

Organic Super Electron Donors

Reporter: Mengmeng Xu

Supervisor: Prof. Quan Cai

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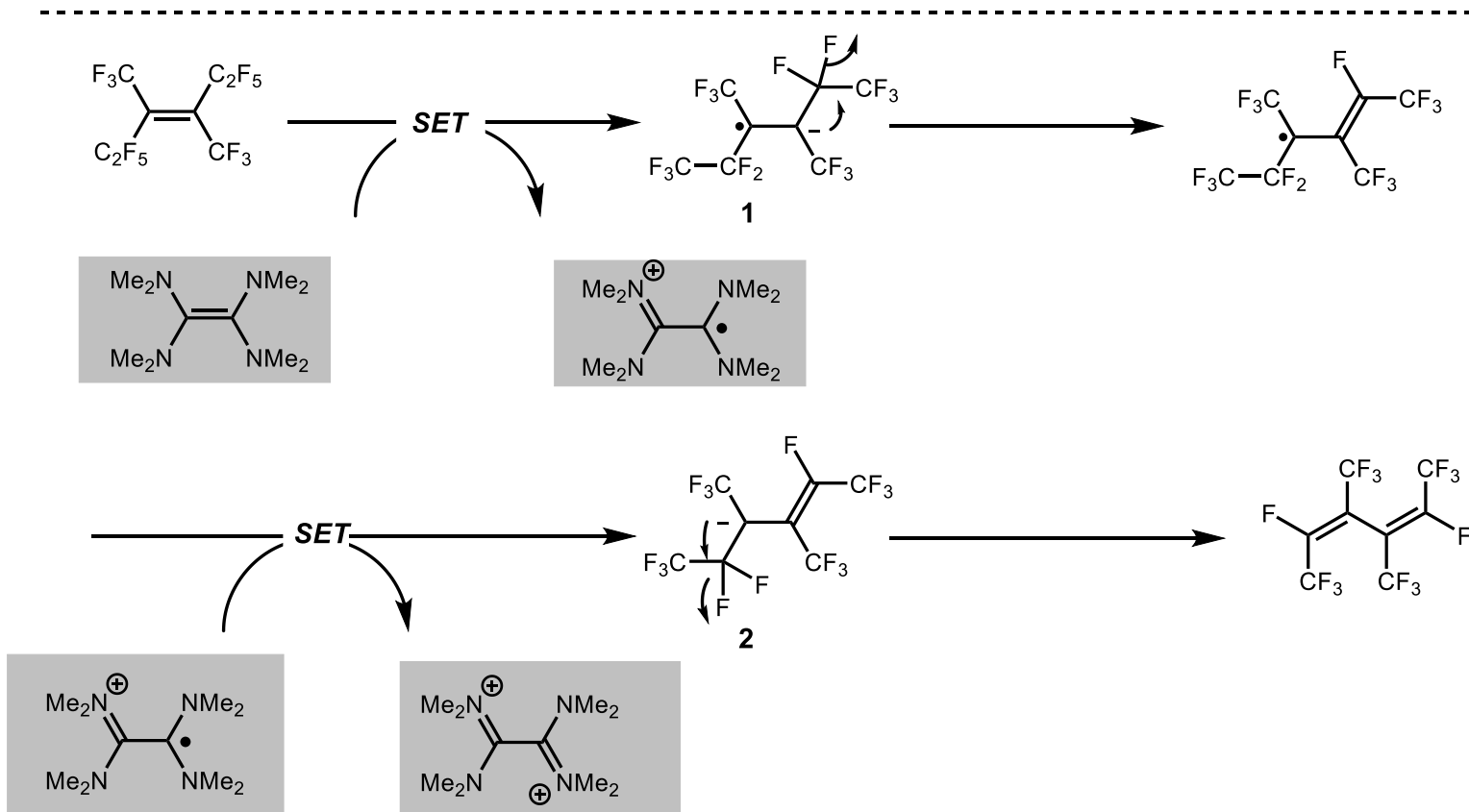
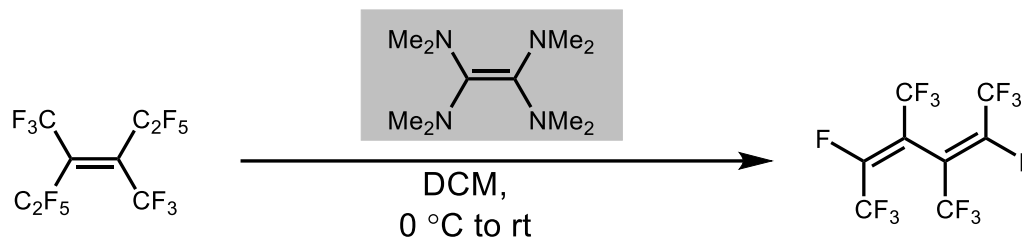
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- **Summary**

Discovery

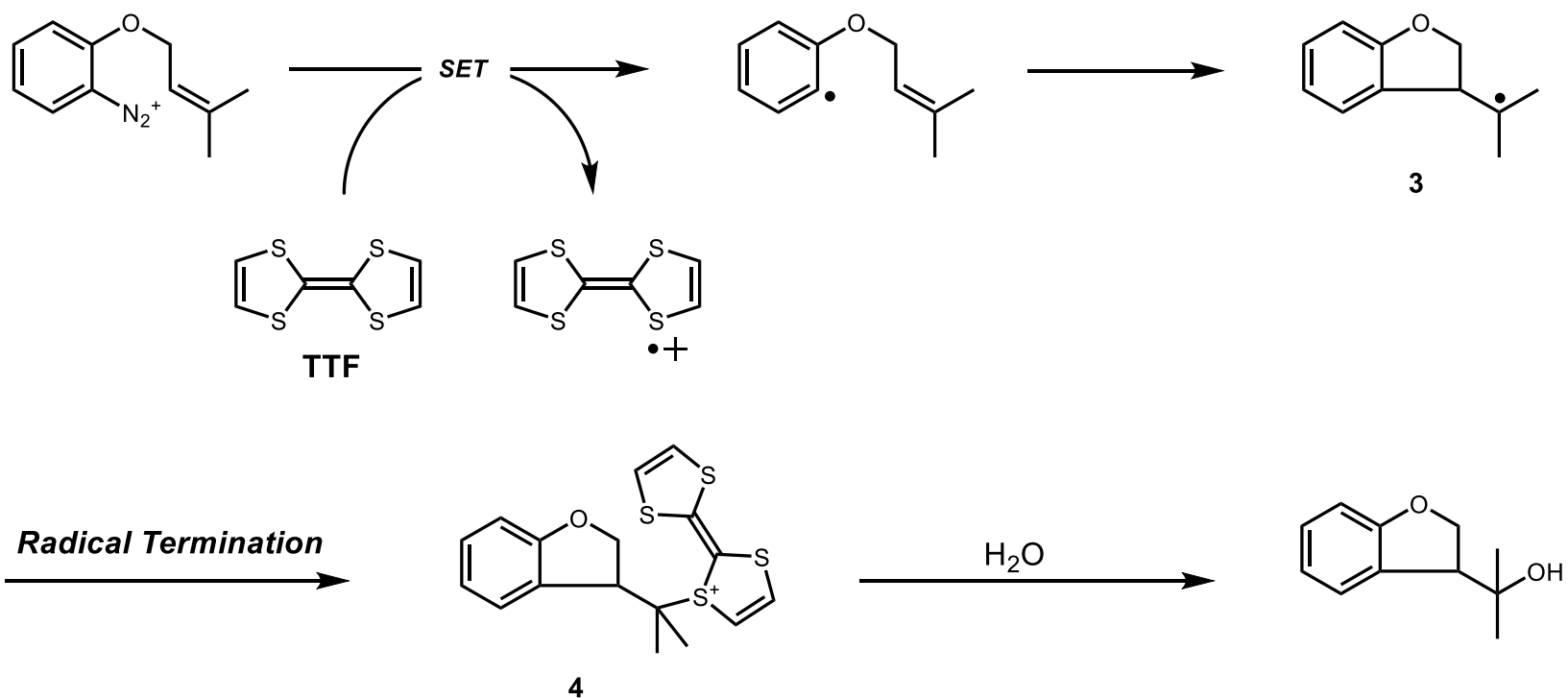
Electron-Donor



Discovery

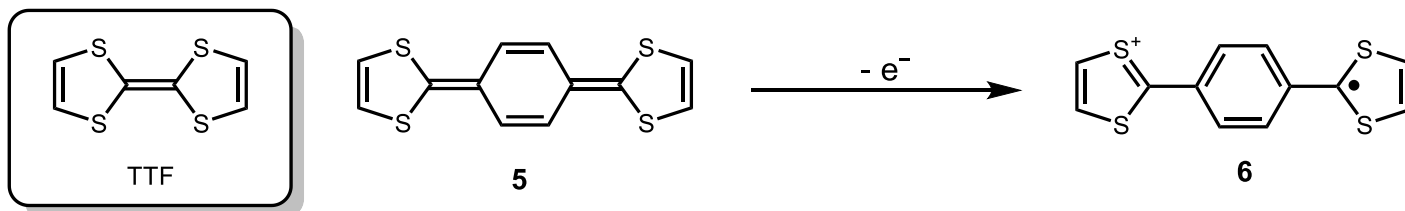
Electron-Donor

Radical-Polar Crossover Reactions



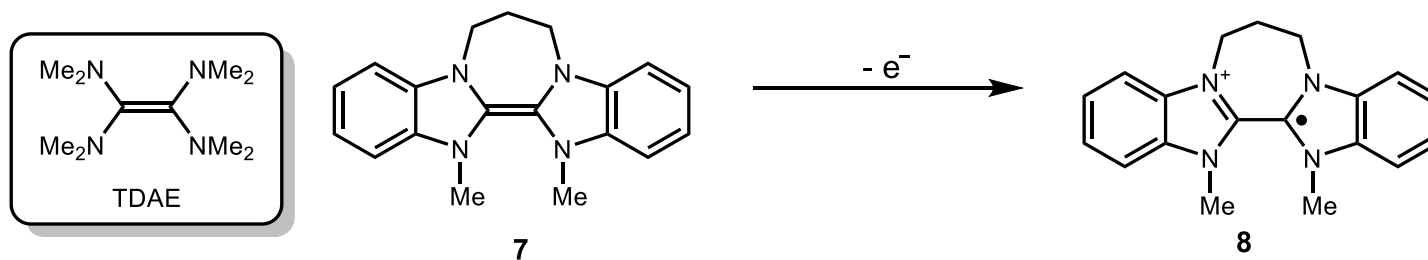
J. A. Murphy, et al. *J. Chem. Soc. Chem. Commun.*, **1993**, 295.

Discovery



Aromatic driving force

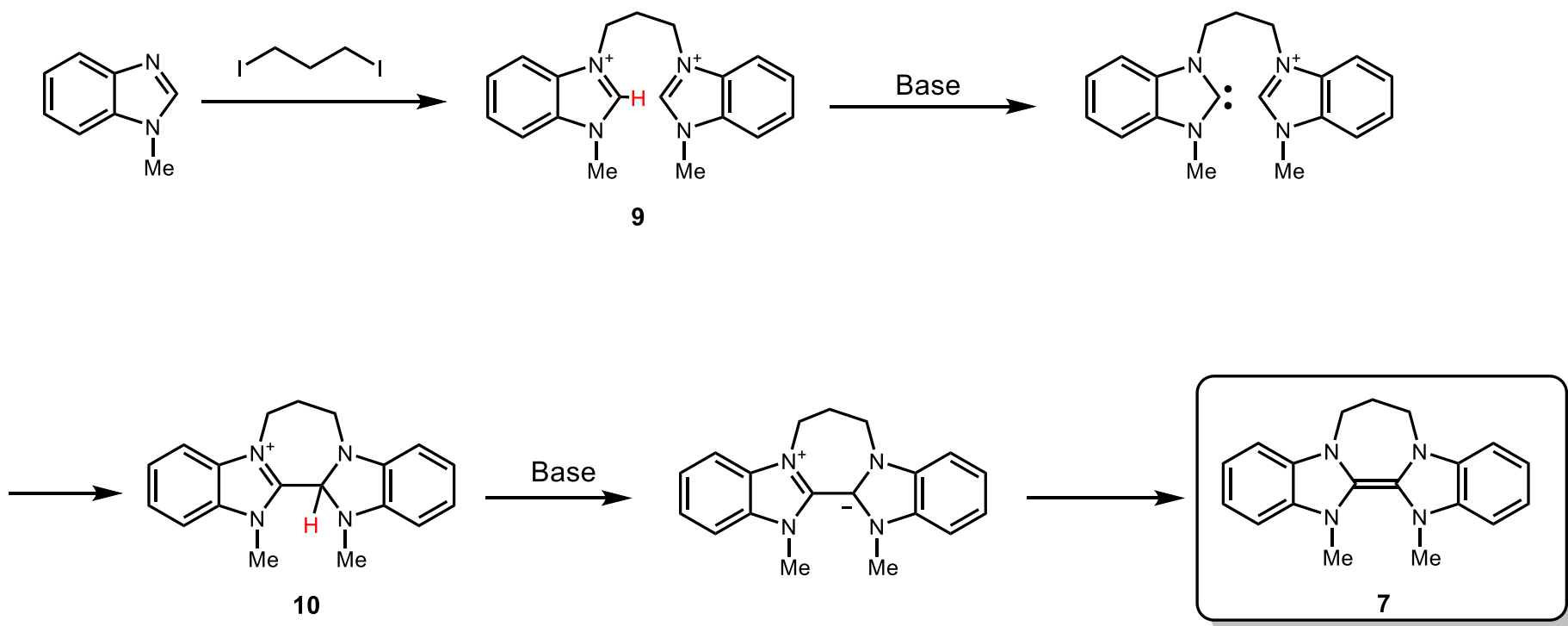
Yamashita Y.; Kobayashi. Y.; Miyashi T. *Angew. Chem., Int. Ed.* **1989**, 28, 1052



J. A. Murphy, et al. *Angew. Chem., Int. Ed.* **2005**, 44, 1356

Discovery

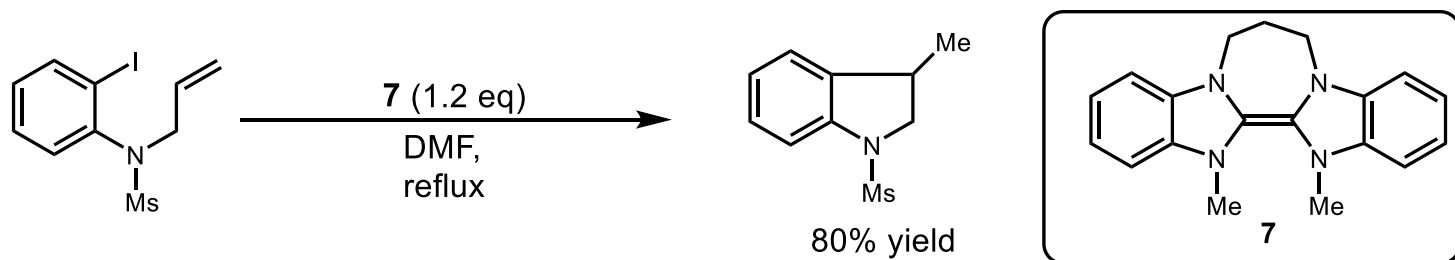
The method to prepare **7**



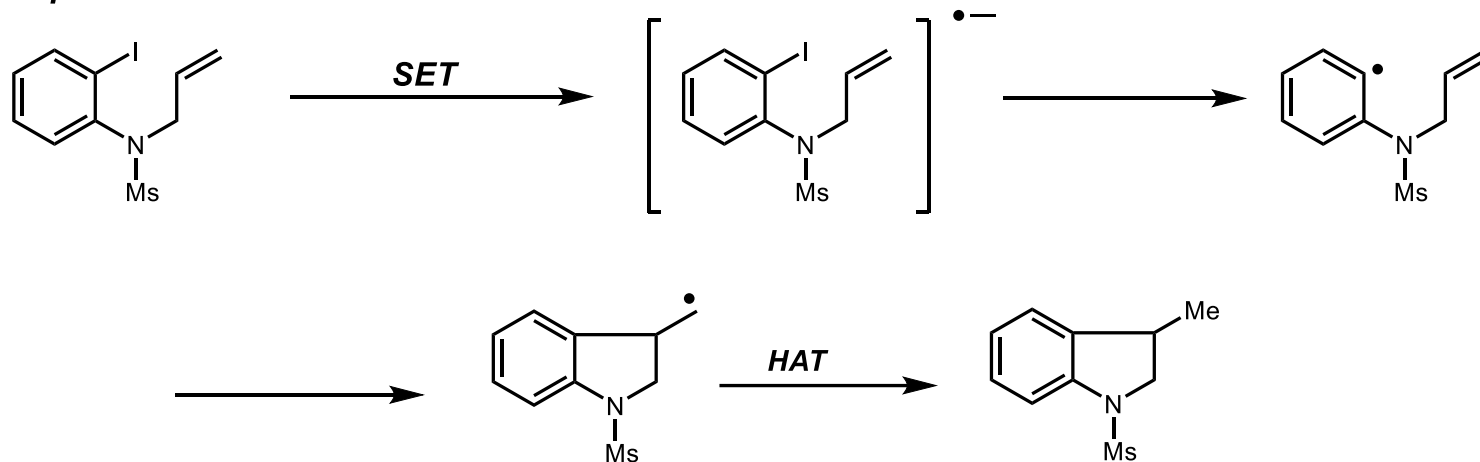
J. A. Murphy, et al. *Angew. Chem., Int. Ed.* **2005**, *44*, 1356

Discovery

The application of **7**



Proposed Mechanism

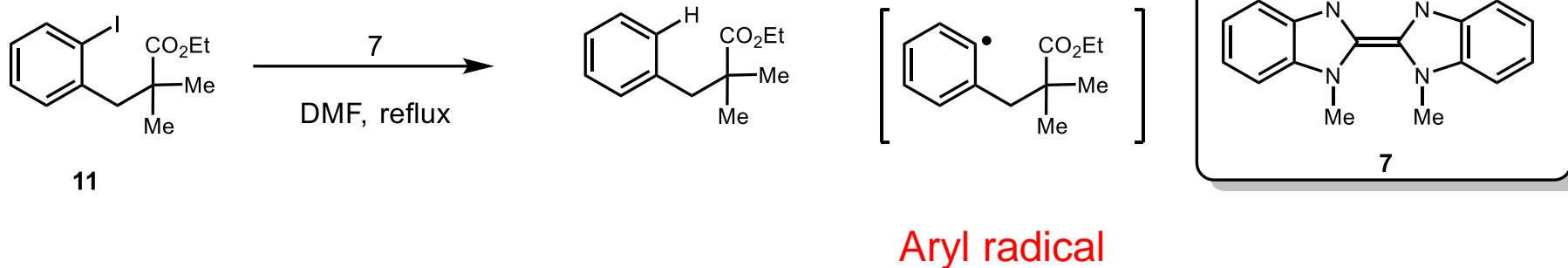
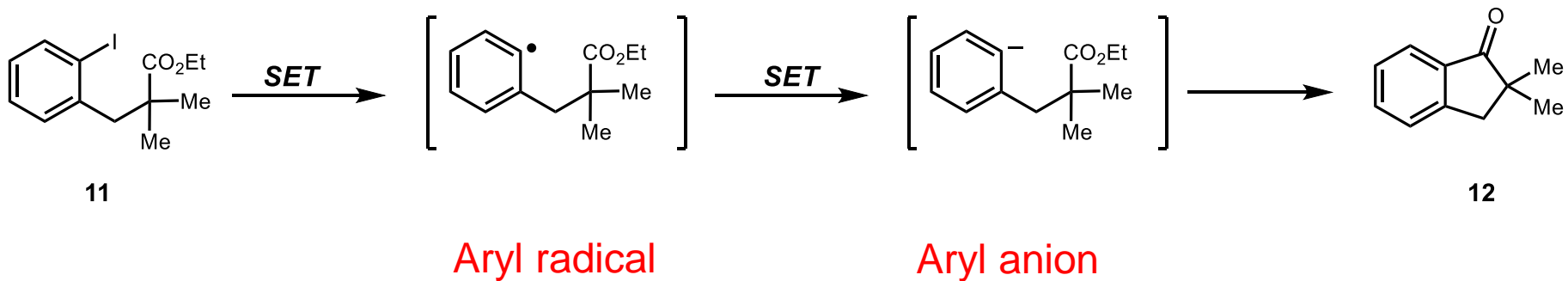


7 is the first neutral organic molecule to form aryl radicals from iodoarenes !

J. A. Murphy, et al. *Angew. Chem., Int. Ed.* **2005**, *44*, 1356

Discovery

Hypothesis

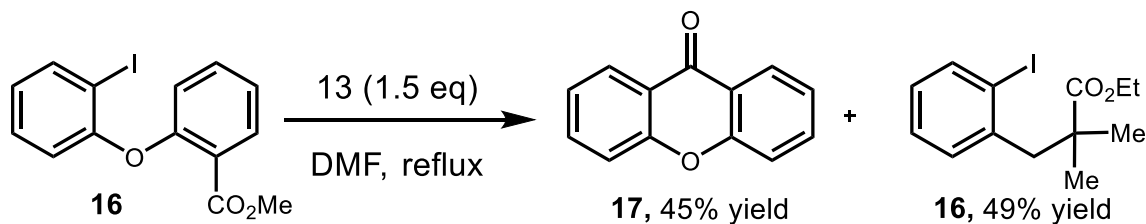
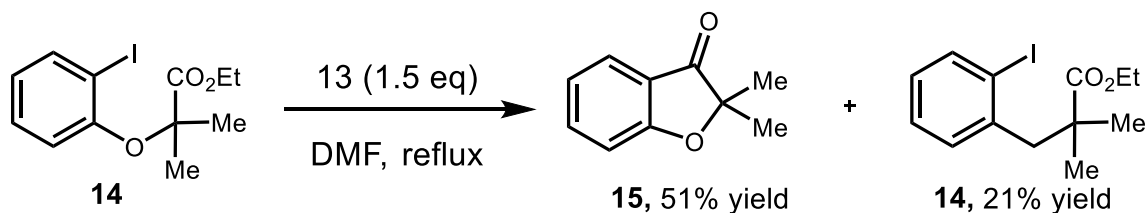
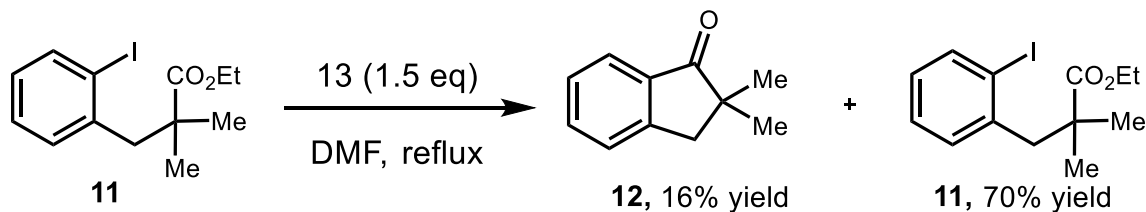
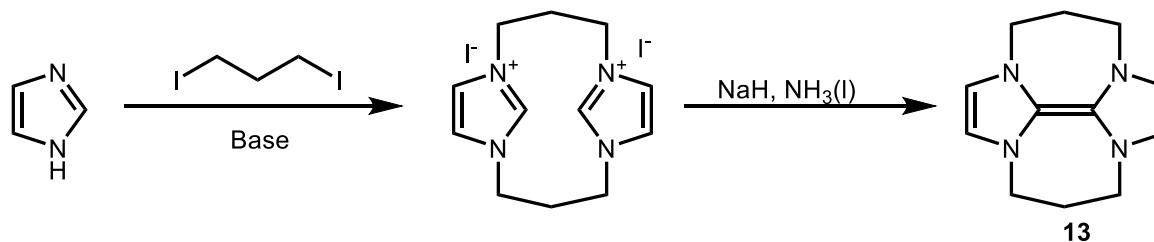


We need a more-reactive Organic Electron Donor !

J. A. Murphy; T. Tuttle; et al. *Angew. Chem., Int. Ed.* **2007**, *46*, 5178

Discovery

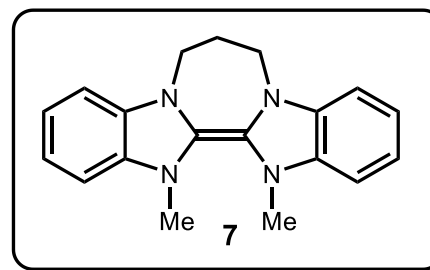
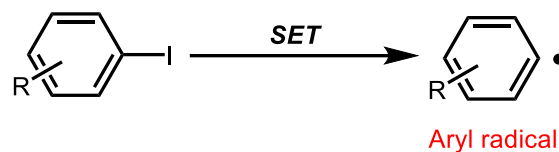
The electron donor 13



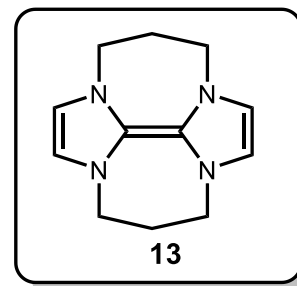
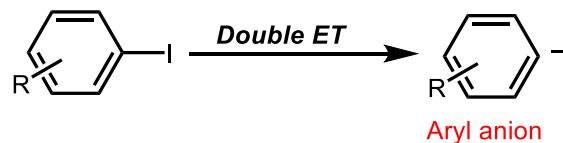
J. A. Murphy, T. Tuttle, et al. *Angew. Chem., Int. Ed.* **2007**, *46*, 5178

Super-electron Donors

Concept



The first neutral organic molecule to form *aryl radicals* from iodoarenes



The first neutral organic molecule to form *aryl anions* from iodoarenes

J. A. Murphy, et al. *Angew. Chem., Int. Ed.* **2005**, *44*, 1356

J. A. Murphy, T. Tuttle, et al. *Angew. Chem., Int. Ed.* **2007**, *46*, 5178

Super-electron Donors

Advantages

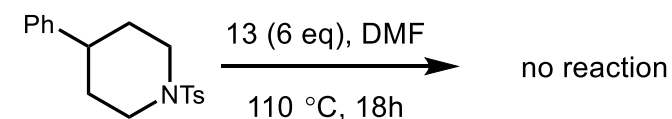
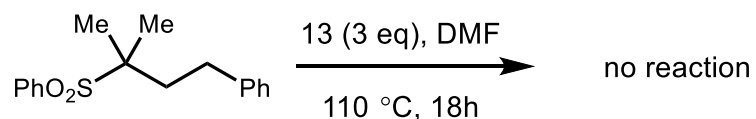
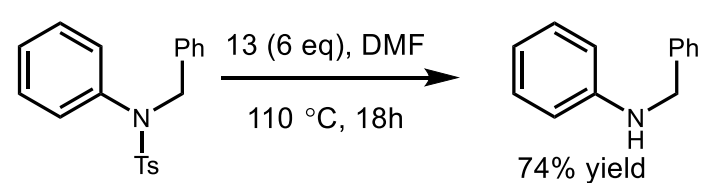
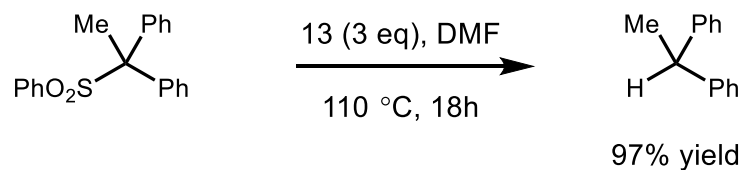
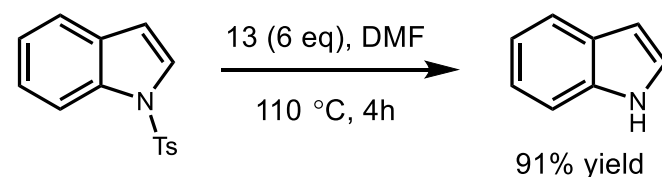
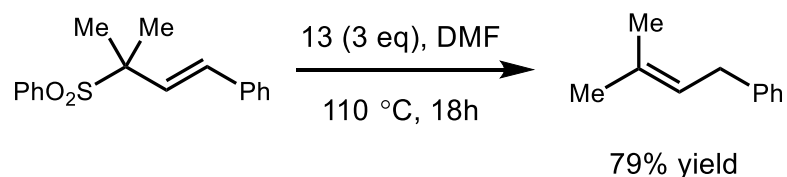
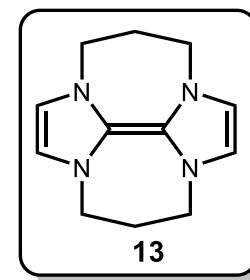
Under very mild conditions because of their neutrality

In the absence of metal ions

With wider applicability than in the case of photochemically assisted reactions

Super-electron Donors

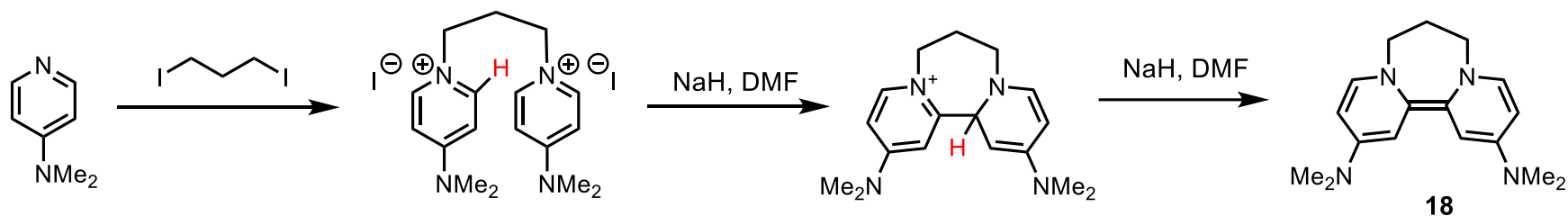
Reductive Cleavage of Sulfones and Sulfonamides



J. A. Murphy; T. Tuttle, et al. *J. Am. Chem. Soc.* **2007**, *129*, 13368

Super-electron Donors

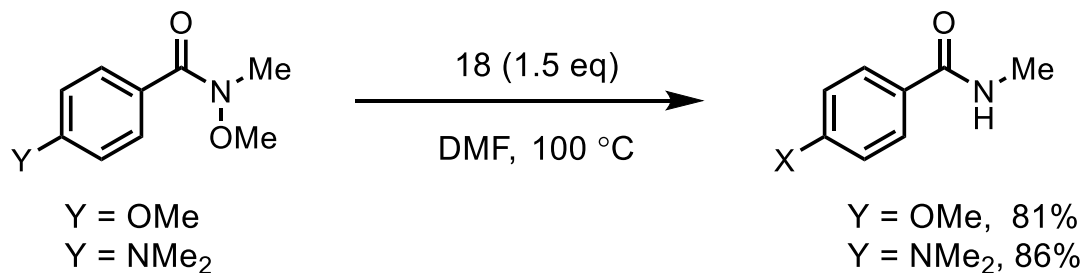
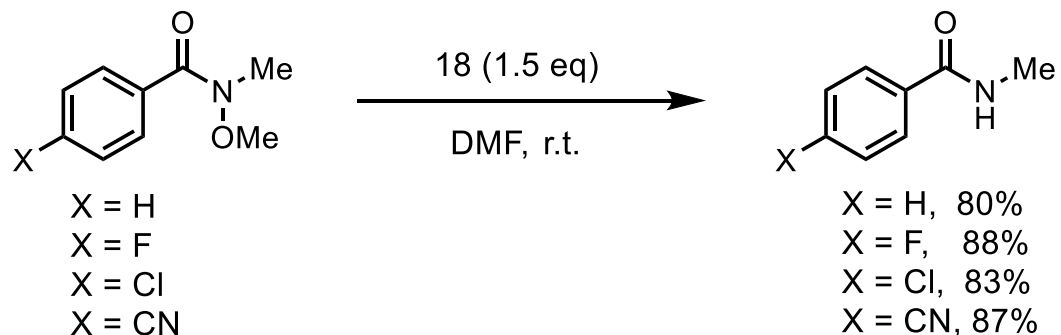
A new *Super-electron-donor* derived from DMAP



18 is a more-reactive *Super-electron donor*

Super-electron Donors

Reductive Cleavage of N–O Bonds in Weinreb Amides

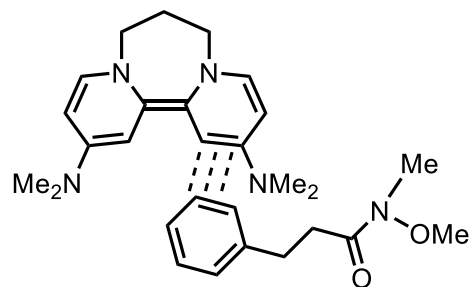
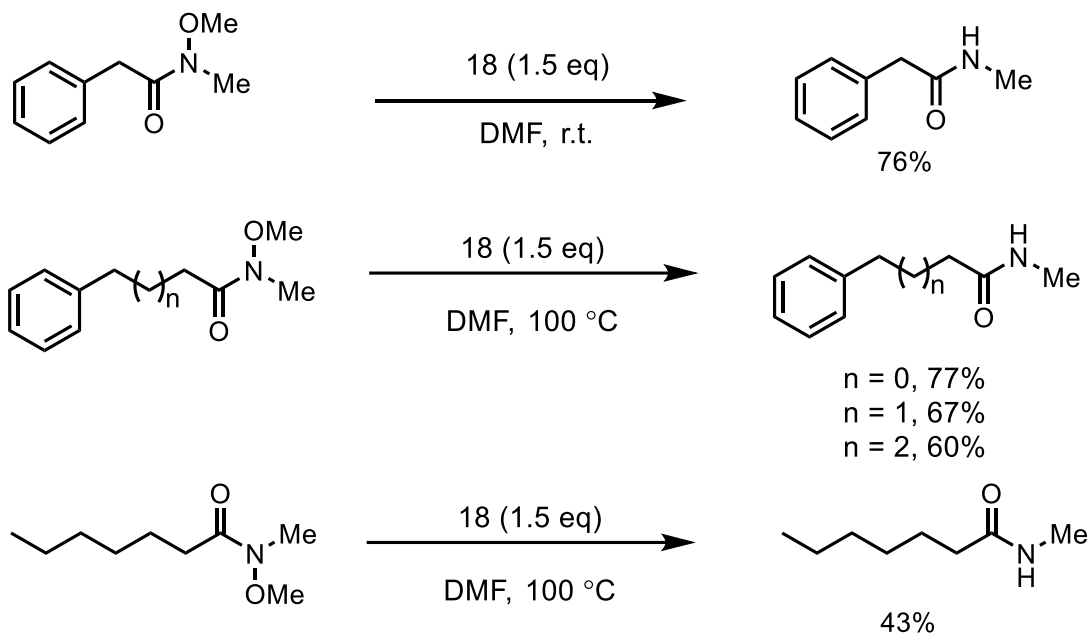


Neighbouring-group electron-transfer effect

J. A. Murphy. et al. *Synlett*, **2008**, 14, 2132

Super-electron Donors

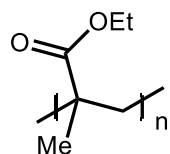
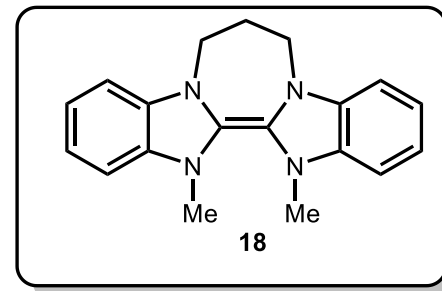
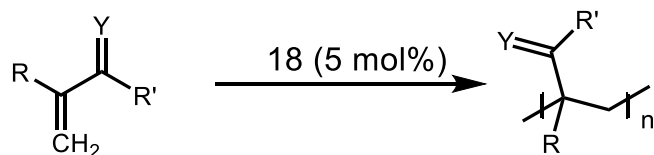
Reductive Cleavage of N–O Bonds in Weinreb Amides



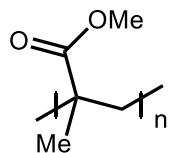
π -stacking interaction

Super-electron Donors

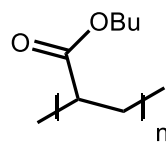
Polymerization Initiated by *SED*



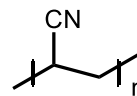
92% Conv.



95% Conv.

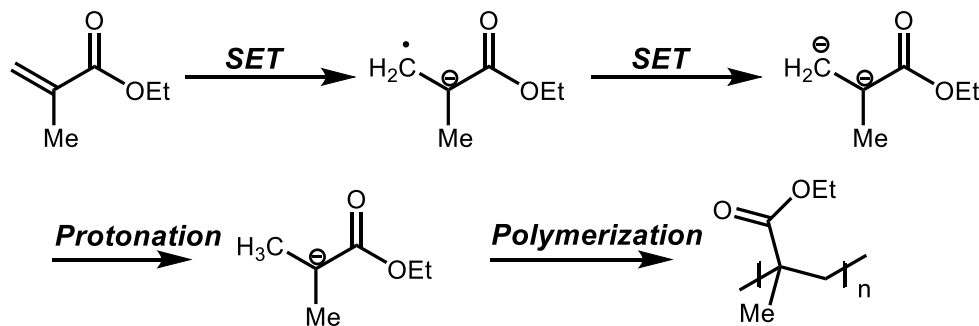


99% Conv.



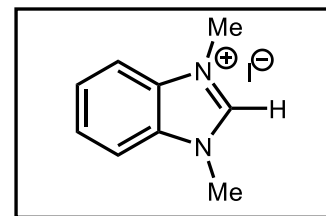
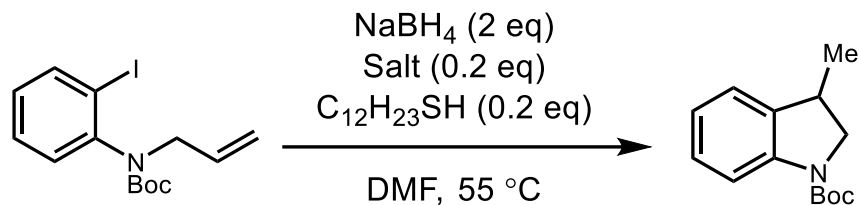
99% Conv.

Proposed mechanism

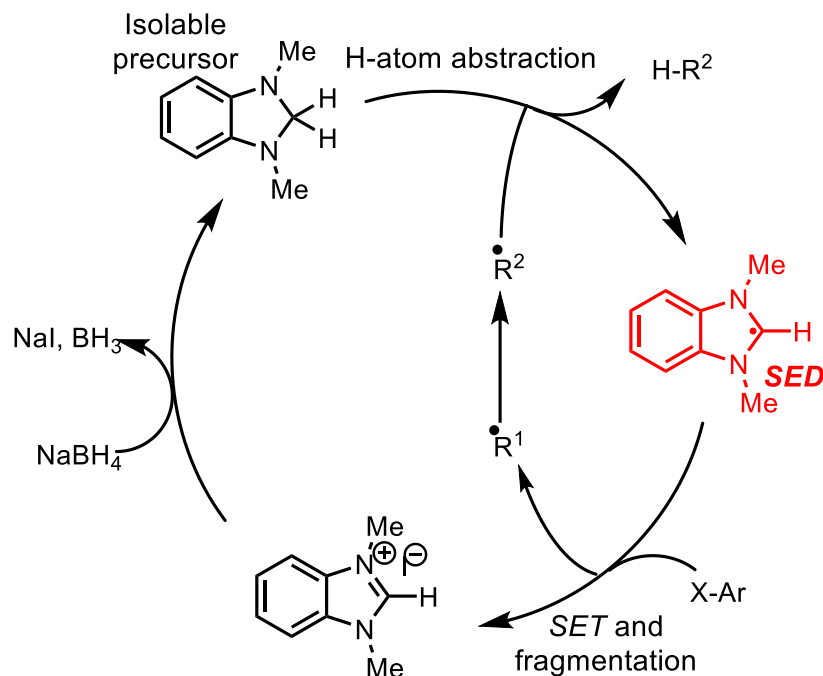


Super-electron Donors

SEDs catalyze radical chain reactions



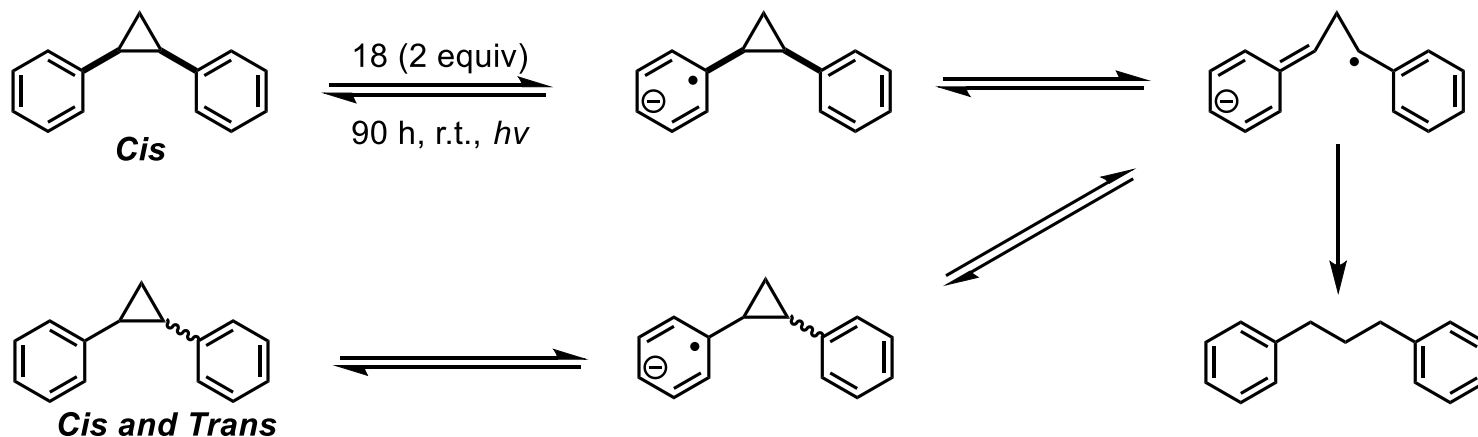
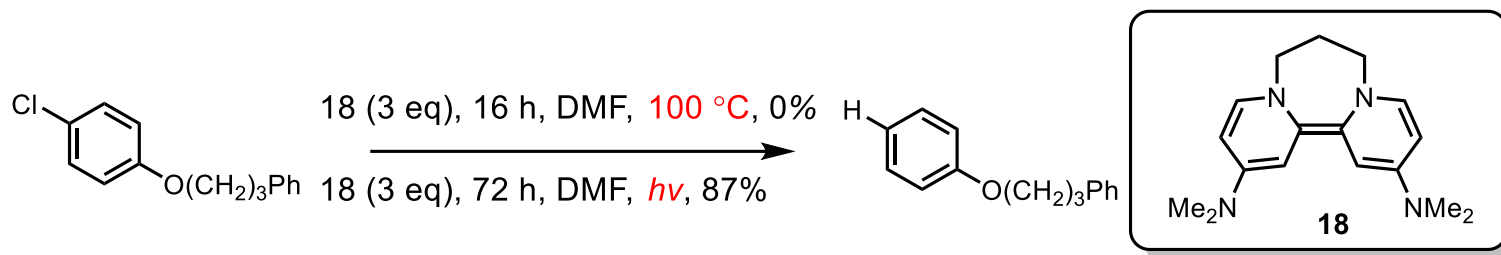
Plausible mechanism



Photoactivation of *SEDs*

Activating Benzenes by photoactivated *SEDs*

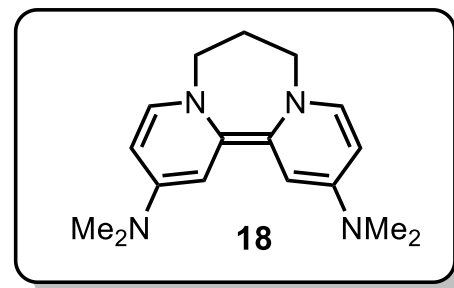
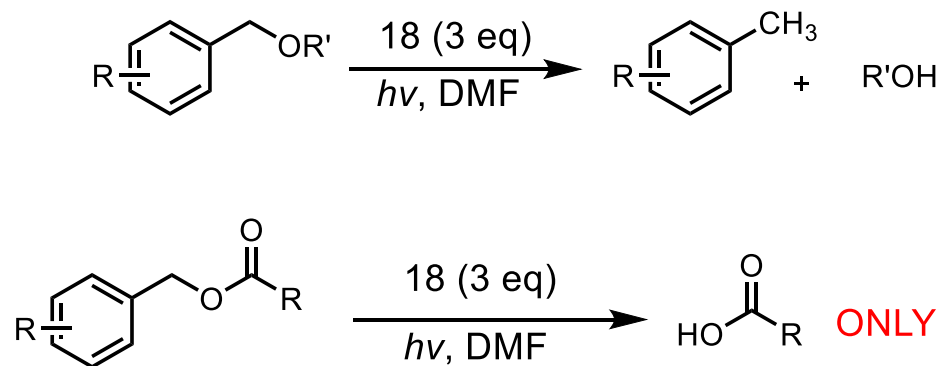
Absorption maxima at 260, 345, and 520 nm



J. A. Murphy, et al. *Angew. Chem., Int. Ed.* **2012**, *51*, 3673

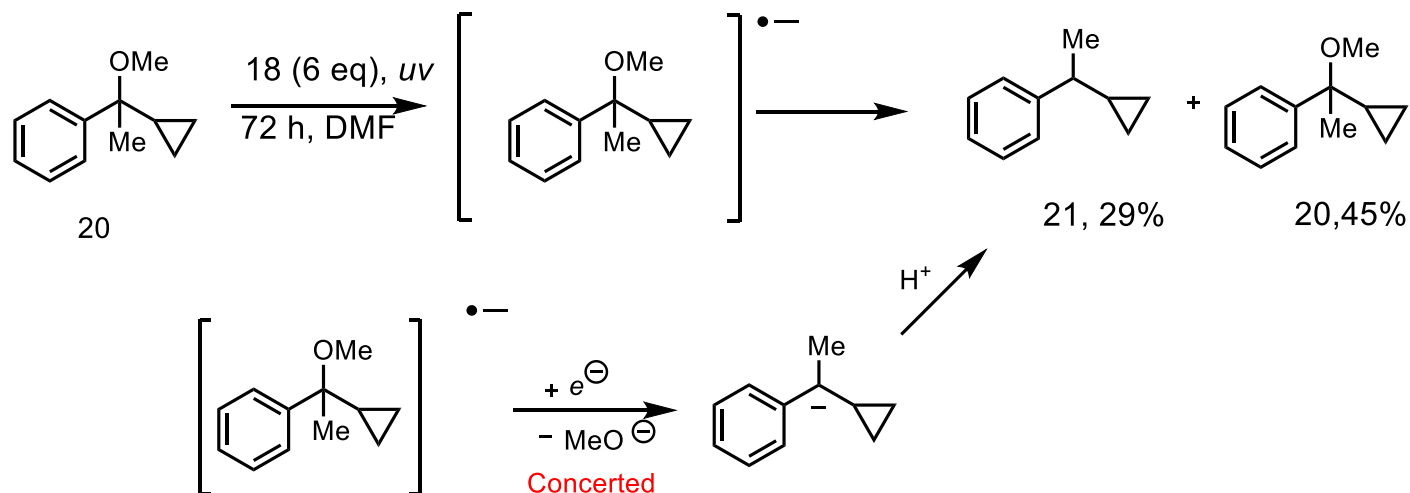
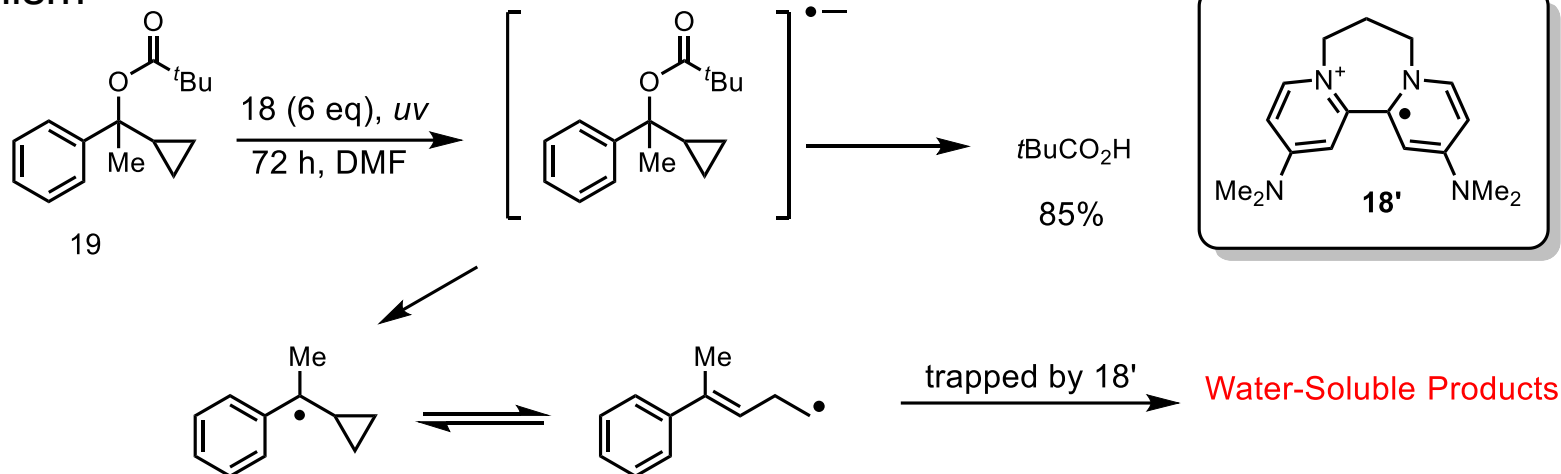
Photoactivation of *SEDs*

Reductive Cleavage of Benzylic Esters and Ethers



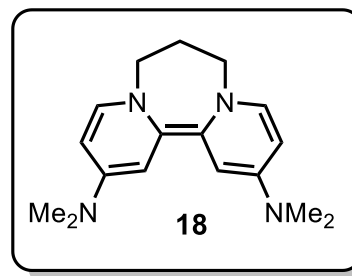
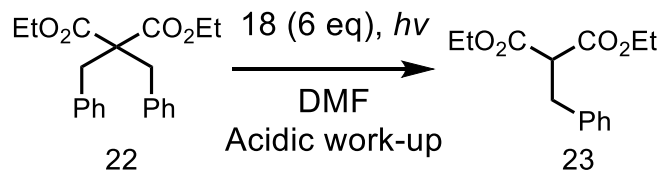
Photoactivation of *SEDs*

Mechanism

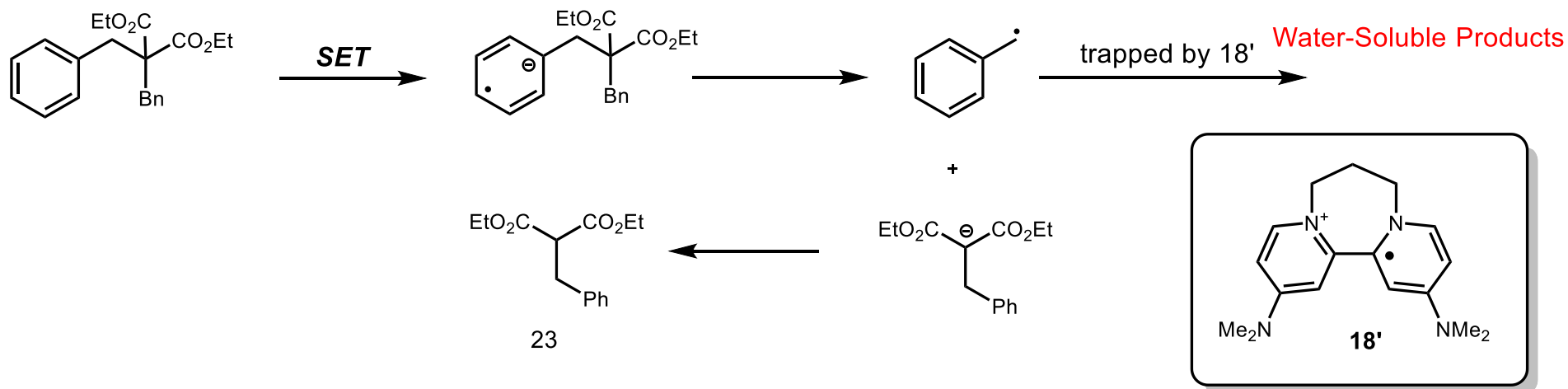


Photoactivation of *SEDs*

Reduction of Arenes by photoactivated *SEDs*

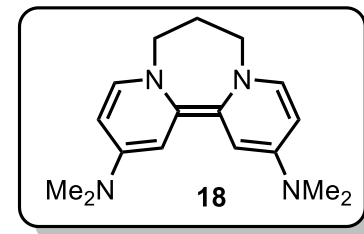
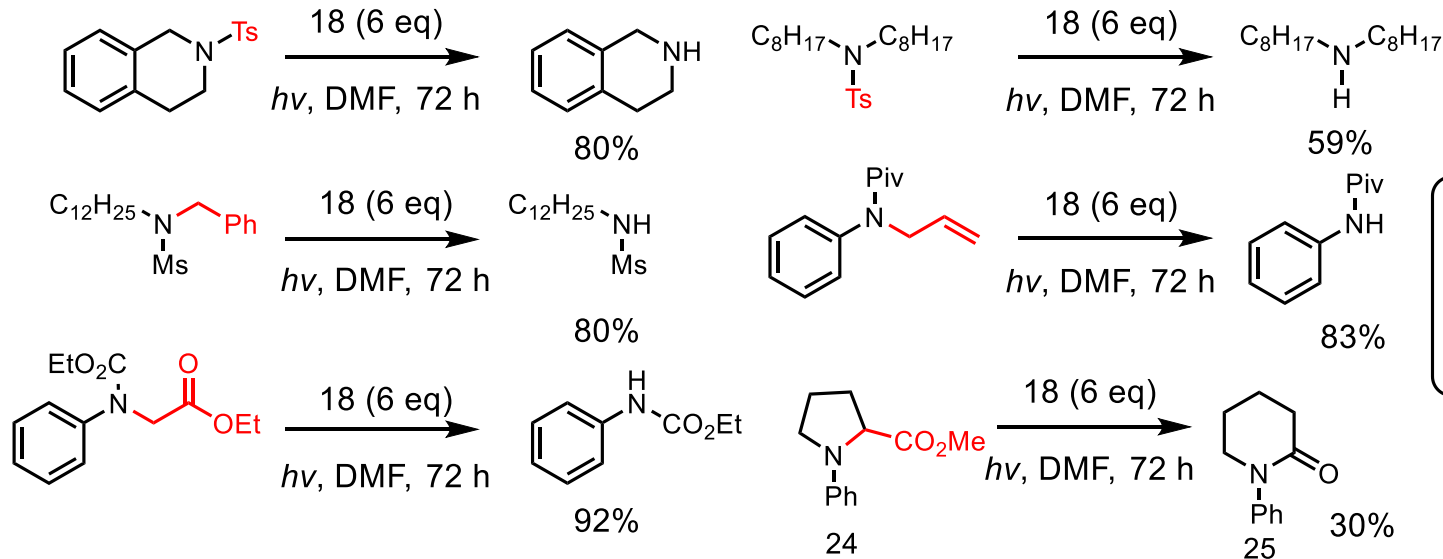


Proposed mechanism

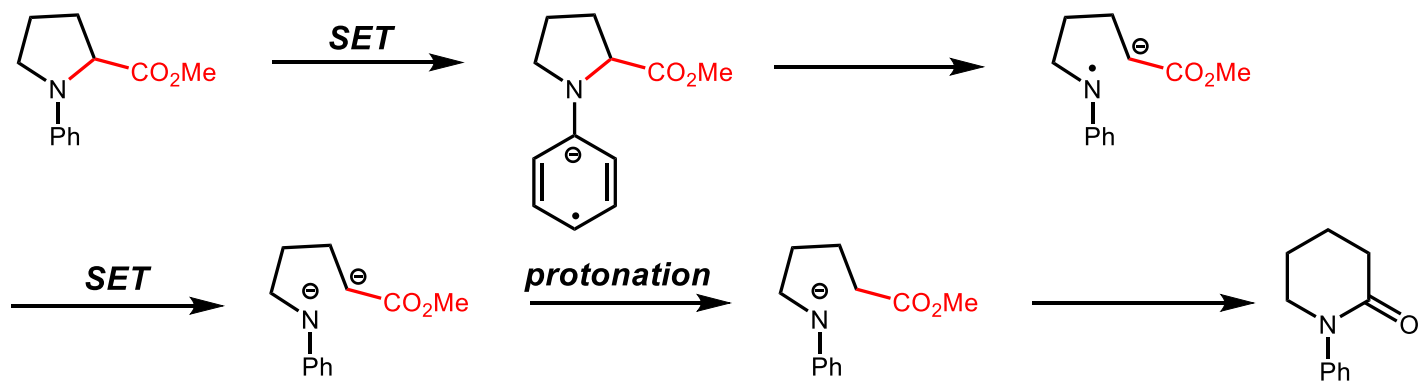


Photoactivation of *SEDs*

Reductive Cleavage of C-N and S-N



Proposed mechanism

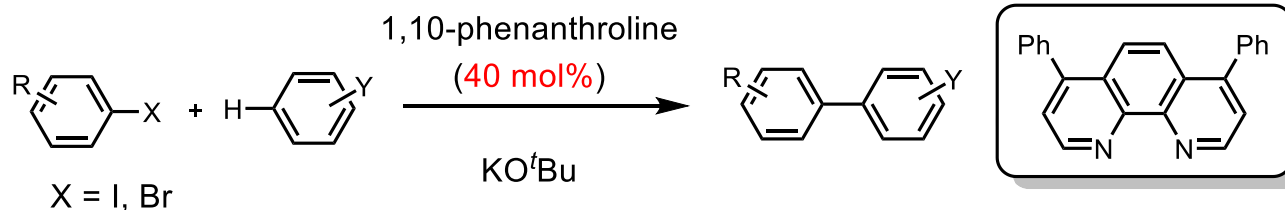


SEDs as initiators in Haloarene–Arene Coupling

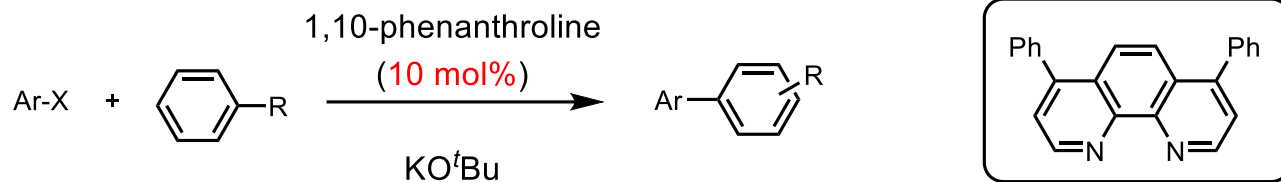
Background



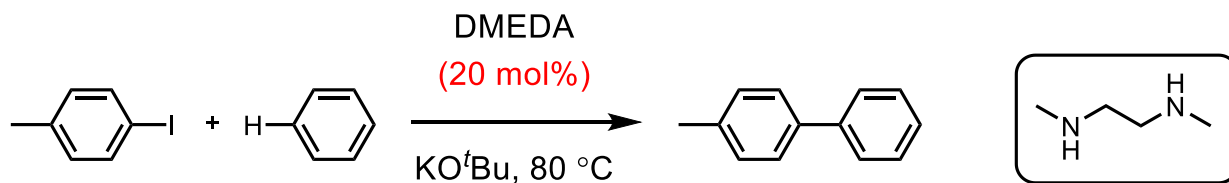
T. Itami, et al. *Org. Lett.* **2008**, 10, 4673.



Shi, Z. J. et al. *Nat. Chem.* **2010**, 2, 1044



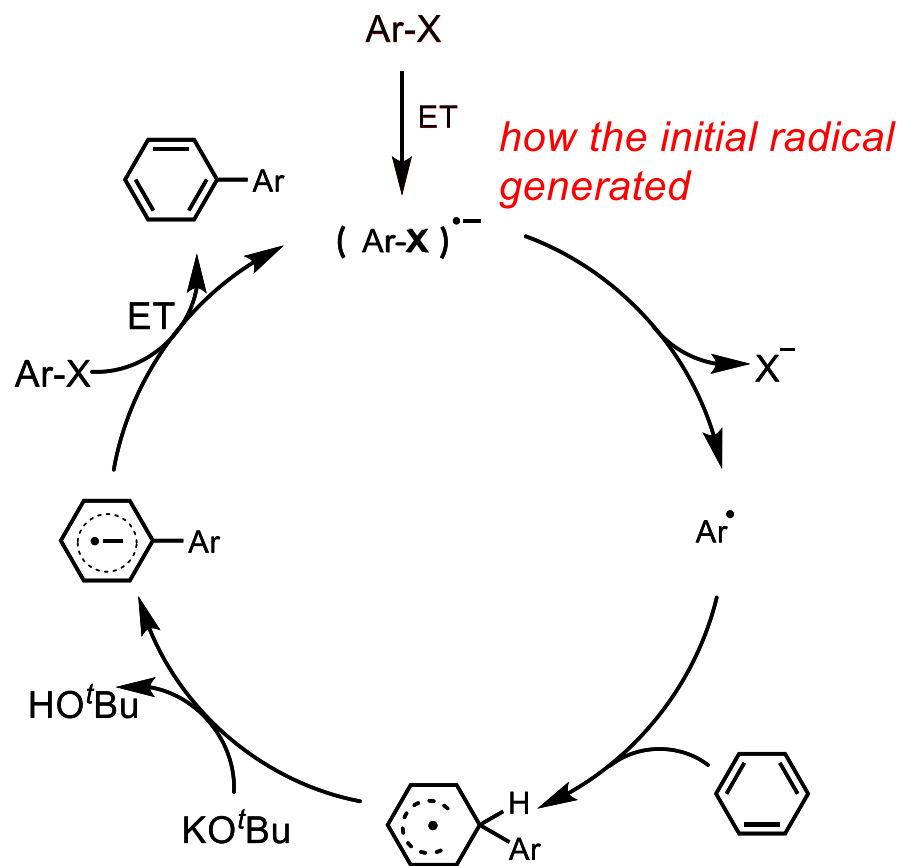
T. Hayashi, et al. *J. Am. Chem. Soc.* **2010**, 132, 15537



Lei, A. W. et al. *J. Am. Chem. Soc.* **2010**, 132, 16737

SEDs as initiators in Haloarene–Arene Coupling

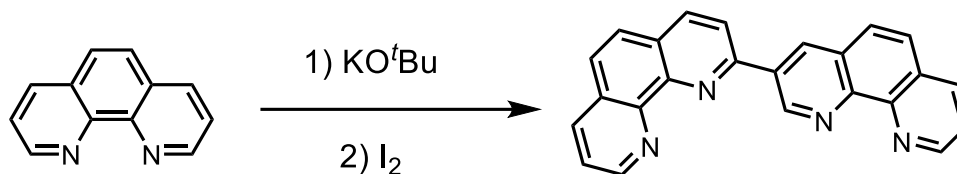
Base-promoted homolytic aromatic substitution (**BHAS**) reactions



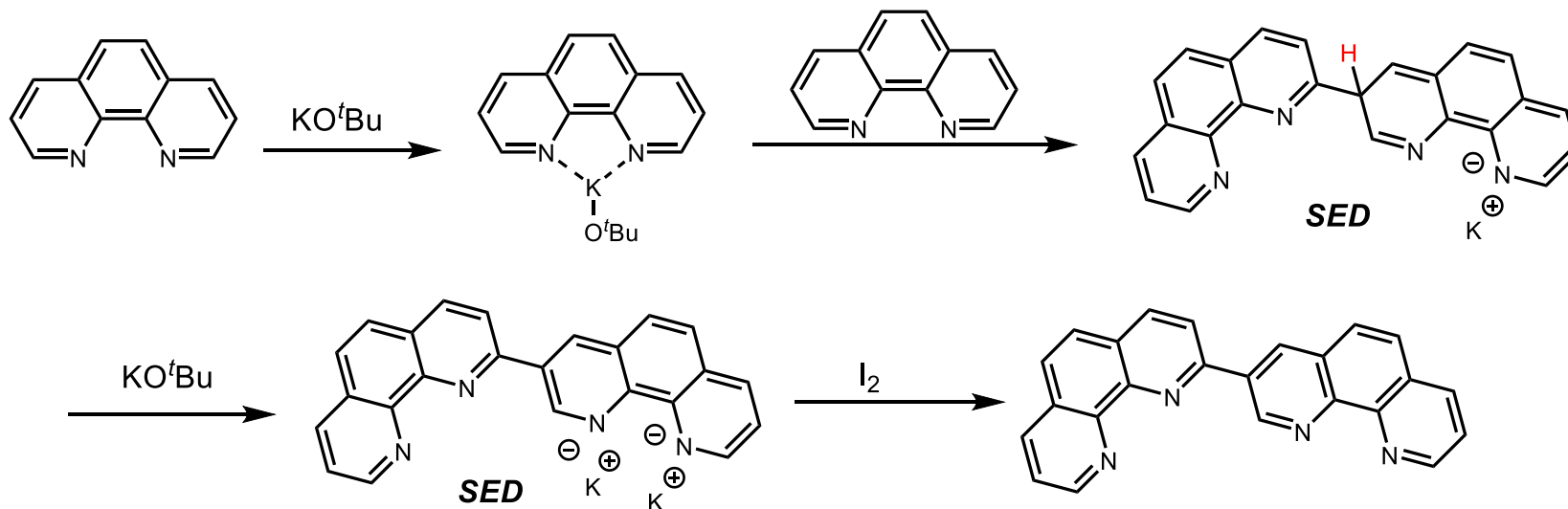
A. Studer; D. P. Curran, *Angew. Chem., Int. Ed.* **2011**, *50*, 5018

SEDs as initiators in Haloarene–Arene Coupling

Initiation with phenanthrolines as additives

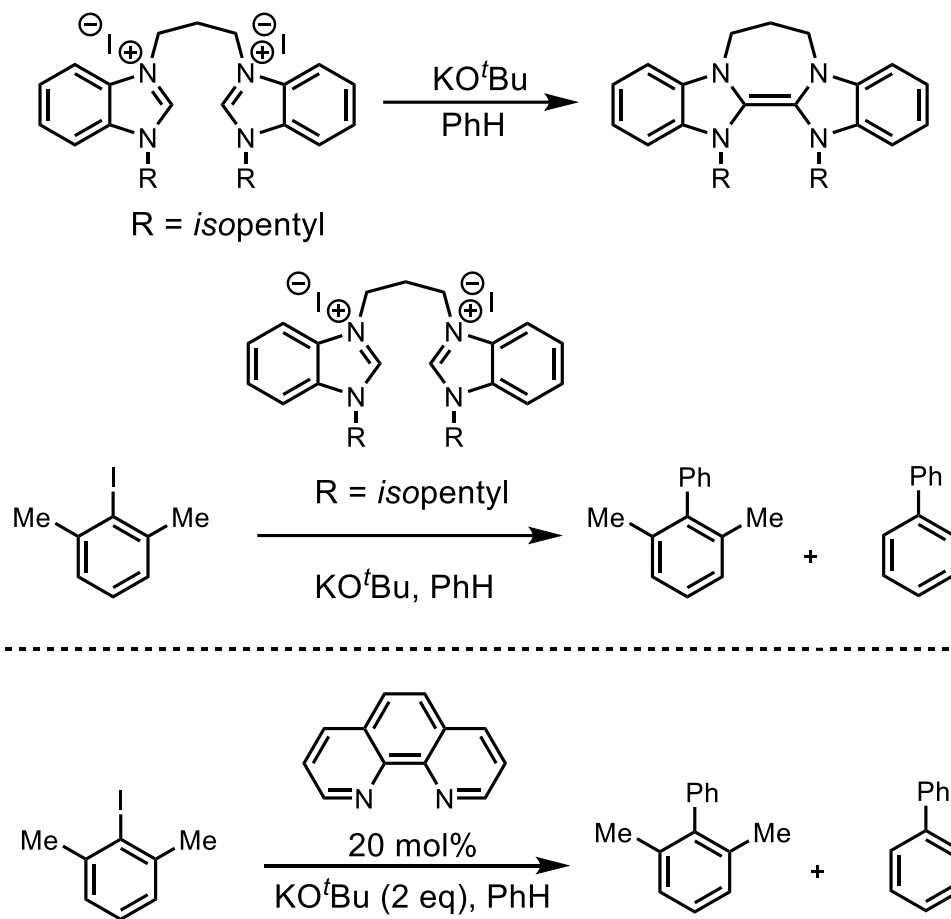


Proposed mechanism



SEDs as initiators in Haloarene–Arene Coupling

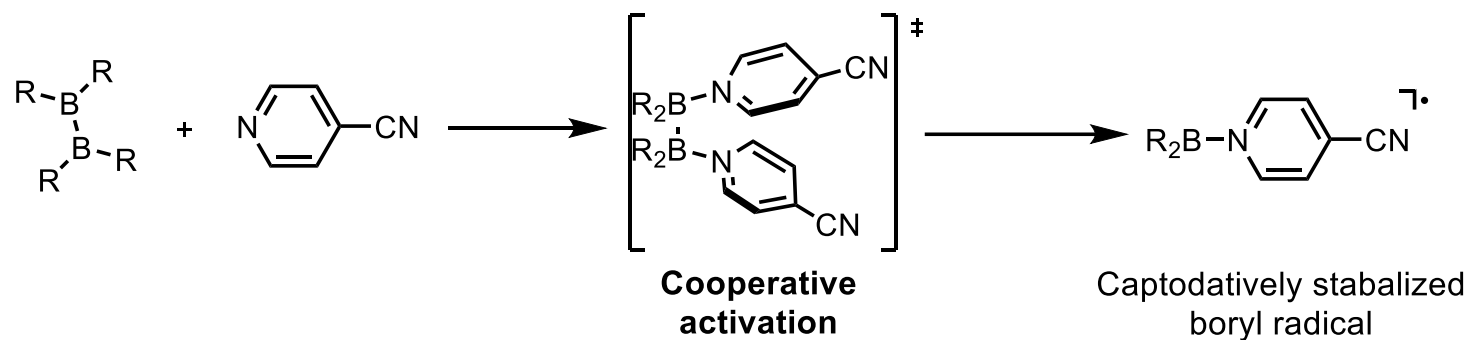
Initiation with phenanthrolines as additives



John A. Murphy, et al. *Chem. Sci.* **2014**, 5, 476

SEDs derived from diborons

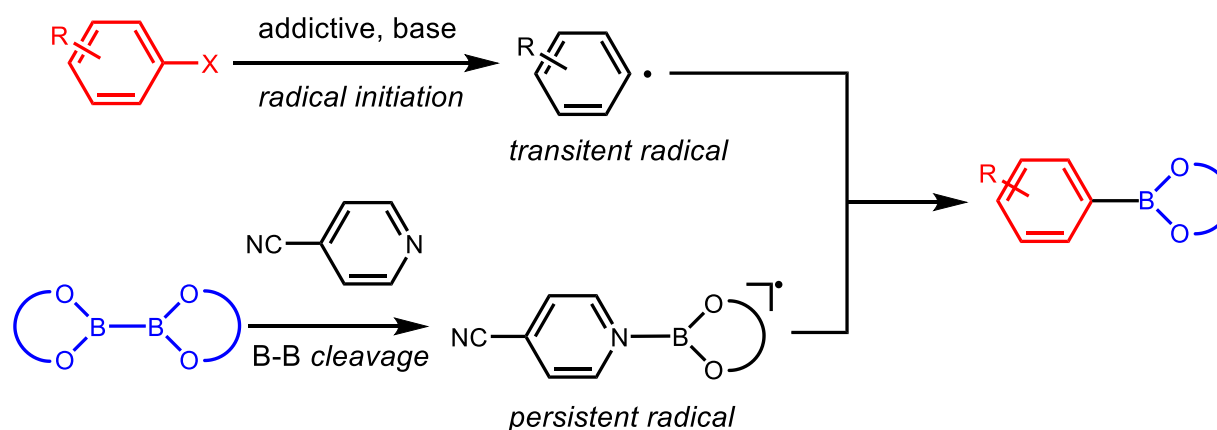
Background



Li, S. H. et al. *Angew. Chem., Int. Ed.* **2016**, *55*, 5985

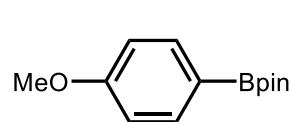
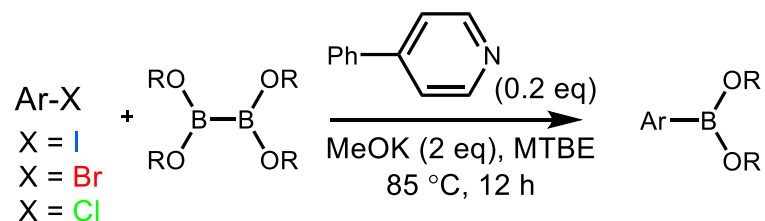
SEDs derived from diborons

Jiao's idea: Radical Borylation of Aryl Halides

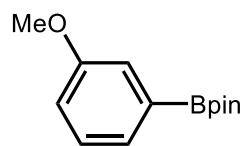


Zhang L.; Jiao L. *J. Am. Chem. Soc.* **2017**, *139*, 607

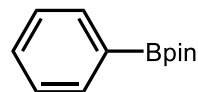
SEDs derived from diborons



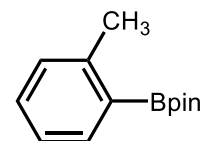
85% X = I
72% X = Br



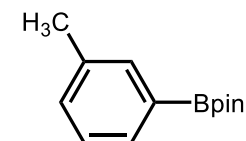
80% X = I
69% X = Br



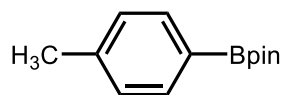
86% X = I
68% X = Br



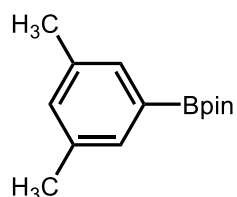
83% X = I
67% X = Br



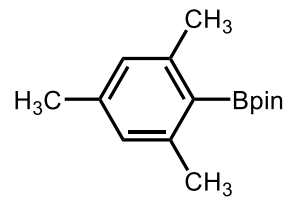
84% X = I



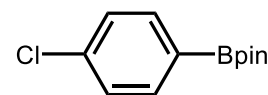
85% X = I



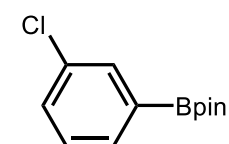
79% X = I
58% X = Br



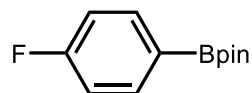
53% X = I



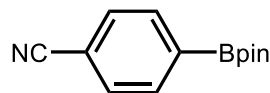
67% X = I



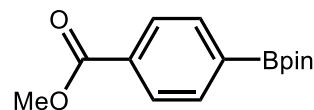
68% X = I



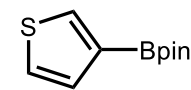
75% X = I
62% X = Br



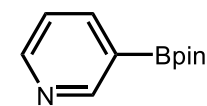
48% X = I
60% X = Cl



62% X = I
61% X = Cl



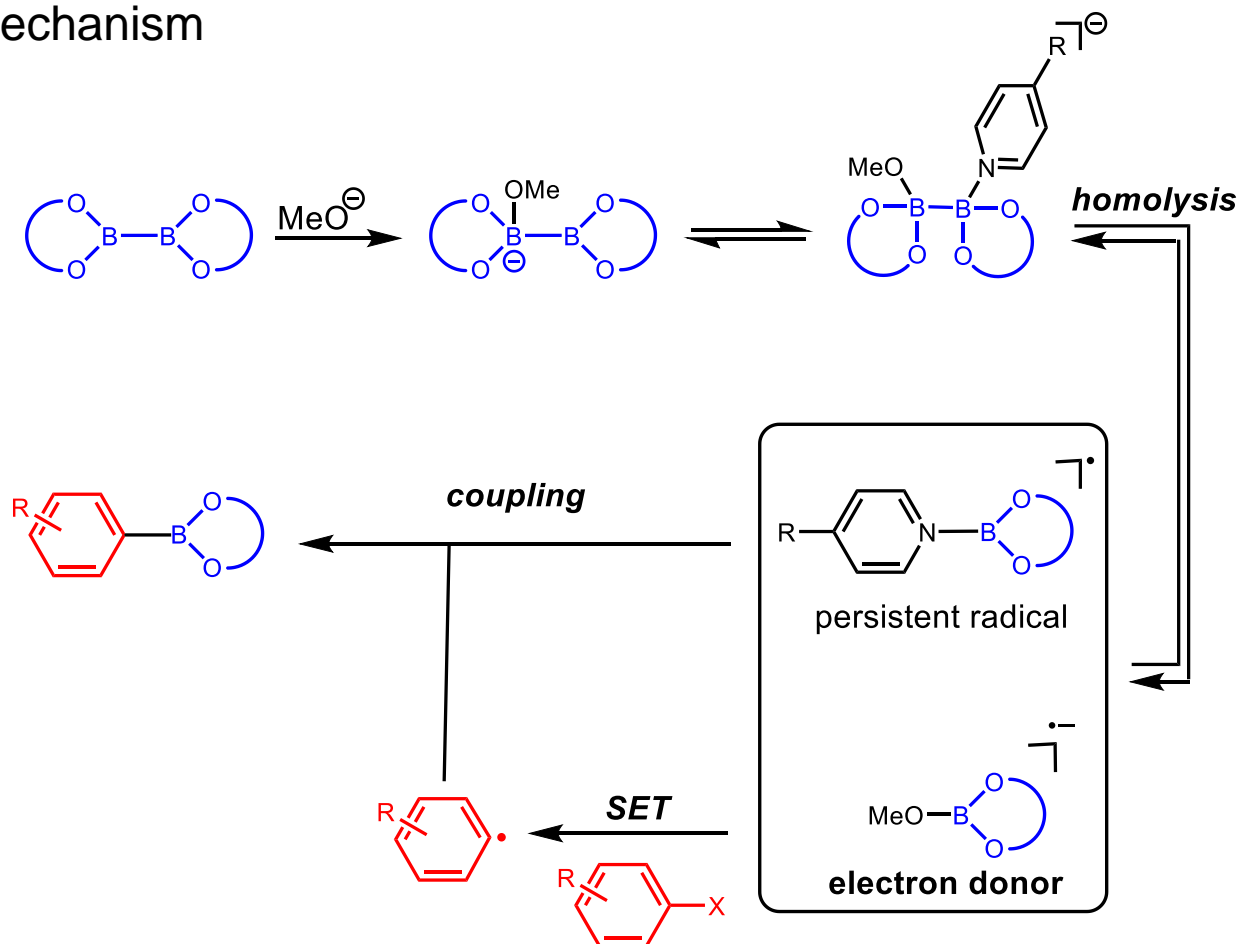
51% X = I



25% X = I

SEDs derived from diborons

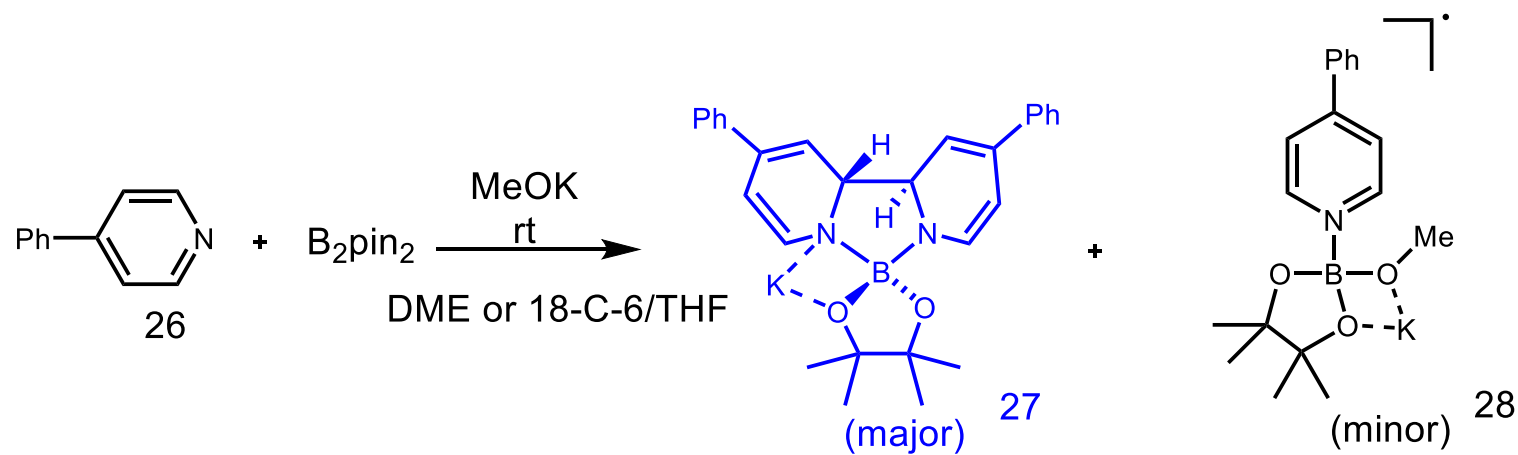
Proposed mechanism



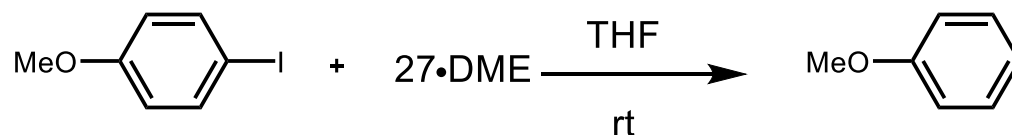
Zhang L.; Jiao L. *J. Am. Chem. Soc.* **2017**, 139, 607

SEDs derived from diborons

The structure of the electron donor



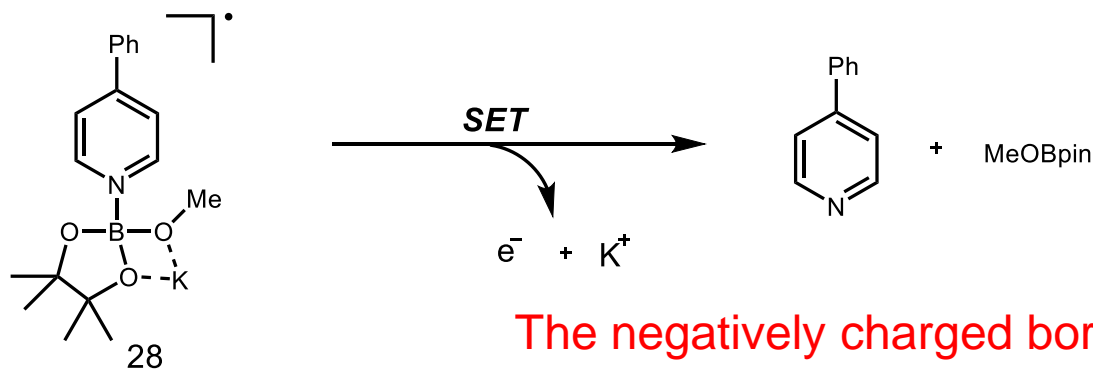
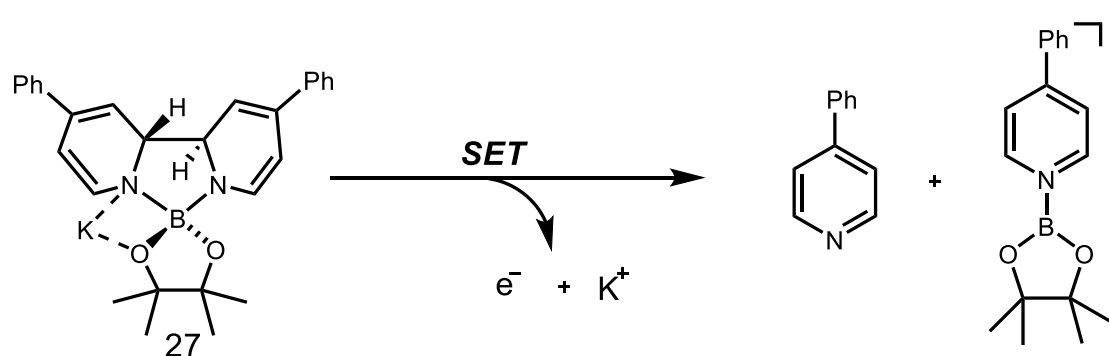
Pyridine- and boron-containing complexes as super electron donors



Zhang L.; Jiao L. *Chem. Sci.* **2018**, 9, 2711

SEDs derived from diborons

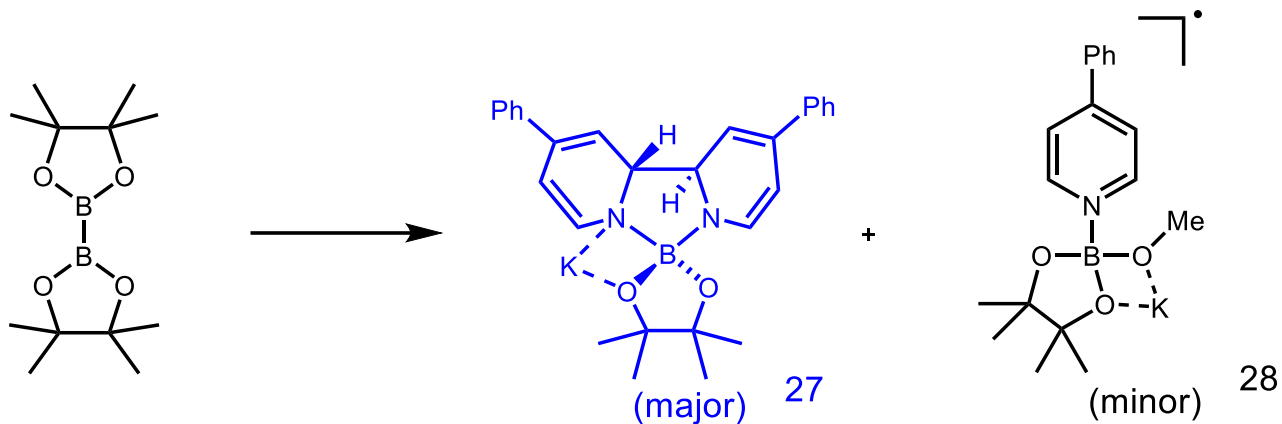
Driving force



The negatively charged boron center
The aromaticity of the pyridine ring

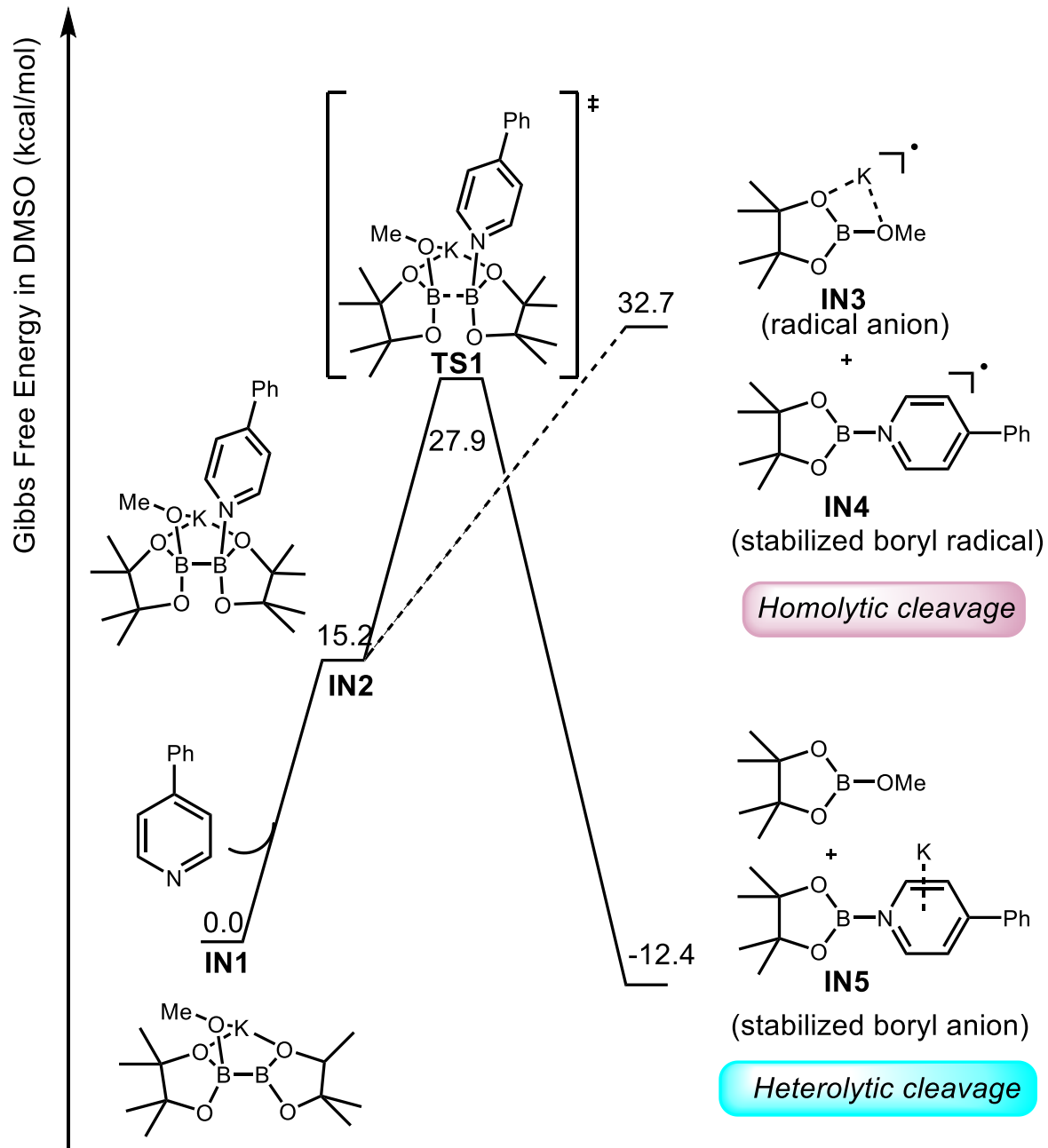
Zhang L.; Jiao L. *Chem. Sci.* **2018**, 9, 2711

SEDs derived from diborons

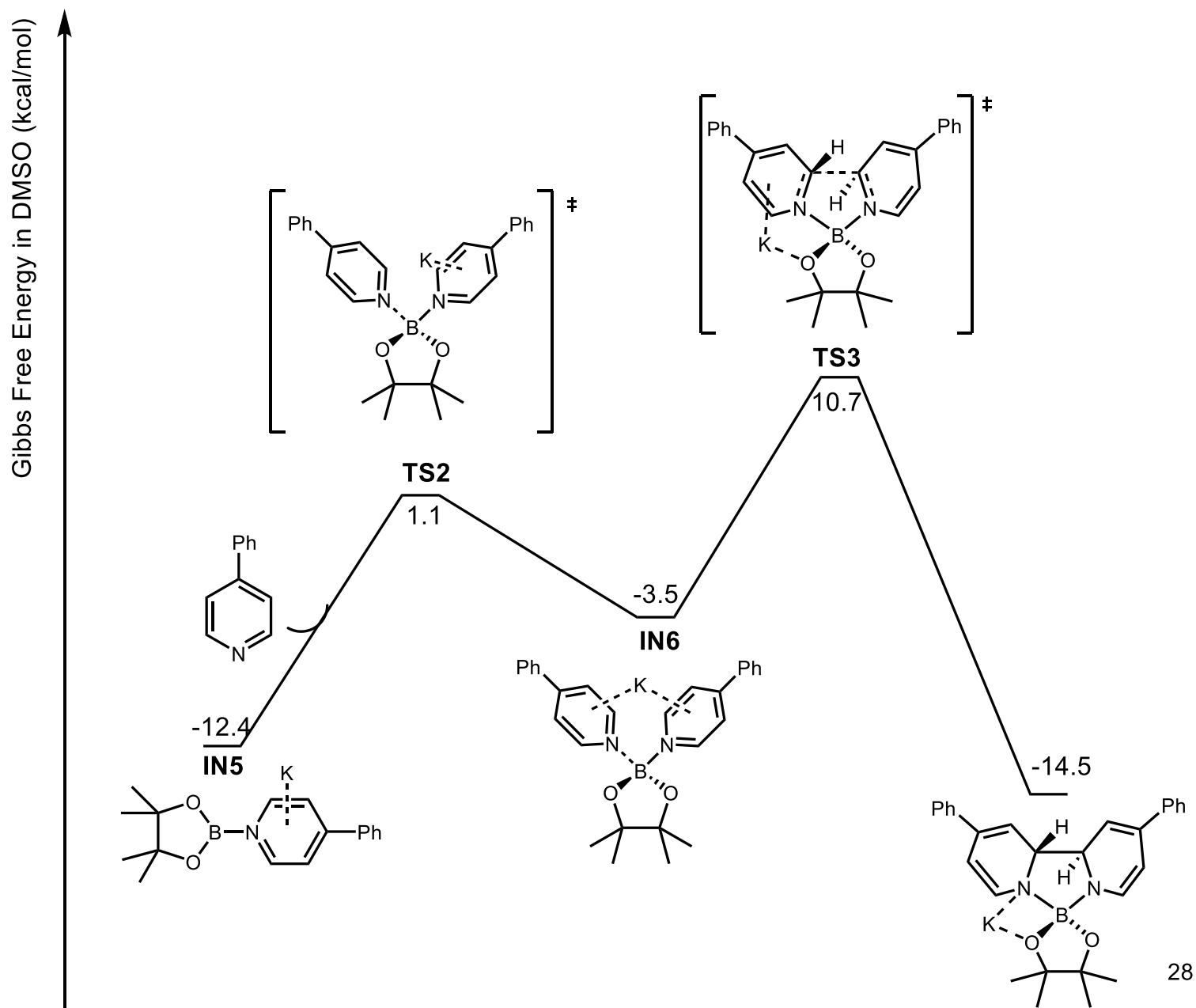


Two questions

- ? The mode of boron–boron bond cleavage
- ? Mechanism for the formation of super-electron-donors



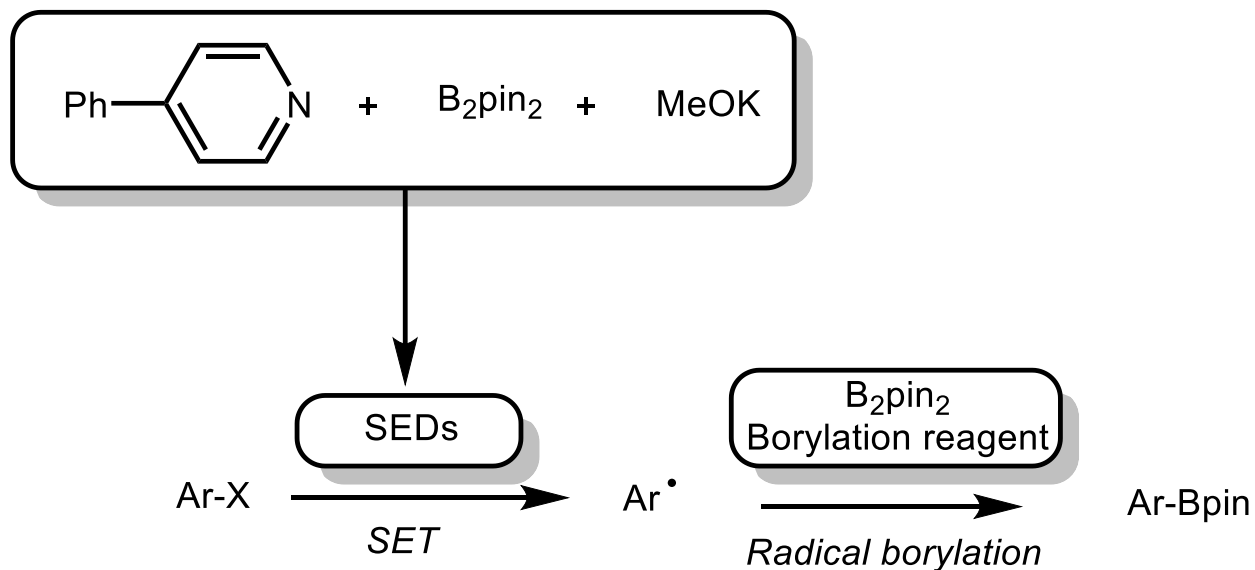
Gibbs free energy profile for the homolytic and heterolytic cleavage pathways of diboron.



Gibbs free energy profile for the formation of the complex

SEDs derived from diborons

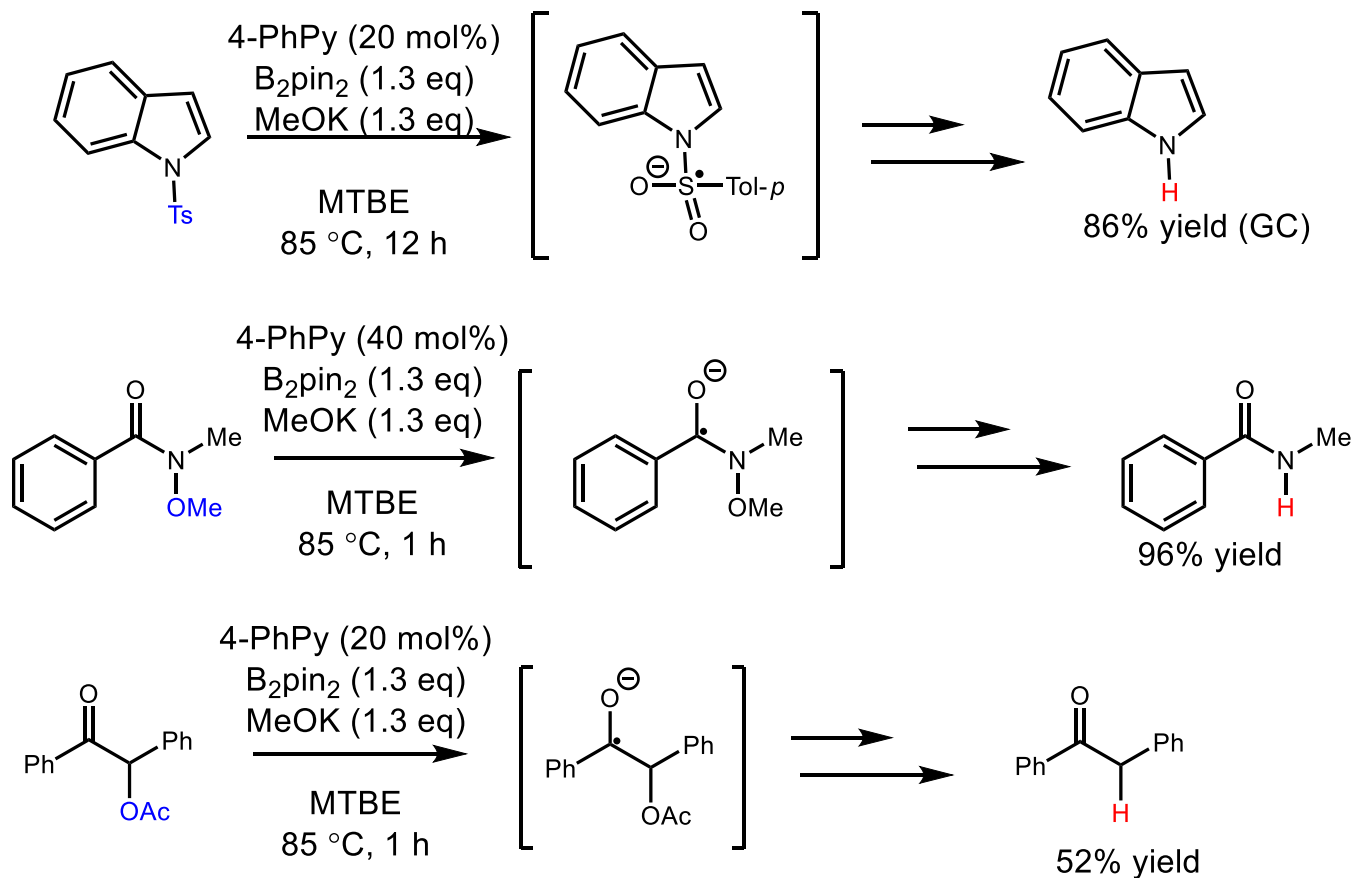
Revised radical borylation mechanism



Zhang L.; Jiao L. *Chem. Sci.* **2018**, 9, 2711

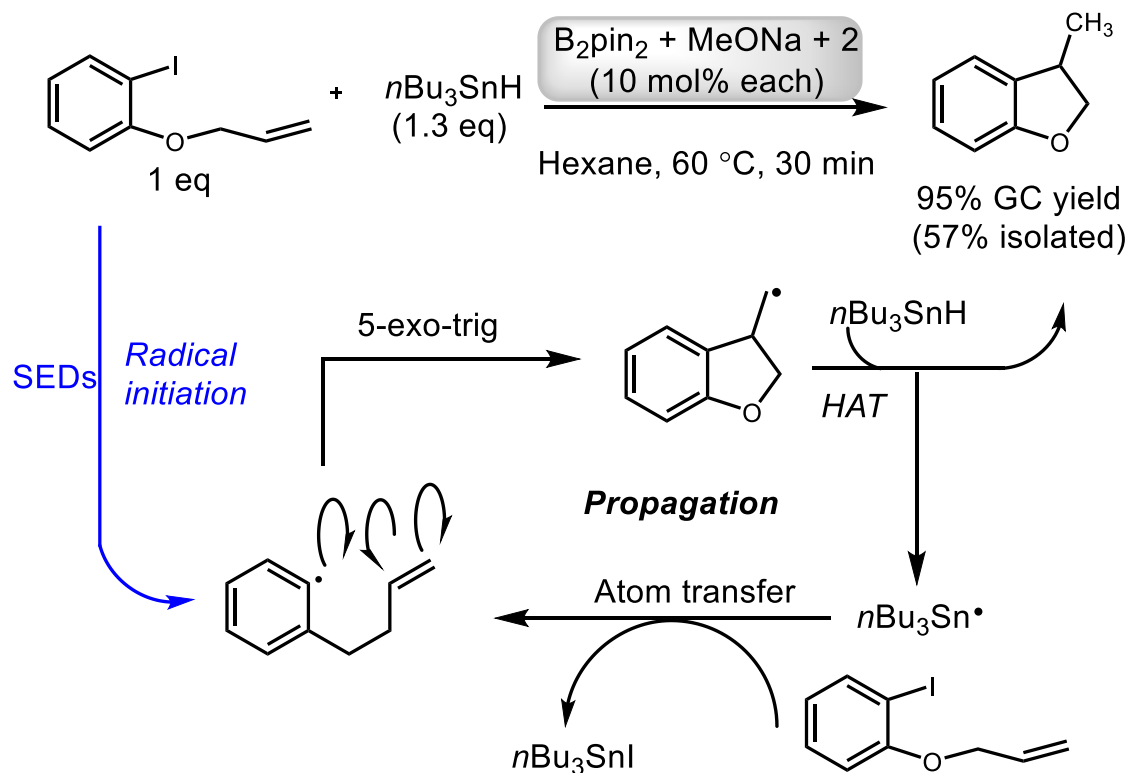
SEDs derived from diborons

Reductive Cleavage of S-N, N-O and C-O



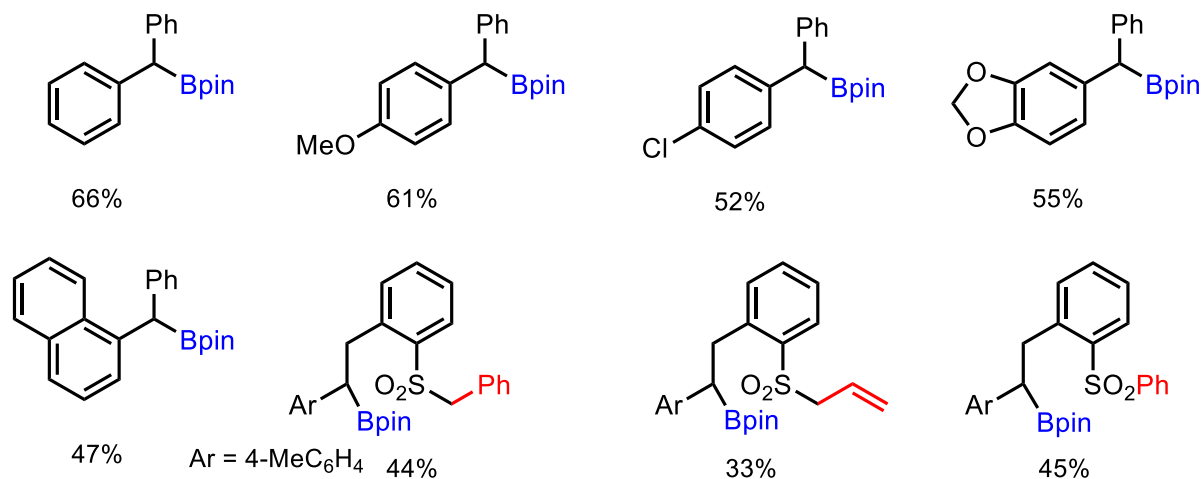
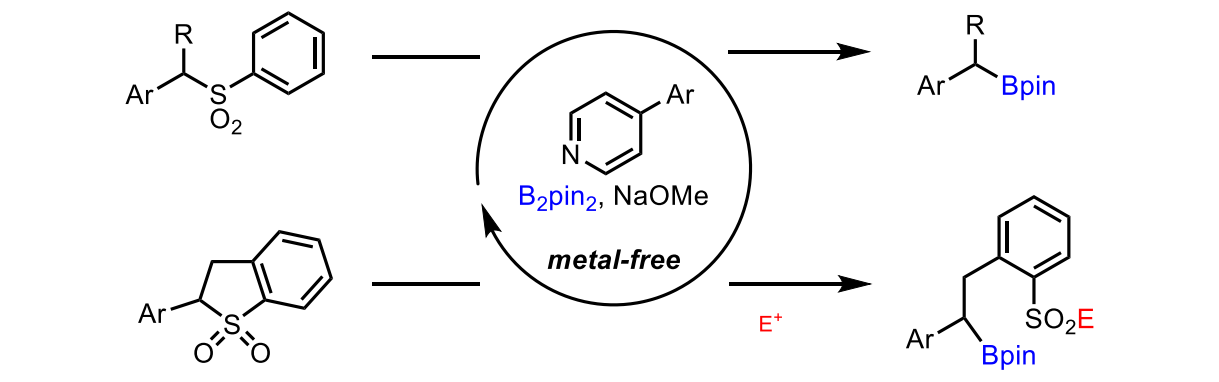
SEDs derived from diborons

SEDs mixture as the radical initiator

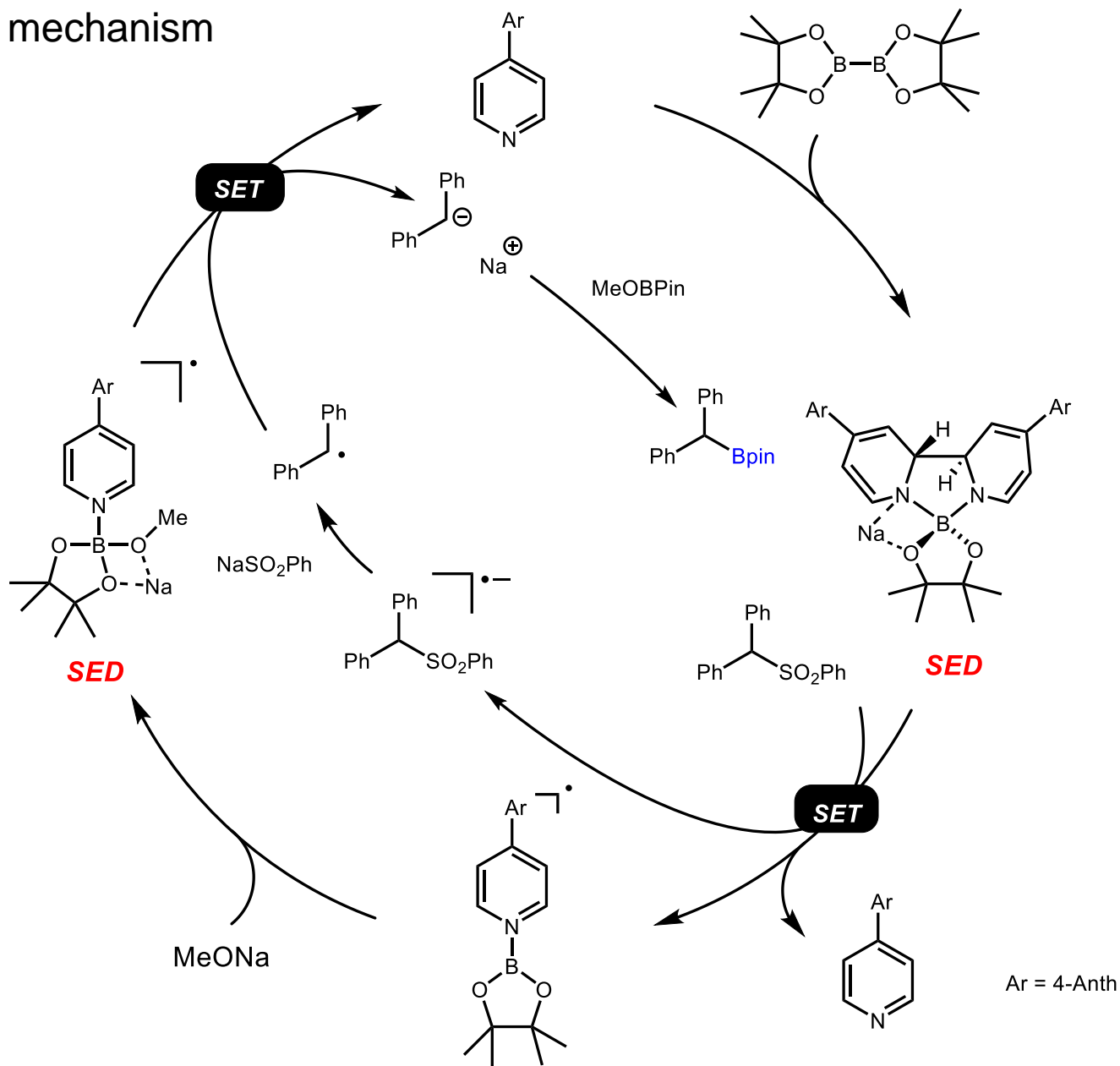


SEDS derived from diborons

Desulfonative borylation of benzyl sulfones

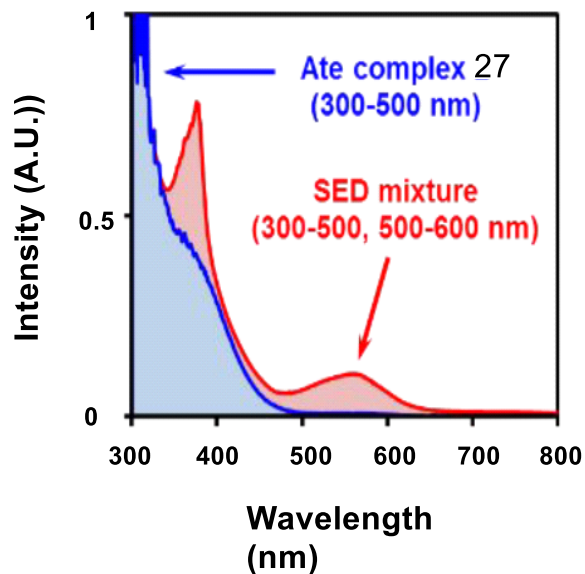
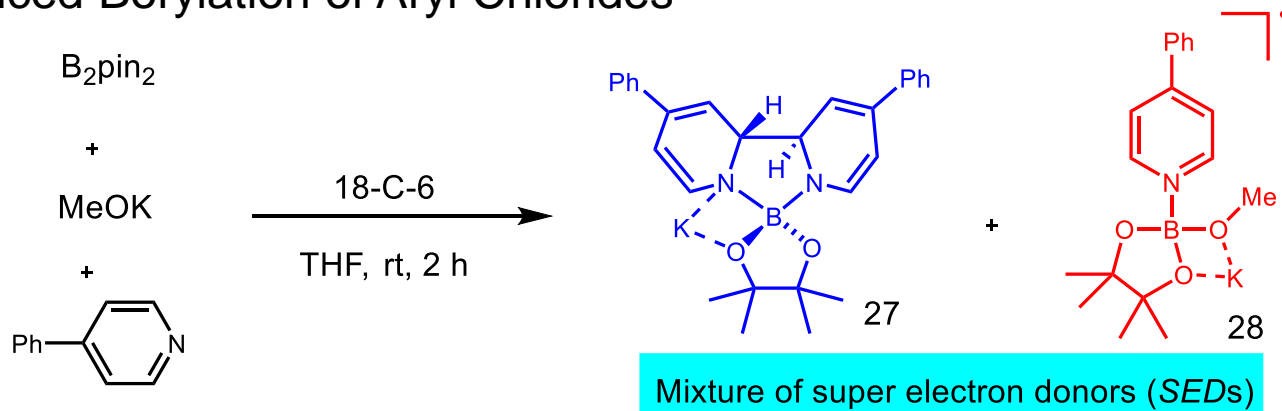


Proposed mechanism



SEDs derived from diborons

Visible-Light-Induced Borylation of Aryl Chlorides

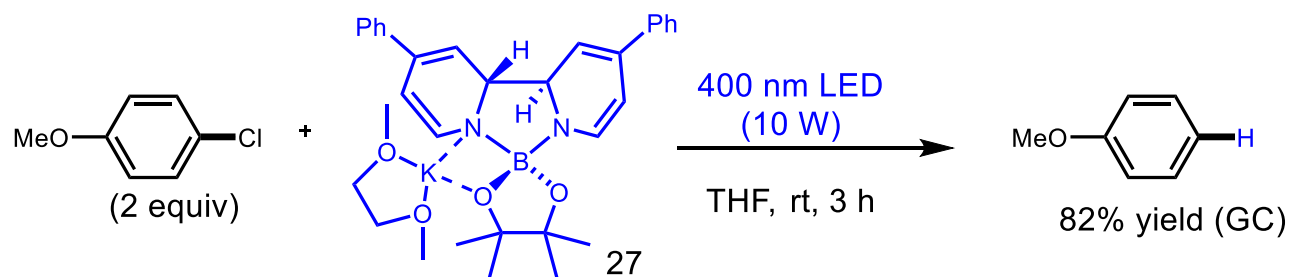
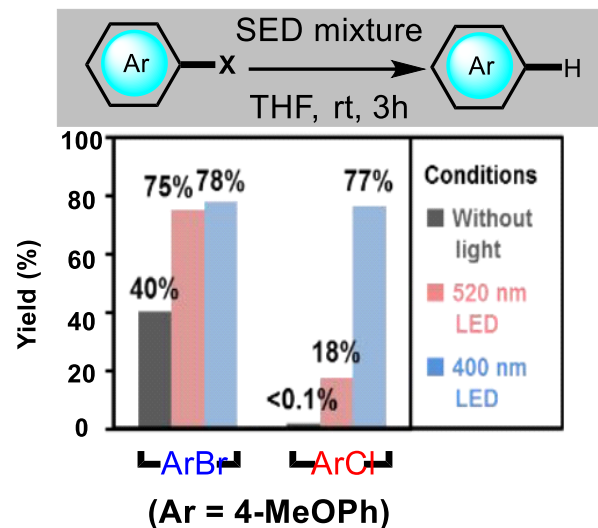


Complex 27 Absorption band 300–500 nm

Complex 28 Absorption band 500–600 nm

SEDs derived from diborons

Visible-Light-Induced Borylation of Aryl Chlorides

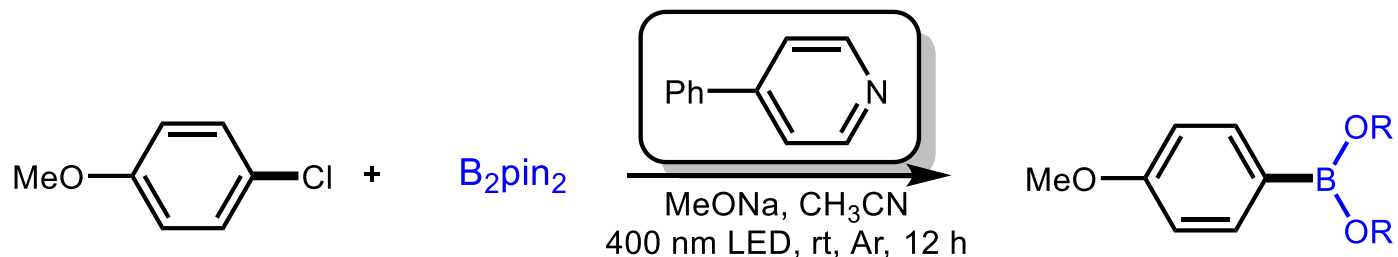


Complex 27 exhibits a superior reactivity!

Zhang L.; Jiao L. *J. Am. Chem. Soc.* **2019**, *141*, 9124

SEDs derived from diborons

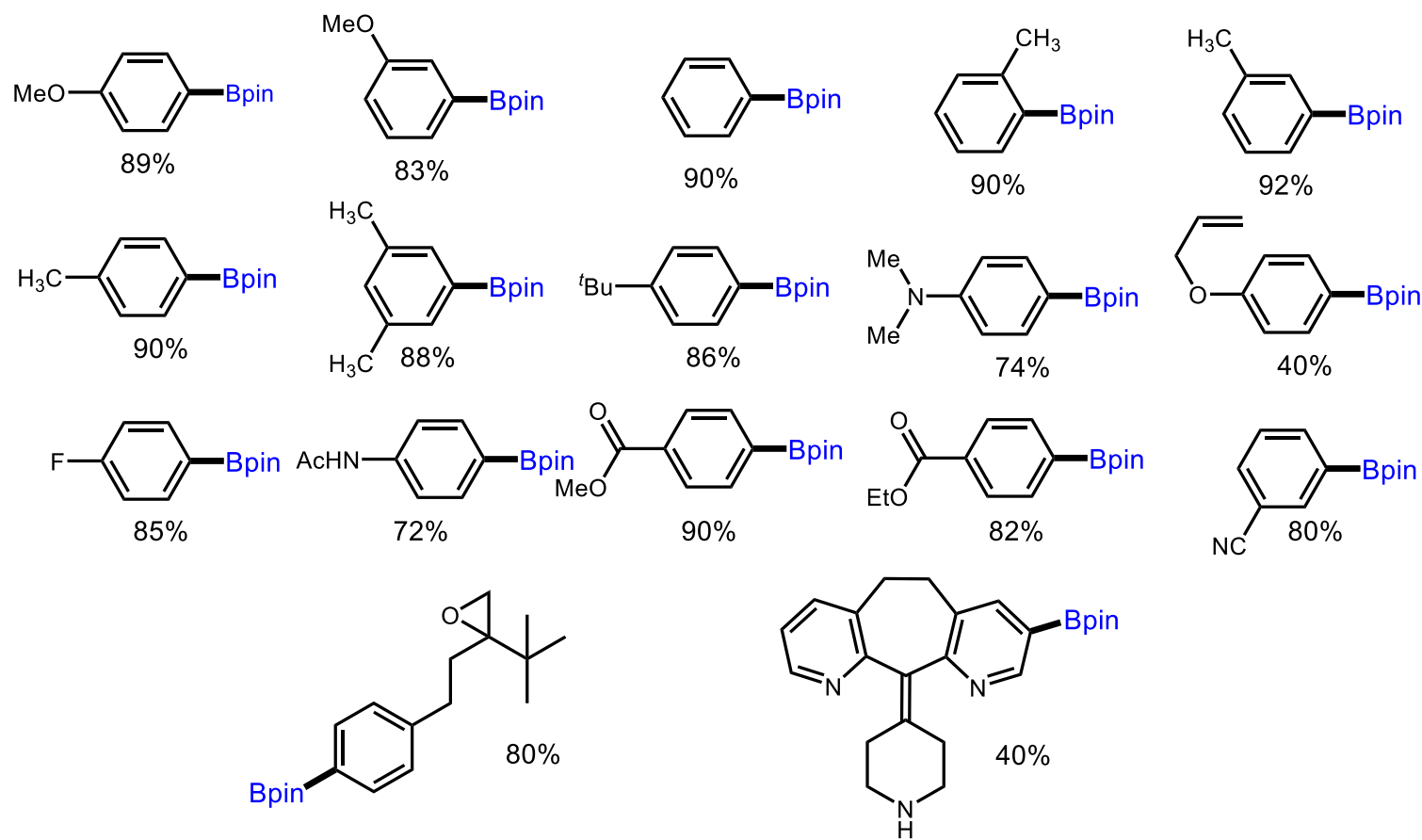
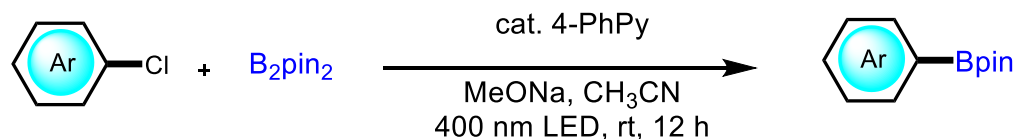
Optimization of Reaction Conditions



entry	change from standard conditions	conv. (%)	yield (%)
1	none	99	92
2	365 nm LED instead of 400 nm LED	40	35
3	450 nm LED instead of 400 nm LED	66	53
4	254 nm Hg lamp (28 W) instead of 400 nm LED	6	5
5	without light	1	1
6	without 4-phenylpyridine	1	1
7	reaction set up under air	94	88

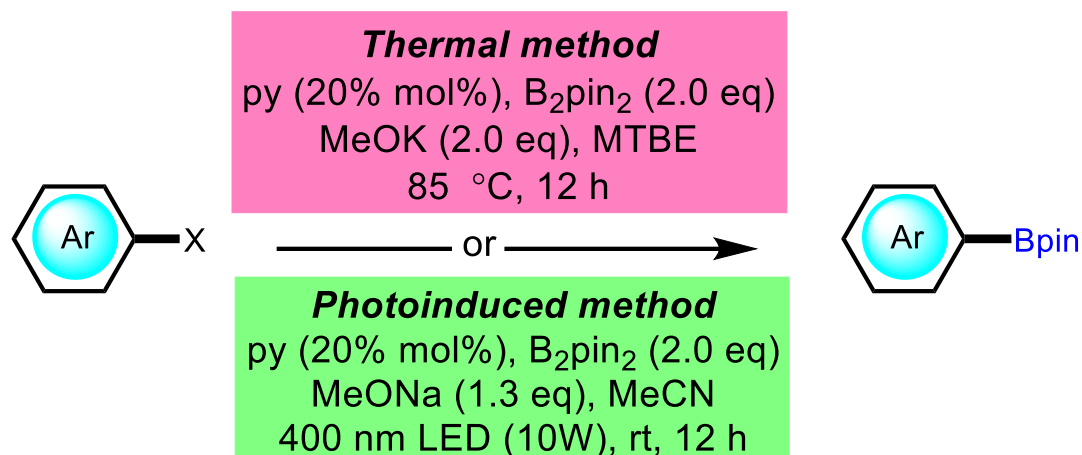
SEDs derived from diborons

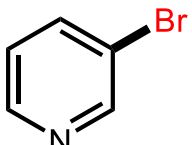
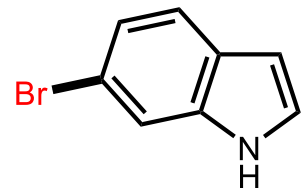
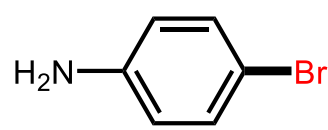
Borylation of Aryl Chlorides



SEDs derived from diborons

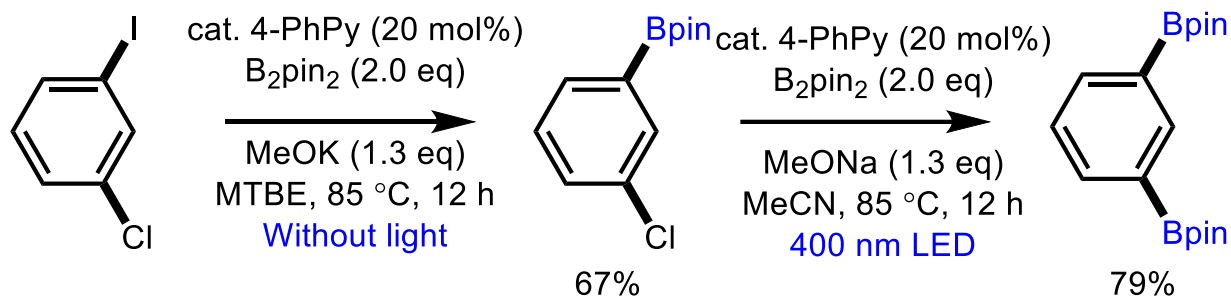
Enabling Difficult Borylation Reactions



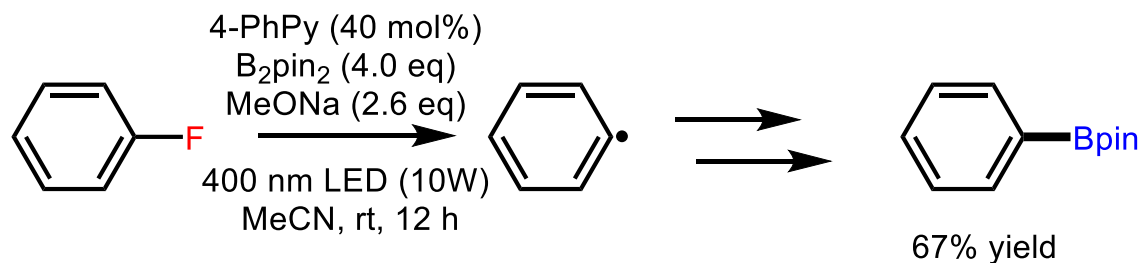
Aryl halide substrates			
<i>Thermal method</i>	10% yield	0% yield	0% yield
<i>Photoinduced method</i>	48% yield	56% yield	78% yield

SEDs derived from diborons

Application in sequential diborylation

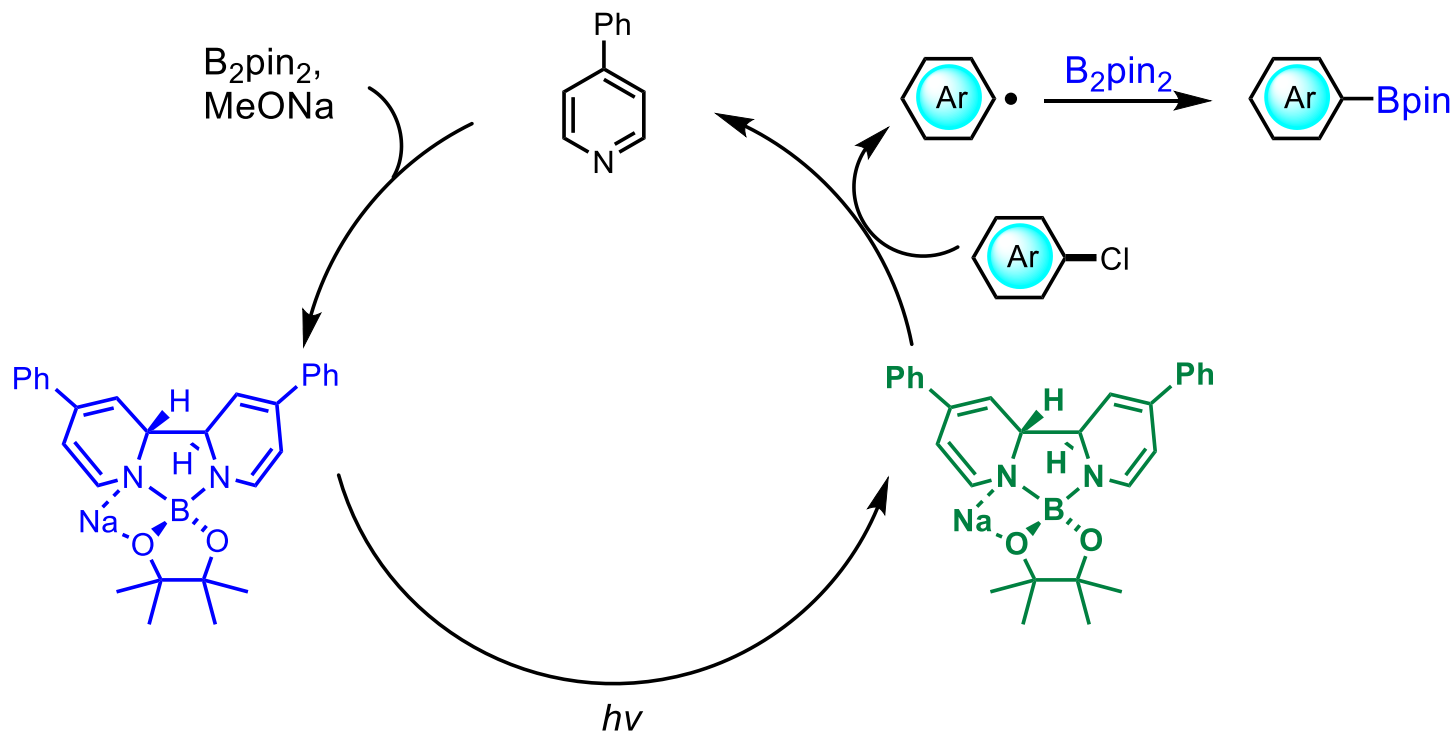


Borylation of Fluorobenzene



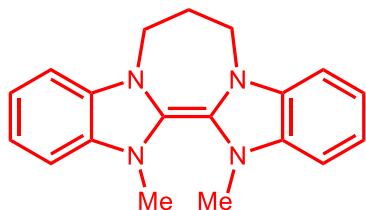
SEDs derived from diborons

Proposed mechanism



Zhang L.; Jiao L. *J. Am. Chem. Soc.* **2019**, *141*, 9124

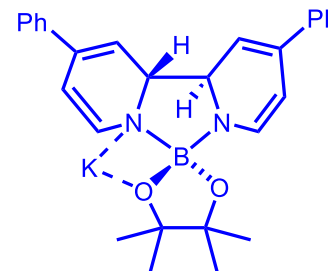
Summary



The electron-rich olefins

2005-2019

- ◆ Reactions in the ground state
- ◆ Reactions in the photo-excited state
- ◆ Haloarene–Arene Coupling



SEDs derived from diborons

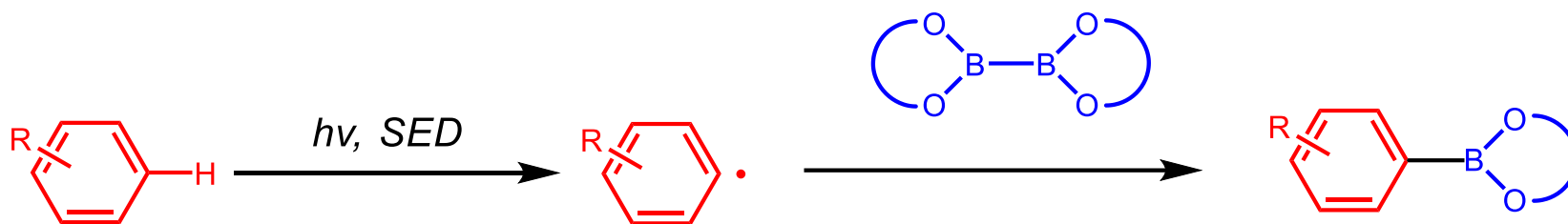
2018-2019

- ◆ Radical Borylation of Aryl Halides
- ◆ Desulfonative borylation of benzyl sulfones
- ◆ Visible-Light-Induced Borylation of Aryl Chlorides

Summary

Hypothesis

Photo-Induced Borylation of Aryl Chlorides



Thanks for your attention