



Trans-cyclooctene and Its Application

Reporter: Peng Yan

Supervisor: Ping Lu

Fudan University

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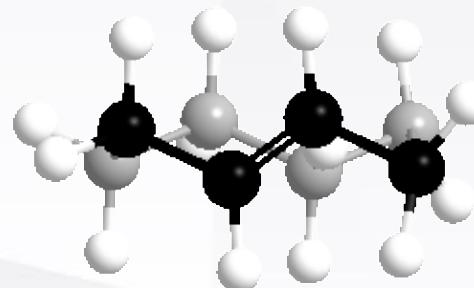
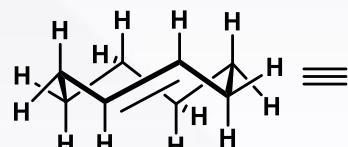
Summary

Part 1

Background

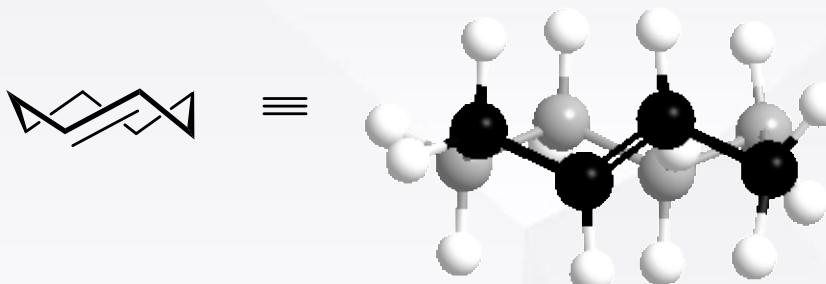
02 The Reaction of TCO

crown conformation

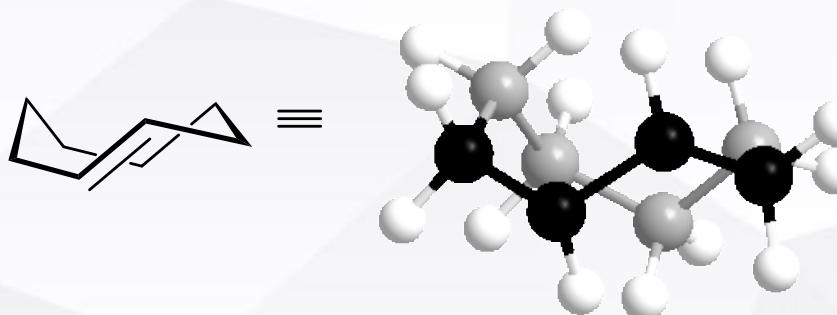


02 The Reaction of TCO

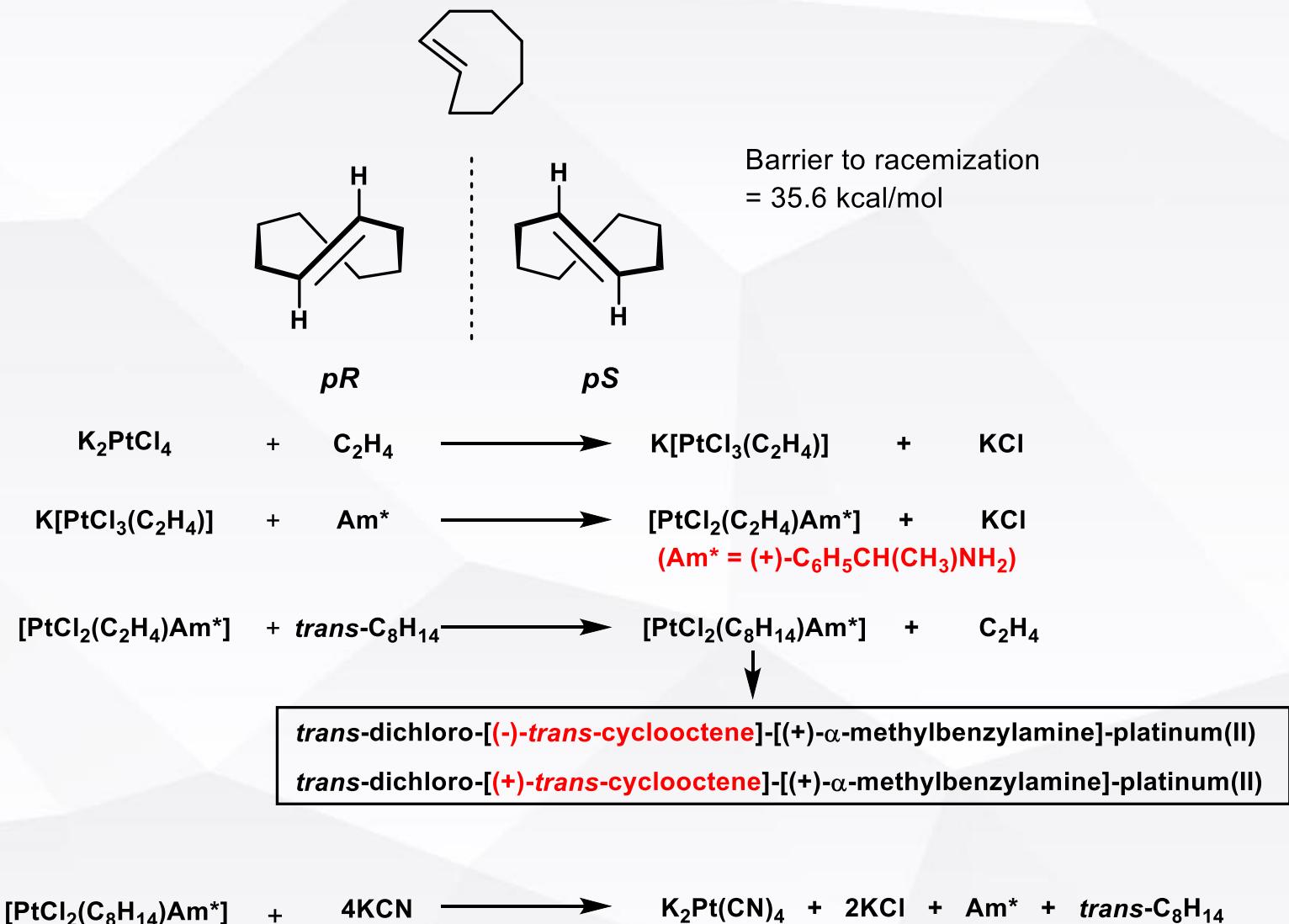
crown conformation



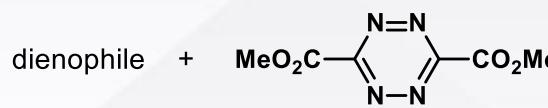
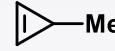
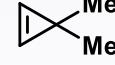
half chair conformation



01 Optically Active TCO

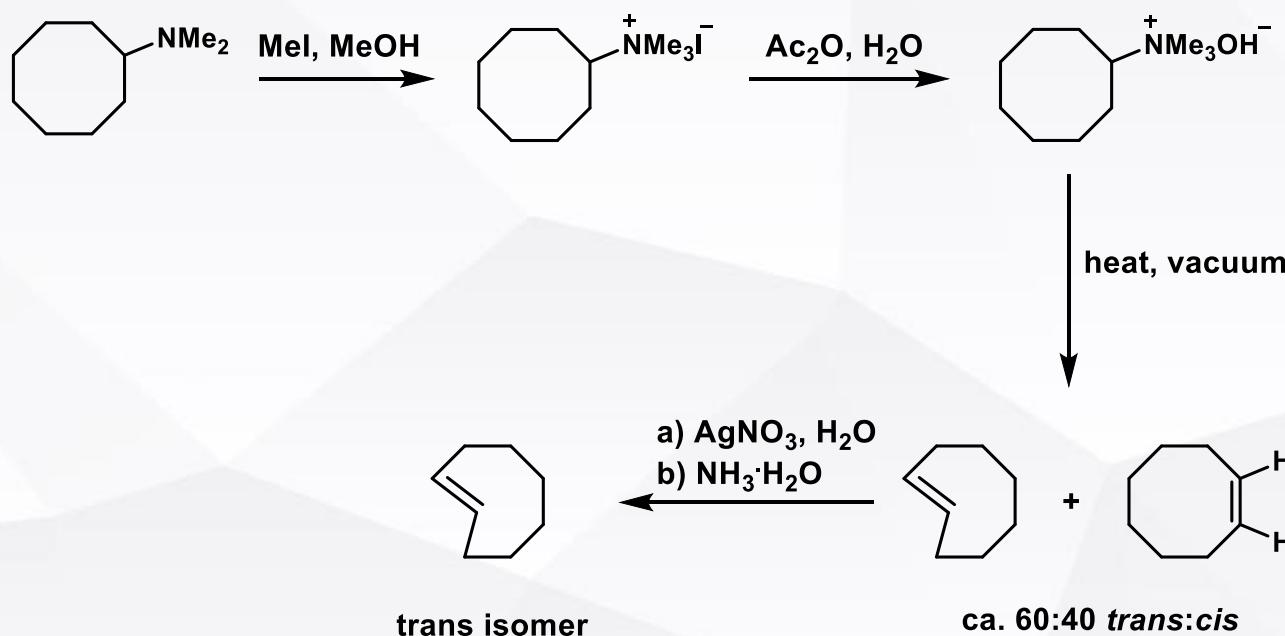
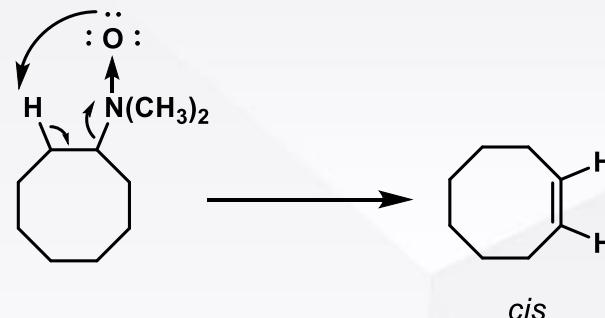


02 The Reaction of TCO

dienophile	+		$20\text{ }^{\circ}\text{C}$, dioxane $k_2 \text{ (l/mol}\cdot\text{s)}$	product
site: $10^5 \cdot k_2 \text{ (l/mol}\cdot\text{s)}$				
				
44800000	329000	23500	86.6	8370
			cis: trans:	3380 1270000000
				
54900000	4.86	104000		2970000
				
7770	6310	136000	769	36.5
				100000
				
35.5	13200	67.7	3.35 (40 °C)	192000

01 Background

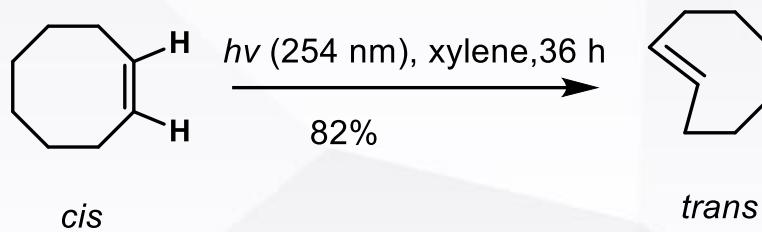
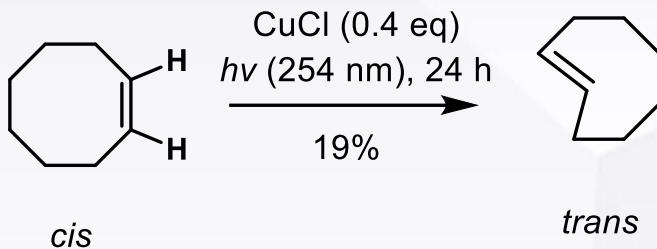
Cope Elimination:



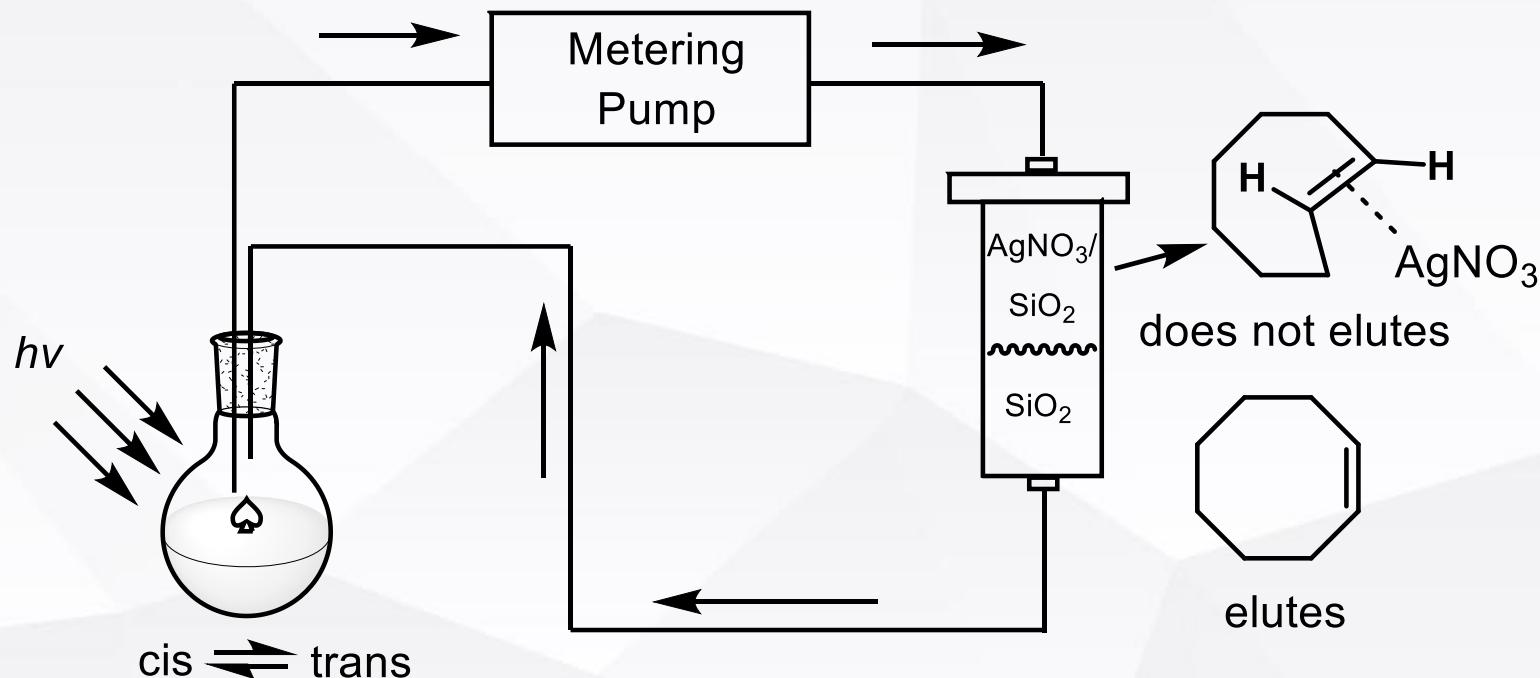
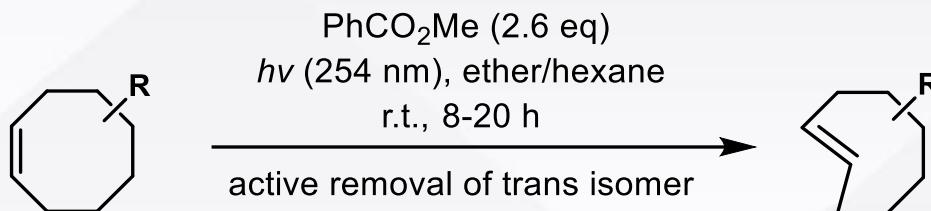
K. Ziegler, H. Wilms, *Ann.*, **1950**, 1, 567.; S. W. Fenton, C. F. Spencer, A. C. Cope, *J. Am. Chem. Soc.*, **1952**, 74, 5884.

R. A. Pike, C. F. Spencer, A. C. Cope, *J. Am. Chem. Soc.*, **1953**, 75, 3212.

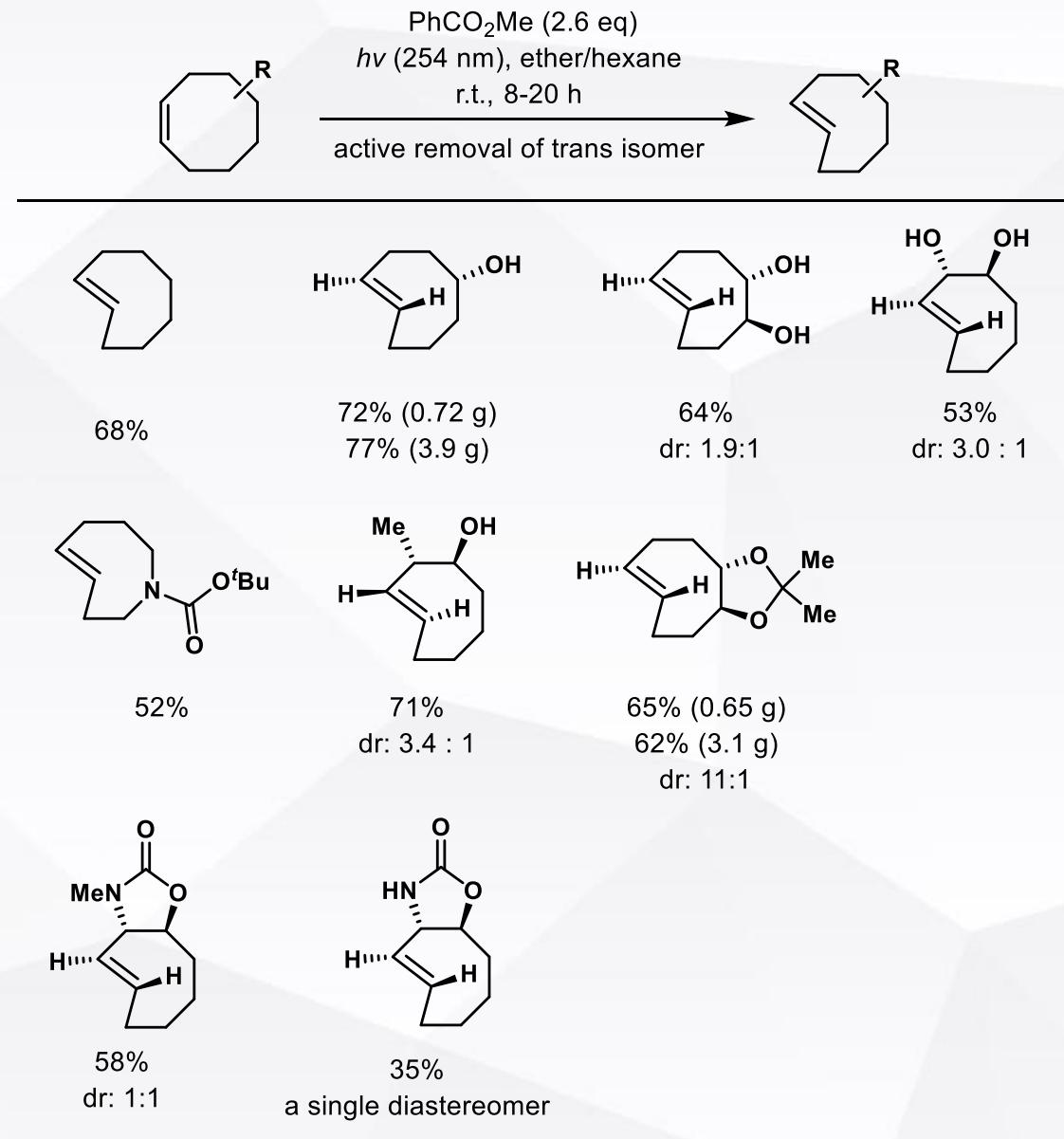
01 New Synthesis Method



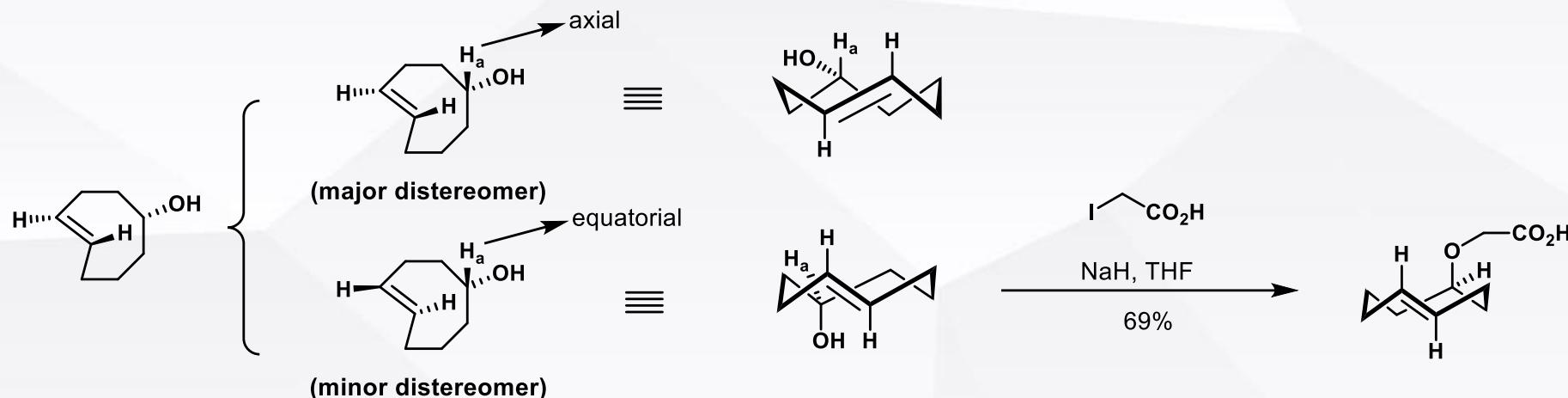
01 New Synthesis Method



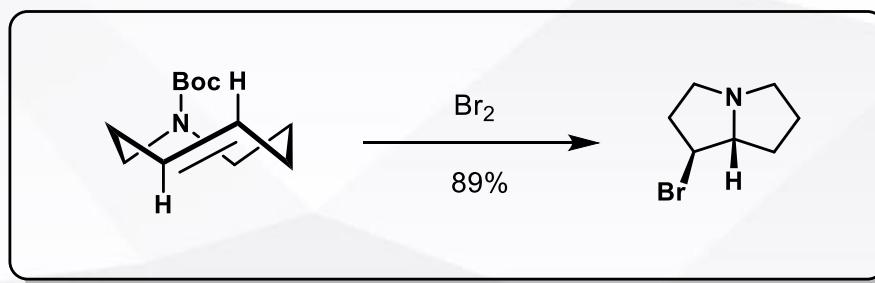
01 New Synthesis Method



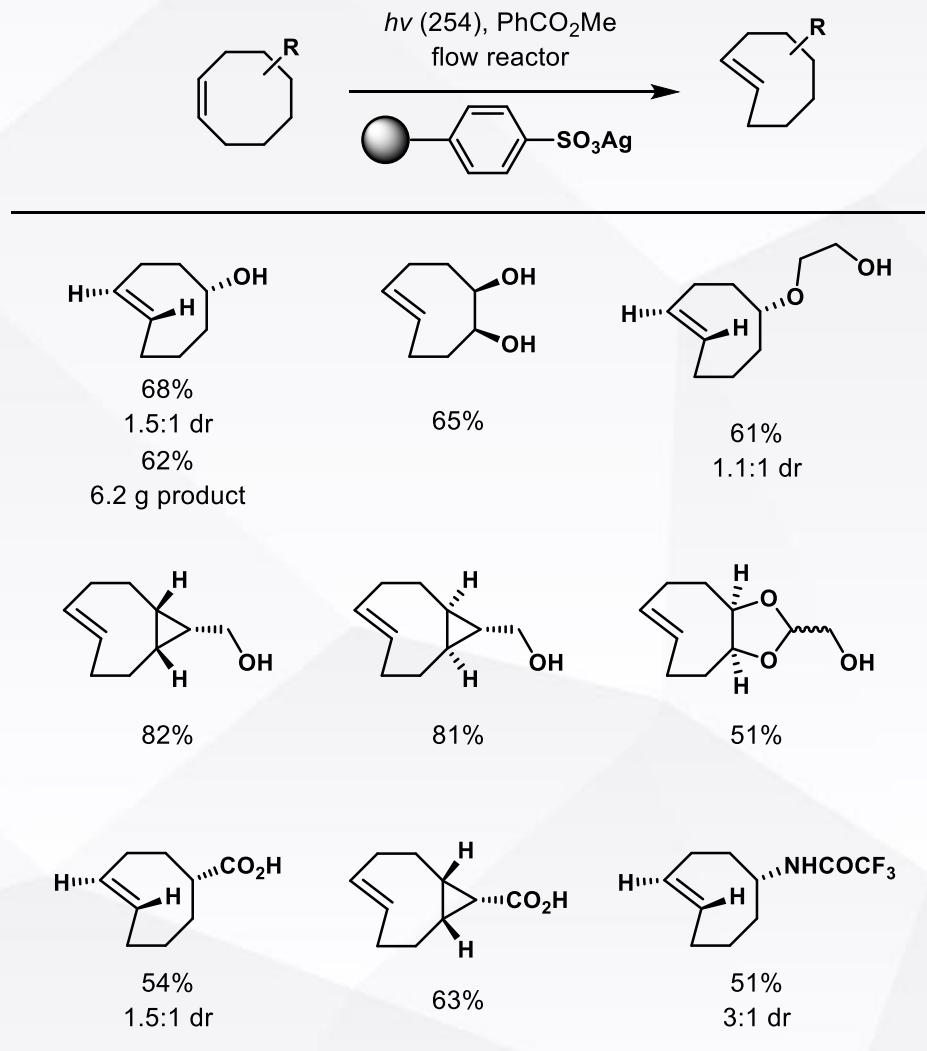
01 New Synthesis Method



Transannular Cyclization Reaction



01 New Synthesis Method



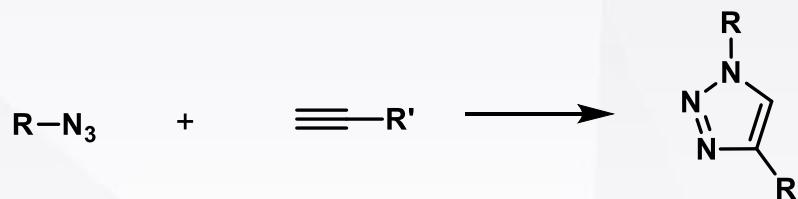
Part 2

The Reaction of TCO

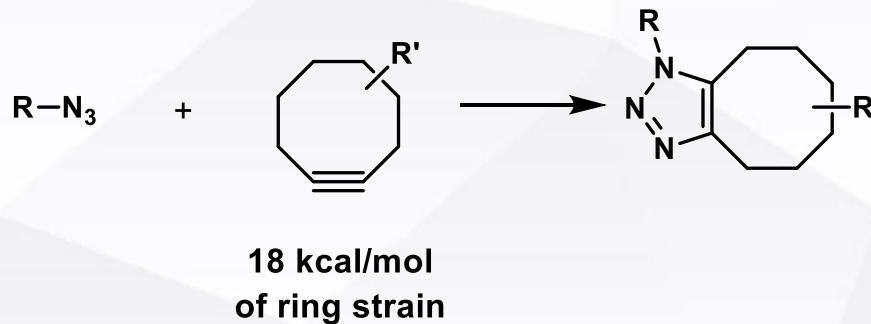
02 The Reaction of TCO

Bioorthogonal chemistry: Any chemical reaction that can occur inside of living systems without interfering with native biochemical processes.

A. Cu(I)-catalyzed Huisgen cycloaddition ("click" chemistry)

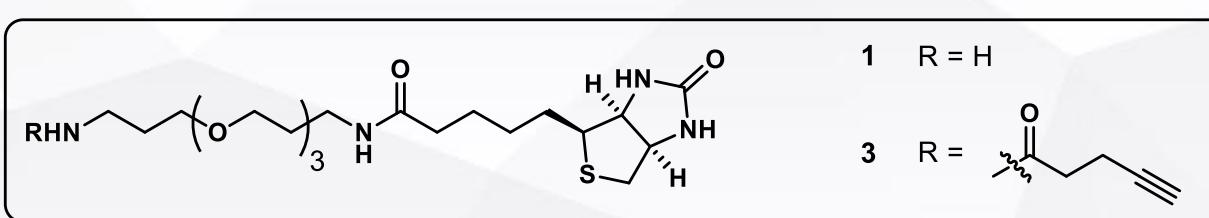
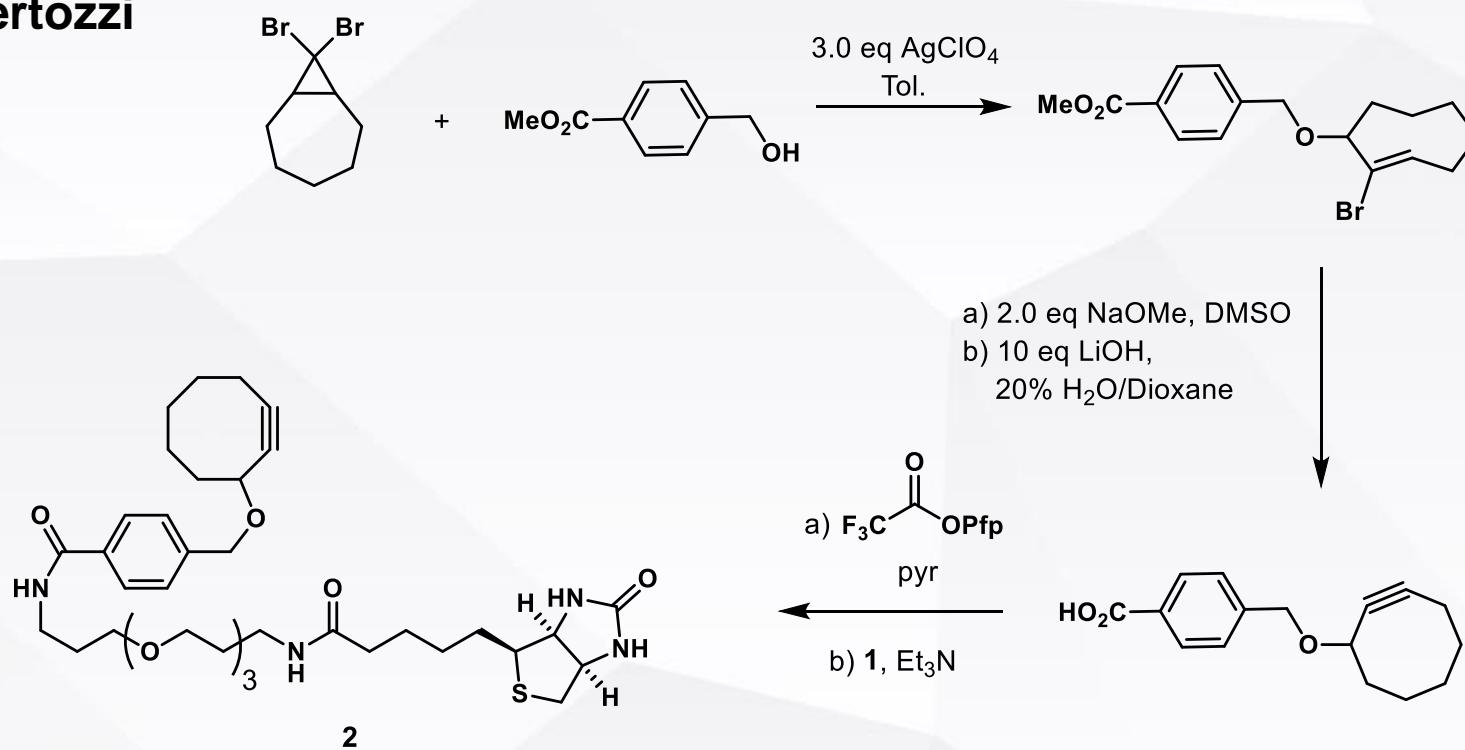


B. Strain-promoted [3+2] cycloaddition of azides and cyclooctynes

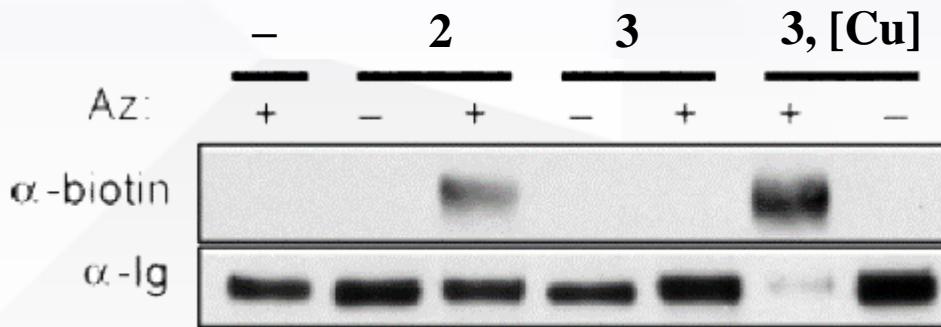
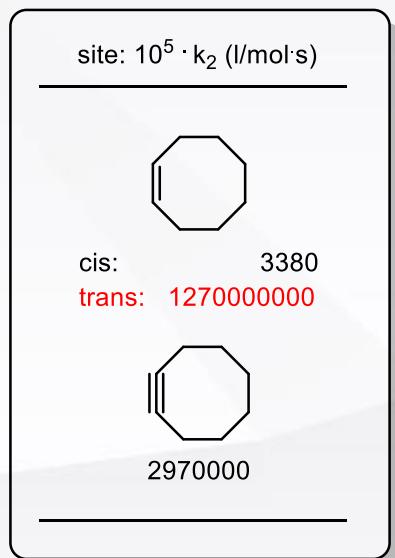
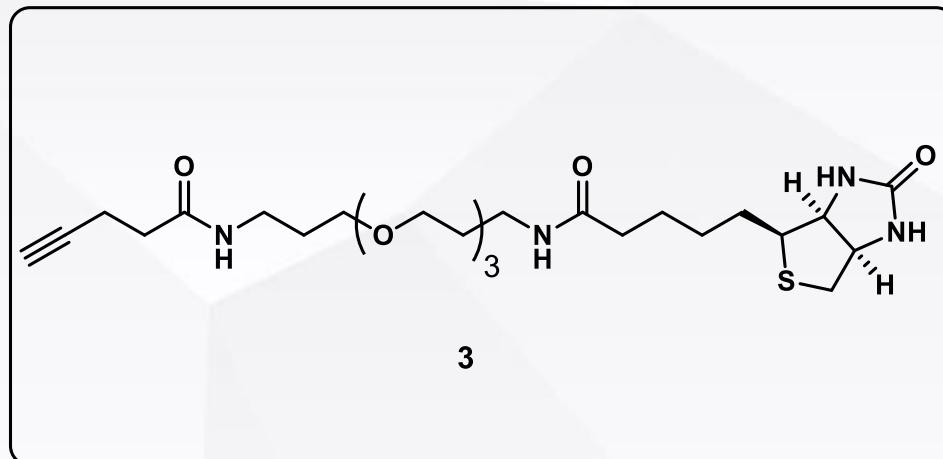
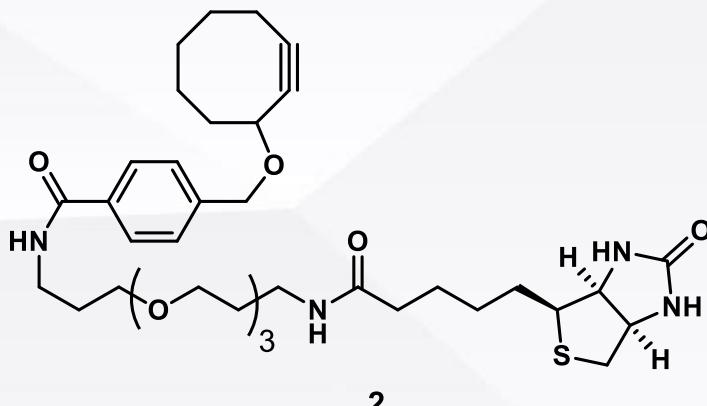


02 The Reaction of TCO

By Bertozzi



02 The Reaction of TCO

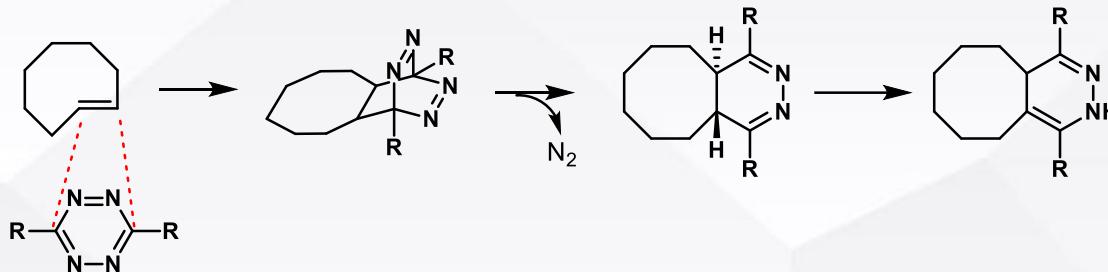


$$k_2 \text{ of SPAAC} = 5 \text{ M}^{-1}\text{s}^{-1}$$

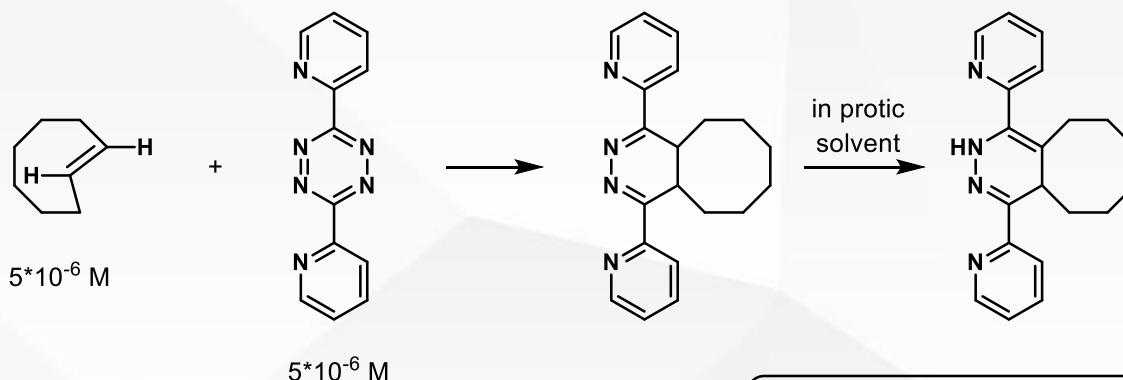
SPAAC = Strain-Promoted Azide-Alkyne Cycloaddition

02 The Reaction of TCO

Diels-Alder Reactions of Tetrazines with *trans*-Cyclooctene



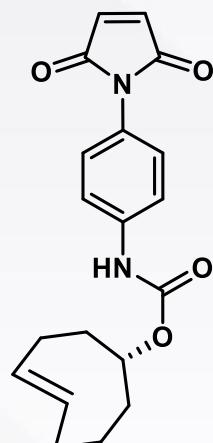
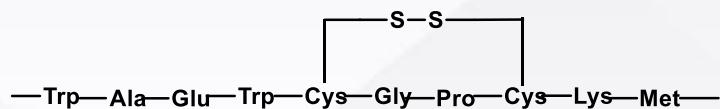
Fast Reactivity at Low Micromolar Concentrations



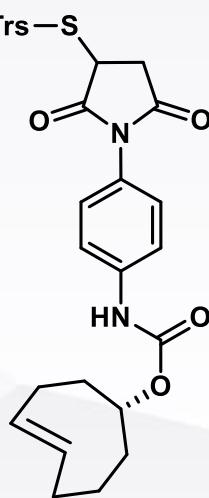
- 100% conv. after 40 min at 25 °C at 5×10^{-6} M
 $t_{1/2} = 7\text{ s}$ when TCO:Tetrazine = 1:10
- ~quantitative yield with $k_2 2000 \text{ M}^{-1}\text{s}^{-1}$
(SPAAC = $5 \text{ M}^{-1}\text{s}^{-1}$)
- Successful reactivity in organic solvents,
water, cell media or cell lysate
- N_2 is the only byproduct

02 The Reaction of TCO

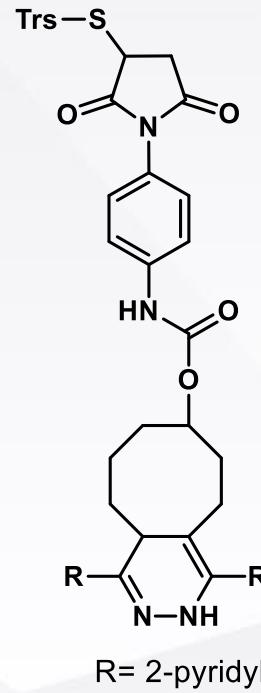
Trx:



Trx (15 μM)
Trx = thioredoxin



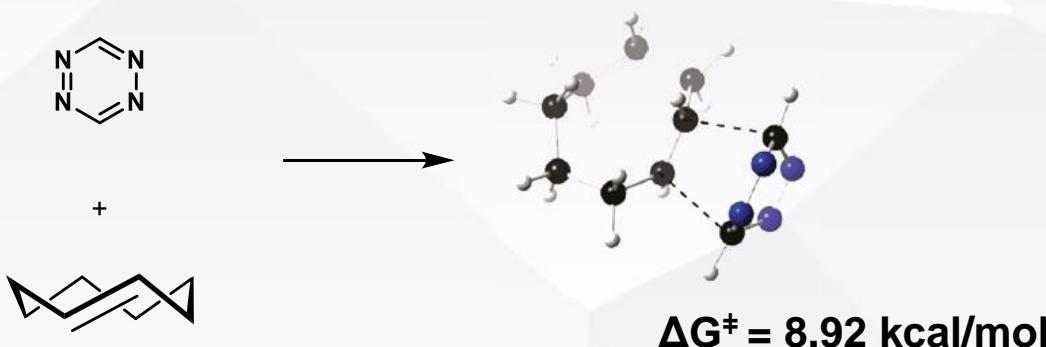
(30 μM)
R = 2-pyridyl
100% conversion
within 5 minutes



R = 2-pyridyl

02 The Reaction of TCO

a

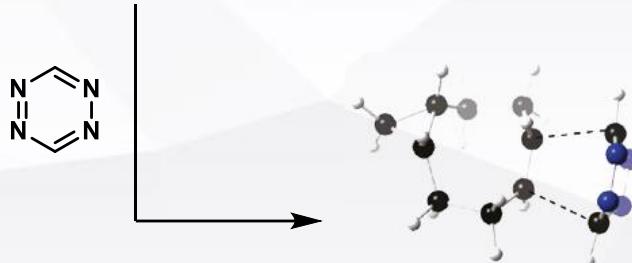


b

s-TCO

cis ring

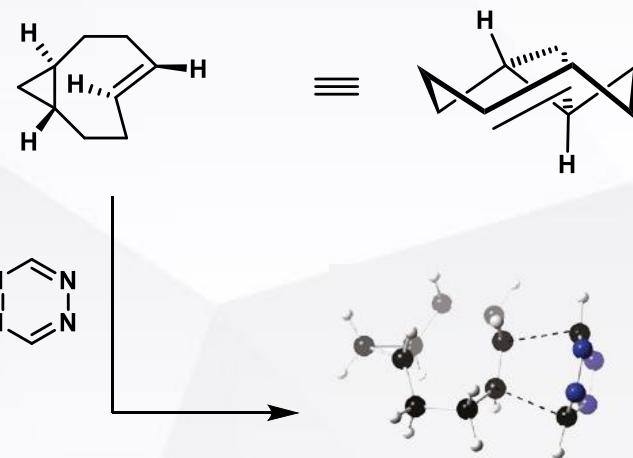
$\Delta G^\ddagger = 6.95 \text{ kcal/mol}$



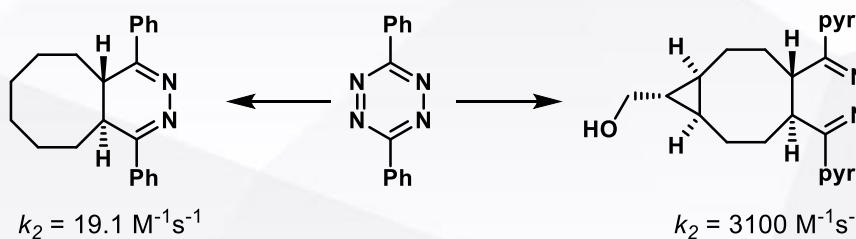
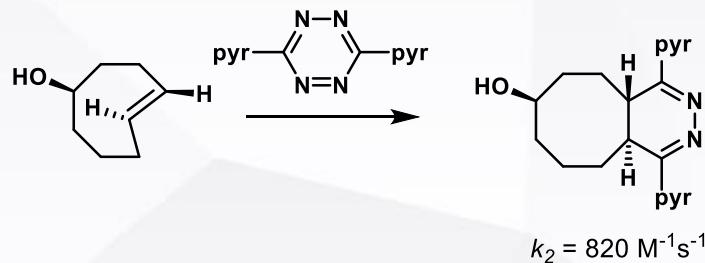
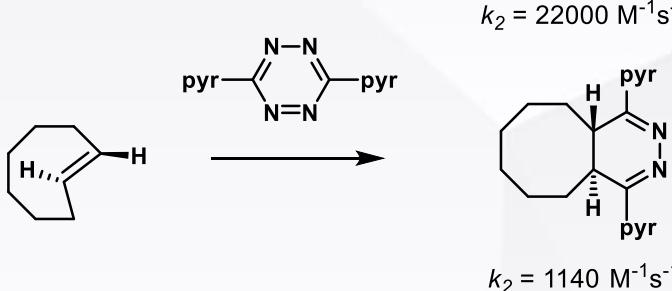
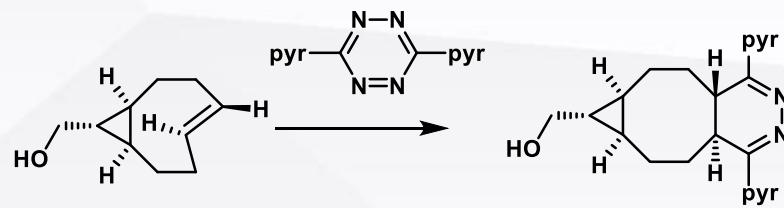
c

trans ring

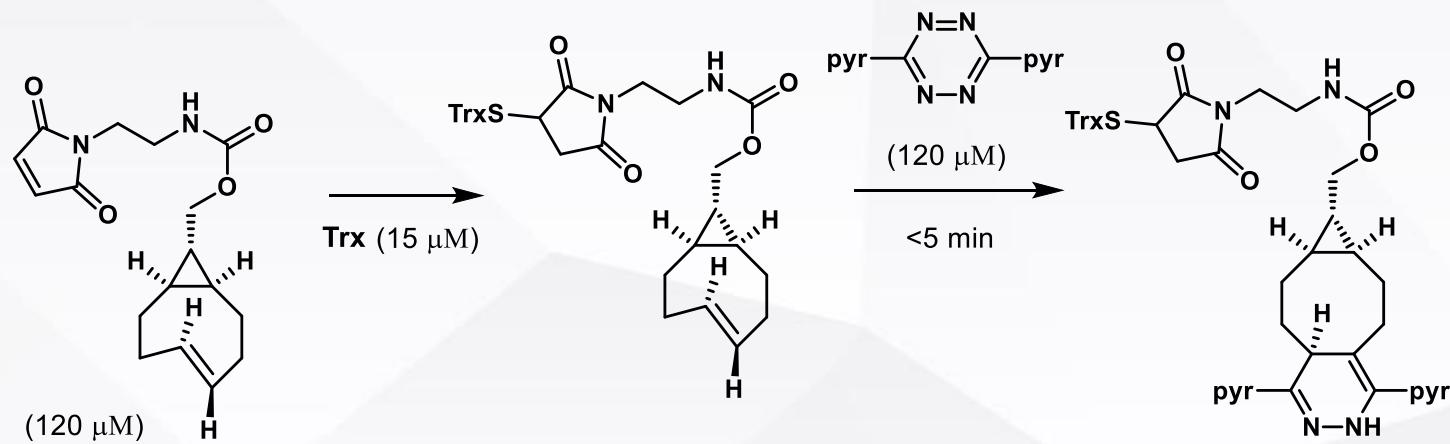
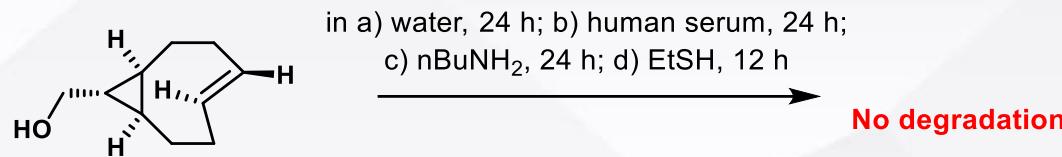
$\Delta G^\ddagger = 8.24 \text{ kcal/mol}$



02 The Reaction of TCO

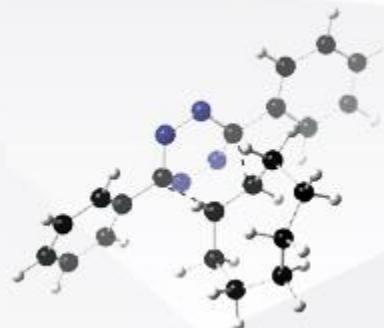
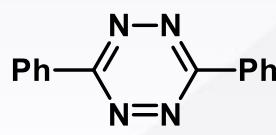


02 The Reaction of TCO



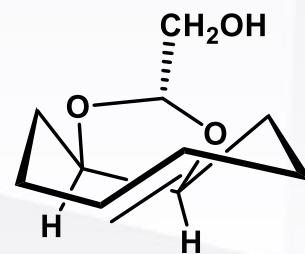
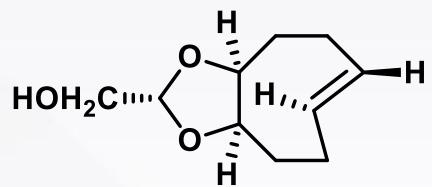
02 The Reaction of TCO

a

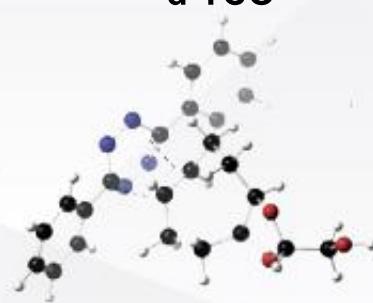
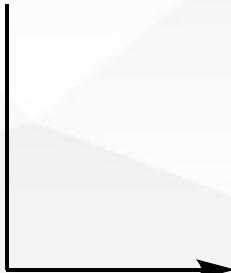
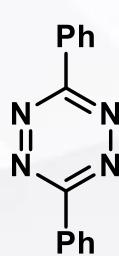


$$\Delta G^\ddagger = 16.09 \text{ kcal/mol}$$

b

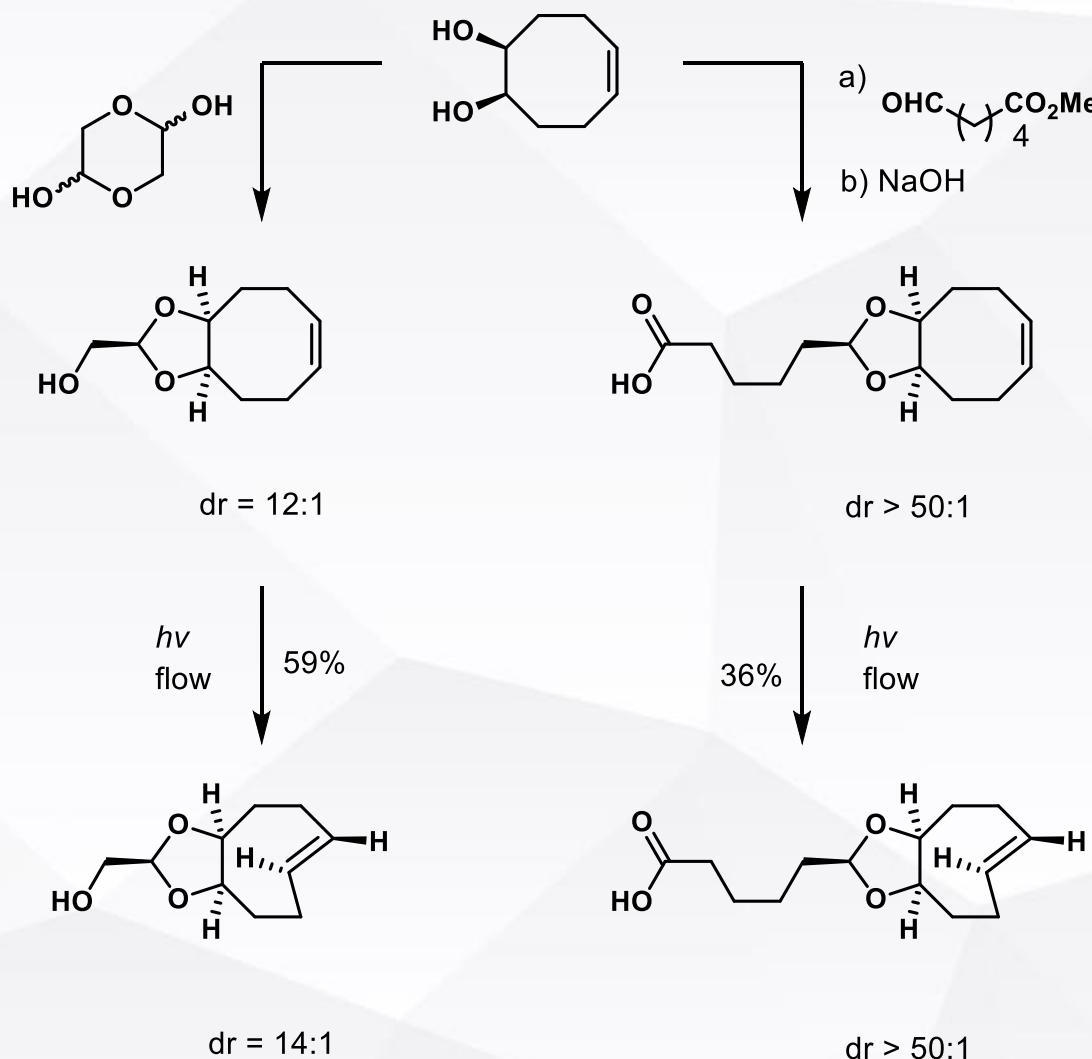


d-TCO

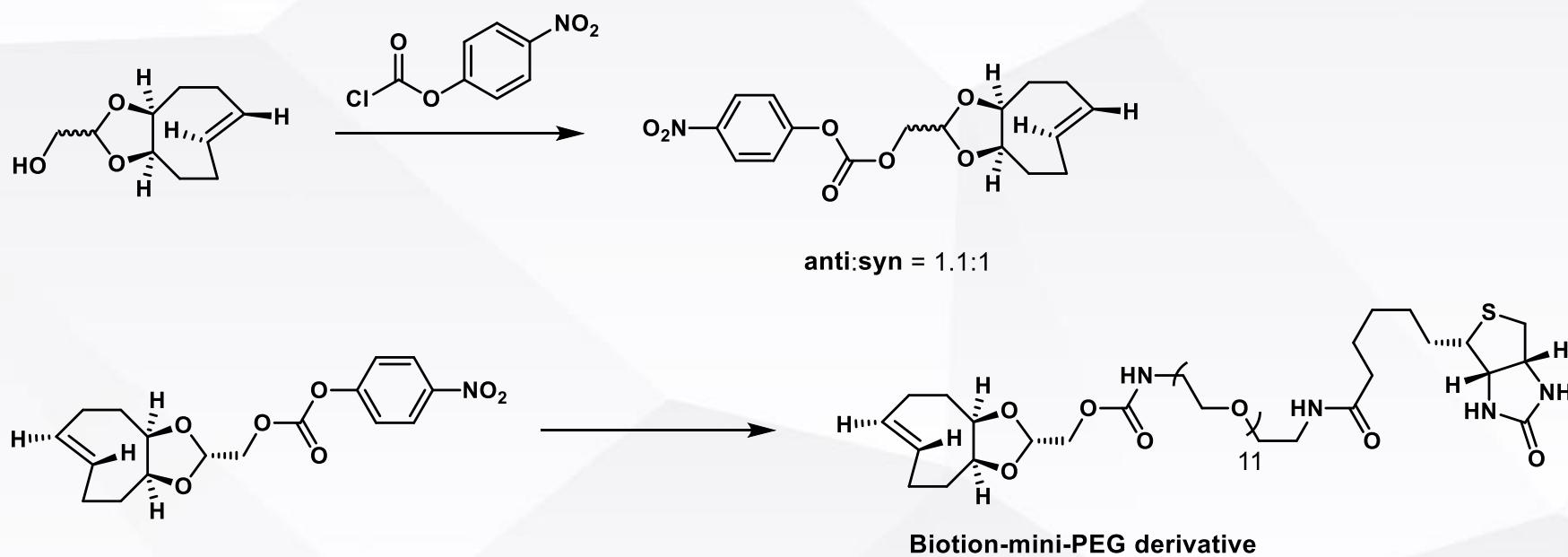


$$\Delta G^\ddagger = 13.27 \text{ kcal/mol}$$

02 The Reaction of TCO



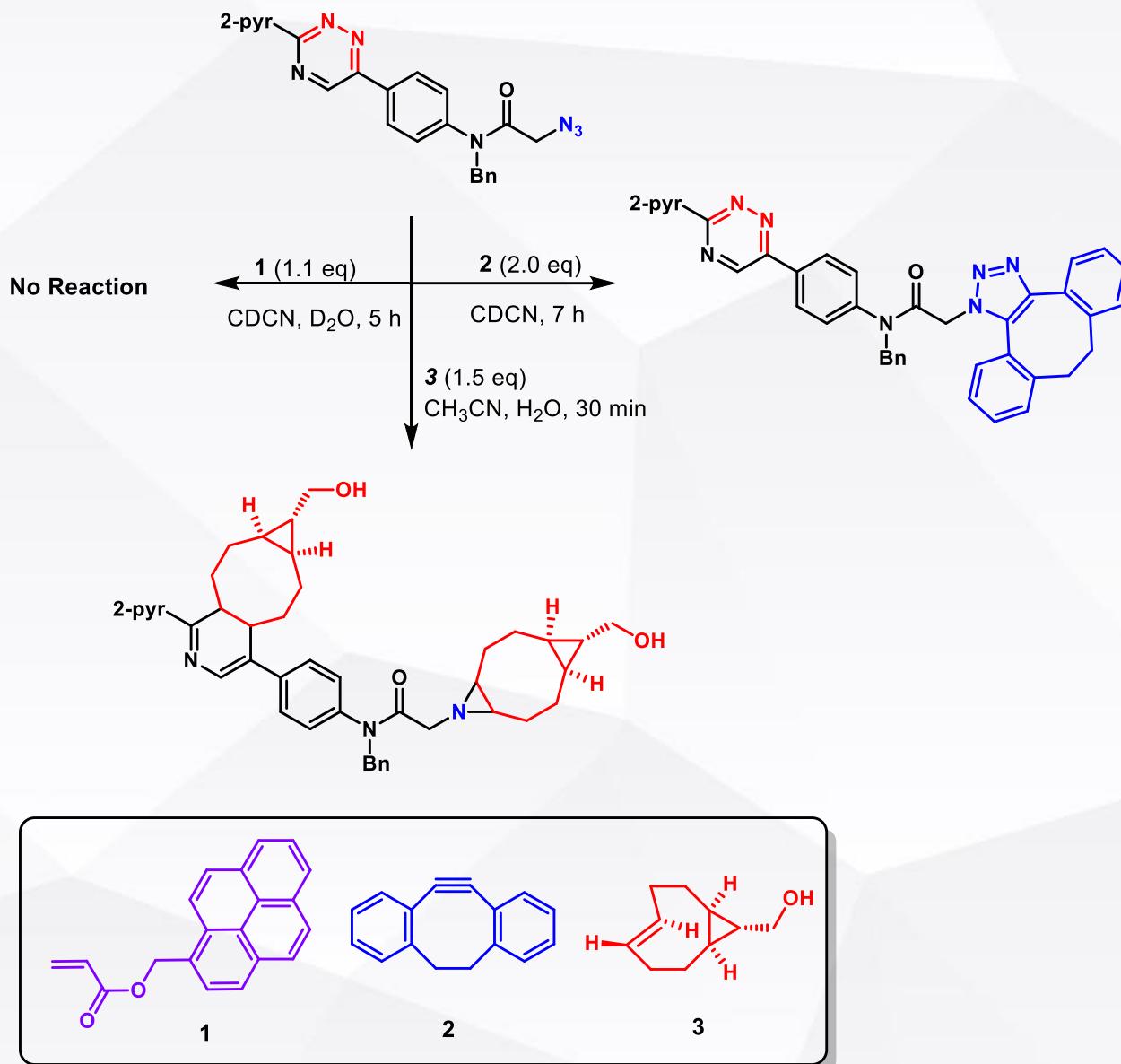
02 The Reaction of TCO

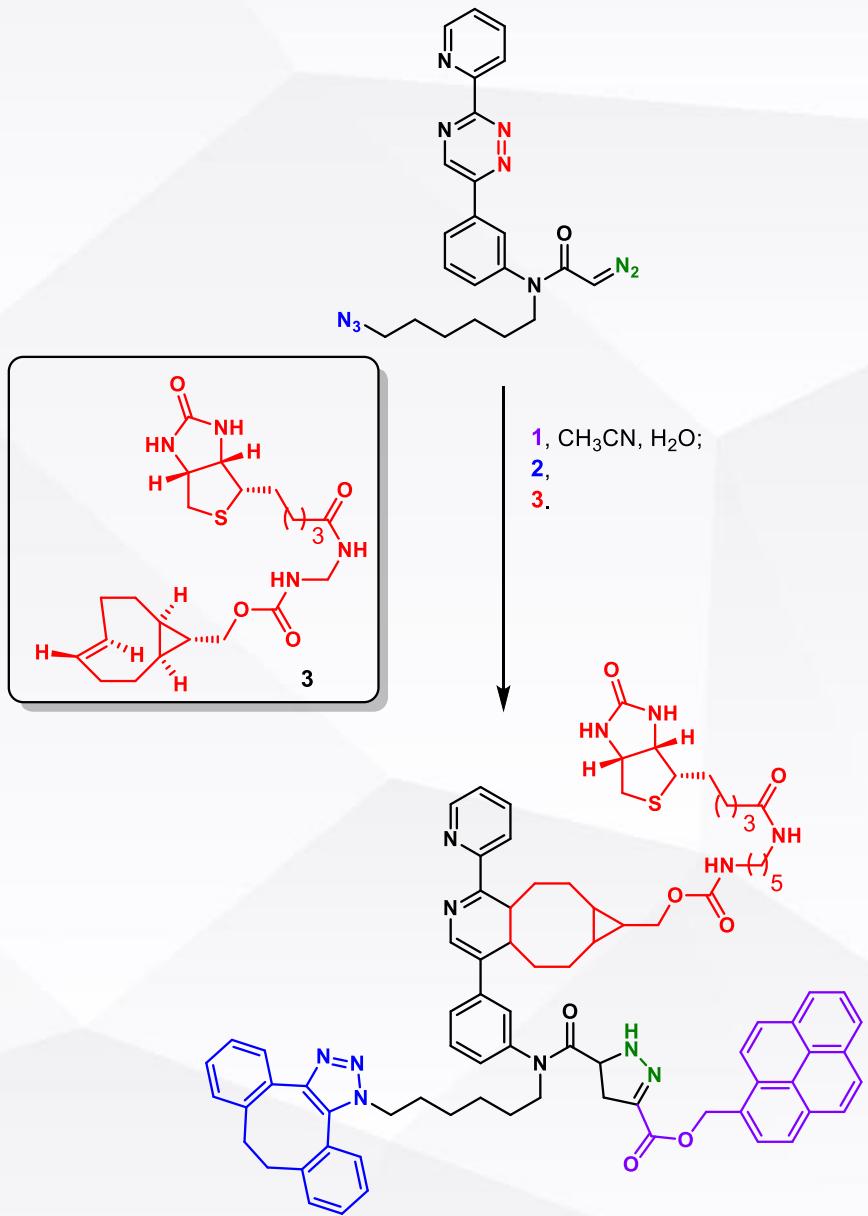


02 The Reaction of TCO

Tz	TCO	k_2 ($M^{-1}s^{-1}$)	Tz	TCO	k_2 ($M^{-1}s^{-1}$)
 MeOH, 25 °C	<p>520</p>		 H_2O , 25 °C	<p>19.1</p>	
 55:45 MeOH:H ₂ O 25 °C	<p>syn, anti 167000 1:1 syn:anti</p> <p>equatorial diastereomer</p> <p>5235</p>		 H_2O 25 °C	<p>41</p>	
			<p>syn 366000 anti 318000</p>	<p>22600</p>	
			<p>80200</p>	<p>3300000</p>	

02 The Reaction of TCO



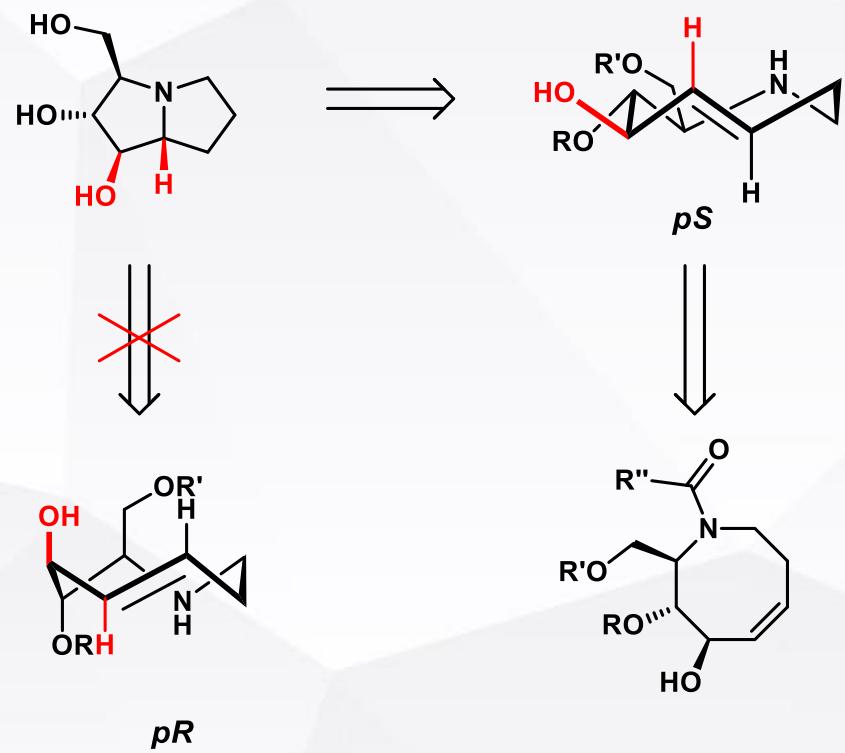
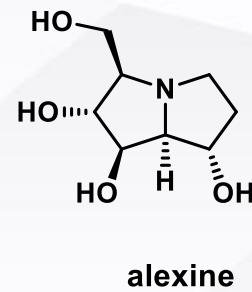
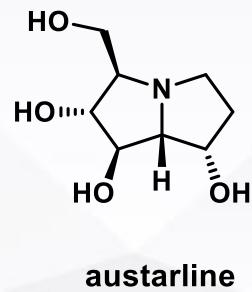
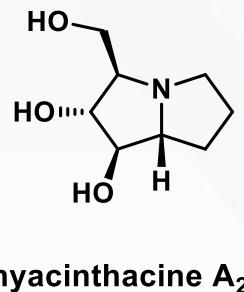
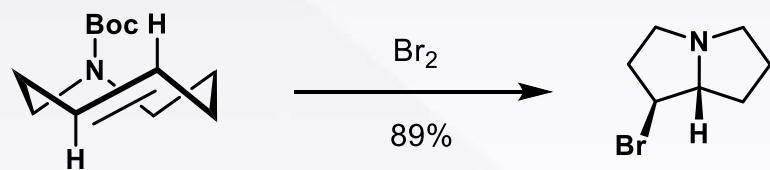


Part 3

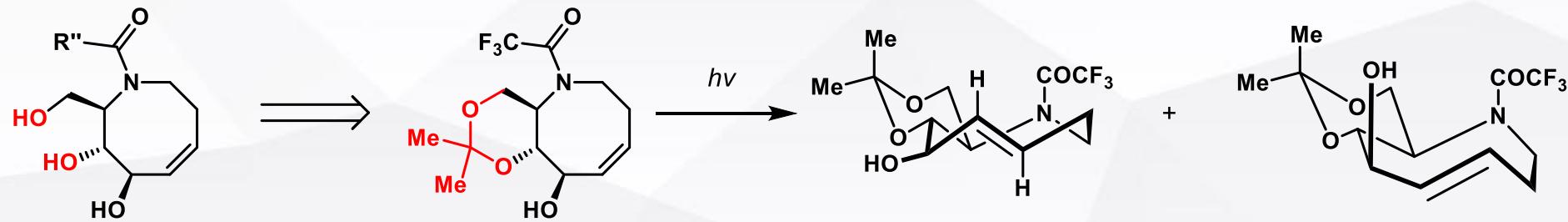
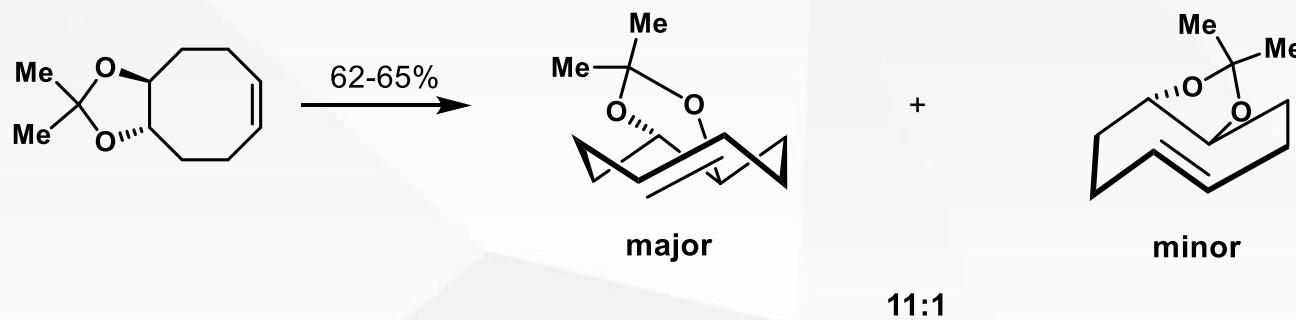
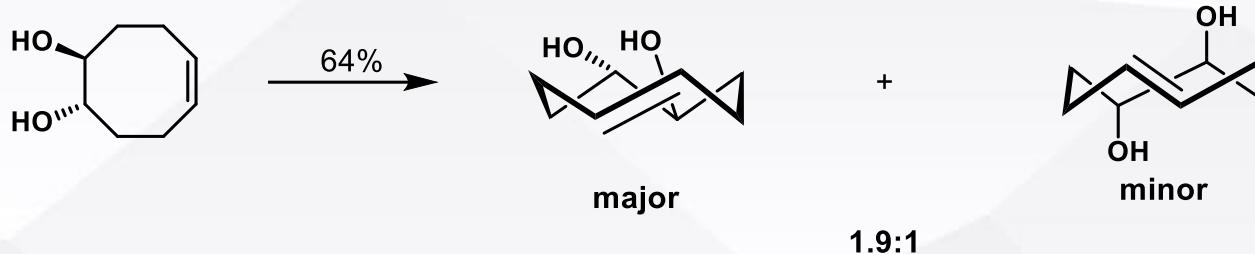
Some New Reactions of TCO

03 Some New Reactions of TCO

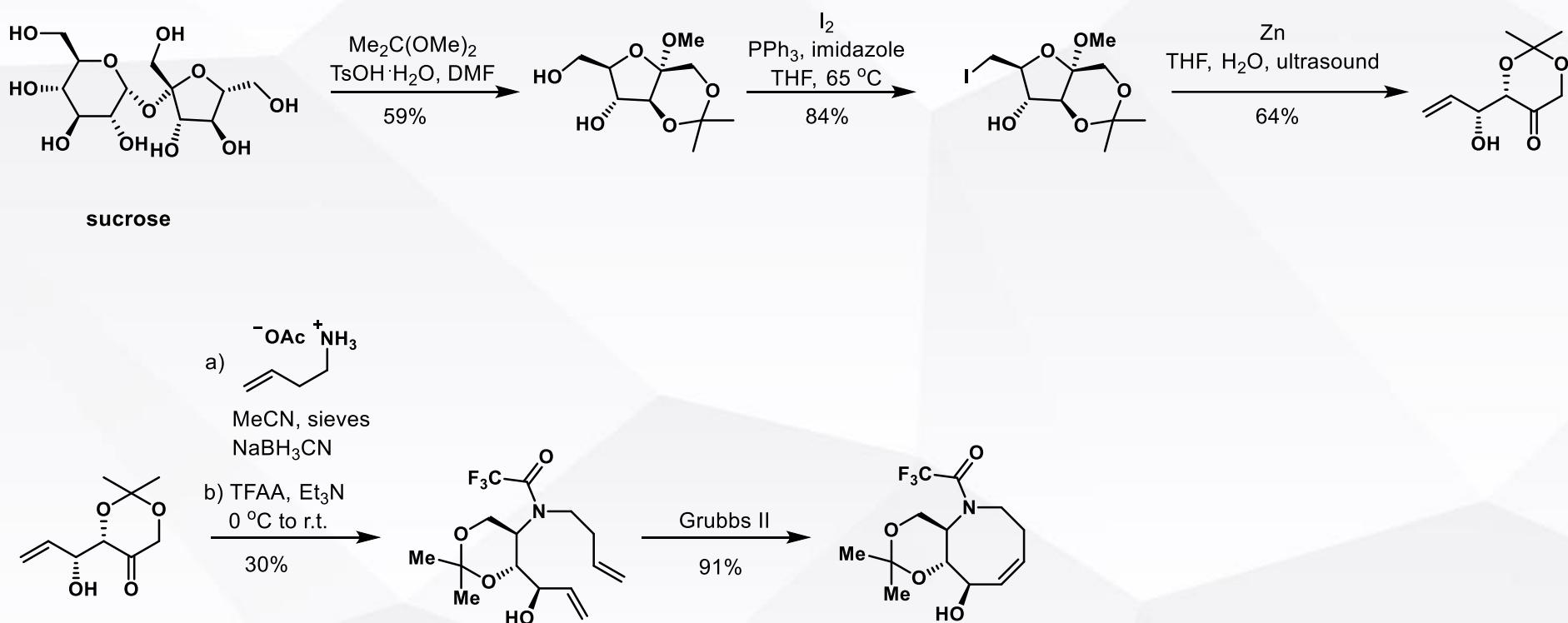
Fox 2008



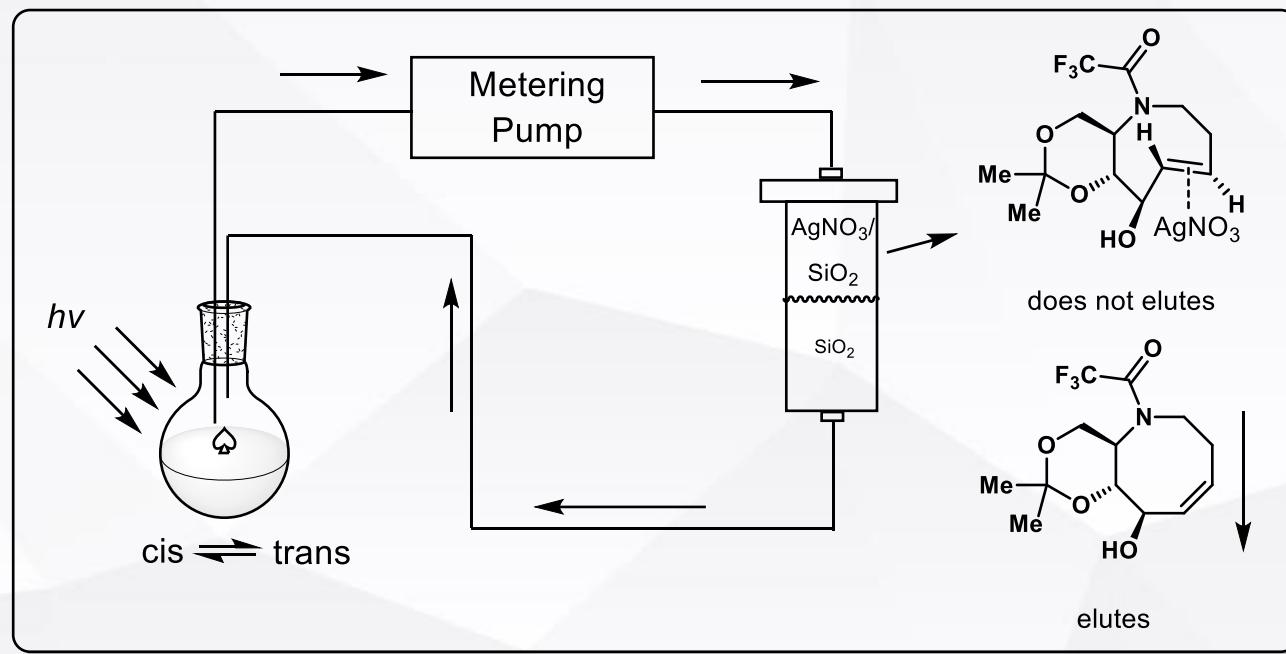
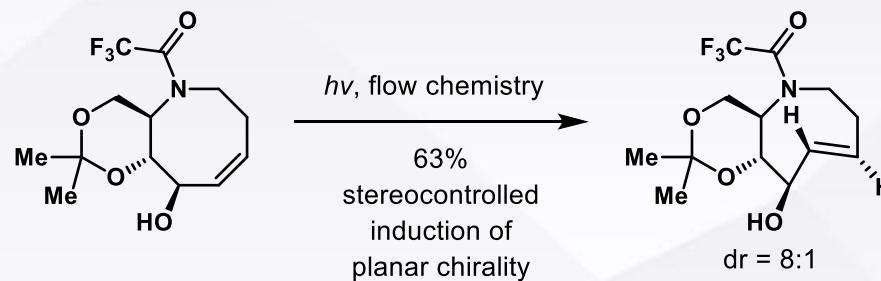
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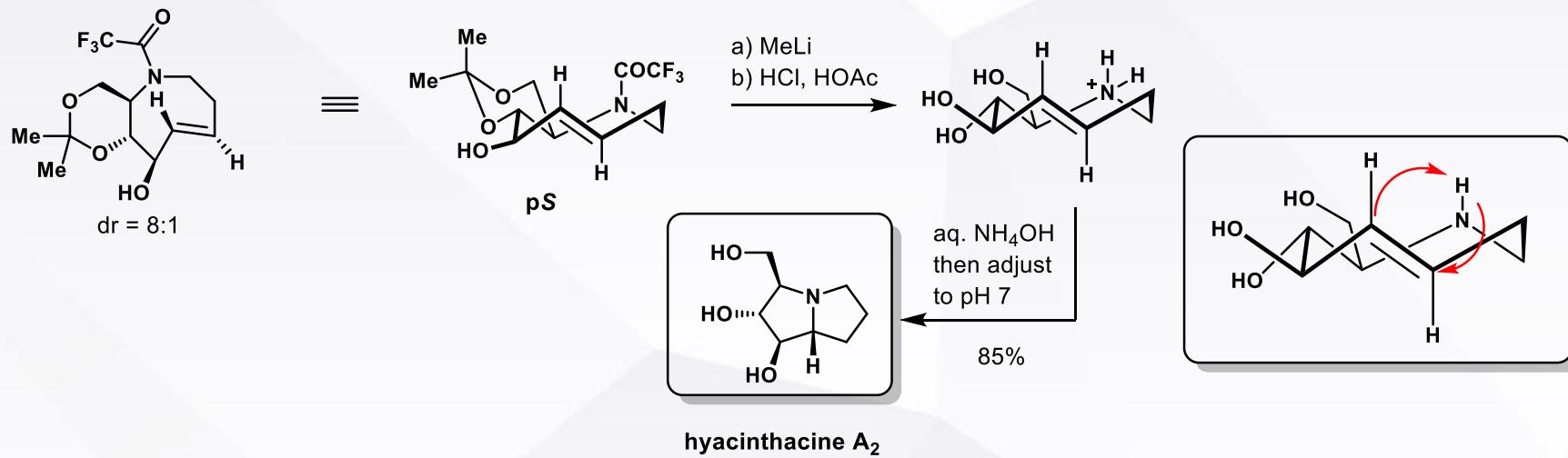
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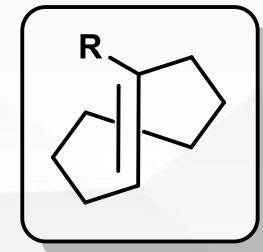
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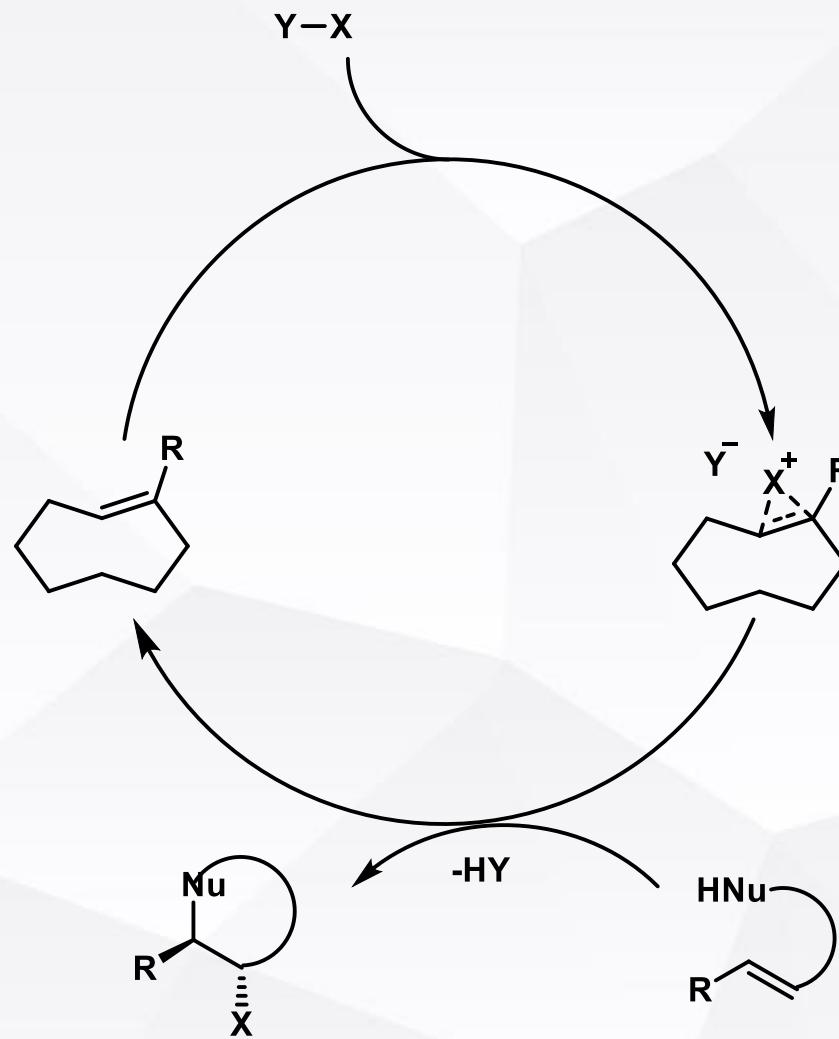
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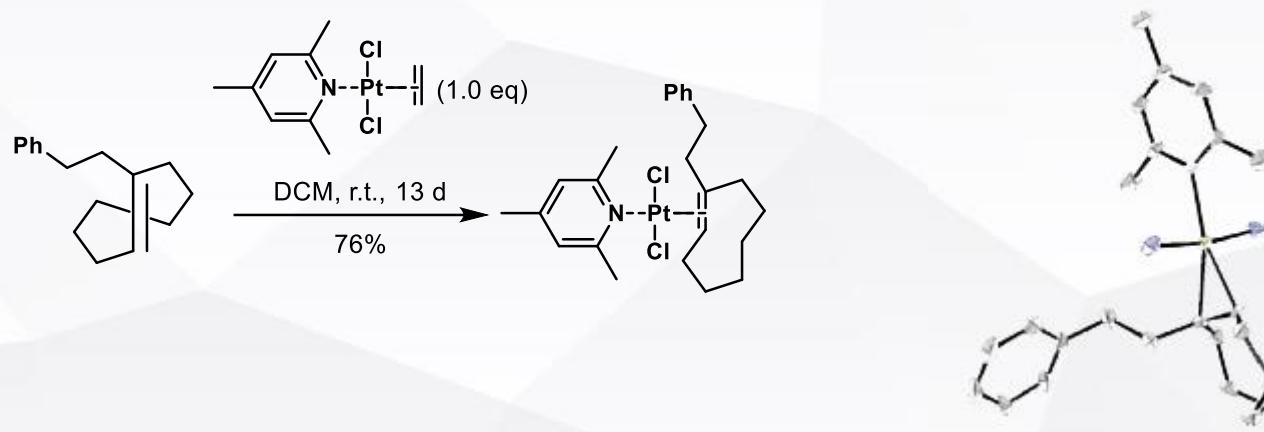
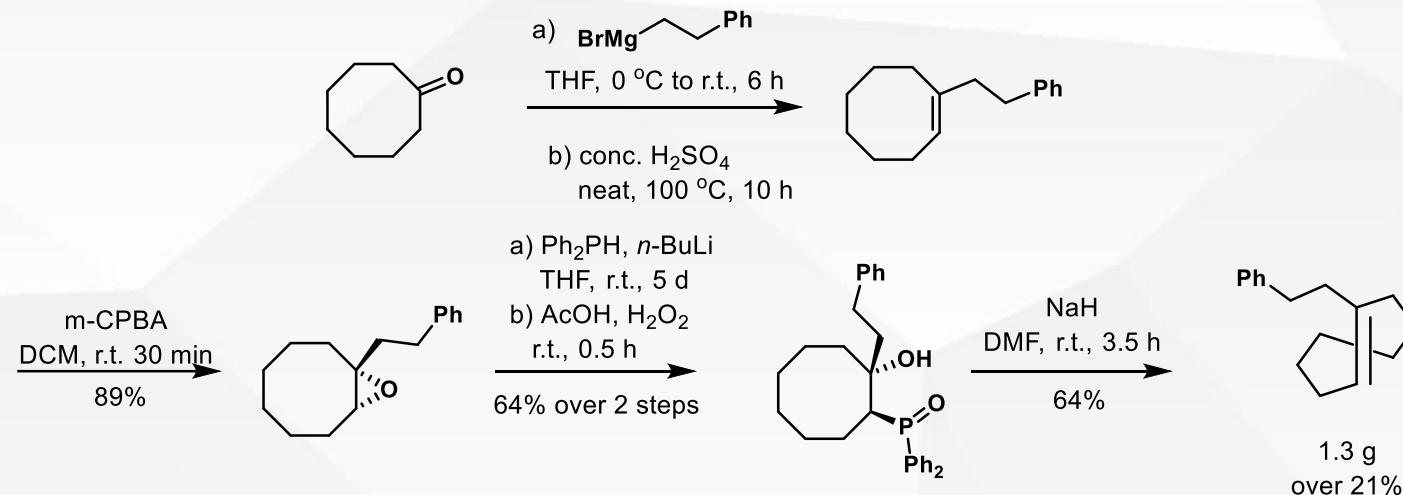
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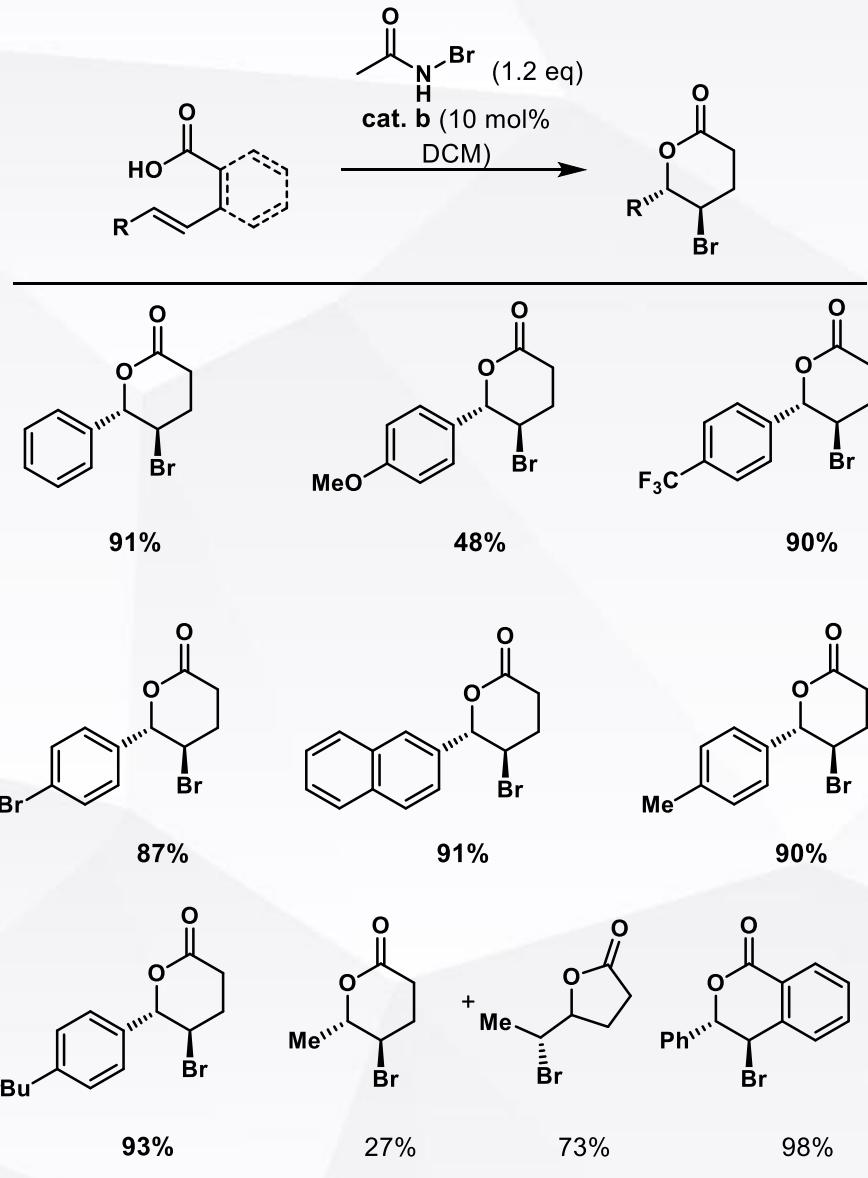
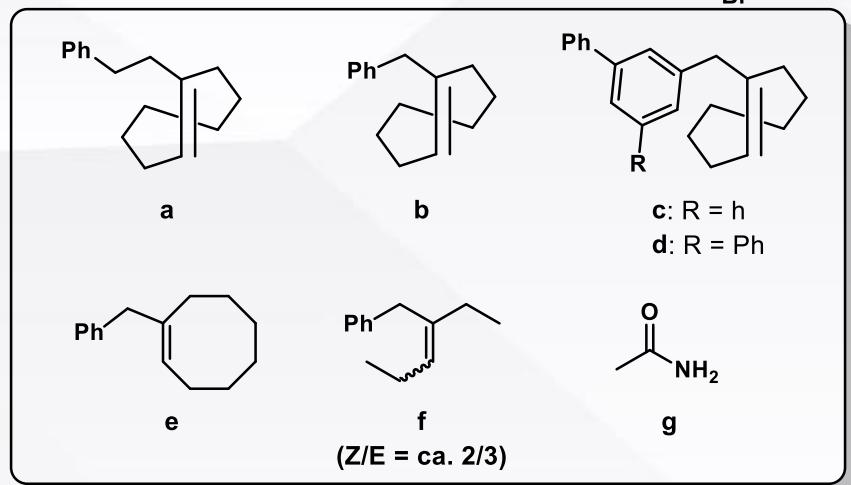
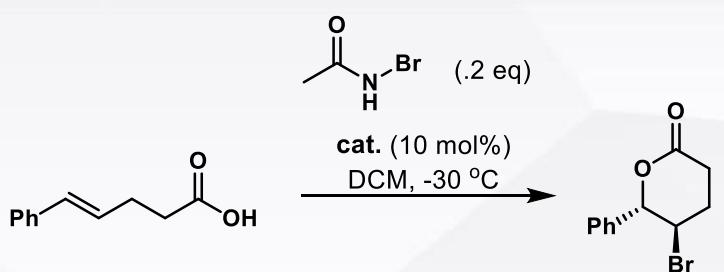


strained olefin



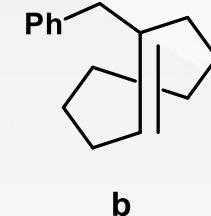
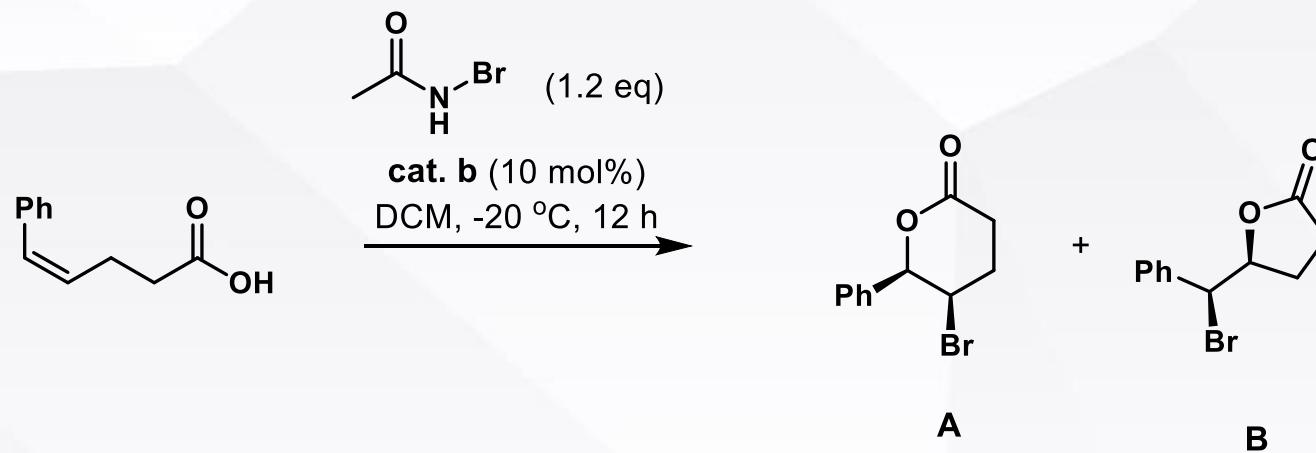
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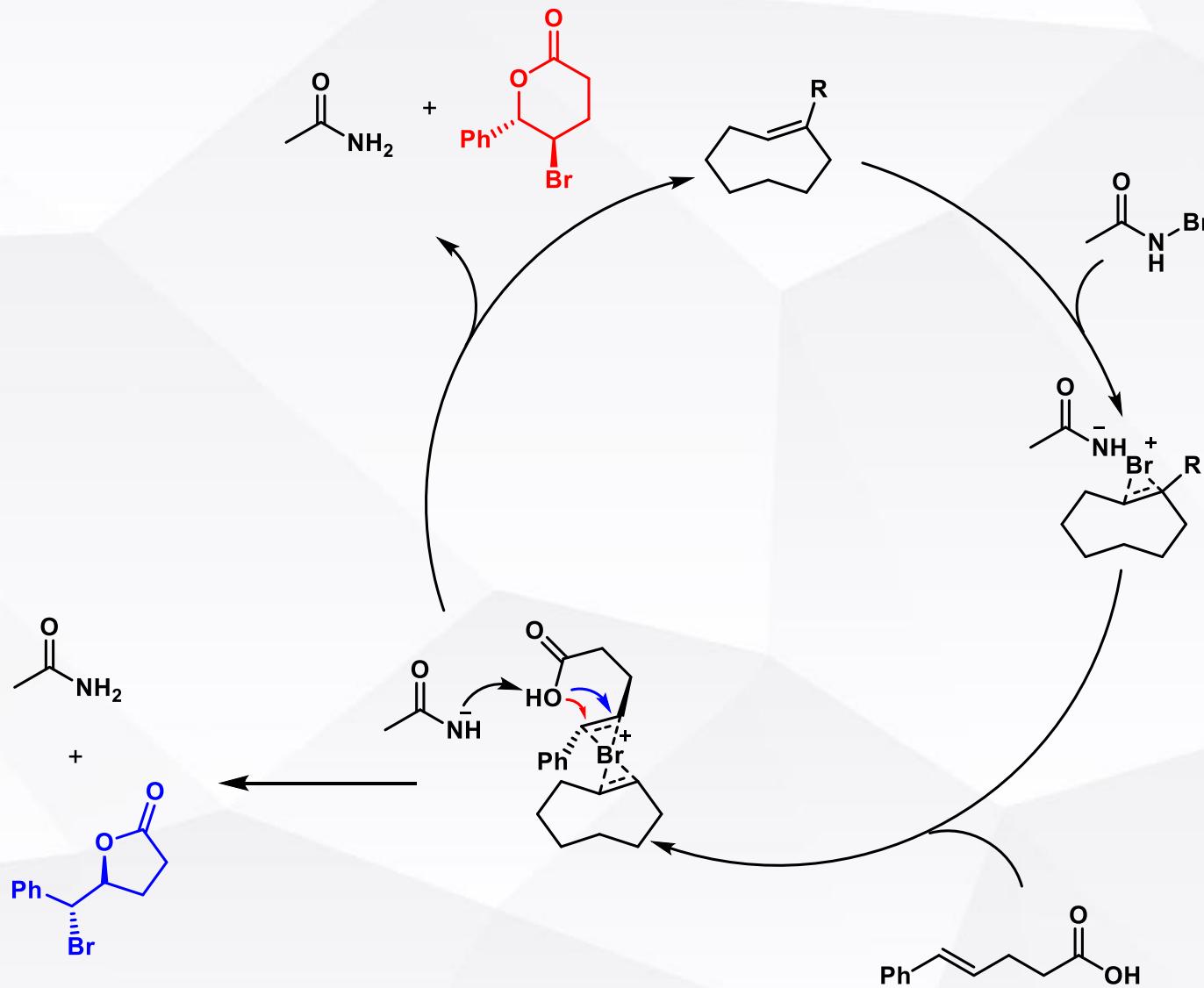


Entry	Cat.	Time [h]	Yield [%]
1	none	6	3
2	none	48	9
3	a	6	2
4	a	48	71
5	b	6	88
6	b	12	99
7	c	6	59
8	d	6	4
9	e	6	1
10	f	6	7
11	g	6	4
12	a+g	6	1

03 Some New Reactions of TCO

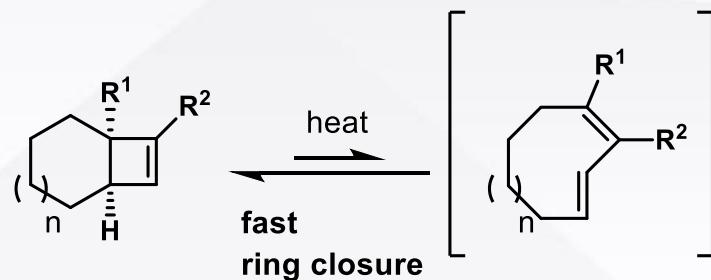


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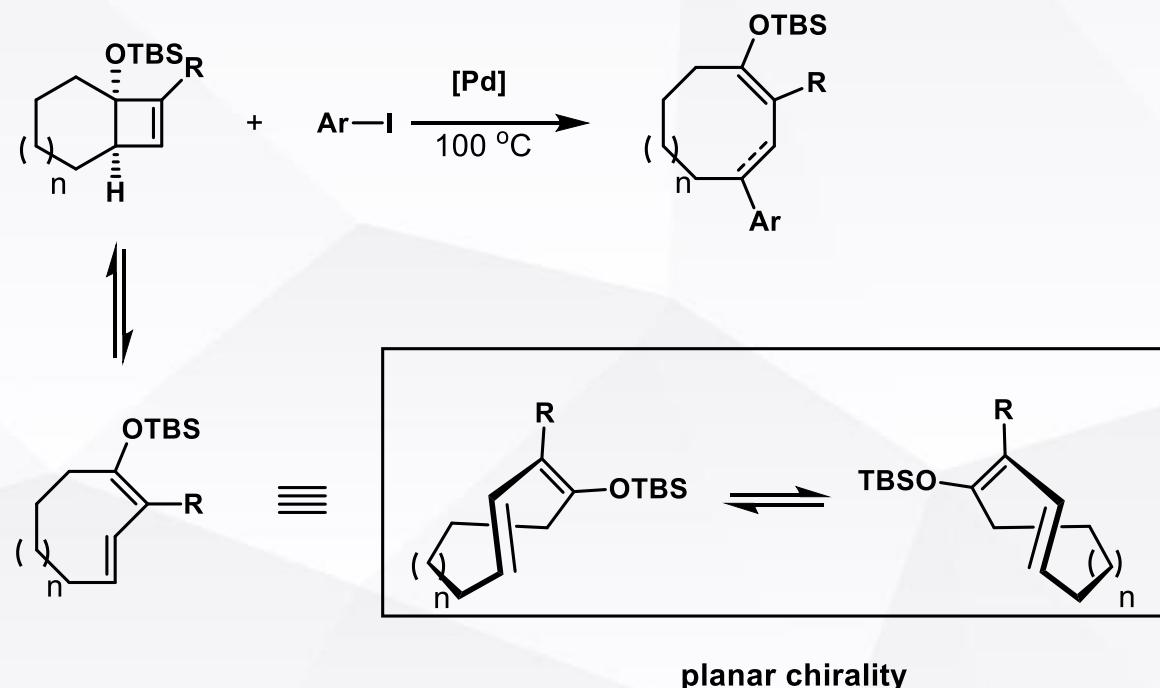


03 Some New Reactions of TCO

In the past

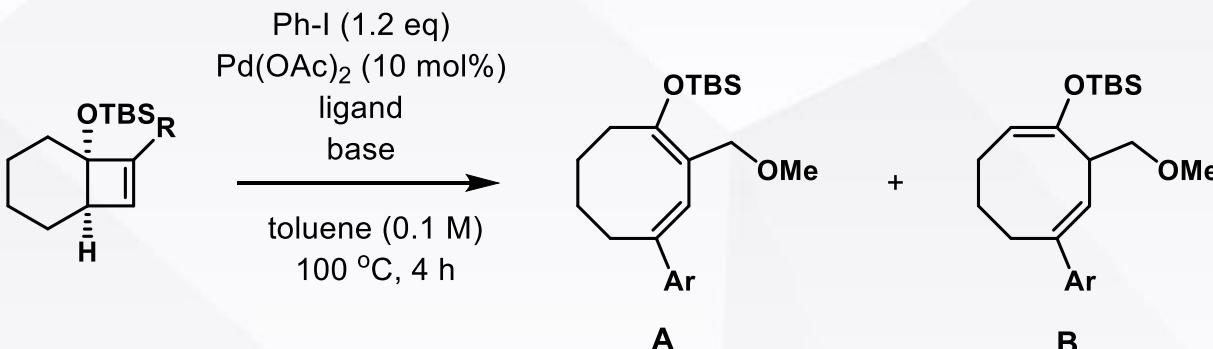


This work



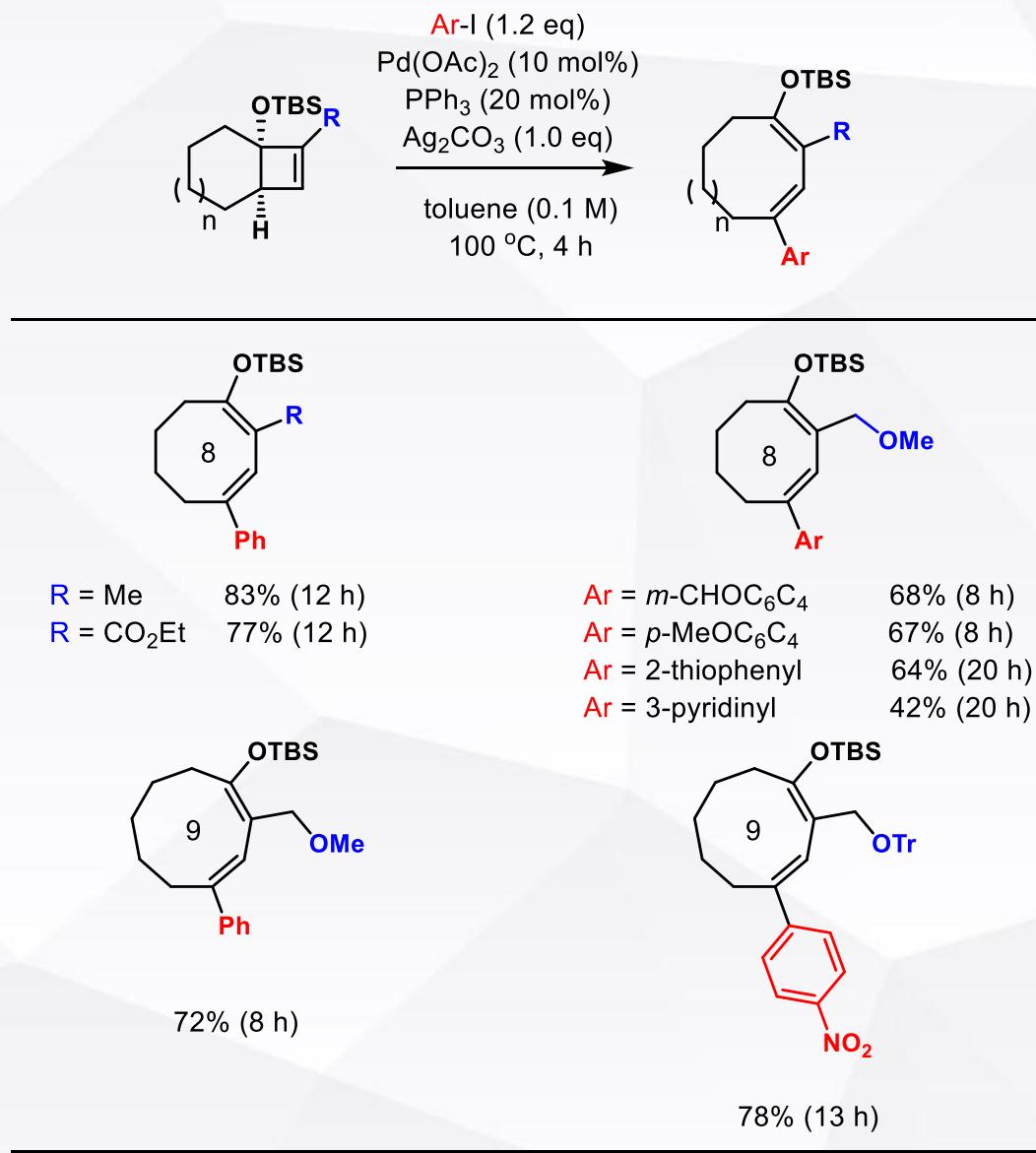
planar chirality

03 Some New Reactions of TCO

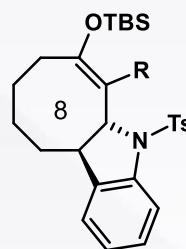
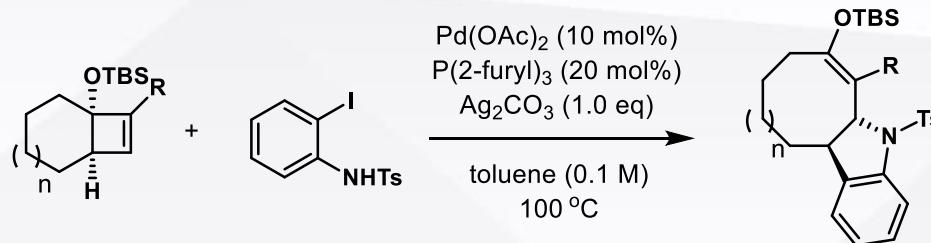


entry	base (eq.)	ligand (mol%)	yield (%)	A:B
1	DIPEA (2.0)	PPh ₃ (20)	34	>20:1
2	CS ₂ CO ₃ (2.0)	PPh ₃ (20)	30	9:1
3	AgOAc (2.0)	PPh ₃ (20)	83	8:1
4	Ag₂CO₃ (1.0)	PPh₃ (20)	90	14:1
5	Ag ₂ CO ₃ (1.0)	PCy ₃ (20)	64	1:2
6	Ag ₂ CO ₃ (1.0)	dppb (10)	61	1:1

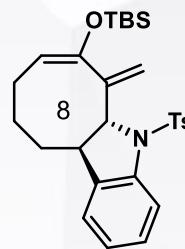
03 Some New Reactions of TCO



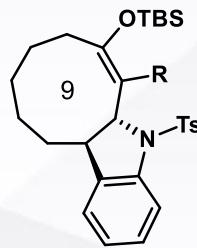
03 Some New Reactions of TCO



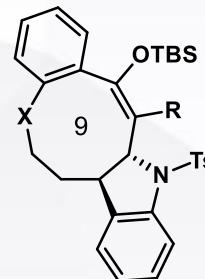
$\text{R} = \text{CO}_2\text{Et}$ 75% (4 h)
 $\text{R} = \text{CH}_2\text{NHNNs}$ 53% (12 h)



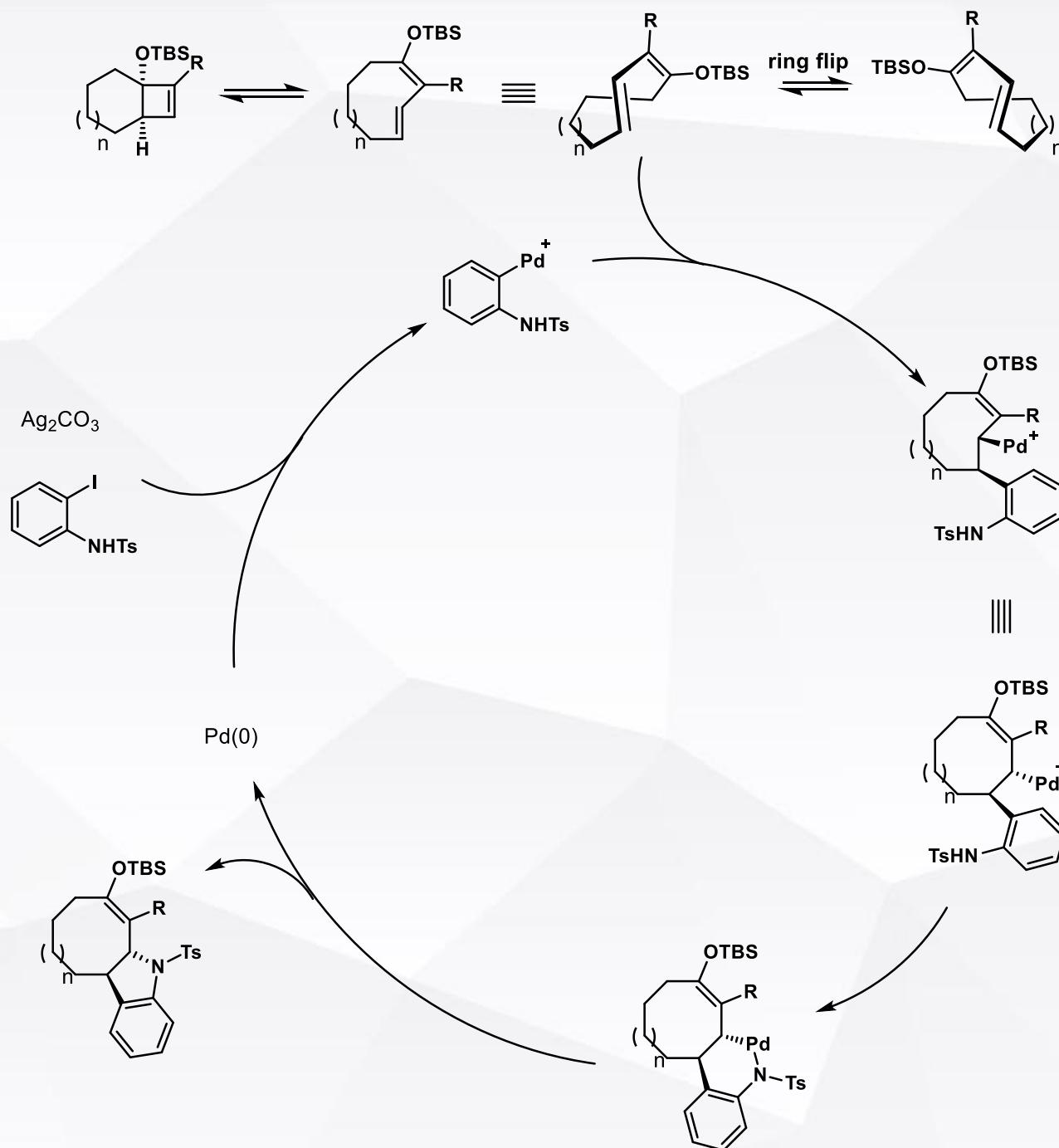
54% (12h)



$\text{R} = \text{CO}_2\text{Et}$ 80% (4 h)
 $\text{R} = \text{CH}_2\text{NHNNs}$ 64% (6 h)

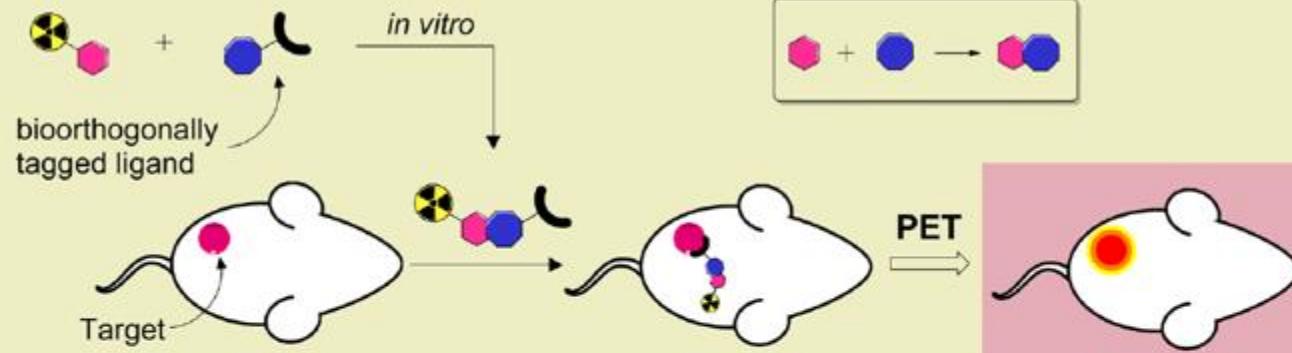


$\text{R} = \text{CO}_2\text{Et}, \text{X} = \text{CH}_2$ 74% (10 min)
 $\text{R} = \text{CH}_2\text{NHNNs}, \text{X} = \text{CH}_2$ 45% (2 h)
 $\text{R} = \text{CO}_2\text{Et}, \text{X} = \text{O}$ 79% (30 min)



03 Some New Reactions of TCO

Rapid Radiolabeling for PET



Requirements for
radiolabeled click agent

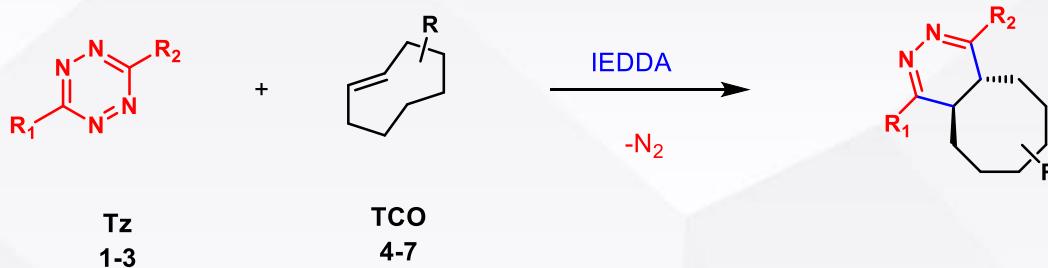
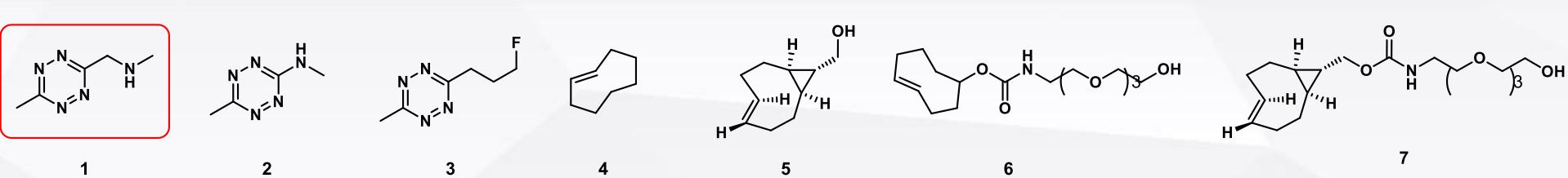
- Fast and/or automated synthesis
- High radiochemical yield
- High purity
- High reactivity

Bioorthogonal PET Imaging



- Homogeneous biodistribution
- High metabolic stability
- Rapid clearance
- High *in vivo* reactivity
- High purity

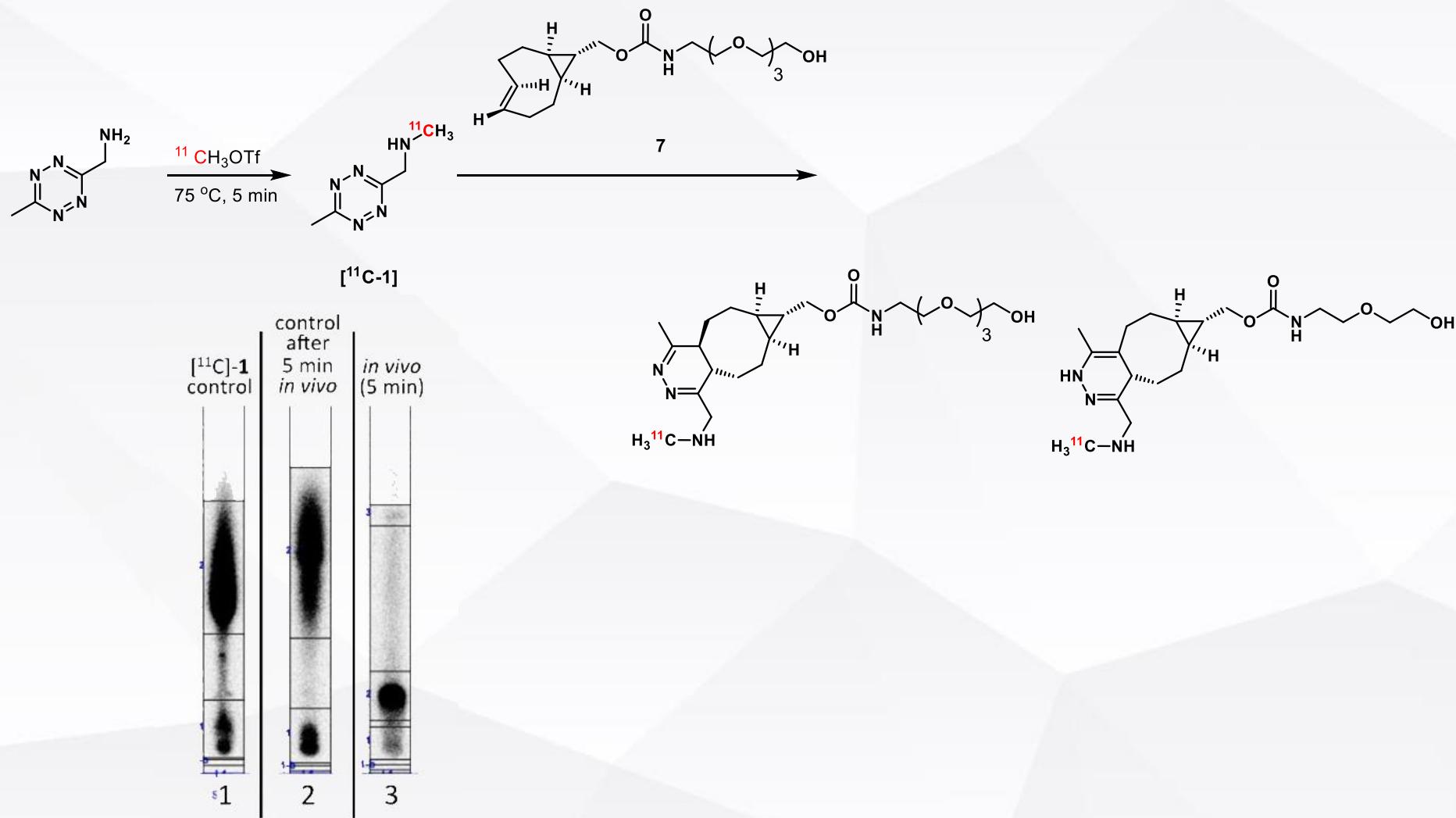
PET: Positron emission tomography



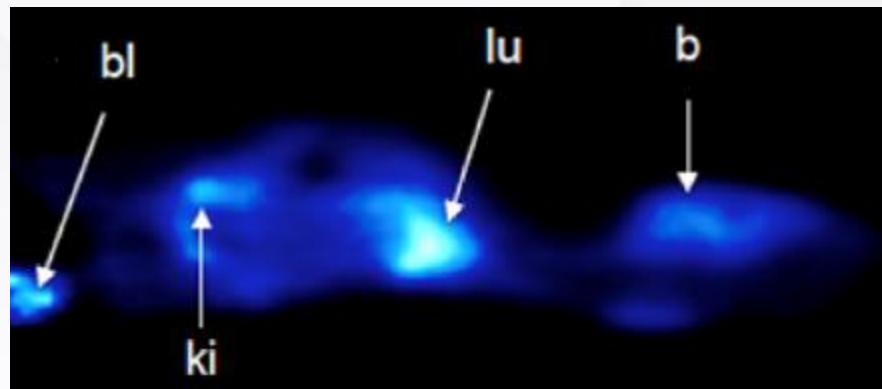
Reaction	k_2 [M ⁻¹ s ⁻¹]	calculated realative rate	realative rate
1+4	1.04 ± 0.03^a	1.32 ^c	0.70 ^c
2+4	n.d.	0.00009 ^c	n.d.
3+4	1.49 ± 0.01^a	1.00 ^c	1.00 ^c
1+5	8.95 ± 1.1^a	1.36 ^d	1.05 ^d
2+5	n.d.	0.0006 ^d	n.d.
3+5	85.5 ± 2.3^a	1.00 ^d	1.00 ^d
1+6	175.4 ± 1.2^b	n.d.	118 ^c
1+7	42500 ± 3100^b	n.d.	497 ^d

^aMeasured in 1,4-dioxane at 37 °C. ^bMeasured in PBS in 37 °C. ^cRelative to the reaction between **3** and **4**. ^dRelative to the reaction between **3** and **5**. n.d. = not determined

03 Some New Reactions of TCO



PET Image



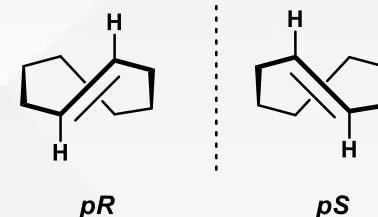
b = brain, bl = bladder, k = kidney, lu = lung

Part 4

Summary

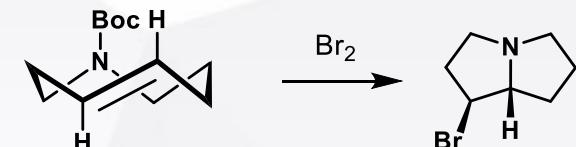
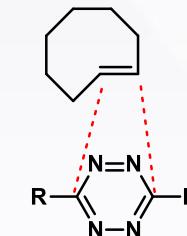
Properties:

- Planar chirality
- High HOMO energy
- High energy of ring strain



Reaction:

- IEDDA reaction
- Click reaction
- Bioorthogonal reaction
- Total synthesis
- The catalyst of transition metal





THANKS