



# Metal-enzyme catalyzes the dynamic kinetic resolution of alcohols and phenols

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**Student:** Kun Wang

**Supervisor:** Can Zhu

**Fudan University**

**2024.05.10**

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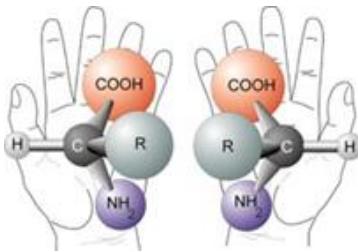
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# Part I: Introduction



Chirality



Medicine



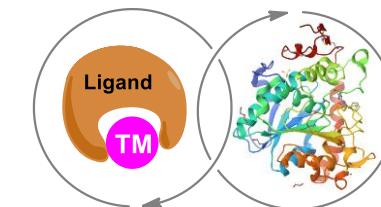
Pesticide



Material

Transition metal catalysis + Enzymatic catalysis → Advantages combination Metal-enzyme catalysis

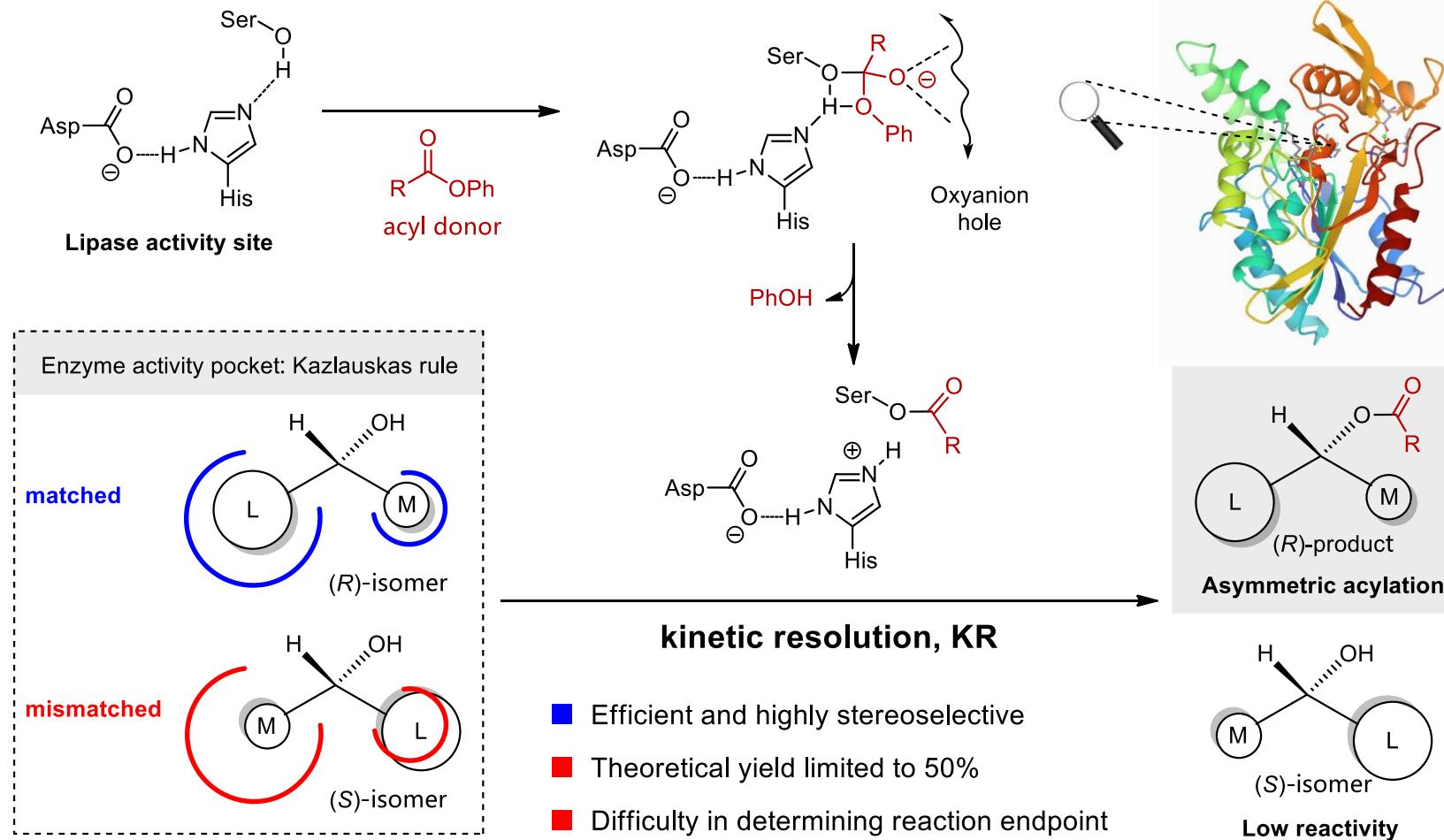
- Diverse types of metals
- Rich catalytic types
- Ligand regulation is variable
- High catalytic efficiency
- High stereoselectivity
- Mild reaction conditions



**Synthesize more challenging molecules**

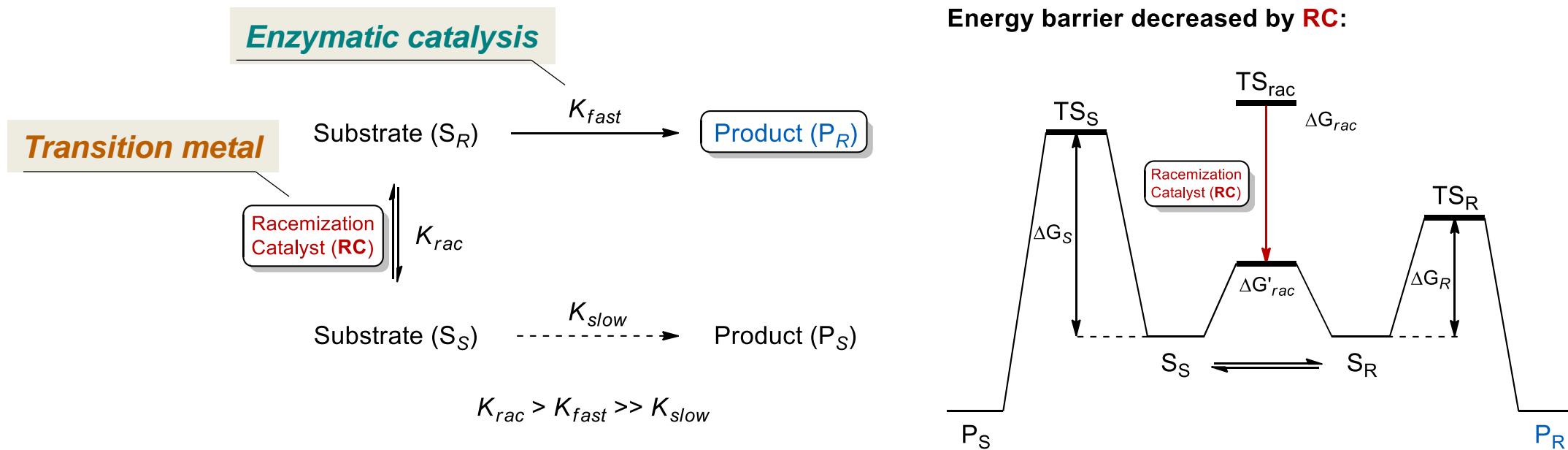
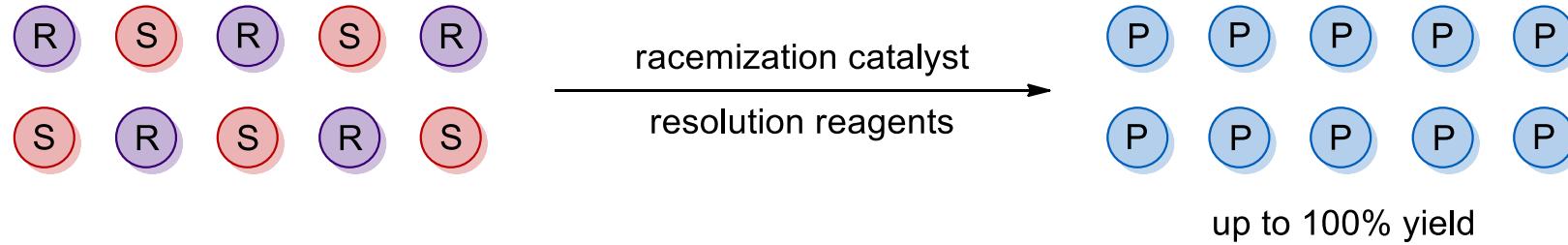
# Part I: Introduction

## Lipase catalyzed kinetic resolution of alcohols (kinetic resolution, KR)



# Part I: Introduction

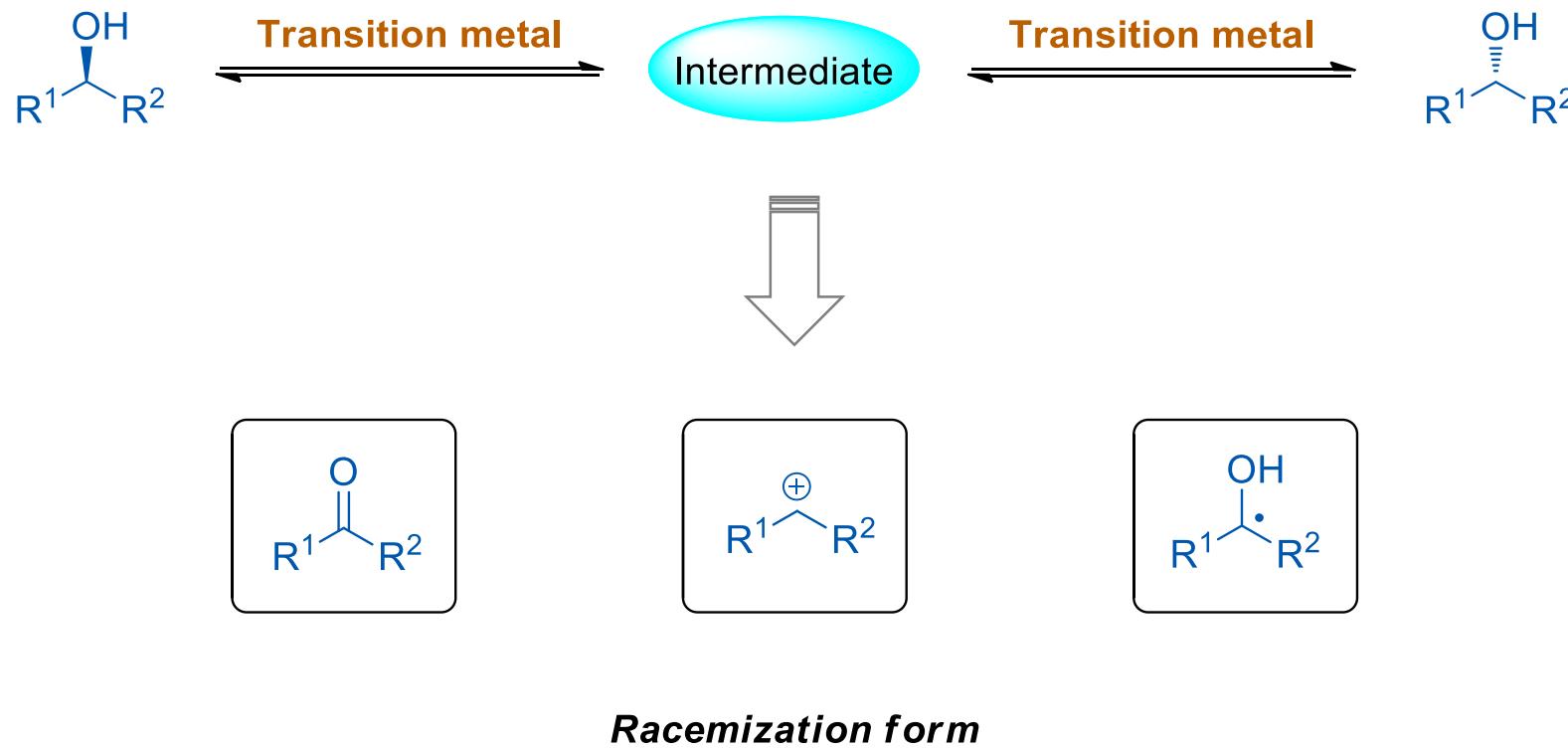
Chemoenzymatic Dynamic Kinetic Resolution (DKR):



- The theoretical yield is 100%;
- High reaction efficiency

# Part I: Introduction

**Transition metal catalyzed racemization of alcohols:**



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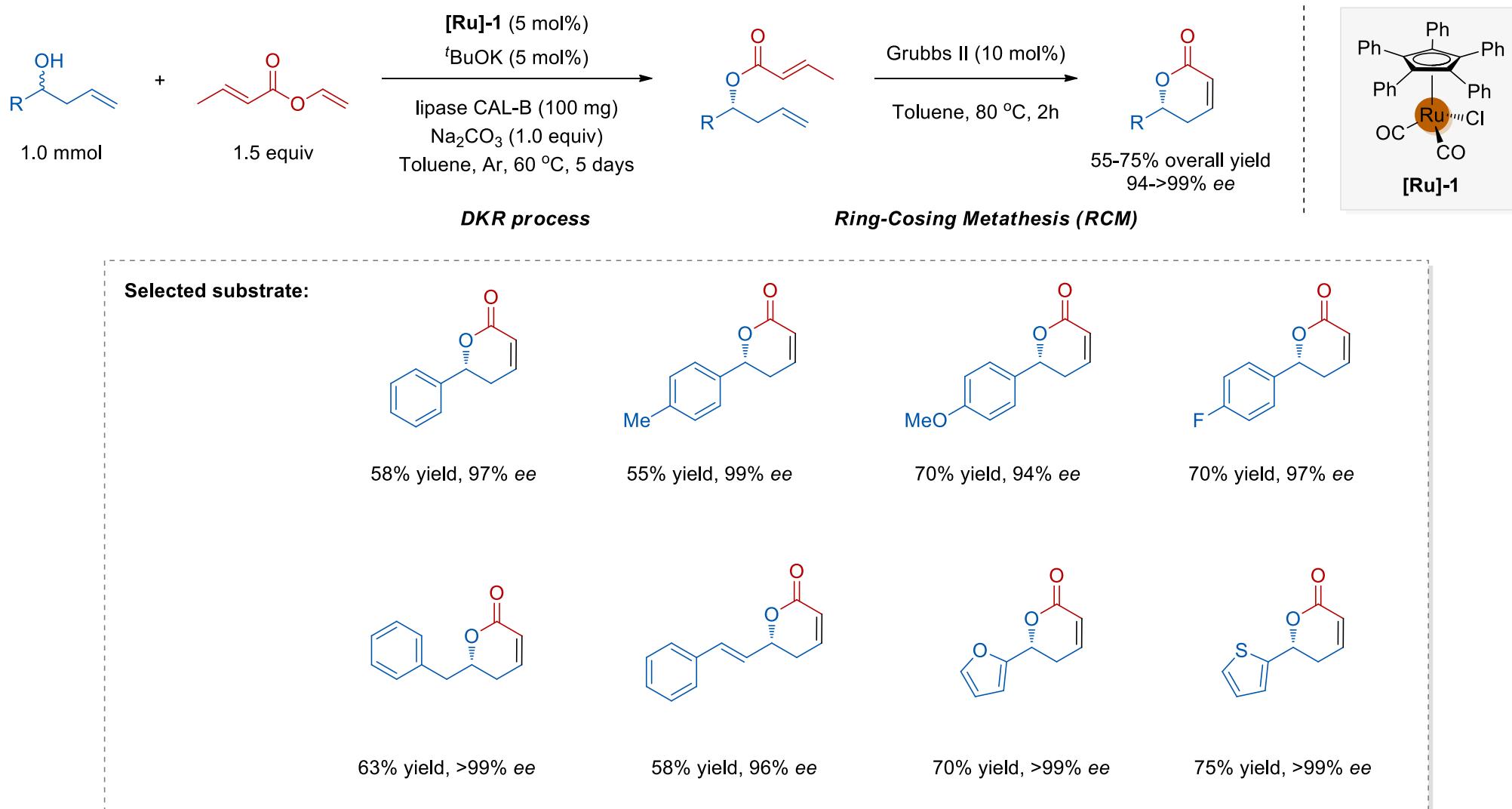
Racemization via radical intermediates

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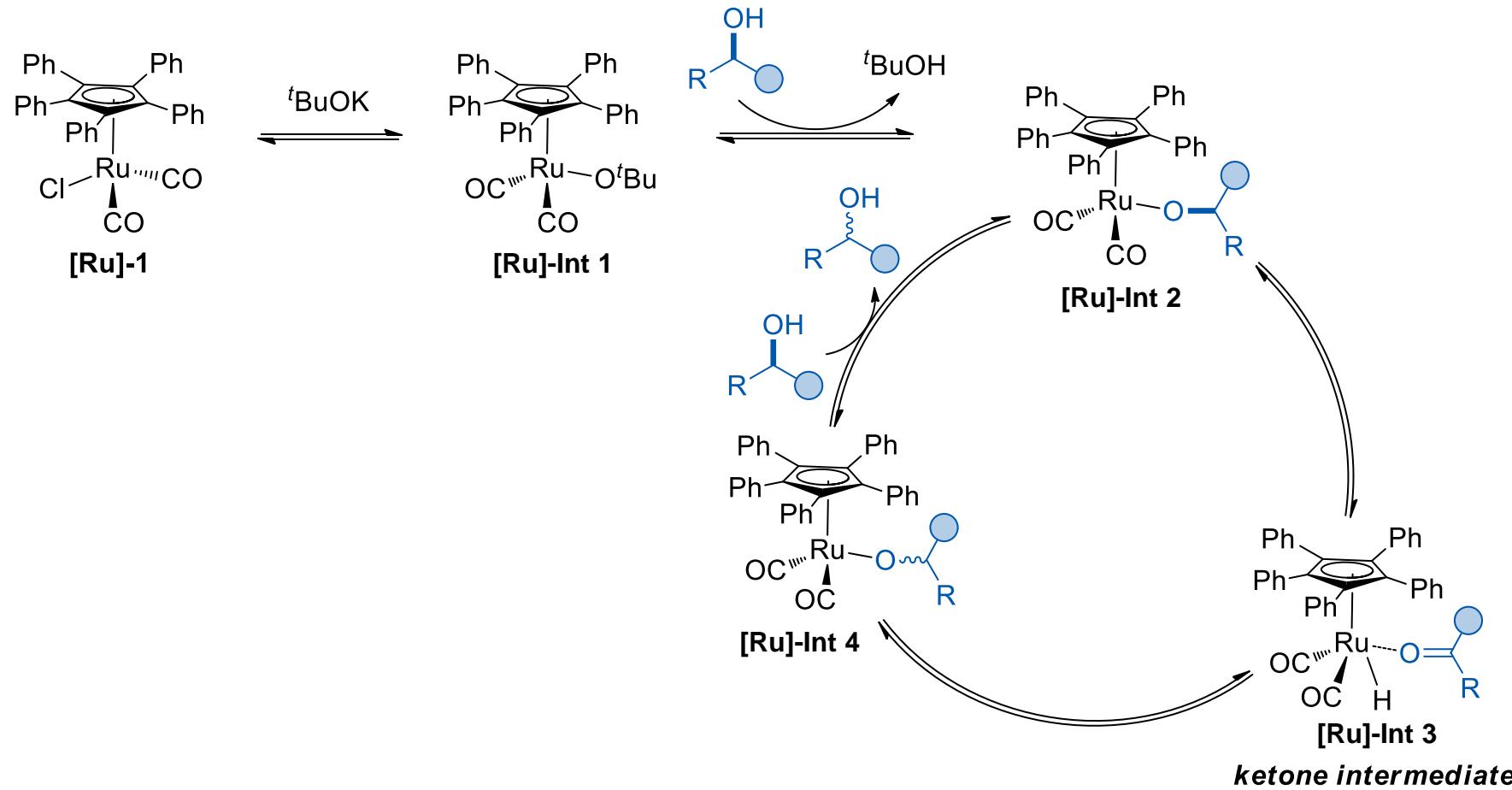
## Part II: Redox racemization via ketone intermediates: *Ru* catalysts

2019, Ryszard Ostaszewski group:



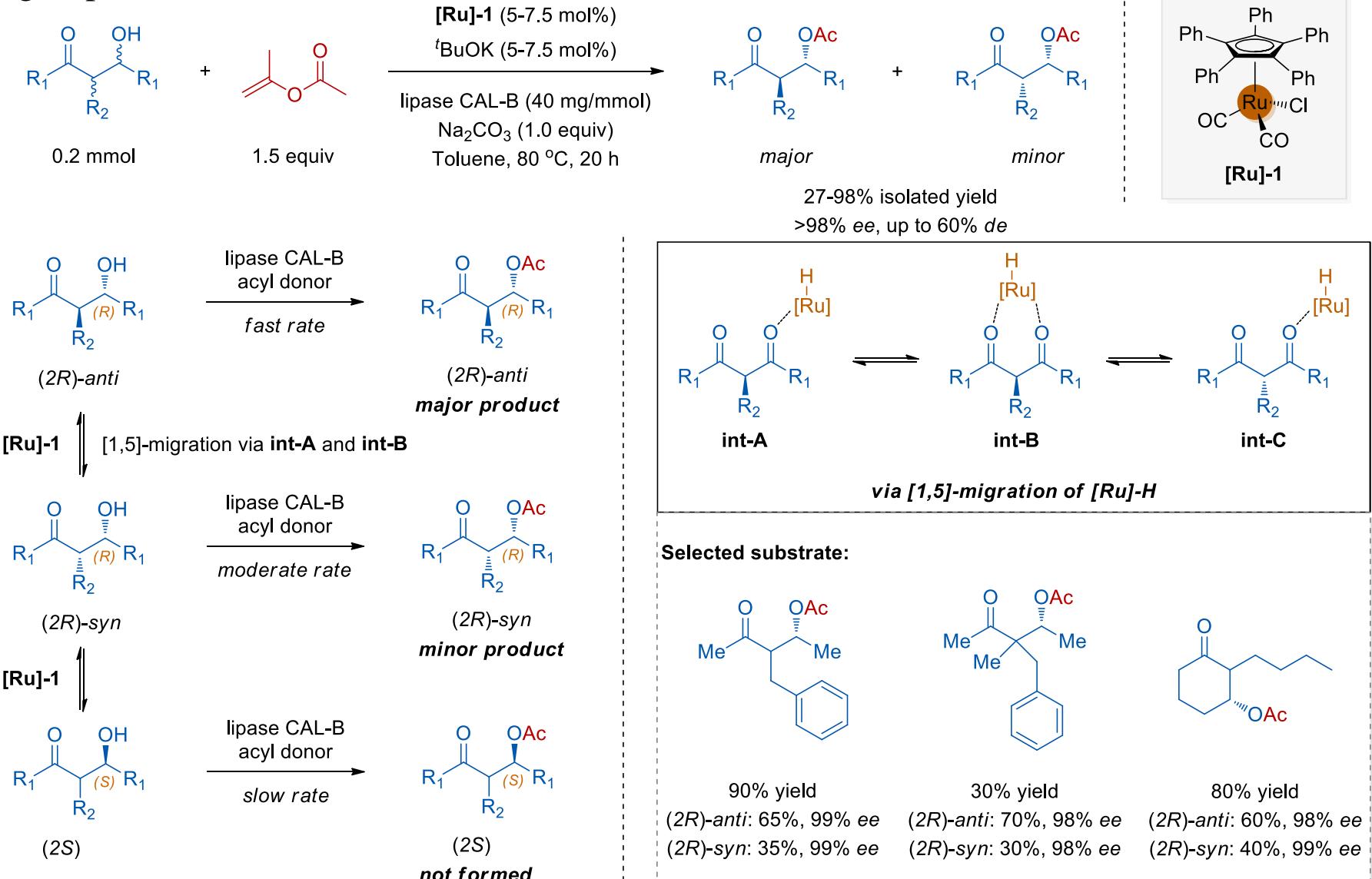
## Part II: Redox racemization via ketone intermediates: *Ru catalysts*

*Proposed mechanism of the racemization process:*

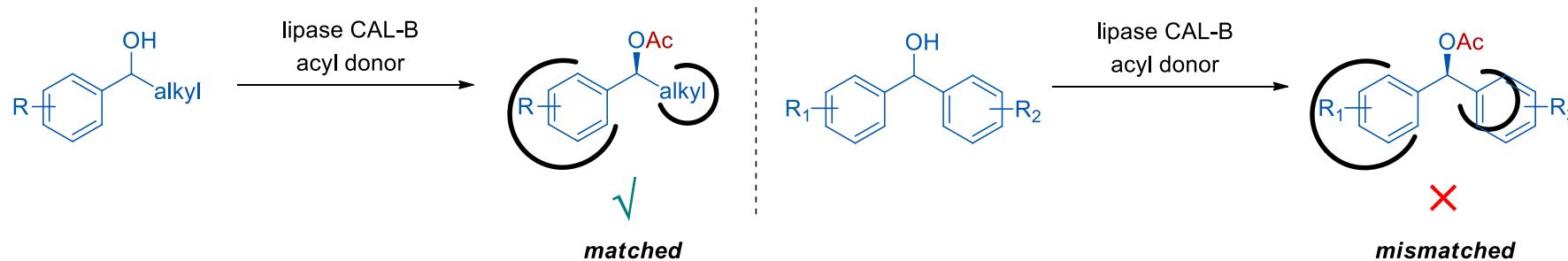


## Part II: Redox racemization via ketone intermediates: *Ru* catalysts

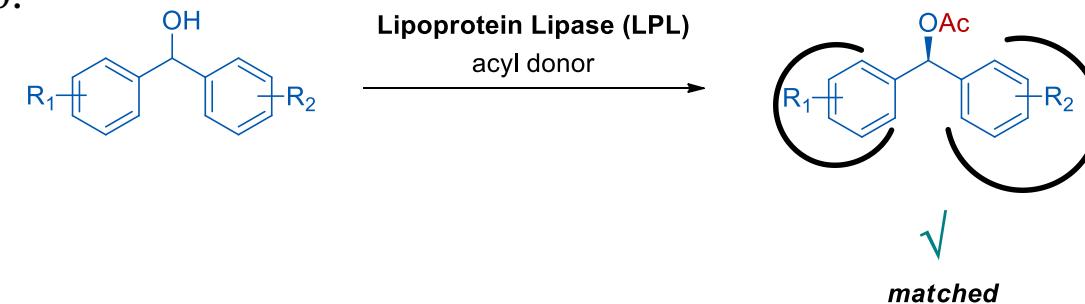
2021, Bäckvall group:



## Part II: Redox racemization via ketone intermediates: *Ru* catalysts



2015, J. W. Park and M. J. Kim group:

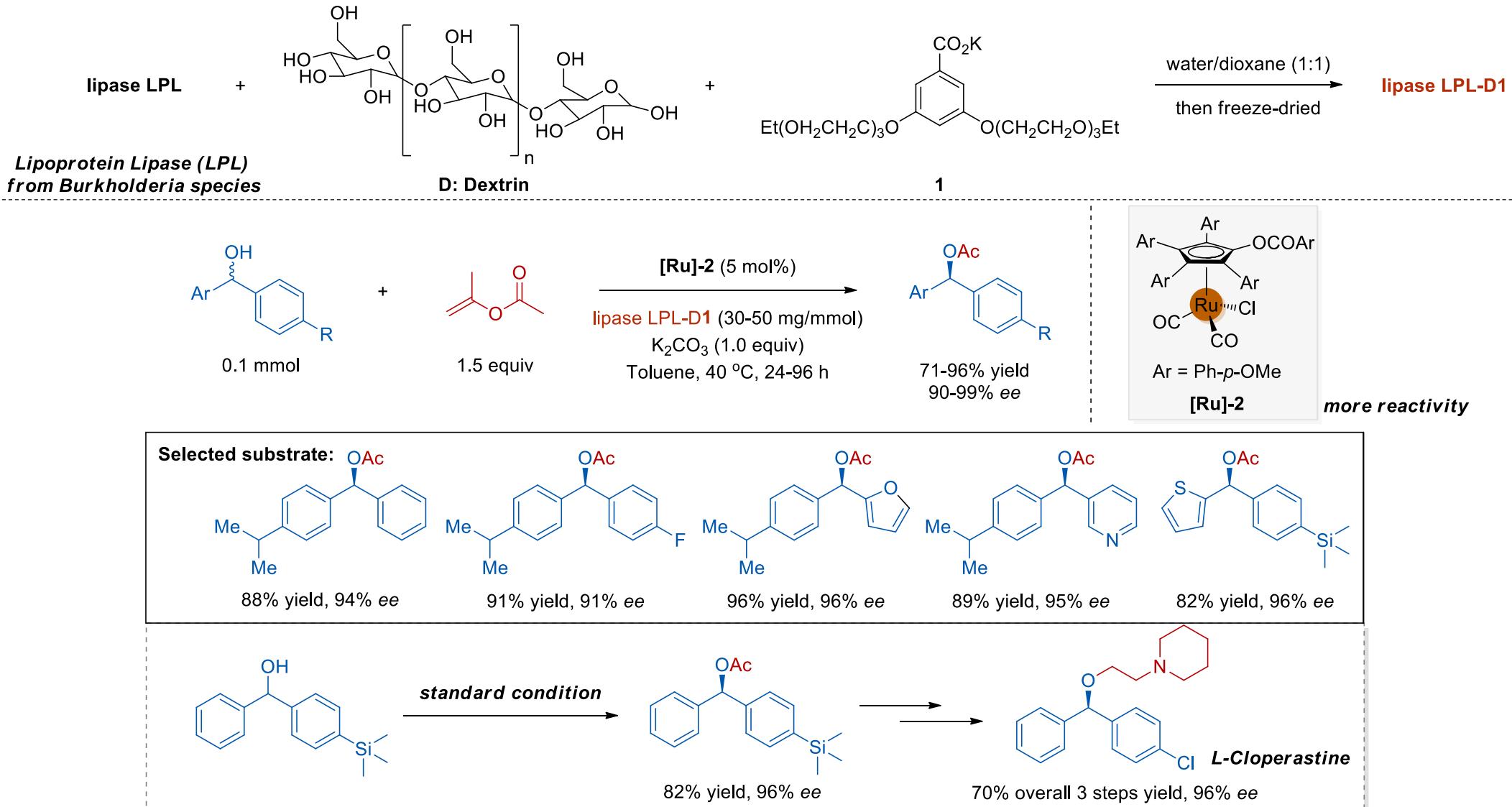


entry	lipase	ROH	solvent	$k_{cat}$ (s <sup>-1</sup> )
1	LPL	H <sub>2</sub> O	phosphate buffer	$2.8 \times 10^2$
2	LPL	PhCH(CH <sub>3</sub> )OH	toluene	$1.7 \times 10^{-2}$
3	LPL-D1	H <sub>2</sub> O	phosphate buffer	$5.7 \times 10^2$
4	LPL-D1	PhCH(CH <sub>3</sub> )OH	toluene	$5.5 \times 10^4$

About 3000 times  
compared to entry 2

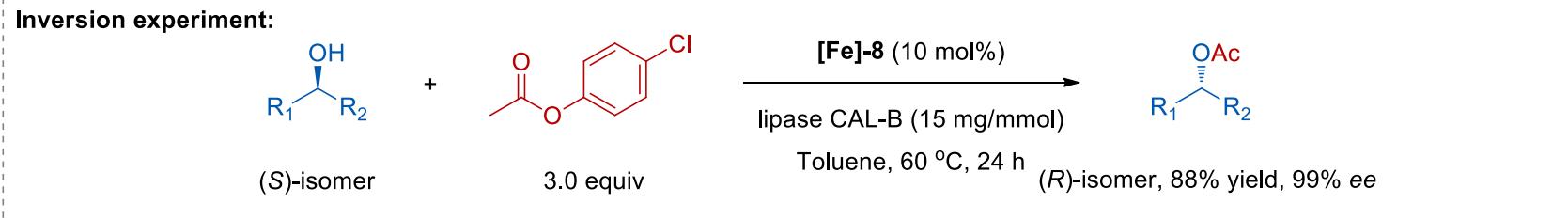
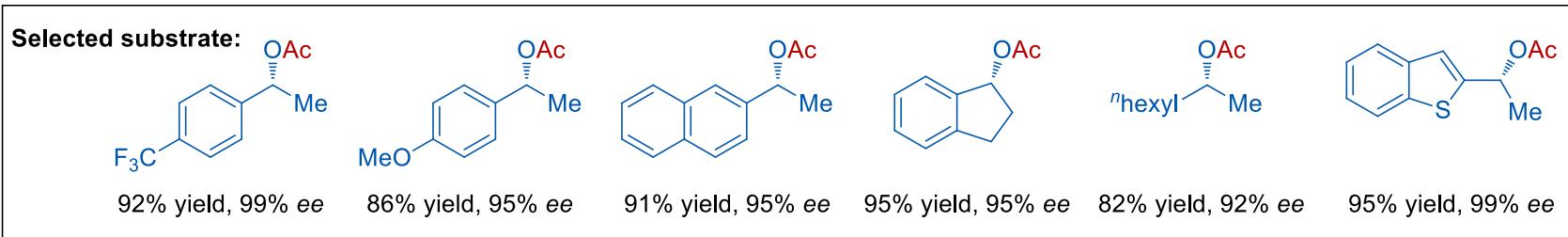
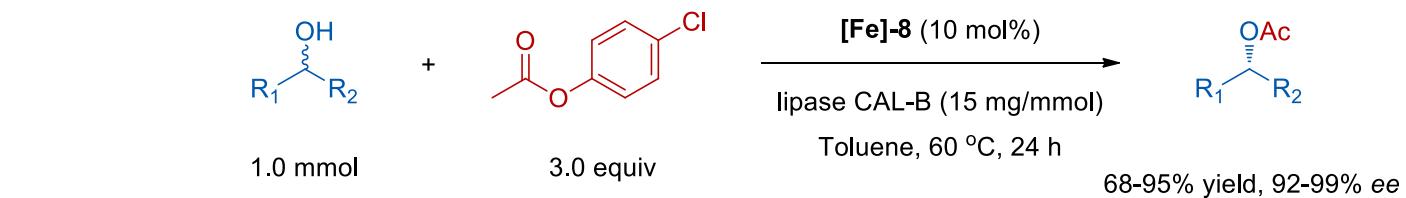
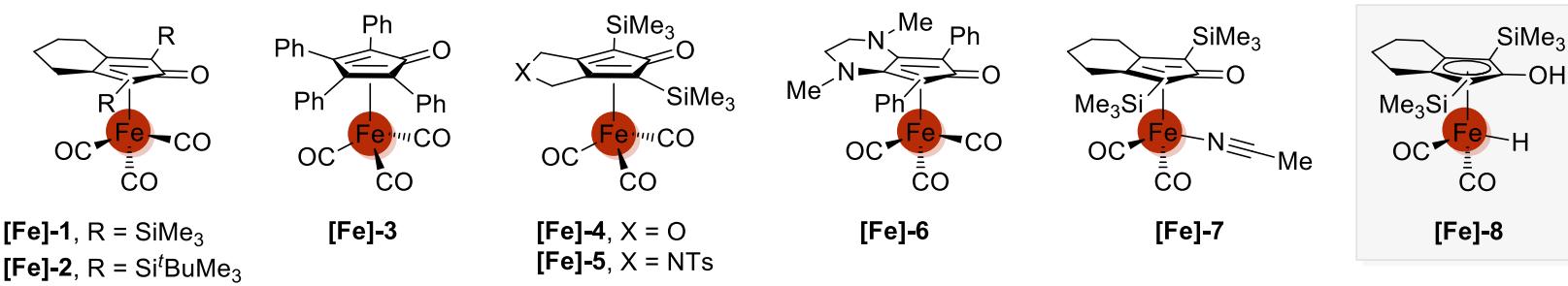
## Part II: Redox racemization via ketone intermediates: *Ru* catalysts

### Preparation of lipase LPL-D1:



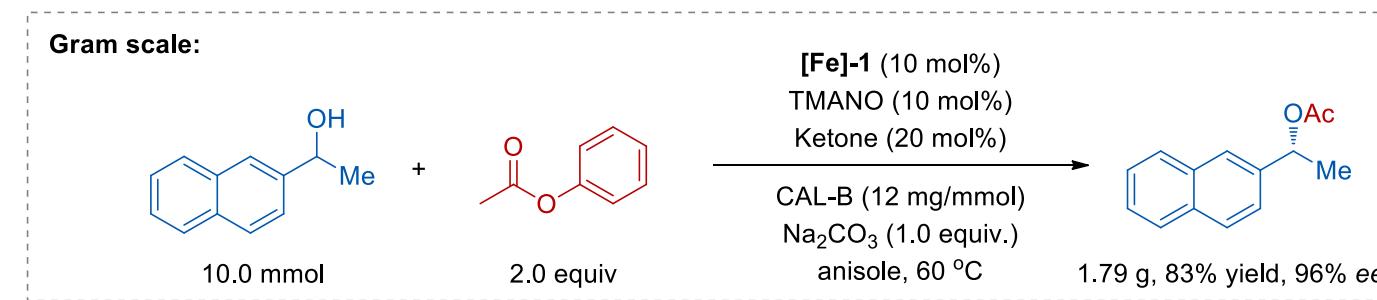
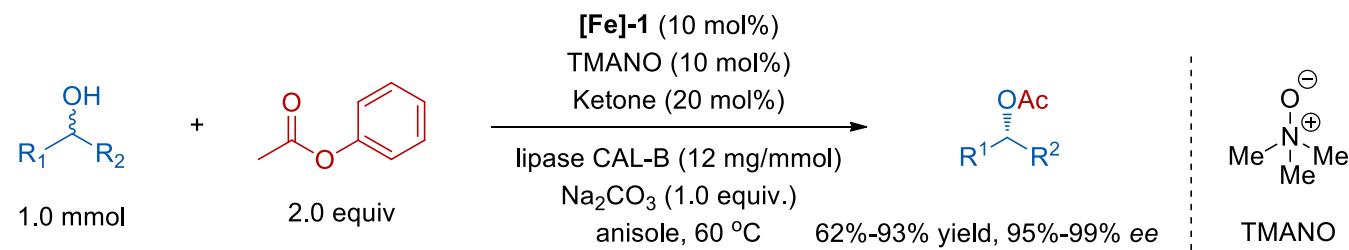
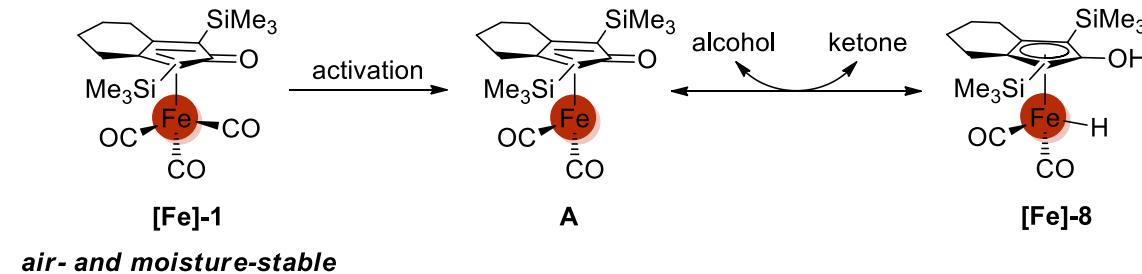
## Part II: Redox racemization via ketone intermediates: *Fe catalysts*

2016, Magnus Rueping group:



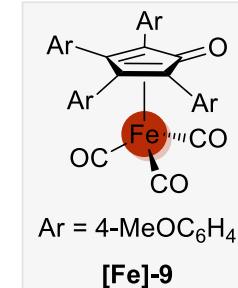
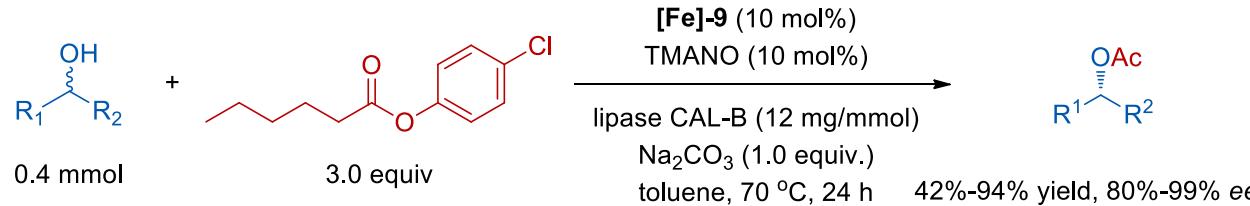
## Part II: Redox racemization via ketone intermediates: *Fe catalysts*

2017, B äckvall group:



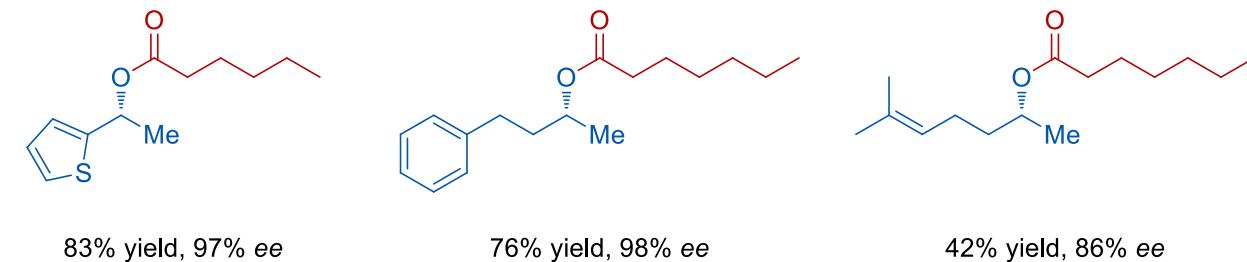
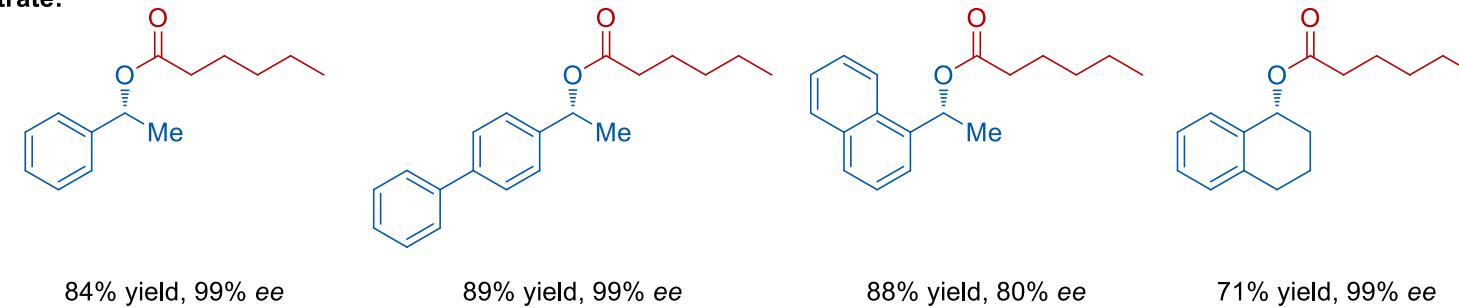
## Part II: Redox racemization via ketone intermediates: *Fe catalysts*

2017, Zhou Shaolin group:



*new air-stable iron catalyst*

**Selected substrate:**



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Racemization via *carbocation* intermediates

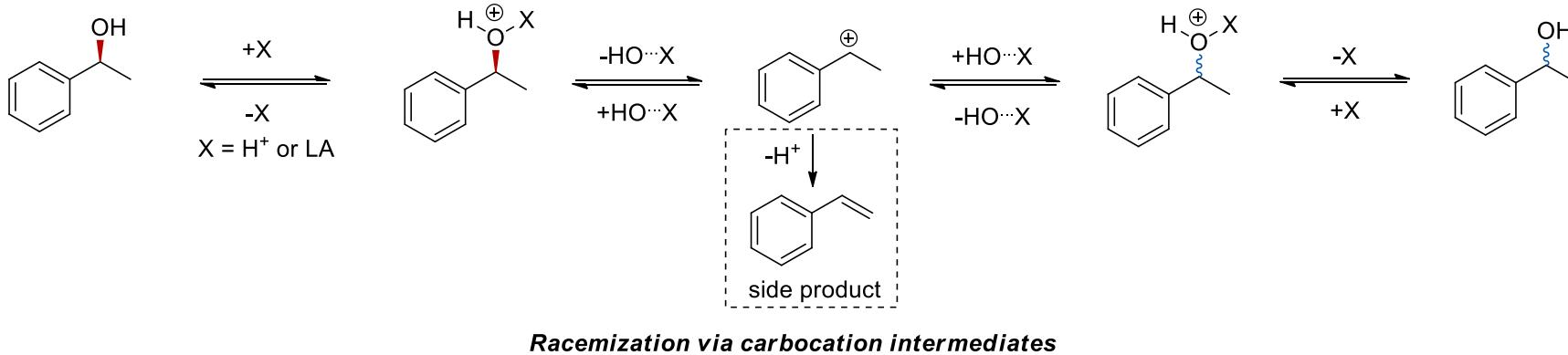
04

Racemization via *radical* intermediates

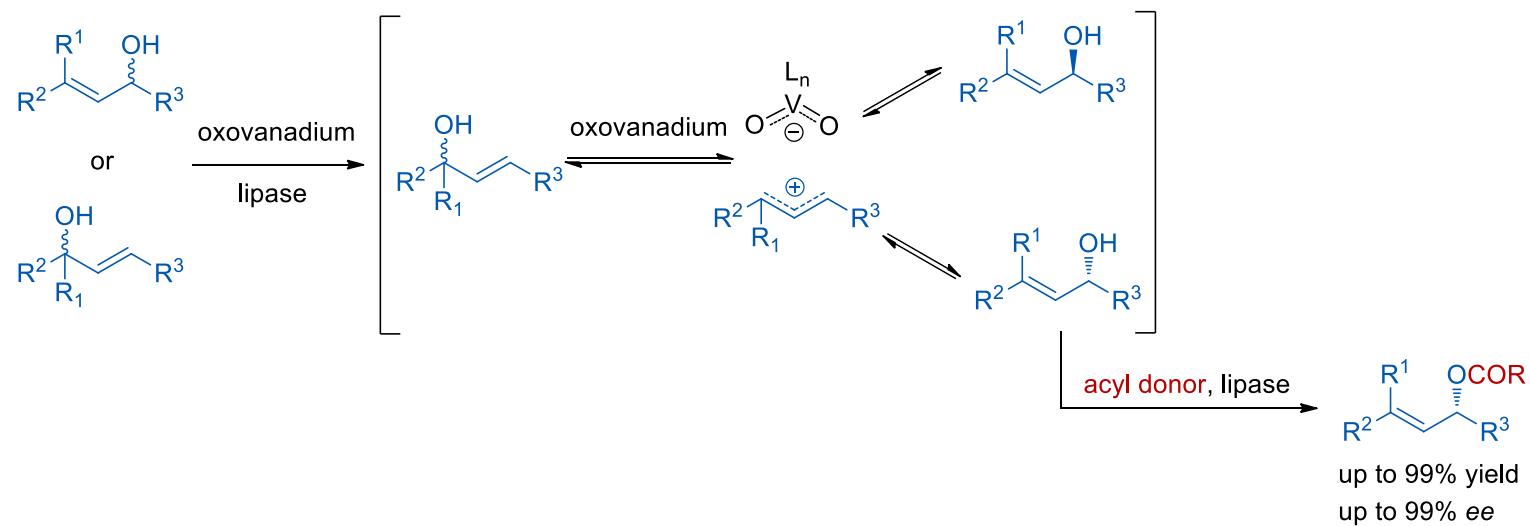
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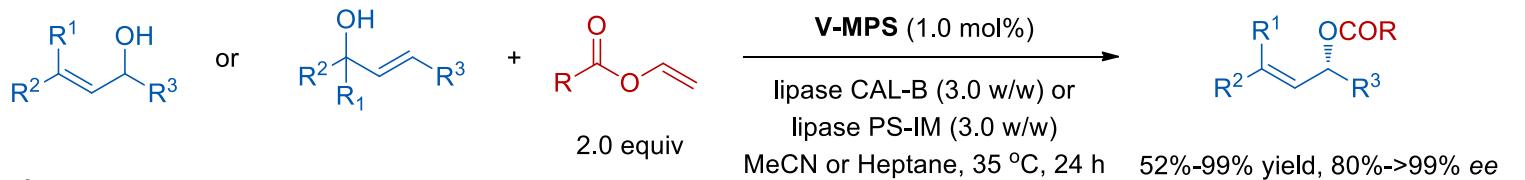
## Part III: Racemization via carbocation intermediates: V catalysts



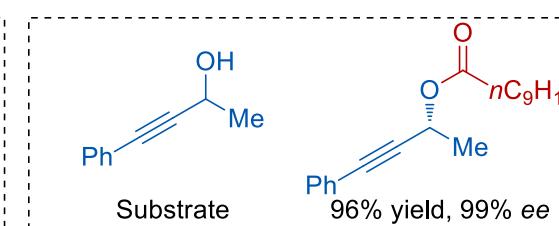
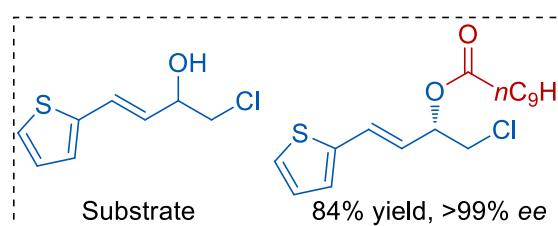
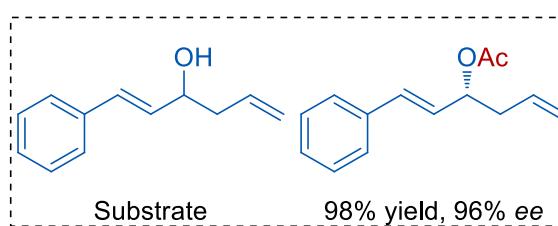
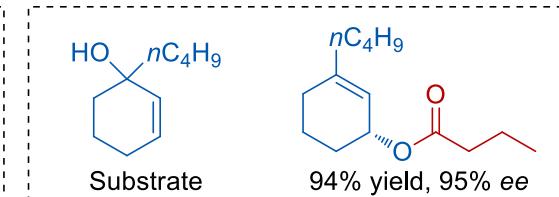
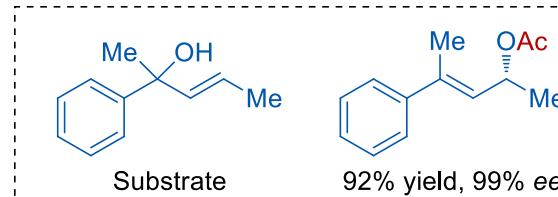
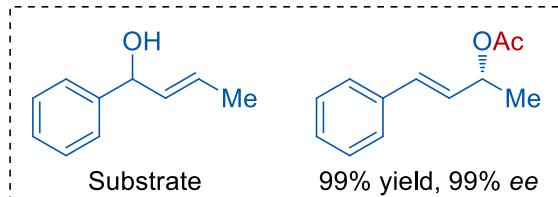
2013, Akai group:



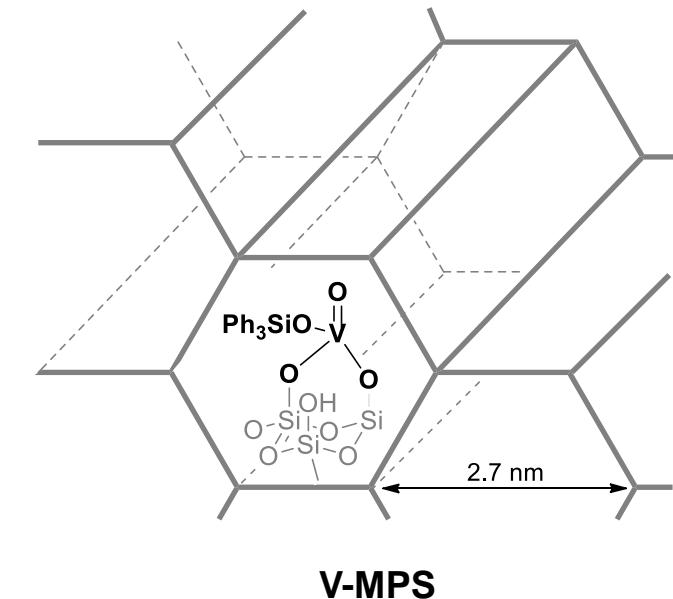
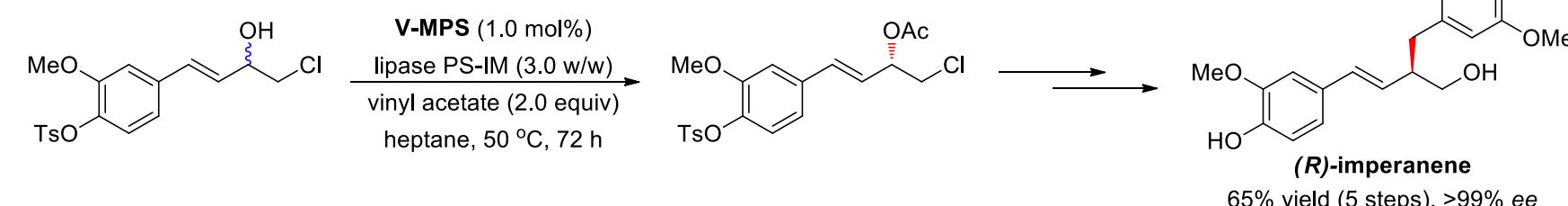
# Part III: Racemization via carbocation intermediates: V catalysts



Selected substrate:



Application of DKR in the synthesis of (*R*)-imperanene

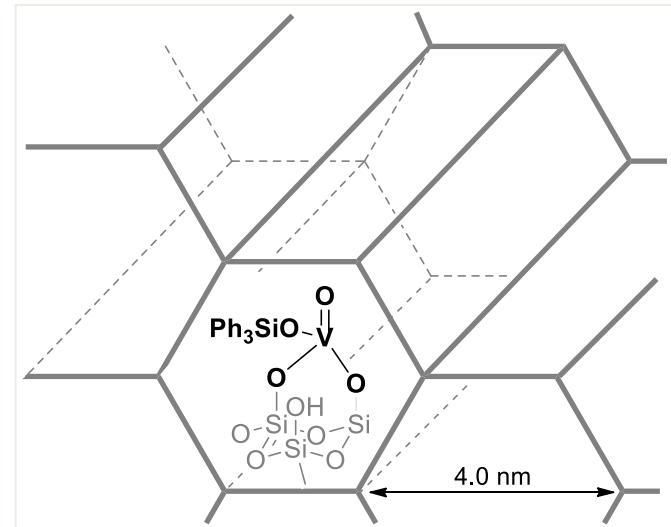
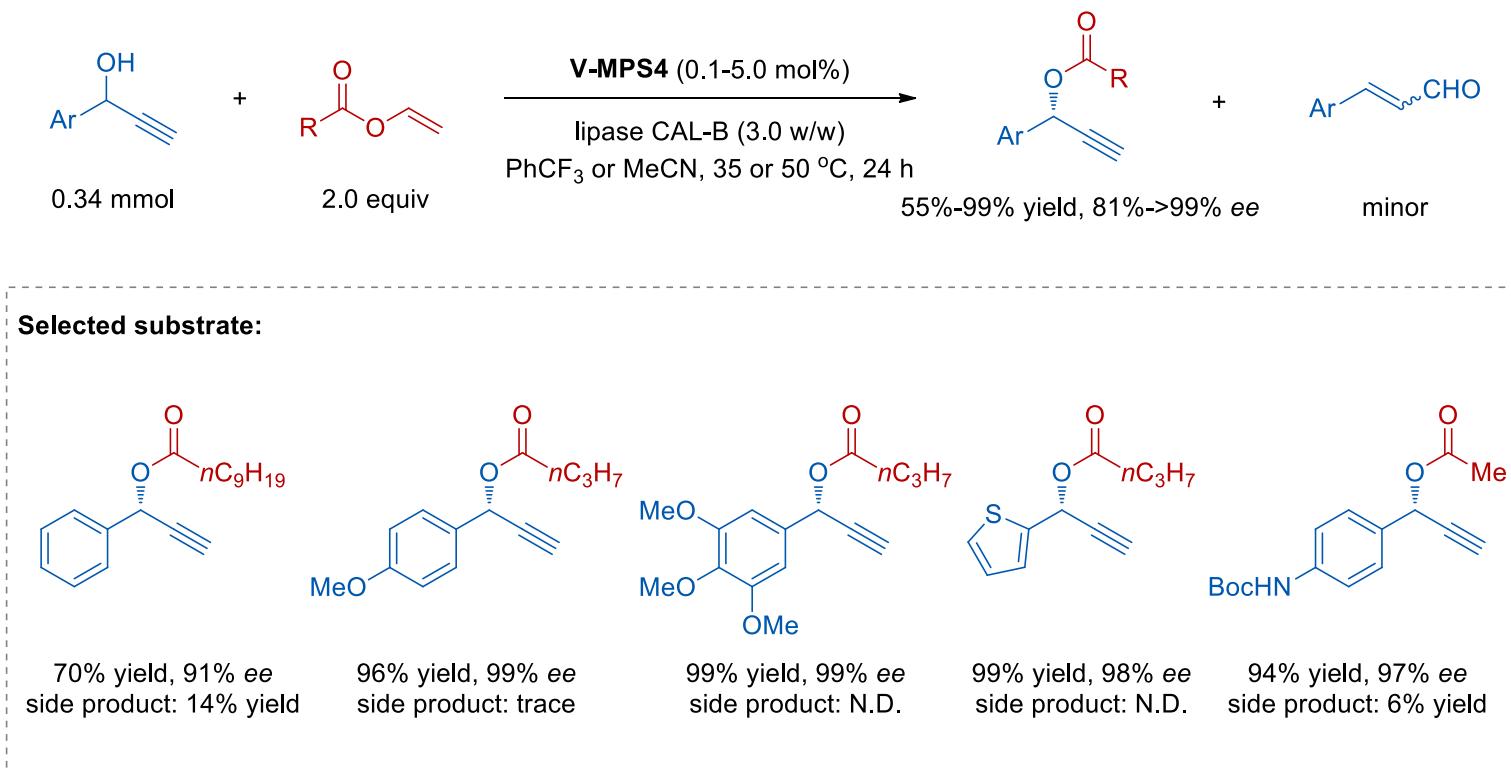


1st run:	99%, 99% ee
2nd run:	100%, 99% ee
3rd run:	99%, 99% ee
4th run:	100%, 99% ee
5th run:	99%, 99% ee
6th run:	99%, 99% ee
7th run:	85%, 99% ee

For standard substrate, V-MPS can be cycled at least 7 times.

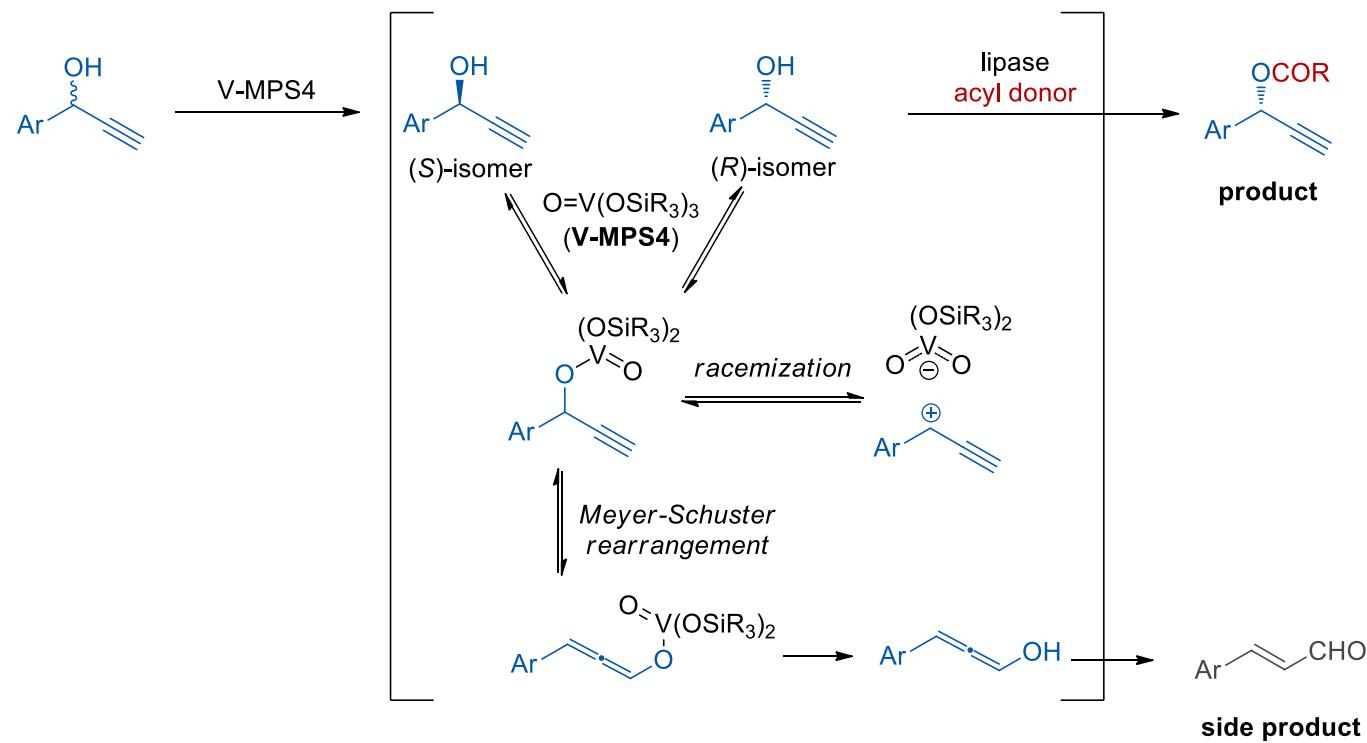
## Part III: Racemization via carbocation intermediates: V catalysts

2019, Akai group:



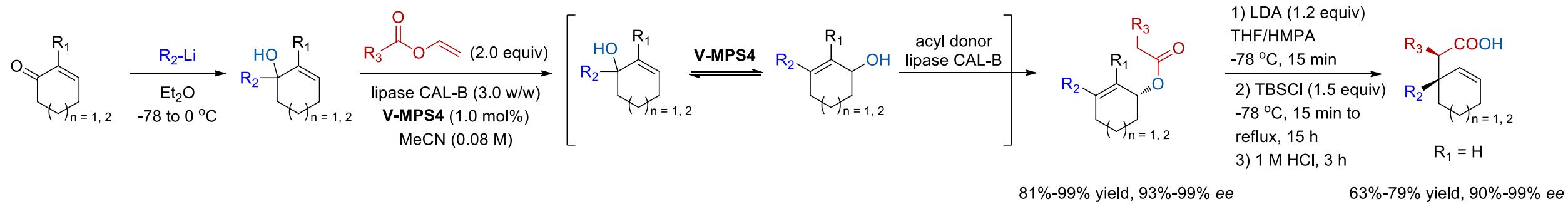
## Part III: Racemization via carbocation intermediates: V catalysts

*Proposed mechanism:*

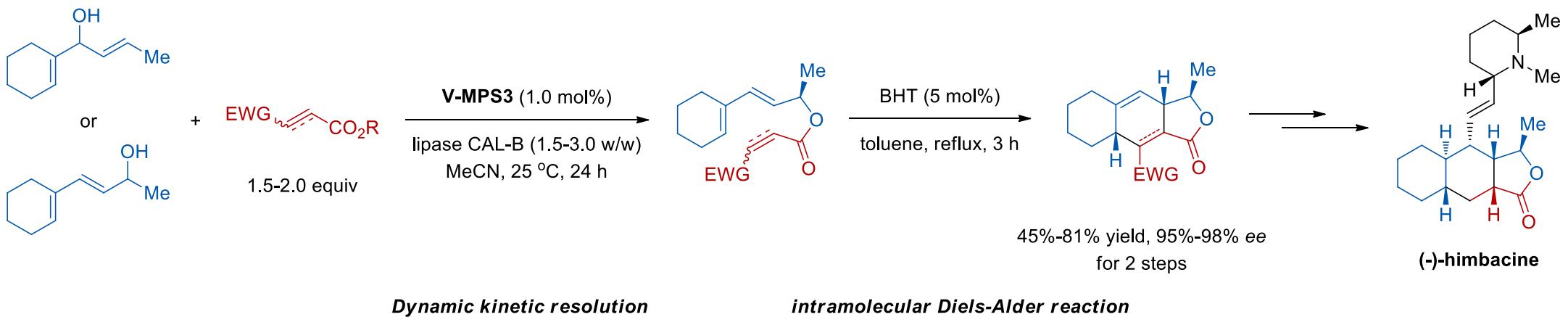


## Part III: Racemization via carbocation intermediates: V catalysts

2017, Akai group:

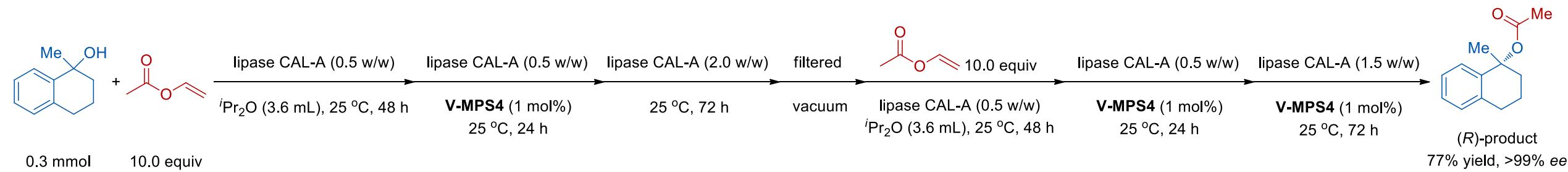


2018, Akai group:

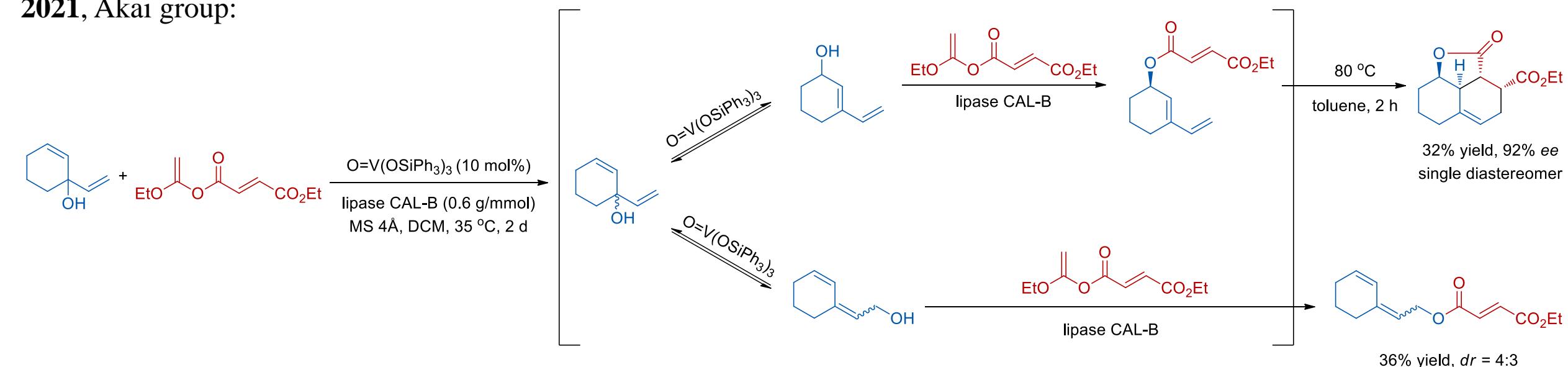


## Part III: Racemization via carbocation intermediates: V catalysts

2020, Akai group:



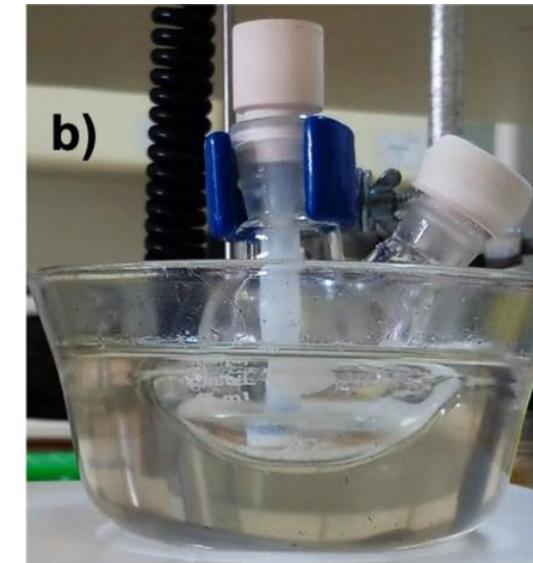
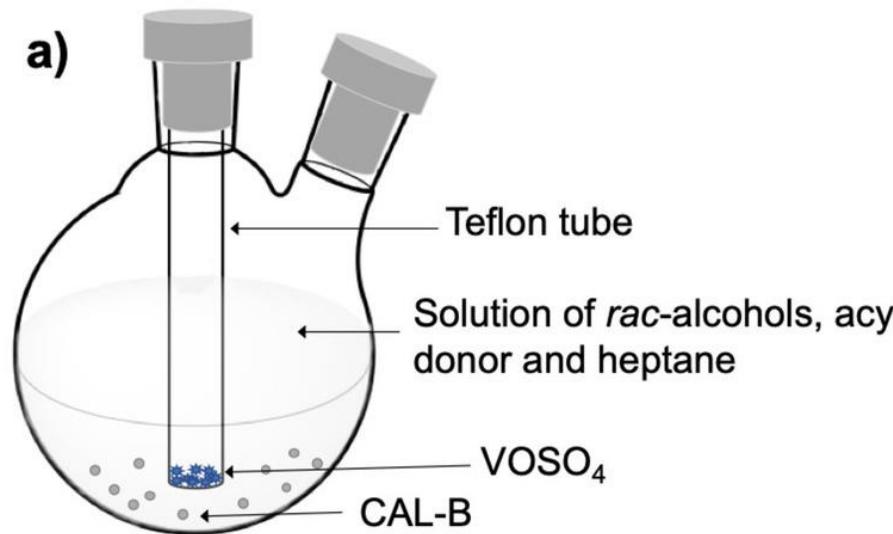
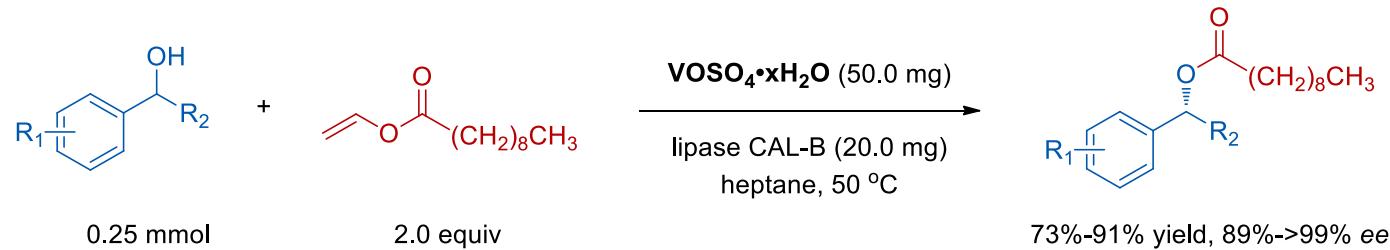
2021, Akai group:



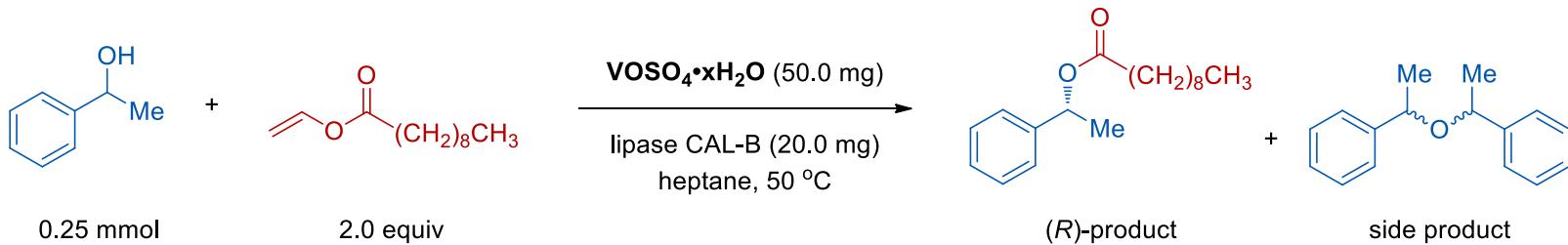
1) S. Akai *et al.*, *Chem. Commun.* **2020**, 56, 2885–2888; 2) S. Akai *et al.*, *Synlett.* **2021**, 32, 822–828

## Part III: Racemization via carbocation intermediates: V catalysts

2020, Milagre group:

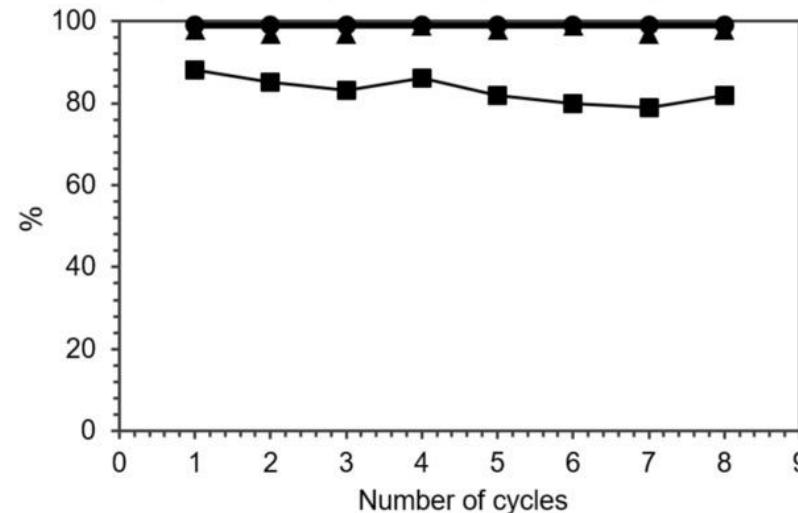


# Part III: Racemization via carbocation intermediates: V catalysts



entry	condition	time [h]	c [%]	ee [%]	Sel [%]
1	Without tube	2	82	>99	92
2	CAL-B into the tube	2	20	>99	>99
3	VOSO <sub>4</sub> ·xH <sub>2</sub> O into the tube	1	91	>99	98
4	VOSO <sub>4</sub> ·xH <sub>2</sub> O into the tube 5.0 mmol scale	2	90	>99	98

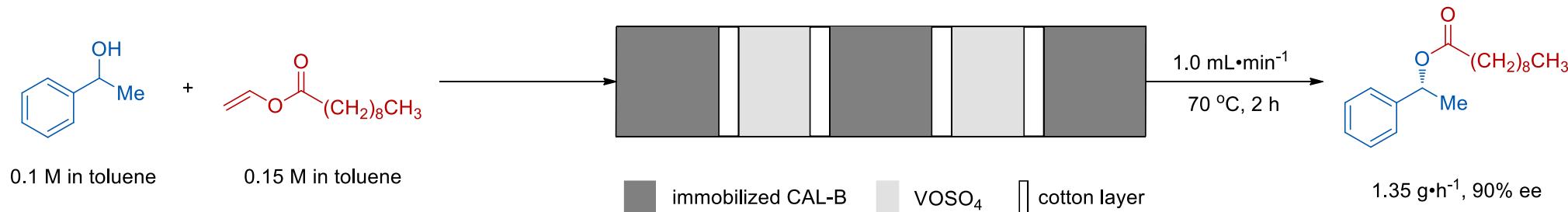
**Recycling of CAL-B and VOSO<sub>4</sub> ·xH<sub>2</sub>O experiment**



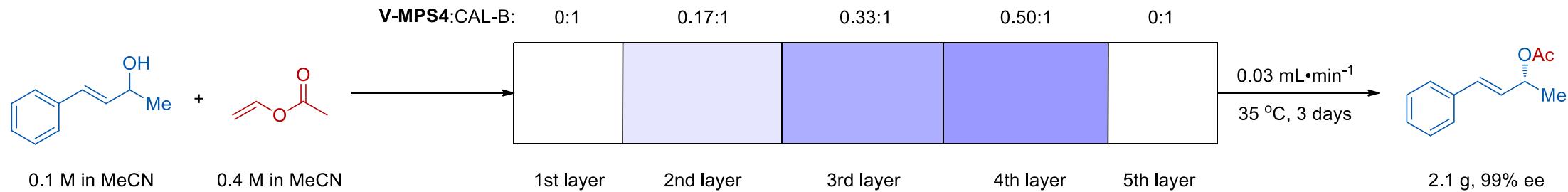
- ee of the (R)-product at the end of each cycle;
- conversion;
- selectivity.

## Part III: Racemization via carbocation intermediates: V catalysts

2017, Souza group:



2020, Akai group:



1) Rodrigo O. M. A. de Souza *et al.*, *React. Chem. Eng.* **2017**, 2, 375–381; 2) S. Akai *et al.*, *Eur. J. Org. Chem.* **2020**, 1961–1967

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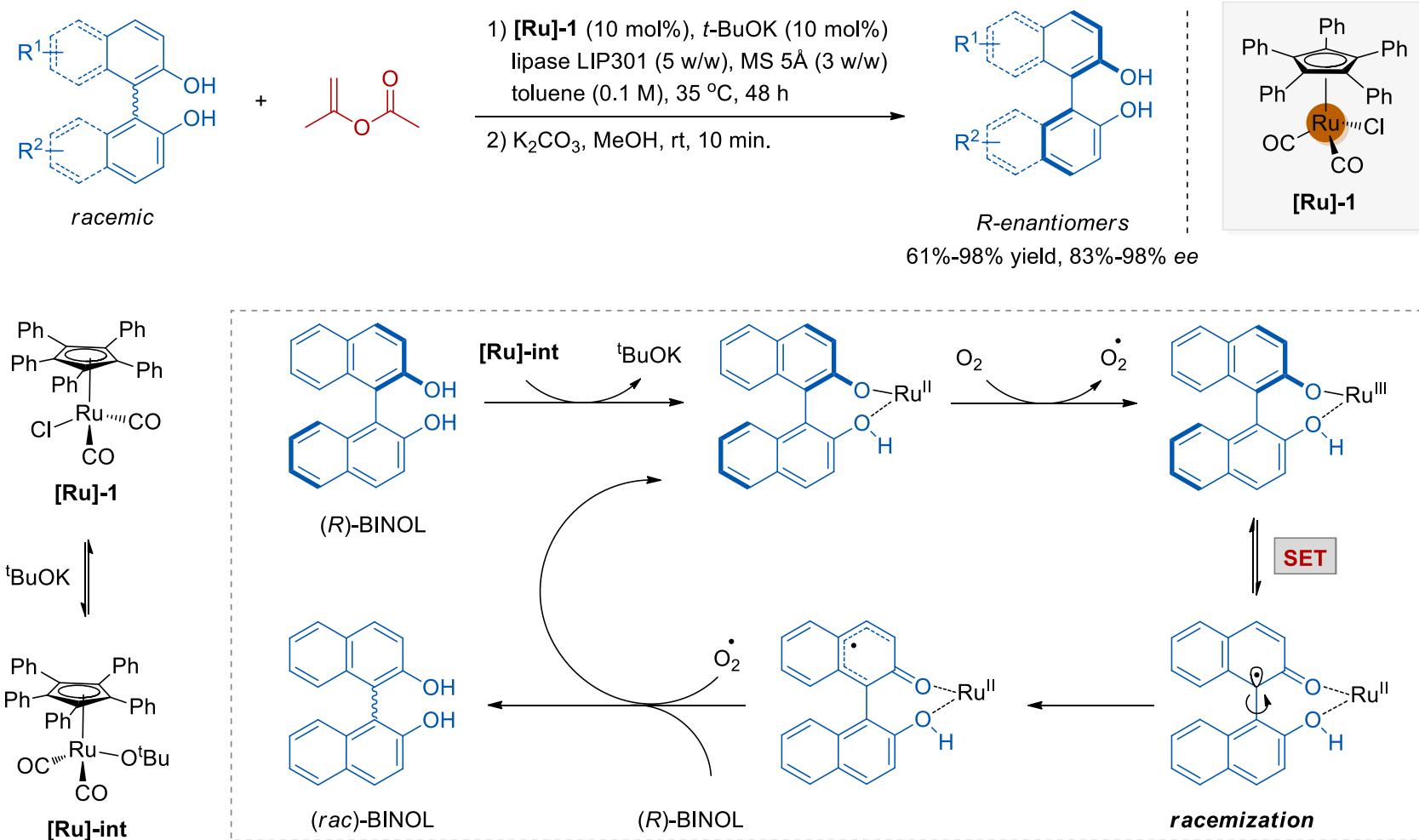
Racemization via *radical* intermediates

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Summary and Outlook

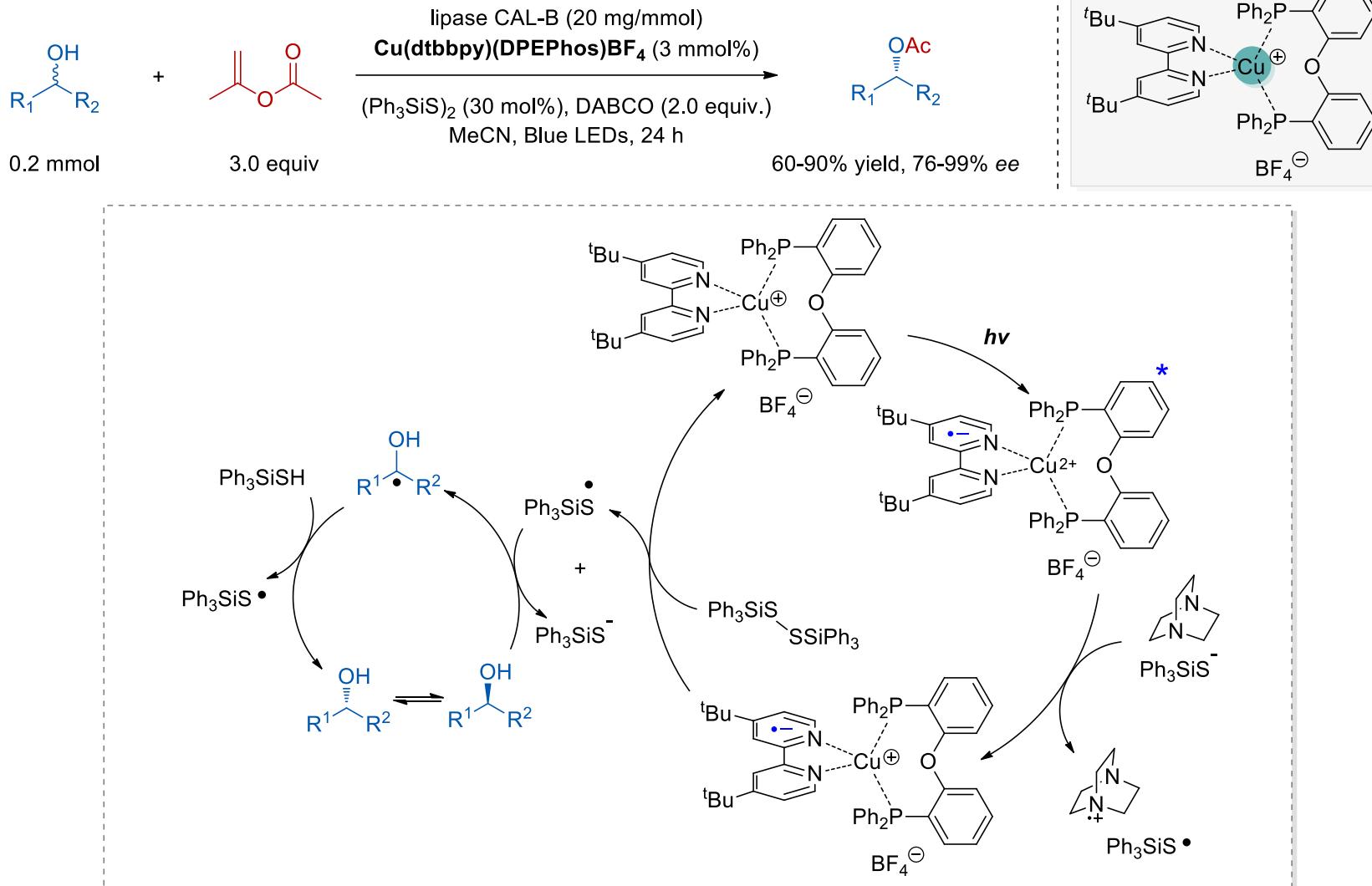
# Part IV: Racemization via radical intermediates: *Ru catalysts*

2018, Shuji Akai group:



# Part IV: Racemization via radical intermediates: Cu catalysts

2023, Collins group:



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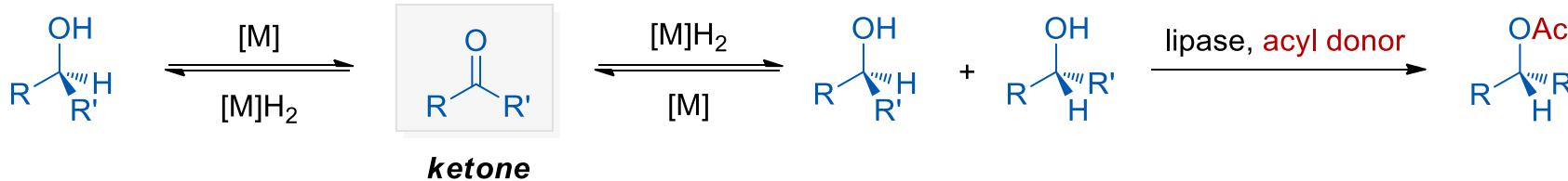
Racemization via *radical* intermediates

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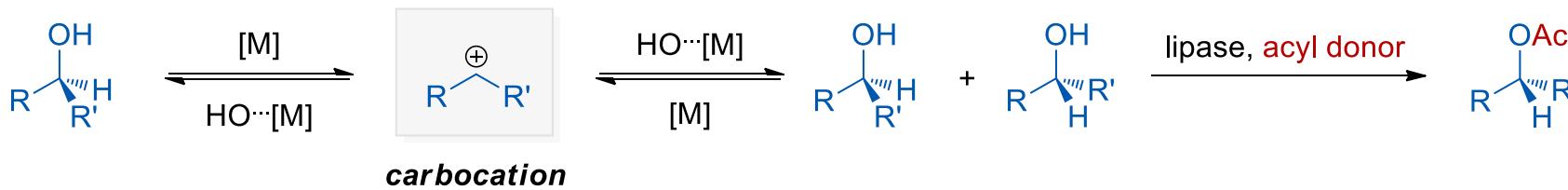
Summary and Outlook

# Part V: Summary and Outlook

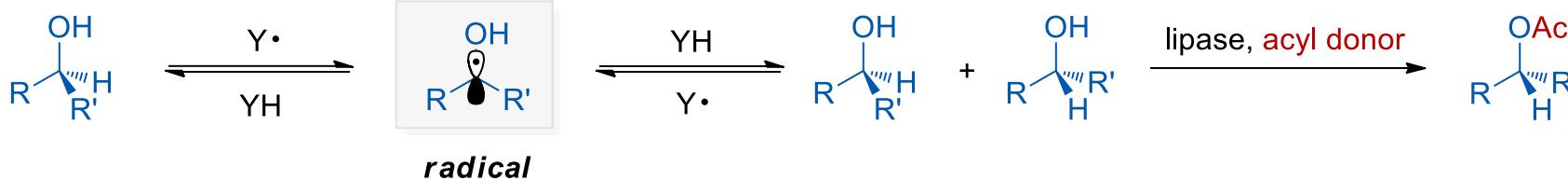
## *Redox racemization via ketone intermediates*



## *Racemization via carbocation intermediates*

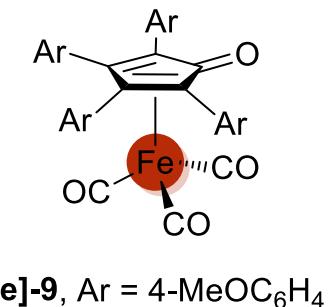
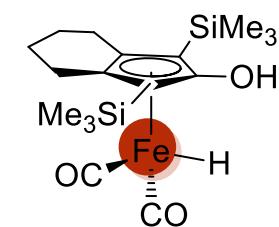
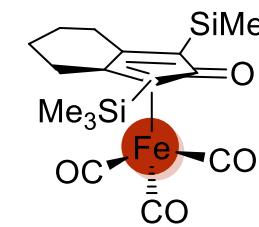
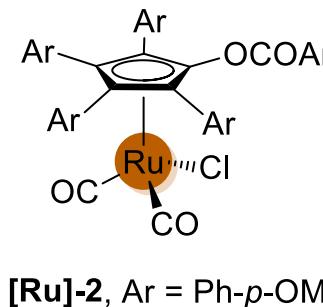
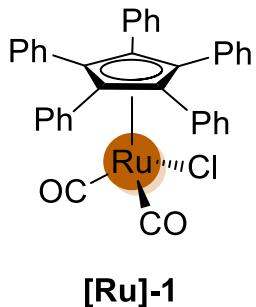


## *Racemization via radical intermediates*



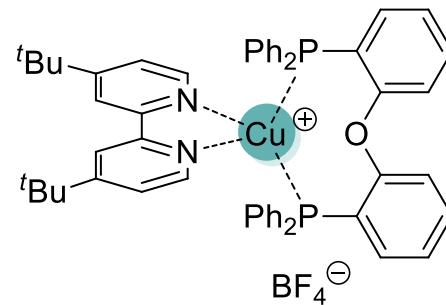
# Part V: Summary and Outlook

*For metal catalysts*

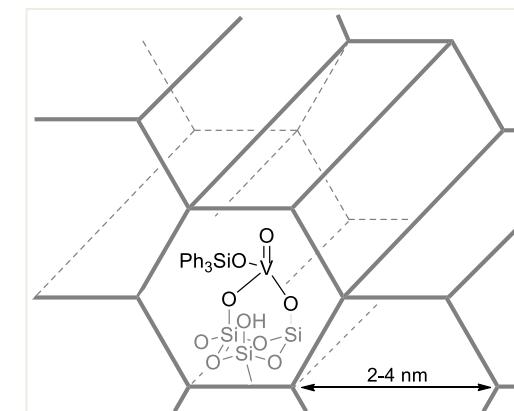


*large ligands protected*

*metal complexes*



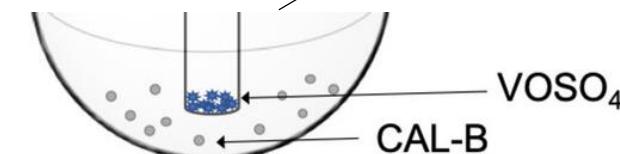
*Ru catalysts*



*Fe catalysts*

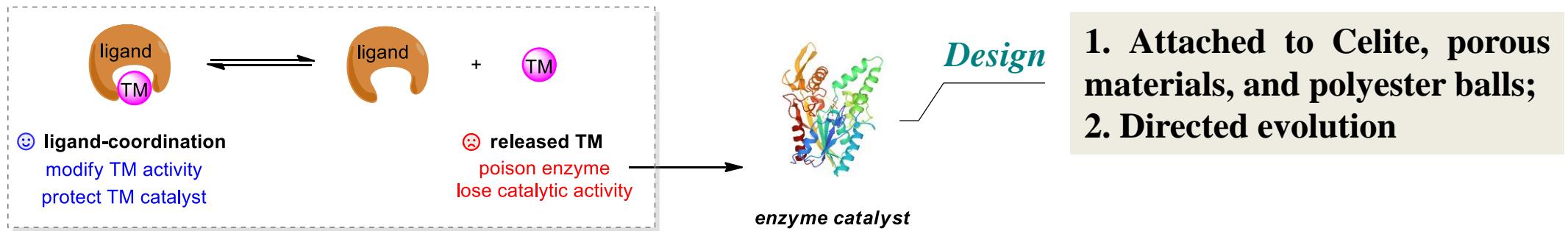
*attached to cellular materials*

*physical separation*



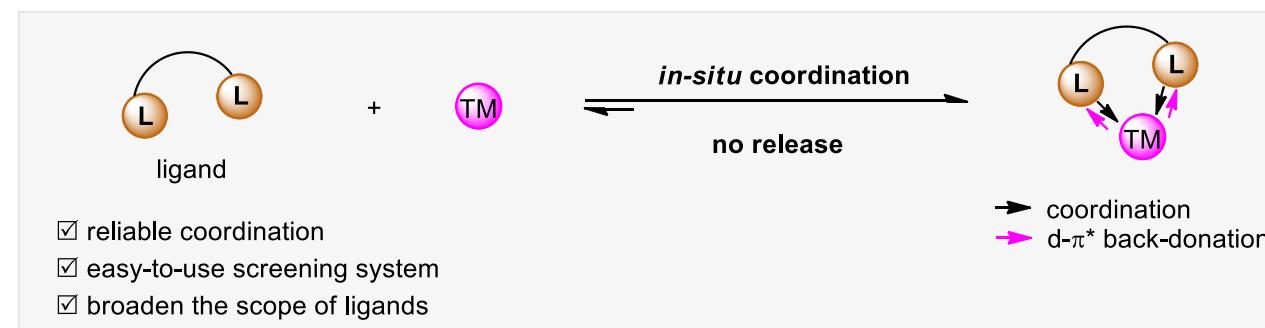
# Part V: Summary and Outlook

**Key issue:** compatibility of TM catalysis with enzyme



1. Attached to Celite, porous materials, and polyester balls;
2. Directed evolution

**Racemization catalysts Design:** complexes with reliable and natural ligand





# Thanks for your attention !

**Student: Kun Wang**

**Supervisor: Can Zhu**

**Fudan University**

**2024.05.10**