



Literature Report

Gold-Catalyzed C-C Coupling Reactions

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Supervisor: Prof. Zhang-Jie Shi

2021-04-09

Background

- Background
- Gold-Catalyzed C-C Coupling Reactions
 - Oxidants-Assisted Coupling Reactions
 - Photo-Assisted Coupling Reactions
 - Dinuclear Gold-Catalyzed Coupling Reactions
- Summary

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Background

- **The source of gold**



- **The role of gold in the life**



- **The role of gold in the chemistry**



Alloy



Catalysts

Background

Relativistic Effects

79	Au
	Gold
	$5d^{10}6s^1$

$$m = m_0 / [1 - (v/c)^2]^{1/2}$$

Au 1s electron as an example

$$\frac{\langle v \rangle}{c} \approx \frac{79}{137} = 0.58$$

$$m = 1.23 m_0$$

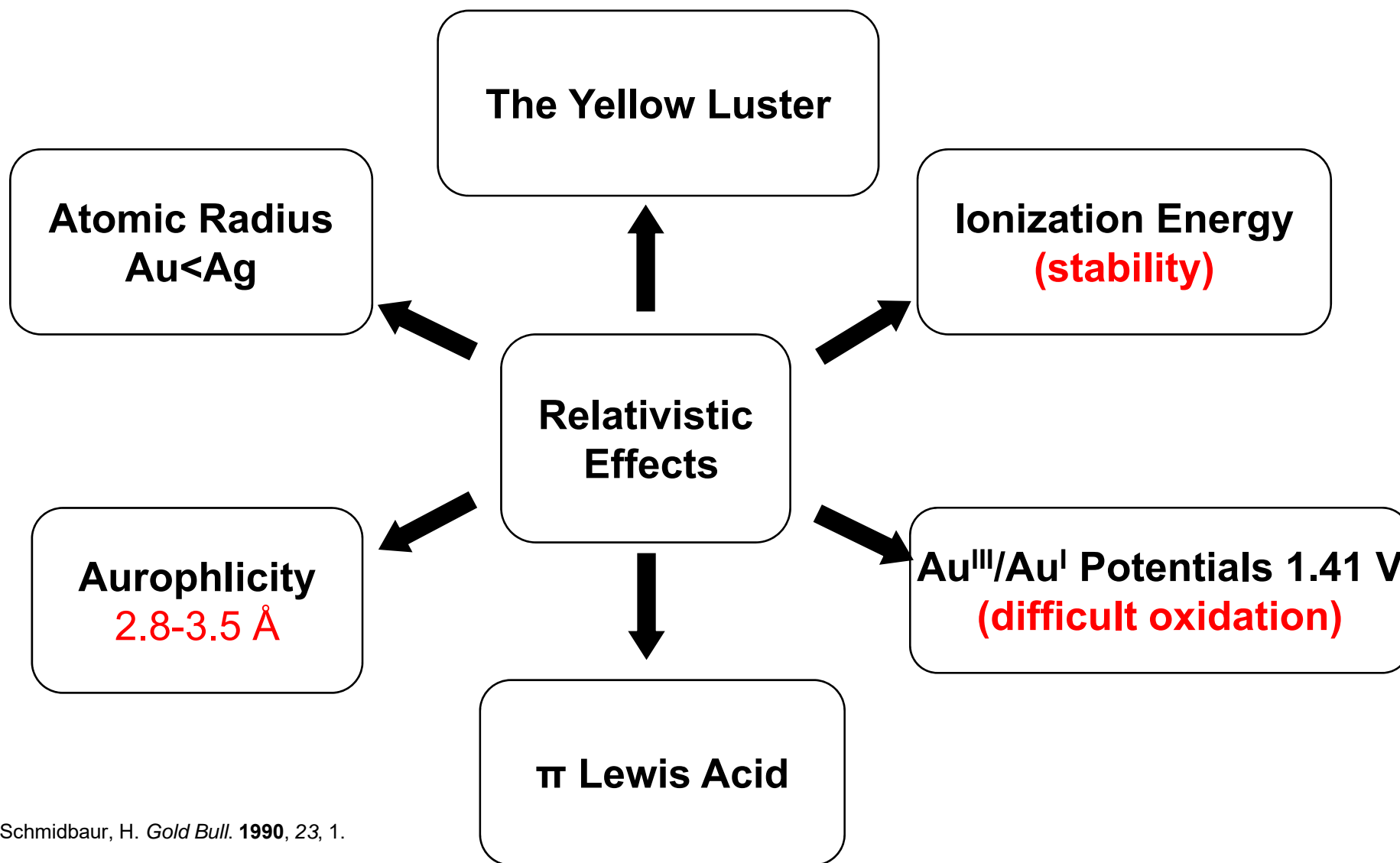
$$r_n = \alpha_0 \frac{n^2}{m_0} \quad \longrightarrow \quad \frac{r_{1s(R)}}{r_{1s(NR)}} = 0.81$$

	0.128 745 Cu 1356 337		
	0.1445 731 Ag 1234 285		
0.1385 866 Pt 2042 469	0.1442 890 Au 1337 343	0.151 1007 Hg 234 59	

a	b
M	
c	d

- Contraction and stabilization of 6s and 6p orbitals
- Expansion and instability of 5d and 4f orbitals

Background



Schmidbaur, H. *Gold Bull.* **1990**, 23, 1.

Schmidbaur, H. *Gold Bull.* **2000**, 33, 3.

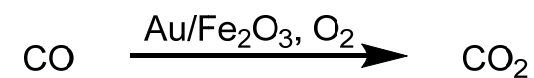
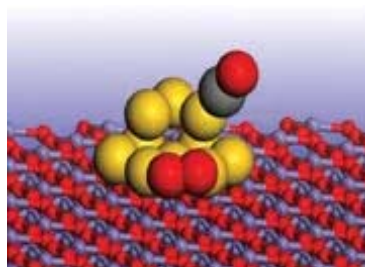
Blinder, S. M. et al. *J. Chem. Educ.* **2011**, 88, 71.

Background

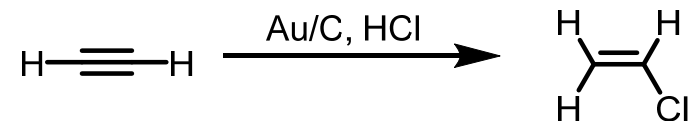
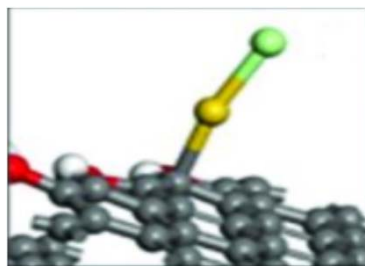
Nano Au as Catalysts



Masatake Haruta



Graham J. Hutchings

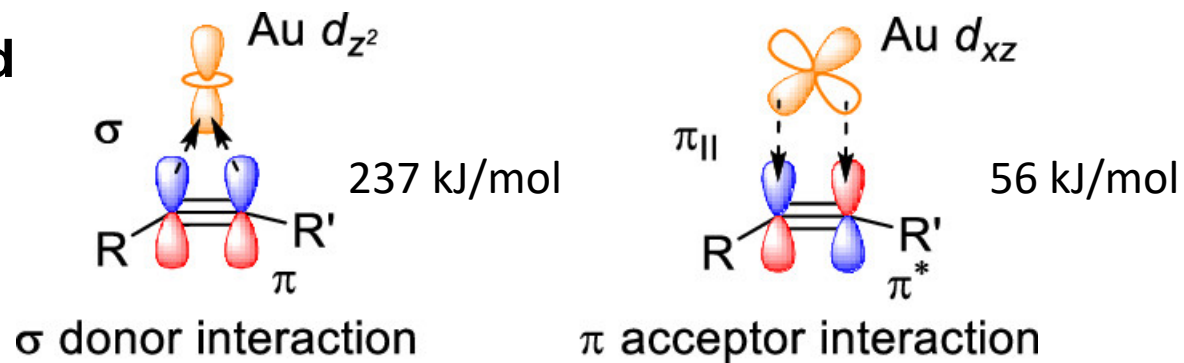


Haruta, M. et al. *J. Catal.* **1989**, *115*, 301.

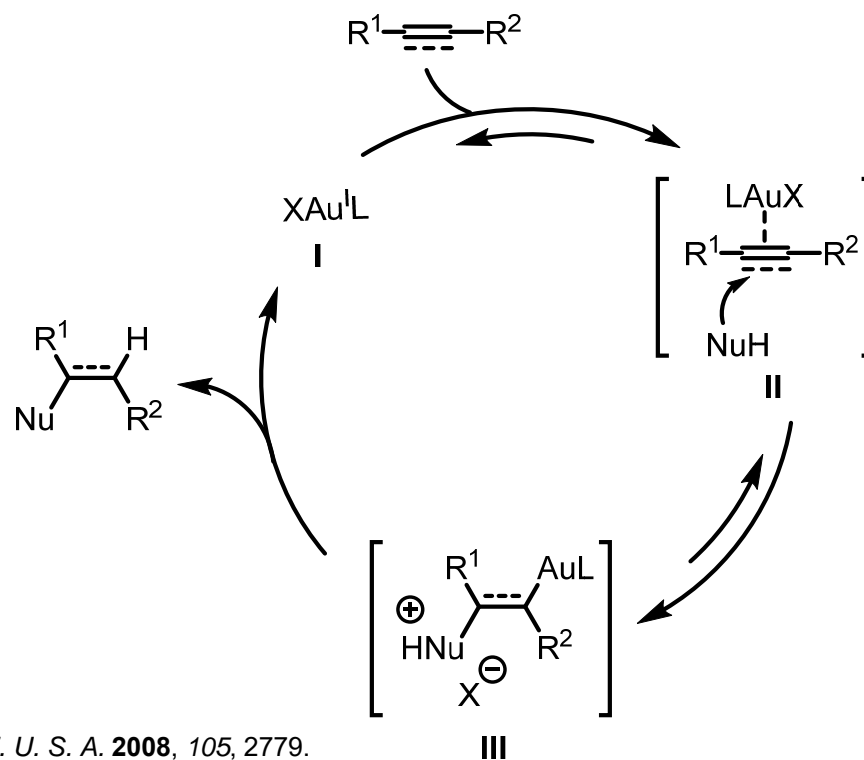
Hutchings, G. J. et al. *J. Catal.* **1985**, *96*, 292.

Background

Au as π Lewis Acid

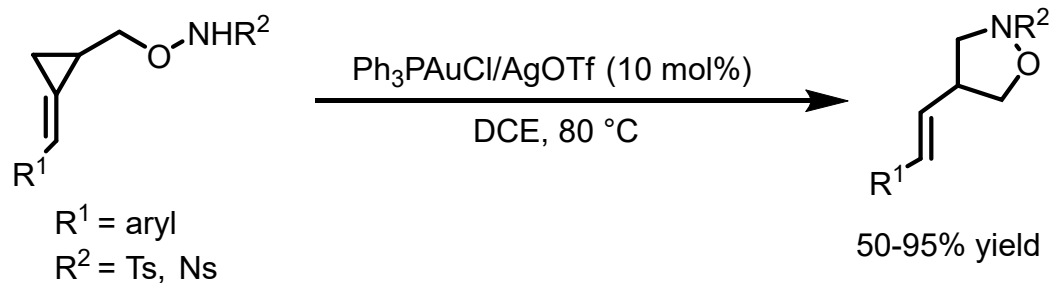


Simplified cationic gold catalytic cycle

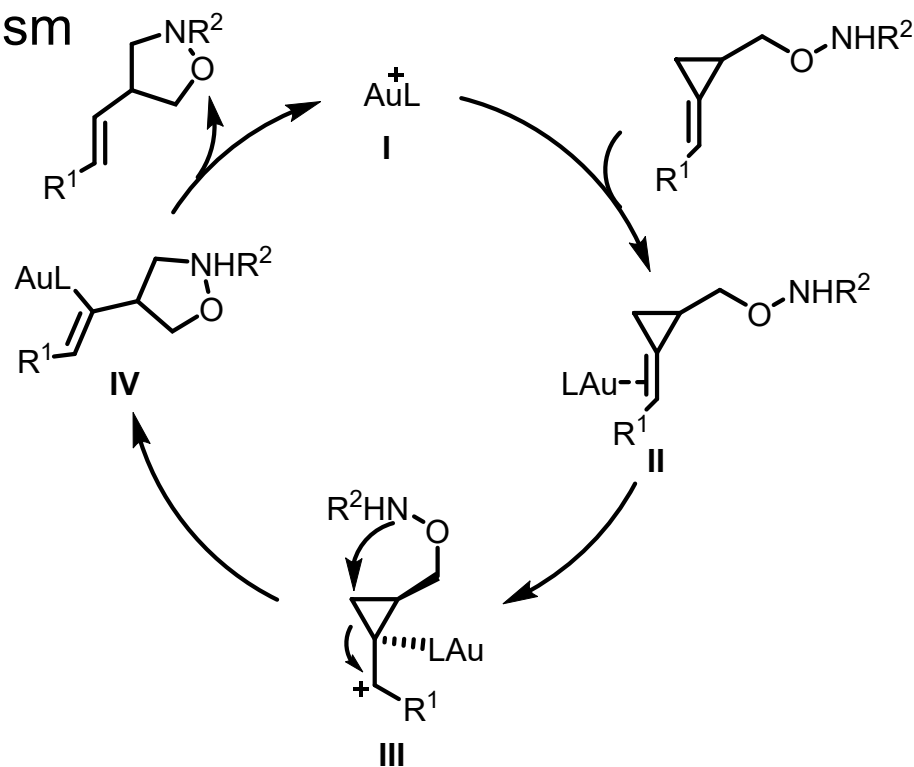


Background

Au^I-Catalyzed intramolecular hydroamination

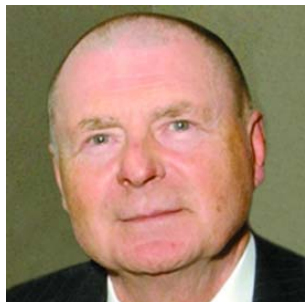


Proposed mechanism



Background

The Nobel Prize in chemistry 2010 for Pd-catalyzed cross couplings



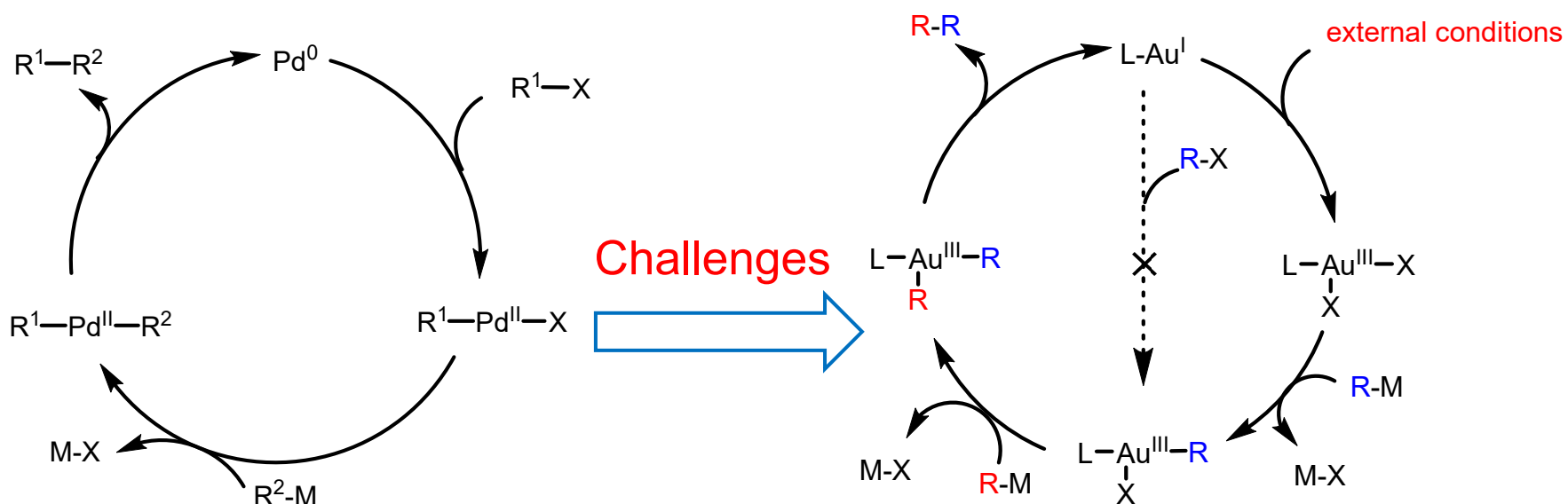
Richard F. Heck



Ei-ichi Negishi



Akira Suzuki

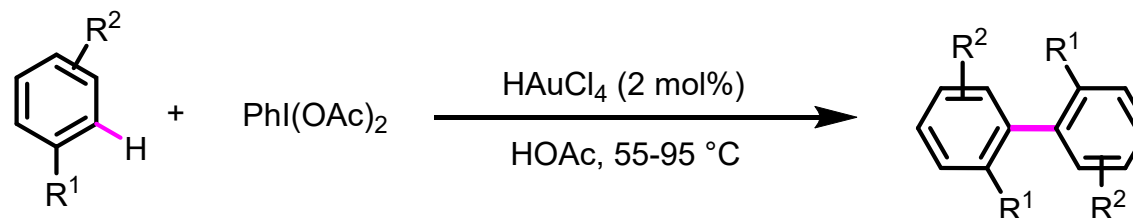


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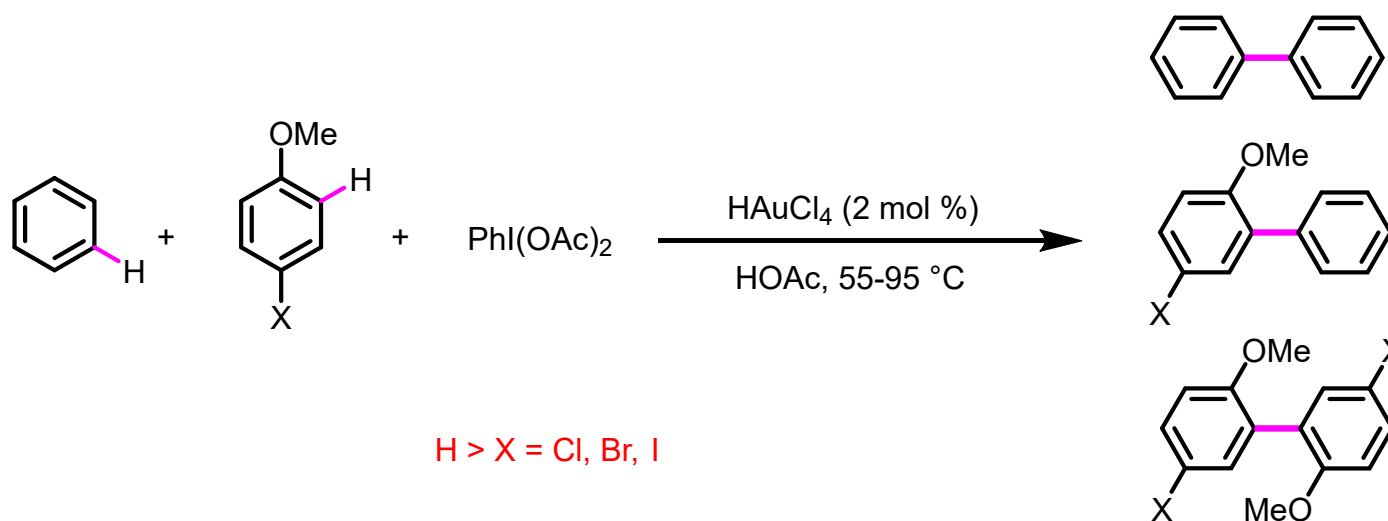
Oxidants-Assisted Coupling Reactions-I³⁺

Au^I and I³⁺-mediated coupling reactions of arenes



R¹ = H, Me, OMe, Cl; R² = H, Me, ^tBu, F, Cl, Br, I, NO₂, COOMe

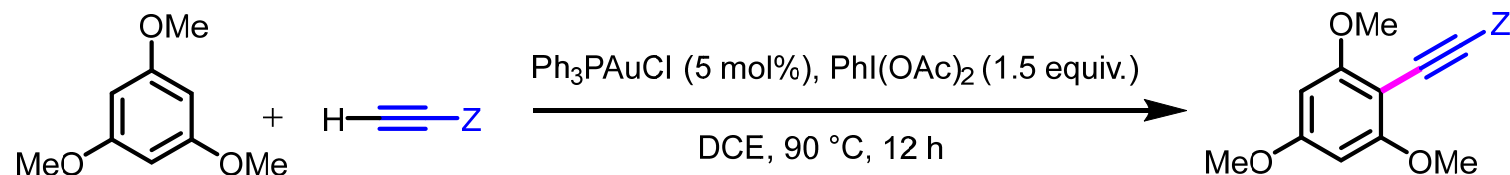
Tse, M. K. et al. *Chem. Commun.* **2008**, 386.



Tse, M. K. et al. *J. Organomet. Chem.* **2009**, 694, 524.

Oxidants-Assisted Coupling Reactions-I³⁺

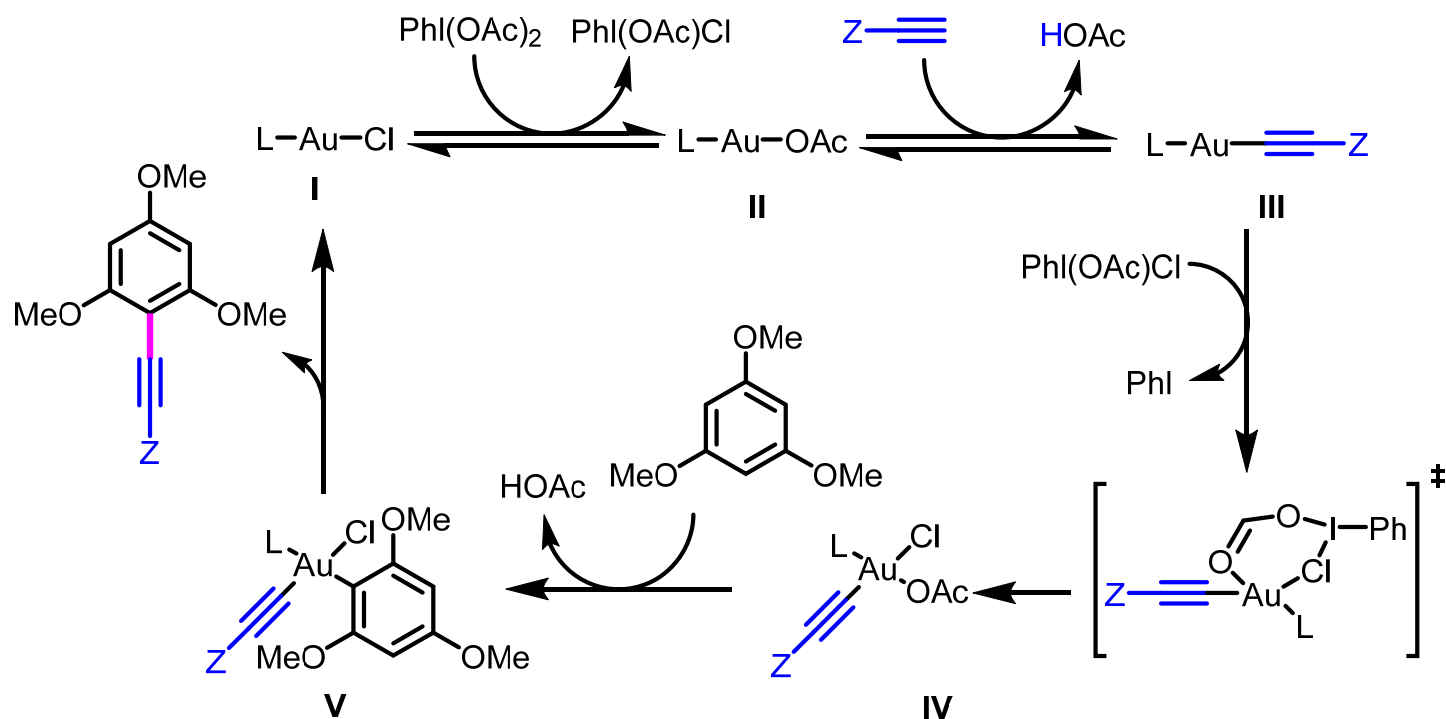
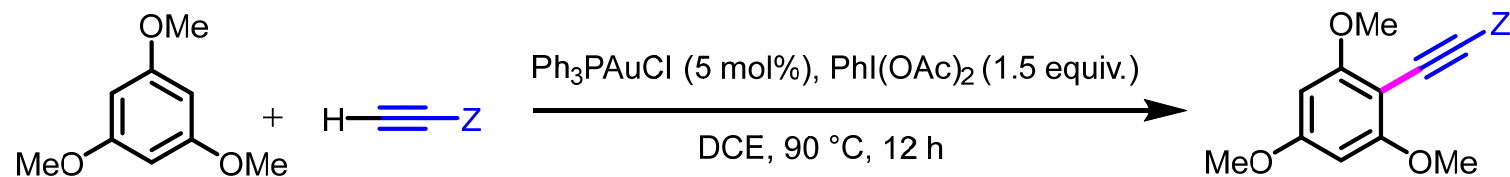
Au^I and I³⁺-mediated coupling reactions of arenes and alkynes



entry	substrate	yield %
1	Z = CO ₂ Et	75
2	Z = CO ₂ ^t Bu	60
3	Z = COPh	72
4	Z = CO(3,5-dimethoxy-phenyl)	68
5	Z = CO(<i>p</i> -CF ₃ C ₆ H ₄)	70
6	Z = CO ^t Bu	31
7	Z = CO(C ₇ H ₁₂)	66
8	Z = (CH ₃)C=CH ₂	48
9	Z = Ph	25

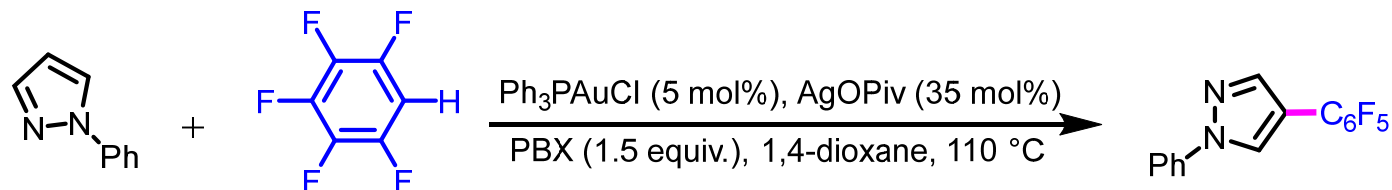
Oxidants-Assisted Coupling Reactions-I³⁺

Au^I and I³⁺-mediated coupling reactions of arenes and alkynes

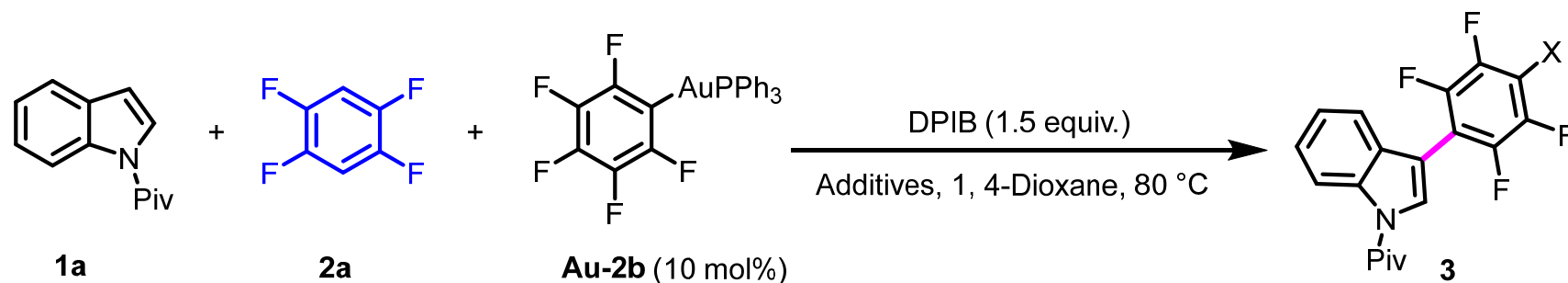


Oxidants-Assisted Coupling Reactions-I³⁺

Au^I/Ag^I bimetallic and I³⁺-mediated coupling reactions of arenes



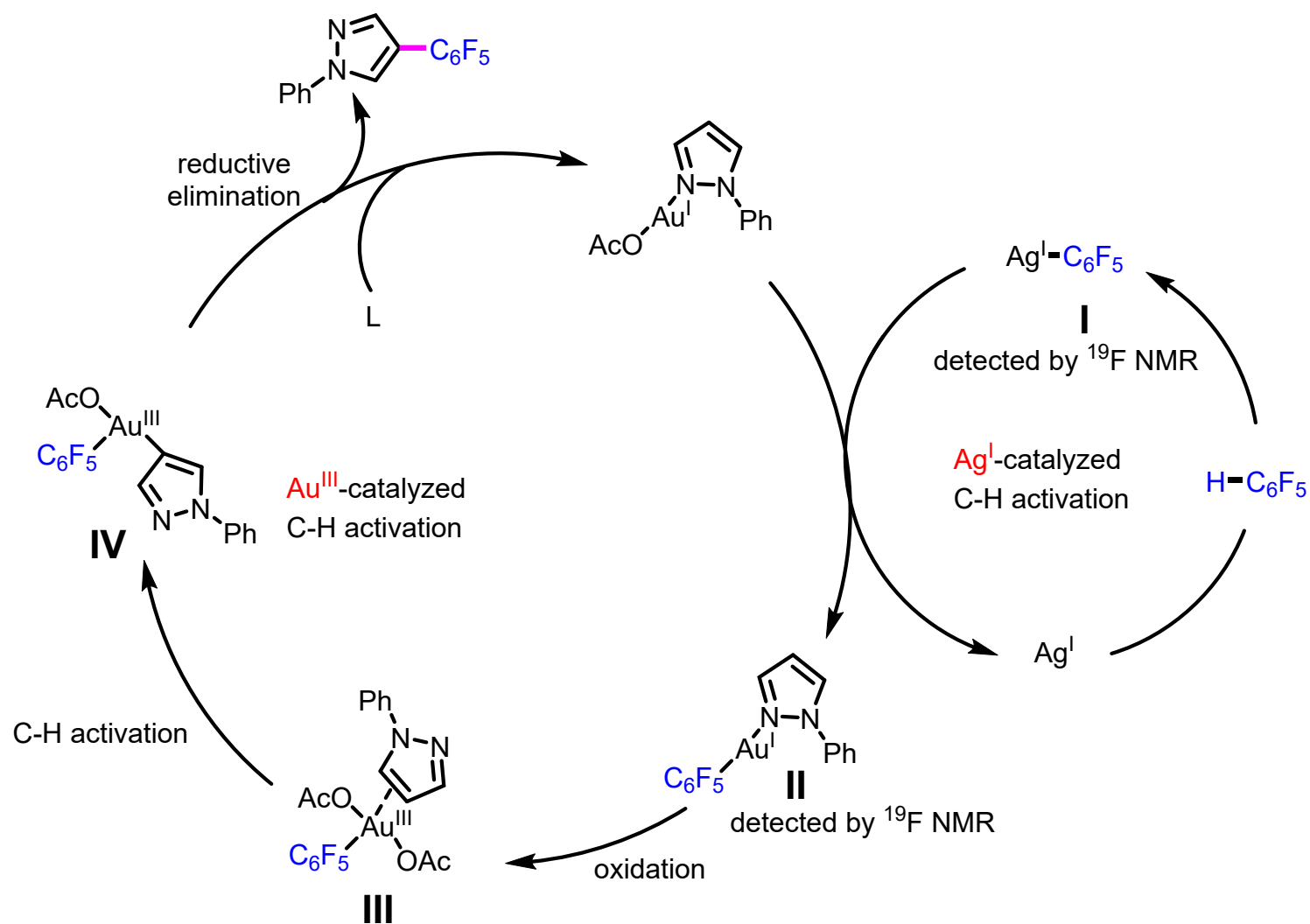
Initial Search for Conditions



	Additives	3a (X = F) %	3b (X = H) %
Stoichiometric reaction of Au-2b	None	9	None
	AgOPiv (20 mol%) K ₂ CO ₃ (2 equiv.)	None	None
Catalytic turnover	AgOPiv (20 mol%)	5	16

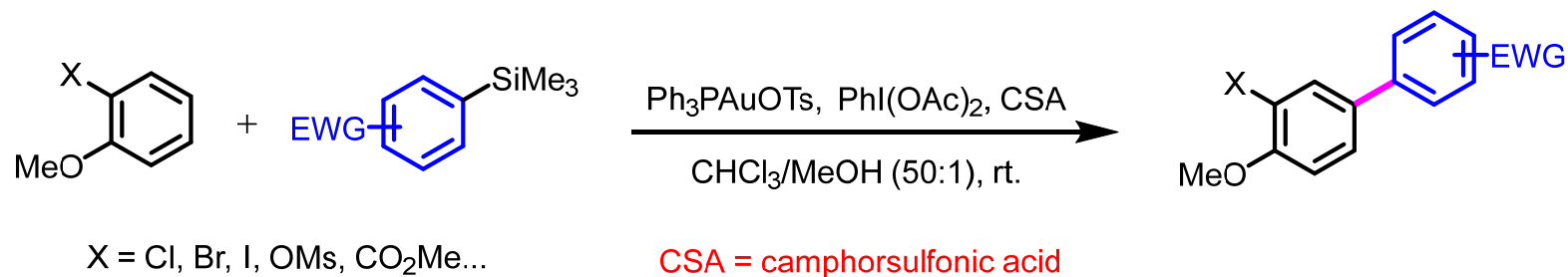
Oxidants-Assisted Coupling Reactions-I³⁺

Proposed mechanism



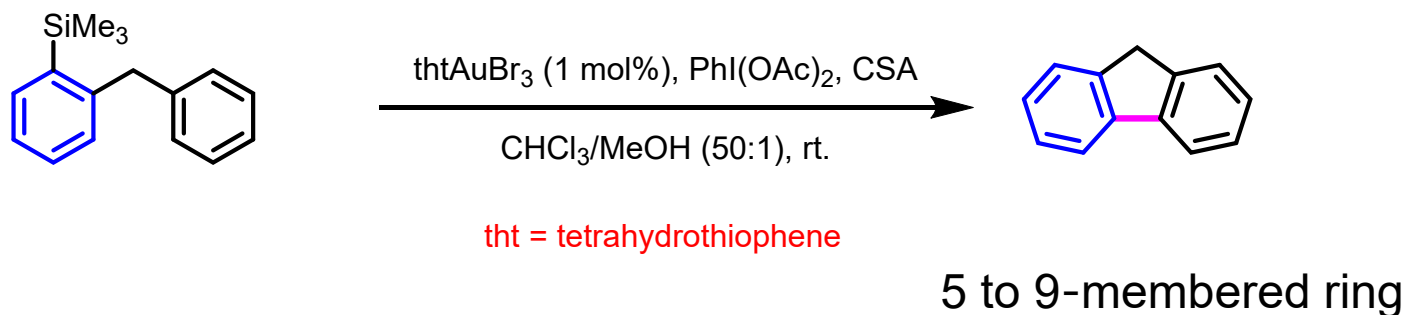
Oxidants-Assisted Coupling Reactions-I³⁺

Au^I and I³⁺-mediated arylation of ArSiMe₃



Mild reaction conditions; High FG tolerance; High selectivity; High yield

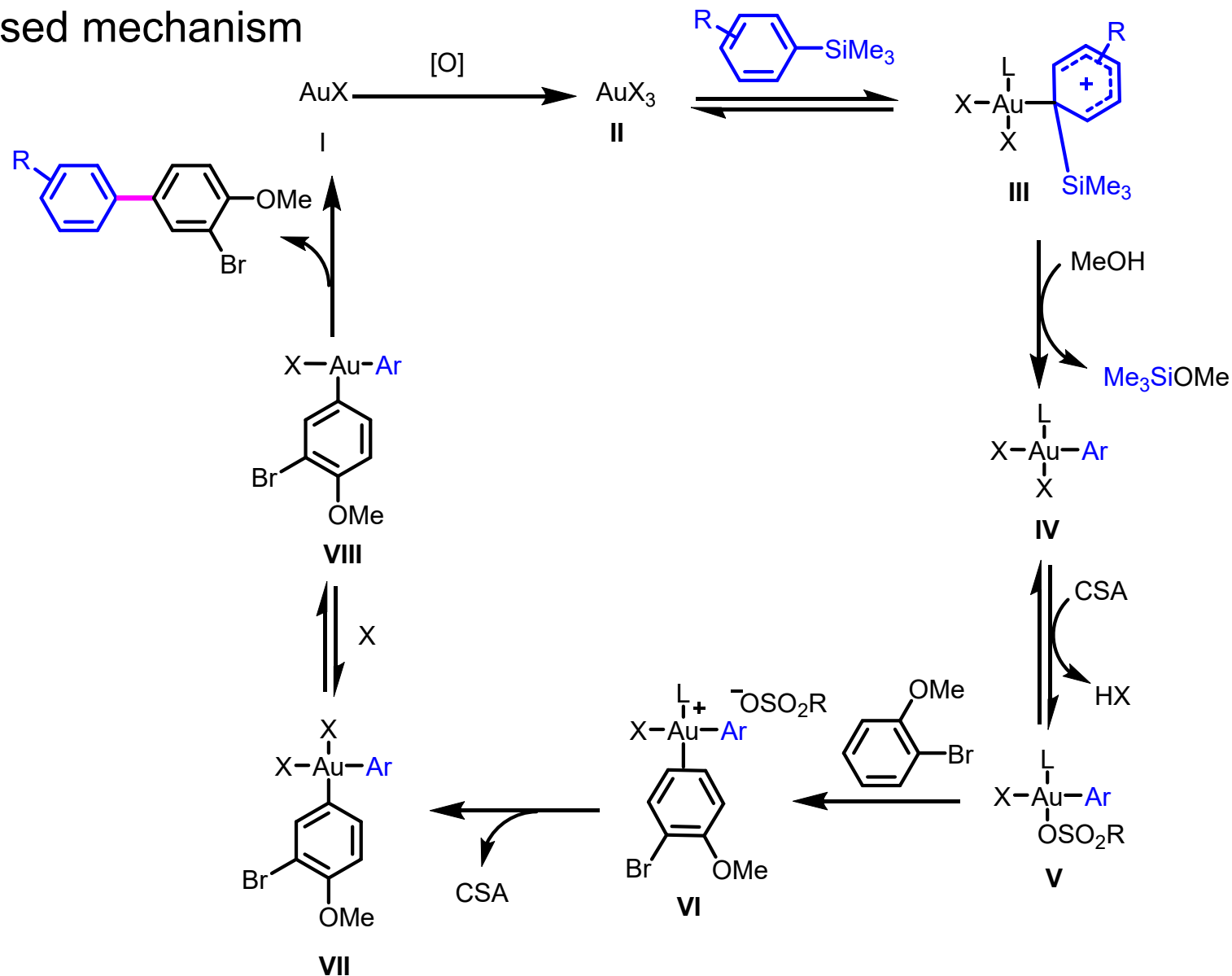
Lloyd-Jones, G. C. et al. *Science* **2012**, 337, 1644.



Lloyd-Jones, G. C. et al. *J. Am. Chem. Soc.* **2017**, 139, 245.

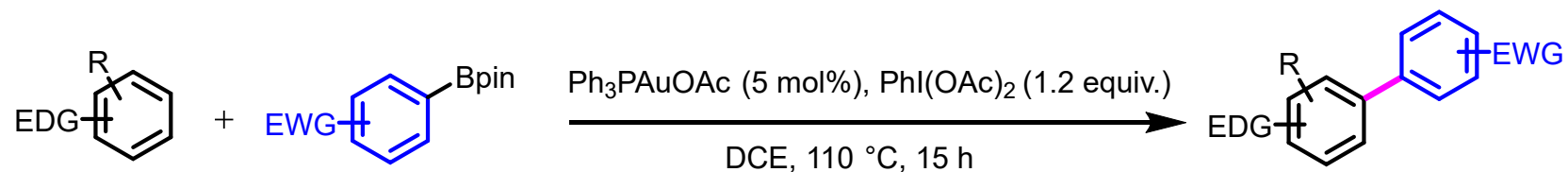
Oxidants-Assisted Coupling Reactions-I³⁺

Proposed mechanism

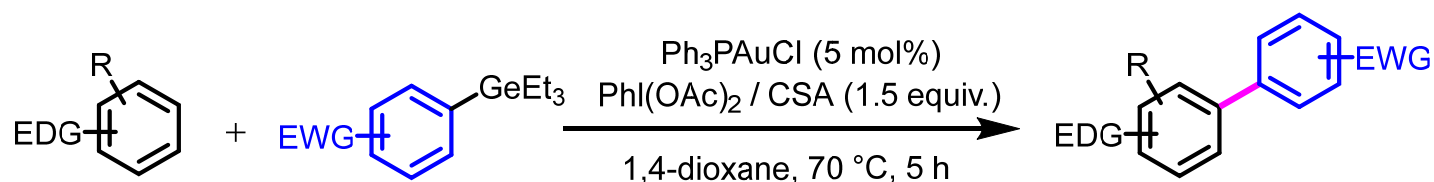


Oxidants-Assisted Coupling Reactions-I³⁺

Au^I and I³⁺-mediated arylation of ArBpin or ArGeEt₃

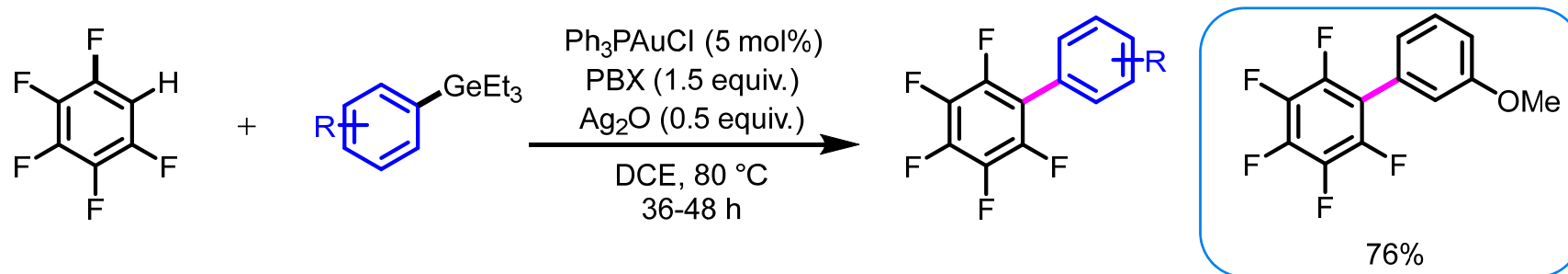


Nevado, C. et al. *Angew. Chem. Int. Ed.* **2017**, 56, 1021.



GeEt₃ > SiMe₃ > BR₂; R = H > I > Br > Cl High selectivity

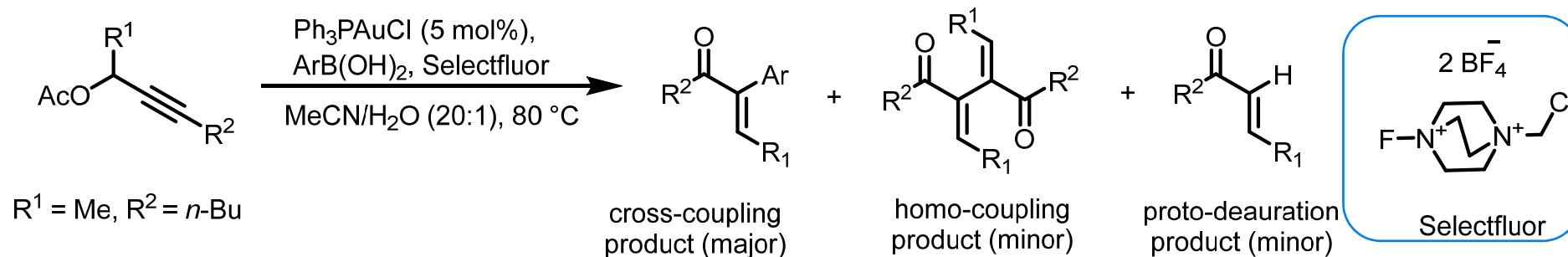
Schoenebeck, F. et al. *ACS Catal.* **2019**, 9, 9231.



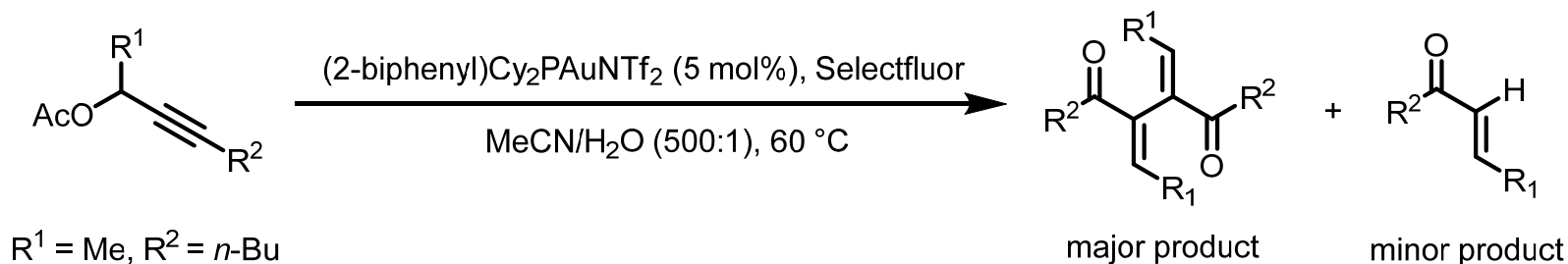
Schoenebeck, F. et al. *J. Am. Chem. Soc.* **2020**, 142, 7754.

Oxidants-Assisted Coupling Reactions-Selectfluor

Au^I- and selectfluor-mediated coupling reactions of propargylic esters



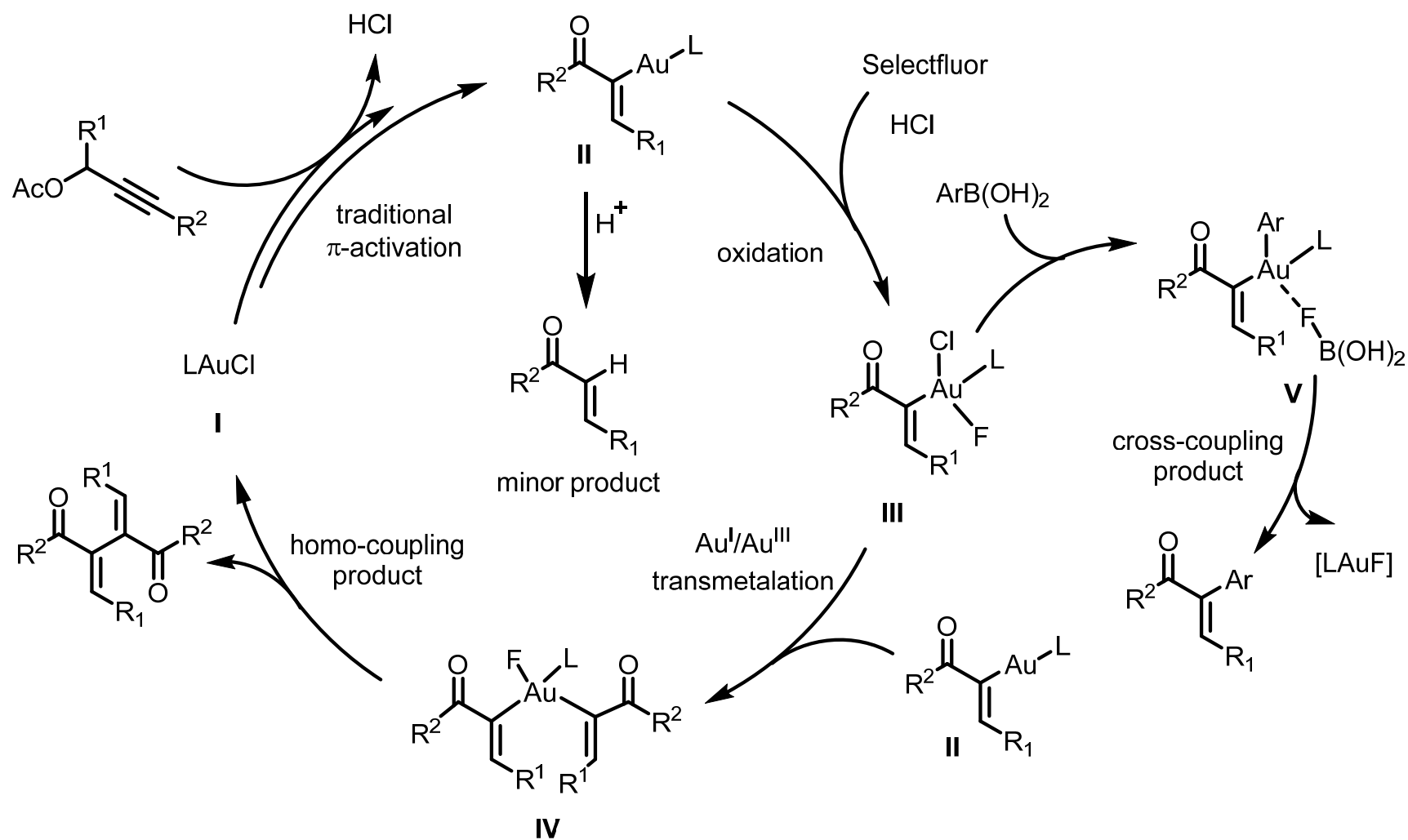
Zhang, L. et al. *Angew. Chem. Int. Ed.* **2009**, *48*, 3112.



Zhang, L. et al. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 3884.

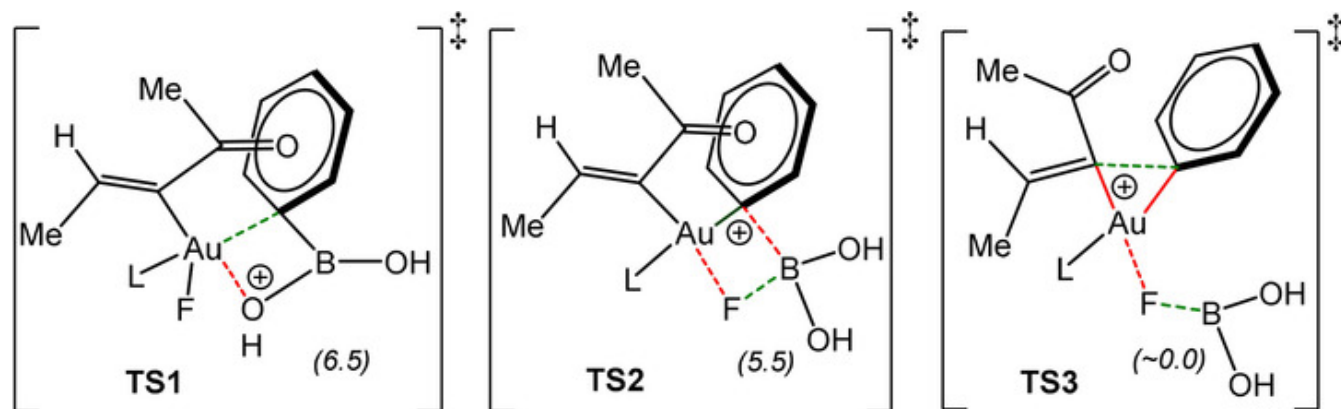
Oxidants-Assisted Coupling Reactions-Selectfluor

Proposed mechanism

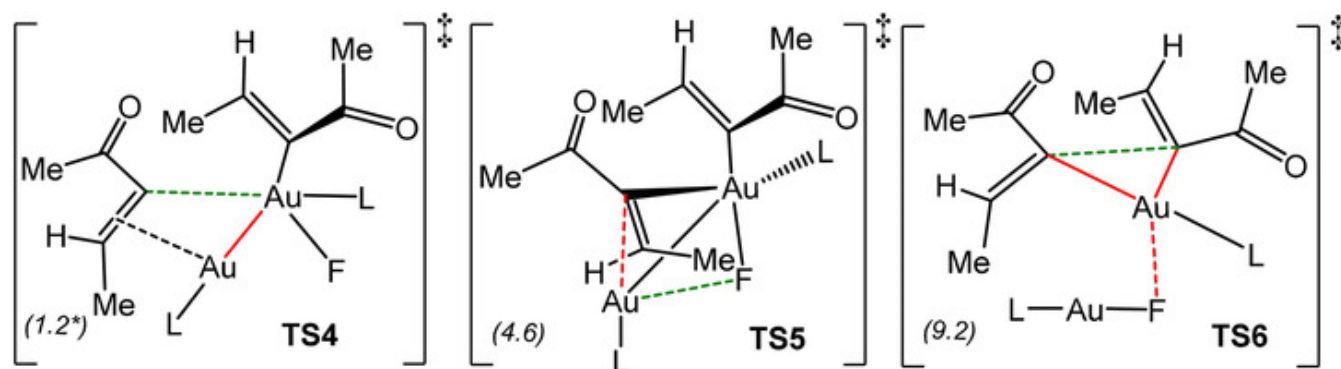


Oxidants-Assisted Coupling Reactions-Selectfluor

DFT calculations in the presence of aryl boronic acids

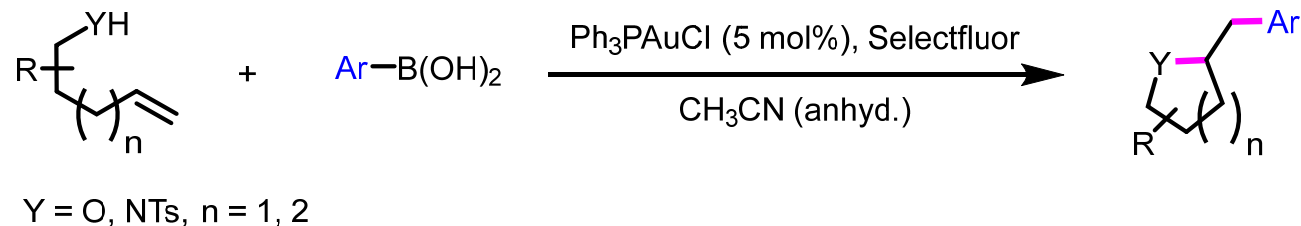


DFT calculations in the absence of aryl boronic acids

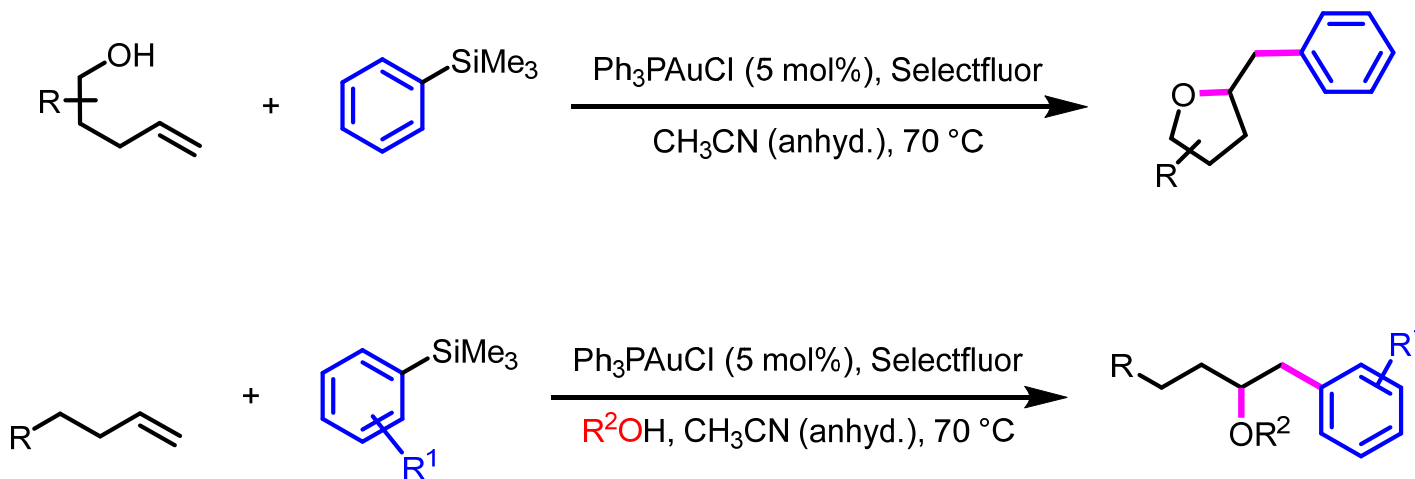


Oxidants-Assisted Coupling Reactions-Selectfluor

Au^I- and selectfluor-mediated coupling reactions of terminal alkenes



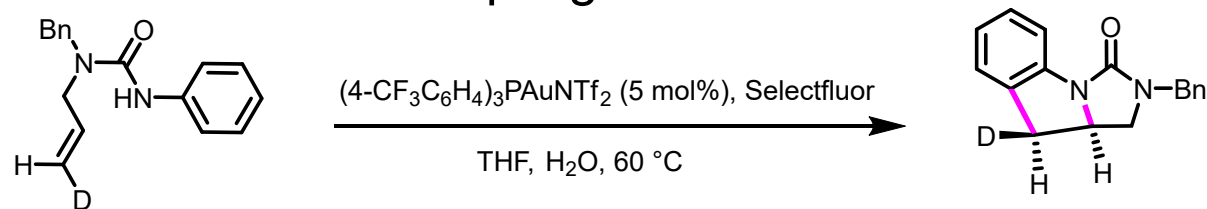
Zhang, L. et al. *J. Am. Chem. Soc.* **2010**, 132, 1474.



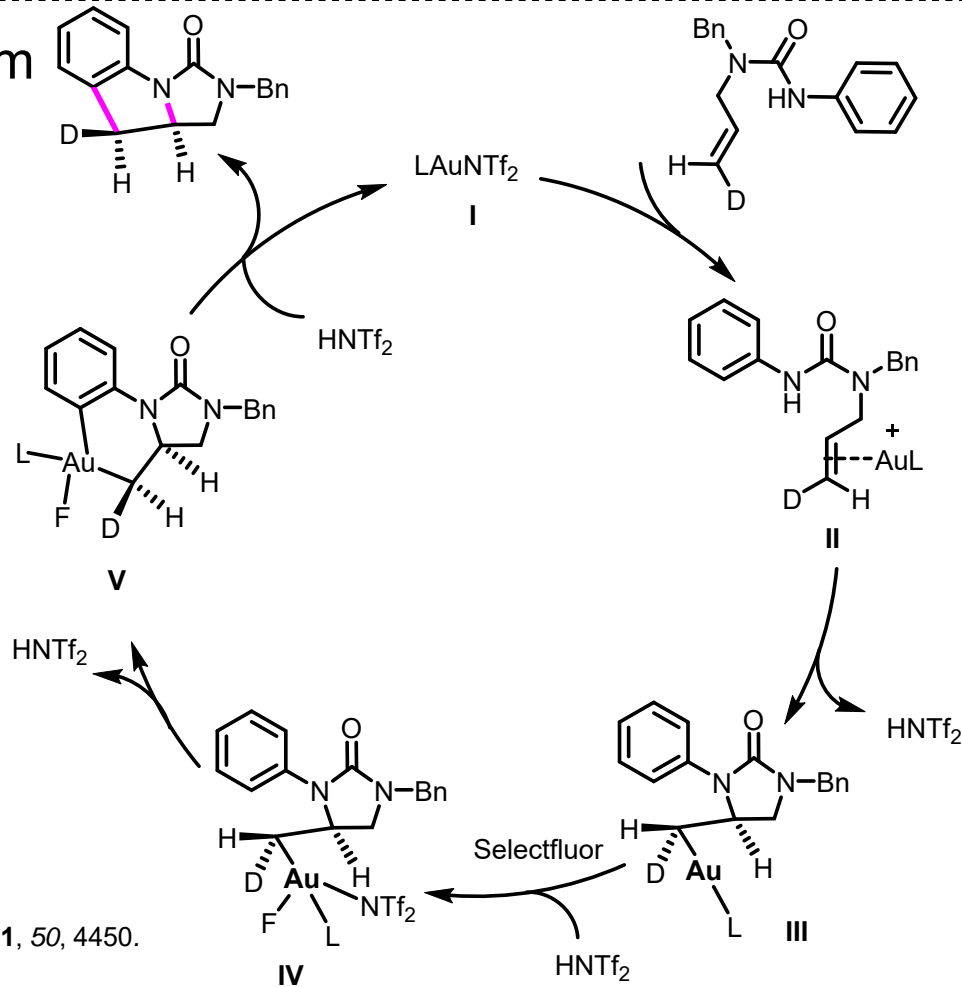
Lloyd-Jones, G. C.; Russel, C. A. et al. *Org. Lett.* **2010**, 12, 4724.

Oxidants-Assisted Coupling Reactions-Selectfluor

Au^I- and selectfluor-mediated coupling reactions of unactivated arenes



Proposed mechanism

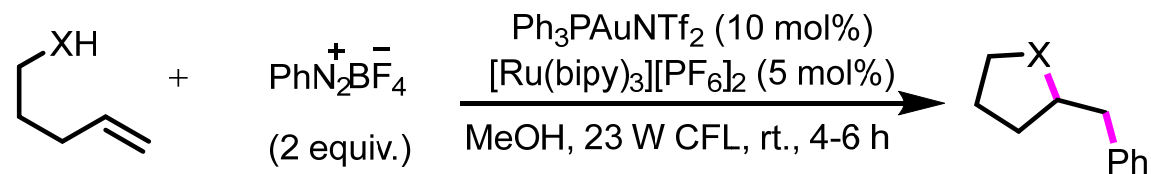


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Photo-Assisted Coupling Reactions

Dual gold/photoredox mediated coupling reactions of terminal alkenes



I. "Transmetalation First"

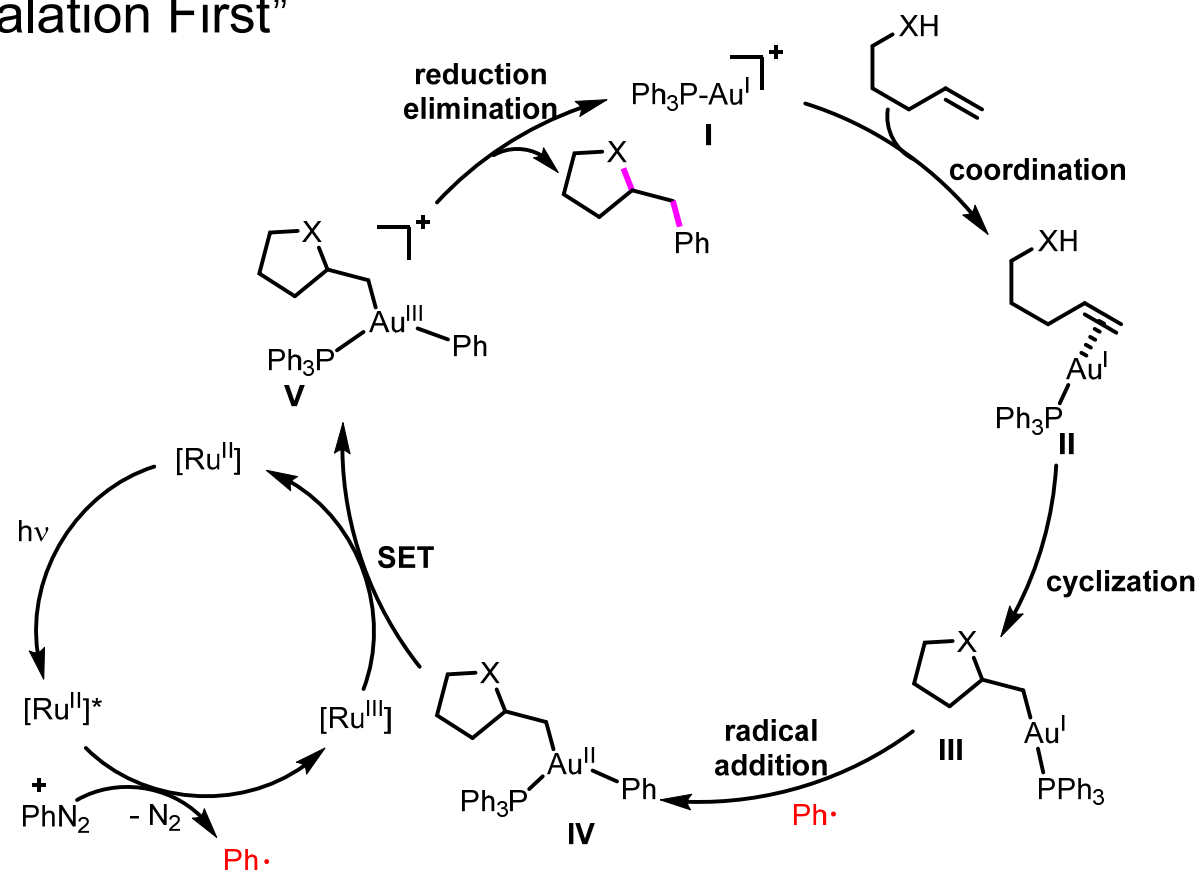
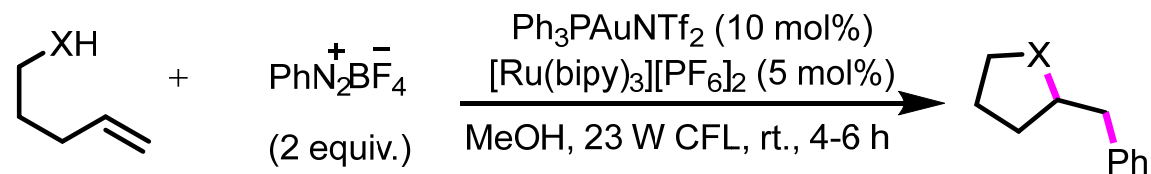


Photo-Assisted Coupling Reactions

Dual gold/photoredox mediated coupling reactions of terminal alkenes



II. "Oxidation First"

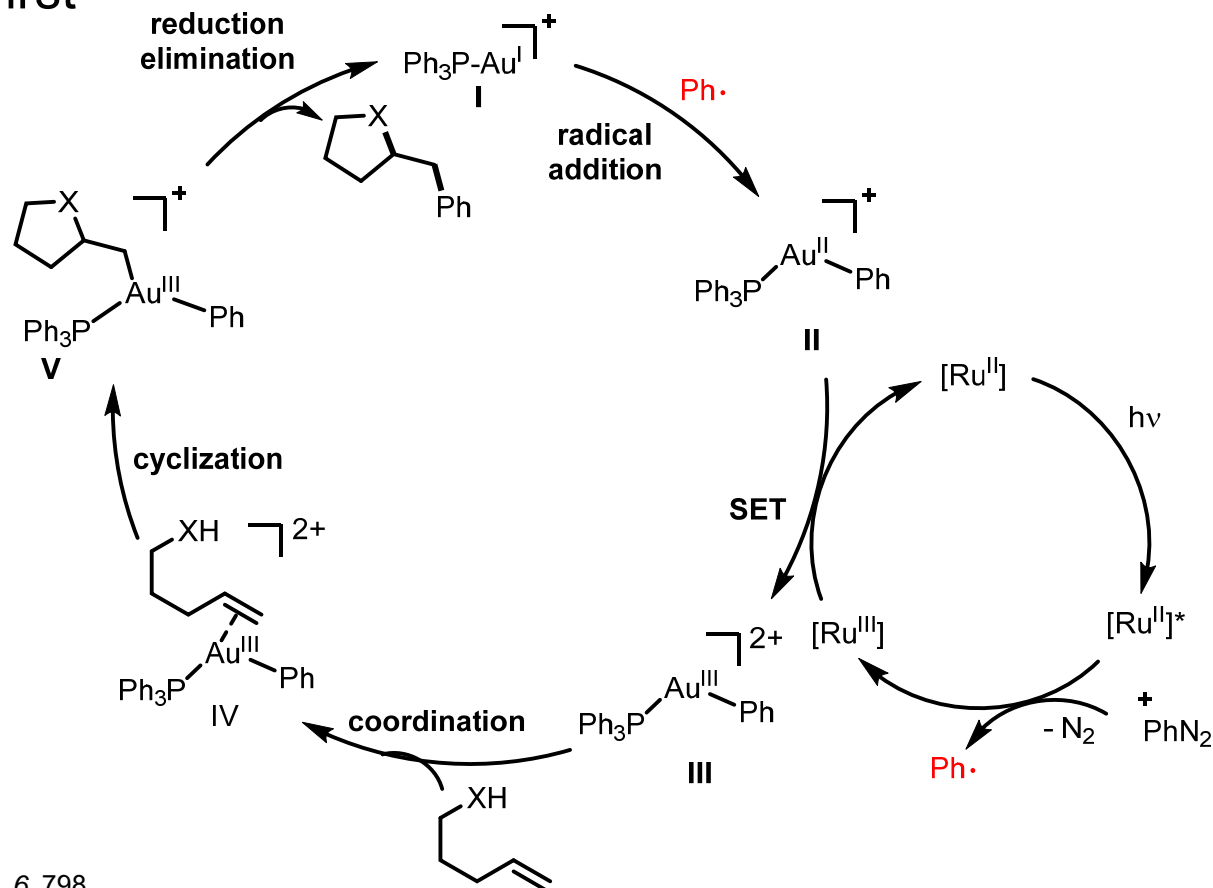


Photo-Assisted Coupling Reactions

Dual gold/photoredox mediated C(sp²)-C(sp²) coupling reactions

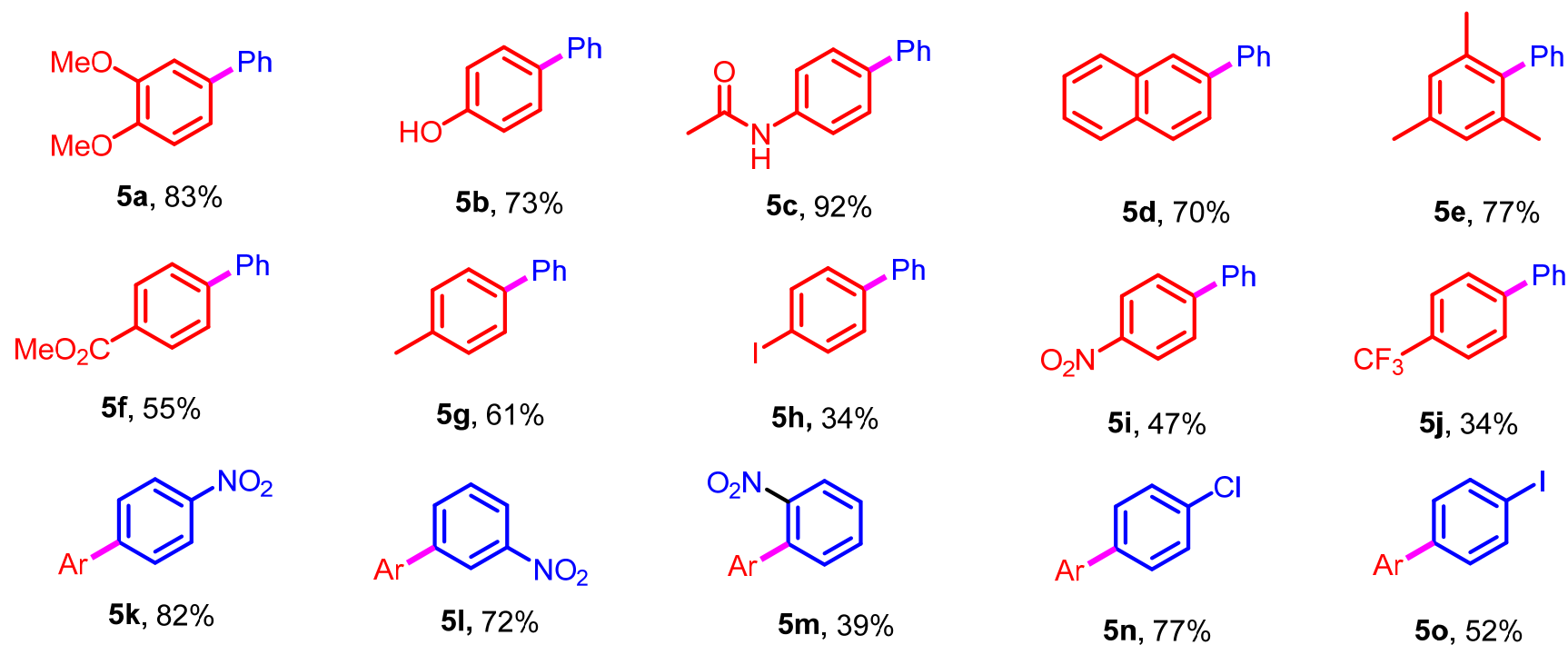
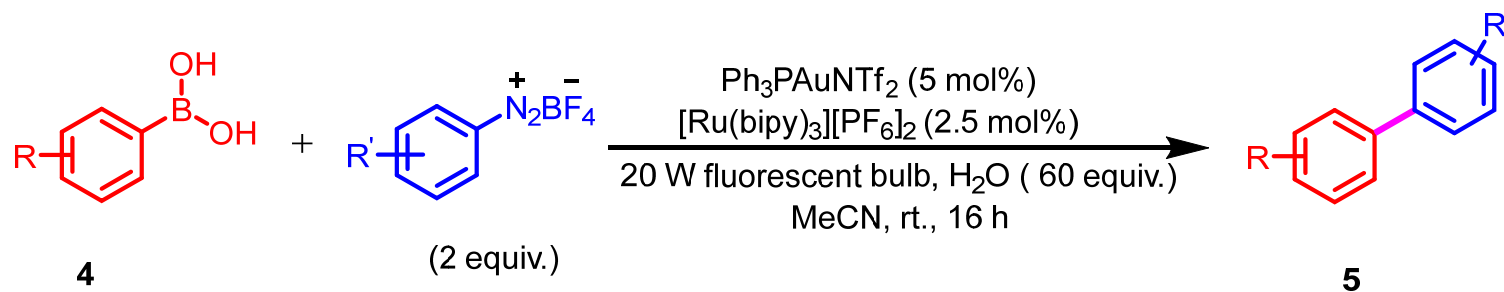
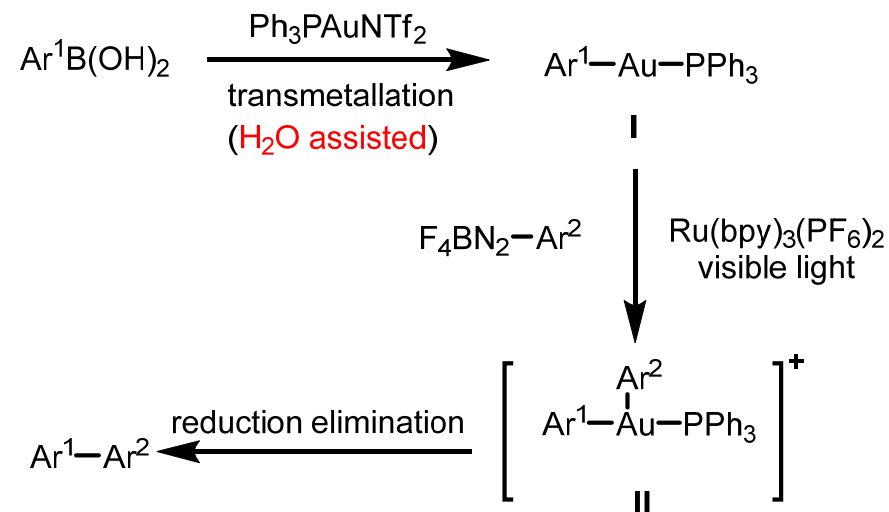


Photo-Assisted Coupling Reactions

Proposed mechanism



- Water is involved in aiding the transmetalation step ($\text{Ar}^1\text{B}(\text{OH})_2 \rightarrow \text{I}$)
- Water affects the homogeneity of the reaction mixture, and therefore the ability of light to efficiently penetrate the mixture to promote photoredox coupling

Photo-Assisted Coupling Reactions

Monometallic light-assisted C(sp²)-C(sp²) coupling reactions

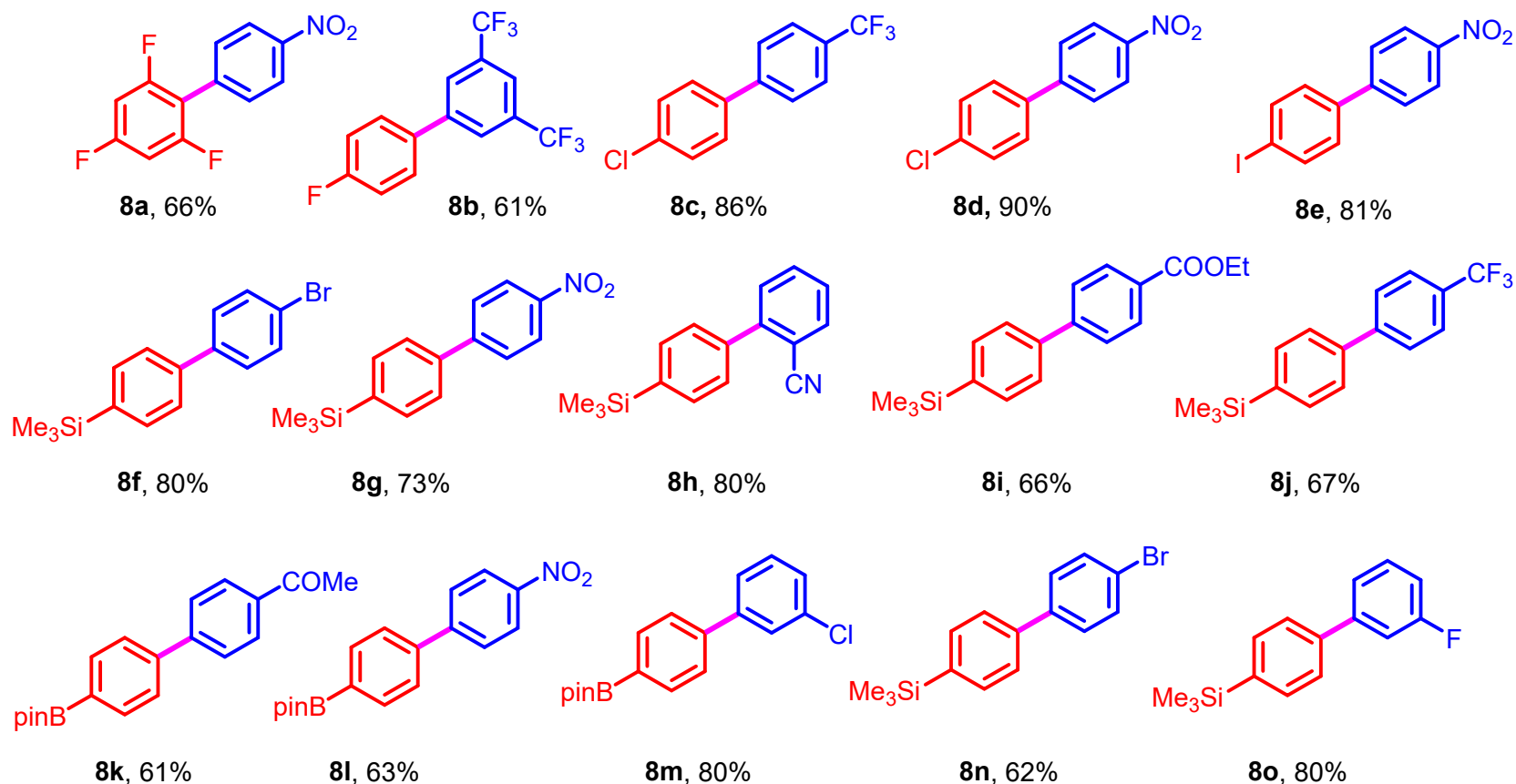
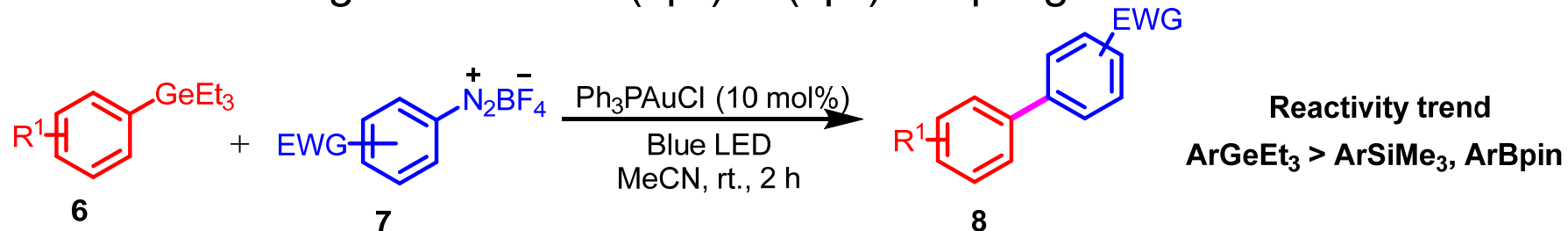
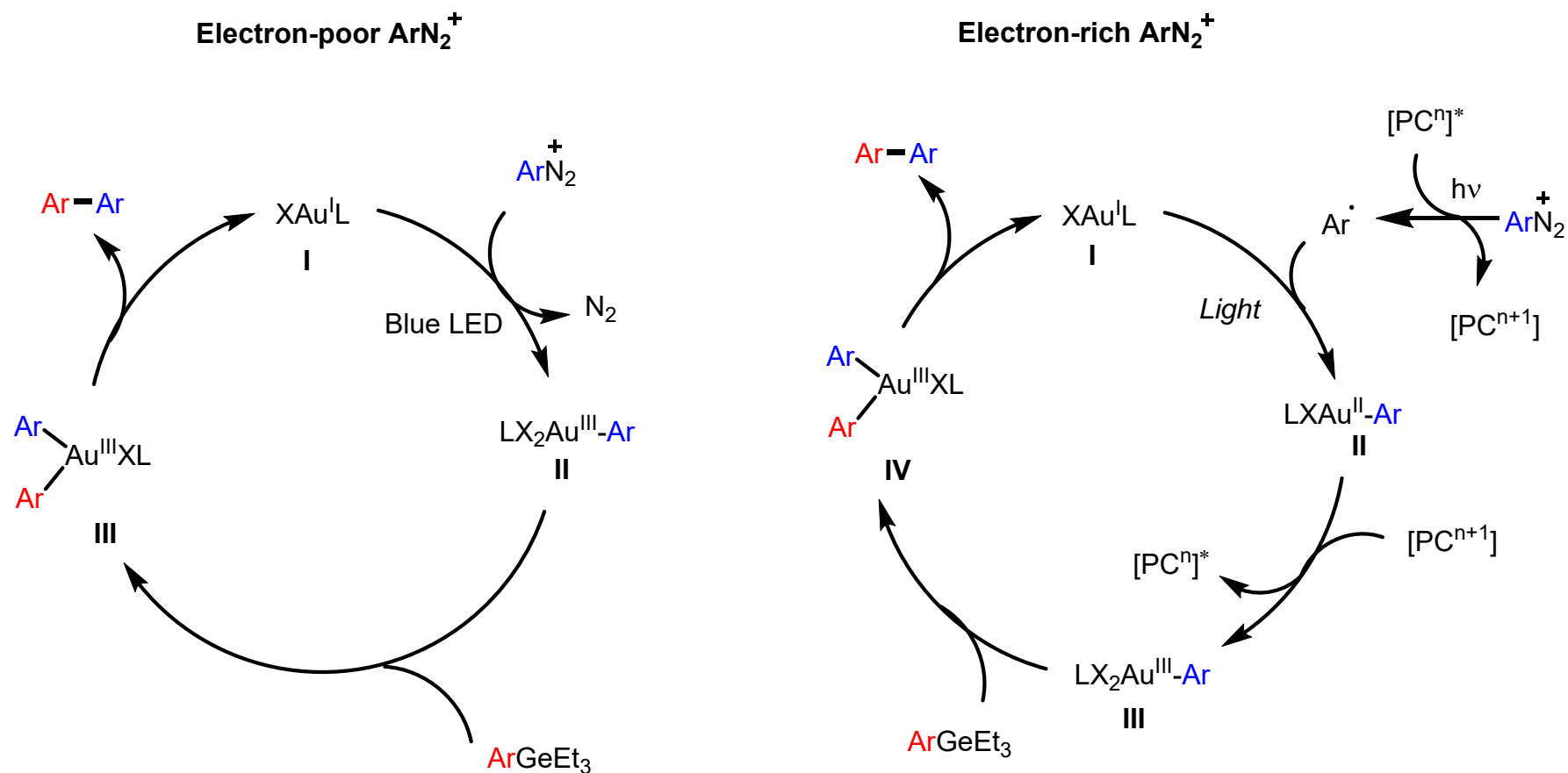


Photo-Assisted Coupling Reactions

Proposed mechanism

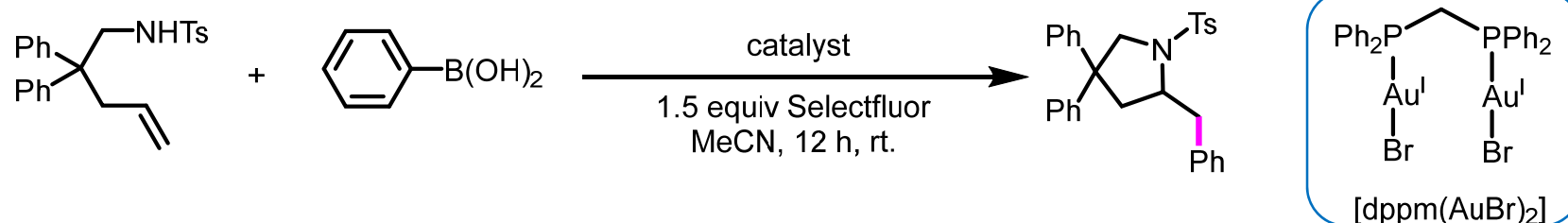


Background

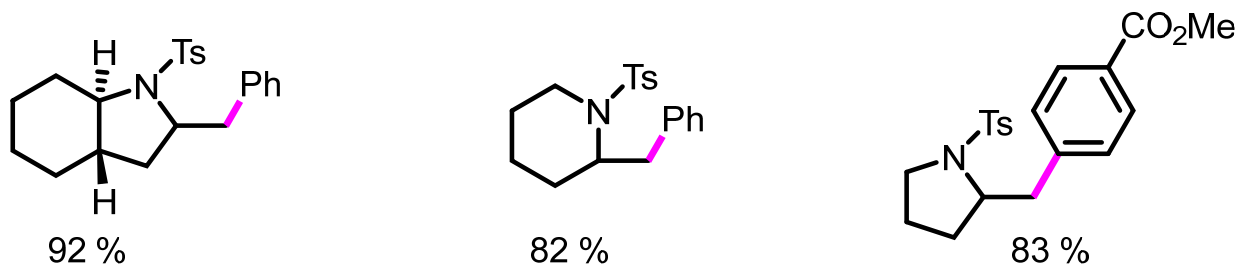
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Dinuclear Gold-Catalyzed Coupling Reactions

Dinuclear gold-catalyzed coupling reactions of terminal alkenes

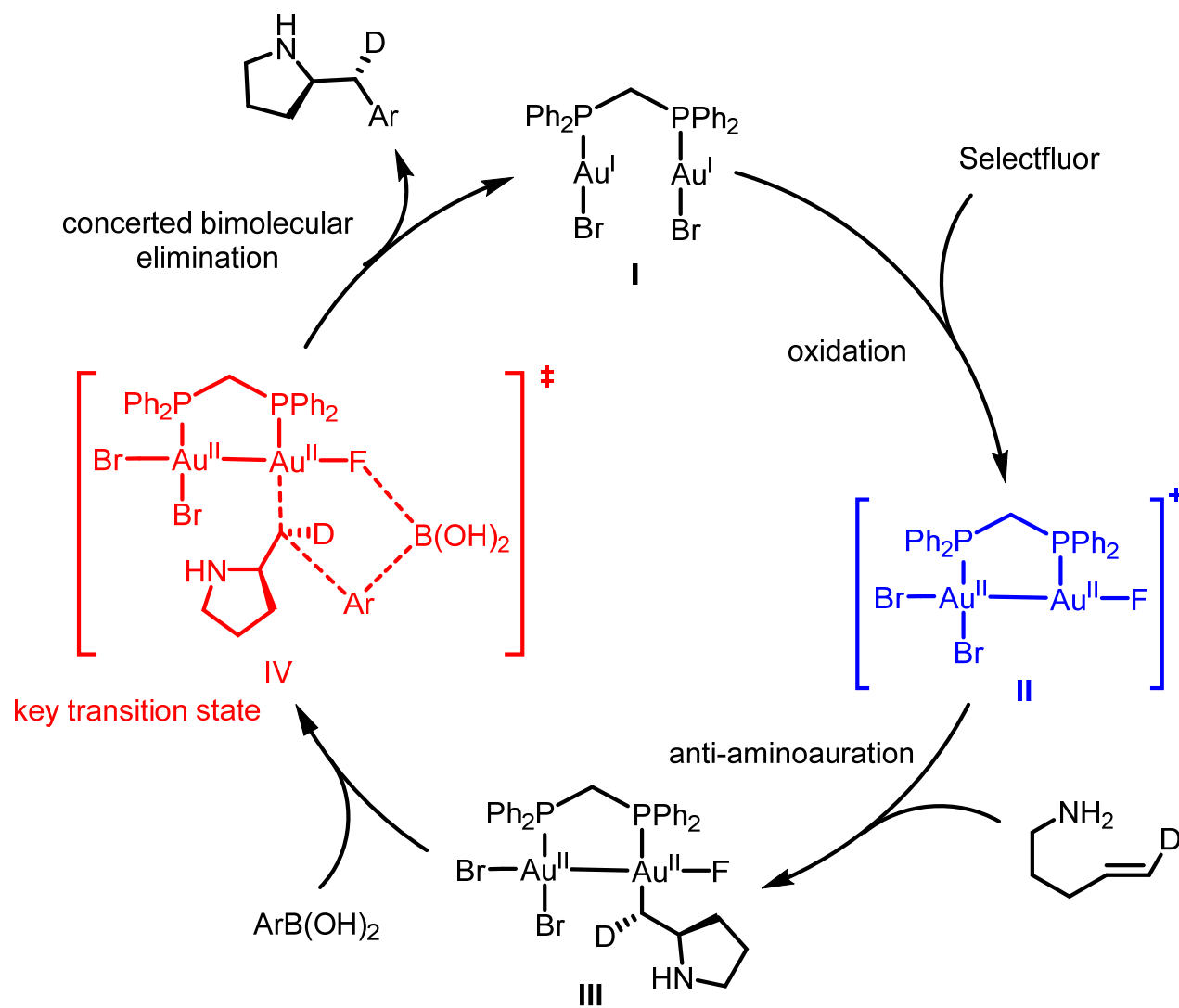


Catalyst (mol%)	Yield [%]
Ph_3PAuX (5), X = Cl, Br, I, OTf (5)	< 47
$[\text{dppm}(\text{AuBr})_2]$ (3)	81



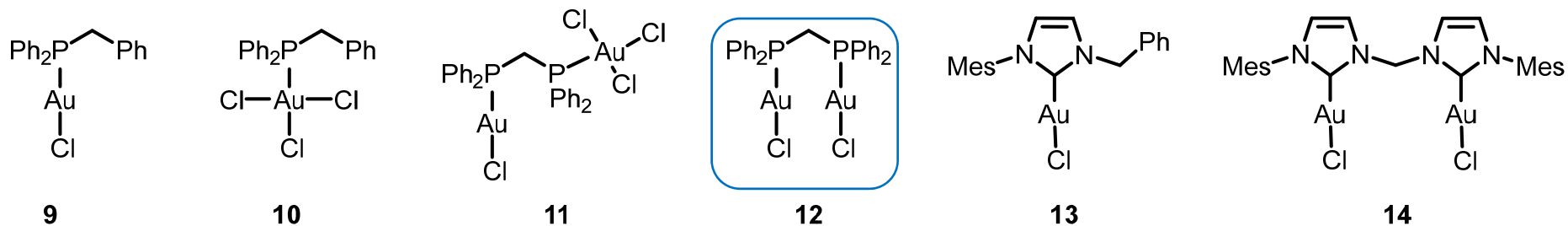
Dinuclear Gold-Catalyzed Coupling Reactions

Proposed mechanism



Dinuclear Gold-Catalyzed Coupling Reactions

Dinuclear gold-complexes

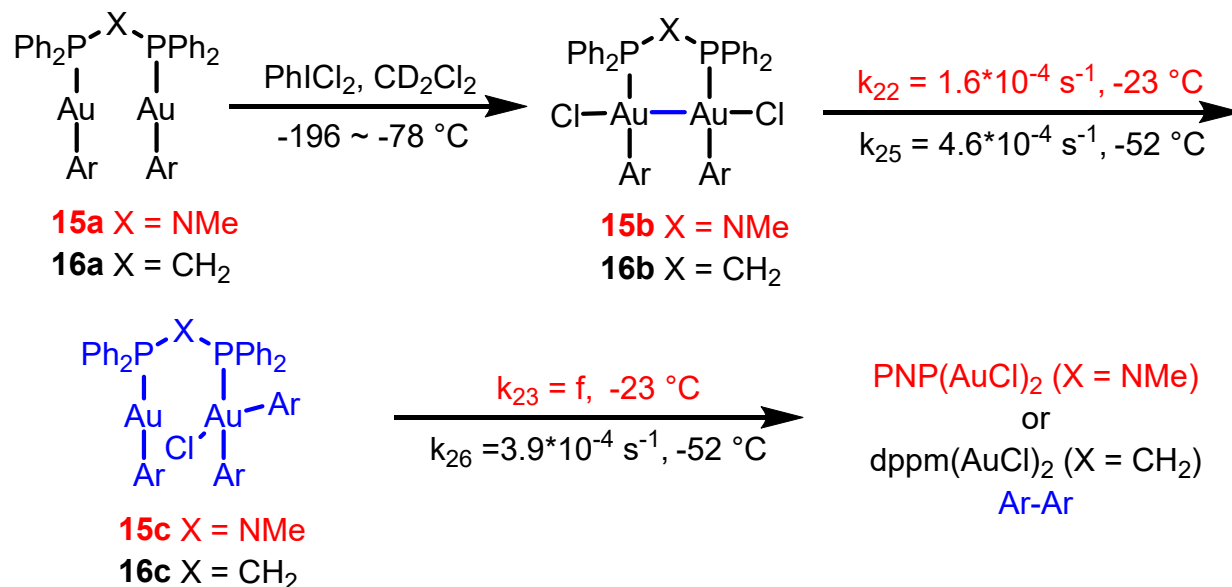


Summary of Electrochemistry Data

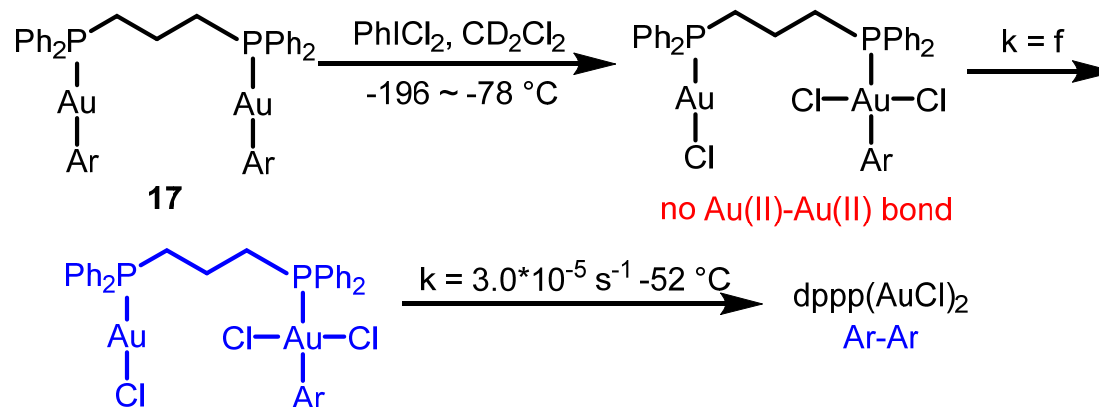
Complex	E_{ox} for Au(I) (V)	E_{red} for Au(III) (V)
9	1.48	/
10	/	-0.69
11	1.48	-0.53
12	1.34	/
13	1.96	/
14	1.64	/

Dinuclear Gold-Catalyzed Coupling Reactions

a) Homocoupling by dinuclear gold with aurophilic interaction

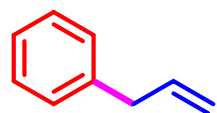
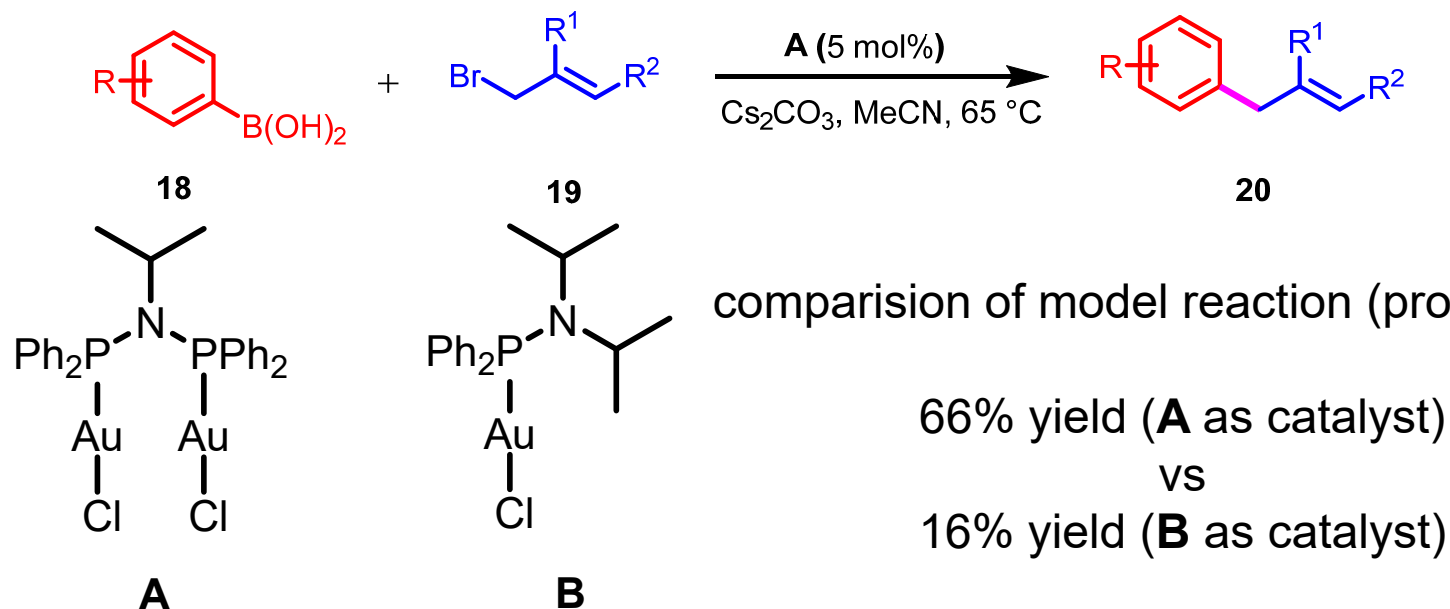


b) Homocoupling by dinuclear gold without aurophilic interaction

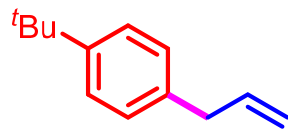


Dinuclear Gold-Catalyzed Coupling Reactions

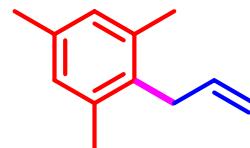
Dinuclear gold-catalyzed coupling of Ar-B(OH)₂ and allyl bromides



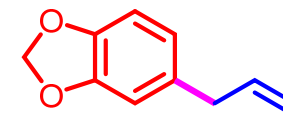
20a, 66%



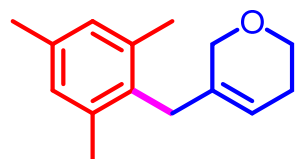
20b, 72%



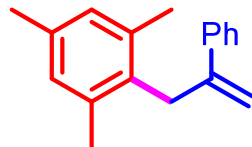
20c, 66%



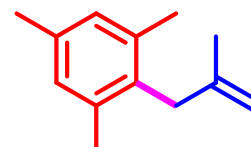
20d, 86%



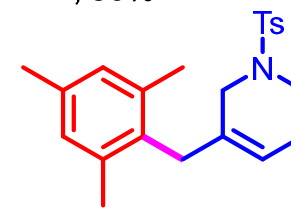
20e, 76%



20f, 57%



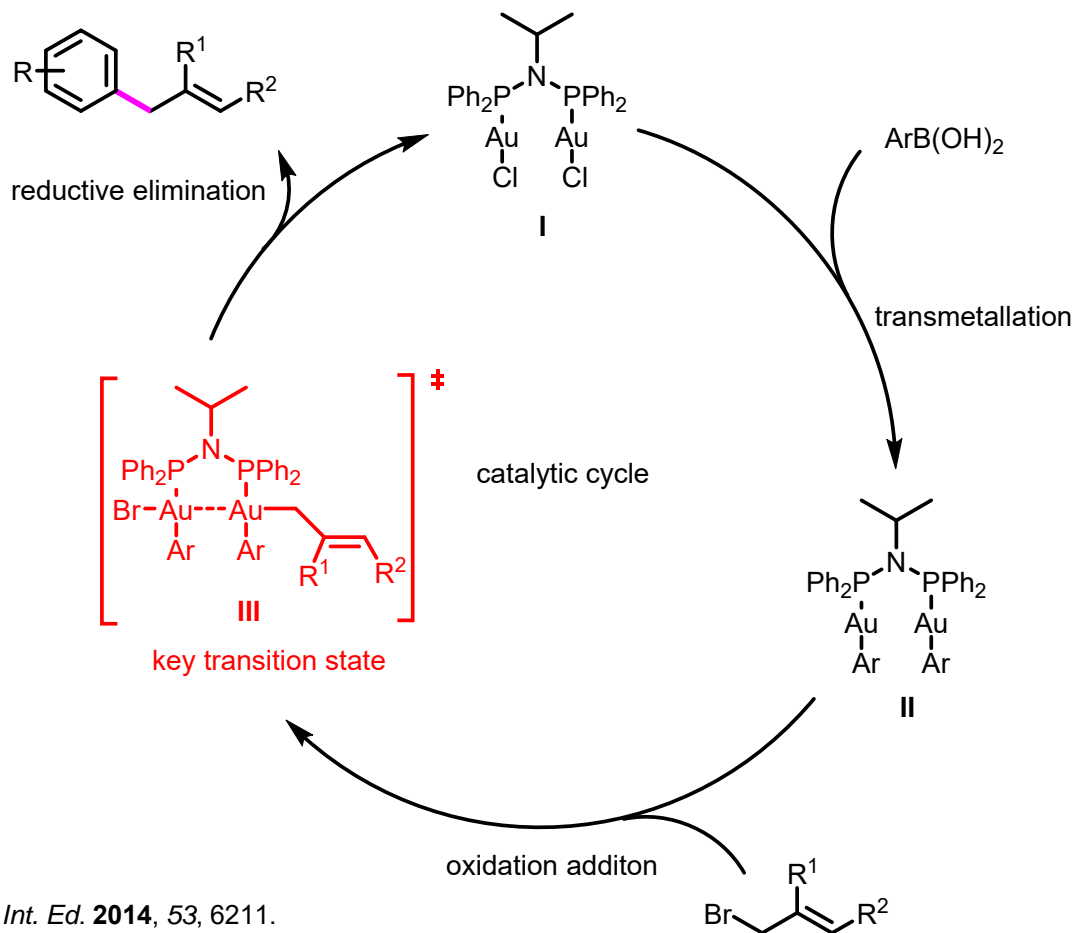
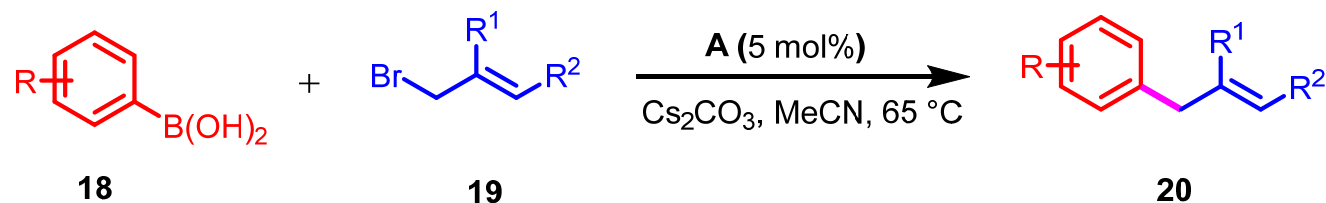
20g, 68%



20h, 58%

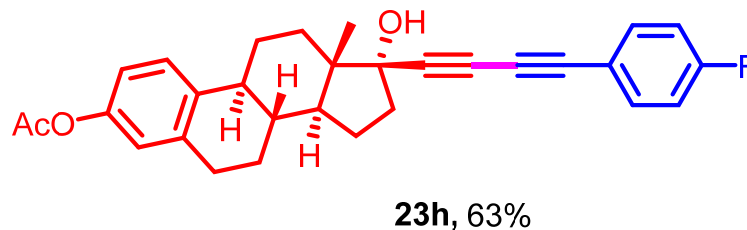
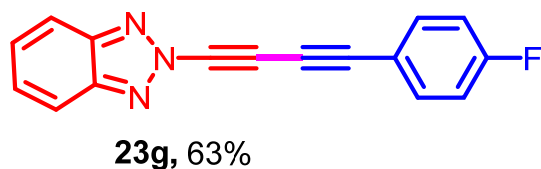
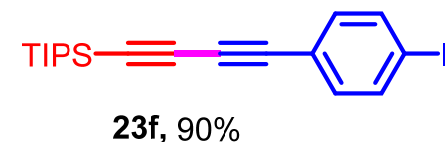
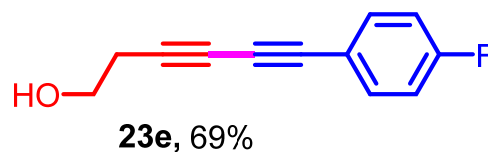
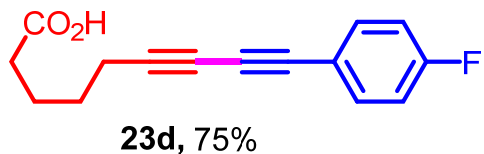
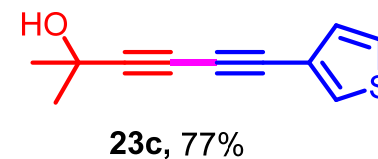
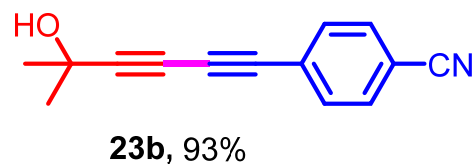
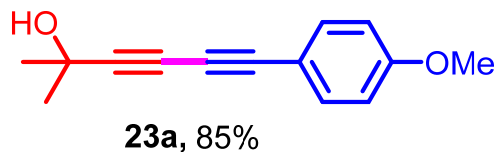
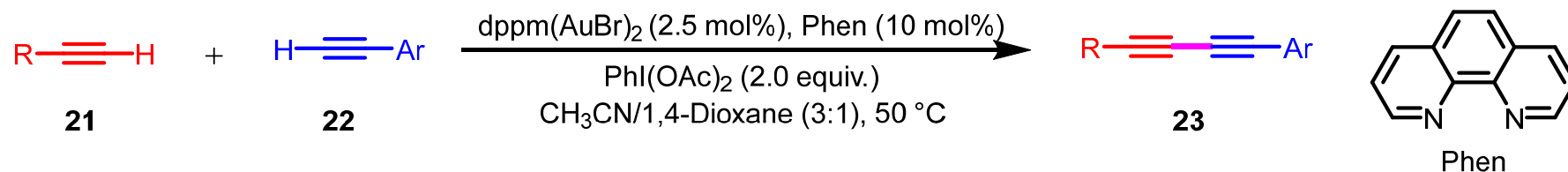
Dinuclear Gold-Catalyzed Coupling Reactions

Dinuclear gold-catalyzed coupling of Ar–B(OH)₂ and allyl bromides



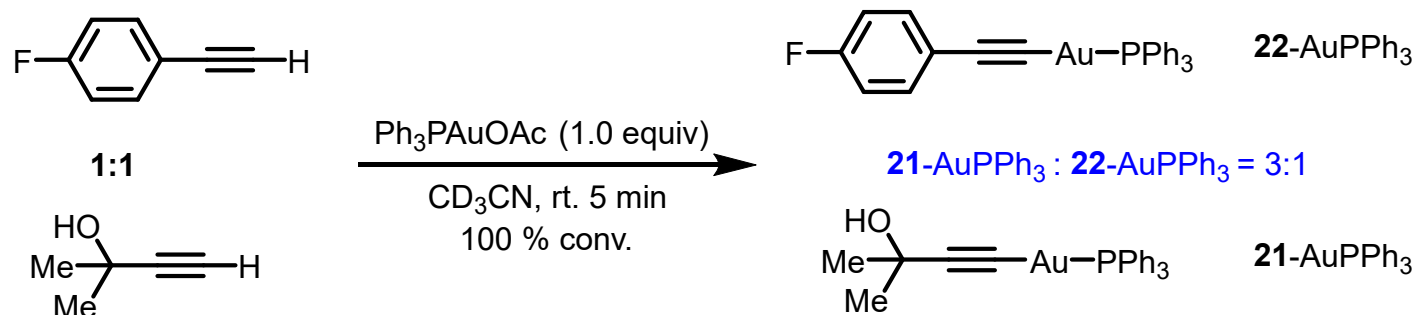
Dinuclear Gold-Catalyzed Coupling Reactions

Dinuclear gold-catalyzed coupling of aliphatic and aromatic alkynes

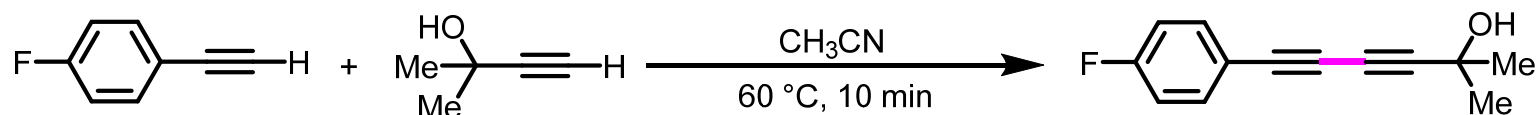


Dinuclear Gold-Catalyzed Coupling Reactions

Discrimination effect



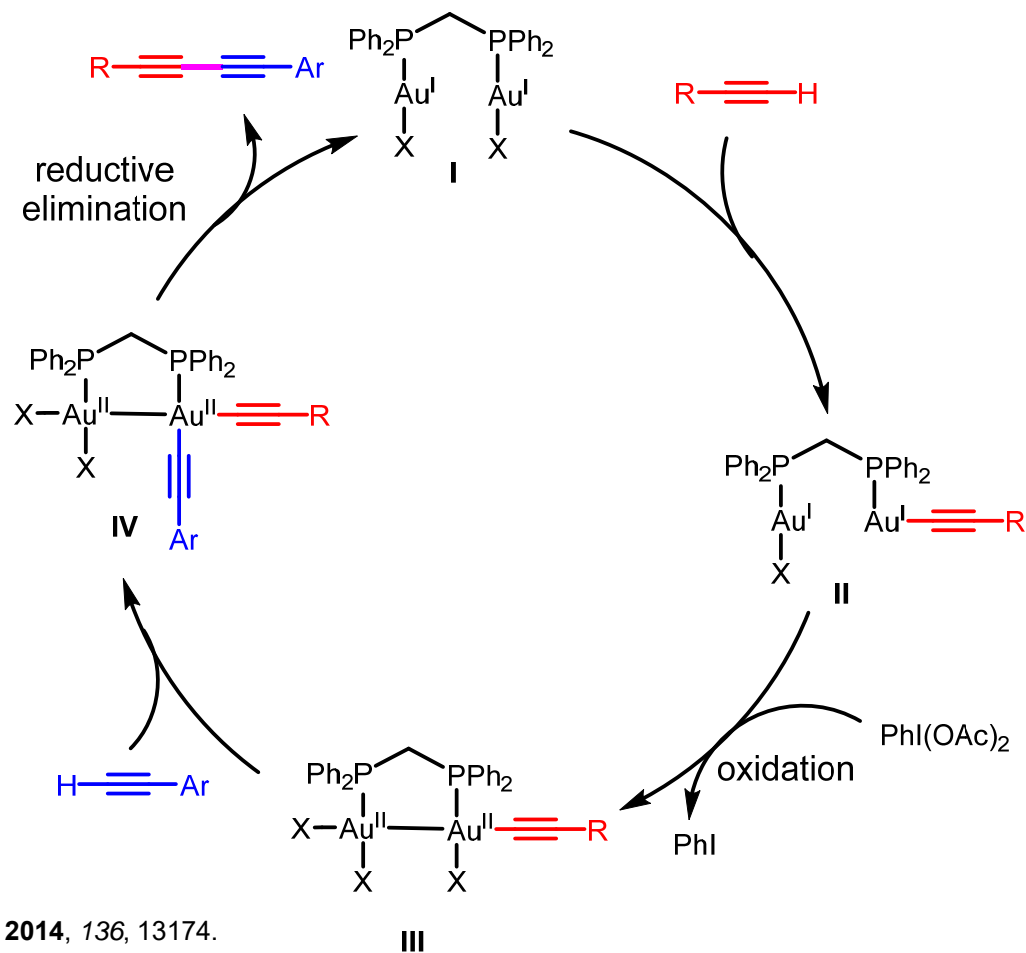
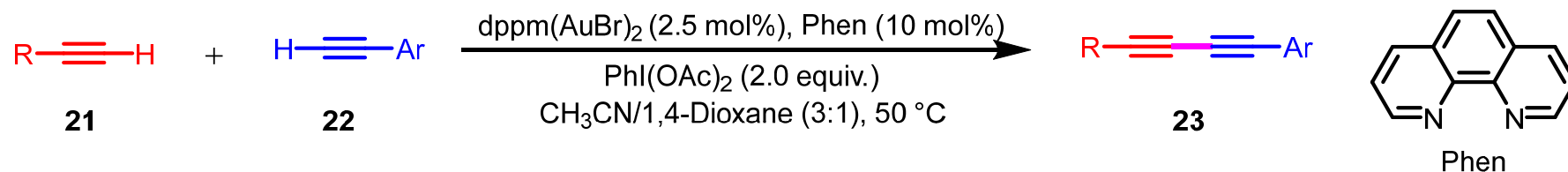
Stoichiometric reactions



Conditions	Yield [%]	Hetero/homo
standard conditions	83	12:1
A: AuCl_3 (1 equiv.)	0	n.a.
B: AuCl_3 (1 equiv.), Phen (1 equiv.)	< 5	n.a.
C: AuCl_3 (1 equiv.), NaOAc (3 equiv.)	12	2:1
D: AuCl_3 (1 equiv.), NaOAc (3 equiv.), Phen (1 equiv.)	65	4:1

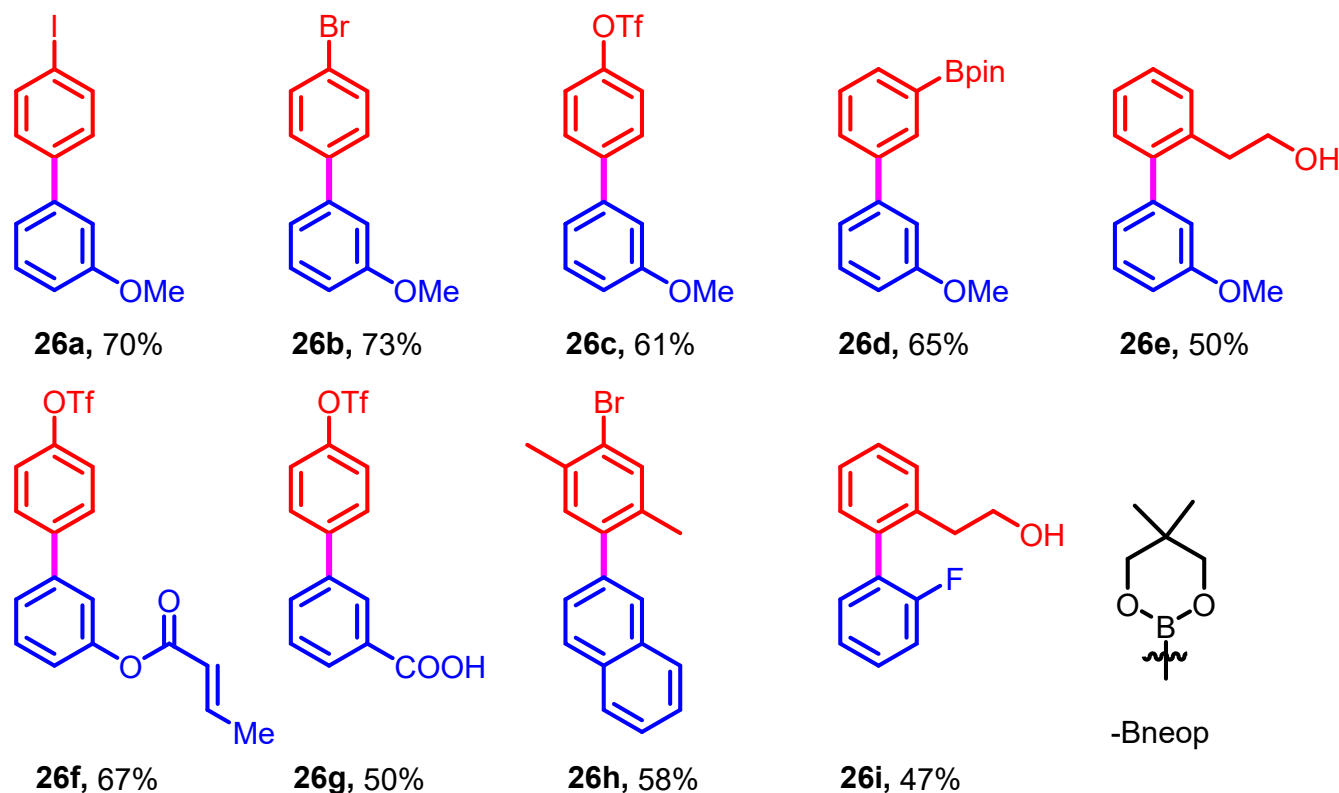
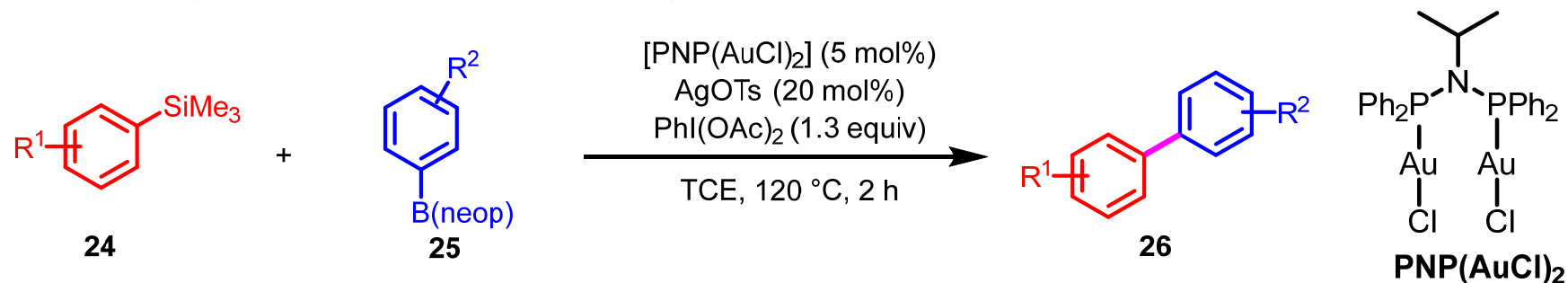
Dinuclear Gold-Catalyzed Coupling Reactions

Dinuclear gold-catalyzed coupling of aliphatic and aromatic alkynes



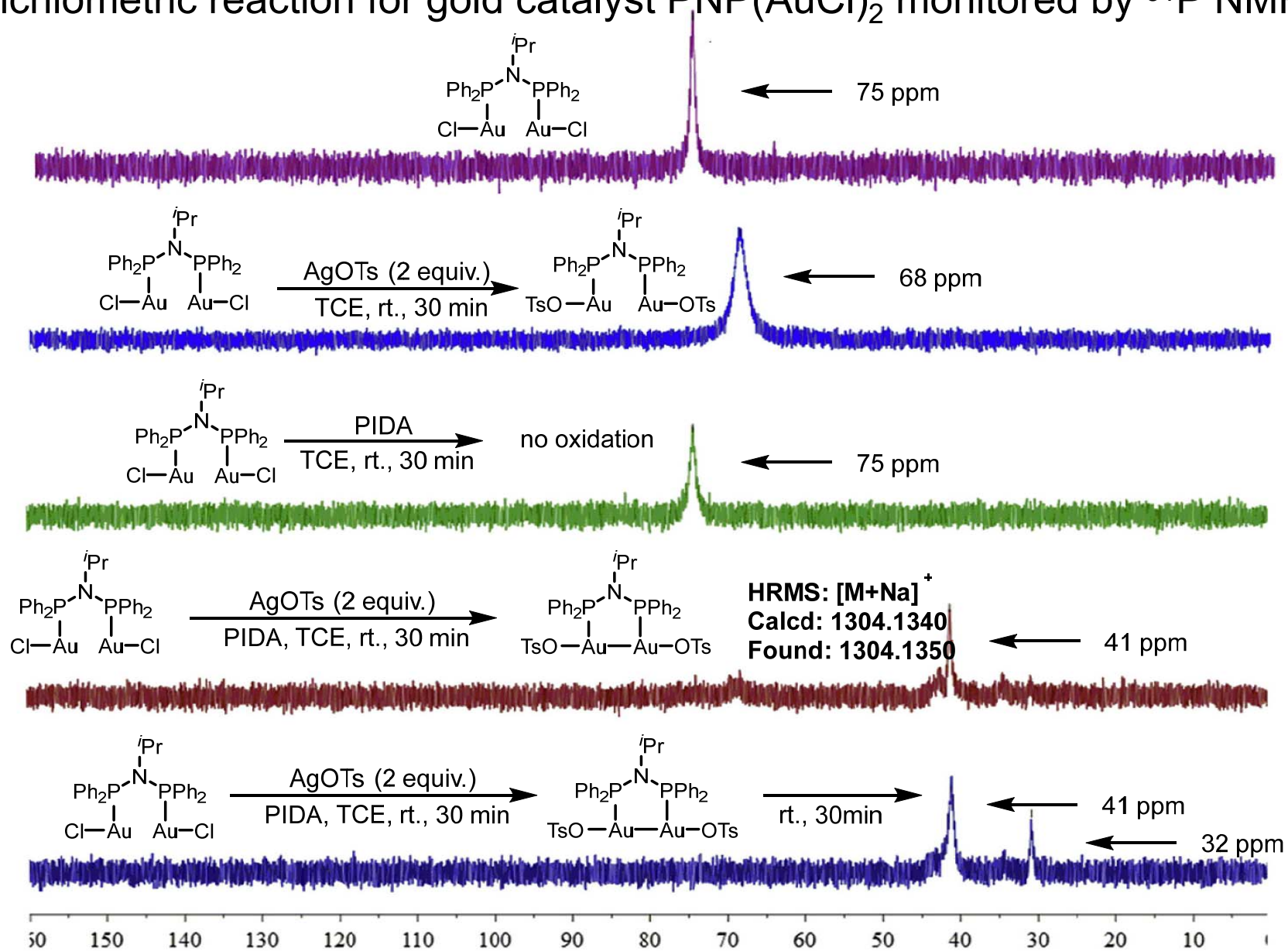
Dinuclear Gold-Catalyzed Coupling Reactions

Dinuclear gold-catalyzed coupling of arylsilanes and arylboronates



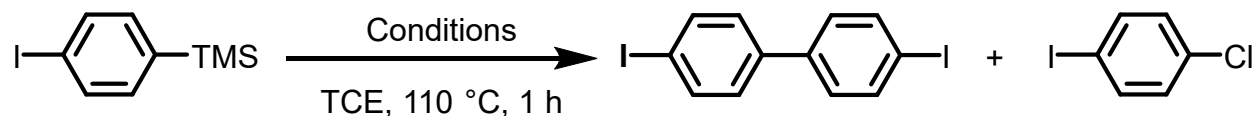
Dinuclear Gold-Catalyzed Coupling Reactions

Stoichiometric reaction for gold catalyst $\text{PNP}(\text{AuCl})_2$ monitored by ^{31}P NMR

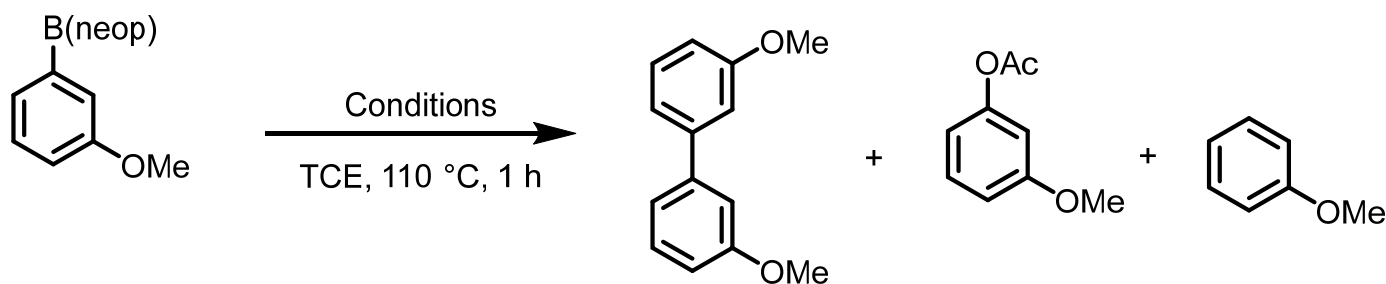


Dinuclear Gold-Catalyzed Coupling Reactions

Control experiments with Ph_3PAuCl



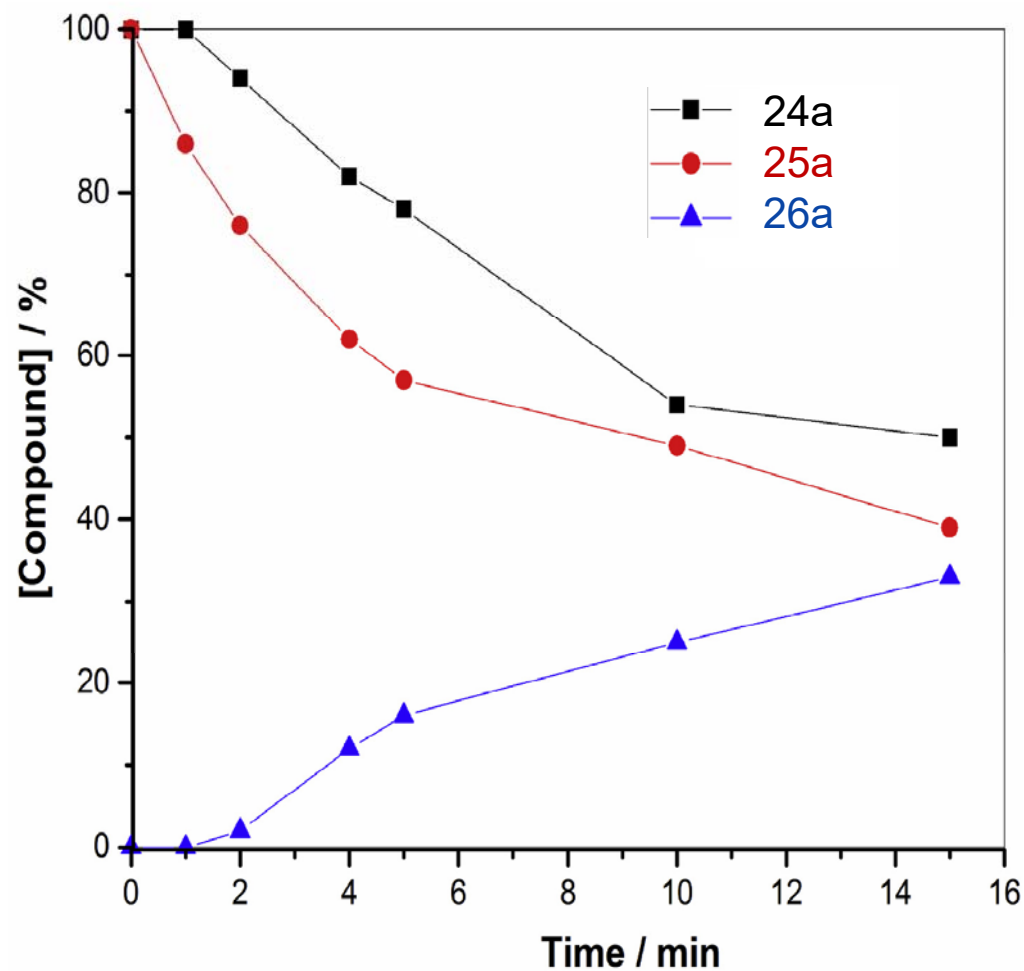
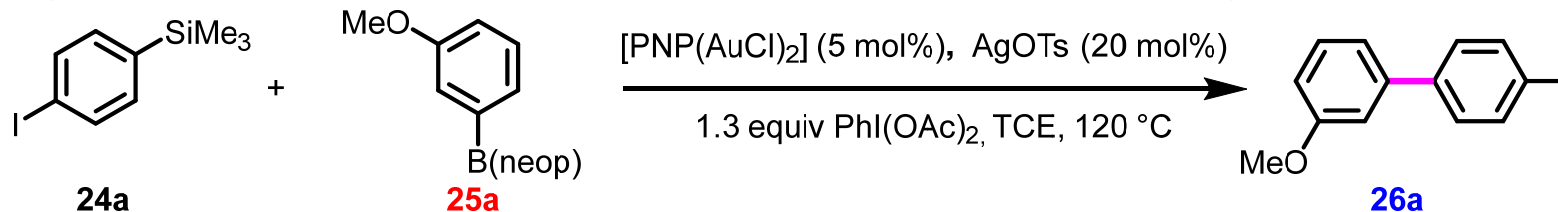
- a. Ph_3PAuCl (1 equiv), AgOTs (1.2 equiv.) conversion: 0%
- b. Ph_3PAuCl (1 equiv), AgOTs (1.2 equiv.), $n\text{Bu}_4\text{NOAc}$ (2.6 equiv.) conversion: 0%
- c. Ph_3PAuCl (1 equiv), AgOTs (1.2 equiv.), PIDA (1.3 equiv.) conversion: 100%



- a. Ph_3PAuCl (1 equiv), AgOTs (1.2 equiv.) conversion: 0%
- b. Ph_3PAuCl (1 equiv), AgOTs (1.2 equiv.), $n\text{Bu}_4\text{NOAc}$ (2.6 equiv.) conversion: 7%
- c. Ph_3PAuCl (1 equiv), AgOTs (1.2 equiv.), PIDA (1.3 equiv.) conversion: 56%

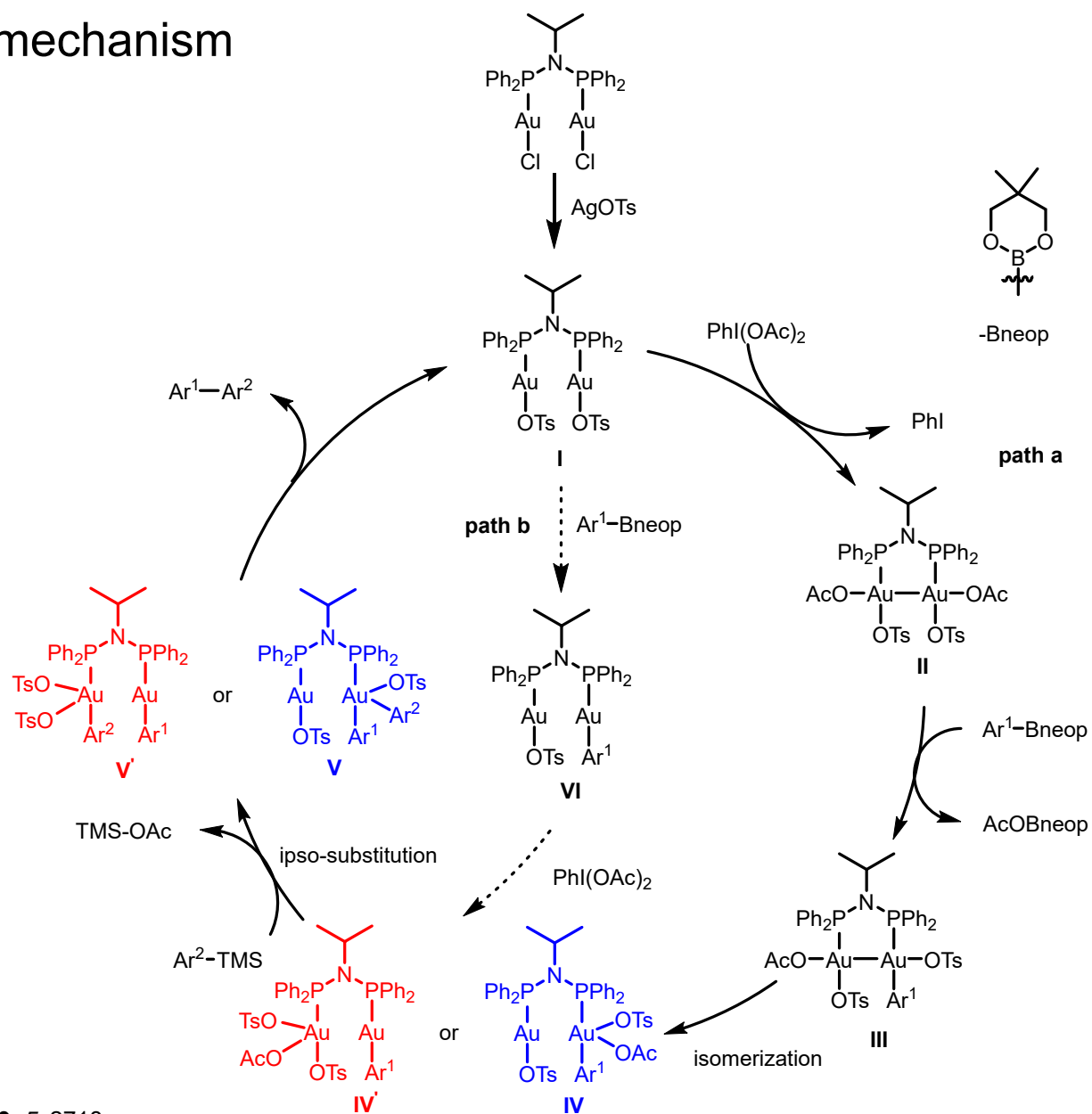
Dinuclear Gold-Catalyzed Coupling Reactions

Investigation of the consumption of **24a** and **25a** and the yield of **26a**



Dinuclear Gold-Catalyzed Coupling Reactions

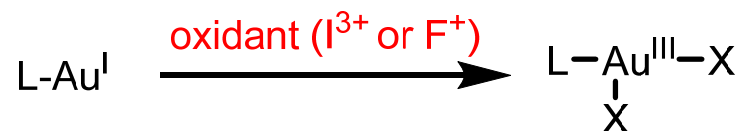
Proposed mechanism



Background

- Background
- Gold-Catalyzed C-C Coupling Reactions
 - Oxidants-Assisted Coupling Reactions
 - Photo-Assisted Coupling Reactions
 - Dinuclear Gold-Catalyzed Coupling Reactions
- Summary

Summary

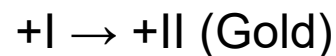


Aryl radicals (diazonium salts)

- Direct functionalization
- High regioselectivity
- High FG tolerance
- Dual role of oxidizing agent and substrate
- +I → +II → +III (Gold)

Dinuclear gold catalysis

Aurophilic interaction between the two gold atoms can reduce its redox potential



Thanks!