
首次实现Ag催化体系下的不对称氮杂环丙烷化反应



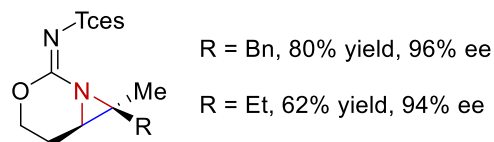
Selected examples



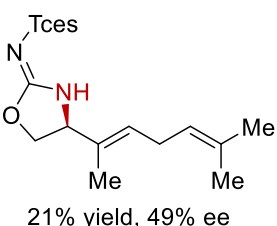
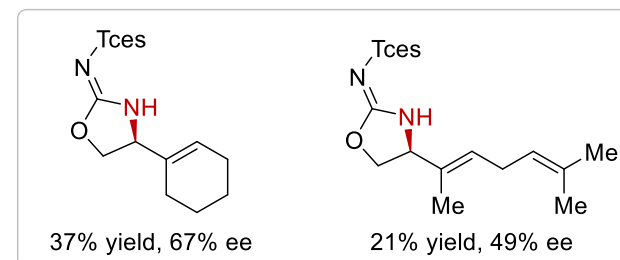
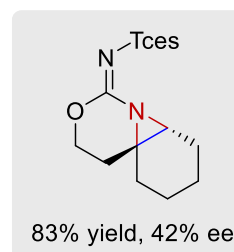
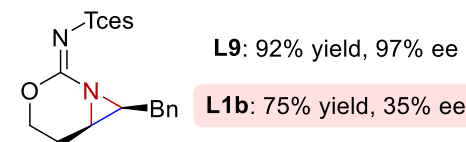
Schomaker (2024)



Selected examples



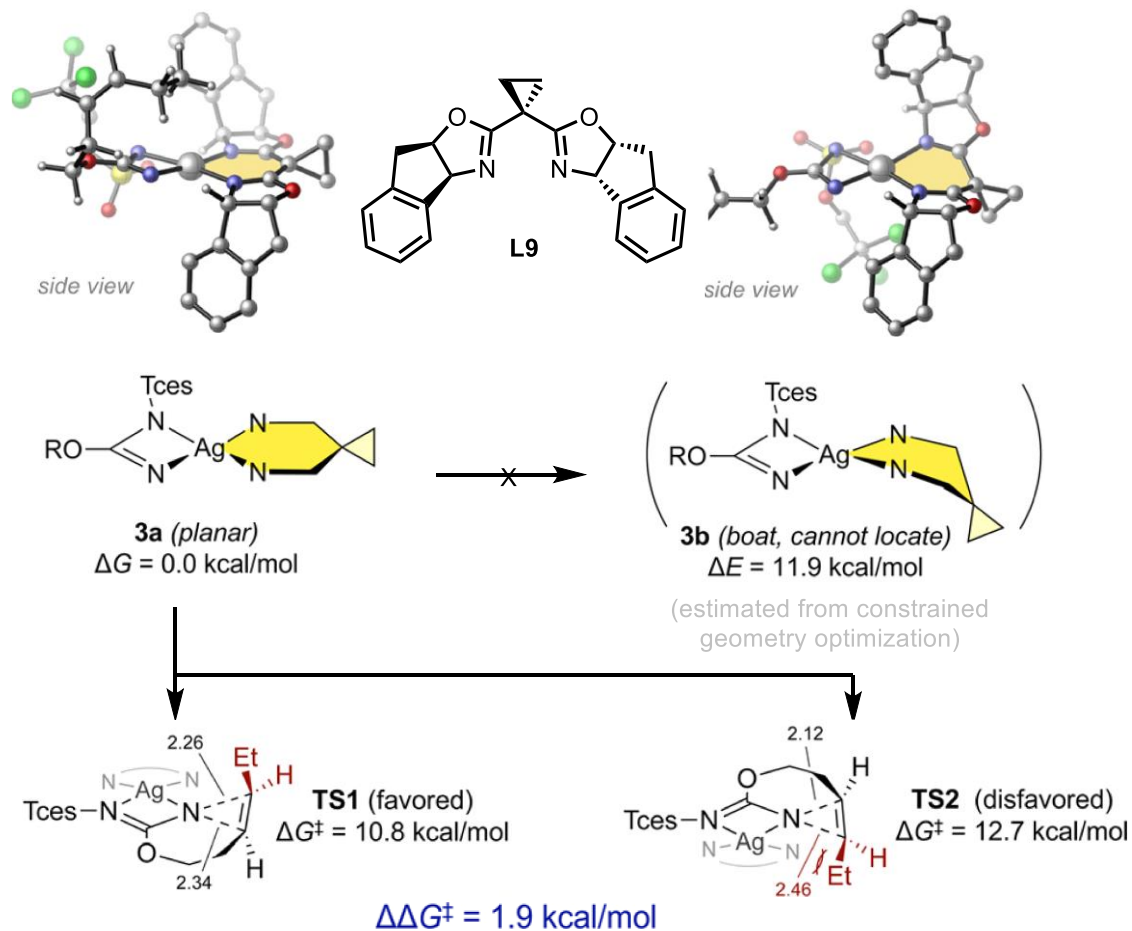
$\text{R} = \text{Bn}$, 80% yield, 96% ee
 $\text{R} = \text{Et}$, 62% yield, 94% ee



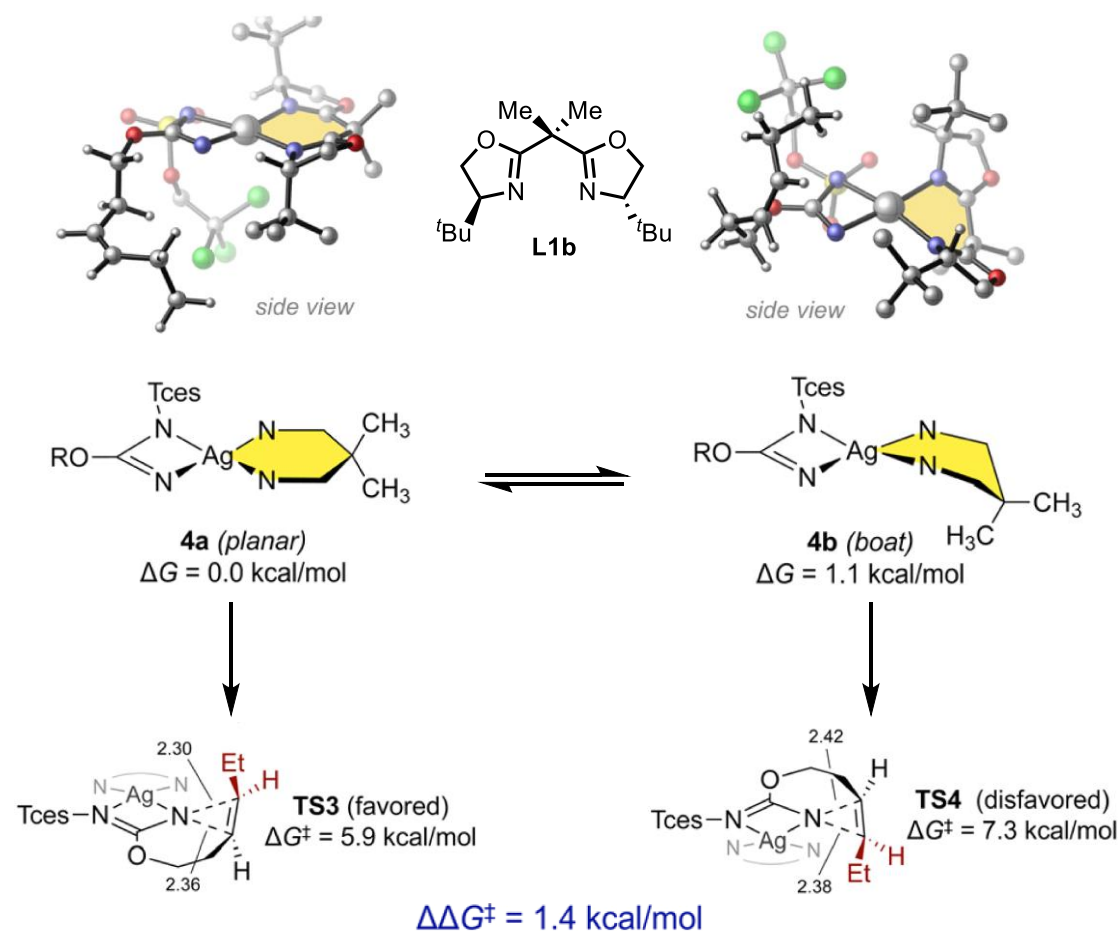
烯丙位胺化副产物

2.2 Metal–Nitrenes from Amides Derivatives

Enantioduction with rigid ligand **L9**



Enantioduction with flexible ligand **L1b**



1 Background

2 Enantioselective Azirdination By Addition of Metal–Nitrenes to Olefins

2.1 Metal–Nitrenes from Iminoiodinanes

2.2 Metal–Nitrenes from Amides Derivatives

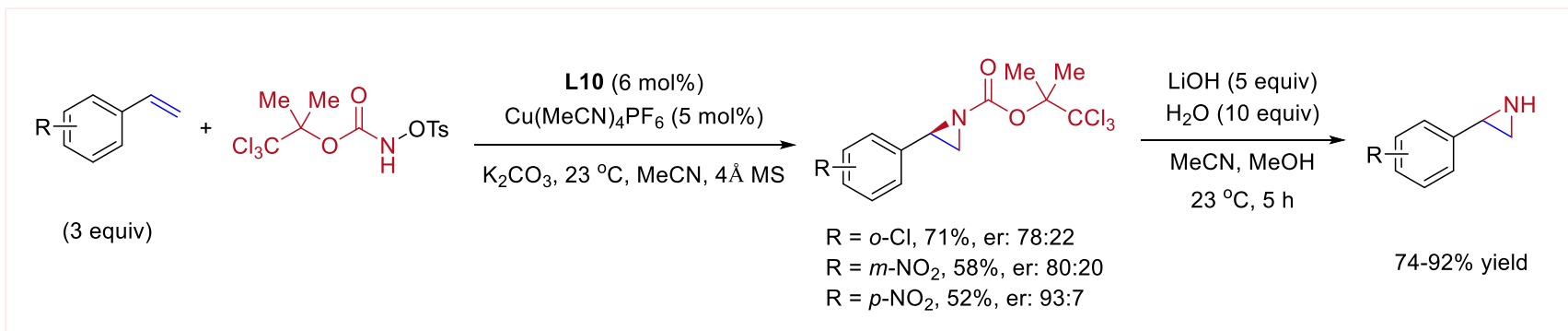
2.3 Metal–Nitrenes from Hydroxylamine Derivatives

2.4 Metal–Nitrenes from Azides

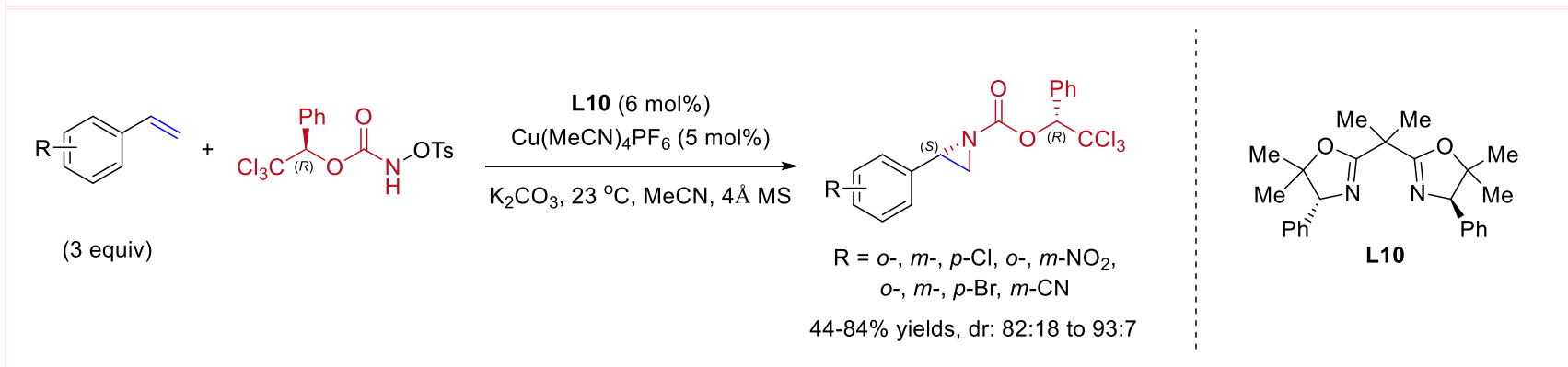
3 Summary and outlook

2.3 Metal–Nitrenes from Hydroxylamine Derivatives

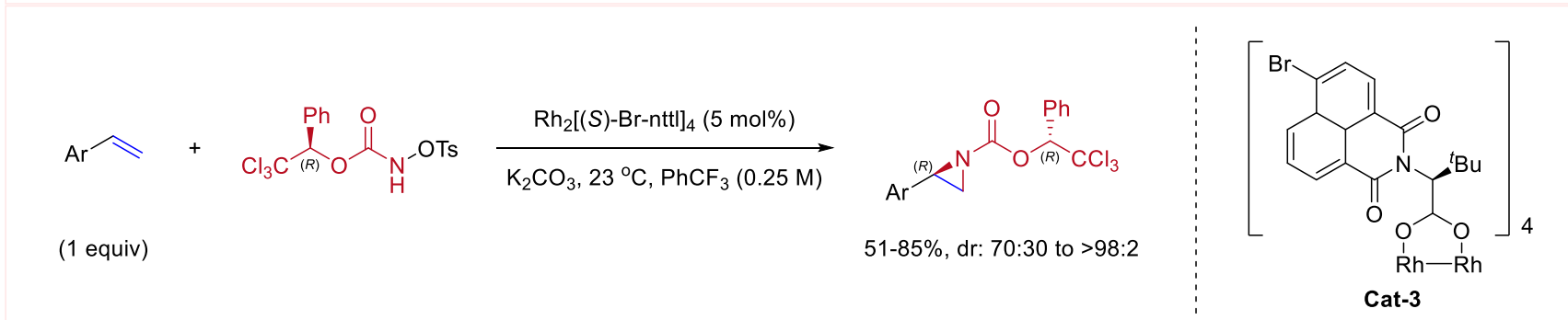
Lebel (2010)



Lebel (2012)



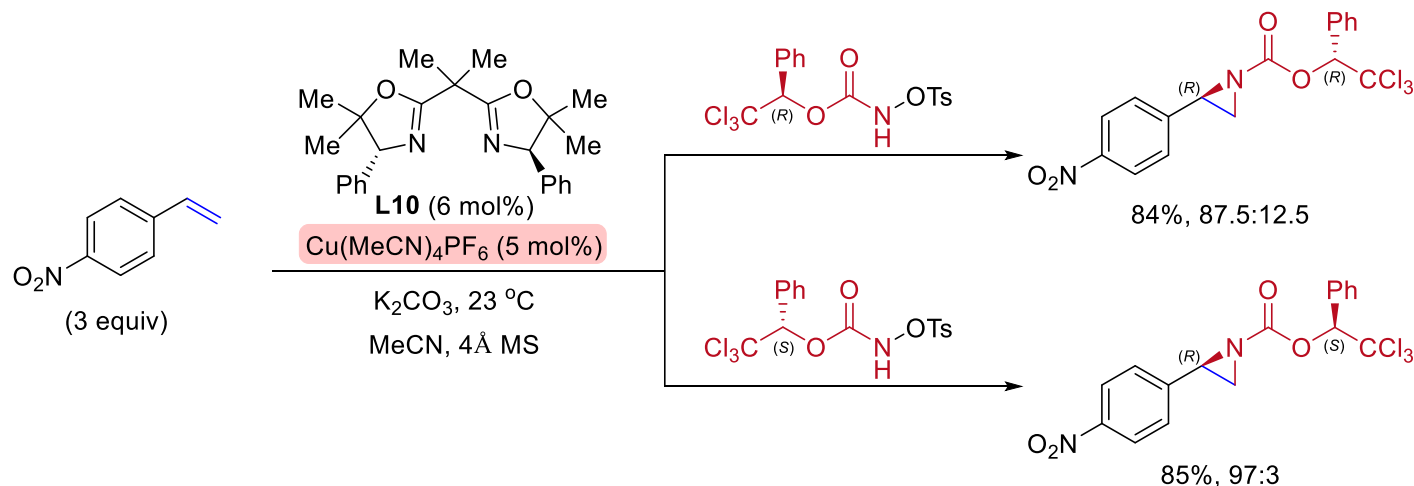
Lebel (2011)



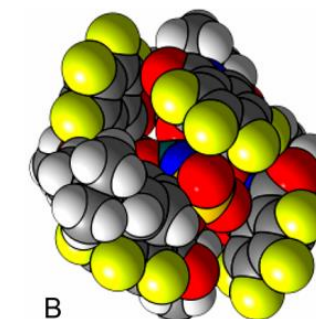
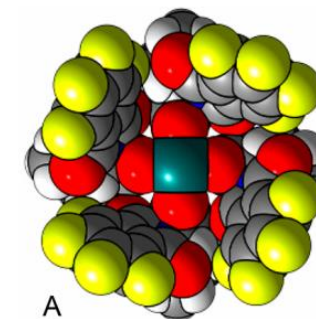
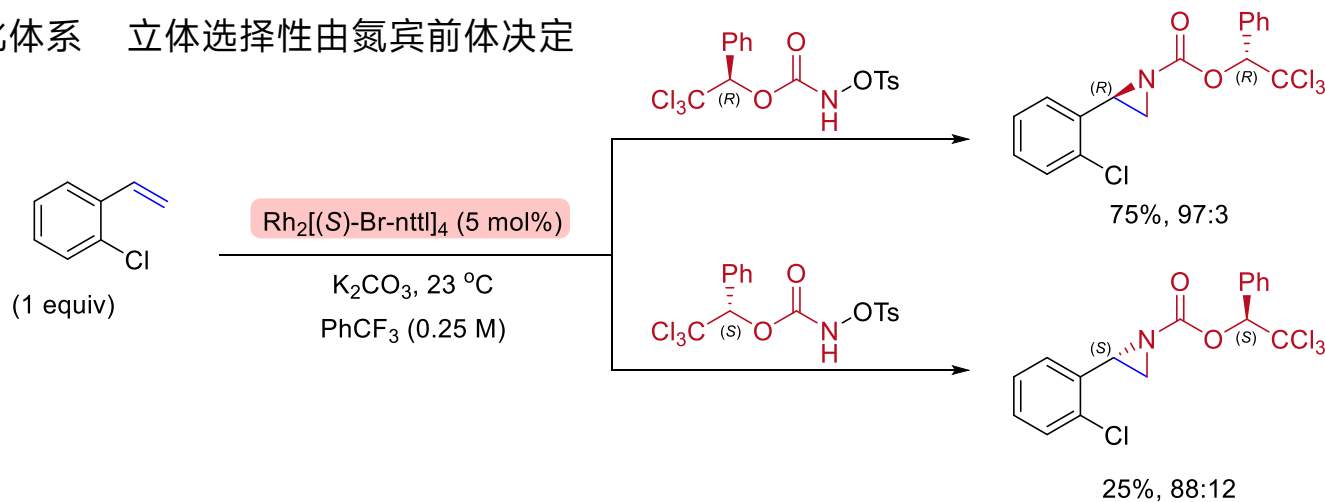
首次实现Rh催化体系下的不对称氮杂环丙烷化反应

2.3 Metal–Nitrenes from Hydroxylamine Derivatives

Cu催化体系 立体选择性由手性配体决定



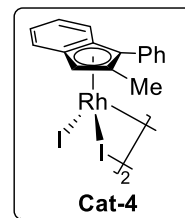
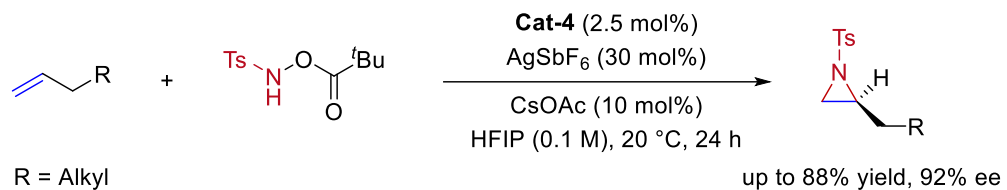
Rh催化体系 立体选择性由氮宾前体决定



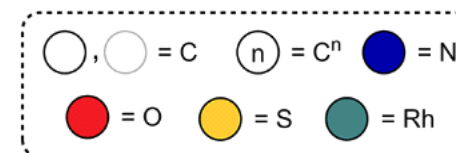
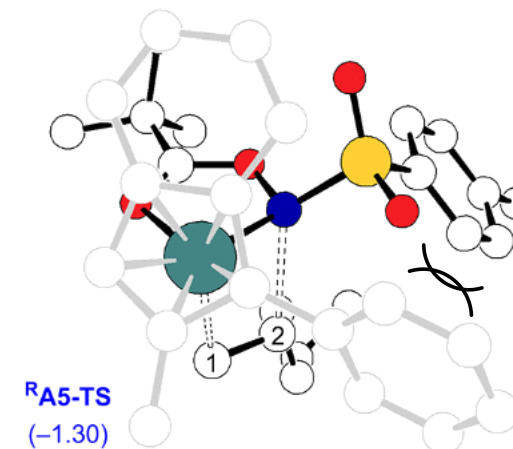
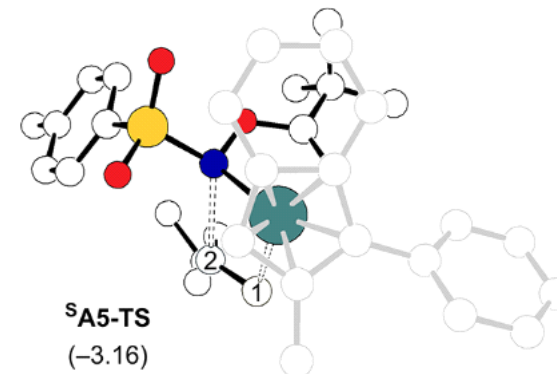
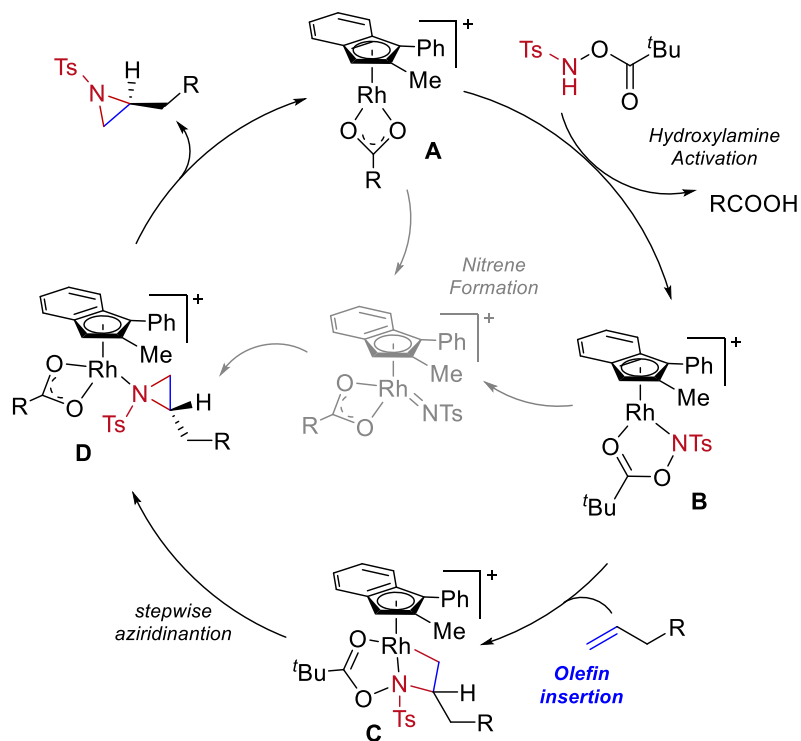
手性Rh催化剂具有手性口袋，
其在和氮宾前体配位后结构相互靠近，
压缩手性口袋的立体空间，
进而影响后续烯烃插入方向

2.3 Metal–Nitrenes from Hydroxylamine Derivatives

Blakey (2024)



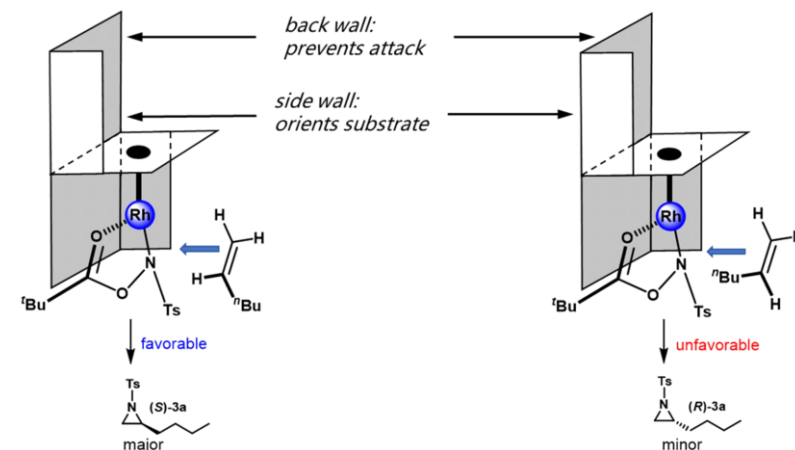
Mechanistic Proposal





$\text{R}^1 = \text{Alkyl, Aryl}$ $\text{R}^2 = \text{Alkyl, Aryl, Thienyl}$ up to 99% yield, 92% ee

99% yield, 88% ee	48% yield, 90% ee	52% yield, 77% ee
90% yield, 90% ee	NR	NR



The diagram illustrates a catalytic cycle for the asymmetric hydrogenation of allyl esters. The cycle begins with a rhodium catalyst complex, $\text{Rh}(\text{Cp}^x)(\text{I})_2$, where Cp^x is a chiral ferrocenyl ligand. This complex reacts with an allyl ester, $\text{R}-\text{CH}=\text{CH}-\text{CO}_2\text{Ts}$, to form a π -allyl rhodium intermediate. The intermediate then undergoes hydride transfer from a $\text{PivO}-\text{H}$ source to yield the saturated ester product, $\text{R}-\text{CH}_2-\text{CH}_2-\text{CO}_2\text{Ts}$, and regenerate the catalyst. The catalyst is shown in two enantiomeric forms, indicating a chiral environment. The ligand structure is a ferrocene derivative with two indenyl rings, one substituted with a methoxy group (OMe).

1 Background

2 Enantioselective Azirdination By Addition of Metal–Nitrenes to Olefins

2.1 Metal–Nitrenes from Iminoiodinanes

2.2 Metal–Nitrenes from Amides Derivatives

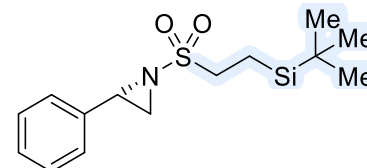
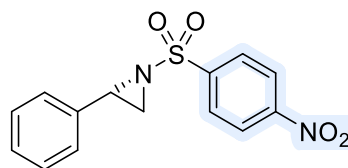
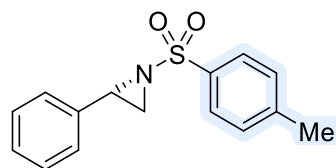
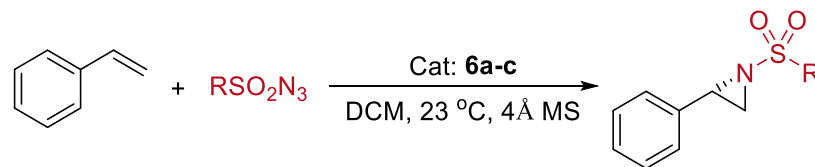
2.3 Metal–Nitrenes from Hydroxylamine Derivatives

2.4 Metal–Nitrenes from Azides

3 Summary and outlook

2.4 Metal–Nitrenes from Azides

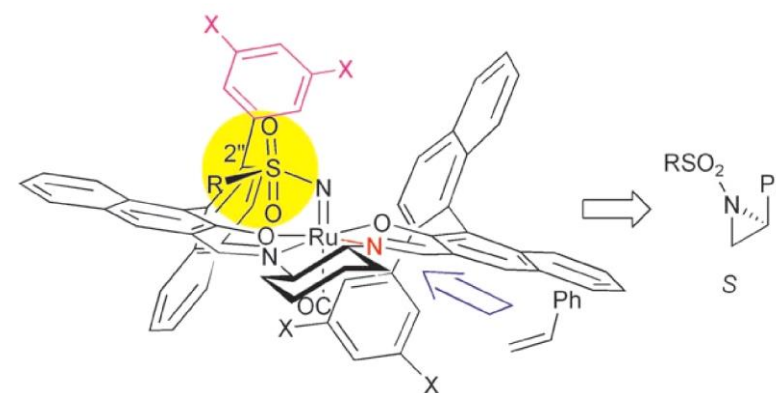
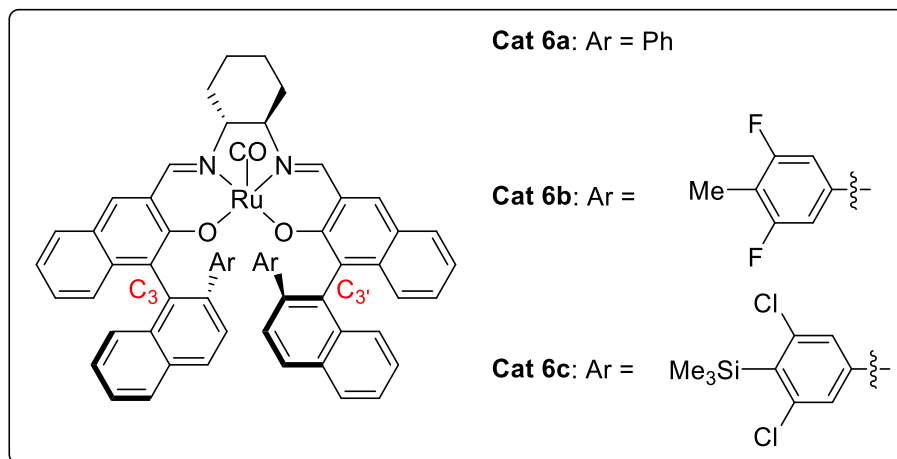
Katsuki (2004 and 2007)



Cat 6a (2 mol%), 71%, er: 94:6 (TON = 36)

Cat 6b (0.09 mol%), 78%, er: 92:8 (TON = 867) **Cat 6b** (1 mol%), 34%, er: 92:8 (TON = 34) **Cat 6c** (1 mol%), 99%, er: 96:4 (TON = 99)

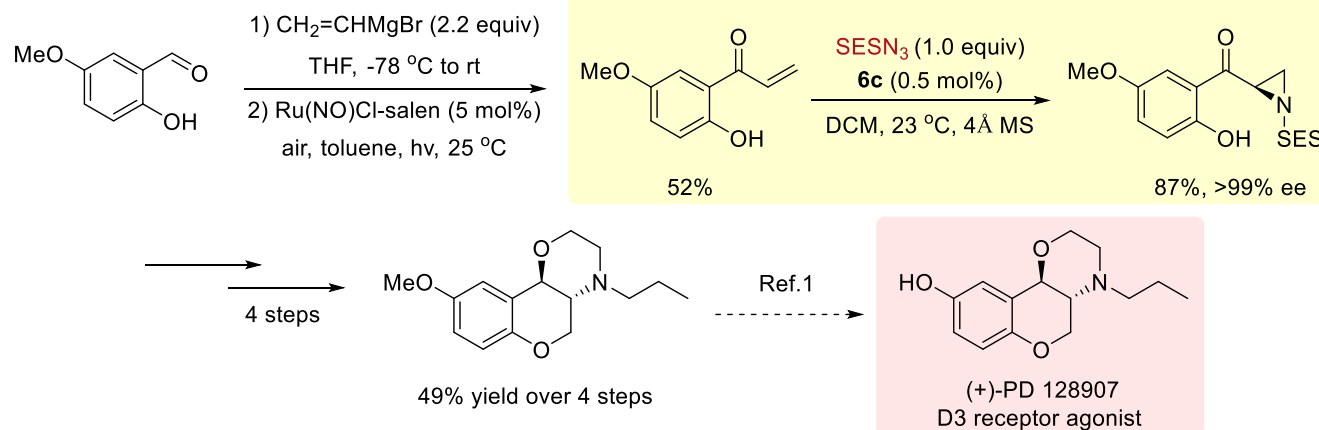
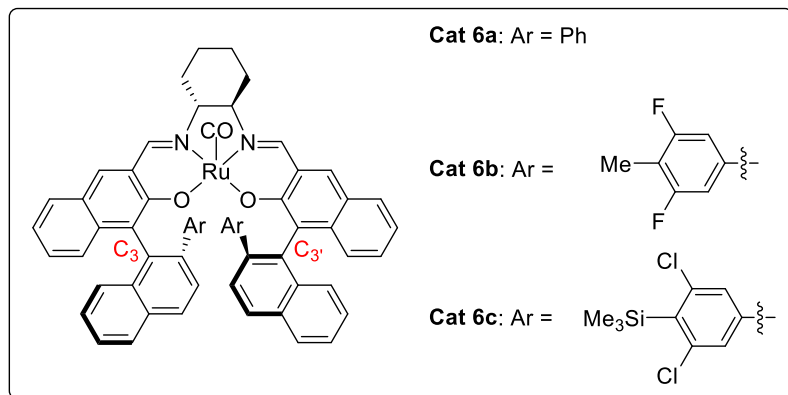
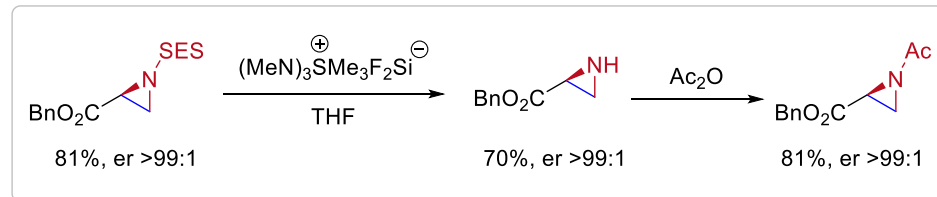
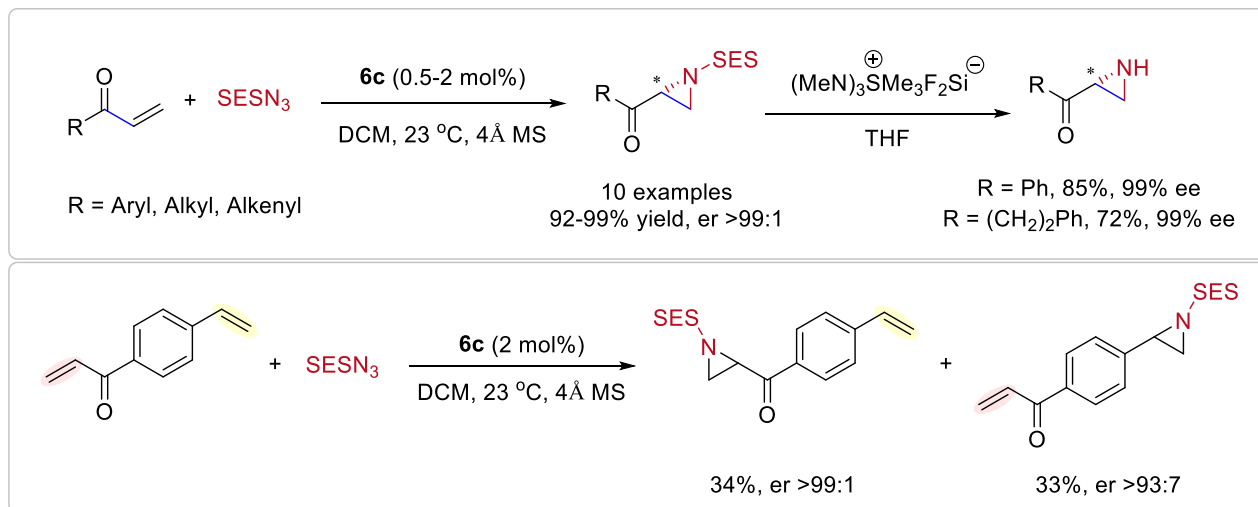
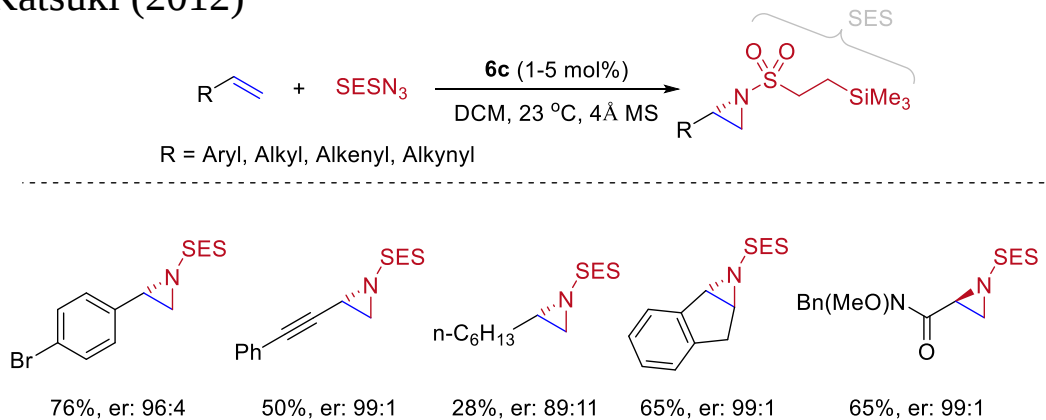
Cat 6c (0.1 mol%), 93%, er: 93:7 (TON = 982)



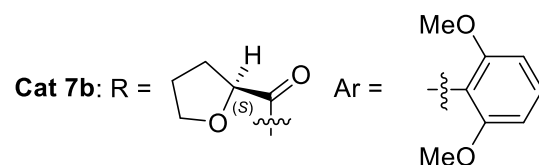
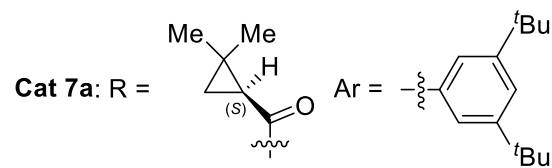
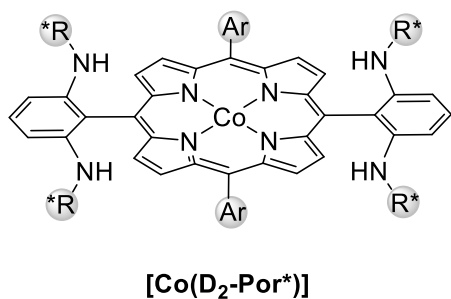
2.4 Metal–Nitrenes from Azides



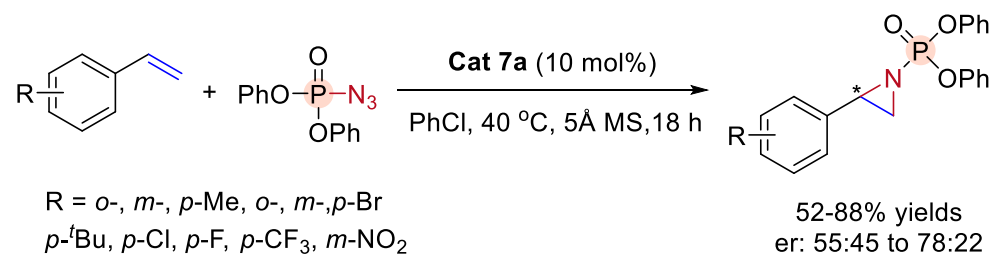
Katsuki (2012)



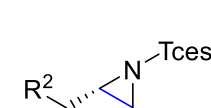
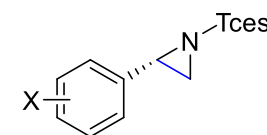
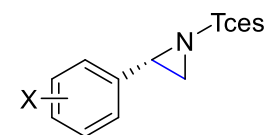
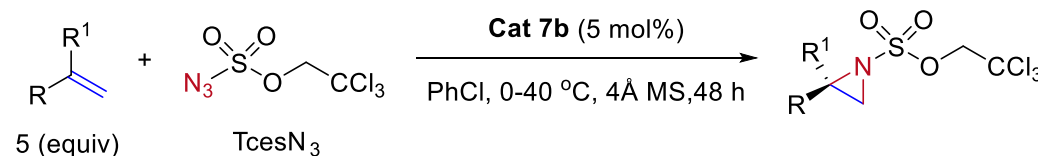
2.4 Metal–Nitrenes from Azides



Peter Zhang (2008)



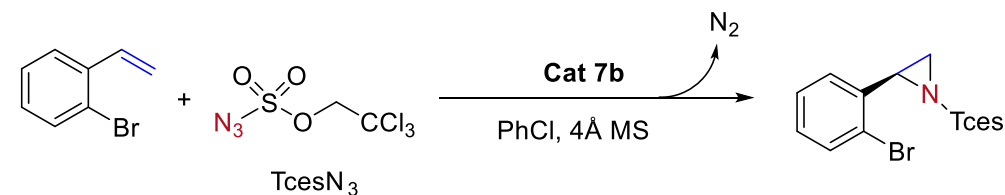
Peter Zhang (2009)



85-92%, er: 90:10 to 97:3
 X = H, EDG, EWG

X = H, 48%, er: 90:10
 X = *p*-Cl, 43%, er: 90:10

R² = Ph, 26%, er: 97:3
 R² = *n*-Pr, 42%, er: 96:4

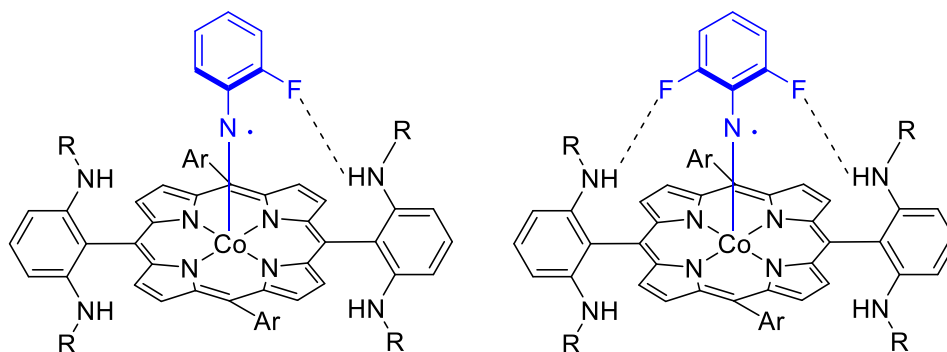
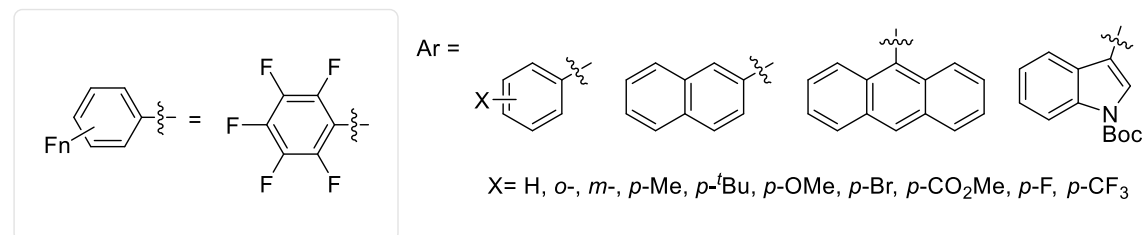
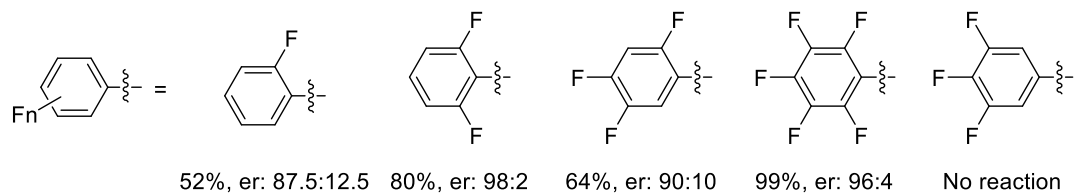
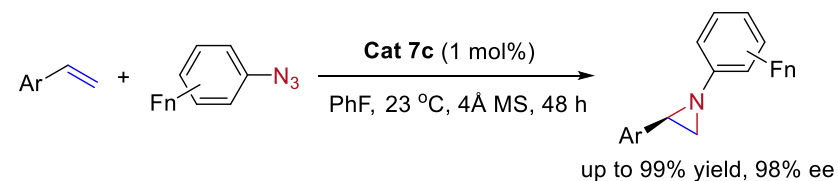
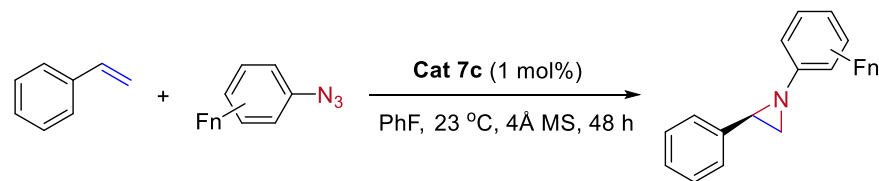


Entry	Cycle	Temp/°C	Yield (%)	ee (%)
1	First	rt	95	96
2	Second	rt	89	94
3	Third	rt	81	94

2.4 Metal–Nitrenes from Azides

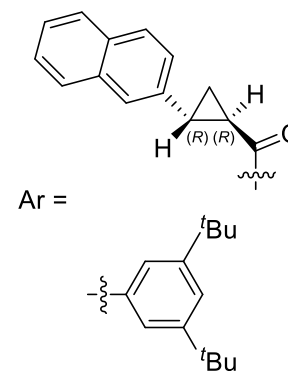


Peter Zhang (2013)



[Co(D₂-Por*)] Cat 7c

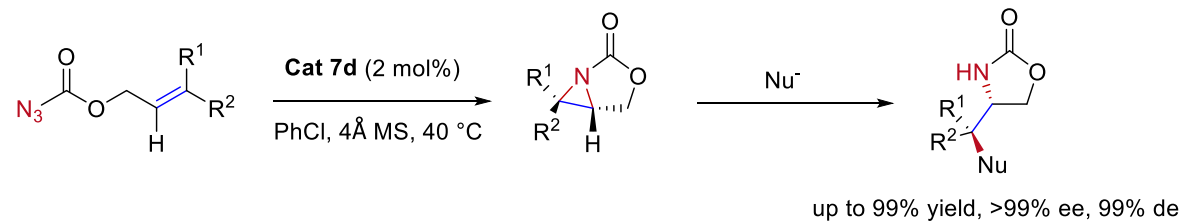
Cat 7c: R =



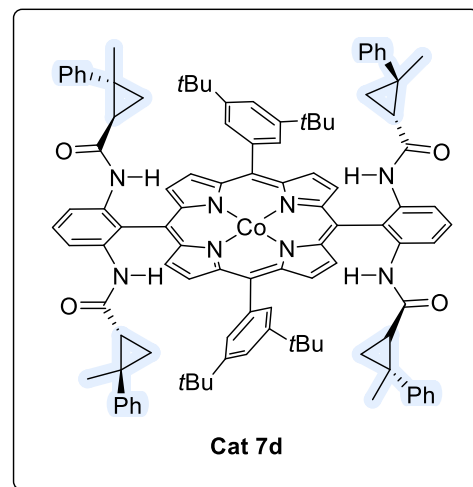
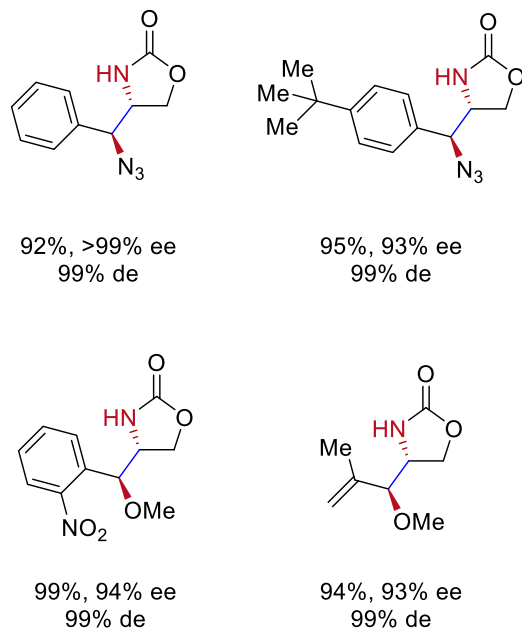
2.4 Metal–Nitrenes from Azides



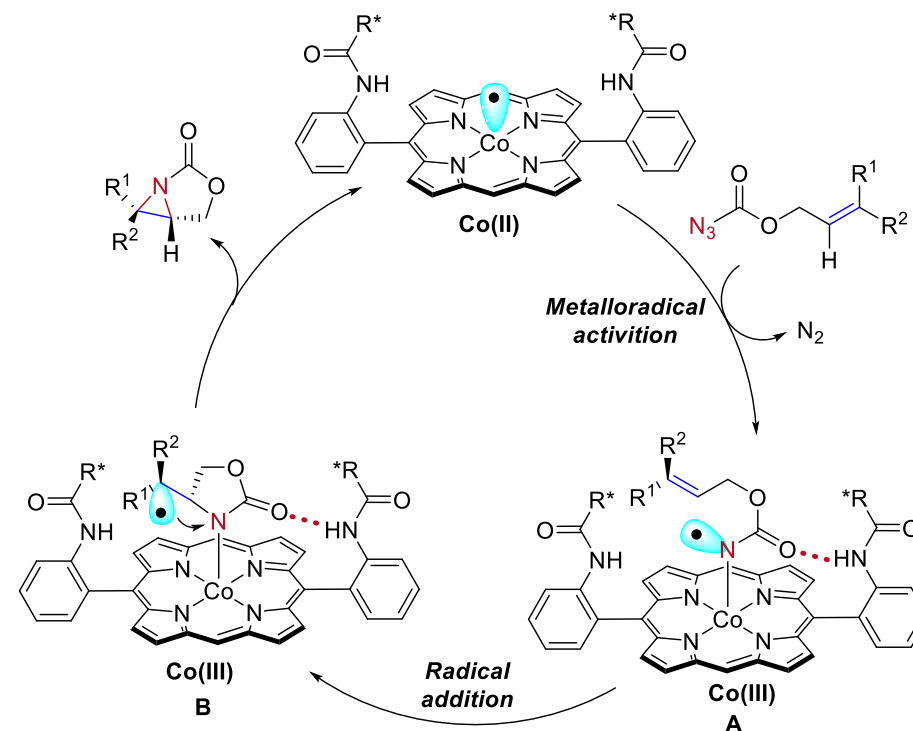
Peter Zhang (2017)



Selected examples



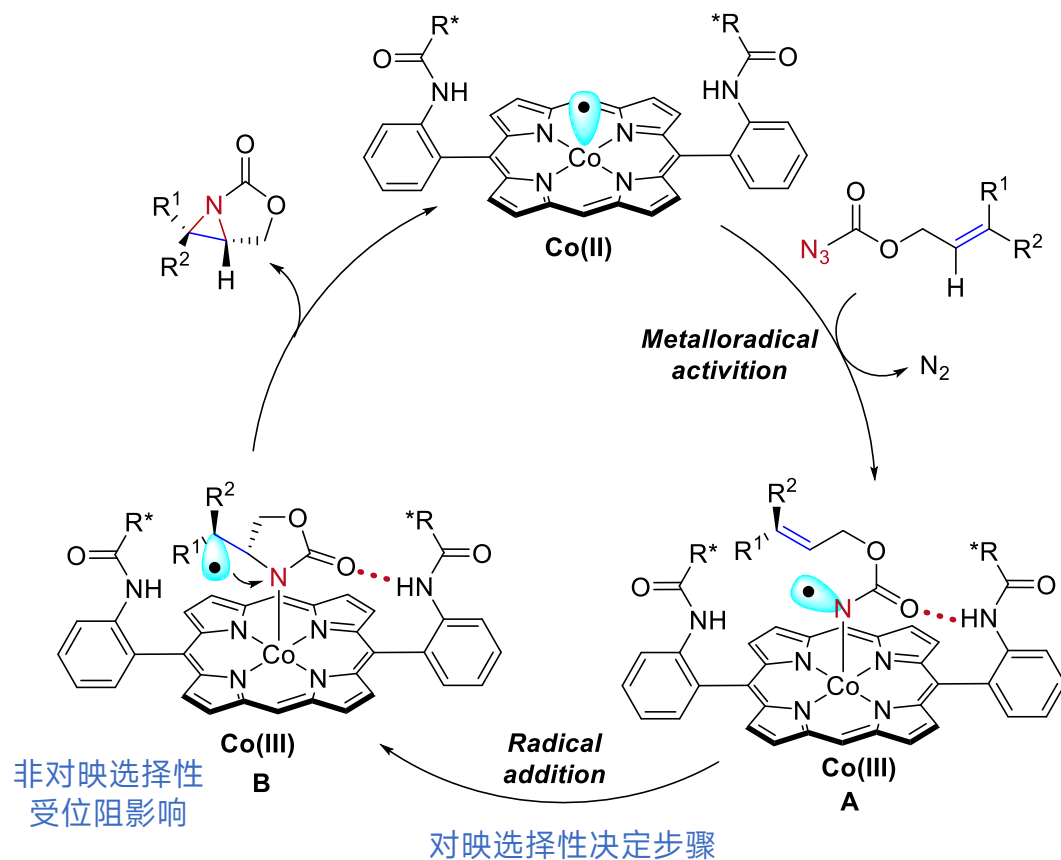
Mechanistic Proposal



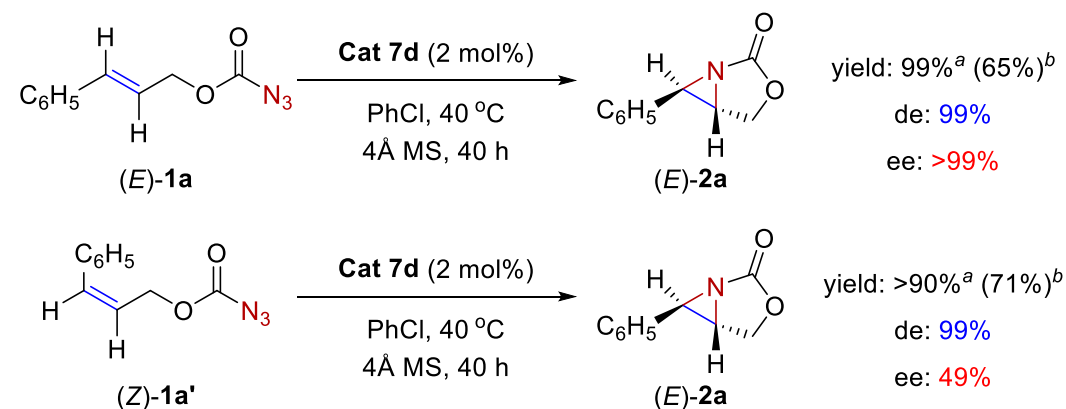
2.4 Metal–Nitrenes from Azides



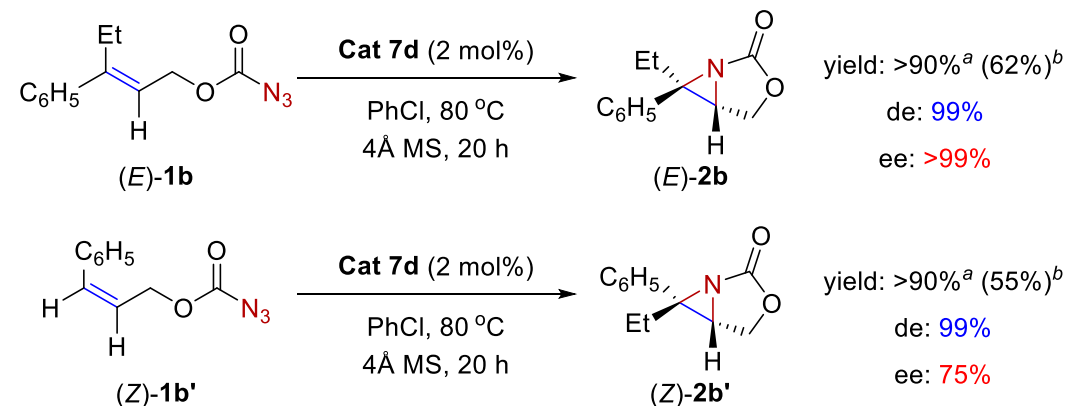
Mechanistic Proposal



A. Diastereoconvergent Asymmetric Radical Bicyclization of Azidoformates



B. Diastereospecific Asymmetric Radical Bicyclization of Azidoformates

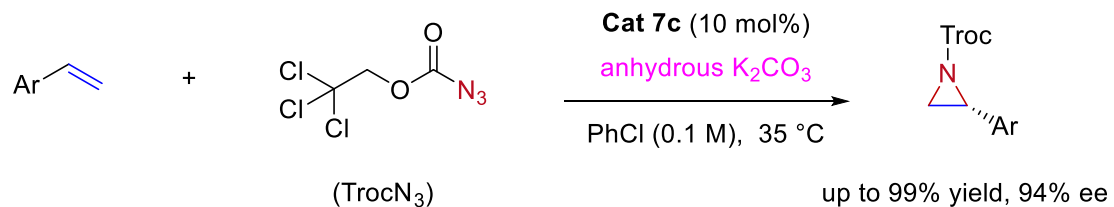


^aNMR yields. ^bIsolated yields. Note: Aziridines were unstable toward workup.

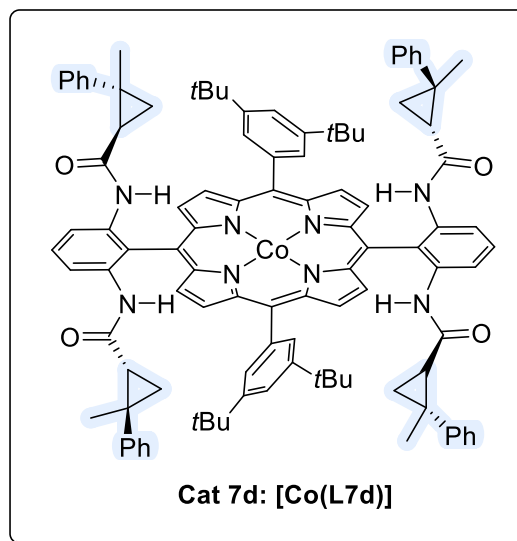
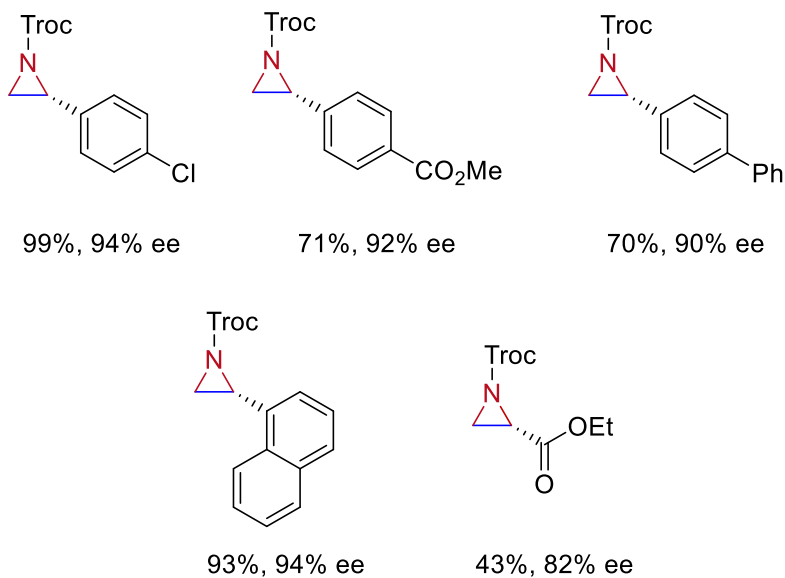
2.4 Metal–Nitrenes from Azides



Peter Zhang (2021)

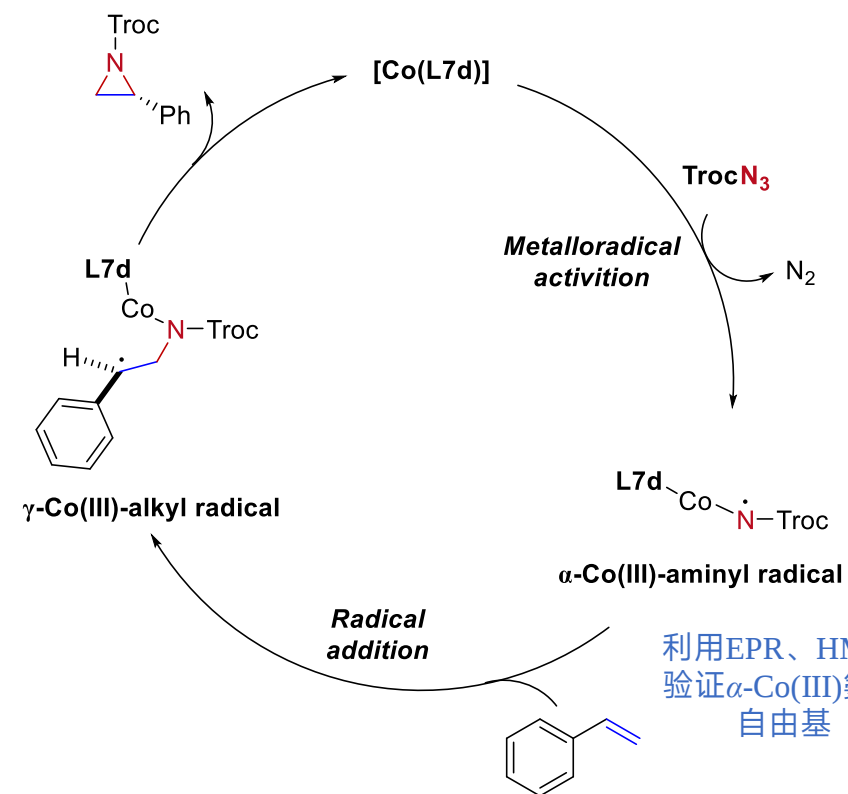


Selected examples



K₂CO₃: 除水、抑制Co(III)路易斯酸性

Mechanistic Proposal

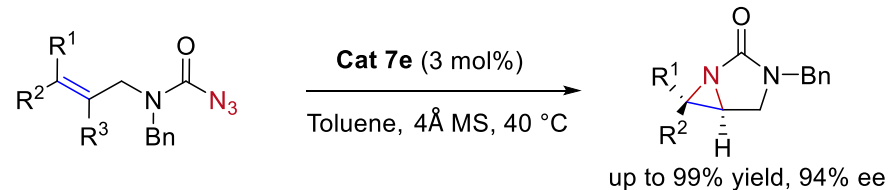


利用EPR、HMRS
验证α-Co(III)氨基
自由基

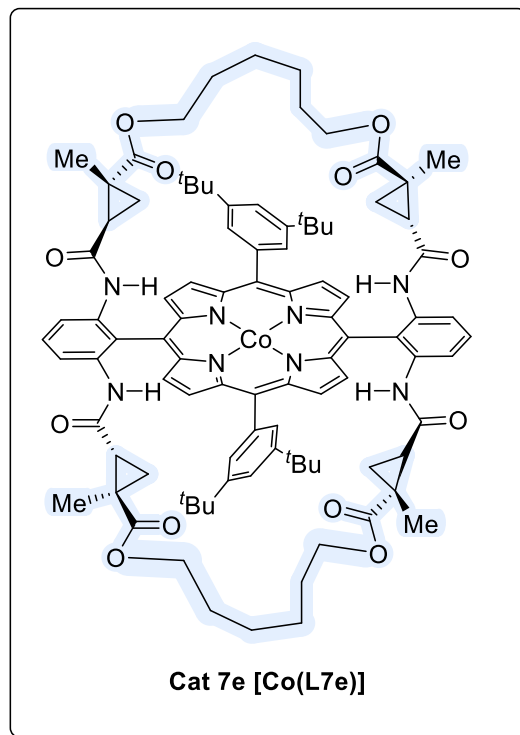
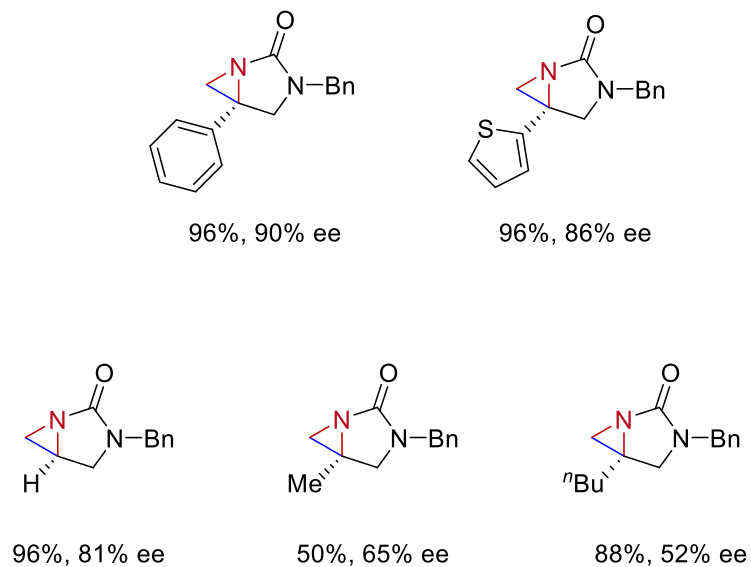
2.4 Metal–Nitrenes from Azides



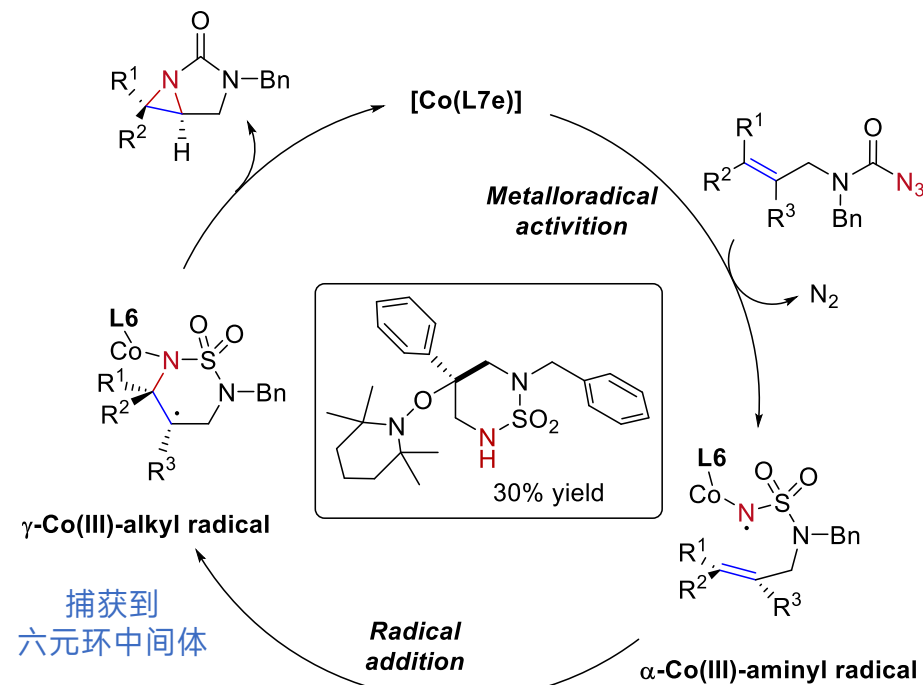
Peter Zhang (2024)



Selected examples



Mechanistic Proposal



1 Background

2 Enantioselective Azirdination By Addition of Metal–Nitrenes to Olefins

2.1 Metal–Nitrenes from Iminoiodinanes

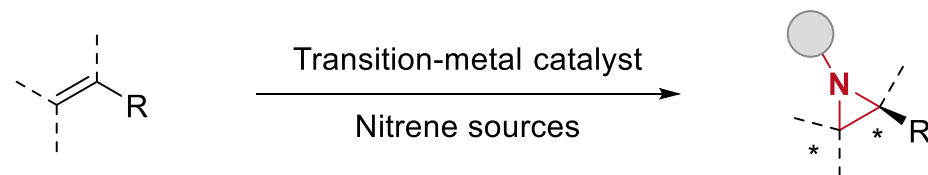
2.2 Metal–Nitrenes from Amides Derivatives

2.3 Metal–Nitrenes from Hydroxylamine Derivatives

2.4 Metal–Nitrenes from Azides

3 Summary and outlook

3 Summary and outlook



R = EWG, Ar, or Alkyl

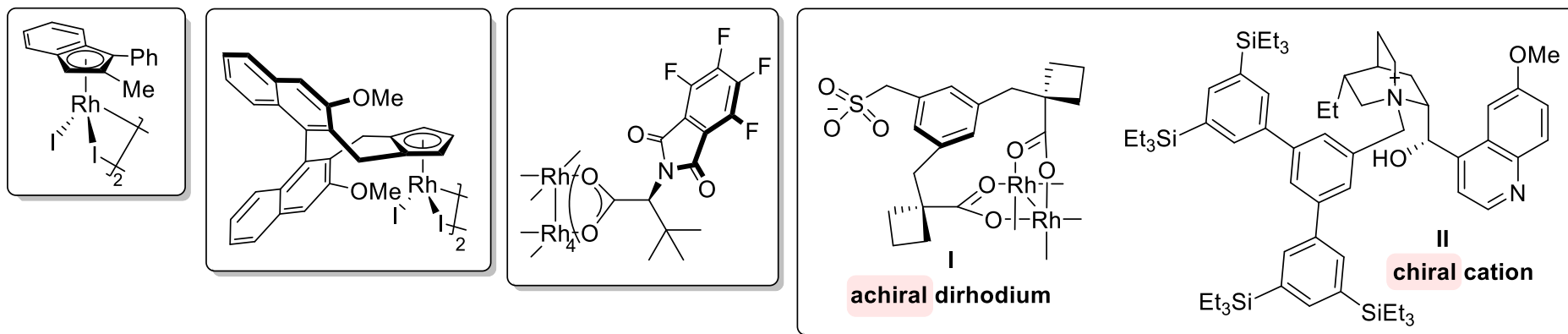
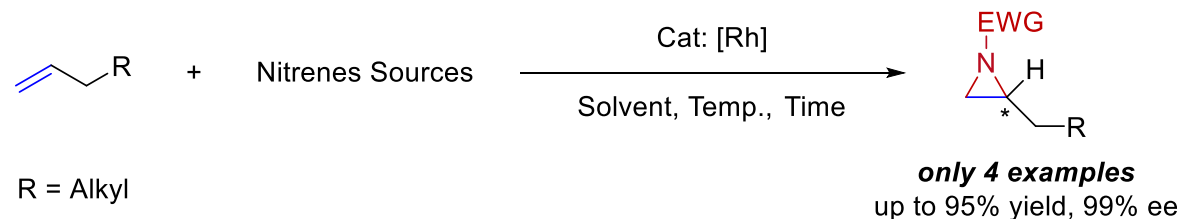
Chiral aziridine

Transition-metal	Activated Alkenes	Chiral Ligand
^{26}Fe $3d^6 4s^2$ 55.85 ^{27}Co $3d^7 4s^2$ 58.93 ^{29}Cu $3d^{10} 4s^2$ 63.55	α, β-unsaturated styrene	
Unactivated Alkenes	Chiral Metal Complex	
Alkyl · Low Polarity · Low Functionalization · Minimal Steric Differentiation	Ru, Rh, Co	

3 Summary and outlook



非活化烯烃分子间不对称氮杂环丙烷化反应



非活化烯烃分子间不对称氮杂环丙烷化研究较少，
现有工作集中在Rh催化体系，体系复杂，
手性催化剂制备难度大



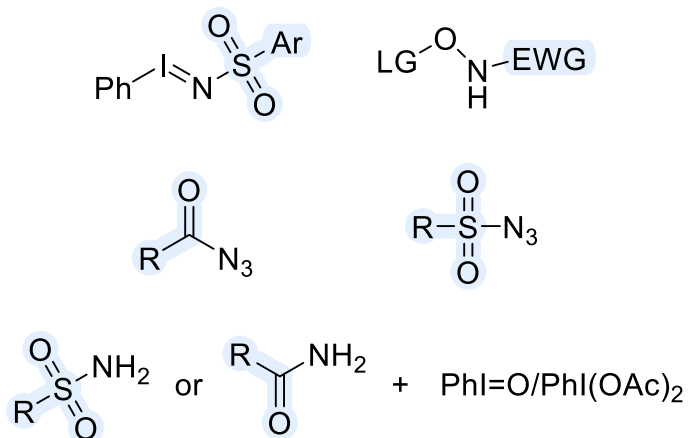
发展丰产金属Fe、Cu体系
合成活性匹配的配体骨架

3 Summary and outlook



无保护基团的氮杂环丙烷不对称构建

Nitrenes Sources

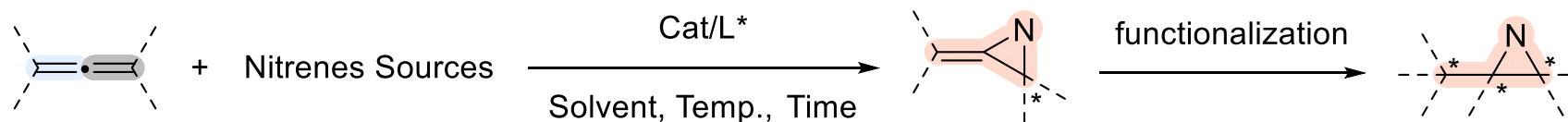


氮宾前体均连有吸电子基，
只能够直接构建具有吸电子N-保护基的
氮杂环丙烷



后续具有NH的氮宾前体
直接实现氮杂环丙烷的不对称构建

联烯的选择性氮宾插入





Thank you for your attention !

4 Supporting Information

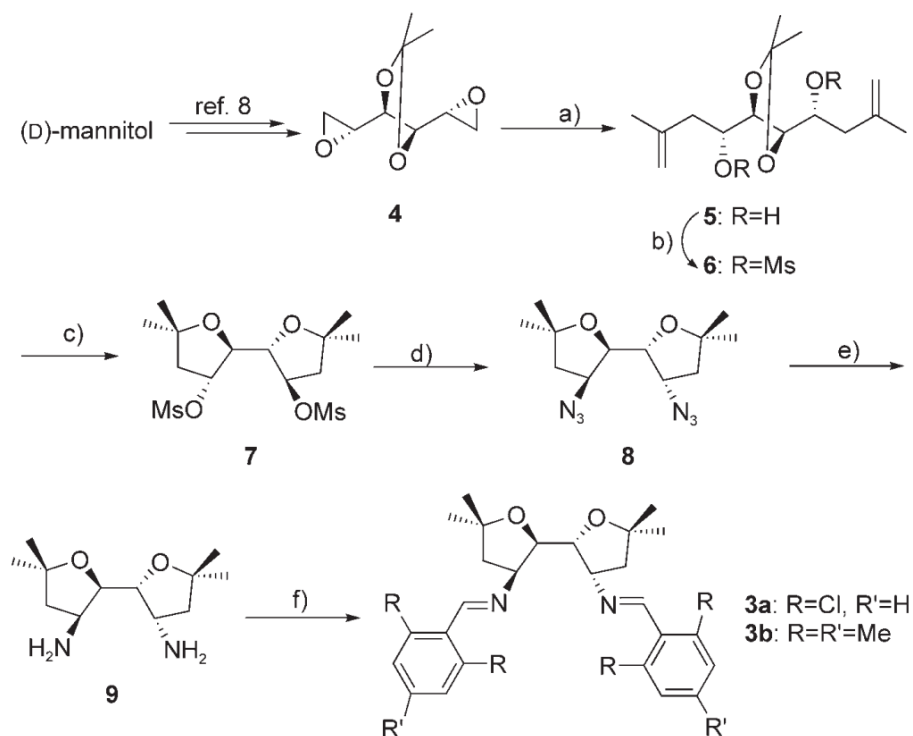
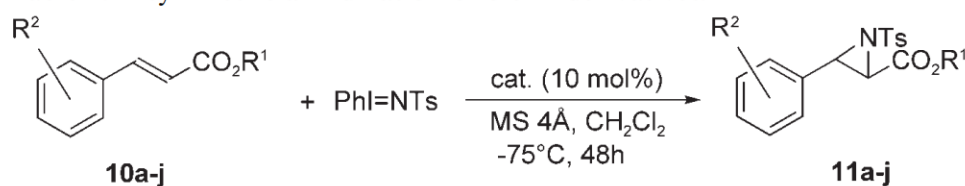


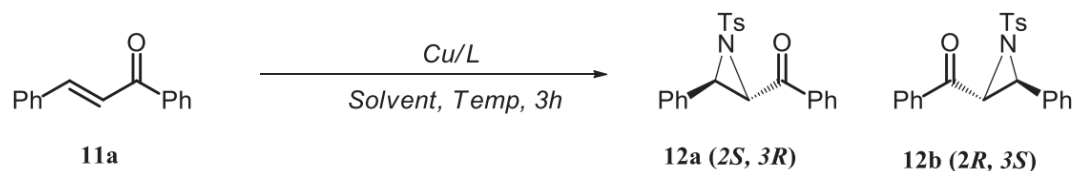
Table 2. Asymmetric aziridination of olefin derivatives.^[a]



Entry	Substrate	R ¹	R ²	Yield [%] ^[b]	ee [%] ^[c]	Config. ^[d]
1	10a	Me	H	97	87.6	2 <i>S</i> ,3 <i>R</i> (–)
2	10b	Ph	H	96	87.1	2 <i>S</i> ,3 <i>R</i> (–)
3	10c	<i>t</i> Bu	H	99	>99	2 <i>S</i> ,3 <i>R</i> (–)
4	10d	<i>t</i> Bu	4-F	99	97.8	(–)
5	10e	<i>t</i> Bu	4-Cl	97	98.1	(–)
6	10f	<i>t</i> Bu	4-Br	97	98.3	2 <i>S</i> ,3 <i>R</i> (–)
7	10g	<i>t</i> Bu	4-Me	86	94.4	(–)
8	10h	<i>t</i> Bu	4-MeO	95	80.1	(+)
9	10i	<i>t</i> Bu	2-NO ₂	63	98.6	(+)
10	10j	<i>t</i> Bu	2,3-MeO	67	97.0	(+)

[a] The reactions were carried out with the ratio of **10**/PhI=NTs/**3a**/Cu 5:1:0.11:0.1 in a 0.125 mmol scale of PhI=NTs (0.042 M) in the presence of 100 mg of MS 4 Å. [b] The yield of isolated product. [c] The enantiomeric excess was determined by HPLC on Chiralcel columns. [d] The absolute configurations of **11a–c** and **11k–l** were determined by comparison of their optical rotations with those of literature data.^[4b] The absolute configuration of (–)-**11f** was determined to be (2*S*,3*R*) by the Bijvoet method based on the anomalous dispersion of Br heavy atom.^[10]

4 Supporting Information



Entry	Ligand	[Cu]	[Cu] load	Solvent	Temp	Yield	%ee ^a
1	9	(CuOTf) ₂ ·PhMe	5%	DCM	rt	91	51
2	10	(CuOTf) ₂ ·PhMe	5%	DCM	rt	81	36
3	9	(CuOTf) ₂ ·PhMe	3%	DCM	rt	82	59
4	9	Cu(OTf) ₂	3%	DCM	rt	33	45
5	9	CuI	3%	DCM	rt	NR	-
6	9	CuCl	3%	DCM	rt	25	43
7	9	Cu(OAc)	3%	DCM	rt	NR	-
8	9	(CuOTf) ₂ ·PhMe	3%	Chloroform	rt	NR	-
9	9	(CuOTf) ₂ ·PhMe	3%	DCE	rt	70	48
10	9	(CuOTf) ₂ ·PhMe	3%	THF	rt	NR	-
11	9	(CuOTf) ₂ ·PhMe	3%	Toluene	rt	NR	-
12	9	(CuOTf) ₂ ·PhMe	3%	Acetone	rt	80	67
13	9	(CuOTf) ₂ ·PhMe	3%	Dioxane	rt	NR	-
14	9	(CuOTf) ₂ ·PhMe	3%	DME	rt	69	69
15	9	(CuOTf) ₂ ·PhMe	3%	CH ₃ CN	rt	93	76
16	9	(CuOTf) ₂ ·PhMe	3%	CH ₃ CN	0°C	93	78

Reaction conditions: Chalcone (100mg, 0.48mmol), Ligand (5%eq), Cu precursor (the load is indicated), PhI=NTs (120mg, 0.32mmol), 3ml dry CH₃CN; a) Determined by HPLC with a chiral OD-H column using heptane/2-propanol (90:10, v/v) as the eluent.

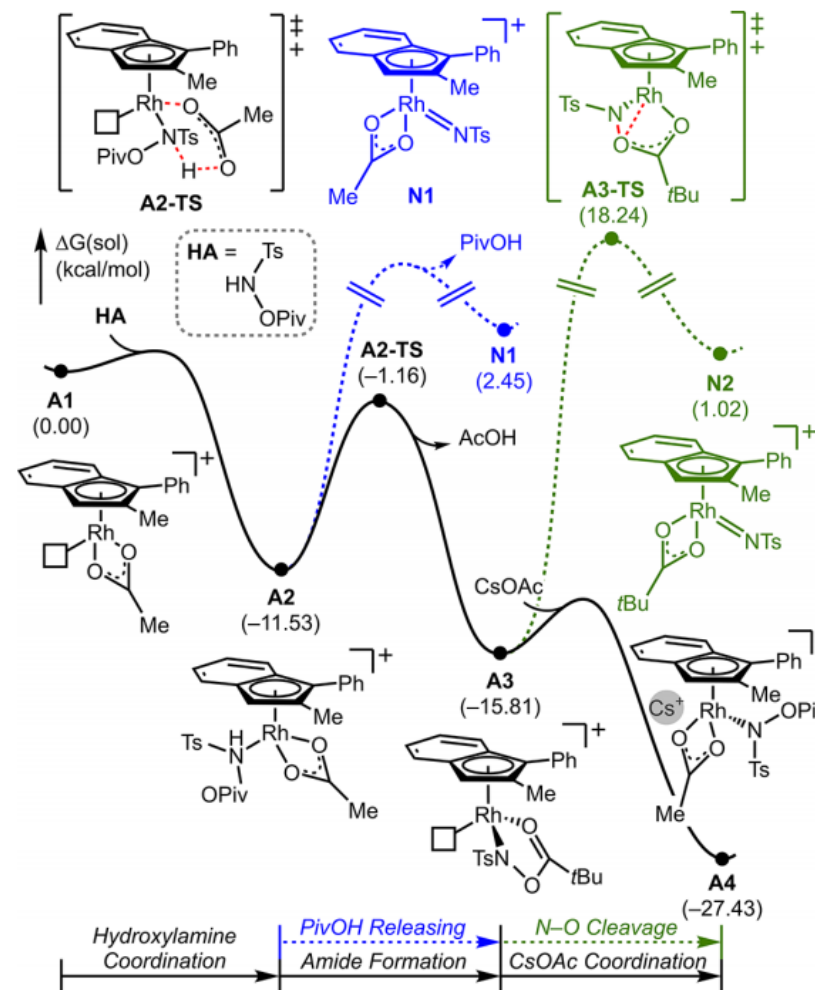


Figure 6. Free energy profile for the formation of A4 from A1. Blue and green traces represent direct nitrene formation via N–O bond cleavage.

4 Supporting Information



Table 1 Optimization of reaction conditions

Reaction scheme showing the conversion of **1a** to **2a** using AgNTf_2 (10 mol %), **L5** (10 mol %), PhIO (2 equiv), 4 Å MS (100 mg), CH_2Cl_2 (0.05 M), -10°C , 24 h.

	Variation from standard Entry conditions	Yield ^a (conversion, %)	ee ^b (%)
1	None	92 (98)	97
2	L1 instead of L5	75 (89)	35
3	L2 instead of L5	63 (77)	9
4	L3 instead of L5	71 (85)	12
5	L4 instead of L5	86 (96)	91
6	L6 instead of L5	78 (89)	93
7	AgClO_4 instead of AgNTf_2	74 (92)	93
8	AgOTf instead of AgNTf_2	73 (90)	76
9	-20°C , 24 h	26 (32)	96
10	AgNTf_2 (5 mol%), L5 (5 mol%)	73 (92)	93
11	2.5 mmol scale	89 ^c (>99)	95

Chemical structures of ligands **L1** through **L6** are shown below the table.

L1 R = *t*Bu
L2 R = Ph

L3

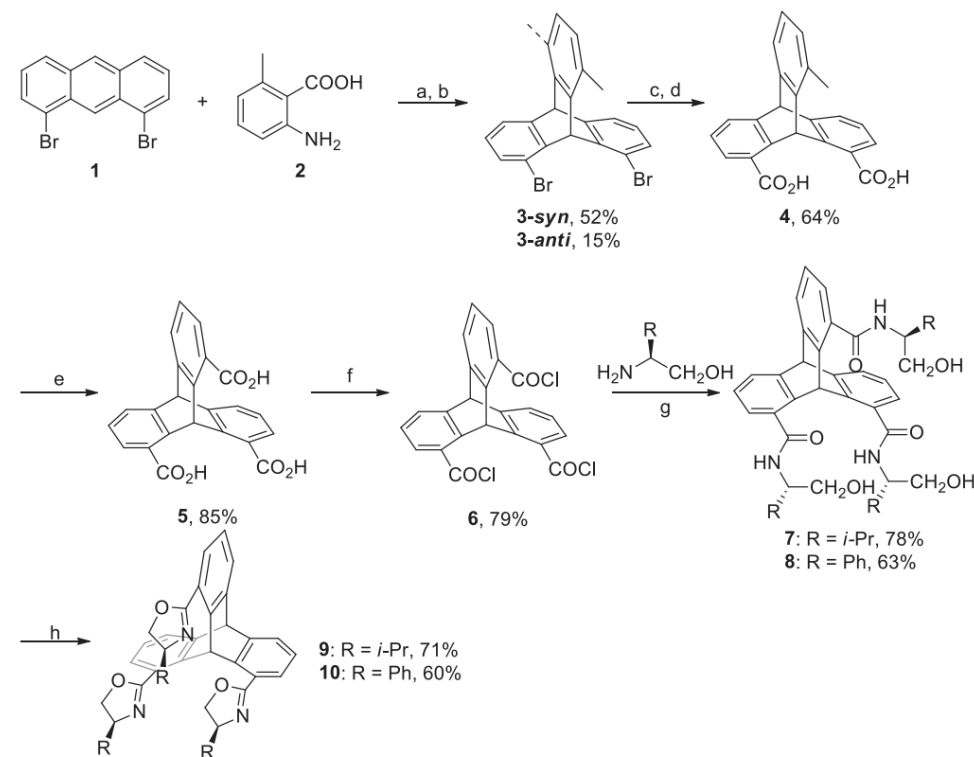
L4

L5 -CH₂CH₂- **L6** -(CH₂)₄-

^a Reactions were carried out on a 0.1 mmol scale unless indicated otherwise. Yields were determined by ¹H NMR using 1,3,5-trimethoxybenzene as an internal standard. ^b The ee was determined by chiral HPLC after purification. ^c Isolated yield.

I. Kisets and D. Gelman

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a) isopentylinitrite, DME, 80 °C; b) maleic anhydride, triglyme, reflux; c) BuLi, TMEDA, THF, -78 °C; d) CO₂, HCl; e) NaOH_{aq} (2%), KMnO₄, reflux; f) SOCl₂, CHCl₃, reflux; g) TEA, DCM, RT; h) DDQ, PPh₃, DCM, reflux.

4 Supporting Information

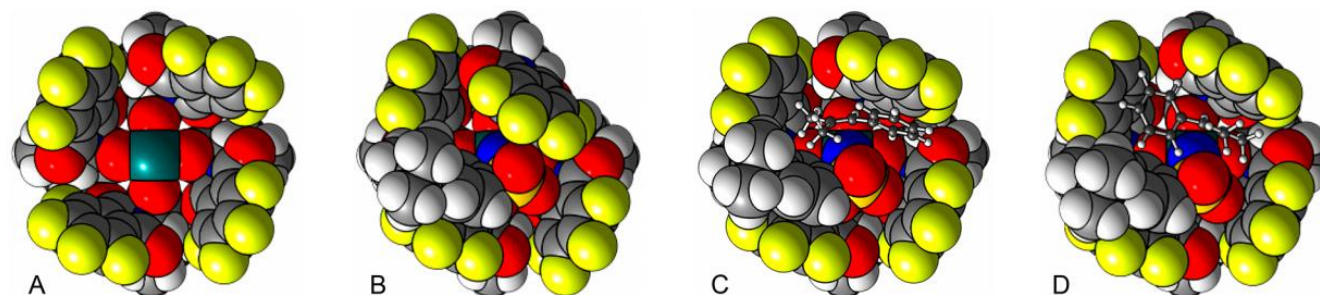


Figure 2. Views showing $^3\text{Rh}_2$ (A), $^3\text{Rh}_2\text{N}$ (B), $^3\text{RhN:Styr}_{\text{Si}}$ (C), and $^3\text{RhN:Styr}_{\text{Re}}$ (D). Different representations are used to provide evidence that the substrate (ball and sticks) fits in the catalytic pocket (van der Waals spheres). H, white; C, gray; N, blue; O, red; F, neon green; S, yellow.

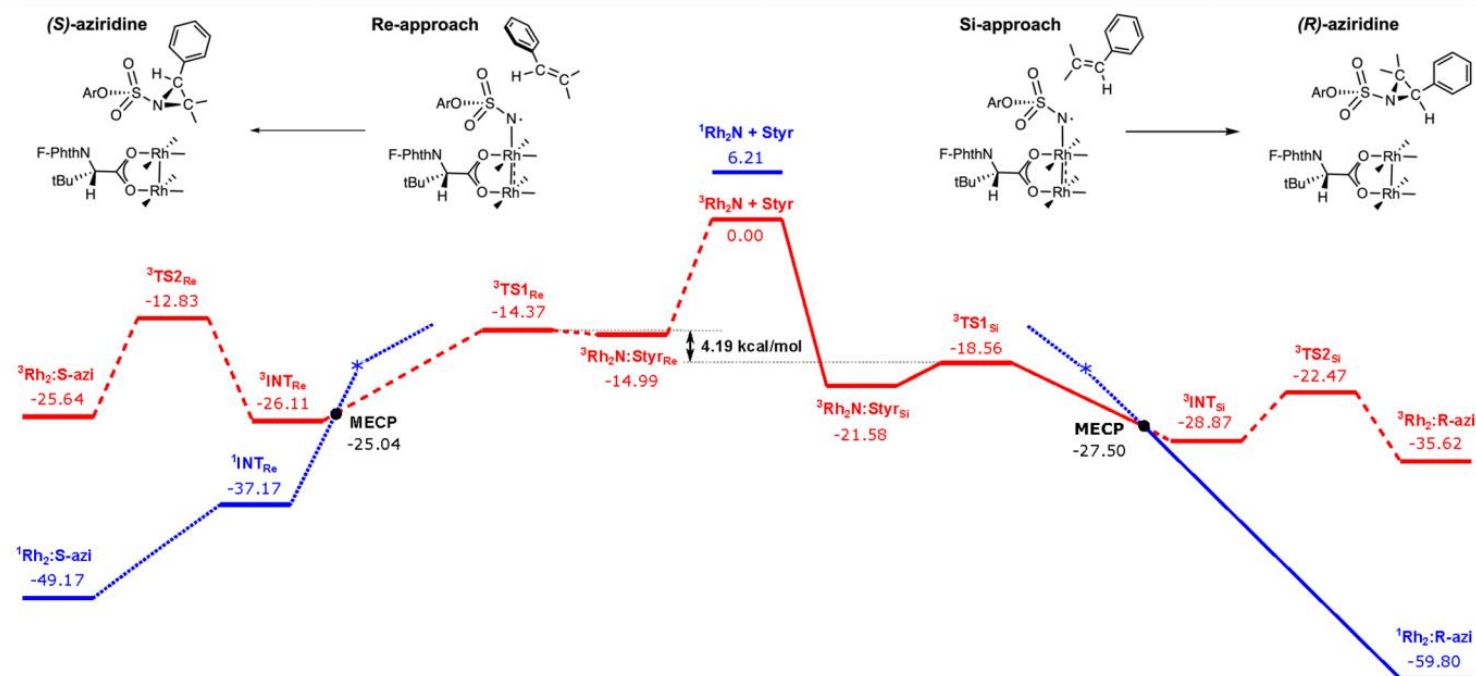
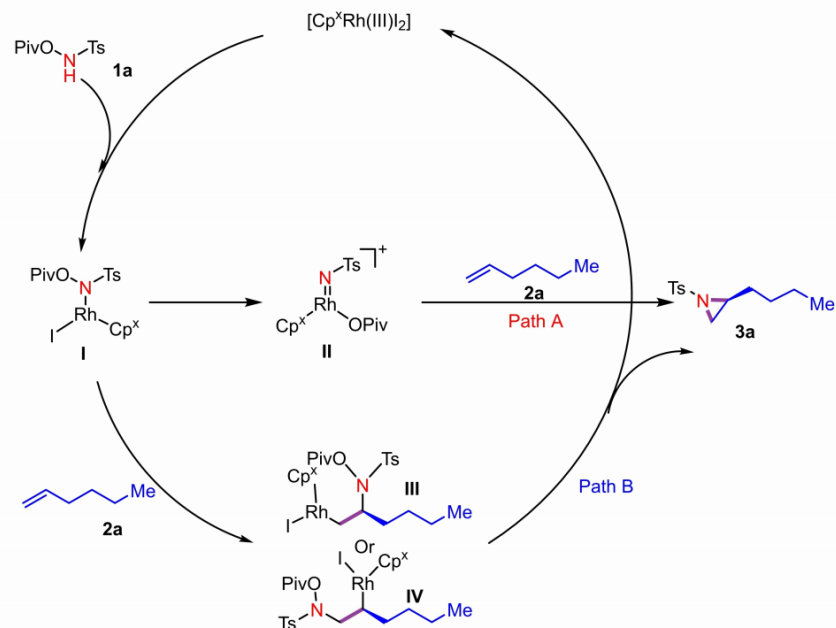


Figure 3. Extract of the computed two spin state mechanism (triplet in red, singlet in blue) showing the favored energy profile in plain line ($\Delta G_{258\text{ K}}$ in kcal/mol); * indicates the position of pseudo-TS.

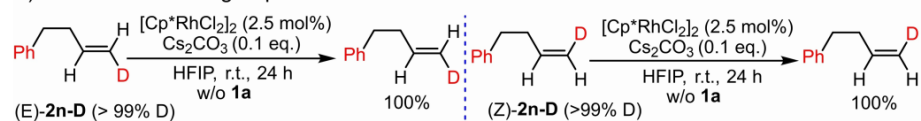
4 Supporting Information



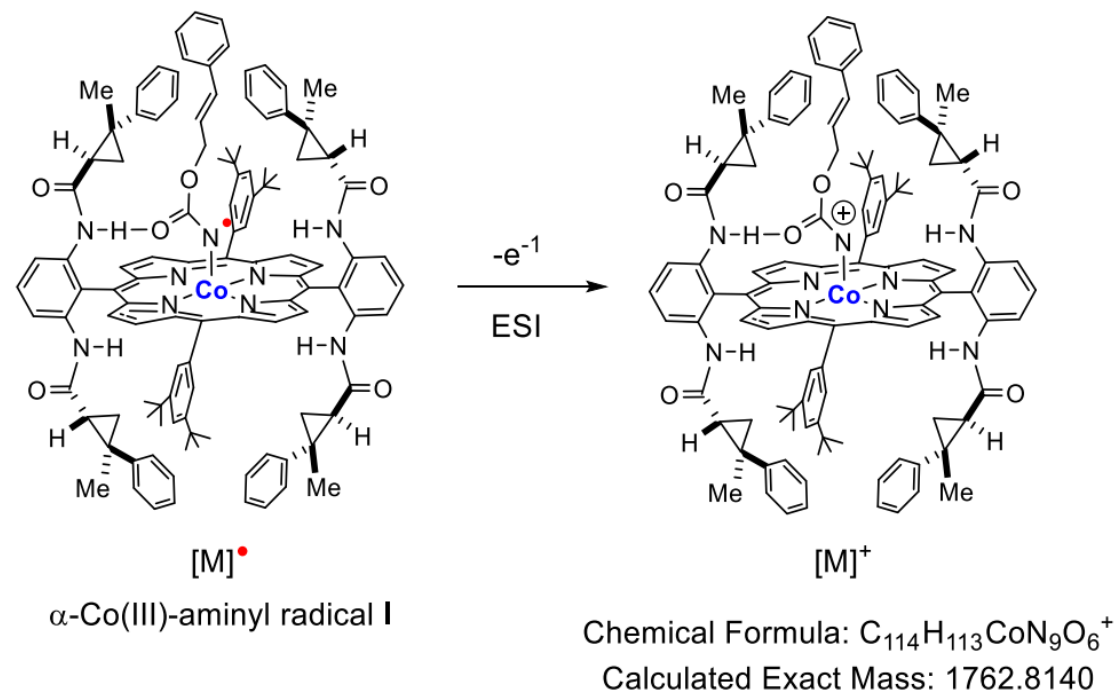
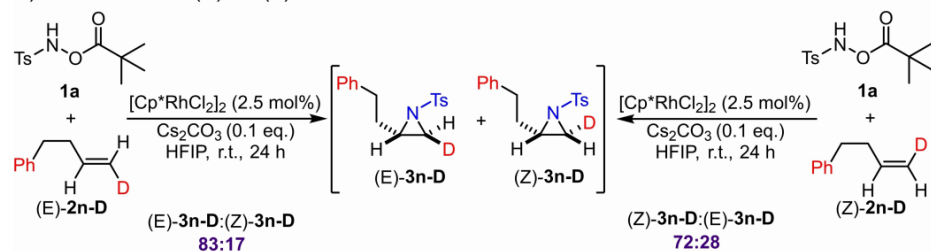
A) Proposed Catalytic Cycle



B) Deuterium Labeling Experiment



C) Aziridination of (E)- or (Z)-2n-D



6

1) $\text{CH}_2=\text{CHMgBr}$ (2.2 equiv)
 THF , -78°C to rt.
 2) $\text{Ru}(\text{NO})\text{Cl-salen}$ (5 mol %),
 air, toluene, hv, 25°C .

7: 52% from **6**

$\text{Ru}(\text{CO})\text{-salen } \mathbf{1}$ (0.5 mol %)
 SESN_3 (1.0 equiv)
 CH_2Cl_2 , N_2 , MS 4 Å, 25°C

8: 87%, >99% ee

Hünig's base (1 mol %)
 1,4-dioxane/ H_2O (4/1), 0°C

9
 (after recrystallization >99% ee)

1) NaBH_4 (2 equiv), EtOH , -78°C
 or
 2) L-selectride (1.5 equiv),
 THF , -78°C

10
 1) 99%, >99% ee, *trans*:*cis* = 7:1
 2) 81%, >99% ee, *trans*:*cis* = 2:>98

$[\text{CH}_2=\text{CHS}(\text{Ph})_2]^+\text{TfO}^-$ (1.1 equiv)
 DBU (2 equiv)
 CH_2Cl_2 , 0°C to rt

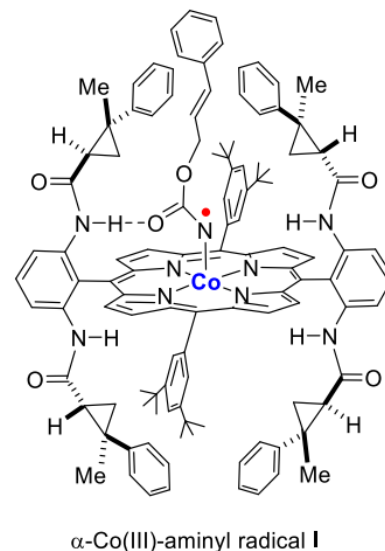
11
trans-**11**: 78% from **9**
cis-**11**: 72% from **9**

TBAF (2 equiv), THF , 60°C for 6 h
 then 1-bromopropane (4 equiv)
 60°C , 12 h
 TBAF = tetrabutylammonium fluoride

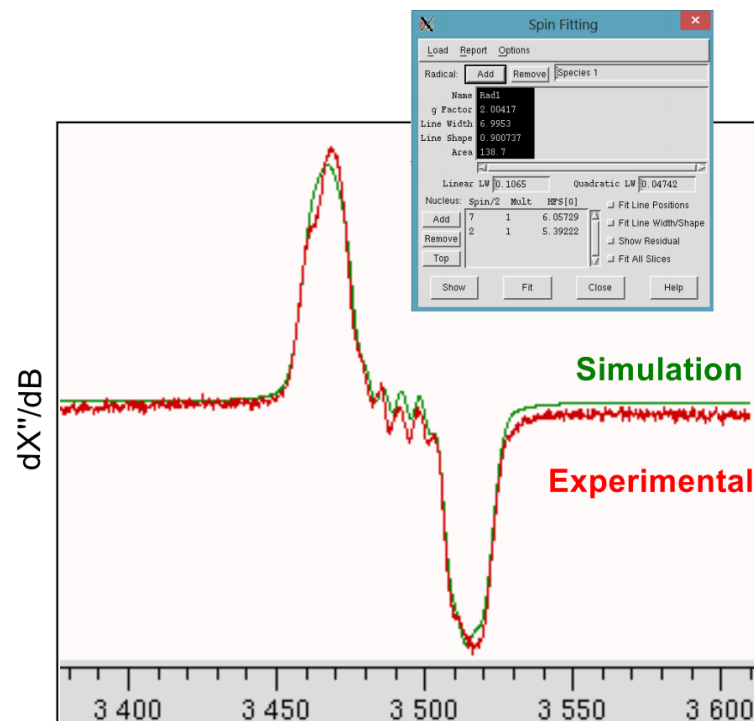
12
trans-**12**: 78%
cis-**12**: 62%

1 step
 from *trans*-**12**

(+)-PD 128907

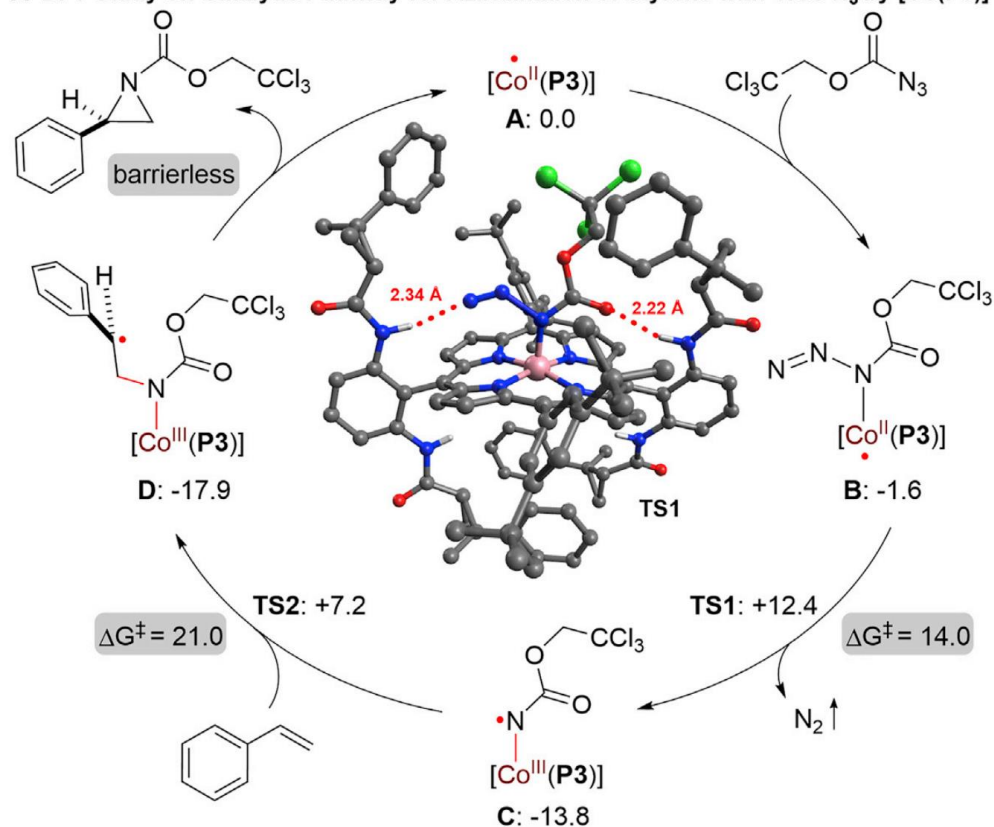


From experimental: g_{iso} : 2.00406



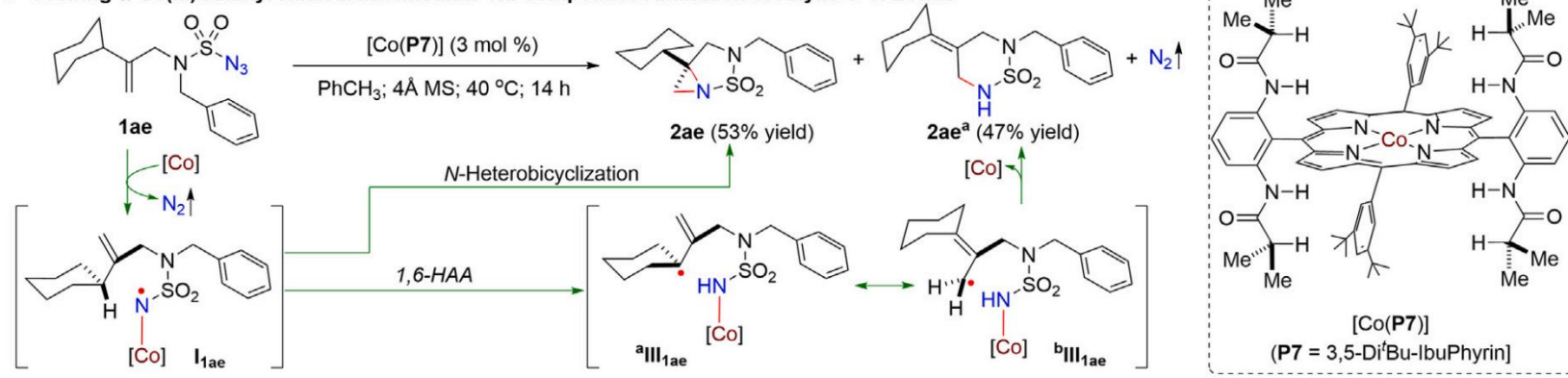
4 Supporting Information

A DFT Study on Catalytic Pathway for Aziridination of Styrene with Troc-N₃ by [Co(P3)]^a

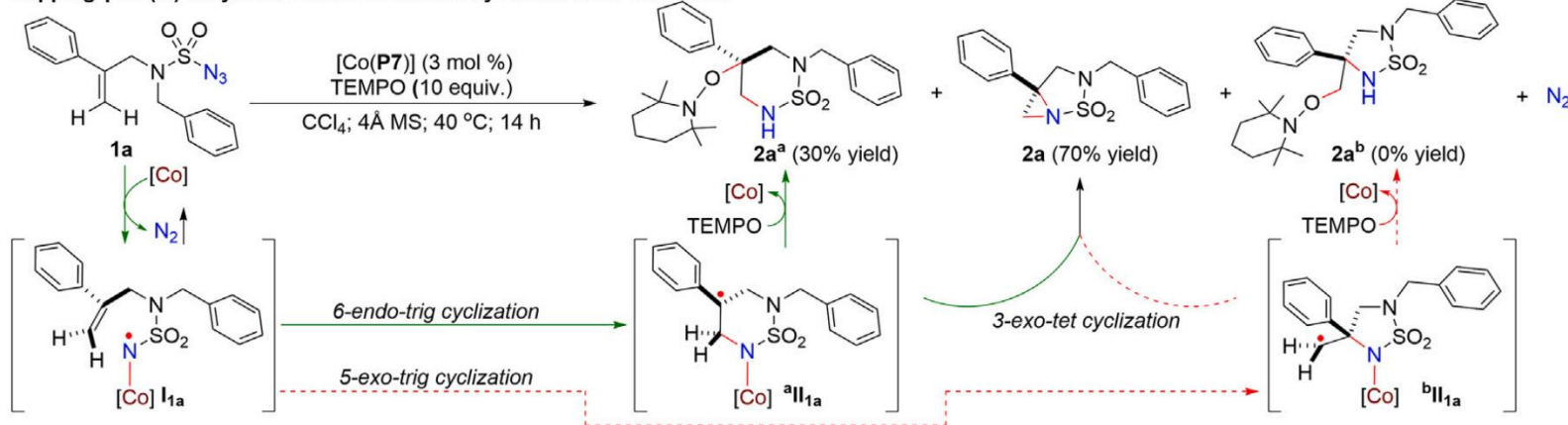


4 Supporting Information

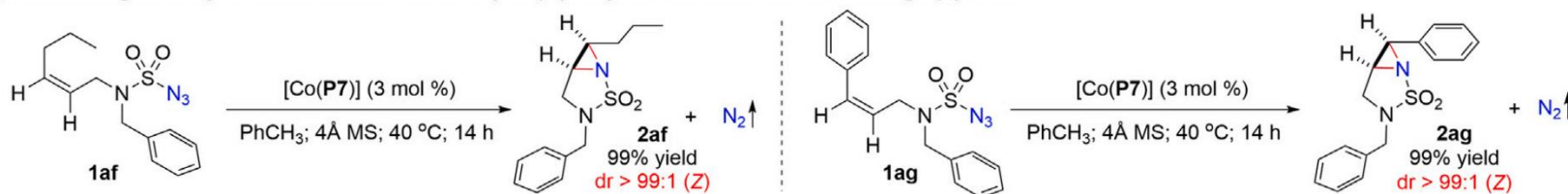
B Probing α -Co(III)-Aminyl Radical Intermediate via Competitive Amination of Allylic C-H Bonds^b



C Trapping γ -Co(III)-Alkyl Radical Intermediates by Stable TEMPO Radical^c



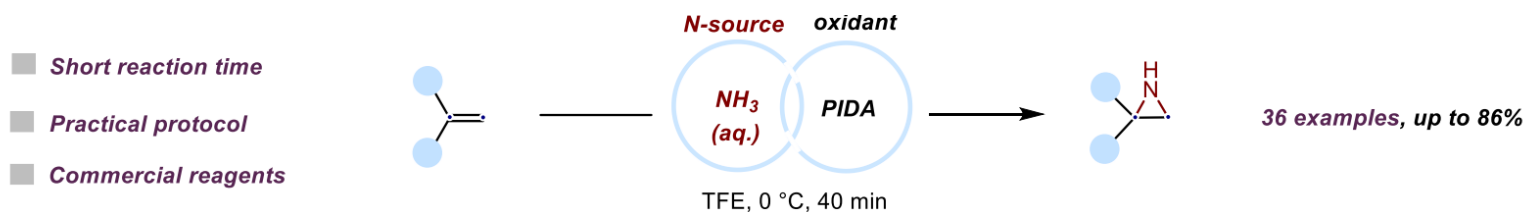
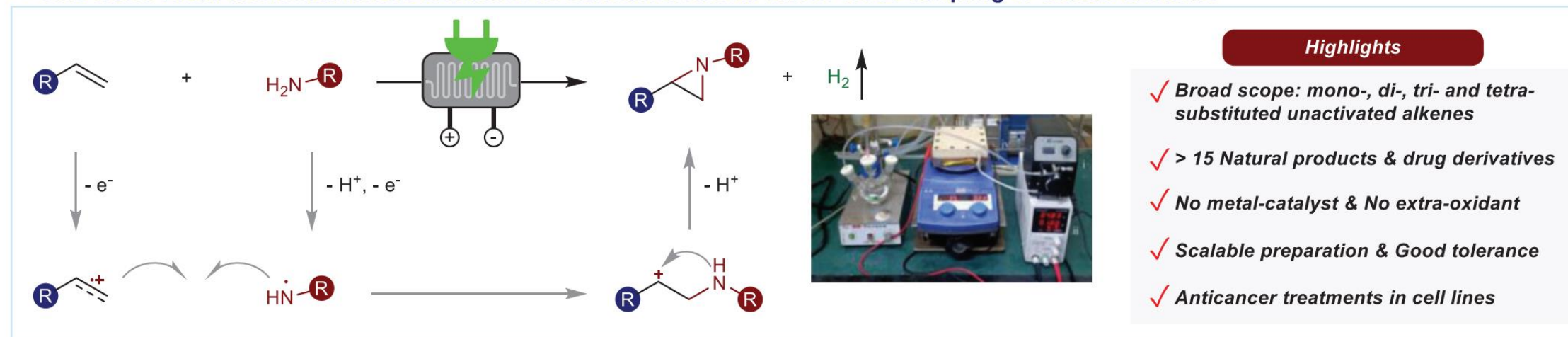
D Examining Stability of Chiral Radical Center in γ -Co(III)-Alkyl Radical Intermediates through (Z)-Olefins^b



4 Supporting Information



D This work: Electrochemical alkene aziridination via radical/radical cation cross-coupling to access aziridine



● Conditions A: PIDA (2 equiv), NH₃ (30 equiv)

